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Data Set

Existing Chemical CAS No. Molecular Formula CAS Name	: ID: 77-86-1 : 77-86-1 : C4H11NO3 : Tris (hydroxymethyl) aminomethane
Producer related part Company Creation date	: The Dow Chemical Company : 11.11.2006
Substance related part Company Creation date	: The Dow Chemical Company : 07.07.2009
Status Memo	:
Printing date Revision date Date of last update	: 07.07.2009 : : 07.07.2009
Number of pages	: 71
Chapter (profile) Reliability (profile) Flags (profile)	 Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 Reliability: without reliability, 1, 2, 3, 4 Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1. General Inform	nation	ld 77-86-1 Date
1.0.1 APPLICANT AN	ID COMPANY INFORMATION	
Type Name Contact person Date Street Town Country Phone Telefax Telex Cedex Email Homepage	 manufacturer Dow Chemical, TERC Dr. William T. Stott 27.08.2006 1803 Building 48674 Midland, MI United States 	
Source 30.08.2006	: Dow Chemical, TERC Midland, MI	
1.0.2 LOCATION OF	PRODUCTION SITE, IMPORTER OR FORMUL	ATOR

Name of plant Street Town Country Phone Telefax Telex Cedex Email Homepage		Dow Chemical Sterlington, Louisiana United States	
Source Reliability 13.11.2006	:	Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data.	(1)
Type Name of plant Street Town Country Phone Telefax Telex Cedex Email Homepage		manufacturer Dow Chemical Ibbenburen Germany	
Source Reliability 13.11.2006	:	Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data.	(1)

1. General Information	n Id 77-86-1 Date	
1.0.3 IDENTITY OF RECIPIE	NTS	
1.0.4 DETAILS ON CATEGO	RY/TEMPLATE	
1.1.0 SUBSTANCE IDENTIF	CATION	
IUPAC Name:Smiles Code:Molecular formula:Molecular weight:Petrol class:	2-Amino-2-hydroxymethyl-1,3-propanediol OCC(N)(CO)CO C4H11NO3 121.14	
Source:Reliability:09.11.2006	Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data.	(2)
1.1.1 GENERAL SUBSTANC	EINFORMATION	
Purity type:Substance type:Physical status:Purity:Colour:Odour:	typical for marketed substance organic solid White Odorless	
Reliability : 13.11.2006	(2) valid with restrictions Data from handbook or collection of data.	(3)
1.1.2 SPECTRA		
Type of spectra : Result :	NMR 13-Carbon NMR spectra of TRIS were measured between 407 and 461	K.
Reliability :	 (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment. 	(4)
1.2 SYNONYMS AND TRA	DENAMES	
Talatrol		
Source : Reliability : 19.03.2004	Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data.	(2)
ТНАМ		
Source :	Dow Chemical, TERC Midland, MI 3 / 71	

1. General Inform	mation	ld 77-86-1 Date 07.07.2009
Reliability	: (2) valid with restrictions	
19.03.2004		(2)
Trimethlol aminom	ethane	
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	
19.03.2004		(2)
TRIS		
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	
19.03.2004		(2)
TRIS AMINO ®		
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data 	
10.11.2006		(2)
Tris Buffer		
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	
19.03.2004		(2)
Tris(hydroxymethy	I)aminomethane	
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	
19.03.2004		(2)
Tris-steril		
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	
19.03.2004		(2)
Trisamine		
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data 	
19.03.2004		(2)
Trizma		
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data 	
19.03.2004		(2)
Trometamol		

1. General Informat	ion Id 77-86-1 Date	
Source Reliability 19.03.2004	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	(2)
Tromethane Source Reliability 09.11.2006	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	(2)
1.3 IMPURITIES Purity CAS-No EC-No EINECS-Name Molecular formula Value	: typical for marketed substance	
Remark Source Reliability 09.11.2006	 The compound is sold as 100% crystal or a 40% aqueous solution of tris(hydroxymethyl)aminomethane. MSDS of the Dow Chemical Company. 2003. Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	
1.4 ADDITIVES		
Purity type CAS-No EC-No EINECS-Name Molecular formula Value Function of additive	 typical for marketed substance 7732-18-5 231-791-2 water H2O ca. 40 % v/v Solvent 	
Source Reliability 09.11.2006	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	(1)
1.5 TOTAL QUANTITY		
1.6.1 LABELLING		
1.6.2 CLASSIFICATION		
1.6.3 PACKAGING		

1. General Informa	ition	ld	77-86-1
		Date	07.07.2009
1.7 USE PATTERN			
Type of use Category	 use other: Emulsifier for cosmetics, mineral of dressings, textiles, cleaners, pharmaceut buffer 	il and wax em icals, and che	ulsions, leather mical intermediate
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 		
10.11.2006			(1) (2)
1.7.1 DETAILED USE P	ATTERN		
1.7.2 METHODS OF MA	NUFACTURE		
Result Source	 May be prepared by reduction or catalytic nitro compound. Preparation by electrolyt 2,485,982 (1949 TO COMM SOLVENTS Hazardous Substances Data Bank. (2) volid with restrictions 	c hydrogenatic tic reduction: CORP), CA 4	n of corresponding McMillan, US Patent 4, 1836B (1950)
Reliability	Data from handbook or collection of data.		
13.11.2006			(5)
1.8 REGULATORY MI	EASURES		
1.8.1 OCCUPATIONAL	EXPOSURE LIMIT VALUES		
1.8.2 ACCEPTABLE RE	SIDUES LEVELS		
1.8.3 WATER POLLUTI	NC		
1.8.4 MAJOR ACCIDEN	T HAZARDS		
1.8.5 AIR POLLUTION			
1.8.6 LISTINGS E.G. CH	EMICAL INVENTORIES		
1.9.1 DEGRADATION/T	RANSFORMATION PRODUCTS		
-			
i ype CAS-No	: degradation product		
EC-No EINECS-Name	<u>.</u>		

I. General Informa	Date	
IUCLID Chapter	:	
Remark Source Reliability	 THAM is stable at room temperature for periods as long as 12 years. Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, well-documented and percenter tenders. 	
09.11.2006	acceptable for assessment.	(6)
1.9.2 COMPONENTS		
1.10 SOURCE OF EXPO	DSURE	
Source of exposure Exposure to the	Human: exposure by productionSubstance	
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	
09.11.2006		(1)
Source of exposure Exposure to the	Human: exposure through intended useSubstance	
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	
09.11.2006		(1)
Source of exposure Exposure to the	Human: exposure of the consumer/bystanderSubstance	
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from bandback or collection of data 	
09.11.2006	Data from handbook of collection of data.	(1)
1.11 ADDITIONAL REM	ARKS	
1.12 LAST LITERATUR	E SEARCH	
Type of search Chapters covered Date of search	 Internal and External 3, 4, 5 03.11.2006 	
13.11.2006		
1.13 REVIEWS		

2.1 MELTING POINT

Value Sublimation Method Year GLP Test substance	ca. 171 - 172 °C 1 1989 2 as prescribed by 1.1 - 1.4	
Source Reliability 19.03.2004	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	(2)
Value Sublimation Method Year GLP Test substance	ca. 171 - 172 °C 1940 as prescribed by 1.1 - 1.4	
Source Reliability 24.03.2004	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment. 	(7)
2.2 BOILING POINT		
Value	: ca. 219 - 220 °C at 101.33 hPa	
Remark Source Reliability 28.06.2004	 @ 10mmHg Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 	(2)
Value	: ca. 219 - 220 °C at	
Source Reliability 24.03.2004	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment. 	(7)
2.3 DENSITY		
2.3.1 GRANULOMETRY		
2.4 VAPOUR PRESSUR	₹E	
Value Decomposition	: .0000027 hPa at 20 °C : 8 / 71	

. Physico-Chemic	Id 77-86-1 Date 14.11.2006
Method	: ASTM method E1719-97
Year	: 2008
GLP	: No
Test substance	:
Source	: Dow Chemical, Analytical Sciences, South Charleston, WV
Reliability	: (1) valid without restriction
28.06.2004	
2.5 PARTITION COEF	FICIENT
Partition coefficient	- octanol-water
Log pow	= -2.31 at 20 °C
pH value	
Method	: OECD Guide-line 107 "Partition Coefficient (n-octanol/water), Flask-
	shaking Method"
Year	: 1996
GLP	: no data
lest substance	: as prescribed by 1.1 - 1.4
Conclusion	: The calculated partition coefficient is consistent with high water solubility and which by definition would be indicative of low log Kow values.
Reliability	: (1) valid without restriction
13.11.2006	Comparable to a guideline study.
Partition coefficient	: octanol-water
Log pow	: <= -3.8 at 20 °C
pH value	
Method	: OECD Guide-line 107 "Partition Coefficient (n-octanol/water), Flask-
Voar	- 1007
GLP	: no data
Test substance	: other TS:AMPD
Test substance	: AMPD (2-Amino-2-methyl-1,3-propanediol) CAS# 000115-69-5. Moleci
	weight = 105.14
13.11.2006	
Partition coefficient	- octanol-water
Log pow	: ca2.22 at $^{\circ}$ C
pH value	: =7
Method	: other (calculated)
Year	: 2004
GLP Test substance	: as prescribed by 1.1 - 1.4
Source	Dow Chemical TERC Midland MI
Test condition	: Model Input Parameters:
	· · · · · · · · · · · · · · · · · · ·
	Data Temperature (°C) = 25
	Chemical Type = 1
	Molecular Mass (g/mol) = 121.14
	Water Solubility (g/m3) = 550,000 Vapor Pressure $@25^{\circ} C (Pa) = 3.0 \times 10.4$
	Water Solubility $(g/m3) = 550,000$ Vapor Pressure @ 25° C (Pa) = 3.0 x 10-4 Melting Point (C) = 171.5
	Water Solubility $(g/m3) = 550,000$ Vapor Pressure @ 25° C (Pa) = 3.0 x 10-4 Melting Point (C) = 171.5 Estimated Henry's Law Constant (H) (Pa m3/mol) = 6.7 x 10-8
	Water Solubility $(g/m3) = 550,000$ Vapor Pressure @ 25° C (Pa) = 3.0 x 10-4 Melting Point (C) = 171.5 Estimated Henry's Law Constant (H) (Pa m3/mol) = 6.7 x 10-8 Log Kow (Octanol-Water Partition Coefficient) = -2.22 / 1.38

Conclusion : This material has very high water solubility, very low vapor pressure, and very low log Kow. In the absence of advective and reactive processes, these physical properties dictate that the material will partition exclusively to the water compartment at equilibrium. Reliability : (2) valid with restrictions Accepted calculation method. (1) 2.6.1 Solubility in : Water Value : ca. 800 gJ at 25 °C pH value : at °C remperature effects : Examine different pol. in pKa : 6.03 at 25 °C : Description : : in pKa : 6.03 at 25 °C : pKa : 8.03 at 25 °C : Description : : in pKa : 6.03 at 25 °C : Description : : in the absence of advective and reactive processes, these physical properties different pol. : pH value : ca. 550 other.mg/mL at 25 °C Temperature effects : : in the absence of advective and reactive physical properties different pol. : pKa : at 25 °C Description : at 25 °	2. Physico-Chemic	al Dala	Date 14.11.2006
Reliability :: (2) valid with restrictions Accepted calculation method. (1: 2.6.1 SOLUBILITY IN DIFFERENT MEDIA (1: Solubility in :: Water value :: ca. 800 g/ at 25 °C pH value (1: concentration :: at °C remporature effects :: is at °C (1: PKa :: 8.03 at 25 °C : pExamine different pol. : pKa : PKa :: 8.03 at 25 °C : : Description :: : is stable :: : : : Solubility in : Water Value :: : :: :: :: :: :: ::: ::: ::: ::: ::	Conclusion	: This material has very high water sol very low log Kow. In the absence of these physical properties dictate that to the water compartment at equilibri	lubility, very low vapor pressure, and advective and reactive processes, the material will partition exclusively
07.04.2004 (1: 2.6.1 SOLUBILITY IN DIFFERENT MEDIA Solubility in : water Value : c.a. 800 g/l at 25 °C pH value : concentration : at °C Temperature effects : : pKa : 8.03 at 25 °C Description :: : stable : : Reliability : (2) valid with restrictions Data from handbook or collection of data. : 13.11.2006 : (1: Solubility in : Water Value : ca. 550 other.mg/mL at 25 °C pH value : ca. 10.4 concentration : 1 other.Molar at 25 °C pKa : at 25 °C Description : : itelability : 1 other.Molar at 25 °C Description : : : itelability : 1 other.Molar at 25 °C Description : :	Reliability	: (2) valid with restrictions Accepted calculation method	un.
 2.6.1 SOLUBILITY IN DIFFERENT MEDIA Solubility in : Water Value : ca. 800 g/l at 25 °C pH value : concentration : at °C Temperature effects : Examine different pol. : pKa : 8.03 at 25 °C Description : : Stable : : Reliability : (2) valid with restrictions Data from handbook or collection of data. 13.11.2006 (11) Solubility in : Water Value : ca. 550 other.mg/mL at 25 °C pH value : ca. 550 other.mg/mL at 25 °C pH value : ca. 550 other.mg/mL at 25 °C pH value : ca. 550 other.mg/mL at 25 °C pH value : ca. 10.4 concentration : .1 other.Molar at 25 °C pKa : at 25 °C Description : Stable : Paramine different pol. : pKa : at 25 °C Description : Stable : Paramine different pol. : pKa : at 25 °C Description : Stable : Paramine different pol. : pKa : at 25 °C Description : Stable : Paramine different pol. : pKa : at 25 °C Description : Stable : Paramine : Paramine : Paramine : Stable : Paramine : P	07.04.2004		(11
Solubility in : Water Value : ca. 800 g/l at 25 °C pH value : concentration : at °C Temperature effects : : Examine different pol. : : pKa : : : Reliability : (2) valid with restrictions : Data from handbook or collection of data. : : 13.11.2006 : : : Solubility in : : : : Yalue : : : : : Yalue : : : : : : Yalue : : : : : : : : : : : : : : : <td>2.6.1 SOLUBILITY IN DI</td> <td>FERENT MEDIA</td> <td></td>	2.6.1 SOLUBILITY IN DI	FERENT MEDIA	
Value : ca. 800 g/l at 25 °C pH value : concentration : at °C Temperature effects : Examine different pol. : pKa : 8.03 at 25 °C Description : Stable : Reliability : (2) valid with restrictions Data from handbook or collection of data. (12 Solubility in : Water Value : ca. 550 other.mg/mL at 25 °C PH value : ca. 550 other.mg/mL at 25 °C pH value : ca. 550 other.mg/mL at 25 °C pH value : ca. 10.4 concentration : 1 other.Molar at 25 °C pExamine different pol. : pKa : at 25 °C Description : : Stable : : PKa : at 25 °C Description : : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midand, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. : 19.03.2004 :	Solubility in	• Water	
intermediate intermediate intermediate intermediate concentration intermediate intermediate intermediate image: transform of the state intermediate intermediate intermediate pKa intermediate intermediate intermediate intermediate pKa intermediate intermediate intermediate intermediate pKa intermediate intermediate intermediate intermediate intermediate pKa intermediate intermediate intermediate intermediate intermediate pKa intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate intermediate inte	Value	$r_{1} = r_{2} = r_{1} = r_{2} = r_{1} = r_{2} = r_{1} = r_{2} = r_{1} = r_{1$	
intermediation at °C Temperature effects : Examine different pol. : pKa : Reliability : Reliability : (2) valid with restrictions Data from handbook or collection of data. 13.11.2006 (12 Solubility in : Value : concentration : : 1 other:Molar at 25 °C pH value : ca. 550 other:mg/mL at 25 °C pH value : : : : : : : : : : : : : : : : : : : : : : : : : : : : : : : : : : : : :	nH value	. ca. 000 g/rat 20 0	
Temperature effects : a. ° ° pKa : 8.03 at 25 °C Description : : Stable : : Reliability : (2) valid with restrictions Data from handbook or collection of data. 13.11.2006 : : (12 Solubility in : Water (2) valid with restrictions Value : ca. 550 other:mg/mL at 25 °C (12 pH value : ca. 10.4 (12 concentration : 1 other:Molar at 25 °C (12 pKa : : : : peg. product : : : : itsble : : : : itsble : : : : : itsble : : : : : 'Year :	concentration	at °C	
Examine different pol. : pKa : 8.03 at 25 °C Description : : Reliability : (2) valid with restrictions Data from handbook or collection of data. (12 Solubility in : Water Value : ca. 550 other:mg/mL at 25 °C pH value : ca. 10.4 concentration : 1. other:Molar at 25 °C pKa : at 25 °C Description : . pKa : at 25 °C Description : . pKa : at 25 °C Description : . istable : . istable : . gLP : . Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, Mil Reliability : (2) valid with restrictions Data from handbook or collection of data. . 19.03.2004 : .	Temperature effects		
Description 2.0.3 at 25 °C Description 2.2 Reliability 2.2 Feliability 2.2 Feliability 2.2 FLAMMABILITY 2.9 FLAMMABILITY	Examine different not		
Description : Instant of the second sec	pKa	8.03 at 25 °C	
Stable : Reliability : Reliability : (12) Valid with restrictions Data from handbook or collection of data. 13.11.2006 Solubility in : Ualue : ca. 550 other:mg/mL at 25 °C pH value : concentration : i. 1 other:Molar at 25 °C pKa : pKa : pgKa : i. : peg. product : Method : Year : Test substance : i. : You'd with restrictions : Data from handbook or collection of data. 19.03.2004 : 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Description	. 0.00 0 20 0	
Reliability : (2) valid with restrictions Data from handbook or collection of data. (12) 13.11.2006 : (12) Solubility in : Water value : (12) Solubility in : Water value : (12) Yalue : : (12) (12) PH value : : (12) (12) Concentration : 1 other:Molar at 25 °C (12) PKa : : (12) (12) pKa : : (12) (12) pKa : : (12) (12) (12) pKa : : (12) (12) (12) pKa : : : (12) (12) (12) Stable :	Stable		
Reliability : (2) valid with restrictions Data from handbook or collection of data. (12) 13.11.2006 (12) Solubility in : Water Ca. 550 other:mg/mL at 25 °C (12) Value : ca. 550 other:mg/mL at 25 °C (12) PH value : ca. 10.4 (12) concentration : 1 other:Molar at 25 °C (12) PKa : at 25 °C (12) Description : (12) Stable : (12) Year : 1989 (12) GLP : (12) Test substance : as prescribed by 1.1 - 1.4 (12) Source : Dow Chemical, TERC Midland, MI (12) Reliability : (2) valid with restrictions Data from handbook or collection of data. (12) 19.03.2004 : (2) valid with restrictions (2) 2.6.2 SURFACE TENSION : (2) 2.8 AUTO FLAMMABILITY : (2) 2.9 FLAMMABILITY : (2)			
13.11.2006 (12) Solubility in :: Water Value : ca. 550 other:mg/mL at 25 °C pH value : ca. 10.4 concentration : 1 other:Molar at 25 °C Temperature effects : ifferent pol. pKa : at 25 °C Description : ifferent pol. Stable : : Peg. product : : Year : 1989 GLP : : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. : 19.03.2004 : : 2.7 FLASH POINT : 2.8 AUTO FLAMMABILITY : 2.9 FLAMMABILITY :	Reliability	: (2) valid with restrictions	data
Solubility in : Water Yalue : ca. 550 other:mg/mL at 25 °C pH value : ca. 10.4 concentration : 1 other:Molar at 25 °C Temperature effects : i pKa : at 25 °C Description : : Stable : : Perform : : Year : 1989 GLP : : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. : 19.03.2004 : : 2.7 FLASH POINT : 2.8 AUTO FLAMMABILITY : 2.9 FLAMMABILITY :	13.11.2006	Data norm handbook of collection of c	uala. (12
Value : ca. 550 other:mg/mL at 25 °C pH value : ca. 10.4 concentration : 1 other:Molar at 25 °C Temperature effects : Examine different pol. : pKa : at 25 °C Description : Stable : Perence : 1 989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source :: Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. : 19.03.2004 : 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Solubility in	· Water	
<pre>pH value : ca. 10.4 concentration : .1 other:Molar at 25 °C Temperature effects : Examine different pol. : pKa : at 25 °C Description : Stable : Deg. product : Method : Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 (2 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY</pre>	Value	c_{a} 550 other mg/mL at 25 °C	
intervention intervention interventinterventinterist interventervention	nH value	: ca 10.4	
Temperature efforts : if outer historia at 25 °C Temperature efforent pol. : pKa : at 25 °C Description : Stable : Deg. product : Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. : 19.03.2004 : (2) valid with restrictions 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	concentration	1 other: Molar at 25 °C	
Examine different pol. : pKa : at 25 °C Description : Stable : Deg. product : Method : Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 (2 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Temperature effects		
pKa : at 25 °C Description : Stable : Deg. product : Method : Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Examine different pol.	:	
Description : Stable : Deg. product : Method : Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 2.6.2 SURFACE TENSION 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	рКа	: at 25 °C	
Stable : Deg. product : Method : Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Description	:	
Deg. product : Method : Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Stable	:	
Method : Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Deg. product	:	
Year : 1989 GLP : Test substance : as prescribed by 1.1 - 1.4 Source :: Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. (2 19.03.2004 (2 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Method	:	
GLP : Test substance : Source : Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY	Year	: 1989	
lest substance : as prescribed by 1.1 - 1.4 Source : Dow Chemical, TERC Midland, MI Reliability : (2) valid with restrictions Data from handbook or collection of data. (2) 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	GLP	:	
Source Reliability : Dow Chemical, TERC Midland, MI : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with restrictions Data from handbook or collection of data. : (2) valid with re	lest substance	: as prescribed by 1.1 - 1.4	
Reliability : (2) valid with restrictions Data from handbook or collection of data. 19.03.2004 (4 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Source	: Dow Chemical, TERC Midland, MI	
Data from handbook or collection of data. 19.03.2004 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	Reliability	: (2) valid with restrictions	
19.03.2004 (: 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	-	Data from handbook or collection of o	data.
 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY 	19.03.2004		(2
 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY 	2.6.2 SURFACE TENSIO	Ν	
 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY 			
2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY	2.7 FLASH POINT		
2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY			
2.9 FLAMMABILITY	2.8 AUTO FLAMMABIL	ΙΤΥ	
	2.9 FLAMMABILITY		

