Cover Sheet for

ENVIRONMENTAL CHEMISTRY METHOD

Pestcide Name: Quinclorac

MRID #: 410635-70

Matrix: Water

Analysis: GC/ECD

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BASF CORPORATION CHEMICALS DIVISION Agricultural Chemicals 100 Cherry Hill Road Parsippany, New Jersey 07054

410635- 70

Study Title

Gas chromatographic determination of 3,7-dichloro-8-quinoline carboxylic acid (quinclorac) in drinking water. Method 245

Data Requirements

EPA Guideline Number: 164-2

Author

Dr. F. Mayer

Study Completed On

October, 1985

Performing Laboratory

BASF Aktiengesellschaft Argicultural Research Station Limbergerhof, W. Germany

Registration Document No. BASF:

85/0488

This report consists of 21 pages.

Quinclorac

MRID NUMBER

PR 86-5 DATA CONFIDENTIALITY CLAIM

No claim of confidentiality is made for any information contained in this study on the basis of its falling within the scope of FIFRA 10 (d) (l) (A), (B), or (C).

BASE CORPORATION Company

Agricultural Chemicals Group

Company Agent:

Jack R. Graham Manager, Registration

Date

PR 86-5 GOOD LABORATORY PRACTICE CERTIFICATION

This study is not required to meet the requirements for 40 CFR 160, Good Laboratory Practice Standards. See the control of the

Company <u>EASF CORPORATION</u> Agricultural Chemicals Group

Company Agent

Jack R Graham Manager Registration

BASE AKTIENGESELLSCHAFT LIMBUGERGERHOF AGRICULTURAL RESEARCH STATION

APE/RU Environmental Research October 1985 Method 245

Dr. F. Hayer

Quinclorac	Gas chromatographical determination
Drinking water	

ALL POSITIVE RESULTS OBTAINED WITH THIS METHOD HAVE IN ANY CASE TO BE VERIFIED BY HASS SPECTROMETRY, SINCE IN THIS ULTRA TRACE RANGE CONTRIBUTIONS FROM DIFFERENT INTERFERING SUBSTANCES HAVE TO BE CONSIDERED. SPECIAL CARE SHOULD BE TAKEN TO AVOID CONTAMINATION IN THE ANALYZING LABORATORY ESPECIALLY WHEN THE INVESTIGATED COMPOUNDS AFE OR WERE USED THERE.

1. INTRODUCTION

Chemical name: 3.7-dichloro-8-quinoline carboxylic acid

Structural formula:

Molecular formula: C: 0 Hs Cl: NO:

Molecular weight: 242.05

Solubility: (g substance in 100 g solvent at $20\,\mathrm{°C}$)

Solvent	water	6.2×10^{-3}		
	acetone	0.2		
	acetonitrile	< 0.1		
•	methylenechloride	< 0.1		
•	, ethanol	0.2		
	n-hexane	<0.1		

Derivatization reaction for the GC determination:

$$C1$$
 $C = 0$
 $C = 0$

2. DESCRIPTION OF THE METHOD

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The active ingredient is absorbed from the water on C_{18} — alkylated silica gel, purified on a precoated silica gel column, methylated with diazomethane and determined by gas chromatography using a capillary column and an electron capture detector (63 Ni-ECD).

3. EQUIPMENT

Gilson-minipuls 2 peristaltic pump (Abimed Analysentechnik GmbH, Ludwigshafener Str. 26, 4000 Düsseldorf, FRG)

PH meter (WTW-Werkstätten, D-8120 Weilheim)

Graduated flasks, 100 ml

Full pipettes, 1, 2, 5 and 10 ml

Measuring cylinders, 500 ml

Flasks, 10 ml NS 10/19 (see Annex I for sketch)

Erlenmeyer flasks, 500 ml

Rapid evaporator according to Rettenberger

(N-EVAP; Labotec, Wiesbaden, Gebr. Rettberg, Rudolf-Diesel-Str. 19, 3400 Göttingen, FRG)

Accessory equipment for column clean-up

Baker-10 extraction system (No. 70 18-0) with empty reservoir τ

Adapter (Order No. 7122-0, Baker, Gross-Gerau) Collective rack for 10 ml flasks

Gas chromatograph: e.g. Perkin Elmer F22 with $^{63}\mathrm{Ni-ECD}$

. REAGENTS

Hexane, dist.

