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SIC  
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# I U C L I D

## Data Set

**Existing Chemical** : ID: 108-08-7  
**CAS No.** : 108-08-7  
**EINECS Name** : 2,4-dimethylpentane  
**EC No.** :  
**TSCA Name** : 2,4-dimethylpentane  
**Molecular Formula** : C<sub>7</sub>H<sub>16</sub>

**Producer related part**  
**Company** : ExxonMobil Biomedical Sciences Inc.  
**Creation date** : 30.06.2008

**Substance related part**  
**Company** : ExxonMobil Biomedical Sciences Inc.  
**Creation date** : 30.06.2008

**Status** :  
**Memo** : U.S. EPA - HPV Challenge Program

**Printing date** : 18.03.2010  
**Revision date** :  
**Date of last update** : 18.03.2010

**Number of pages** : 5

**Chapter (profile)** : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10  
**Reliability (profile)** : Reliability: without reliability, 1, 2, 3, 4  
**Flags (profile)** : Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE),  
Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

1.1.0 SUBSTANCE IDENTIFICATION

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type :  
Substance type : petroleum product  
Physical status : liquid  
Purity :  
Colour :  
Odour :

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1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

1.3 IMPURITIES

1.4 ADDITIVES

1.5 TOTAL QUANTITY

1.6.1 LABELLING

1.6.2 CLASSIFICATION

1.6.3 PACKAGING

## 1. General Information

**Id** 108-08-7  
**Date** 30.06.2008

### 1.7 USE PATTERN

#### 1.7.1 DETAILED USE PATTERN

#### 1.7.2 METHODS OF MANUFACTURE

### 1.8 REGULATORY MEASURES

#### 1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

#### 1.8.2 ACCEPTABLE RESIDUES LEVELS

#### 1.8.3 WATER POLLUTION

#### 1.8.4 MAJOR ACCIDENT HAZARDS

#### 1.8.5 AIR POLLUTION

#### 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES

#### 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS

#### 1.9.2 COMPONENTS

### 1.10 SOURCE OF EXPOSURE

### 1.11 ADDITIONAL REMARKS

### 1.12 LAST LITERATURE SEARCH

### 1.13 REVIEWS

## 2.1 MELTING POINT

**Value** : = -119.5 °C  
**Sublimation** :  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.  
**Reliability** : (2) valid with restrictions  
The CRC Handbook of Chemistry and Physics is a peer reviewed publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.  
**Flag** : Critical study for SIDS endpoint  
03.07.2008 (13)

## 2.2 BOILING POINT

**Value** : = 80.4 °C at 1013 hPa  
**Decomposition** :  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.  
**Reliability** : (2) valid with restrictions  
The CRC Handbook of Chemistry and Physics is a peer reviewed publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.  
**Flag** : Critical study for SIDS endpoint  
03.07.2008 (13)

## 2.3 DENSITY

**Type** : density  
**Value** : = .668 g/cm<sup>3</sup> at 20 °C  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.  
**Reliability** : (2) valid with restrictions  
Data supplied by the experimental database associated with EPISuite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-reviewed publication.  
**Flag** : Critical study for SIDS endpoint  
03.07.2008 (16)

## 2.3.1 GRANULOMETRY

## 2.4 VAPOUR PRESSURE

**Value** : = 10.59 hPa at 25 °C  
**Decomposition** :  
**Method** :

## 2. Physico-Chemical Data

**Id** 108-08-7  
**Date** 30.06.2008

**Year** :  
**GLP** : no data  
**Method** : Method not specified.  
**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data were not reviewed for quality, however, the reference is from a peer-reviewed handbook.  
**Flag** : Critical study for SIDS endpoint  
03.07.2008 (4)