Γ

2. Physico-Chemic	al Data	ld Date	77-86-1 14.11.2006
2.10 EXPLOSIVE PROF	PERTIES		
2.11 OXIDIZING PROPE	ERTIES		
2.12 DISSOCIATION CO	DNSTANT : 8.21		
Method	:		
Year GLP	- 1989		
Test substance	:		
Source Reliability 28.06.2004	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Data from handbook or collection of data. 		(13)
2.13 VISCOSITY			

2.14 ADDITIONAL REMARKS

Memo	: pKa= 8.3 @ 20°C; 8.03 @ 25°C; 7.8 @ 37°C
Source Reliability	 L. Troester (2006) ANGUS data. (2) valid with restrictions Data from handbook or collection of data
13.11.2006	

3. Environmental Fate and Pathways

3.1.1 PHOTODEGRADATION

3.1.2 STABILITY IN WATER

Туре	: abiotic
t1/2 pH4	: at °C
t1/2 pH7	: at °C
t1/2 pH9	: at °C
Deg. product	:
Method	:
Year	:
GLP	: no data
Test substance	: as prescribed by 1.1 - 1.4
Conclusion	: TRIS AMINO is stable for several years at ambient temperatures according to ANGUS files.
Reliability	: (2) valid with restrictions
2	Data from handbook or colletion of data.
13.11.2006	(14)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type Media Air Water Soil Sediment	 Fugacity n water – air and use p <0.1% 100 % <0.1% <0.1 	nodel level I r (assumes atterns)	II water is emissi	on route, bas	sed on physical properties
Method	: other: calc University	ulated Cana , Peterborou	adian Environm Igh, Ontario, Ca	iental Modeli anada.	ng Centre, Trent
rear	: 2004				
Result	: Level III data	a is reported	as follows:		
	<u>Scenario</u>	<u>% to Air</u>	<u>% to Water</u>	<u>% to Soil</u>	% to Sediment
	1000kg/hr to Air	<0.1%	59.4%	40.6%	<0.1%
	1000kg/hr to Water	<0.1%	100.0%	<0.1%	0.1%
	1000kg/hr	<0.1%	58.3%	41.7%	<0.1%
		12 /	71		

3. Environmental	Fate and Pathways	Id 77-86-1 Date 14.11.2006
	to Soil	
	1000kg/hr <0.1% 68.6% 31 simultaneously to air, water, soil.	.4% <0.1%
Conclusion	: Tris(hydroxymethyl)aminomethane has hi pressure, and low log Kow. The substance adsorption to soil or sediments, and a low or soil to the atmosphere. If released to a hydroxyl radicals. If released directly to w route based on physical properties and us will remain in the water compartment and released to soil, the substance is expected	igh water solubility, a low vapor ce has a low potential for <i>i</i> potential to volatilize from water air, the substance will react with vater, the most probable emission se patterns, most of the substance is expected to be biodegraded. I ed to be biodegraded.
Source Test condition	: Dow Chemical, TERC Midland, MI : Input Parameters for Level III Model:	
	Data Temperature (°C) = 25 Chemical Type = Type 1 indicates chemic environmental compartments Molecular Mass (g/mol) = 121.14 Water Solubility (g/m3) = 800,000 Vapor Pressure @ 25° C (Pa) = 3.0×10^{-10} Melting Point (C) = 171.5 Estimated Henry's Law Constant (H) (Pa Log Kow Octanol-Water Partition Coeffic Simulated Emission Rate (kg/hr)= 1,000 Simulated Environment = Level III Default	cal can partition into all -4 m3/mol) = 4.54 x 10-8 tient = -2.3 t environment
	Reaction Half-lives (hr.) Input to Level III I Air (vapor phase) = 3.8 Water (no susp. solids) = 3600^{*} Soil = 7200^{*} Sediment = 7200^{*} Suspended Sediment = $1.0 \times 10^{11**}$ Fish = $1.0 \times 10^{11**}$ Aerosol = $1.0 \times 10^{11**}$	Model
	*Half-lives extrapolated based on inheren according to Technical Guidance Docume consistent with predicted biodegradability **Default value used in Level III model wh negligible in this compartment.	t biodegradability classification, ent of the European Commission, based on BIOWIN program. nen reaction is expected to be
Reliability	: (2) valid with restrictions	
09.11.2006	Acceptable calculation method.	(15
3.3.2 DISTRIBUTION		
3.4 MODE OF DEGF	ADATION IN ACTUAL USE	
3.5 BIODEGRADAT	ION	
Туре	: aerobic	

3. Environmental Fate and Pathways

Id 77-86-1

Date

Deg. product Method Year GLP Test substance Results	 OECD Guide-line 301 D "Ready Biodegradability: Closed Bottle Test" 1990 no data as prescribed by 1.1 - 1.4 No biodegradation measured after 28 days at test substance concentrations of 8.7 and 13 mg/L. :
Conclusion Reliability 13.11.2006	 TRIS AMINO is not readily biodegradable. (1) valid without restriction Meets national standard methods (AFNOR/DIN).
Type Inoculum Concentration	 aerobic activated sludge, domestic, non-adapted 2 mg/l related to Test substance related to
Contact time Degradation Result Deg. product Method Year GLP	: ca. 40 (±) % after 28 day(s) other: not readily biodegradable OECD Guide-line 301 D "Ready Biodegradability: Closed Bottle Test" 1989 no
Test substance	 Other: AMP The study followed OECD Guide-line 301 D "Ready Biodegradability: Closed Bottle Test".
	AMP was dissolved in mineral nutrient medium at a concentration of 2 mg/L. The mineral medium had been prepared from aerated de-ionized water amended with nutrient salts and trace element stock solutions. The reaction mixtures were inoculated with a few drops of inoculum from the discharge of a community wastewater treatment plant. Corresponding control mixtures were prepared with the nutrient medium; nutrient medium with inoculum; and nutrient medium with inoculum and the reference substance sodium acetate to correct for background oxygen consumption and confirm the viability of the microbial inoculum. Equal volumes of the various reaction mixtures were dispensed into replicate bottles, sealed and incubated in the dark for 28 days at 20-21 °C.
	The dissolved oxygen concentration in duplicate samples was measured with an oxygen electrode on days 0, 5, 15, and 28. The percent biodegradation of the test substance was determined using the following formula:
	Dt = (mg of BODx/L) o 100 (mg of substance/L) o TOD
Result	 Dt = percent biodegradation of the test substance BODx = biological oxygen demand after x days. TOD = Theoretical oxygen consumption The results for replicate samples, expressed as percent biodegradation, are as follows:
	AMP: 5 days (0.1%, 0.1%); 15 days (1.9%, 1.9%); 28 days (40.7%, 38.9%) Average biodegradation after 28 days was 40%.
	Sodium acetate: 5 days (51.5%, 51.5%); 15 days (70.9%, 58.0%); 28 days (77.3%, 83.6%)
	14 / / 1

3. En	vironmental Fa	ate and Pathways	ld Date	77-86-1 14.11.2006	
Reli	ability	 (2) valid with restrictions 2 (meets generally accepted scientific standards acceptable for assessment) 	, well-c	locumented, and	l
01.0	06.2006				(17)
3.6	BOD5, COD OR BOD	D5/COD RATIO			
3.7	BIOACCUMULATION	N			
3.8	ADDITIONAL REMA	RKS			

4. Ecotoxicity

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type Species Exposure period Unit Method Year GLP Test substance	 other: aerated and static were both tested other:29 species were tested 1958 as prescribed by 1.1 - 1.4
Method Result	 Fish were exposed to the buffer for 2 days, then placed in untreated water and observed for 30 days. Fresh-water species were left in the buffer solutions for 9 days. The vessels were tested under static and aerated conditions separately. Fish were monitored for signs of toxicity. Increases in weight occurred in all species. The pH decline over the 9 day fresh-water test was 0.02 pH units. One specimen of chiclid (Aquidens portalegrensis) died during the third day, likely die to starvation since none were fed until the fourth day. There was no other mortality in any other fish throughout the study period.
Source Test condition	 The only observations of toxicity noted were to specimens of the opaleye (Girella nigricans), which exibited a marked interruption of opercular movements after 5 hours in the TRIS buffer. Irregularities appeared in solutions ranging from 5-20g TRIS per gallon water. The irregularities were characterized by two or three large opercular beats, followed by a pause lasting 5-10 seconds. Total beats varied from 8-20 per minute. The upset in opercular rhythm did not seem to have an adverse effect upon the fish. They appeared normal in all other aspects. Dow Chemical, TERC Midland, MI The following marine species were tested: Heterodontus francisci Holocentris microstomus Myripristis murdjan Pseudopeneus bifasciatus Amphiprion percula Lepidaplois bilunulatus Iniistius pavo Girella nigricans Kuhlia marginatus Hippocampus punctulatus Hepatus olivaceus Hepatus bariene Microcenthis strigatus Zebrasoma flavescens Zanclus canescens Heterostichus rostratis Leptocottus armatus Clinocottus analis Balistes vidua
Conclusion	 The following fresh water species were tested: Lebistes reticulatus Mollienesia sp. Aequidens portalegrensis Cichlasoma nigrofasciatus Tests indicate that 29 species of fish can stand high concentrations of buffer (20g / gallon) dissolved in transport water. Light dosages (2-5g / gallon) are adequate to stabilize pH during transport of fish, regardless of

