# Cover Sheet for

# **ENVIRONMENTAL CHEMISTRY METHOD**

Pestcide Name: Fomesafen

*MRID* #: 450482-02

Matrix: Water

Analysis: HPLC/UV

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Volume 2

# Study Title

Fomesafen: Determination of Fomesafen in Water by Liquid-Solid Extraction and High Performance Liquid Chromatography with Ultraviolet Detection

<u>Data Requirements</u> Series 164

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Study Date April 16, 1999

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Richmond, CA

Report ID TMR0883B Fomesafen: Determination of Fomesafen in Water by Liquid-Solid Extraction and High Performance Liquid Chromatography with Ultraviolet Detection

WRC study number: TMR0883B

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# **Certification of Authenticity**

I, the undersigned, declare that this study was performed under my direction and that this report represents a true and accurate record of the results obtained.

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Approved By:

David R. Dohn Group Leader.

Dietary Exposure Function

Zeneca Inc.

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Report Title: Fomesafen: Determination of Fomesafen in Water by Liquid-Solid

Extraction and High Performance Liquid Chromatography with

Ultraviolet Detection

Authors: Sabrina C. Leung and Stephen L. Schmidt

# 1 Scope

This method is intended for the determination of residues of formesafen in drinking water and surface water sources. The limit of quantitation for this method is 0.1  $\mu$ g/L (ppb) in water. The method has been validated for trace analysis of formesafen at concentrations of 0.1 to 10  $\mu$ g/L in water. The Chemical Abstracts Name for formesafen is 5-[2-chloro-4-(trifluoromethyl)phenoxyl-N-methylsulfonyl)-2-nitrobenzamide (9Cl), CAS No. 72178-02-0. The chemical structure is given below.

# 2 Summary

A measured volume of water of 500 mL is extracted using a C18 liquid-solid extraction disk. Fomesafen residue is eluted from the disk using methanol. The methanol eluate is evaporated to dryness and re-dissolved in acetonitrile/water mobile phase for final determination by high performance liquid chromatography (HPLC) with UV detection at 290 nm.

# 3 Materials/Methods

The equipment and reagents described below were used to generate the data and chromatograms presented in this report. Equipment capable of providing equivalent sensitivity and selectivity, and reagents of comparable purity can be used.

# 3.1 Apparatus

# 3.1.1 Liquid Chromatograph

Hewlett-Packard (HP) Model 1090 equipped with a UV diode array detector (DAD), an automatic injector, 250-µL injection volume option, and a HPLC 3D ChemStation DOS Series data acquisition system. Instrument with the above specifications is available from Hewlett-Packard Company, Wilmington, DE.

# 3.1.2 Injection Volume Modification for HP1090 HPLC (Optional)

To obtain sufficient instrumental sensitivity with detectable HPLC responses from the lowest calibration standard solution, it may be necessary to increase injection volume in excess of 250  $\mu$ L. The following modifications are required to allow injections up to 500  $\mu$ L. Other instruments may require different (or no) modifications.

Replace 250-µL syringe with a 500-µL Hamilton series 1700 gas-tight syringe (Hamilton #81230). The waste sleeve hole for the 250-µL syringe plunger will have to be enlarged to accommodate the larger diameter of the 500-µL syringe plunger (HP 79846-24502). Increase the total volume of the injection loop to at least 500 µL by adding an additional 250-µL sample loop (HP #79846-87613) between the existing loop and the auto-injector needle. Note: It is important to remember that the HP ChemStation software controlling the HPLC does not recognize injection volumes greater than 250 µL. After this modification is made all injection volumes are actually two times (2X) the volume entered in the ChemStation method.

# 3.1.3 Analytical Column

Spherisorb ODS-2, 5  $\mu$ m particle size, 100A pore, 150 mm long x 4.6 mm i.d., Alltech Associates, Inc. Cat No. 8545.

### 3.1.4 Analytical Balance

Sartorius Analytic Model A200S, capable of accurately weighing 0.0001g.

# 3.1.5 Extraction Apparatus

Liquid-solid extraction disks – C18 (Octadecyl) Empore<sup>TM</sup> extraction disk, 47 mm, 3M catalog No. 98050300155.

Vacuum pump – Welch® GEM™ Model 8890A vacuum pump.
Millipore standard filter apparatus – All glass, to hold extraction disk.