Acetone, dist.

Dichloromehtane, dist.

Methanol, dist.

H₂SO₄ 1 M

One-way separating columns, silica gel Si
(Baker, Order No. 70 86-3)

BONDED-PHASE-Octadecyl C 18 (Baker, Order No. 70 31-0)

Empty columns, 3 ml (Baker, Order No. 71 21-3)

Instructions for preparing diazomethane

The following amounts of reagents are required for the preparation of approx. 100 ml ethereal, about 1.5% diazomethane solution in a distillation apparatus with a descending condenser and a dropping funnel: 3 g KOH, dissolved in 5 ml water, diluted with 45 ml methanol, 6 g N-methyl-N-nitroso-P-toluene sulfonamide (diactin) dissolved in 100 ml ether.

3 g KOH dissolved in 50 ml of 90% methanol in the reaction vessel are connected to the apparatus. The receiver vessel cooled with ice and water and filled with 10 ml ether is connected to the descending condenser. 6 g diactin, dissolved in 100 ml ether, are filled into the 250-ml dropping funnel and slowly introduced dropwise into the solution of KOH in water and methanol maintained at 60°C. The whole amount of diactin solution is to be dripped in during about 30 minutes. The mixture of ether and diazomethane is to be distilled off at the same rate in order to avoid an over-concentration of diazomethane in the reaction mixture. Finally, a further 10 ml ether are added via the dropping funnel and distilled off with the remaining diazomethane. Rubber or Latex gloves must always be worn when diactin and diazomethane are being handled. 5 10 -556 3.3

Standard solutions for additional tests
Quinclorac > 99.5% (Dr. Ohnsorge, BASF AG; APE/RU)

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Prepare 100 $\mu g/ml$ in acetone; dilute 1 $\mu g/ml$ and 10 ng/ml with water

Standard solutions for gas chromatography

3,7-Dichloro-8-quinoline carboxylic acid methyl ester > 99.5% (Dr. Ohnsorge, BASF AG, APE/RU)
100 μg/ml; l μg/ml; l0 ng/ml in acetone.

3-Ethyl-7-chloro-8-quinoline carboxylic acid (internal standard) > 99.5% (Dr. Ohnsorge, BASF AG, APE/RU)
500 μg/ml; 5 μg/ml; 50 ng/ml in acetone.

In order to obtain comparable signal levels, the concentration of the internal standard must be five times as great as that of the active ingredient derivative.

-6-

STABILITY OF THE STANDARD SOLUTIONS.

Storage(). Days	inptherlight (a Project				
6.5 3.00 (2.00)	3,7-dichloro-8-quinoline carboxylic acid 200 μg/ml in acetone				
5 71	100% xuli y 1 110%	ਪੂਰ ਫੋ ਰਹੁਣ 100% ਨੂਟ 100%			
	3,7-dichloro-8-quinoli coll 0.2 μg/ml in; ace	ne carboxylic acid methyl ester			
1	· · · · · · · · · · · · · · · · · · ·	λ ⁴ · · · · 100% · · · · _			
1 , 7	100%	100%			
20	100%	100%			
272	100%	100%			
	3-ethyl-7-chloro-8-qu	inoline_carboxylic acid methyl			
	ester in acetone				
17	96%	96%			
36 /	1.03%	109%			
52	. 110%	103%			

Quinclorac is stable in water for 30 days at pH 5, pH 7 and pH 9 at 70°C .

5. ANALYTICAL PROCEDURE

The expressions in brackets used below refer to the symbols used in the calculation formula in 7.2. All pressure specifications are to be understood as the difference to atmospheric pressure. The method is described for one sample. However, 8 samples can be handled in parallel.