the presence of aaration (provided the weight of fich transported does not exceed 25-50g / gallon in a closed system). The data indicate that TRIS buffer is nontoxic to the 29 species tested under these conditions. Buffered solutions are stable for up to three months at normal air temperatures. Reliability (2) valid with restrictions 00 11 2006 (18) Type : static (18) Species : static (18) CSO : 96 hour(s) (18) Type : static (18) Species : 1997 (18) CSO : > 10000 measured/nominal (18) Method : OECD Guide-line 203 "Fish, Acute Toxicity Test" Year Year : 1997 GLP : 10000 Test substance : ofter TS/AMPD Result : Hour LCSO (mgL) Z4 > 10000 28 > 10000 29 Test substance : OHHT DS/AMPD (19) (19) Type :: Subcole ine 203 "Fish, Acute Toxicity Test" (19) Test substance : obto(r(s) : 10000 (19) Test substance : Exposure period : 06 hour(rs) (19) <	4. Ecotoxicity	ld 77-86-1 Date 14.11.2006
Reliability Euffered solutions are stable for up to three months at normal air temperatures. 09:11.2006 (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment. 09:11.2006 (18) Type :: Species :: Exposure period :: Byperiod :: System period :: Byperiod :: System period :: System period :: Byperiod :: System period :: CLP :: Test substance :: Conclusion :: Test substance :: Conclusion :: Type :: System period :: System period <t< td=""><td></td><td>the presence of aeration (provided the weight of fish transported does not exceed 25-50g / gallon in a closed system). The data indicate that TRIS buffer is nontoxic to the 29 species tested under these conditions.</td></t<>		the presence of aeration (provided the weight of fish transported does not exceed 25-50g / gallon in a closed system). The data indicate that TRIS buffer is nontoxic to the 29 species tested under these conditions.
Méets generally accepted scientific standards, well-documented and acceptable for assessment. (18) Type : : static Sposure period : 96 hour(s) Unit : mg/ LCS0 : :> 10000 measured/nominal Method :: OECD Guide-line 203 'Fish, Acute Toxicity Test" Year : :1997 GLP :: Test substance : other TS:AMPD Result : Hour LCS0 (mg/L) 24 > 10000 72 > 10000 73 > 10000 74 > 10000 75 > 10000 76 > 10000 76 > 10000 77 > 10000 78 > 10000 79 > 10000 79 > 10000 70 = 10000 70 = 10000 70 = 10000 70 = 10000 71 = 10000 72 > 10000 72 > 10000 73 > 10000 74 = 10000 75 = 10000 75 = 10000 76 = 10000 76 = 10000 76 = 10000 77 = 10000 77 = 10000 70 = 100000 70 = 1000000 70 = 10000000 70 = 10000000000 70 = 1000000000000000	Reliability	Buffered solutions are stable for up to three months at normal air temperatures. (2) valid with restrictions
$\begin{array}{rcl} 09.112006 \\ 0.112006 \\ 0.112006 \\ 0.112006 \\ 0.112006 \\ 0.112006 \\ 0.112006 \\ 0.112006 \\ 0.112000 \\ 0$		Meets generally accepted scientific standards, well-documented and acceptable for assessment.
Type:staticSpecies:Erachydanio rerio (Fish, fresh water)Exposure period:96 hour(s)Unit:mg/LC50:> 10000Method:OECD Guide-line 203 'Fish, Acute Toxicity Test'Year:1997GLP::Test substance:other TS:AMPDResult:HourLC50 (mg/L)24> 1000072> 1000096> 1000097::14.11.2006:(19)Type::Species:Leuciscus idus (Fish, fresh water)Species:Leuciscus idus (Fish, fresh water)Species:Leuciscus idus (Fish, fresh water)Species:Leuciscus idus (Fish, fresh water)Species:1990GLP::Test substance:as prescribed by 1.1 - 1.4Source:IWL labs 1990Conclusion:The 96-h LCS0 in golden orfe is >10,000 mg/L.Reliability::(2) valid with restrictions Data from handbook or collection of data.13.11.2006:Year:Year:Species::Pleuronectes platessa (Fish, marine)Species:::Species:::::::::::<	09.11.2006	. (18)
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	Type	: static
Exposure period : 96 hour(s) Unit : mg/ LC50 : > 10000 measured/nominal Method : OECD Guide-line 203 'Fish, Acute Toxicity Test" Yoar : 1997 GLP : Test substance : other TS:AMPD Result : Hour LC50 (mg/L) 24 >10000 72 >10000 72 >10000 73 >10000 Test substance : AMPD: 2-Amino-2-methyl-1,3-propanedial Conclusion : The acute LC50 96-hour for Brachydania rerio is > 10,000 mg/L for AMPD. Reliability : (2) valid with restrictions Guideline study with acceptable restrictions. 14.11.2006 (19) Type : Species : Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) Unit : mg/ LC50 : > 10000 adueted Method : Year : 1990 GLP : Type : sensistatic Source : IWL labs 1990 Conclusion : The 96-h LCS0 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions GLP : Type : sensistatic Species : Leuciscus idus (Fish, marine) Exposure period : 96 hour(s) Unit : mg/ LC50 : > 10000 calculated Method : Year : 1990 GLP : Type : sensistatic Species : Pleuronectes platessa (Fish, marine) Exposure period : 96 hour(s) Unit : mg/ LC50 : = 184 calculated Method : Year : 1983 GLP : Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed moratily, at a 95% Fiducial CI: 24 hour LC50 = 121 mg/L (13-340 mg/L) 47 hour LC50 = 144 mg/L (31-340 mg/L) 27 tour LC50 = 144 mg/L (31-340 mg/L)	Species	: Brachydanio rerio (Fish, fresh water)
Unit : mg/l LCS0 : > 10000 measured/nominal Method : OECD Guide-line 203 "Fish, Acute Toxicity Test" Year : 1997 GLP : Test substance : other TS:AMPD Result : Hour LCS0 (mg/L) 24 > 10000 48 > 10000 72 > 10000 96 > 10000 Test substance : AMPD:2-Aminoy-1,3-propanediol Conclusion : The acute LC50 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD. Reliability : (2) valid with restrictions Guideline study with acceptable restrictions. 14.11.2006 (19) Type : Species : Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) Unit : mg/l LCS0 : > 10000 calculated Method : Year : 1990 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : IWL labs 1990 Conclusion : The 96-h LCS0 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions Data from handbook or collection of data. 13.11.2006 (20) Type : semistalic Species : Pleuronectes platesa (Fish, marine) Exposure period : 96 hour(s) Unit : mg/l LCS0 : 1942 (20) Type : semistalic Species : Pleuronectes platesa (Fish, marine) Exposure period : 96 hour(s) Unit : mg/l LCS0 : = 184 calculated Method : Year : 1983 GLP : Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LCS0 = 184 mg/L (13-340 mg/L) 72 hour LCS	Exposure period	: 96 hour(s)
LC50 : > 10000 measured/nominal Method : OECD Guide-line 203 'Fish, Acute Toxicity Test" Year : 1997 GLP : Test substance : other TS:AMPD Result : Hour LC50 (mg/L) 24 > 10000 48 > 10000 72 > 10000 96 > 10000 Test substance : AMPD: 2-Amino-2-methyl-1,3-propanediol Conclusion : The acute LC50 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD. Reliability : (2) valid with restrictions Guideline study with acceptable restrictions. 14.11.2006 (19) Type : Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) Unit : mg/l LC50 : > 10000 calculated Method : Year : 1990 Conclusion : The 96-h LC50 in golden orfe is >10,000 mg/L. (20) Type : Reliability : (2) valid with restrictions Guideline study with acceptable restrictions Guideline study with acceptable restrictions. 13.11.2006 (2-P) : Test substance : as prescribed by 1.1 - 1.4 Source : IWL labs 1990 Conclusion : The 96-h LC50 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions Data from handbook or collection of data. 13.11.2006 (20) Species : Pleuronectes platessa (Fish, marine) Exposure period : 96 hour(s) Unit : mg/l LC50 : 1940 acutated Method : Year : 1983 GLP : Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 134 mg/L (31-340 mg/L) 72 hour LC50 = 134 mg/L (31-34	Unit	: mg/l
Method : OECD Guide-line 203 'Fish, Acute Toxicity Test" Year : 1997 GLP : : Test substance : other TS:AMPD Result : Hour LC50 (mg/L) 24 >10000 48 >10000 48 >10000 72 >10000 72 >10000 96 >10000 72 >10000 96 >10000 72 >10000 96 >10000 72 >10000 96 >10000 74 >10000 96 >10000 75 >10000 96 >10000 76 Conclusion : The acute LC50 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD. Reliability : (2) valid with restrictions Guideline study with acceptable restrictions. (19) Type : : : Species : Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) : 190 GL GL GLP : : 19	LC50	: > 10000 measured/nominal
Year : 1997 GLP : Test substance : other TS:AMPD Result : Hour LC50 (mg/L) 24 >10000 48 >10000 72 >10000 Test substance : AMPD: 2-methyl-1,3-propanediol Conclusion : The acute LC50 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD. Reliability : (2) valid with restrictions Guideline study with acceptable restrictions. 14.11.2006 Type : Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) Unit : mg/l LC50 : > 10000 calculated Method : Year : 1990 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : IN/L labs 1990 Conclusion : The 96-h LC50 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions Data from handbook or collection of data. 13.11.2006 Type : semistatic Species : Pleuronectes platessa (Fish, marine) Exposure period : 96 hour(s) Unit : mg/l LC50 : = 184 calculated Method : Year : 1990 Conclusion : The 96-h LC50 in golden orfe is >10,000 mg/L. (20) Type : semistatic Species : Pleuronectes platessa (Fish, marine) Exposure period : 96 hour(s) Unit : mg/l LC50 : = 184 calculated Method : Year : 1983 GLP : Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 184 mg/L (31-340 mg/L) 72 ho	Method	: OECD Guide-line 203 "Fish, Acute Toxicity Test"
GLP : other TS:AMPD Result : Hour LCS0 (mg/L) 24 >10000 48 >10000 72 >10000 96 >10000 96 >10000 96 >10000 96 >10000 96 >10000 96 >10000 96 >10000 96 >10000 96 >10000 96 >10000 96 >10000 97 :: 14.11.2006 (19) Type :: Species : LC50 : >10000 Call : 1990 GLP : Test substance : as prescribed by 1.1 - 1.4 Source Source : !WL labs 1990 Conclusion : The 96-h LC50 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions Data from handbook or collection of data. 13.11.2006 (20) <t< td=""><td>Year</td><td>: 1997</td></t<>	Year	: 1997
Test substance:Other TS:AMPDResult:HourLCS0 (mg/L) 24 < 10000 48 > 10000 72 > 10000 96 > 10000Test substance:AMPD: 2-Amino-2-methyl-1,3-propanediol ConclusionConclusion:The acute LCS0 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD. ReliabilityReliability:(2) valid with restrictions Guideline study with acceptable restrictions.14.11.2006(19)Type:.Species:Leuciscus idus (Fish, fresh water) Exposure periodExposure period:96 hour(s) 10 thitUnit:mg/LTest substance:Year:1990 GLPGLP:Test substance:a s prescribed by 1.1 - 1.4Source:WL labs 1990 Conclusion:Conclusion:The 96-h LC50 in golden orfe is >10,000 mg/L. ReliabilityReliability:(20)Type:semistaticSpecies:Pleuronectes platessa (Fish, marine)Exposure period::?Data from handbook or collection of data.13.11.2006:'' Year:'' Year:'' Result:LC50 values were calculatedMethod:'' Year:'' Year:'' Year:'' Year:'' Year:'' Year	GLP	
Result:HourLC50 (mg/L) 24 > 10000 72 72 240000 $72 < 1000072 < 1000072 < 1000072 < 1000072 < 1000072 < 10000Test substance::AMIPD: 2-methyl-1,3-propanediolConclusion:Reliability::(2) valid with restrictionsGuideline study with acceptable restrictions.10,000 mg/L for AMPD.14.11.2006::(2) valid with restrictionsGuideline study with acceptable restrictions.(19)Type:::.(19)Species:Leuciscus idus (Fish, fresh water)Exposure period:(20) valid with acceptable restrictions.Unit::mg/l.LC50:> 10000 calculated.Method::(20) add with restrictionsData from handbook or collection of data.Source:!WL labs 1990Conclusion.Conclusion:::(20) add with restrictionsData from handbook or collection of data.13.11.2006::semistaticregister.Type::semistatic.Conclusion::::Species:Pleurometes platessa (Fish, marine)Exposure period:::Species:Pleurometes platessa (Fish, marine)Exposure period::Species::Conclusion::Type::Glab:<$	lest substance	: other IS:AMPD
Nestin 1000 24 >10000 24 >10000 48 >10000 72 >10000 96 >10000 96 >10000 96 >10000 72 >10000 96 >10000 96 >10000 96 >10,000 mg/L for AMPD. Reliability (2) valid with restrictions Conclusion : The acute LC50 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD. Reliability : (2) valid with restrictions Guideline study with acceptable restrictions. 14.11.2006 (19) Type : Species : Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) Unit : mg/l LC50 : 1990 GLP : Conclusion : The 96-h LC50 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions Data from handbook or collection of data. : (20)<	Pocult	\cdot Hour \downarrow C50 (mg/l)
$\begin{array}{rcl} 24 & 10000 \\ 72 & 100000 \\ 72 & 10000 \\ 72 & 10000 \\ 72 & 10000 \\ 72 & 10000 \\ 72 & 100$	Result	24 > 1000
Total 10000 96 Yest substance :: Conclusion :: The acute LC50 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD. Reliability :: (2) valid with restrictions Guideline study with acceptable restrictions. 14.11.2006 (19) Type :: Species :: Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) Unit :: mg/l LC50 :: > 10000 calculated Method :: : Year :: 1990 Conclusion :: The 96-h LC50 in golden orfe is >10,000 mg/L. Reliability :: : Test substance :: as prescribed by 1.1 - 1.4 Source :: !! Type :: semistatic Species :: ! Type :: semistatic Species :: Pleuronectes platessa (Fish, marine) Exposure period :: : Ype ::		48 >10000
Test substance : AMPD: 2-Amino-2-methyl-1,3-propanediol Conclusion :: The acute LCS0 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD. Reliability :: (2) valid with restrictions Guideline study with acceptable restrictions. (19) Type :: Species : Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) Unit : mg/l LC50 : > 10000 calculated Method :: Year : 1990 GLP : Test substance : as prescribed by 1.1 - 1.4 Source : IWL labs 1990 Conclusion : The 90-h LC50 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions Data from handbook or collection of data. (20) Type : semistatic Species : Pleuronectes platessa (Fish, marine) Exposure period : 96 hour(s) Unit : mg/l LC50 : = 184 calculated Method :: Year : 1983 GLP :: Test substance :		72 >10000
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Conclusion:The acute LCS0 96-hour for Brachydanio rerio is > 10,000 mg/L for AMPD.Reliability:(2) valid with restrictions14.11.2006(19)Type:Species:Leuciscus idus (Fish, fresh water)96 hour(s)Unit:Type:Species:Leuciscus idus (Fish, fresh water)Exposure period:96 hour(s)Unit:Year:1990GLP:Test substance:as prescribed by 1.1 - 1.4Source:WL labs 1990Conclusion:Conclusion:The 96-h LC50 in golden orfe is >10,000 mg/L.Reliability:(2) valid with restrictionsData from handbook or collection of data.13.11.2006:Type:Species:Pleuronectes platessa (Fish, marine)Exposure period:96 hour(s)Unit:Type:Test substance:other TS:AMP-95Result:LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial Cl: 24 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	Test substance	: AMPD: 2-Amino-2-methyl-1.3-propanediol
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Species : Leuciscus idus (Fish, fresh water) Exposure period : 96 hour(s) Unit : mg/l LCS0 : > 10000 calculated Method : . Year : 1990 GLP : . Test substance : as prescribed by 1.1 - 1.4 Source : IWL labs 1990 Conclusion : The 96-h LC50 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions Data from handbook or collection of data. . 13.11.2006	Type	
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Conclusion : ITLe Bde-h LC50 in golden orfe is >10,000 mg/L. Reliability : (2) valid with restrictions Data from handbook or collection of data. 13.11.2006 (20) Type : semistatic Species : Pleuronectes platessa (Fish, marine) Exposure period : 96 hour(s) Unit : : Year : : Year : : Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 	Source	• IW/L Jabs 1990
Reliability:(2) valid with restrictions Data from handbook or collection of data.13.11.2006(20)Type:semistatic(20)Species:Pleuronectes platessa (Fish, marine)Exposure period:96 hour(s)Unit:mg/lLC50:Year:1983GLP:Test substance:other TS:AMP-95Result:LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial Cl: 24 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	Conclusion	: The 96-h LC50 in golden orfe is >10.000 mg/L.
Data from handbook or collection of data.(20)13.11.2006(20)Type: semistaticSpecies: Pleuronectes platessa (Fish, marine)Exposure period: 96 hour(s)Unit: mg/lLC50: = 184 calculatedMethod:Year: 1983GLP:Test substance: other TS:AMP-95Result: LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial Cl: 24 hour LC50 = 184 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	Reliability	: (2) valid with restrictions
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Species : Pleuronectes platessa (Fish, marine) Exposure period : 96 hour(s) Unit : mg/l LC50 : = 184 calculated Method :	Туре	: semistatic
Exposure period : 96 hour(s) Unit : mg/l LC50 : = 184 calculated Method : Year : 1983 GLP : Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	Species	: Pleuronectes platessa (Fish, marine)
Unit : mg/l LC50 : = 184 calculated Method : Year : 1983 GLP : Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial Cl: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	Exposure period	: 96 hour(s)
LC50 : = 184 Calculated Method : Year : 1983 GLP : Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	Unit	: mg/l
Year : 1983 GLP : . Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	LC50 Motheral	: = 184 calculated
Image: GLP : Test substance : Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	Wethod	1082
Test substance : other TS:AMP-95 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	rear CIP	. 1903
 Result : LC50 values were calculated to be the following, based on observed mortality, at a 95% Fiducial CI: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L) 	Test substance	other TS:AMP-95
mortality, at a 95% Fiducial CI: 24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L)	Recult	 I C50 values were calculated to be the following, based on observed
24 hour LC50 = 231 mg/L (138-385 mg/L) 48 hour LC50 = 184 mg/L (31-340 mg/L) 72 hour LC50 = 184 mg/L (31-340 mg/L) 17/71	NGOUIL	mortality at a 95% Fiducial CI.
48 hour LC50 = 184 mg/L (31-340 mg/L) $72 hour LC50 = 184 mg/L (31-340 mg/L)$ $47 / 74$		24 hour C50 = 231 mg/l (138-385 mg/l)
72 hour LC50 = 184 mg/L (31-340 mg/L)		48 hour LC50 = 184 mg/L (31-340 mg/L)
17 / 71		72 hour LC50 = 184 mg/L (31-340 mg/L)
• • • • •		17 / 71

1. Ecotoxicity	ld 77-86-1 Date
	96 hour LC50 = 184 mg/L (31-340 mg/L)
Test condition	 Nominal temperature of the test solution was 15C (14.4-15C during the test). Dissolved O2 range during the test was 7.0-8.2 mg/L. The pH range during the test was 8.1-10.3. In a 96-hour, semi-static seawater test system, the toxicity of AMP-95 to Plaice was studied. Fish (20 per dose level, mean weight of 4.1g and mean length of 63 2mm) were placed in 18L class vessels containing 10L
Reliability	of the test solution. The pH of the system was 8.08-8.14, and salinity was 34.76-34.82 o/oo. Concentrations tested were 100, 320, 560, 1000 mg/L. : (2) valid with restrictions
19.03.2004	2e (21)
Type	• static
Snecies	: Leuciscus idus (Fish fresh water)
Exposure period	• 48 hour(s)
Unit	: 40 hour(3)
	• _ 220
	= 320
	. = 331
Mothod	= 540
Voar	• 1096
	. 1900
GLF Tost substance	· Other: AMP
Test substance	. Other. AMP
Source	: ANGUS Chemie GmbH Ibbenburen European Commission - European Chemicals Bureau Ispra (VA)
Reliability	: (4) not assignable 4e
05.03.2004	
4.2 ACUTE TOXICITY T	O AQUATIC INVERTEBRATES
Туре	: semistatic
Species	: Crangon crangon (Crustacea)
Exposure period	: 96 hour(s)
Unit	: mg/l
FC50	· ca 179 calculated

Unit EC50 Analytical monitoring Method Year GLP Test substance	mg/l ca. 179 calculated no data 1983 Cother: AMP-95
Method	: Concentrations of AMP-95 tested were 0, 56, 100, 320, and 560 mg/L under semi-static conditions for 96 hours. The 18L glass vessel was filled with 10L of the test solution, and 20 brown shrimp were added to each vessel after an on-site acclimation period of 5 days on a specially-prepared diet. Food was withheld during the test. The test solution was comprised of local filtered seawater (pH=8.10-8.16, salinity 34.86-34.90 o/oo), and was maintained within a temperature range of 14.7-15.8C. The solutions were aerated with compressed air and dissolved O2 ranged from 7.2-8.4 mg/L. The pH range during the test was 8.0-10.1.
Remark	: Detailed information regarding pH and dissolved oxygen were not provided. Only target ranges were given.
Result	 Results were as follows, with a 95% Fiducial CI: 24-hour LC50 = 241 mg/L (17.8-292 mg/L) 48-hour LC50 = 179 mg/L (98.2-329 mg/L) 72-hour LC50 = 179 mg/L (98.2-329 mg/L) 96-hour LC50 = 179 mg/L (98.2-329 mg/L)

Ecotoxicity	Id 77-86-1
2	Date 14.11.2006
Test substance	 The control mortality was 0% at 96 hours. The purity and identification of the test material is not noted. The test material was confirmed by the authors to be AMP-95 (2-Amino-2-methyl-1 propanol).
Reliability	2e
19.03.2004	(2
Type Species Exposure period Unit NOEC EC50 Analytical monitoring Method Year GLP	 static Daphnia magna (Crustacea) 48 hour(s) mg/l = 100 measured/nominal = 193 calculated yes other 1983 yes
Test substance Method	 Other: AMP Test procedures per "Methods of Acute Toxicity Tests with Fish, Macroinvertebrates, and Amphibians" (USEPA, 1975) were followed. The daphnia used in the test were cultured at the testing facilities, and the adu daphnia were fed a suspension of trout chow and alfalfa daily until 24 hou prior to testing.
	The static Daphnia bioassay was conducted in 250mL beakers containing 200mL of test solution. The vessels were kept at 19-21C, with a 16-hour light / 8 hour dark photocycle.
Remark Result	 An initial range-finding study was conducted using 10 Daphnia per concentration level. The range was found by beginning at 0.1 mg/L and increasing the amount of test material by a factor of 10 until a toxic level was found. Once this level had been determined, five concentrations, in duplicate, of the test compound with 10 Daphnia per beaker were selected for their respective bioassay. These concentrations were a logarithmic series ranging from 100 to 1000 mg/L. There was no additional data provided on the results of the range-finding study referenced in the methods section. A computerized program calculated the LC50 for Daphnia based on the method of Standard ed at a provided.
	to be 240 mg/L, and 48-hour LC50 was calculated to be 193 mg/L.

4. Ecotoxicity	ld 77-86-1 Date
	mg/L AMP) at 48 hours. The pH of the control at study start was 7.8. At 48 hours, the pH data collected was: 8.1 in the control, 8.3 (100mg/L AMP), 8.9 (320 mg/L AMP), and 9.4 (1000 mg/L AMP).
Test condition	 The study was conducted following the intent of the Good Laboratory Practice Regulations. Daphnias (Daphnia magna) first instar < 24 hours old. Test conditions were a temperature range of 19-20C, and a photoperiod of 16 hours light. Test solution (200mL) was placed in 250mL beakers, and ten daphnias were added per test solution. Concentrations tested were in duplicate at 0, 100, 180, 320, 560, and 1000 mg/L AMP-95. AMP-95 (94.31% Active 5.69% water. NVM 0.0004%
Reliability	 AMP-95 (94.31% Active, 5.09% water, two 0.0004%. Lot 216, the same as referenced in the report, "Acute Toxicity Effects of AMP-95 on Bluegill Sunfish, Ninety-Six Hour LC50" (C. Parekh, 1980). Received as a clear liquid and stored at room temperature. (2) valid with restrictions
19.03.2004	2e (22)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species Endpoint Exposure period Unit NOEC Method Year GLP Test substance	 Selenastrum capricornutum (Algae) growth rate 4 day(s) mg/l ca. 100 measured/nominal 1985 as prescribed by 1.1 - 1.4
Method	: Dilutions of the actively-growing cultures were added to test flasks to give an approximate initial count of 1000 cells per mL, and incubated 24 hours. An appropriate volume of test material was added to achieve the desired concentration, and the culture volume made up to 100mL with particle-free deionized water. Cell counts were made every 24 hours. The cell counts were used to calculate the growth parameters to establish quantitative toxic values.
Source Test condition	 Dow Chemical, TERC Midland, MI Stock cultures of Selenastrum were maintained at 22C under constant illumination from 2 fluorescent tubes, on slopes of Oxoid agar containing Bold's basal medium. About 7 days prior to use, 2 conical flasks of 90mL particle-free BBM were aseptically inoculated with Selenastrum from stock cultures. The cultures were incubated at 22C, and shaken at 175 rpm under constant illumination. Sub-cultures were made into fresh medium 2 days prior to inoculation of the test flasks, and incubated under identical conditions.
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment.
09.11.2006	(23)
Species Endpoint Exposure period Unit NOEC Method	 Selenastrum capricornutum (Algae) growth rate 96 hour(s) mg/l ca. 100 measured/nominal

4. Ecotoxicity	ld 77-86-1 Date
Year GLP Test substance	: 1985 : : as prescribed by 1.1 - 1.4
Method	: Dilutions of the actively-growing cultures were added to test flasks to give an approximate initial count of 1000 cells per mL, and incubated 24 hours. An appropriate volume of test material was added to achieve the desired concentration, and the culture volume made up to to 100mL with particle- free deionized water. Cell counts were made every 24 hours. The cell counts were used to calculate the growth parameters to establish quantitative toxic values.
Source Test condition	 Dow Chemical, TERC Midland, MI Stock cultures of Selenastrum were maintained at 22C under constant illumination from 2 fluorescent tubes, on slopes of Oxoid agar containing Bold's basal medium. About 7 days prior to use, 2 conical flasks of 90mL particle-free BBM were aseptically inoculated with Selenastrum from stock cultures. The cultures were incubated at 22C, and shaken at 175 rpm under constant illumination. Sub-cultures were made into fresh medium 2 days prior to inoculation of the test flasks, and incubated under identical conditions.
Reliability	 (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment
09.11.2006	(23)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