# 3.1.6 Vacuum Rotary Evaporator

Buchi Model assemblies "RE-121 Rotovapor" with thermostatically controlled water bath, for evaporating methanol eluate in 50-mL round-bottom flask.

# 3.1.7 Solvent Filtering Unit

Millipore All-Glass Filter Holder for 47 mm diameter filters, catalog No. XX15 047 00, equipped with Millipore 0.45  $\mu$ m pore size HV filter, catalog No. HVLP 04700, for vacuum filtering of HPLC solvents.

#### 3.1.8 Glass Flasks

50-mL capacity, round-bottom, Kimax No. 25285 for collecting the methanol eluate.

# 3.1.9 Glass Bottles

16-oz clear narrow-mouth, with PTFE-lined caps for collecting water samples; 1-, 2-, and 4-oz capacity, narrow-mouth, with PolySeal®-lined screw caps, for storing standard solutions.

# 3.1.10 Pasteur disposable glass transfer pipettes

# 3.1.11 Glass Pipettes

1-, 2-, 5-, and 10-mL disposable, graduated glass pipettes for general use (Kimble).

# 3.1.12 Syringes

10-, 25-, 100-, and 250-μL capacity (Hamilton gas-tight 1700 series) for sample fortifications.

# 3.1.13 pH Indicator Strips

EM Science ColorpHast® strips, 0 to 6 pH range, catalog No. EM-9586-3.

# 3.2 Reagents and Standards

#### 3.2.1 Water

HPLC grade, Fisher Scientific Catalog No. W5-4.

#### 3.2.2 Methanol

HPLC grade or High Purity Solvent (supplied by Burdick & Jackson Catalog No. 230-4).

# 3.2.3 Glacial Acetic Acid

Certified ACS grade, Fisher Scientific Catalog No. A38C-212.

#### 3.2.4 Acetonitrile

HPLC grade, Burdick & Jackson Brand from Baxter, Catalog No. 015-4.

#### 3.2.5 Potassium Nitrate

Analytical Reagent grade, Mallinckrodt Catalog No. 7028.

# 3.2.6 Phosphoric Acid

Reagent grade, 85%.

#### 3.2.7 Sodium Sulfate

Anhydrous, Certified ACS grade.

#### 3.2.8 Preparation of Mobile Phase Solutions

Mobile Phase Solution 'A' (Acetonitrile: water 10:90):

Dissolve 1.01 g of potassium nitrate in 900 mL of HPLC grade water. Adjust to pH 3 with phosphoric acid as indicated by pH sticks. Add 100 mL of HPLC grade acetonitrile. Mix well and vacuum filter through a Millipore 0.45 µm pore size HV filter prior to use.

Mobile Phase Solution 'B' (Acetonitrile: water 90:10):

Dissolve 1.01 g of potassium nitrate in 100 mL of HPLC grade water. Adjust to pH 3 with phosphoric acid as indicated by pH sticks. Add 900 mL of HPLC grade acetonitrile. Mix well and vacuum filter through a Millipore 0.45 µm pore size HV filter prior to use.

Mobile Phase Solution 'C' (Acetonitrile: water 30:70): .

Dissolve 0.700 g of potassium nitrate in 700 mL of HPLC grade water. Adjust to pH 3 with phosphoric acid as indicated by pH sticks. Add 300 mL of HPLC grade acetonitrile. Mix well and vacuum filter through a Millipore 0.45 µm pore size HV filter prior to use.

# 3.2.9 Fomesafen Analytical Reference Standard

Zeneca Analytical Standard ASJ10035-01S, 98.3% w/w purity or equivalent, available from Zeneca Ag Products, 1200 South 47th Street, Richmond, CA 94804-4610.

#### 3.2.10 Calibration Standard Solutions

To prepare a stock calibration solution at a concentration of  $1000 \,\mu\text{g/mL}$ , weigh accurately a known quantity ( $100 \, \text{mg} \pm 2 \, \text{mg}$ ) of primary standard fomesafen of known purity into a clean beaker. Add a sufficient volume of acetonitrile to the beaker to dissolve the fomesafen. Quantitatively transfer the fomesafen solution to a clean  $100 \, \text{mL}$  volumetric flask, and dilute to volume. Stopper the volumetric flask and mix the contents thoroughly. Calculate the concentration of the stock solution as follows:

 $C = W \times P / 100$ 

Where

C = the concentration of formesafen in final solution (mg/mL)

W = the weight of primary standard taken (mg)

P = the purity of the primary standard (e.g. 1.00 for 100.0% w/w purity)

100 = the volume of solvent (mL)

Transfer the contents into a glass bottle. Cap the bottle with a PolySeal®-lined cap, and keep refrigerated when not in use.