5.1. Solid phase extraction

5.1.1. Preparation of the column

l g bonded phase octadecyl is introduced into an empty column with a fritted glass disc, the material is covered with the second fritted disc, and it is placed on a Luer fitting on the cover of the extraction system. A reservoir is fixed on the column by means of an adaptor. 10 ml methanol is introduced in each case in order to wet the \mathbf{C}_{18} material and it is sucked through the column at 20 kPa until it has reached the surface of the column packing. The methanol is displaced by 10 ml water with a pH of 2.5 (adjusted with 1 M $_{2}$ SO $_{\Delta}$).

5.1.2. Enrichment of the active ingredient on the column

500 n water (= G) are adjusted to a pH of 2.5 with 1 M $\rm H_2SO_4$ (about 1.5 ml) in a 500-ml Erlenmeyer flask and degasified for a quarter of an hour with helium. The column prepared according to 5.1.1. is connected up to the tube of the pump and placed in the water with its opening

downward. By gently shaking the tube, the air bubble between water and C_{18} material is displaced. Then 2.5 ml water/min. is drawn through the column with the peristaltic pump. 66

5.1.3. Prewashing of the column

The active ingredient-lader C₁₈ column is replaced on the extraction system and each of the unused openings is closed with a plastic stopper. Air is sucked through the column at 40 kPa for half a minute.

4 ml of a mixture of acetone + hexane = 3 + 97 are drawn through the column by means of the reservoir. The cover of the extraction system is removed together with the columns and the stainless steel rack with the 10-ml flasks (receivers) are placed in the basin. Before the cover is put back on, small amounts of washing liquid that may be hanging on the outlets of the Luer fittings are wiped off.

5.1.4. Elution of the active ingredient

7 ml of a mixture of acetone + hexane = 25 + 75 are applied via the reservoir and sucked through the column at 20 kPa. The eluate in the 10-ml flasks is evaporated to dryness in a stream of nitrogen on the N-EVAP at 40°C. The residue in the ultrasonic bath is dissolved in 1 ml dichloromethane.

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5.2. Purification of extract on silica gel

The extract from 5.1.4. is now applied to a Baker 3-ml silica gei separating column. The flask is rinsed out with 0.5 ml DCM, and the whole solution and then, for 10 sec., air at 20 kPa are sucked through the column.

Preliminary washing is carried out by drawing 4 mi methanol + DCM = 20 + 80 through the column. Then the stainless steel rack with the 1C-ml flasks as receiver is placed in the basin, residues of washing liquid are wiped off the outlets and the cover with the columns is replaced. The active ingredient is eluted with 3 ml methanol + DCM = 40 + 60 (vacuum 20 kPa) and the eluate is evaporated to dryness at 40° C in a stream of nitrogen.

5.3. Methylation

The residue from 5.2. is dissolved in 0.5 ml acetone. 2 ml of ethereal diazomethane solution are added and the whole is left to stand for one hour. The solution must then still be colored yellow; otherwise a further 2 ml diazomethane solution are added and a further hour is maited. Then the solvent is blown off in a stream of nitrogen at 30°C and the residue is dissolved in 0.5 ml of a solution of the internal standard in acetone.

6. GC conditions

Equipment: e.g. Perkin-Elmer, F 22 with 63Ni-

SE54, 19 m MCOT, 0.28 mm i.d., film 1 / 2 / 3 / 3 / 3 / 4 thickness 0.5 μm

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Injector: 260°C of an ord figure 4 country

Oven: 200°C

PAR Detector: 10 % 370°C or such the jack to the as

Carrier gas: He, 1000 mbar

Make-up gas: Ar + CH₁ = 90 + 10; 30 ml/min.

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EVALUATION FOR SUBJECT OF THE PROPERTY OF

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The signals are evaluated via the peak height. In the calibration curve the peak height (in counts or mm) is plotted against the weight of the injected standard.