Type Species Exposure period Unit EC10 Method Year GLP Test substance	 aquatic Pseudomonas putida (Bacteria) mg/l ca. 33447 measured/nominal other: Acute Bacteia Cell Multiplication Inhibition Test 1990 yes other TS: TRIS AMINO 40%
Method :	: A small amount of bacteria from a 7-day old stock culture of Pseudomonas putida was inoculated in fluid nutrient medium in Erlenmeyer flasks. The preliminary cultures were incubated at 25°C for 16-20 hours. Subsequently, the extinction of the monochromatic radiation at 436 nm for a 10 mm layer of the bacterial suspension was determined by photoelectric measurement. On the basis of the values measured, the final turbidity value if the bacterial suspension was adjusted.
	Four parallel dilution series in 300 ml Erlenmeyer flasks stoppered with aluminium caps were prepared from the formulated test substance stock solution and sterile Milli-Q water. Each flask contained 80 ml of liquid at the start.
	Each flask of three dilution seried to be inoculated, was made up to a final volume of 100 ml by adding 5 ml of a stock solution I, 5 ml of a stock solution II, and 10 ml of the prepared bacterial suspension from the preliminary culture having a known adjusted extinction value.
	The flasks of the dilution series that are not inoculated, were made up to 100 ml by adding 5 ml of stock solution I, 5 ml of stock solution II, and 10 ml of saline.
	Five culture flasks of the reference series were made up with 80 ml of the
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Ecotoxicity	Date 14.11.2006
	methanol concentration, 5 ml stock solution I, 5 ml stock solution II, and ml prepared bacterial solution.
	Ten control culture flasks with 80 ml sterile Milli-Q water, 5 ml stock solution I, 5 ml stock solution II and 10 ml prepared bacterial solution we also included.
	All flasks were left at 25°C for 16-20 hours. Subsequently, cell suspensions were homongenized and the extinction of the monochroma radiation at 436 nm in a 10 mm layer in all series were measured.
	Because, after termination of the test period, coloration or turbidity occurred in the dilution series for chemical-physical reasons, the analog steps of dilution of the non-inoculated series were used as photometric blank values for turbidity fo the inoculated dilutions series.
Remark	 The study procedure was based on the following guideline: Umweltbundesamt (UBA) Guidelines: wassergefährdender Stoffe, III Bestimmung der akuten Bakterientoxizität, ad-hoc-Arbeidsgruppe I (Ohmenn Dr. Niemite), ITMO, Nr. 40, Contembor 4070.
Result	 Comann Dr. Niemitz), LTWS, Nr. 10, September 1979. TRIS AMINO 40% was investigated for its ability to inhibit the cell multiplication of the bacteria species Pseudomonas putida. Cultures of Pseudomonas putida bacteria were exposed to concentrations ranging from 432 to 885.6x10³ TRIS AMINO 40% per litre. Based on the solut of TRIS AMINO 40% in water, a toxicity threshold value of TRIS AMINO 40% of 23 4v1042 mg/l for Depudements putida could be determined
Conclusion	 40% of 33.4x10'3 mg/1 of PSeudomonas putida could be determined. The Assessment figure or value for bacteria toxicity is 1.5
Reliability	: (1) valid without restriction
13.11.2006	CEI guideline study.
Tumo	
i ype Species	: Aqualic : Pseudomonas putida (Bacteria)
Exposure period	
Unit	: ma/l
EC10	: = 400 measured/nominal
EC50	: = 1600 measured/nominal
Method	: DIN 38412, part8
Year	: 1990
GLP	: no data
Test substance	: other TS: TRIS AMINO 40%
Result	: EC50 = 1600 mg/l EC10 = 400 mg/l
Test condition	: Concentrations (g/l): 10, 5, 2.5, 1.25, 0.625 and 0.315 Temperature: 20-22°C
Reliability	: (2) valid with restrictions Meets national standard methods with acceptable restrictions
13.11.2006	

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4. Ecotoxicity

ld 77-86-1 Date 14.11.2006

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

5. Toxicity

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type Value Species Strain Sex Number of animals Vehicle Doses Method Year GLP Test substance	 LD50 ca. 5000 ml/kg bw mouse no data no data 10 10 10000, 7000, 5000, 3500 and 2000 mg/kg bw 1955 no as prescribed by 1.1 - 1.4
Method Conclusion Reliability	 Ten mice per dose were administered TRIS AMINO solutions via gavage at a constant volume of 0.05 ml/gm bw. The oral LD50 in mice is 5500 mg/kg. (2) valid with restrictions
13.11.2006	(26)
Type Value Species Strain Sex Number of animals Vehicle Doses Method Year GLP Test substance	 LD50 > 3000 mg/kg bw other: rat and mice other:Wistar and Swiss 1000, 2000, and 3000 mg/kg as a solution 1962 no data as prescribed by 1.1 - 1.4
Method Result Conclusion Reliability 13.11.2006	 Solutions of 20% and 5% of TRIS AMINO were given via gastric intubation to rats and mice respectively at doses of 1000, 2000, and 3000 mg/kg. There was no toxicity noted in either species at the top dose levels, although abundant urine output was noted for some animals The LD50 for rats and mice are estimated to be >3000 mg/kg in solution. (2) valid with restrictions
Type Value Species Strain Sex Number of animals Vehicle Doses Method Year GLP Test substance	 LD50 ca. 3350 mg/kg bw mouse no data no data 7000, 5000, 3530, 2500, and 2000 mg/kg bw 1955 no other TS:AMPD

	Date
Test substance	: AMPD: 2-Amino-2-methyl-1-propanol
Reliability	: (2) valid with restrictions
13.11.2006	(26
.1.2 ACUTE INHALATI	ON TOXICITY
.1.3 ACUTE DERMAL	ΤΟΧΙCΙΤΥ
Туре	: LD50
Value	: > 1000 mg/kg bw
Species	: rat
Strain	
JULA Number of animals	. 5
Vehicle	
Doses	500 mg/kg (delivered via 5% solution)
Method	:
Year	: 1962
GLP	:
Test substance	: as prescribed by 1.1 - 1.4
Method Result	 Five rats were injected with 500 or 1000 mg/kg (via 5% solution) of THAM. 500 mg/kg caused irritation at the injection site.
	1000 mg/kg caused the formation of lesions.
C	There were no other observations noted.
Source Poliability	: Dow Chemical, LERC Midiand, Mi
Reliability	Meets generally accepted scientific standards, well-documented and
09 11 2006	acceptable for assessment.
00.11.2000	
Туре	: LD50
Value	: > 1000 mg/kg bw
Species	: mouse
Strain	
Jex Number of enimels	:
Number of animals	. U
Doses	
Method	
Year	: 1962
GLP	:
Test substance	: as prescribed by 1.1 - 1.4
Method Result	 Mice were injected with 500 or 1000 mg/kg THAM (via 5% solution). 500 mg/kg caused irritation at the injection site.
	1000 mg/kg caused skin lesions at the site.
	There were no other observations of toxicity noted.
Source	: Dow Chemical, IERC Midland, MI
Dallah 114 -	(2) valid with restrictions
Reliability	
Reliability	Meets generally accepted scientific standards, well-documented and acceptable for assessment

5. Toxicity	ld 77-86-1 Date
09.11.2006	(27)
Type Value Species Strain Sex Number of animals Vehicle Doses Method Year GLP	 LD50 > 2000 mg/kg bw rabbit no data male/female 12 other: no vehicle 1000, 1500, and 2000 mg/kg other: Pharmacology Lab Protocol 1980 yes
Test substance	: other TS:AMP
Method	 Each group of rabbits was treated with 1000, 1500, or 2000 mg of test material per kg body weight (mg/kg). The desired dose was spread over the prepared abdominal skin area (abraded or smooth as designated). The skin was covered with a gauze and a sheet of impervious rubberized cloth to prevent any loss of the test material. The trunk was further enclosed with a flexible wire screen held in place by tape. The animals were returned to individual cages. After 24 hours dermally exposed, the bindings and patches were removed, the exposed areas gently cleaned, and observed for skin irritancy. The animals were observed for another 14 days for any gross symptoms of toxicity. At the end of the 14 day observation period, the animals were weiged, sacrificed, and the organs examined for gross pathology. At the end of the 24 hour exposure period, the intact and abraded treated skin sites were severely irritated and black in color. The sites became necrotic within two to three days and remained necrotic for the 14 days. The treated sites had severe eschar formation by the 14th day. The rabbits in the three treatment groups lost body weight over the 14 day observation period. The animals in all treated groups showed no signs of toxicity or apprend barmacelogical behavior. At page on the rappet and the organs in all rabbits.
Test condition	 were grossly normal. The treated skin sites in all rabbits were necrotic. 12 Rabbits weighing 3.0 +/- 0.5 kg were divided into 3 groups of 4 each, and their abdomens were shaved free of hair. The skin of 2 rabbits were further prepared by abrasions. The abrasions were made 2-3 cm apart over the area of exposure with a blunt bypodermic needle without bleeding.
Test substance	 AMP:2-Amino-2-methyl-1-propanol; CAS# 124-68-5; molecular formula C4H11NO: molecular weight 89 14
Conclusion	 The acute dermal LD50 for P-1826 for the rabbit was >2000 mg/kg. The test material was dermally pontoxic, but was a severe skin irritant.
Reliability	 (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
13.11.2006	acceptable assessment. (28)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

Type:Value:Species:Strain:Sex:Number of animals:Vehicle:Doses:Route of admin.:	LD50 ca. 3350 ml/kg bw mouse no data no data 4000, 3600, 3250, 2500 and 2000 mg/kg bw i.p.
--	--

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Toxicity	IQ //-00-1
-	Date 14.11.2006
Exposure time	
Method	:
Year	: 1955
GLP	: no
Test substance	: as prescribed by 1.1 - 1.4
Method	 Ten mice per dose were administered TRIS AMINO solutions via intraperitonal at a constant volume of 0.015 ml/gm bw.
Conclusion	: The acute LD50 i.p. of TRIS AMINO is 3350 mg/kg bw.
Reliability	: (2) valid with restrictions
	Meets national standard methods with acceptable restrictions.
13.11.2006	(20
Туре	: LD50
Value	: ca. 16.5 other: mM TRIS/kg
Species	: mouse
Strain	:
Sex	:
Number of animals	: 10
Vehicle	: no data
Doses	:
Route of admin.	: i.v.
Exposure time	: .5 minute(s)
Method	:
Year	: 1961
GLP	:
Test substance	: as prescribed by 1.1 - 1.4
Method	: IV injections of Tris over 30 seconds using un-neutralized and neutralized
Beault	SOlutions of 0.3M Tris.
Result	Miss that were given a lathel does convulsed immediately before doeth
Source	• Dow Chemical TEPC Midland MI
Jource Test condition	 Dow Chemical, TERC Initialia, Mi Ten mice for each dose, injected IV. Mice were observed for 24 hours and
	LD50 calculated from the per cent of mice that died.
Reliability	: (2) valid with restrictions
	Meets generally accepted scientific standards, well-documented and
~	acceptable for assessment.
09.11.2006	(2)
Туре	: LC50
Value	: 3.28 - 4.04 other:g/kg
Species	: rat
Strain	: Sprague-Dawley
Sex	: male/female
Number of animals	: 6
Vehicle	: physiol. saline
Doses	: 2.0, 2.5, 3.0, 3.5 g/kg of 0.6M THAM, 4.0 and 4.5 g/kg of 0.9M THAM.
Koute of admin.	1.V.
⊏xposure time	: minute(s)
wethoa Voor	- 1005
	- CORI
GLY Tost substance	:
iest substance	as prescribed by 1.1 - 1.4
Method	: Group 1
	Males and females (3 rats each sex) were administered various doses of
	THAM via the tail vein The dose levels of THAM administered were 2.0
	2.5, 3.0, 3.5 g/kg of 0.6M THAM, 4.0 and 4.5 g/kg of 0.9M THAM.
	2.5, 3.0, 3.5 g/kg of 0.6M THAM, 4.0 and 4.5 g/kg of 0.9M THAM. Group 2
	2.5, 3.0, 3.5 g/kg of 0.6M THAM, 4.0 and 4.5 g/kg of 0.9M THAM. Group 2 A replication of experiments conducted with Group 1. Instead of

5. Toxicity	ld	77-86-1
	Date	14.11.2006
	one day, the different dosage levels were given at one s rats, and repeated, on subsequent days, until all 36 ani had been used. Ultimately, 3 male and 3 female rats fre received THAM at each of the dosage levels shown for Group 3 As a partial control (to rule out adverse effects of rapid volumes of fluid), 4 pairs of male and female rats receive the tail vein, of physiological saline (0.9g NaCl per 100r rates and volumes of saline were similar to the fluid infu- volumes required for each dose level of THAM.	sitting to pairs of mals in this group om Group 2 Group 1. infusions of large red IV infusions, via nL). The infusion usion rates and
Result :	In all cases, the rate of administration of the test solution individually based each rat's body weight. Each animal material such that they receive 0.45g THAM per kg bod administered in one minute. Sterile equipment was use study, and each animal was treated with a separate syn Animals that survived the infusions were held and obse post-infusion and were sacrificed and necropsied. Nec- likewise performed on animals dying spontaneously foll Specimens from all organs and tissues were preserved stained. All histopathologic and histochemical methods according to established methods. In most cases, rats that did not survive the post-infusion the infusion or very soon afterward. Those animals that than 10 minutes following treatment survived the entire and recovered with no grossly-observable ill effects. A survived the infusion. There were no significant gross lesions noted at necrop animals, with the exception of the liver and kidneys. Per nephrosis was noted consistently in the kidneys up to 2 infusion. The lesion varied in that it was limited to a mo pyknosis of the nuclei of isolated segments of the renal in rats infused with 2 and 2.5 g/kg THAM doses, and ind with the dose. In higher dose levels, the lesions were of animals and the puscle of isolated segments of the renal in rate infusion with 2 and 2.5 g/kg THAM doses, and ind with the dose. In higher dose levels, the lesions were of animals and the puscle of isolated segments of the renal in rate infused with 2 and 2.5 g/kg THAM doses, and ind with the dose. In higher dose levels, the lesions were of animals infused with 2 and 2.5 g/kg THAM doses, and ind with the dose. In higher dose levels, the lesions were of an animals infused with 2 and 2.5 g/kg THAM doses, and ind with the dose. In higher dose levels, the lesions were of an animals infused with 2 and 2.5 g/kg THAM doses, and ind with the dose. In higher dose levels is found to a more provide the puscles is an animal of an animal second to a more puscles to the puscles is an animal second to a more pusc	ns was established received the test y weight would be ed throughout the inge and needle. rved for 2 hours ropsies were owing treatment. , embedded, and were performed n period died during t survived longer post-infusion period All 8 control animals osy for any of the eracute toxic hours post- oderate degree to tubular epithelium creased in severity characterized by
	severe pyknosis of the nuclei of swollen renal tubular ep carried segments of the cortex. The cytoplasm of the a coagulated, distinctly granular, and intensely eosinophil affected tubules were distended with eosinophilic, amor and secretions. Affected tubules were observed adjace normal tubules. The incidence of lesions increased with levels, and were noted in a similar dose-response patter spontaneously or at scheduled sacrifice.	bithelial cells of ffected cells was ic. Lumens of the phous tissue debris ant to apparently h increasing dose irn in animals dying
Source :	Lethargy was noted sporadically in rats at 3 -4 g/kg dos were all noted with lesions of acute toxic hepatitis. The characterized by pyknosis of the nuclei of the hepatocyt swelling of the cytoplasm of hepatocytes. Although the to be related to THAM administration, they did not cons characteristic lesion as did the peracute toxic nephrosis Dow Chemical, TERC Midland, MI	e levels. There lesion was tes and cloudy lesions are thought titute a consistent s.
Test condition :	Eighty, 200-300g Sprague-Dawley rats (males & female for 14 days prior to study start for health evaluation. Th into 3 groups: 2 experimental groups of 36 animals eac fameles), and any control group (4 males and 4 females)	es) were observed by were segregated h (18 males and 18
Conclusion :	The actual LD50 of THAM given intravenously was calc g/kg +- 0.1 for Group 1, and 3.6 g/kg +- 0.2 for Group 2 evaluation calculated a maximum likelihood estimate of of body weight respectively, with a fiducial probability of LD50 in groups 1 and 2 were expected to be 3.28-3.83 g/kg, respectively.	s). sulated to be 3.5 . Statistical 3.55 and 3.64 g/kg f 95%. The value of g/kg and 3.28-4.04
	A NOAEL was not observed. Observations, treatment-	related, were noted
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5. Toxicity	ld 77-86-1 Date 14.11.2006
Reliability	 at the lowest dose level tested. (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment
09.11.2006	(30)
Type Value Species	: LC50 : ca. 6000 mg/kg bw
Species Strain Sex	
Vehicle Doses	: 100, 200, 400, 500, 1000, 3000, 5000, 6000, 7000 mg/kg (1 & 20%
Route of admin. Exposure time Method	solutions) : i.v. :
Year GLP	: 1962 :
Test substance	: as prescribed by 1.1 - 1.4
Result Source Reliability	 There were no notations of any toxicity at doses lower than 3000 mg/kg. At 5000 mg/kg, 30% mortality was noted, at 6000 mg/kg, 60% mortality was noted, and at 7000 mg/kg, 70% mortality was noted. Dow Chemical, TERC Midland, MI (2) valid with restrictions
09.11.2006	Acceptable for assessment. (27)
Type	. 1 C50
Value Species Strain	ca. 6100 mg/kg bw mouse
Sex Number of animals Vehicle	: : 10
Doses Route of admin.	 100, 200, 400, 500, 1000, 2000, 3000, 5000, 6000, 7000 mg/kg (1% solution) i.v.
Exposure time Method Year	1962
GLP Test substance	: as prescribed by 1.1 - 1.4
Result	There was no mortality noted at doses less than 5000 mg/kg. At 6000 mg/kg, 40% mortality was noted, and at 7000 mg/lg, 100% mortality was recorded. Animals experienced muscle weakness accompanied by respiratory difficulty prior to death.
Source Reliability	 Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
09.11.2006	acceptable for assessment. (27)
Туре	: LC50
Value Species Strain	: rabbit :
Sex	: 29 / 71

ld 77-86-1 Date 14.11.2006

5. Toxicity

Number of animals	: 5
Venicie	= 250 E00 mg/kg (5% colution)
Doses Bouto of admin	. 250, 500 mg/kg (5% solution)
Exposure time	. I.V.
Method	
Yoar	. 1962
GIP	. 1302
Test substance	as prescribed by 1.1 - 1.4
Method	: Marginal vein in the ear was used
Result	: There was no treatment-related mortality. Changes in respiratory rate and
nooun	amplitude were the only observations noted.
Source	: Dow Chemical, TERC Midland, MI
Reliability	: (2) valid with restrictions
2	Meets generally accepted scientific standards, well-documented and
	acceptable for assessment.
09.11.2006	(27)
_	
Туре	: LC50
Value	: > 125 ml/kg bw
Species	: dog
Sov	
Number of animals	
Vehicle	
Doses	: 125 mg/kg (5% solution)
Route of admin.	: i.v.
Exposure time	
Method	:
Year	: 1962
GLP	:
Test substance	: as prescribed by 1.1 - 1.4
Method	: Saphenous vein was used.
Result	: Alterations in respiratory rate and amplitude were the only observations
	noted.
Source	: Dow Chemical, TERC Midland, MI
Reliability	: (2) valid with restrictions
	Meets generally accepted scientific standards, well-documented and
09 11 2006	acceptable for assessment. (27)
09.11.2000	(27)
Туре	: LD50
Value	: > 3000 mg/kg bw
Species	: rat
Strain	: Wistar
Sex	
Number of animals	: 10
Vehicle	
Doses Bouto of admin	:
Exposure time	. other gastric tube
Method	
Year	. 1962
GLP	
Test substance	: as prescribed by 1.1 - 1.4
Method	: Doses of 1000 and 3000 mg/kg were administered by gastric tube in a 20%
	solution.
Result	: I here were no signs of acute toxicity noted.

Toxicity	ld 77-86-1 Date 14.11.2006	
	Abundant urine output was recorded for some animals.	
Source	: Dow Chemical, TERC Midland, MI	
Reliability	: (2) valid with restrictions	
-	Meets generally accepted scientific standards, well-documented and	
	acceptable for assessment.	
09.11.2006		27
Type	· 1 D50	
Value	> 3000 mg/kg bw	
Species	· mouse	
Strain	: Swiss	
Sex		
Number of animals	: 10	
Vehicle		
Doses	:	
Route of admin.	: other: gastric tube	
Exposure time	:	
Method	:	
Year	: 1962	
GLP	:	
Test substance	: as prescribed by 1.1 - 1.4	
Method	: Doses of 1000, 2000, and 3000 mg/kg were delivered by gastric tube in a	à
	5% solution.	
Result	: There was no significant toxicity noted, other than increased urine output	•
Source	: Dow Chemical, TERC Midland, MI	
Reliability	: (2) valid with restrictions	
	Meets generally accepted scientific standards, well-documented and	
00 11 2006	acceptable for assessment.	דר
03.11.2000	(2	<u> </u>

5.2.1 SKIN IRRITATION

Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP Test substance		rabbit moderately irritating Draize Test 1961 as prescribed by 1.1 - 1.4
Method Result	:	A saturated solution (pH 10.8), 25% solution, and the crystalline product were tested on intact and abraded skin of rabbits. The saturated solution and the crystalline product produced comparable primary irritation scores. There was no noticeable irritation produced by any of the test concentrations on unabraded skin. All signs of irritation
		were completely resolved in 48 hours. Solution Intact Abraded Total crystals 0 0.83 0.4 saturated 0.16 0.83 0.5 25% soln 0 0.16 0.08 The test material is considered a mild irritant to the skin.