### 2.5 PARTITION COEFFICIENT

**Partition coefficient** : octanol-water  
**Log pow** : = 3.63 at 25 °C  
**pH value** :  
**Method** : other (calculated)  
**Year** :  
**GLP** :  
**Method** : Calculated values using KOWWIN version 1.67, a subroutine of the computer program EPIWIN version 3.20  
**Test condition** : Octanol / Water Partition Coefficient is calculated by the KOWWIN subroutine, which is based on an atom/fragment contribution method of W. Meylan and P. Howard in "Atom/fragment contribution method for estimating octanol-water partition coefficients". 1995. J. Pharm. Sci. 84:83-92.  
**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown  
**Reliability** : (2) valid with restrictions  
The value was calculated based on chemical structure as modeled by EPI Suite™. This robust summary has a reliability rating of 2 because the data are calculated and not measured.  
**Flag** : Critical study for SIDS endpoint  
03.07.2008 (17)(19)

### 2.6.1 SOLUBILITY IN DIFFERENT MEDIA

**Solubility in** : Water  
**Value** : = 5.5 mg/l at 25 °C  
**pH value** :  
**concentration** : at °C  
**Temperature effects** :  
**Examine different pol.** :  
**pKa** : at 25 °C  
**Description** :  
**Stable** :  
**Deg. product** :  
**Method** : other: no data  
**Year** :  
**GLP** : no data  
**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown  
**Reliability** : (2) valid with restrictions  
Data supplied by the experimental database associated with EPI Suite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-reviewed publication.  
**Flag** : Critical study for SIDS endpoint  
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2. Physico-Chemical Data

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- 2.6.2 SURFACE TENSION
- 2.7 FLASH POINT
- 2.8 AUTO FLAMMABILITY
- 2.9 FLAMMABILITY
- 2.10 EXPLOSIVE PROPERTIES
- 2.11 OXIDIZING PROPERTIES
- 2.12 DISSOCIATION CONSTANT
- 2.13 VISCOSITY
- 2.14 ADDITIONAL REMARKS

## 3.1.1 PHOTODEGRADATION

Type : air  
 Light source :  
 Light spectrum : nm  
 Relative intensity : based on intensity of sunlight  
 Conc. of substance : at 25 °C  
**INDIRECT PHOTOLYSIS**  
 Sensitizer : OH  
 Conc. of sensitizer : 1500000 molecule/cm<sup>3</sup>  
 Rate constant : = .00000000000685 cm<sup>3</sup>/(molecule\*sec)  
 Degradation : = 50 % after 18.7 hour(s)  
 Deg. product :  
 Method : other (calculated): Calculated values using AOPWIN version 1.92, a subroutine of the computer program EPI Suite™ version 3.20  
 Year :  
 GLP :  
 Method : Calculated values using AOPWIN version 1.92, a subroutine of the computer program EPI Suite™ version 3.20

Indirect photodegradation, or atmospheric oxidation potential, is based on the structure-activity relationship methods developed by R. Atkinson under the following conditions:  
 Temperature: 25°C  
 Sensitizer: OH- radical  
 Concentration of Sensitizer: 1.5E6 OH- radicals/cm<sup>3</sup>  
**Remark** : 2,4-dimethylpentane has the potential to volatilize to air, based on a relatively high vapor pressure, where it is subject to atmospheric oxidation. In air, 2,4-dimethylpentane can react with photosensitized oxygen in the form of hydroxyl radicals (OH-). The computer program AOPWIN (atmospheric oxidation program for Microsoft Windows) (EPI Suite™, 2000) calculates a chemical half-life for a 12-hour day (the 12-hour day half-life value normalizes degradation to a standard day light period during which hydroxyl radicals needed for degradation are generated), based on an OH- reaction rate constant and a defined OH- concentration. Based on a 12-hour day, a rate constant of 6.85 E-12 cm<sup>3</sup>/molecule\*sec, and an OH- concentration of 1.5 E6 OH-/cm<sup>3</sup>, 2,4-dimethylpentane has a calculated half-life in air of 1. 6 days or 18.7 hours of daylight.  
**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.  
**Reliability** : (2) valid with restrictions  
 The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data are calculated and not measured.  
**Flag** : Critical study for SIDS endpoint  
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Deg. product :  
 Method :  
 Year :  
 GLP :  
 Method : Technical discussion  
**Remark** : Direct photochemical degradation occurs through the absorbance of solar radiation by a chemical substance in aqueous solution. If the absorbed energy is high enough, then the resultant excited state of the chemical may undergo a transformation. A prerequisite for direct photodegradation is the ability of one or more bonds within a chemical to absorb ultraviolet (UV)/visible light in the 290 to 750 nm range. Light wavelengths longer than 750 nm do not contain sufficient energy to break chemical bonds, and wavelengths below 290 nm are shielded from the earth by the stratospheric