To prepare working standard solutions for calibration purposes, dilute the stock calibration solution with mobile phase solution 'C' (Acetonitrile:water 30:70) to give 10, 5, 2, 1, 0.2, 0.1 and 0.05 µg/mL solutions. Transfer working standard solutions to glass bottles with PolySeal®-lined caps and keep refrigerated when not in use.

#### 3.2.11 Fortification Standard Solutions

Fortification standard solutions are to be used for fortification purposes and should be prepared from stock standard solutions different from those used for preparation of calibration solutions. To prepare a stock fortification standard solution at a concentration of 1000  $\mu$ g/mL, weigh accurately a known quantity (100 mg  $\pm$  2 mg) of primary standard fomesafen of known purity into a clean beaker. Add a sufficient volume of acetonitrile to the beaker to dissolve the fomesafen. Quantitatively transfer the fomesafen solution to a clean 100-mL volumetric flask, and dilute to volume. Stopper the volumetric flask, and mix the contents thoroughly. Prepare diluted fortification solutions (100, 10 and 1  $\mu$ g/mL) by diluting appropriate portions of the stock fortification solution with mobile phase solution 'C' (Acetonitrile:water 30:70). Transfer solutions to glass bottles with PolySeal®-lined caps and keep refrigerated when not in use.

# 3.3 Analytical Procedure

# 3.3.1 Preparation of Fortified Samples

Fortified and unfortified control samples are analyzed with each sample set to demonstrate method recovery and performance. Fortify 500-mL aliquot of water samples by adding known volumes of the fortification standard solution of fomesafen (as prepared in Section 3.2.11) to the control samples before extraction. For example, add 50.0  $\mu$ L of the 1.0  $\mu$ g/mL fortification standard solution to the water sample to produce 0.1  $\mu$ g/L (ppb) fortification. Extract the fortified samples as detailed below.

# 3.3.2 Liquid-Solid Extraction of Water Samples

Before sample extraction, precondition the C18 Empore<sup>TM</sup> extraction disk by passing through 10 mL of methanol under vacuum at approximately 10 mL/minute. Remove the vacuum immediately after all the methanol has passed through the filter in order to prevent the filter from drying out. Pass 500 mL of water sample through the filter disk under vacuum at a rate of approximately 100 mL/min. Allow the disk to dry and discard the water.

Elute the C18 Empore<sup>TM</sup> extraction disk under vacuum with 10 mL of methanol. Collect the methanol eluate in a 50-mL round-bottom flask. Transfer the flask to a rotary evaporator and evaporate the sample to dryness with the water bath temperature set at 30°C (temperature not to exceed 40°C). Re-dissolve the residuum in 1.0 mL of Mobile Phase 'C' (Acetonitrile: water 30:70). Final sample to solvent ratio is 500 mL/mL for

water to Mobile Phase 'C'. Use a disposable glass transfer pipette to transfer the sample final extract into an autosampler vial for analysis by HPLC.

#### 3.3.3 HPLC Conditions

	HPLC			
Instrument:	Hewlett-Packard (HP) Model 1090 equipped with diode array detector and an automatic injector			
Column:	Spherisorb ODS-2, 5 µm, 100A pore, 150 mm long x 4.6 mm i.d., from Alltech Associates, Inc., Cat. No. 8545			
Mobile Phase:	<ul> <li>a) Acetonitrile: Water, 10:90, 0.01N KNO<sub>3</sub> to pH 3</li> <li>b) Acetonitrile: Water, 90:10, 0.01N KNO<sub>3</sub> to pH 3</li> </ul>			
Data Acquisition:	HP HPLC 3D ChemStation DOS Series			
Flow Rate:  Column Temperature:	0.75 mL/min 65% Mobile Phase (A) 35% Mobile Phase (B) Time Table: Time (min) %B 0.00 35 7.00 100 7.50 100 8.00 35 11.00 35			
Detector Wavelength:	290 nm			
Injection Volume: Sampling Interval:	120 - 300 μL 0.640 sec			
Run Time:	11 min			

Using the above conditions, the elution time for fomesafen was 7.2 to 7.3 minutes. See Figures 1 to 5 for typical chromatograms.