5. Toxicity	ld 77-86-1 Date 14.11.2006	
Source Reliability 09.11.2006	 Dow Chemical, TERC Midland, MI (4) not assignable Documentation insufficient for assessment. 	(31)
Species Concentration	: rabbit :	
Exposure Exposure time Number of animals		
Vehicle PDII Result	: : highly irritating	
Classification Method	: irritating : Draize Test	
GLP Test substance	no : other TS:AMP	
Source	: ANGUS Chemie GmbH Ibbenburen European Commission - European Chemicals Bureau Ispra (VA).	
Test substance	 AMP: 2-Amino-2-methyl-1-propanol; Molecular formula C4H11NO; Molecular weight 89.14 	
Reliability	: (4) not assignable Documentation insufficient for assessment	
13.11.2006		(32)

5.2.2 EYE IRRITATION

Species	:	rabbit	
Concentration	:	undiluted	
Dose	:		
Exposure time	:		
Comment	:		
Number of animals	:	6	
Vehicle	:	none	
Result	:	not irritating	
Classification	:	not irritating	
Method	:	1075	
Year	:	1975	
GLP Toot out of one of	:		
lest substance	:	as prescribed by 1.1 - 1.4	
Result	:	Combined Average Score = 0/110	
Source Reliability	:	Classification = Non-irritating Dow Chemical, TERC Midland, MI (4) not assignable Documentation insufficient for assessment.	
13.11.2006			(33)
			、 /
Species	:	rabbit	
Concentration	:	40 %	
Dose	:		
Exposure time	:		
Comment	:		
Number of animals	:	6	
	:	water	
Result Olassifiantian	:	not irritating	
Classification	:	not irritating	

5. Toxicity	D	ld 77-86-1 ate 14.11.2006
Method Year GLP Test substance Source Reliability 13.11.2006	1992 as prescribed by 1.1 - 1.4 L. Troester (2006). Personal communication. (2) valid with restrictions Data from handbook or collection of data.	(34)
5.3 SENSITIZATION		
Type Species Concentration Number of animals	Intracutaneus test guinea pig 1 st : Induction .1 % 2 nd : Challenge .05 % 3 rd : Challenge .01 % 30	
Vehicle Result Classification Method Year GLP Test substance	water not sensitizing not sensitizing 1982 other TS:AMP	
Method	One group was treated with 0.05mL of 1% P-1826 s control group was treated with saline, and a positive treated with dinitrochlorobenzene (DNCB solubilized to volume with saline). After 24 hours, sites were cl erythema and edema according to Draize (Draize, C Safety of Chemicals in Foods, Drugs, and Cosmetic Drug Officials of the United States, p. 48, 1957). At application was repeated with each group, and cont week until 10 applications were made. Animals were recovery period, and then challenged at a virgin site control animals were challenged with 0.1mL of 0.05 of P-1826. Positive and negative control animals w with 0.3% and 0.03% DNCB solution. After 24 hour and three hours later scored for erythema and edem again at 48 hours.	solution, a negative e control group was d in alcohol and made leaned and scored for IH, "Appraisal of the cs". Assoc. of Food and 48 hours, the cinued 2-3 times per re allowed a 2 week e. The test and negative % and 0.01% solutions ere also challenged rs, they were depilated, na. Sites were scored
Result	Test material is considered a sensitizer if the challer reactions in a large number of test animals when co- control. During the induction phase, the first injection at 1% 0.5% P-1826 induced necrotic lesions, so the remain made with 0.1% solutions. The 0.3% DNCB sites we entire 10 injections.	nge elicits skin ompared to the negative and second injection at ining 8 injections were vere necrotic for the
	At challenge with 0.05% and 0.01% P-1826, one an showed mild reactions with 0.05%, but none of the challenged with P-1826 showed any reactions at 24 repeat challenge, none of the animals in the test gro reactions with P-1826, but of the negative control gr 0.05% and 1 animal at 0.01% showed skin reaction	imal in the test group negative controls or 48 hours. In the oup showed any roup, 4 animals at s at 24 hours.
	A challenge with the positive control induced skin re control group. The 0.03% solution did not elicit any hours.	eactions in the positive skin reaction at 48

5. Toxicity	Id 77-86-1
····,	Date 14.11.2006
Test condition	 Thirty male guinea pigs (250-300g each) were divided into 3 groups of 10 each. The animals' backs and flanks were shaved free of hair. The guinea pigs were intradermally-injected with the solutions.
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
44.44.0000	acceptable for assessment.
14.11.2006	(35)
Туре	: Patch-Test
Species	: guinea pig
Concentration	: 1 st : Induction 5 % 2 nd : Challenge 2.5 % 3 rd : Challenge 5 %
Number of animals	: 30
Vehicle	: water
Result	: not sensitizing
Classification	: not sensitizing
Year	- 1982
GLP	
Test substance	: other TS: AMP
Method	: One group was treated with 0.5mL of 10% P-1826 solution, a negative control group was treated with saline, and a positive control group was treated with dinitrochlorobenzene (DNCB solubilized in alcohol and made to volume with saline). After 24 hours, the patches were removed, and sites were cleaned and scored at 24 and 48 hours for erythema and edema according to Draize (Draize, JH, "Appraisal of the Safety of Chemicals in Foods, Drugs, and Cosmetics". Assoc. of Food and Drig Officials of the United States, p. 48, 1957). At 48 hours, the application was repeated with each group, and continued 2-3 times per week until 10 applications were made. Animals were allowed a 2 week recovery period, and then challenged at a virgin site. The test and negative control animals were challenged with 0.5mL of 2.5% and 5% solutions of P-1826. Positive and negative control animals were also challenged with 0.03% DNCB solution. After 24 hours, the test material was cleaned away, depilated, and three hours later scored for erythema and edema. Sites were scored again at 48 hours.
Result	 I est material is considered a sensitizer if the challenge elicits skin reactions in a large number of test animals when compared to the negative control. During the induction, the 10% P-1826 solution was found to be mildly irritating to all animals in the test group, so the remaining 8 doses during the induction were made with a 5% solution. The positive control, DCNB, elicited a mild to strong reaction during the 10 applications. There was one death in the positive control group, but was deemed not treatment-related via necropsy.
	At challenge with 2.5% and 5% solutions of P-1826, none of the animals in the test or positive control groups showed any skin reactions at 24 hours, but the positive control animals showed mild skin reactions at 48 hours. Nine of ten positive controls when challenged with DNCB at 24 hours showed skin reactions, and 7/10 showed reactions at 48 hours. Only 4/10 negative controls showed reactions when challenged with DNCB at 24
Test condition	 hours, and none at 48 hours. Thirty male guinea pigs (250-300g each) were divided into 3 groups of 10 each. The animals' backs and flanks were shaved free of hair. The guinea pigs were topically treated with the solutions applied under an occlusive patch.
Conclusion	: 2-Amino-2-methyl-1-propanol was a non-sensitizer in the topical 34 / 71

. Toxicity	ld 77-86-1 Date 14.11.2006
Reliability	 sensitization test in guinea pigs, under these test conditions. (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment
14.11.2006	(36
Type Species Number of animals Vehicle Result Classification	: other : human : :
Method Year GLP Test substance	: : no data : other TS: AMPD and AMP
Result	 In patch tests with humans, cosmetic formulations containing either 0.22% AMP or 0.5% AMPD or 0.075% AMPD were negative for dermal sensitization.
Test condition	: AMP The skin irritation potential of a cosmetic formulation containing 0.22% AMP-95 was examined using a single insult occlusive patch test on 15 panelists. One panelist formulation containing 0.22% AMP-95 had a negligible primary skin irritation potential.
	AMPD A cosmetic formulation containing 0.073% AMPD was tested for sensitization potential in a group of 30 human test subjects using a repeated insult open patch test. The material was applied to the arm daily 4 days/week for 2 weeks, alternating arms daily. In addition, an occlusive patch was was applied on the first day of the test. After the 2-week application period, there was a 2-week nontreatment period. After this 2- week period, the test subjects received a reapplication of the formulation o the formulation to the test site along with an occlusive patch at an adjacent site. The original patch, challenge patch, and open challenge test sites were read at 24, 48, and 96 hour. No reactions were observed in any of the test subjects. The formulation containing 0.073% AMPD was neither a primary irritant, nor a fatiguing agent, nor a sensitizer, and the formulation was safe under the conditions of the study.
	A modified repeated insult patch test of a cosmetic formulation containing 0.5% AMPD was performed on a panel of 39 women and 20 men. The test material, 0.5 ml, was applied to a semiopen patch on the arm of each panelist every Monday, Tuesday, Wednesday, and Thursday for two weeks. The patch sites were graded approximately 24 h after application. In addition, a closed patch was applied to each panelist on the first day fo the study and on the day of challenge. No patches were applied for 2 weeks after the induction phase. On Monday following the nontreatment period, challenge patches were applied to the original test site and an adjacent site; the second closed patch patch was also applied at this time. The challenge sites were graded 1, 2, and 4 days after application. Slight erythema was noted at one adjacent applicatin site at each of the grading times, but it was clear whether these reactions occurred in the panelist. The formulation containing 0.5% AMPD was not a sensitizer under the conditions fo the test.
Reliability	: (2) valid with restrictions Data from handbook or collection of data
14 11 2006	(37) (38

5. Toxicity

Date

5.4 REPEATED DOSE TOXICITY

Type Species Sex Strain Route of admin. Exposure period Frequency of treatm. Post exposure period Doses Control group		Sub-chronic rabbit i.v. 1-99 days once daily 5-100 mL of 0.3M Tris per kg bodyweight, given at pH 7.4 and at pH 5.5
Result	:	Treatment-related mortality was first noted a few days after start of treatment. Results of the study indicate that neutralized Tris is less toxic. Toxic symptoms noted included anorexia, bloody urine, hindleg paralysis, and irregular respiration. Gross observations at necropsy included abnormally red lungs, necrosis at the point of infusion, bleached liver, darkened spleen, bloated stomach, and lesions on the heart and kidney. Histological examinations of the organs were negative.
		increased upon successive treatments, and remained high the following day. Results indicate an accumulation of Tris at 25mL/kg (0.3M), the top dose evaluated for this parameter, but not at doses lower.
		Blood sampling following treatment with Tris (0.3M neutralized and non- neutralized) could not establish a direct relationship between Tris concentrations and glucose concentrations in the blood due to lost sampling data. The glucose concentrations, however, dropped significantly during the infusions, but returned to normal or above normal following the end of the infusions (Tris-induced hypoglycemia persisted longer than the Tris-neutralized). Tris and Tris-Neutralized both caused transient hypoglycemia. Blood analysis on extracted blood (Tris added to blood droplets at varying levels) also determined that there was no deleterious effect on erythrocytes. Urinalysis (urine collected via Foley catheter measured every hour for 7 hours following start of infusion) revealed that the amount of Tris excreted in the urine reached a maximum at the end of infusion, and dropped rapidly after infusion stopped. Only a small quantity of chloride was excreted. With Tris pH 5.5, a larger amount of chloride than with Tris was excreted. At the end of the 7 hours, 44% of the infused Tris was found in the urine, while with Tris pH 5.5, 77% was found.
		Causes of local necrosis around the infusion site were investigated using Tris pH 5.5 and 7.4. Using injected Trypan dye, the irritation caused by the solutions was evaluated by observing the amount of extravasated dye. Neutralization of the Tris reduced the irritation, suggesting that the pH of the Tris is the probable cause of the dermal irritation. Dow Chemical TERC, Midland, MI
Test condition	:	Two or three rabbits for group, injected IV over 3-6 hours for 1-99 consecutive days (depending on the endpoint being measured) at a doses ranging 5-100 mL of 0.3M Tris per kg bodyweight. The doses were given at pH 7.4 and at pH 5.5 to investigate differences related strictly to pH.
Conclusion	:	 Acute IV toxicity of 0.3M Tris is 55mL/kg. Neither neutralization of addition of glucose or NaCl decreases the toxicity. Maximum subchronic non-lethal dose of 0.3M Tris in rabbits is 75- 90mL/kg over 5 hours. Neutralization appears to decrease the toxicity. If high doses of Tris are given frequently, Tris will accumulate. Injection of Tris causes hypoglycemia during infusion and afterward. Tris neutralized causes hypoglycemia only during infusion. There is no direct relationship between Tris and glucose concentrations in the blood.

Toxicity	ld 77-86-1 Date 14.11.2006
	5. Dermal irritation is likely caused by pH, and not toxicity per se of Tris.
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment
09.11.2006	(29)
Туре	: Sub-acute
Species	: rabbit
Sex Strain	
Route of admin.	· : infusion
Exposure period	: 10 days
Frequency of treatm.	once daily
Doses	
Control group	:
Method	:
rear GLP	
Test substance	as prescribed by 1.1 - 1.4
Result	: No treatment-related mortality was noted in either species. There were no
Source	: Dow Chemical. TERC Midland. MI
Test condition	 Three of each species for each dose, injected IV over 30 seconds for 10 consecutive days at a dosage of 50 mL and 10 mL of 0.155M Tris per kg bodyweight
Reliability	 (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
09.11.2006	acceptable for assessment. (29)
Туро	
Species	: rat
Sex	:
Strain	: Sprague-Dawley
Exposure period	: Iniusion : 20 days
Frequency of treatm.	: 1 infusion daily
Post exposure period	: 24 hour or 7 day recovery period without infusions prior to necropsy
Doses Control group	: U.3WI FHAM (U.5, 1.5g/Kg FHAM: 10 or 20 infusions)
NOAEL	: ca. 500 mg/kg
Method	:
Year	: 1965
Test substance	as prescribed by 1.1 - 1.4
Method	 Group 1 Twelve rats (6 each sex) were given 20 daily infusions via the tail vein of 0.3M THAM at 0.5 g/kg administered at 0.45g/kg/min. Group 2 Twelve rats (6 each sex) were given 20 daily infusions via the tail vein of 0.3M THAM at 1.5 g/kg administered as a single injection in a 1-2 minute
	period.
	Group 3 Twelve rats (6 each sex) were given 10 daily infusions via the tail vein of 0.3M THAM at 0.5 g/kg administered at 0.45g/kg/min. On the 11th day, the infusions were discontinued and IP injections were commenced at the same dose level. The IP injections were given daily for 10 days as a single
	same dose level. The injections were given daily for 10 days as a single 37 / 71

5. Toxicity	ld 77-86-1
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	injection in a 1-2 minute period. Group 4
	As controls, 4 rats (2 each sex) received 20 daily IV infusions, via the tail vein, of a solution containing 9g NaCl and 0.37g KCl per liter of water. The rate of injection was 1mL/kg/min. Group 5
	As controls, 4 rats (2 each sex) received 20 daily IP injections of a solution containing 9g NaCl and 0.37g KCl per liter of water. The rate of injection was 1mL/kg/min.
	Half of the surviving rats of Group 1 & 2 and of the rats of the remaining groups were observed for 24 hours after the 20th infusion and were sacrificed and necropsied. The remaining rats of groups 1 & 2 were observed an additional 7 days post-infusion before a scheduled necropsy. At necropsy, tissue and organ specimens were taken, preserved, processed, and evaluated according to established histopathologic and histochemical methods.
Result	: There were no findings noted in any animal in Group 1. Group 1 &3 animals experienced dry gangrene at the sites of the tail injections. Approximately half of the rats in Group 2 & 3 were noted with a mild inflammation of various parts of the visceral peritoneum, or fat necrosis and hemorrhage of the serosa of various parts of the stomach, intestine, and peritoneum. No gross lesions were noted in any control animal. Microscopic examination of tissues revealed a chronic cellulites at the injection sites, and peracute toxic nephrosis of the kidneys (5/6 rats of Group 1 necropsied 24 hours after injection, but not seen in animals allowed the 7 day recovery period). In group 2, all rats necropsied at 24 hours and 5/6 rats in the 7-day recovery group presented similarly.
Source Test condition	 Dow Chemical, TERC Midland, MI Sprague-Dawley, 200-300g, rats (19 males & 19 females) were observed for 14 days prior to study start for health evaluation. They were segregated into 5 groups
Conclusion Reliability	 NOAEL = 0.5g/kg of 0.3M THAM (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment
09.11.2006	(30)
Туре	: Sub-acute
Species Sox	: mouse
Strain	
Route of admin.	: infusion
Exposure period	: 10 days
Frequency of treatm.	once dally
Post exposure period	:
Control group	
Method	:
Year	: 1961
GLP Test substance	: as prescribed by 1.1 - 1.4
Result	No treatment-related mortality was noted in either species. There were no
Result	toxic symptoms, and a histological study of the organs was negative.
Source	: Dow Chemical, TERC Midland, MI
lest condition	 I nree animals for each dose, injected IV over 30 seconds for 10 consecutive days at a dosage of 50 mL and 10 mL of 0.155M Tris per kg bodyweight
Reliability	 (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment.
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5. Toxicity	ld 77-86-1 Date 14.11.2006
09.11.2006	(29)
Type Species Sex Strain Route of admin. Exposure period Frequency of treatm. Post exposure period Doses Control group	 Sub-acute rabbit male/female New Zealand white i.v. 4 weeks, 5 days per week once daily 24 hours or 20 day recovery periods 0.3M THAM at 0.5g/kg at a rate of 0.5 mL/min. yes, concurrent vehicle
Method	 Group 1 Each animal in the first group was given 20 daily infusions (5 days per week for 4 weeks) of 0.3M THAM at 0.5g/kg at a rate of 0.5 mL/min. Group 2 Animals were given 20 daily IV infusions of normal saline to which 5 m Eq of KCI were added per liter. The dosage and rate of administration were equal in volume and time, on a kg bodyweight basis, to the first group. Two males and 2 females from each group were sacrificed and necropsied within 24 hours of the completion of the last treatment. The remaining animals in each group were held for an additional 20 days post-infusion prior to necropsy. Bodyweight, food and water intake, and urine output were recorded daily, urinalysis weekly. Rectal temperature was recorded twice daily during the pre- and post-infusion periods, and immediately before and hourly until temperatures returned to normal after all infusions given over the 20 day treatment period. Urinalysis (pH, albumin, glucose, benzidine test for hemoglobin on centrifuged sediment and supernate, and microscopic examination for red and white blood cells and casts) was performed on the first day of infusions, and twice a week for the remaining 3 weeks. Two days a week during the infusion period, a post-infusion urinary THAM excretion was measured on 2 males and 2 females in each group. Heart blood specimens were collected for BSP retention tests once during the pre- and post-infusion periods, and during the second week of the infusion period. Blood specimens were collected for m an ear vein for the following tests, once during the pre- and post-infusion periods, and twice a week during the infusions; total serum proteins, A/G ratios, serum bilirubin, cephalin flocculation, and serum transaminase. Also, blood specimens were collected at the same time for hemoglobin and hematocrit determinations, and for white blood cell, red blood cell, differential, and platelet counts. At necropsy, specimens from all organs and tissues were extracte
Result	 formalin-calcium for the following: alkaline phosphatase, acid phosphatase, esterase, peptidase, DPN diaphorase, and PPN diaphorase. Food and water consumption and body temperature was not effected by treatment in any group, nor was water diuresis produced. Body weight fluctuated throughout study in all animals, including control, but not in any treatment-related pattern. Seven of 8 rabbits receiving THAM had inflammatory lesions of the external ear. The lesions carried from swelling and redness to dry gangrene and erosion. Weekly analyses on blood samples were normal for the following parameters: total serum proteins, A/G ratio, serum bilirubin, cephalin flocculation, serum transaminase, RBC, differential counts, hemoglobin, microhematocrit, and platelet counts. White blood cell counts in excess of 13,000 were seen in 5/8 rabbits receiving THAM. In all cases, elevated WBC coincided with the observations of dry gangrene in the external ear. No significant findings were noted in any of the parameters tested during