ozone layer (Harris, 1982).

An approach to assessing the potential for a substance to undergo photochemical degradation is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by constituent molecules (Zepp and Cline, 1977). Saturated and unsaturated hydrocarbons do not absorb light above 290 nm. Consequently, 2,4-dimethylpentane is not subject to photolytic processes in the aqueous environment.

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown  
**Reliability** : (2) valid with restrictions  
 This robust summary has a reliability of 2 because it is a technical discussion and not a study.

**Flag** : Critical study for SIDS endpoint

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### 3.1.2 STABILITY IN WATER

**Type** : abiotic  
**t1/2 pH4** : at °C  
**t1/2 pH7** : at °C  
**t1/2 pH9** : at °C  
**Deg. product** :  
**Method** : other: Technical discussion  
**Year** :  
**GLP** : no data  
**Result** : Hydrolysis of an organic chemical is the transformation process in which a water molecule or hydroxide ion reacts to form a new carbon-oxygen bond. Chemicals with leaving groups that have a potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Gould, 1959). The lack of a suitable leaving group renders a compound resistant to hydrolysis. Heptane is resistant to hydrolysis because it lacks a functional group that is hydrolytically reactive and Harris (1982) identifies hydrocarbons as generally resistant to hydrolysis. Therefore, hydrolysis will not contribute to the removal of 2,4-dimethylpentane from the environment.

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown  
**Reliability** : (2) valid with restrictions  
 This robust summary has a reliability of 2 because it is a technical discussion and not a study.

**Flag** : Critical study for SIDS endpoint

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### 3.1.3 STABILITY IN SOIL

### 3.2.1 MONITORING DATA

### 3.2.2 FIELD STUDIES

### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

**Type** :  
**Media** : other: air - biota - sediment(s) - soil - water  
**Air** : % (Fugacity Model Level I)  
**Water** : % (Fugacity Model Level I)



### 3. Environmental Fate and Pathways

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**Soil** : % (Fugacity Model Level I)  
**Biota** : % (Fugacity Model Level II/III)  
**Soil** : % (Fugacity Model Level II/III)  
**Method** : other: Calculation according Mackay, Level I  
**Year** :  
**Remark** : The EQC Level I is a steady state, equilibrium model that utilizes the input of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Physicochemical data used in the calculation:

Parameter	Value w/ Units
Molecular Weight	100.21
Temperature	25° C
Log Kow	3.63
Water Solubility	5.5 g/m3
Vapor Pressure	10,586 Pa
Melting Point	-119.5° C

**Result** : Using the Mackay Level I calculation, the following distribution is predicted for heptane:

%Distribution	Compartment
99.98	Air
<0.01	Water
0.01	Soil
<0.01	Sediment
<0.01	Suspended Sediment
<0.01	Biota

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data are calculated.

**Flag** : Critical study for SIDS endpoint  
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**Type** : fugacity model level III  
**Media** : other  
**Air** : % (Fugacity Model Level I)  
**Water** : % (Fugacity Model Level I)  
**Soil** : % (Fugacity Model Level I)  
**Biota** : % (Fugacity Model Level II/III)  
**Soil** : % (Fugacity Model Level II/III)  
**Method** : other: Level III simulation using the Mackay Multimedia Environmental Model (Mackay, 2001)  
**Year** :  
**Method** : The EQC Level III model is a steady state model that is useful for determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also

used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 100.21 g/mol

Water solubility = 5.5 g/m<sup>3</sup>

Vapour pressure = 10,586 Pa

log Kow = 3.63

Melting point = -119.5 deg C

Degradation half-lives:

Air - 18.7 hrs

Water - 360 hrs

Soil - 720 hrs

Sediment - 7200 hrs

Environmental Properties (EQC standard environment)

Dimensions (all defaults)

Densities (all defaults)

Organic carbon & Advection (all defaults)

Transport Velocities (all defaults)

Emission and Inflows (defaults used)

Air 1000 kg/hr

Water 1000 kg/hr

Soil 1000 kg/hr

Sediment 0 kg/hr

**Result**

: Output:

	Mass%	Emissions(kg/hr)
Air	20.0	1000
Water	70.6	1000
Soil	2.8	1000
Sediment	6.6	0

**Test substance**

: CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Conclusion**

: The majority of 2,4-dimethylpentane is calculated to partition into the water phase, with smaller but significant amounts into air, soil, and sediment based on the modeling parameters used in this calculation. 2,4-dimethylpentane is considered to be a Type 1 chemical with potential to partition into all environmental compartments.

**Reliability**

: (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are calculated.

**Flag**

: Critical study for SIDS endpoint

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### 3.3.2 DISTRIBUTION

### 3.4 MODE OF DEGRADATION IN ACTUAL USE

### 3.5 BIODEGRADATION

**Type**

: aerobic

**Inoculum**

: Freshwater (un-acclimated pond water)

**Test duration**

: 31 days

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**Degradation Result** : Median half-life = 9.1 days  
**Deg. product** : other: primary biodegradation half-life  
**Method** :  
**Year** : No guideline followed  
**GLP** : 2007  
**Result** : no  
: The median half-life of total detectable gasoline hydrocarbons was 5.0 days, and the mean of these estimates was 5.9 days (n=102). No obvious differences were noted between inocula collected during different times of the year. Although laboratory samples were kept at "summer" temperatures.

Biodegradation followed a semi-sequential process: n-alkanes degraded first followed by iso-alkanes, with preference for the 2-methyl form over the 3-methyl and 4-methyl forms. More complex branching slowed down biodegradation rates some. The larger n-alkanes and iso-alkanes degraded faster than their smaller carbon number counterparts. Dimethyl forms degraded similarly to 3-methyl forms. Overall, within 5 days the majority of n-alkanes and iso-alkanes had been degraded. Below are results based on the calculation presented in the publication:

The percent depletion of individual gasoline hydrocarbons was calculated using the equation:

$$\% \text{ Loss} = [((A_0/C_0) - (A_s/C_s))/(A_0/C_0)] \times 100$$

Where  $A_s$  and  $C_s$  are the concentrations of the target analyte and conserved compound in the sample, respectively, and  $A_0$  and  $C_0$  are the concentrations in the sterile controls.

Calculated apparent half-lives,  $\tau$ , for the disappearance of the total detectable gasoline and the individual hydrocarbons from the fraction remaining  $((100 - \% \text{ loss})/100)$ ,  $A$ , at time  $t$  from the equation

$$\tau = \ln 2 \cdot (-t/\ln A)$$

Substance	Median half-life (days)
2,4-dimethylpentane	9.1
n-hexane	6.5
n-heptane	4.5
cyclohexane	8.2

**Test condition** : The test substance was a complex hydrocarbon, unleaded regular gasoline, obtained from the American Petroleum Institute (API 91-1). The experiments were performed in sealed 40 ml vials with enough headspace to ensure adequate oxygen was available for complete biodegradation of the added gasoline, which was approximately 70 ppm in 10 ml of water. Test concentration selected to be in the range typically used in ready biodegradation studies following OECD 301 test methods. Gasoline was added to each test vial separately. A purge-and-trap analysis was used to detect individual hydrocarbons in both the air and water phases.