# 3.3.4 Calibration

Before starting to calibrate the chromatographic system, first made one or two injections of the 10  $\mu$ g/mL standard solution to condition the inlet and column. Calibrate the liquid chromatograph with the daily-use calibration standard solutions (Section 3.2.10). Inject the entire range of solutions, from 0.05  $\mu$ g/mL to 5  $\mu$ g/mL, at the beginning and at the end of each run. After every 6 to 8 samples, inject one or more of the calibration standards to assure that the fomesafen response is stable.

# 3.3.5 Analysis of Sample Extracts

Analyze the final mobile phase extract from each water sample on the same day of calibration. Inject the sample extracts using the same conditions and injection volumes as those used for the calibration standards. The identity of the fomesafen peak in the sample chromatogram is assigned based upon the coincidence of the retention time ( $\pm$  0.10 minute) with that of the fomesafen peak in the calibration standard chromatogram. Dilute the extract with Mobile Phase 'C', if necessary, to keep the fomesafen response within the calibration range.

#### 3.4 Calculations

The concentration of fomesafen in the original sample is calculated by using the external standard method, i.e., the response obtained for fomesafen in the sample extract is compared to the response obtained from a separate injection of fomesafen calibration solution. To use the linear response calculation method shown below, the injection volumes for all calibration solutions and sample extracts must be fixed at the same volume.

#### 3.4.1 Calibration Response Factor

Calculate the response factor, RF, for injection of a calibration solution as follows:

RF = — Rstd

Where

Cstd = concentration in  $\mu$ g/mL of the calibration solution

Rstd = response units (e.g., peak height, peak area, electronic units) from detector for the calibration solution

#### 3.4.2 Fomesafen in Sample

Determine the concentration of fomesafen in the original sample, Cs (in  $\mu$ g/L), from the average response factor, RFavg, and the sample response, Rsample, as follows:

Cs ( $\mu$ g/L) = Rsample x RFavg x D x  $10^3$ 

Where

Rsample = response unit from detector for the sample final extract

RFavg = average response factor over the entire range of calibration

C = Concentration of sample in final extract (sample to solvent ratio, in

mL/mL) = 500 mL/mL for water

D = dilution factor required if final extract is diluted to keep in calibration range

# 4 Results/Discussion

# 4.1 Precision and Accuracy

Fortified water samples were prepared as described under Section 3.3.1 and analyzed for fomesafen according to this method to establish recoveries. Recoveries from 9 drinking water samples and 3 surface water samples fortified at levels of 0.1  $\mu$ g/L to 10  $\mu$ g/L are shown in Table 1. Recoveries ranged from 91 to 133% with a mean recovery of 101% and a coefficient of variation (CV) of 11%.

# 4.2 Interferences/Matrix Effects

Control tap and river water samples were extracted and taken through the entire analytical procedure. No interferences/matrix effects have been observed.

# 4.3 Confirmatory Techniques

If it is necessary to confirm the identities of the peaks assigned as fomesafen, the confirmation can be achieved using an LC column of different polarity. The untreated control should be analyzed to demonstrate freedom from matrix interferences for this technique to be applicable.

#### 4.4 Limit of Quantitation

The limit of quantitation (LOQ) was assessed by carrying out recovery experiments at low fortification levels. In this laboratory, the LOQ has been established at  $0.1 \mu g/L$  for water.

# 4.5 Limit of Determination

The limit of determination (LOD) has not been fully established. The LOD was calculated using the standard deviation from the 0.1  $\mu$ g/L recovery results. The LOD was estimated as three times the standard deviation (3s) of the results of the analysis of six samples. In this method validation study, the calculated statistics support an LOD of 0.05  $\mu$ g/L.

# 4.6 Time Required for Analysis

The analysis can be completed by one person in one 8-hour work day. The procedures within this method can be stopped at any point, excluding the liquid-solid extraction step. The final extracts can be retained at room temperature overnight.

# 4.7 Quality Assurance

Method blanks of water should be extracted and analyzed along with each set of samples to demonstrate freedom from interferences. Where true control samples are unavailable, use untreated water collected from a neighboring area believed to be free of fomesafen

residues, or use deionized water. Fortified samples should be prepared as described in Section 3.3.1 and analyzed with each sample set.