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Source Test condition Conclusion Reliability	 urinalysis. No gross lesions were noted in control animals at necropsy, however 2/4 test rabbits necropsied 20 days post-infusion presented grossly visible infarcts in the kidneys. No gross lesions were noted in any other organ or tissue in any other animal. In 7/8 test animals in which there were gross lesions of the ear, there were microscopic lesions of chronic cellulites and necrosis at sites of injection in the subcutaneous tissues of the ear. The gross kidney lesions in the 2 animals were confirmed microscopically. They were also found to have chronic interstitial nephritis. Infiltrations of lymphocytes were seen in tissue sections of the liver and kidney of 3 additional test rabbits. The infiltrations were seen in animals that were allowed to recover 20 days post-infusion, as well as those sacrificed immediately following the treatment series. Peracute toxic nephrosis was observed in 1 rabbit, which also presented urolithiasis. Lesions of peracute toxic nephrosis were not observed in any of the other 7 rabbits receiving THAM. Dow Chemical, TERC Midland, MI New Zealand rabbits (16 adult, 3-4kg) were used. After a 2 week observation period, the animals per group. Other than the necrotizing effects at the injection site, and transient body temperature changes, the IV administration of up to 20 daily doses of 0.5g/kg of 0.3M THAM produces no readily detected deleterious effects in mature rabbits. (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment.
09.11.2006	(30)
Type Species Sex Strain Route of admin. Exposure period Frequency of treatm. Post exposure period Doses Control group LOAEL Method Year GLP Test substance Method	 Sub-acute Rat male/female Sprague-Dawley oral feed 8 weeks Continuous 0, 1000, 2000, 4000, 8000, 16000 ppm yes, concurrent vehicle ca. 1000 ppm 1976 other TS: P-1826 (assumed to be AMP-regular) The test material was fed in the diet at dosage levels of 1000, 2000, 4000, 8000, 16000 ppm. Ten male and 10 female rats were used at each dose level and also in a control group. The compound was mixed with distilled water so that the same total volume of liquid was mixed with the food for each group. The compound-water premix was mixed with a small amount of diet, and the food premix was mixed with the total diet. Control rats also received distilled water in their diet at the same volume as the treated diets. Rats were observed daily for changes in general behavior and appearance. Individual body weights and sex group food consumption were recorded weekly. At the termination of the study, all surviving rats at the 8000ppm and 16000ppm doses were necronsied. In ardition 2 males and 2 females at
Remark	 each of the other dose levels plus control were likewise necropsied. Livers plus any gross lesions from all animals necropsied were paraffin embedded, sectioned, and stained and examined microscopically. Estimated test material intake on week 4, based on mean body weights
Source Test condition Conclusion Reliability 09.11.2006 Type Species Sex Strain Route of admin. Exposure period Frequency of treatm. Post exposure period Doses Control group LOAEL Method Year GLP Test substance Method	 Dow Chemical, TERC Midland, MI New Zealand rabbits (16 adult, 3-4kg) were used. After a 2 week observation period, the animals were segregated into 2 groups compose of 4 male and 4 female animals per group. Other than the necrotizing effects at the injection site, and transient bor temperature changes, the IV administration of up to 20 daily doses of 0.5g/kg of 0.3M THAM produces no readily detected deleterious effects mature rabbits. (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment. Sub-acute Rat male/female Sprague-Dawley oral feed 8 weeks Continuous 0, 1000, 2000, 4000, 8000, 16000 ppm yes, concurrent vehicle ca. 1000 ppm 1976 other TS: P-1826 (assumed to be AMP-regular) The test material was fed in the diet at dosage levels of 1000, 2000, 400 8000, 16000 ppm tevel and also in a control group. The compound was mixed with the food for each group. The compound-water premix was mixed with the food for each group. The compound-water premix was mixed with the small and 10 fenale rats were used at each doo level and the food premix was mixed with the small and of diet, and the food premix was mixed with the total diet. Control rats received distilled water in their diet at the same volume as the treated of diet, and the food premix was mixed with the total diet. Control rats received distilled water in their diet at the same volume as the treated of diet, and the food premix was mixed with the total diet. Control rats received distilled water in their diet at the same volume as the treated of diet, and the food premix was mixed with the total diet. Control rats received distilled water in their diet at the same volume as the treated of diet, and the food premix was mixed with the total diet. Control rats received distilled water in their diet at the same volume as the treated of diet,

5. Toxicity	ld 77-86-1
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	and food consumption:
	MALESDose GroupMaterial Intake (mg/kg/day)Control01000ppm2442000ppm4744000ppm9128000ppm190116000ppm4698
Result	FEMALES Dose Group Material Intake (mg/kg/day) Control 0 1000ppm 436 2000ppm 806 4000ppm 1628 8000ppm 3424 16000ppm 7579 In-life observations made beginning week 3 included emaciation, rough hair coat, and scattered 1x1mm lesions over the dorsal surfaces of a few rats in the 16000ppm group. After week 3, the incidences increased so that all rats in the high dose showed similar clinical signs. Hair loss on the dorsal surface was also noted in rats in this group beginning at week 5 and continuing thereafter. One female rat at the high dose showed general paleness and tremors prior to death at week 2. Two female rats in the 16000ppm group died during the study.
Test condition	 Rats in the 16000ppm group gained slightly less weight as compared to control rats. Gains in the other dose groups were similar to the control. Food consumption patterns were similar for all animals. At necropsy, all 16000ppm surviving rats, and all rats at the 8000ppm dose level were necropsied. Two males and two females from each of the other groups were also examined. Alopecia and focal skin erosions and ulcerations found in rats at the 16000ppm dose level were viewed as possibly compound related. Hepatocyte vacuolation descrived in rats from all the treated groups, with increasing severity at the higher dose levels, was considered compound related. Microscopic skin lesions found at the 16000ppm dose level, like the corresponding gross lesions, were viewed as possibly compound related. Sixty male (133-183g at study start) and female (115-146g at study start)
Conclusion	 CD rats were received from Charles River Labs, and were housed individually in hanging canges, maintained in a temperature and humidity controlled room. Purina rodent chow and water were provided to the animals ad libitum. : LOAEL = 1000ppm
Doliohility	LOAEL under these test conditions was 1000ppm. Observed effects at 1000 ppm included alopecia and focal skin erosions.
Renability	2e (30)
Type Species Sex Strain Route of admin. Exposure period Froguency of treatm	: Sub-acute : Mouse : male/female : CD-1 : oral feed : 8 weeks : Continuous
Post exposure period	

5. Toxicity	ld 77-86-1
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Doses Control group NOAEL Method Year	 0, 200, 400, 800, 1600, 3200 ppm yes, concurrent vehicle > 3200 ppm 1976
GLP Tost substance	:
Method	 other TS: P-1826 (assumed to be AMP-regular) The test material was fed in the diet at dosage levels of 200, 400, 600, 800, 1600, 3200 ppm. Ten male and 10 female mice were used at each dose level and also in a control group. The compound was mixed with distilled water so that the same total volume of liquid was mixed with the food for each group. The compound-water premix was mixed with a small amount of diet, and the food premix was mixed with the total diet. Control mice also received distilled water in their diet at the same volume as the treated diets.
	Mice were observed daily for changes in general behavior and appearance. Individual body weights and sex group food consumption were recorded weekly.
Remark	 At the termination of the study, all surviving mice at the 3200ppm dose were necropsied. In addition, 2 males and 2 females at each of the other dose levels plus control were likewise necropsied. Livers plus any gross lesions from all animals necropsied were paraffin embedded, sectioned, and stained and examined microscopically. Estimated test material intake for the animals were calculated based on 4 week body weight and food consumption:
	MALES Estimated mg/kg/day Control 0 200ppm 1.244 400ppm 2.088 800ppm 3.360 1600ppm 8.950 3200ppm 14.380
Result	FEMALES Estimated mg/kg/day Control 0 200ppm 1.600 400ppm 3.146 800ppm 6.070 1600ppm 12.870 3200ppm 27.742 : No changes considered to be related to compound were seen in the general behavior and appearance. There was no treatment-related mortality. Increases in body weight over the course of the study were comparable for control and all treated animals. Food consumption values were also similar for the control and treated animals.
Test condition	 There were no compound-related gross lesions observed in any of the necropsied mice. There were likewise no microscopic lesions that were considered treatment-related in the livers of any mice necropsied. Sixty male (20-31g at study start) and female (18-27g at study start) CD-1 mice were received from Charles River Labs, and were housed individually in hanging canges, maintained in a temperature and humidity controlled room. Purina rodent chow and water were provided to the animals ad libitum.
Conclusion	: The NOAEL for CD-1 mice under these conditions is >3200ppm. 42 / 71

Date 14.11.2006	
Reliability : (2) valid with restrictions 2e (40)	
Type:Sub-chronicSpecies:RatSex:male/femaleStrain:other: CDRoute of admin.:GavageExposure period:13 weeksFrequency of treatm.:5 days / weekPost exposure period:0, 500, 750, 1100, 1700 mg/kg at pH of 7 & 11+Control group:yes, concurrent vehicleMethod::Year:1977GLP:Test substance:other TS: P-1826 (assumed to be AMP-Regular)Method::Dose solution for pH=11+ was prepared by adding the neat test material to distilled water in order to achieve a concentration adequate to dose the animals 1-5mL of dose solution of pH=7 was prepared by adding the solution to pH=7. Because the resulting stock solution is purple in color, the solution was decolored using activated charcoal. The stock solution was then diluted as appropriate to obtain solutions adequate to dose the animals with 1-5mL solution according to the inmals with 1-5mL solution according to the inmals	
Dose Administration The solutions were administered 5 days per week for 13 weeks, once per day by oral gavage. The dose was adjusted to the body weight of the rat by administration of different volumes of solution (1-5mL). In-life Observations & Measures Animals were observed 5 days/week for pharmacological effects of the test material. The observations were general, including signs of altered food and water consumption, irritability, sedation, hair and cutaneous alterations, external inflammation, hyperventilation, convulsions, neuromuscular paralysis, and altered activity. Body weights were recorded weekly, and dead animals were recorded and autopsied as soon as discovered.	
Hematology & Clinical Chemistry Measures The following blood parameters were measured: hemaglobin, packed cell colume (hematocrit), erythrocyte count, leucocyte count, differential count, transaminases, glutamic-oxalacetic and glutamic-pyruvic, total protein, blood urea nitrogen (BUN), Sodium, Potassium, Chloride, and Creatinine. For urine, Ph, protein, occult blood, glucose, bilirubin, sediment, and specific gravity were measured.	
Result Pathology Only gross pathology was reported. : Mortality A high mortality rate was recorded for animals with the pH=11 doses. No treatment-related mortality was noted for the pH=7 groups. Behavior and Appearance General observations on the pH=11 groups included hyperventilation and respiratory difficulty, neither of which could be confirmed at necropsy. Non- treatment related pneumonia was noted in some rats. Rats generally	

5. Toxicity	ld 77-86-1 Date 14.11.2006
	In the pH=11 rats receiving more than 750mg/kg, abdominal swelling, nasal and mouth bleeding were noted the first few days of the study. The abdominal swelling was always followed by death. Some loss of hair around the mouth and face was noted for pH=11 rats, and assumed to be treatment-related. General observations for rats receiving the dose at pH=7 were hyperventilation (lesser extent than pH=11 rats), and limited hyperactivity and hyperirritability. Some of the rats also showed loss of hair around the mouth and face. In a few rats, the effect was accompanied by skin ulceration due likely to excessive salivation, or a local irritancy by the compound.
	Body Weight, Food Consumption Trends There was no significant difference in the rate of body weight change between all dose groups relative to the controls. Food and water consumption were qualitatively noted. Food consumption appeared normal, while AMP-administered rats appeared to drink more frequently than controls.
	Hematology & Clinical Chemistry Neither the pH=7 or pH=11 females showed significant differences in hematocrit, hemoglobin, erythorcyte counts, or leukocytes. The males from the pH=11 group in the 1100 mg/kg dose group showed a significant decrease in hematocrit, hemoglobin, and red blood cell count due to a loss of blood.
	pH 11 group males Group HCT HgB RBC(x10^6) WBC Control (10) 44+-1 16.5+-0.4 7.41+-0.29 8.2+-1.2 500mg/kg (8) 43+-1 16.7+-0.9 6.20+-0.47 13.6+-2.6 750mg/kg (7) 43+-1 16.2+-0.5 6.31+-0.34 9.4+-2.5 1100mg/kg(2) 40+-1 14.9+-0.3 5.94+-0.10 9.0+-1.6 1700mg/kg (none surviving)
	A slight but significant decrease is seen in the RBC of rats receiving 500 and 750 mg/kg AMP. There was no significant effect on hematocrit at these exposure levels. The pH=7 group (only at 1700mg/kg) showed significant decreases in hematocrit and hemoglobin.
	pH 7 group Group HCT HgB Control (10) 46+-1 17.1+-0.1 500mg/kg (9) 43+-2 16.4+-0.5 750mg/kg (10) 42+-1 16.4+-0.6 1100mg/kg(10) 42+-1 16.4+-0.8 1700mg/kg (9) 40+-1 15.1+-0.2
	No significant differences were measured in Na, K, Cl, Ca, total serum protein, BUN, or creatinine for either pH group at any dose level. SGPT values were within normal ranges for all dose groups except for pH=7 females at 1700mg/kg, and pH=7 males at 1000 and 1700 mg/kg, where the values were significantly elevated.
	Urinalysis Urinalysis was performed only for the pH=11 groups. The data indicated some proteinuria.
Test condition	 Gross Pathology There were no treatment-related grossly-recognizable changes observed at necropsy. CD rats were received from Charles River weighing 90-100g, and were acclimated a minimum of one week to the laboratory prior to the experimental start. The rats were divided into groups of 10 per sex per 44 / 71

5 Toxicity	ld 77-86-1
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Conclusion	 dose level, for each of 2 pH's. They were housed in groups of 2-3 per plastic cage, and fed via overhead racks. Food and water was provided ad libitum and animals were housed in rooms with controlled temperature, humidity, and photocycles. LOAEL and NOAEL cannot be established because the authors do not specifically state, or provide detailed data regarding the lowest dose level at which in-life observations (respiratory difficulty, hyperirritability,
Reliability	hyperactivity, etc.) were seen. : (2) valid with restrictions 2e (41)
Type Species Sex Strain Route of admin. Exposure period Frequency of treatm. Post exposure period Doses Control group NOAEL Method Year GLP Test substance Remark	 Chronic Dog male/female Beagle oral feed 1 year Continuous 0, 1.1, 11, 110 ppm yes, concurrent vehicle > 110 ppm 1990 other TS: AMP-HCI (47.1% AMP) Test material intake is estimated based on week 36 body weights and food consumption. Males
	Control 0 1.1ppm .031 11ppm 0.31 110ppm 2.98
	Females
	estimated (mg/kg/day) Control 0 1.1ppm .029 11ppm 0.31 110ppm 2.55
Test substance	 2-amino-2-methyl-1-propanol supplied by Angus Chemical Co, Northbrook, IL. Supplied as an aqueous solution of AMP-HCI. Concentration of AMP in the solution was 47 1%
Conclusion	: Based on the findings under these study conditions, there is no effect at any dose level on general appearance, behavior, body weight, food consumption, ophthalmoscopic exams, clinical chemistry, hematology, organ weights, or tissue histopathology. Based on the absence of statistically and biologically significant findings in dose-response patterns, the No-Observed Effect Level for AMP in the diets of Beagle dogs in greater than 110 ppm.
Reliability	 (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment
13.11.2006	(42)

5. Toxicity	Id 77-86-1
	Date
5.5 GENETIC TOXICIT	Y 'IN VITRO'
Туре	: Gene mutation in Saccharomyces cerevisiae
System of testing	:
Test concentration	:
Cvcotoxic concentr.	:
Metabolic activation	:
Result	: Negative
Method	:
Year	: 1990
GLP	
Test substance	: other TS
Result	: No evidence of ketorolac tromethamine-induce mutagenesis in in vitro Saccharomyces cerevisiae [Syntex Laboratories, Inc. Toradol IM (ketorolac tromethamine) prescribing information, product monograph and formulary facts. Palo Alto, CA; 1990]
Source	: Dow Chemical, TERC Midland, MI
Test substance	: Ketorolac is commerically available as the tromethamine salt. Ketorolac tromethamine is commercially available as a racemic mixture.
Reliability	: (4) not assignable
13.11.2006	(43)
Туре	: Bacterial gene mutation assay
System of testing	: E.coli
Test concentration	:
Cycotoxic concentr.	:
Metabolic activation	
Result	: Negative
Method	:
Year	: 1990
GLP	:
Test substance	: as prescribed by 1.1 - 1.4
Result	 No evidence of ketorolac tromethamine-induce mutagenesis in in vitro Escherichia coli [Syntex Laboratories, Inc. Toradol IM (ketorolac tromethamine) prescribing information, product monograph and formulary facts. Palo Alto, CA; 1990]
Source	: Dow Chemical, TERC Midland, MI
Test substance	tromethamine is commercially available as a racemic mixture.
Reliability	: (4) not assignable Documentation insufficient for assessment.
13.11.2006	(43)
Type	: other: Mechanism of nucleolysis in DNA
System of testing	: Streptomyces lividans
Test concentration	
Cycotoxic concentr	
Metabolic activation	
Result	- Positive
Method	
Year	: 1992
GLP	
Test substance	as prescribed by 1.1 - 1.4
Method	 The authors investigated reactivity using a test system of electrophoretic activation of electrophoresis buffer conponents and a combination of the activated samples with S.lividans plasmid DNA. After termination of the reaction, DNA samples were assayed for double-strand cleavage. The 46 / 71

5. Toxicity	ld 77-86-1
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Result	 assay involved gel electrophoresis of the DNA samples in HEPES buffer, components of which are non-reactive with the DNA modifications, Southern transfer, and hybridization. The authors conclude that two finctional groups of TRIS are involved in strand scission. Initial electrophoretic activation at the anode is absolutely required for the formation of a TRIS derivative, a presumptive oxygencentered radical species which can be scavenged by thiourea. This species reacts with the DNA modifications in the first instance. As a result, the lesions are then susceptible to further attack, resulting in strand cleavage. The authors infer that it is the amine group in particular which confers on TRIS its nucleolytic activity.
Source	: Dow Chemical, TERC Midland, MI
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
09.11.2006	(44)
Type System of testing Test concentration Cycotoxic concentr. Metabolic activation Result Method	Salmonella typhimurium reverse mutation assay
Year	: 1990
GLP	
lest substance	: as prescribed by 1.1 - 1.4
Result Source Test substance Reliability 13.11.2006	 No evidence of ketorolac tromethamine-induce mutagenesis in in vitro Salmonella typhimurium [Syntex Laboratories, Inc. Toradol IM (ketorolac tromethamine) prescribing information, product monograph and formulary facts. Palo Alto, CA; 1990] Dow Chemical, TERC Midland, MI Ketorolac is commercially available as the tromethamine salt. Ketorolac tromethamine is commercially available as a racemic mixture. (4) not assignable Documentation insufficient for assessment.
Type	Bacterial reverse mutation assay
System of testing Test concentration Cycotoxic concentr. Metabolic activation Result Method Year GLP Test substance Method	 S. typhimurium (tester strains TA98, TA100, TA1535, TA1537) and E.coli (tester strain WP2 uvrA) 100-5000 ug per plate with and without Negative OECD Guide-line 471 1996 Yes Other TS:AMP Positive controls, appropriate to the given tester strains, were used, and the concentration varied according to positive control compound. The sterility of the test article was verified. Preliminary Toxicity Assay Ten dose levels of the test article were plated, one plate per dose, with overnight cultures of TA100 and WP2 uvrA on selective minimal agar in the presence and observed of ret liver S0 activities
	Mutagenicity Assay An initial and confirmatory assay was used to evaluate the mutagenic potential of the test material. A minimum of 5 dose levels of the test article, 47 / 71