Water samples used as inocula were collected from a New Jersey rainwater retention pond (approx. 4000 m<sup>2</sup> surface area, up to 3 m deep) approximately every month throughout the year, from the New Jersey coastline in June and November (seawater samples), and from an activated sludge wastewater treatment facility treating only domestic wastewater in August. None showed any detectable hydrocarbons by the methods used here (detection limit ~2 ppb in 10 ml water). The waters

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were amended with 1% Bushnell Haas medium to provide approx.100  $\mu$ M biologically available nitrogen and phosphorus, and 10ml was added to a 40ml vial, which already contained approx.0.7  $\mu$ l of gasoline. The vials were sealed with a Teflon membrane, and incubated with gentle swirling (100 rpm) or almost horizontal rotation (1 rpm) at laboratory room temperature (approx. 21 °C). At the end of the incubation, some vials were briefly opened to add sufficient HCl to bring the pH to 2. Others were analyzed without killing the organisms. Each vial was analyzed by purge-and-trap gas chromatography coupled with mass spectrometry.

Samples were taken at various intervals after the initiation of the tests, with a poisoned control included at each sample point.

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane  
**Conclusion** : 2,4-dimethylpentane is expected to have a primary aerobic biodegradation half-life of 9.1 days in freshwater.  
**Reliability** : (2) valid with restrictions

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**Type** : aerobic  
**Inoculum** : activated sludge  
**Contact time** : 28 day(s)  
**Degradation** : 74 ( $\pm$ ) % after 28 day(s)  
**Result** : other: readily biodegradable  
**Deg. Product** :  
**Method** : OECD Guide-line 301 F "Ready Biodegradability: Manometric Respirometry Test"  
**Year** : 1996  
**GLP** : yes  
**Result** : Test material was readily biodegradable. Half-life was reached by day 10. By day 28, 74% degradation of the test material was observed. 10% biodegradation was achieved on day 4. By day 10, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were noted. Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an elemental analysis of the test material.

	% Degradation*	Mean % Degradation
<u>Sample</u>	<u>(day 28)</u>	<u>(day 28)</u>
Test Material	72.5, 74.0, 76.8	74.4
Na Benzoate	88.7, 88.9	88.8

**Test condition** : \* replicate data  
: Non acclimated activated sludge and test medium were combined prior to test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate, Calcium chloride).  
Test vessels were 1L glass flasks placed in a waterbath and electronically monitored for oxygen consumption.  
Test material was tested in triplicate, controls and blanks were tested in duplicate.  
Test material concentration was approximately 45 mg/L. Sodium benzoate (positive control) concentration was 50mg/L.  
Test temperature was 22 +/- 1 Deg C.  
All test vessels were stirred constantly for 28 days using magnetic stir bars and plates.  
**Test substance** : CAS #142-82-5; heptane; 99% pure.

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**Conclusion** : Heptane is readily biodegradable.  
**Reliability** : (1) valid without restrictions

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**Type** : aerobic  
**Inoculum** : other: soil, non-adapted  
**Contact time** : 20 day(s)  
**Degradation** : 70 (±) % after 20 day(s)  
**Result** : other: readily biodegradable  
**Deg. Product** :  
**Method** : other: Standard Methods for the Examination of Water and Waste Water  
**Year** : 1971  
**GLP** : no  
**Result** : 70% degradation was measured after 20 days incubation with an unacclimated inoculum.  
% Biodegradation of test substance after days:  
2 days = 28 %  
5 days = 63 %  
10 days = 70 %  
20 days = 70 %

**Test condition** : American Public Health Association, Standard Methods for the Examination of Water and Waste Water, using 1.0 mg/l of test substance. Biodegradation was determined by measuring biological oxygen demand (BOD). Each 300 ml BOD bottle received 1.0 mg of heptane, 1.0 ml of a 1:10 suspension Hudson-Collamer silt loam soil in distilled water, and a mineral salts solution prepared as described in the test method. Bottles were incubated in the dark at 25C. The test substance was obtained from Aldrich Chemical Co.