A calibration curve is prepared by injecting at least three standard amounts. For each series of analyses two samples of water that are free of the active ingredient (blank samples) and two additional tests are subjected to the analytical procedure.

One of the blank samples is taken up in pure acetone (not in solution with the internal standard) for the GC injection. If the chromatogram

of this sample is clean at the retention time of the internal standard, the series of samples is evaluated via the internal standard. For this purpose, instead of the peak heights, the quotients of the peak heights of samples (or additional tests and calibration curve) and internal standard are used. The concentration of the internal standard in the final volume must always be the same within a series of analyses!

For the additional tests water that is free of the active ingredient is mixed with a known amount of quinclorac that is in the same order of magnitude as the expected residues. The yield factor (= f) is determined from the additional tests. The evaluation can also be carried out by means of an appropriate computer program.

7.2. Calculation of the content

The content of quinclorac (R) in µg/kg is calculated by means of the following formula:

$$R = \frac{v_E \cdot W_A \cdot F}{G \cdot v_I}$$

G = sample weight in g

 V_c = final volume of the extract before injection in ml

 V_T = partial volume injected from V_E in μl

F = yield factor determined by investigator through additional tests

 W_A = amount of active ingredient derived from the calibration curve in pg 85/0488 - 0014

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8. YIELD, REPRODUCIBILITY AND LOWEST CONCENTRATION DETERMINED

-Quinclorac in drinking water

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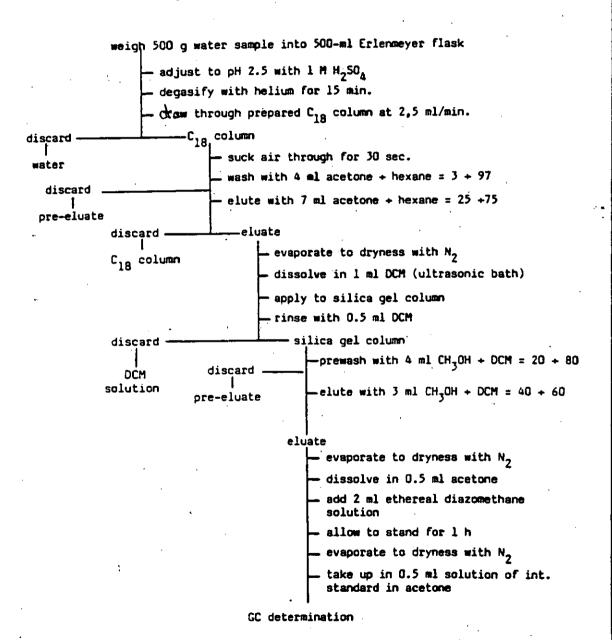
Addition of quinclorac ng/kg (ppt)	Evaluated Yield according to % standard	* "	Standard deviation	Variation coefficient ± %
	internal 86.5; 89.0; 82.1; 84.8; 84.3	85.34	2.58	3.02
10 •	91.1	35.20	(4) 8.90	10.45

* limit of, determination of the process of tagents

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9. ANALYTICAL SCHEME



10. REMARKS

All glass apparatus must be rinsed with acetone before being used for analyses at the limit of determination.

In the event that traces of \mathbb{C}_{18} material go into solution in the elution of the active ingredient, they are removed again by the following silica gel column clean-up.

Technical procedure: H. Sträßner

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Annexes 120 Annexes

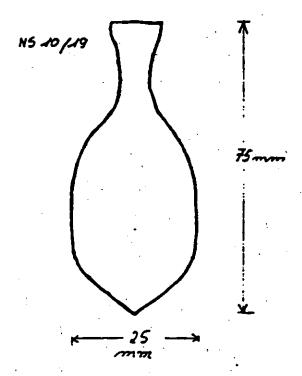
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1. 20-ml flasks for Baker high-grade steel rack

for $\lambda_{i} \in \mathbb{R}^{n}$, $\lambda_{i} \in \mathbb{R}^{n}$. (Cas chromatograms