# 4.8 Safety Considerations

Personnel untrained in the routine safe handling of chemicals must not attempt to use this procedure. Information on any first aid procedures can be found in the Material Safety Data Sheets accompanying the chemical or available from the chemical supplier. In general, always wear safety glasses with side shields. Work in well ventilated areas. Avoid inhaling particulates, aerosols and/or vapors; and avoid contact of the chemicals with skin and clothing. Keep flammable solvents away from potential sources of ignition.

#### 4.9 Method Notes

To avoid cross contamination between samples during extraction, it is important that all extraction and concentration glassware be thoroughly decontaminated and cleaned prior to its use for water samples. Clean all glassware with HPLC grade acetone and subsequently with HPLC grade methanol. Be sure to also decontaminate the coarse-frit glass base support of the extraction apparatus by vacuum rinsing it with a minimum of 10 mL of HPLC grade methanol.

Water sample extraction procedure was adapted from an earlier residue method developed for analyzing fluazifop and fomesafen in water (Reference 1). LC conditions were adapted from an earlier method developed for analyzing fomesafen in soil (Reference 2).

# 5 Tables and Figures

Table 1.	Recoveries of Fomesafen from Water	•
Figure 1	Typical liquid observators for for form	

Figure 1. Typical liquid chromatogram for fomesafen analysis (0.05 µg/mLcalibration standard solution)

Figure 2. Typical liquid chromatogram for fomesafen analysis (control tap water)
Figure 3. Typical liquid chromatogram for fomesafen analysis (tap water sample fortified at 1.0 µg/L)

Figure 4. Typical liquid chromatogram for fomesafen analysis (control river water)

Typical liquid chromatogram for fomesafen analysis (control river water sample fortified at 0.1 µg/L)

Table 1. Recoveries of Fornesafen from Water

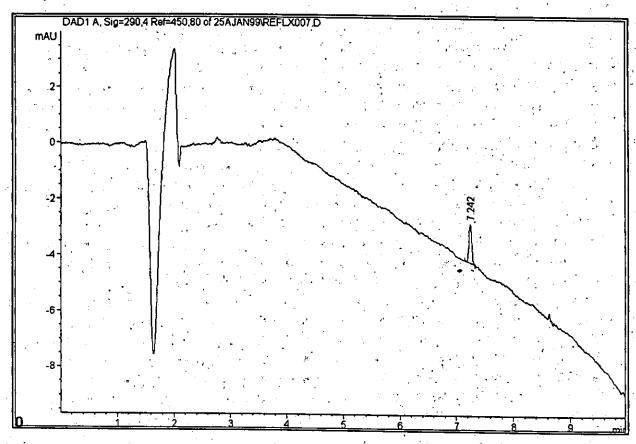
Sample ID (WRC #16592-)	Matrix Type	Fomesafen µg/L Added	Fomesafen µg/L Found	Fomesafen % Recovery
47-1 (control)	Drinking water	0.0	0.0	
48-1 (control)	Surface water	0.0	0.0	
47-2 (F0.1A)	Drinking water	0.1	0.0965	97
47-3 (F0.1B)	Drinking water	0.1	0.0922	92
47-4 (F0.1C)	Drinking water	0.1	0.1057	106
48-2 (f0.1-a)	Surface water	0.1	0.1333	133
48-3 (f0.1-b)	Surface water	0.1	0.1056	106
48-4 (f0.1C)	Surface water	0.1	0.0913	91
	•		• • •	<u>L</u> .
47-5 (F1.0A)	Drinking water	1.0	0.9675	97
47-6 (F1.0B)	Drinking water	1.0	0.9470	95
47-7 (F1.0C)	Drinking water	1.0	1.0358	104
47-8 (F10.0A)	Drinking water	10.0	9.613	96
47-9 (F10.0B)	Drinking water	10.0	10.05	101
47-10 (F10.0C)	Drinking water	10.0	9.800	98
Mean Recovery for a	ll fortifications ± CV	= . ,	A A V	101 ± 11%

Notes: 1.  $s = standard deviation of the 6 analyses at the 0.1 <math>\mu g/L$  fortification level  $= 0.0155 \ \mu g/L$ 

# 2. Sources of test water matrix:

Drinking water: tap water taken from laboratory at WRC, Richmond, CA Surface water: river water collected from the American River, CA, logged in reference as WRC 16441-22-1

Figure 1. Typical liquid chromatogram for fomesafen analysis (0.05  $\mu g/mL$  calibration standard solution)



(Retention time of fomesafen = 7.24 min.)