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	with vehicle and positive controls, were plated with tester strains in the presence and absence of rat liver S9 activation. All dose levels of test article, vehicle controls, and positive controls were plated in triplicate. The test systems were exposed to the test article via the plate incorporation methodology described by Ames, et.al.((1975) and updated by Maron and Ames (1983). Plates were coded according to the testing laboratory's standard procedures. Plates were incubated for 48-72 hours at 35-39C. Plates not counted immediately following the incubation period were stored at 2-6C until the colonies could be counted.
	Toxicity and degree of precipation were scored relative to the vehicle control plate using a well-defined 9-point scale. Colonies were counted either by an automated colony counter or by hand unless the assay was preliminary or the plate exhibited toxicity. Plates exhibiting test article precipitate that interferes with an automated colony counter were counted by hand.
	Mean and standard deviation of the number of revertants per plate were calculated and reported. For a positive finding, the test material must cause a dose-related increase in mean revertants per plate of at least one tester strain with a minimum of 2 increasing concentrations of test article. Positive finding for TA1535 and TA1537 is an increase in mean revertants at the peak of dose-response greater than, or equal to, 3x mean control value. Positive finding for TA98, TA100, WP2 uvrA is an increase in mean revertants at the peak of dose-response greater than, or equal to, 2x mean control value.
Remark Result	 GLP's were followed except for the following exceptions: 1. The identity, strength, purity, and composition or other characteristics to define the test or control article were not determined by the testing facility. 2. Analyses to determine the uniformity, concentration, or stability of the test or control mixtures were not performed by the testing facility. 3. The stability of the test or control article under the test conditions was not determined by the testing facility. Water was selected as the solvent of choice. The test material was soluble
	in water at approximately 100ug.mL, the maximum concentration tested. Preliminary Toxicity Assay The maximum dose tested was 5000ug/plate, achieved by using a concentration of 100mg/mL and a 50uL plating aliquot. Based on the findings, the maximum dose plated in the mutagenicity assay was 5000ug/plate. Neither precipitate nor appreciable toxicity was observed.
Test condition	 Mutagenicity Assay Neither precipitate nor appreciable toxicity was observed in the initial or confirmatory assays. In neither the initial nor confirmatory assay was there a positive response with any of the tester strains at any dose level, in the presence or absence of S9 activation. Salmonella tester strains were recieved directly from Dr. Bruce Ames, University of California, Berkeley, and the E.coli was received from the National Collection of Industrial and Marine Bacteria, Aberdeen, Scotland.
	Tester strains TA98 and TA1537 are reverted from histidine dependence (auxotrophy) to histidine independence (prototrophy) by frameshift mutagens. Tester TA1535 is reverted by mutagens that cause basepair substitutions. Tester TA100 is reverted by mutagens that cause both frameshift and basepair substitution mutations. Specificity of the reversion mechanism in E.coli is sensitive to basepair substitution mutations, rather than frameshift mutations.
	To assure that cultures were harvested in late log phase, the length of incubation was controlled and monitored. Each culture wasmonitored spectrophotometrically for turbidity and was harvested at a percent 48 / 71

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	transmittance yielding a titer of approximately 10^9 cells per milliliter.	
Conclusion	Acoclor 1254-induced rat liver S9 was used for the metabolic activation system. The S9 was prepared from male Sprague-Dawley rats induce with a single IP injection of Acroclor 1254, 500mg/kg, five days prior to sacrifice. Each batch was assayed for its ability to metabolize.	n d
Reliability	test material did not cause a positive response with any of the tester st in the presence and absence of Acrolor-induced rat liver S9. (1) valid without restriction	rains
·····,	1a	(45)
Type System of testing Test concentration Cycotoxic concentr. Metabolic activation Result Method Year GLP Test substance Method	Mammalian cell gene mutation assay L5178Y/TK+- Mouse Lymphoma 5000 ug/mL with and without Negative OECD Guide-line 476 1996 Yes other TS: AMP-95 Aroclor 1254-induced rat liver S9 was used as the metabolic activation system. A preliminary toxicity assay was used to establish the optimal dose lew for the mutagenesis assay. L5178Y cells were exposed to the vehicle alone and to concentrations of the test article ranging from 0.5-5000 ul both the absence and presence of S9-activation. Cell population dens was determined at 24 and 48 hours after exposure to the test article. Toxicity was measured as suspension of growth relative to the growth the solvent controls. The mutagenesis assay, initial and independent repeat, was used to evaluate the mutagenic potential of the test article. L5178Y mouse lymphoma cells were exposed to the vehicle alone and concentrations of the test article in both the absence and presence of S Positive controls, with and without S9, were tested concurrently.	els _ in ity of 1 10 S9.
Remark	Positive controls treated with MMS and 7,12-DMBA were also prepared Treatment tubes were gassed with CO2 and incubated in the dark for 4 hours. Cells were washed twice, resuspended, gassed with CO2 in air again, and placed on a roller. Cells were counted, and numbers were adjusted if necessary. For expression of TK+- cells, cells were placed cloning meduim of granulated agar, mixed with Trifluorothymidine (TFT viable count, prewarmed and filled with cloning media. The tubes were centrifuged, supernatant decanted, and cells resuspended in cloning media. Tests were done in duplicate. TFT or Viable Count were add to the floask and placed on a shaker. The suspension was placed in pe dishes, placed in cold storage for 30 minutes, then incubated for 10-14 days. Diameters and numbers of colonies were recorded, and analyzed relat to positive and solvent controls. Previous published data indicates that large colony mutants received localized damage, likely in the form of a point mutation or small deletion within the TK locus, and small colony mutants received damage to collateral loci concordant with the loss of activity. Study followed GLP's except for the following:	d. 4 · in a [) or e Jed etri tri t the TK
remark	 The identity, strength, purity, and composition or other characteristic define the test or control article were not determined by the testing faci 	cs to lity.

5. Toxicity	ld 77-86-1 Date
Result	 Analyses to determine the uniformity, concentration, or stability of the test or control mixtures were not performed by the testing facility. The stability of the test or control article under the test conditions was not determined by the testing facility. Sterile distilled water was the solvent of choice. The test material was miscible in water at 500 mg/mL, the maximum tested concentration.
	Preliminary Toxicity Assay The high dose tested was 5000ug/mL. Osmolality of the solvent control was 300mmol/kg and of the highest dose was 396mmol/kg. Suspension growth relative to the solvent control was 13% at 5000ug/mL without S9 activation, and 31% with S9 activation. Based on these results, the dose levels chosen for the mutagenesis assay ranged from 500-5000ug/mL for both the non-activated and S9-activated cultures.
Conclusion	Mutagenesis Assay In the non-activated system, cultures treated with 1000-500ug/mL were cloned and produced a range in suspension growth of 11-55%. In the S9- activated system, cultures treated with 500-5000ug/mL were cloned and produced a range in suspension growth of 9-98%. No treated cultures exhibited mutant frequencies more than 55 mutants per 10^6 clonable cells over the solvent control. A dose-response trend was not observed in either the activated or non-activated systems. Over the concentration ranges, total growth for the non-activated cultures ranged from 12-60%, and 10-137% for the S9-activated cultures. Results of the independent repeat assay are similar (12-62% and 32-124% for non- activated and S9-activated system suspension growth, respectively; 14- 67% and 29-191% for non-activated and S9-activated system total growths, respectively). There was likewise no increase in mutant frequencies over the solvent control. Colony sizing for the positive control yielded the expected increase in small colonies, thereby verifying the methods used to detect small colony mutants.
Ballatilla	indicate that, under the conditions of this study, 2-amino-2-methyl-1- propanol did not cause a positive response in the non-activated and S9- activated systems, and was concluded to be negative.
Reliability	: (1) valid without restriction 1a (46)

5.6 GENETIC TOXICITY 'IN VIVO'

Type Species Sex Strain Route of admin. Exposure period	Micronucleus assay Mouse
Doses Result Method Year GLP	Negative 1990
Test substance	: as prescribed by 1.1 - 1.4
Result	: There was no evidence of mutagenicity when the micronucleus assay was

5. Toxicity	ld 77-86-1
	Date 14.11.2006
Source Test substance Reliability 13.11.2006	 used to test for chromosome breaks in vivo mice that had received ketorolac tromethamine [Syntex Laboratories, Inc. Toradol IM (ketorolac tromethamine) prescribing information. Palo Alto, CA; 1990] Dow Chemical, TERC Midland, MI Ketorolac is commercially available as the tromethamine salt. Ketorolac tromethamine is commercially available as a racemic mixture. (4) not assignable Documentation insufficient for assessment.
Type Species Sex Strain Route of admin. Exposure period Doses Result Method Year 	 Micronucleus assay Mouse ICR i.p. 1 ip injection vehicle control, 16, 30, 60 mg/kg, and positive control CP (50mg/kg) Negative OECD Guide-line 474 "Genetic Toxicology: Micronucleus Test" 1998 Yes other TS: AMP-95 (lot 35662) In each the Pilot & Toxicity Studies (designed to set appropriate dose levels for the micronucleus assay), and the Micronucleus Assay, mice were weighed immediately prior to dose administration and the dose volume was based on individual body weights. They were observed after dosing, and daily thereatter for 3 days for clinical signs of chemical effects. Body weights were recorded prior to dose administration and the dose volume was based on individual body weights. They were observed after dosing, and daily thereatter for 3 days for clinical signs of chemical effects. Body weights were recorded prior to dose administration and the dose volume was based. Pilot Study: 2 male mice were dosed with 1, 10, 100, or 1000 mg test material /kg body weight, and 5 males and 5 females were dosed with 2000 mg/kg. Toxicity Study: 5 male and 5 female mice were dosed with 200, 400, 600, 800 mg/kg. Based on the results of the pilot and toxicity studies, the high dose for the micronucleus Assay. Micronucleus Assay: Male and female mice were dosed with 16, 30, or 60 mg/kg via a single IP injection of the test material. At the scheduled sacrifice times, up to 5 mice / sex / dose were sarificed, the femure seposed, and bone marrow aspirated into a syringe of fetal bovine serum. The bone marrow cells were presurgender and a small amount of suspension spread onto a glass side. Two to four slides were prepared per animal. The slides were fixed in methanol, stained, and permanently mounted. Slides were coded and scored folindy. Using oil immersion, 2000 polychromatic erythrocytes were scored for the presence of micronuclei. The number of micronucleated mor
	observed and one or more doses were statistically-elevated relative to the

5. Toxicity	I	d 77-86-1
	Dat	e 14.11.2006
Remark	vehicle control (p>0.05) at any sampling time. If a sir significantly elevated without a dose-response trend, considered a suspect or unconfirmed positive and a r recommended. The test article would be considered statistically significant increase in micronucleated pol erythrocytes above the concurrent vehicle control wa sampling time. Exceptions to GLP's included:	Igle dose group was the assay was epeat assay negative if no ychromatic s observed at any
Beerli	 The identity, strength, purity, and composition or or define the test or control article were not determined Analyses to determine the uniformity, concentration test or control mixtures were not performed by the test The stability of the test or control article under the not determined by the testing facility. 	ther characteristics to by the testing facility. In, or stability of the sting facility. test conditions was
Result	Mortality occurred within 4 hours following dosing: 5/5 females at 2000 mg/kg, and 2/2 males at 1000 mg/kg clinical observations noted.	5 males and 5/5 J. There were no
	Toxicity: Mortality occurred within 3 days of dosing: 5/5 males 400, 600, and 800 mg/kg, and 3/5 males and 5/5 fem Clinical signs included: lethargy in males and females piloerection in males and females, and crusty eyes in The LD50/3 was calculated by probit analysis to be a mg/kg for male and female mice.	and 5/5 females at ales at 200 mg/kg. s at all dose levels, males at 200 mg/kg. pproximately 73.7
Test condition	Micronucleus Assay: There was no treatment-related mortality at any dose noted on the day of dosing included: lethargy in male mg/kg (highest dose). All other mice treated with test articles appeared normal during the study. The numb polychromatic erythrocytes per 2000 polychromatic e article-treated groups was not statistically increased r respective vehicle control in either male of female mid level or bone marrow collection time. ICR mice from Harlan Sprague-Dawley, INC, were 6- initiation. For the pilot, toxicity, and MNT assay, male from 27.4g to 31.9g, and females ranged from 23.0g to the lab, the animals were monitored for parasites a were quarantined a minimum of 5 days. They were of general health, food and water consumption patterns The animals were deemed healthy prior to placement	level. Clinical signs s and females at 60 material and control per of micronucleated rythrocytes in test elative to their ce, regardless of dose 8 weeks old at study es ranged in weight to 28.0g. After arrival and infections, and observed daily for , and other conditions. t on study.
	Mice were housed in an AALAC-accredited facility wir temperature, humidity, and photocycle for the species up to 5 per cage, and given food and water ad libitum contained no contaminants that influenced the study.	th appropriate 3. They were housed 1. The feed and water
Conclusion	The mice were randomized and placed into assigned were identified by a uniquely numbered ear tag. Under the conditions of the assay described in this re methyl-1-propanol did not induce a sugnificant increa micronucleated polychromatic erythrocytes in bone m concluded to be negative in the micronucleus test usi ICR mice	dose groups, and port, 2-amino-2- se in the incidence of arrow, and was ng male and female
Reliability	(1) valid without restriction1a	/ / _ \
		(47)

5. Toxicity

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5.7 CARCINOGENICITY

Species Sex Strain Route of admin. Exposure period Frequency of treatm. Post exposure period Doses Result Control group Method Year GLP Test substance	 Dog male/female Beagle oral feed 1 year Continuous 0, 1.1, 11, 110 ppm Negative yes, concurrent vehicle 1990 other TS: AMP-HCI (47.1% AMP)
Remark	: Test material intake is estimated based on week 36 body weights and food consumption. Males estimated (mg/kg/day) Control 0 1.1ppm .031 11ppm 0.31 110ppm 2.98
	Females estimated (mg/kg/day) Control 0 1.1ppm .029 11ppm 0.31 110ppm 2.55
Test substance	 2-amino-2-methyl-1-propanol supplied by Angus Chemical Co, Northbrook, IL. Supplied as an aqueous solution of AMP-HCI. Concentration of AMP in the solution was 47.1%.
Conclusion	: Based on the findings under these study conditions, there is no effect at any dose level on general appearance, behavior, body weight, food consumption, ophthalmoscopic exams, clinical chemistry, hematology, organ weights, or tissue histopathology. Based on the absence of statistically and biologically significant findings in dose-response patterns, the No-Observed Effect Level for AMP in the diets of Beagle dogs in greater than 110 ppm.
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment
13.11.2006	(42)
Species Sex Strain Route of admin. Exposure period Frequency of treatm. Post exposure period Doses Result	Syrian hamster Male other: intratracheal
Control group Method	: yes, concurrent vehicle

. Toxicity	ld 77-86-1 Date 14.11.2006
Year	: 1979
GLP	: no data
Test substance	: other TS: TRIS AMINO and 0.9% NaCl
Method	: Thirty 8-week old male Syrian golden hamsters were used in the control group. The animals received a pelleted diet and water ad libitium. Weekly intratracheal instillations of a 0.2 ml mixture of TRIS buffer and 0.9% NaCl were performed for the life of the hamster.
	Animals were observed for life or sacrificed when moribund. Body weights, average survival time and tumor incidence for the respiratory tract (trachea larynx and lungs) were analyzed. Complete necropsies were performed and the organs fixed in a 10% buffered formalin.
Result	 TRIS AMINO did not induce tumors when tested in male Syrian golden hamsters.
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5.8.1 TOXICITY TO FERTILITY

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

5.9 SPECIFIC INVESTIGATIONS

Endpoint Study descr. in chapter	:	other: Dermal Absorbtion
Reference Type	:	Wepierre, J. and Noel-Hudson, M.S.
Species Sex	:	other: human skin
Strain Route of admin. No. of animals	:	100 uL applications to the skin.
Vehicle Exposure period	:	24 hour(s)
Frequency of treatm. Doses	:	Once 0.1% and 10% solutions
Control group Observation period	:	
Result Method	:	less than 1% of applied dose is absorbed into the skin.
GLP Test substance	:	other TS ⁻ C-14 labeled Tromethamine hydrochloride
Method	•	In vitro percutaneous absorption was studied using dermatomized human
	-	skin (obtained from plastic surgery biopsies of the abdomen and frozen until use) placed in Franz diffusion cells to permit contact of the dermis with a preservation liquid of sodium chloride and bovine serum albumin. The skin was thawed, defatted, and cut to 0.3mm thickness. A cutaneous biopsy is maintained horizontally between the dermis and epidermis. The system is surrounded with a warm water jacket to maintain the temperature at 37C. The air and liquid in the system are circulated continuously to insure consistent temperature.

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Result	The test material, C-14- labeled tromethamine hydrochloride, was furnished by L'Oreal under the same S1 (0.1%) and S2 (10%). The test material (100 uL) was applied to 0.635 cm^2 skin. At regular time interval (2, 4, 6, 8, 10 hours), the totality of the liquid contained in the dermal compartment is taken by lateral adjustment and replaced by new liquid. A t=24 hours, the preservation liquid is removed and the surface of the biopsy is washed with 100 uL of different solvents: 1st wash- Cetavlon / doubly-distilled water 2nd wash- doubly-distilled water 3rd wash- Cetavlon / doubly-distilled water 4th wash- doubly-distilled water 5th wash- doubly-distilled water The application zone was wiped with cotton rolls, the dermis and epidermi were separated mechanically with a scalpel, and digested in Soluene TM for 24 hours at 37C. The detection of remaining radioactivity was determined via liquid scintillation counter. The preservation liquid removed, cotton rolls, and glass cylinder are also counted for radioactivity. Counting values were corrected by the method of external standard to obtain the dpm.
Source Conclusion	 another. At the end of 24 hours, 0.506 +- 0.765 for the 0.1% solution and 0.797 +- 0.691 for the 10% solution were determined. There is no significant difference between the percentages, showing that the increase in test material concentraion does not alter cutaneous permeability under these test conditions. The fluxes in the 2 cases reaches a maximum value after 4 hours and remain constant during the rest of the experiment. After washing, the retention of tromethamine hydrochloride in the dermis and epidermis is low (0.13-0.14% for the 2 solutions in the epidermis and 0.69- 0.22% in the dermis). The test material is not retained in the horny layer. The washing waters contained more than 90% of the applied dose. Dow Chemical, TERC Midland, MI Percutaneous absorption through human skin in vitro is low, on average less than 1% of the applied dose remains at the end of 24 hours, and the formation of the provide that the end of 24 hours, and the formation of the provide that the test of the end of 24 hours, and the formation of the provide that the end of 24 hours, and the formation of the provide that the test of the end of 24 hours, and the formation of the provide that the test of the end of 24 hours, and the formation of the provide that the test of the end of 24 hours, and the formation of the provide that the test of test of the test of test of test of the test of the test of test of the test of test of the test of test of test of test of test of the test of test
	finding is independent of the concentration of tromethamine hydrochloride applied. The test material is almost totally eliminated by washing the skin after 24 hours.
Reliability	 The test material is not retained in the horny layer. (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
09.11.2006	acceptable for assessment. (4
Type of experience	: Direct observation, clinical cases
Method	: THAM (36g in 5% dextrose water) was infused in 4 of the subjects over a period of 30 minutes to 24 hours. One patient received 9g THAM over 30 minutes.
Result	 In all cases, there was a substantial rise in the blood pH and a significant increase in the total CO2 content of the blood. However, none of the patients showed any clinical improvement. The increase in blood pH and total CO2 content appeared to persist for >48 hours in most of the patients and continued to remain elevated for a period of several days in 2 of the patients. Three of five patients showed 2 major complications following
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Source Test condition	 THAM administration: hyperkalemia and oliguria. Dow Chemical, TERC Midland, MI Five patients manifesting 3 types of renal disease: postpartum bilateral renal cortical necrosis, acute and chronic glomerulonephritis, and diabetic intracapillary glomerulonephrosclerosis. All subjects had severe impairment of renal functions as manifested clinically by oliguria, elevated blood urea nitrogen, and abnormal urinary findings. Two were anuric and 2
Conclusion	 In 2 of the cases of complications, the subjects were acutely ill and manifested a severe renal failure that could have resulted in the hyperkalemia and oliguria. However, the third patient, who received the THAM administration more rapidly than the other 4 patients, presented initially with normal urine output and normal serum potassium, and within 12 hours of THAM administration showed marked oliguria and hyperkalemia and an elevation of blood urea nitrogen level, although the mechanism of such complications was not known at the time of publication. THAM is an effective agent in the prompt correction of metabolic acidosis. In comparison with sodium bicarbonate or lactate, it may also offer the advantage of being able to buffer the intracellular hydrogen ion. THAM may therefore be a more suitable agent to correct acidosis when sodium intake is a concern. The authors suggest a maximum dose of 500mg/kg (0.3M THAM in dextrose) given over 24 hours with careful monitoring of urinary output, serum potassium, and blood urea nitrogen for 48 hours following THAM administration
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
23.03.2004	(50)
Type of experience	: Direct observation, clinical cases
Remark	 Results of this study (Samiy, et al., 1961) are typical of many publications evaluating clinical treatment of ill patients with THAM as a buffer for acidosis. The following reference supports the general findings of the authors of this article. Authors support THAM's use as treatment of metabolic acidosis, noting generally positive outcomes and no adverse effects associated with THAM. Dow Chemical TERC, Midland MI
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
23 03 2004	acceptable for assessment.
T ana C arran ¹ and C	
i ype of experience	: Direct observation, clinical cases
Remark	 Results of this study (Samiy, et al., 1961) are typical of many publications evaluating clinical treatment of ill patients with THAM as a buffer for acidosis. The following reference supports the general findings of the authors of this article. The authors support THAM as treatment of respiratory acidosis, and additionally noting a slight decrease in respiration rate following treatment.
Source Reliability	: Dow Chemical, TERC Midland, MI : (2) valid with restrictions
-	Meets generally accepted scientific standards, well-documented and acceptable for assessment.
23.03.2004	(52)
Type of experience	: Direct observation, clinical cases
Remark	 Results of this study (Samiy, et al., 1961) are typical of many publications evaluating clinical treatment of ill patients with THAM as a buffer for acidosis. The following reference supports the general findings of the 56 / 71