**Test substance** : CAS #142-82-5; heptane; 99% pure.

**Conclusion** : Heptane is readily biodegradable.

**Reliability** : (2) valid with restrictions

A standard test method was used. The study was conducted prior to GLP.

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**Type** : Computer model  
**Inoculum** :  
**Test duration** :  
**Degradation** : half-life = 5.6 days  
**Result** : other: primary biodegradation half-life  
**Deg. product** :  
**Method** : BioHCWin version 1.01  
**Year** : 2008  
**GLP** : no  
**Result** : **Regression Analysis**

Initially, each compound from the 121 compound training set was fragmented into the applicable fragment descriptors of the [BIOWIN Program](#)'s MITI model (Biowin5) and Linear model (Biowin1). The number of instances of each fragment occurring in each compound was placed in mathematical matrix suitable for a multiple-linear regression analysis. The solution column of the matrix is the log recommended Half-life (in days).

Successive multiple-linear regressions were performed to remove and add fragment descriptors. New fragments were suggested by comparing the results from the multiple linear regression with the recommended half-lives. As new fragments were added and others removed, new regressions were run and the results examined. Several new fragments were added to better define common structures found in petroleum hydrocarbons, such as those for ring systems (e.g. number of cyclic rings, number of 5-member aromatic

rings, number of 6-member aromatic rings). Initial regressions also included molecular weight and log P values as descriptors; both descriptors were dropped in the final model as neither had much statistical significance.

Because of the use of fragments as correction factors in this model, all applicable fragments are counted for each compound. Therefore, the equation defining the linear model is as follows:

$$\log HL = \sum a_0 + a_1 f_{i1} + a_2 f_{i2} + \dots + a_m f_{im}$$

where

$f_{ij}$ =number of  $i^{\text{th}}$  substructures in  $j^{\text{th}}$  chemical

$a_0$ =intercept (equation constant = 0.48976)

$a_i$ =regression coefficient for  $i^{\text{th}}$  substructure

$m$ =number of substructures in model

HL=half-life in days

The model considers any half-life that is less than one day to be equal to one day.

Substance	Half-life (days)
2,4-dimethylpentane	5.6
n-hexane	4.7
n-heptane	5.5

**Test condition** : This model is a subroutine of EPISuite version 4.0 from US EPA.

This model uses a fragment-based approach that is similar to several other biodegradation models, such as those within the Biodegradation Probability Program (BIOWIN) estimation program. In the present study, a half-life in days is estimated using a multiple linear regression against counts of 31 distinct molecular fragments. The model was developed using a data set consisting of 175 compounds with environmentally-relevant experimental data that was divided into training and validation sets. The original fragments from the Ministry of International Trade and Industry BIOWIN model were used initially as structural descriptors and additional fragments were then added to better describe the ring systems found in petroleum hydrocarbons and to adjust for nonlinearity within the experimental data. The training and validation sets had  $r^2$  values of 0.91 and 0.81, respectively.

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane  
**Conclusion** : 2,4-dimethylpentane is expected to have a primary aerobic biodegradation half-life of 5.6 days in freshwater.  
**Reliability** : (2) valid with restrictions

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### 3.6 BOD5, COD OR BOD5/COD RATIO

### 3.7 BIOACCUMULATION

### 3. Environmental Fate and Pathways

**Id** 108-08-7  
**Date** 30.06.2008

**Species** : other: see remark  
**Exposure period** : at 25 °C  
**Concentration** :  
**BCF** : = 125  
**Elimination** :  
**Method** : other: calculation  
**Year** :  
**GLP** : no  
**Remark** : A log bioconcentration factor (BCF) of 2.097 is calculated (BCF = 125).  
With respect to a log Kow = 3.63, which was used to calculate the BCF,  
2,4-dimethylpentane in the aquatic environment is expected to have a low  
potential to bioaccumulate.  
**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data are  
calculated and not measured.  
**Flag** : Critical study for SIDS endpoint  
03.07.2008 (19)