Figure 2. Typical liquid chromatogram for fomesafen analysis (control tap water)

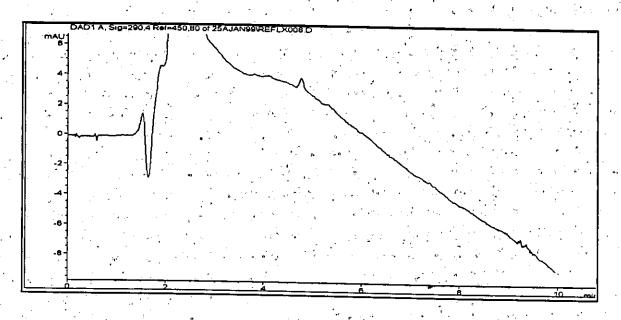


Figure 3. Typical liquid chromatogram for fomesafen analysis (tap water sample fortified at 1.0 μg/L)

(Retention time of fomesafen = 7.26 min.)

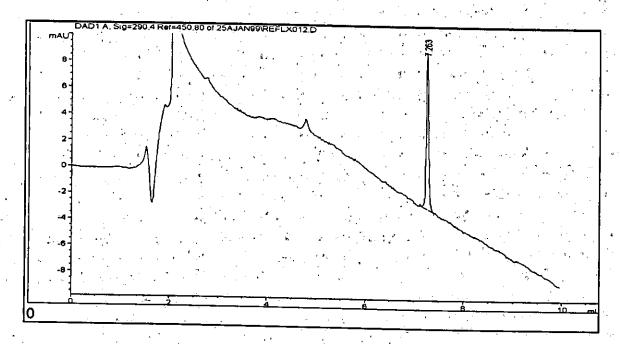


Figure 4. Typical liquid chromatogram for fomesafen analysis (control river water)

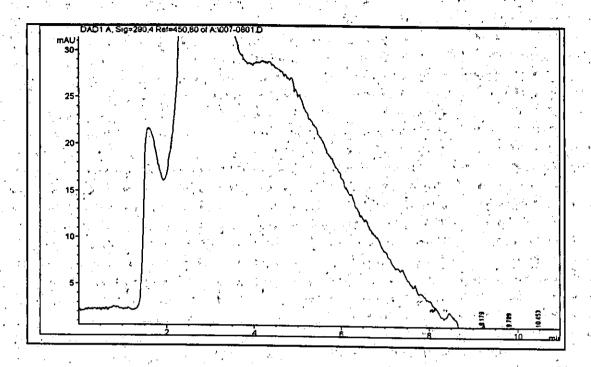
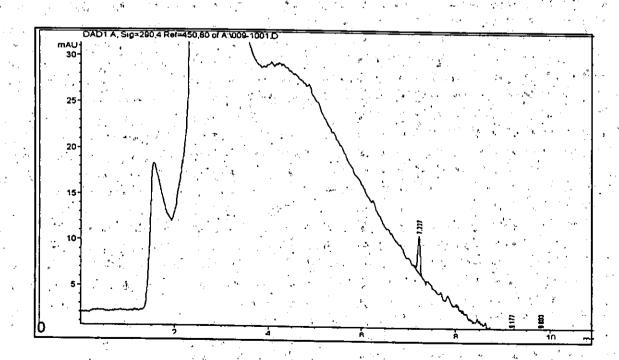


Figure 5. Typical liquid chromatogram for fomesafen analysis (river water sample fortified at 0.1 µg/L) (Retention time of fomesafen = 7.23 min.)



# 6 References

- 1. French, D.A. "The simultaneous Determination of Residues of Fluzziflop and Fomesafen in Water" Zeneca Agrochemicals Standard Operating Procedure (SOP) Residue Analytical Method (RAM) 208/03 (1994).
- 2. Leung, S.C. "Fomesafen: Determination of Fomesafen in Soil or Water (WRC-97-110)" Zeneca Ag Products TMR0741B (1997).
- 3. Raw data referenced in WRC Laboratory Notebook 16592-47 to -50.