5. Toxicity	ld 77-86-1
	Date 14.11.2006
	authors of this article.
Source Reliability	 For treatment of acidosis during cardiac bypass procedures and during total circulatory arrest, these authors additionally note that THAM is even more effective when CO2 is allowed a means to leave the blood (via respiration, etc), and can be used when sodium bicarbonate cannot be (when a patient is on total cardiopulmonary bypass). THAM has not been associated with any toxic effects and is rapidly excreted in the urine. It has however, been noted to have transient effects on blood sugar, serum potassium, increased urinary (water) loss, and depressed respiratory rates. Dow Chemical, TERC Midland, MI (2) valid with restrictions
	acceptable for assessment.
09.11.2006	(53)
Type of experience	: Direct observation, clinical cases
Remark	: Results of this study (Samiy, et al., 1961) are typical of many publications evaluating clinical treatment of ill patients with THAM as a buffer for acidosis. The following reference supports the general findings of the authors of this article.
Source Reliability	 Authors note that Tromethamine NF is an effective blood buffer for metabolic and respiratory acidosis. They suggest administering sufficient amounts of water with treatment may prevent hyperosmolarity and avoid tissue dehydration, and that caution should be used to avoid hyperkalemia, hypoglycemia, and depression of the respiratory center. Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, well-documented and escenteble for accepted scientific standards, well-documented and escenteble for accepted scientific standards.
23.03.2004	acceptable for assessment. (54)
Type of experience	: Direct observation, clinical cases
Remark	: Results of this study (Samiy, et al., 1961) are typical of many publications evaluating clinical treatment of ill patients with THAM as a buffer for acidosis. The following reference supports the general findings of the authors of this article.
Sourco	Authors recommend THAM for treatment of diabetic acidosis, when patients are unable to tolerate sodium bicarbonate or lactate therapy due to severe cardiac and/or sodium-retaining renal failure. The authors suggest that THAM may reduce the insulin requirements for management of diabetic comas. The patients in this article, while responding favorably to THAM treatment, eventually succumbed to other complications of their conditions, and upon post-mortem examinations there were no findings that suggested THAM contributed to the causes of death.
Reliability	 2) valid with restrictions (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment
23.03.2004	(55)
Type of experience	: Direct observation, clinical cases
Method	: The patient breathes oxygen in a nonbreathing system to remove nitrogen from the lungs, after which a barbiturate is used to render the patient unconscious during the procedure. THAM is administered IV (0.33M solution in 0.2% NaCl) to investigate the buffering effect of THAM. Succinylcholine is injected and the bronchoscope introduced when the

5. Toxicity		ld	77-86-1
,		Date	14.11.2006
Result	:	patient is completely paralyzed. Oxygen is continuously the patients to ensure complete apneic oxygenation of t 6-minute procedure. Likewise, no carbon dioxide is allo lungs. Arterial blood samples were drawn before and ir following the 6-minute apneic period, and at 5 minutes a procedure. All patients were maintained on ventilation in normal spontaneous respiration. Blood was analyzed for anaerobically and two at known CO2 tensions) and pCC was calculated according to Henderson-Hasselbalch for determined with a modified Clark electrode. Brachial ar pressure was measured using an inflatable cuff and me During the 6 minute procedure, the PaCO2 increased fr	y administered to he blood during the wed to exit the nmediately and 2 hours post- until recovery of or pH (one D2. Actual HCO3- rnula, and pO2 was terial blood rcury manometer. om 38 to 66 mmHg,
		with a fall in pH from 7.41 to 7.24 and a slight increase bicarbonate from 24 to 28 mEq/L. When THAM was ad arterial pH was kept nearly constant, with a rise of almo proportions in H2CO3 and HCO3- in all cases. The Pat 42 mmHg and the HCO3- from 24 to 28 mEq/L. In all s blood was completely saturated with oxygen throughour	In actual Iministered, the st identical CO2 rose from 37 to ubjects, the arterial t the procedure.
		The two subjects in poor condition prior to the procedur when breathing air, however breathing pure O2 for 5 mi cyanosis. Analysis of arterial blood in both subjects sho compensated respiratory acidosis. PaO2 during breath shunting of nonoxygenated blood in the lungs. Six minu ventilation pre-procedure increased the PaO2 slightly a PaCO2 to normal values with a marked alkaline shift in similarly to the subjects in better initial health during the Both showed improved respiratory status upon removal the airways.	e were cyanotic nutes relieved the owed an almost-fully ing suggested a utes of controlled nd caused a drop in pH. Both faired apneic period. of secretions from
		In patients not receiving THAM, there was a consistent blood pressure, not seen when the respiratory acidosis THAM. No patient receiving THAM reported any compl complications that could be attributed to the buffer or th	rise in diastolic was buffered with aints or e apnea.
Source Test condition	:	Dow Chemical, TERC Midland, MI Twelve adult subjects were treated with THAM in the stu in acid-base status were compared with those occurring receiving THAM. Ten of the subjects had lung tumors of cavities and underwent bronchoscopy as a routine preo They were all in good general condition without respirat subjects, dealt with separately, were in poor condition w insufficiency prior to the bronchoscopy.	udy. The changes in 6 subjects not or small tuberculous perative procedure. ory distress. Two <i>v</i> ith ventilatory
Conclusion	:	Because all CO2 produced by the body remained in the apneic period, a nearly-perfect stoichometric relationshi between CO2 produced and the THAM needed to buffe found to not only reduce the increase in arterial blood p reduce the typical increase in cerebrospinal fluid pressu respiratory acidosis. THAM was not found to cause any were any toxic manifestations noted following THAM tre THAM is considered by the author to be helpful in count acidosis especially in patients where a blood pressure r rise is not desired, during procedures where adequate v maintained at all times.	body during the p was revealed r it. The buffer was ressure, but also re seen in y hypoglycemia, nor eatment. teracting respiratory ise or CSF pressure yentilation cannot be
Reliability	:	 (2) valid with restrictions Meets generally accepted scientific standards, well-doc acceptable for assessment. 	umented and
23.03.2004			(56)
Type of experience	:	Direct observation, clinical cases	
Remark	:	Results of this study (Holmdahl, 1961) are typical of ma evaluating clinical treatment of ill patients with THAM as 58 / 71	ny publications a buffer for

5. Toxicity	ld 77-86-1 Date 14.11.2006
	acidosis. The following references support the general findings of the authors of this article.
Source Reliability 23.03.2004	 For treatment of respiratory acidosis caused by chronic alveolar hypoventilation, the following authors caution use of THAM, because, even though an effective buffer, they noted it caused in some patients an accentuation of anoxemia that became very severe, and suggested that THAM should not be used to treat such cases unless other means of oxygenation are supplied (mechanical aid). The authors do not support general use of THAM as a treatment for respiratory acidosis caused by chronic alveolar hypoventilation. Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment.
Type of experience	: Direct observation, clinical cases
Result	: Results of this study (Holmdahl, 1961) are typical of many publications evaluating clinical treatment of ill patients with THAM as a buffer for acidosis. The following reference supports the general findings of the authors of this article.
Source Reliability	 For treatment of respiratory acidosis, the authors likewise caution the use of THAM to buffer the blood unless supplemental oxygen is given due to a resulting decrease in minute ventilation rate. The authors advance the theory that the decrease in minute ventilation rate is due to the increase in pH of the blood and not due to a reduction of CO2 tension per se. They report spurious findings of kidney (swelling and hydropic degeneration of the lining of proximal tubules) and/or liver (hydropic degeneration of the cells) changes in 2 patients receiving THAM, however the authors are reluctant to attribute the findings directly to THAM based on others in their sampling that do not present such signs. Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
23 03 2004	acceptable for assessment. (58)
Type of experience	: Direct observation. clinical cases
Remark	: Results of this study (Holmdahl, 1961) are typical of many publications evaluating clinical treatment of ill patients with THAM as a buffer for acidosis. The following reference supports the general findings of the authors of this article.
Source Reliability	 For treatment of respiratory acidosis, the authors find THAM an effective blood buffer, noting also a decrease in ventilation rate, and additionally a consistent rise in arterial blood pH and a considerable increase in urinary pH with an associated increase in urinary excretion of bicarbonate. They noted no adverse effects attributed to THAM treatment, and did not note hypoglycemia as other researches have, likely due to lower dose levels. Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, well-documented and acceptable for assessment.
23.03.2004	(59)
Type of experience	: Direct observation, clinical cases
Remark	: Results of this study (Holmdahl, 1961) are typical of many publications 59 / 71

5. Toxicity		Id 77-86-1
		Date 14.11.2006
	evaluating clinical treatment of ill patients with Thacidosis. The following reference supports the g authors of this article.	HAM as a buffer for eneral findings of the
Source Reliability	For treatment of respiratory acidosis, the authors blood buffer, noting also a decrease in ventilation consistent rise in arterial blood pH and a conside pH with an associated increase in urinary excreti noted no adverse effects attributed to THAM treat hypoglycemia as other researches have, likely de Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, w acceptable for assessment.	e find THAM an effective in rate, and additionally a erable increase in urinary on of bicarbonate. They atment, and did not note ue to lower dose levels. ell-documented and
23.03.2004		(00)
Type of experience	Direct observation, clinical cases	
Remark	Results of this study (Holmdahl, 1961) are typica evaluating clinical treatment of ill patients with TH acidosis. The following reference supports the g authors of this article.	l of many publications HAM as a buffer for eneral findings of the
Source Reliability	In the treatment of artificially-induced respiratory recovering from pulmonary tuberculosis with a ner respiratory center), THAM is found to be an effect hyperventilation while breathing CO2. No toxicity respiratory toxicity was suggested by the authors rapidly-reversible effects on ventilation rates. Dow Chemical, TERC Midland, MI (2) valid with restrictions Meets generally accepted scientific standards, w acceptable for assessment	acidosis (in patients ormally sensitive ctive buffer that prevents y has been noted, and to be unlikely due to ell-documented and
09.11.2006		(61)
Type of experience	Direct observation, clinical cases	
Remark	Tromethamine occurs in Ketorolac Tromethamin the formulation to improve solubility. While it is n the side effects of the active drug from Trometha clinically-significant findings that would suggest in	e at approximately 32% of not possible to differentiate ne, there were no t is inherently toxic.
Result	Results of this study are typical of many publication treatment of post-operative patients with Ketorola Ketorolac Tromethamine is a non-steroidal anti- indicated for the treatment of acute pain, recently (U.S. Food and Drug Administration) for IM (intra It is a pyrrolo-pyrrole compound related to tolmet tromethamine salt in Ketorolac enhances its solution ketorolac in 10mg ketorolac tromethamine.	ions evaluating clinical ac Tromethamine. nflammatory drug (NSAID) / approved by the FDA imuscular) administration. in and zomepirac. The bility; there are 6.8mg
	Clinical trials of Ketorolac Tromethamine have preffects in post-operative patients with quick onset than comparable doses of morphine. In addition treatment provides a non dependency-forming all Authors noted that some patients experienced a and bleeding time that was statistically, but not c although it is not recommended for treatment of pleeding disorders. It has, like most NSAIDS, the mucosal injury. Diarrhea, dizziness, sweating, a were noted in 1-3% of patients. There have bee noted. Although animal studies have shown no enter the formation of the statement of the statem	roduced positive analgesic t of relief lasting longer , Ketorolac Tromethamine ternative to opiates. decreased platelet count linically, significant, patients with existing e potential to cause gastric nd pain at the injection site n no drug interactions evidence of mutagenesis,

. Toxicity	Id 77-86-1
•	Date 14.11.2006
	carcinogenesis, or fertility impairment, it is not recommended for pregnant or lactating women. Respiratory depression has not been noted in patients using the drug.
Source	While IM administration has been approved by the USFDA, intravenous administration was not approved when the article was published.
Conclusion	 Ketorolac Tromethamine has been evaluated by the USFDA to be a viable alternative to opiates for pain relief. While side-effects have been noted similar to other NSAIDS, it can be safely used as a non-habit forming drug to relieve pain in most patients.
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
23.03.2004	(62)
Type of experience	: Direct observation, clinical cases
Remark	: Results of this study (Resman-Targoff, 1990) are typical of many publications evaluating clinical treatment of post-operative patients with Ketorolac Tromethamine. The following reference supports the general findings of the authors of this article.
	Noting an increase in liver function test results, the following authors have additionally noted that the drug may not be ideal for use in patients with prior liver damage
Source	: Dow Chemical, TERC Midland, MI
Reliability	: (2) valid with restrictions Meets generally accepted scientific standards, well-documented and
23.03.2004	acceptable for assessment. (63)
Type of experience	: Direct observation, clinical cases
Remark	: Results of this study (Resman-Targoff, 1990) are typical of many publications evaluating clinical treatment of post-operative patients with Ketorolac Tromethamine. The following reference supports the general findings of the authors of this article.
	A transient renal insufficiency was noted by the following authors, directly attributed to Ketorolac Tromethamine treatment. Decreased urine output and an increased serum creatinine concentration were measured. There was no hyperkalemia noted. These observations, however, are in discord with observations of hyperkalemia and increased urine output noted with THAM infusion (minus Ketorolac) in other patients, suggesting that the observations noted by these authors could be primarily attributed to the Ketorolac.
Source Roliability	: Dow Chemical, TERC Midland, MI
Reliability	Meets generally accepted scientific standards, well-documented and
23.03.2004	(64)
Type of experience	: Direct observation, clinical cases
Method	: Four healthy adults received THAM via IV over 30-60 minutes in 0.03M
Result	 64% of THAM administered was eliminated in the urine over 2 days, and 77% over 3 days.
	Major side effects noted in the subject given 8.8 mM/kg bodyweight included: hypotension, hypoxia, hunger, sweating, periodic breathing, 61 / 71

5. Toxicity		ld 77-86-1 Date
Source Reliability 24.03.2004	:	weakness, somnolence, diarrhea, intense wretching and vomiting, sensations of heat, swelling and numbness of face. In all other subjects, water diuresis was noted with decrease blood serum glucose concentrations. Urine pH and CO2 content rose as the THAM was eliminated. Dow Chemical, TERC Midland, MI (4) not assignable Documentation insufficient for assessment.
Type of experience	:	Direct observation, clinical cases
Method	:	Venous blood samples were taken from the other arm 10, 20, and 30 minutes after starting the infusion, and 5, 10, 20, and 40 minutes at 1, 2, 4, 8, 12, 20, and 24 hours after it had ended.
		Urine was collected from the beginning of the infusion until 30 minutes after its end, from 30 min to 4 hours, from 4 to 8 hours, and from 8 to 24 hours.
		Five patients required haemodialysis or haemofiltration as a result of acute anuria. In two patients, gastric juices were drained continuously via a plastic tube and were collected for 48 hours and analyzed for TRIS. In one patient, a drain had been implanted in the common bile duct diring choledocholithotomy and the bile was collected over two 24 hour periods.
Result	:	All samples were analyzed for TRIS concentration via gas chromatography. In three patients, unidentified peaks were seen with the same retention times as TRIS due to additional medication. The haemofiltrates of these patients was subsequently not analyzed for TRIS concentrations.
		The results from the 6 healthy subject were pooled. At the end of infusion, the TRIS plasma concentration averaged 565 ug/mL. There was a biexponential decline of plasma TRIS levels and after 24 hours, the level was only 3.8 ug/mL. The half-life of the terminal phase was 5.6 hours. TRIS concentrations in erythrocytes rose more slowly, reaching a maximum 20 minutes after the end of infusions. After 2 hours, drug levels in erythrocytes were about 1.5 times greater than those in plasma, and they remained well above the corresponding plasma levels during the rest of the observation period.
		Pharmacokinetic parameters were calculated from the individual concentration-time curves using a two-compartment model with elimination from the central compartment.
		TRIS is mainly excreted by the kidney. Already 30 minutes after the end of infusion, 25% of the TRIS was found in the urine, and after 24 hours, 82% of the TRIS had been eliminated in that way.
		Infusion of the strongly alkaline solution (pH 10.9) was well tolerated by all patients without adverse reactions.
		The half-life of TRIS in normuric patients (including some with poor renal function) was longer than in healthy patients (16-45 hours), and the volumes of distribution were much larger. Up to 72% of the TRIS was eliminated in the urine after 24 hours, and an additional 2-5% excreted during the next 24 hours.
		The half-life of TRIS in anuric patients ranged 15-58 hours. 25-66% of the infused TRIS left the plasma in the first 24 hours, and the clearance averaged 16.7 mL/kg/hr. Periods of haemodialysis or haemofiltration did not affect plasma TRIS level. The amount of TRIS eliminated via these procedures could not be measured as the fluid could not be collected for

5. Toxicity	ld 77-86-1 Date 14.11.2006
	analysis.
Source	Less than 0.2% of the infused dose of TRIS was found during 24 hours in gastric juice or bile.
Test condition	 Six healthy volunteers (5 males, 1 female, ages 27-37, weighing 50-90 kg) and 20 patients in a surgical intensive care unit (diagnoses ranged from traumatic lesions, intestinal bleeding, perforated appendicitis, and pancreatitis to rectal and gastric cancer, and aortic aneurysm) were infused with 121 mg/kg TRIS of 0.3 mol/L solution at a pH of 7.4 over 30 minutes in an antebrachial vein
Conclusion	 TRIS is primarily eliminated via the kidneys in the urine. There were no adverse effects related to TRIS treatment in any subject.
Reliability	 Accumulation of TRIS may occur in the body if patients with impaired renal function are repeatedly given TRIS treatment. (2) valid with restrictions
	Meets generally accepted scientific standards, well-documented and

5.11 ADDITIONAL REMARKS

6. Analyt. Meth. for Detection and Identification	Id 77-86-1
······································	Date 14.11.2006

6.1 ANALYTICAL METHODS

6.2 DETECTION AND IDENTIFICATION

7. Ef	f. Against Target Org. and Intended Uses	ld Date	77-86-1	
7.1	FUNCTION			
7.2	EFFECTS ON ORGANISMS TO BE CONTROLLED			
7.3	ORGANISMS TO BE PROTECTED			
7.4	USER			
7.5	RESISTANCE			

8.	Meas.	Nec.	to Prot.	Man,	Animals,	Environment
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- 8.1 METHODS HANDLING AND STORING
- 8.2 FIRE GUIDANCE
- 8.3 EMERGENCY MEASURES
- 8.4 POSSIB. OF RENDERING SUBST. HARMLESS
- 8.5 WASTE MANAGEMENT
- 8.6 SIDE-EFFECTS DETECTION
- 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER
- 8.8 REACTIVITY TOWARDS CONTAINER MATERIAL

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10.1 END POINT SUMMARY

10.2 HAZARD SUMMARY

10.3 RISK ASSESSMENT