3./4th/calculation

10-ml flasks for Baker stainless steel rack



1: standard 10 pg 2: standard 5 pg 3: standard 2.5 pg 4: blank sample $V_F = 0.5$ ml, $V_T = 1$ μ l 5: addition of ng/kg $V_F = 0.5$ ml, $V_T = 1$ μ l +: active ingredient derivative =: internal standard; on account of the following negative peak

not suitable here for evaluation

BASF Limburgerhof, -			Evaluation	Environmental Chemistry APE/RU .			
	Queue file name"	:	ST85037.LST	Report number	:	02/01	,
	Investigator	:	STRAESSNER	Date of report	:	30/10/85	07:33:47
	Sample material	:	WATER	Method number	:	245/01/01	٠.
	Culture	:		Injection volume (µ1)	:	1.0	
	Act. ingred./metab.	:	BAS514H	Detection limit (NG/kg):	1.0	
	Internal standard	:		Blank value	:	ST231 .01	
	Derivative determ.	:	QUINCLORAC M.ESTER	Derivat. factor	:	0.945	
	Subst. concerned	:	QUINCLORAC	Conversion factor	:	1.0	
	Eval. accord. to	:	Peak height	Mean recovery	:	82.0	
	Remarks	:		•			, , ,
	Protocol	:					
	Study	:		•			
•	Study section	:	0	C		0	-
	Sample number	:	ST231 .01	ST232 .01	S.	1233 .01	
	Descr. of sample	:	U	Z1 ·	Z	Z	
	Sample weight (g)	:	500.0	500.0		500.0	•
	Aliquot (%)	:	100.0	100.0		100.0	
	Final dilution (ml)	:	0.5	0.5		0.5	*
	Peak area subst.	: ,	2818	12600		14617	•
	Peak height subst.	:	66	402		481	
	Peak area i.std.	:	,	•		4	
	Peak height i.std.	:	•				
	Quotient sub/i.std.	:					
	Residue (µg/kg)	:	< detect. limit				
	Addition (NG)	:		5.0		5.0	*
	Addition (µg/kg)	:		0.0100		0.0100	

Recovery (%)

72.8

91.1

. BASF Limburgerhof	- Evaluation Er	nvironmental Chemistry	, APE/RU
Queue file name	: ST85037.LST	Report number	: 03/01
Investigator	: STRAESSNER	Date of report	: 29/10/85 13:52:04
Sample esterial	: WATER	Method number	: 245/01/01
Culture		Injection volume (µl	
Act. ingred./metab.	BASS14H	Detection limit (NG/	kg): 1.0
Internal standard		Blank value	: ST231 .01
Derivative determ.	QUINCLORAC M.ESTER	Derivat. Factor	: 0.945
Subst. concerned	QUINCLORAC	Conversion factor	1.0
Eval. accord. to	Peak height	Mean recovery	: 88.5
Remarks			
Values identified with	· * lie outside the ca	libration series	
Protocol :	,	•	
Study :			
Study section _ :	o _o	° 0 <u>0</u>	0
•	برين (ST231 ،01 من (ST231)	07074	57,57237 .01
Descr. of sample :	U .	Z3	_Z4
Sample weight (g) ';	500.0	500.0 6.763	500.0
Aliquot (%)	100.0	100.0	100 0
Final dilution (ml) :	0.5 g.S	0.5	100.0 (; ; ;; ;; ;; ;; ;; ;; ;; ;; ;; ;; ;; ;
Peak area subst :	2818	14997	18312*
Peak height subst. :			1
ون ا	66	507	546*
Peak area i.std. :	66	507	• 546•
Peak area i.std. : Peak height i.std. :	66	507	
	66	507	*** 546* ***********************************
Peak height i.std. : Quotient sub/i.std. :	<pre>66 < detect. limit</pre>		
Peak height i.std. : Quotient sub/i.std. : Residue (µg/kg) :	< detect. limit	al et an	*
Peak height i.std. : Quotient sub/i.std. : Residue (µg/kg) :			