#### 3.8 ADDITIONAL REMARKS

## 4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type	:	
Species	:	other: fish
Exposure period	:	96 hour(s)
Unit	:	mg/l
LC50	:	= 2.2
Method	:	other: ECOSAR version 0.99h, US EPA
Year	:	
GLP	:	
Method	:	ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.
		To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.
Result	:	Calculated 96-hr LC50 for fish = 2.2 mg/L
Test condition	:	Experimental water solubility = 5.5 mg/l @ 25°C (Yalkowsky and Dannenfelser, 1992), log Kow = 3.63 @ 25°C (USEPA, 2000), and melting point = -119.5°C (Lide et al., 1997-1998) were entered into the program. Class: Neutral organics
Test substance	:	CAS #108-08-7; 2,4-dimethylpentane; purity is unknown
Reliability	:	(2) valid with restrictions This robust summary has a reliability rating of 2 because the data are calculated and not measured.

03.07.2008

(5)

## 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type	:	
Species	:	other: Daphnia
Exposure period	:	48 hour(s)
Unit	:	mg/l
LC50	:	= 2.6
Method	:	other: ECOSAR version 0.99h, US EPA



## 4. Ecotoxicity

Id 108-08-7

Date 30.06.2008

**Year** :  
**GLP** :  
**Method** : ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

**Result** : Calculated 48-hr LC50 for a daphnid = 2.6 mg/L  
**Test condition** : Experimental water solubility = 5.5 mg/l @ 25°C (Yalkowsky and Dannenfelser, 1992), log Kow = 3.63 @ 25°C (USEPA, 2000), and melting point = -119.5°C (Lide et al., 1997-1998) were entered into the program. Class: Neutral organics

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data are calculated and not measured.

03.07.2008 (5)

**Type** : static  
**Species** : other: Daphnia  
**Exposure period** : 48 hour(s)  
**Unit** : mg/l  
**EC50** : = 1.5  
**Method** : other: based on discussions in GESAMP/MARPOL meetings held in 1973  
**Year** :  
**GLP** : no data  
**Method** : Individual treatment concentrations were prepared by mixing the test substance in freshwater for 24 hours in a conical flask. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were 250 ml conical flasks with 25 daphnids per flask. Two replicates of each treatment level and control were evaluated.

Organisms supplied by testing lab; age = <24 hours old; parents age = approximately 21 days old.  
Statistical method:  
Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981.  
Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

## 4. Ecotoxicity

Id 108-08-7

Date

	119.).																		
Result	: 48-hr EC50 for a daphnid = 0.423 mg/L																		
Test condition	: Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 20 Deg C.																		
Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L. Only the following analytical data were reported:																			
	<table><tr><td>Nominal Conc. (mg/L)</td><td>Initial Measured Conc. (mg/L)</td><td>48-hr Measured Conc. (mg/L)</td></tr><tr><td>0.32</td><td>0.04</td><td>Not Determined</td></tr><tr><td>1.0</td><td>0.04</td><td>Not Determined</td></tr><tr><td>3.2</td><td>0.5</td><td>Not Determined</td></tr><tr><td>5.6</td><td>2.1</td><td>1.7</td></tr><tr><td>10</td><td>2.2</td><td>Not Determined</td></tr></table>	Nominal Conc. (mg/L)	Initial Measured Conc. (mg/L)	48-hr Measured Conc. (mg/L)	0.32	0.04	Not Determined	1.0	0.04	Not Determined	3.2	0.5	Not Determined	5.6	2.1	1.7	10	2.2	Not Determined
Nominal Conc. (mg/L)	Initial Measured Conc. (mg/L)	48-hr Measured Conc. (mg/L)																	
0.32	0.04	Not Determined																	
1.0	0.04	Not Determined																	
3.2	0.5	Not Determined																	
5.6	2.1	1.7																	
10	2.2	Not Determined																	
Test substance	: CAS #142-82-5; heptane																		
Reliability	: (2) valid with restrictions This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline, OECD 202, and used acceptable methods to prepare exposure solutions.																		
Flag	: Critical study for SIDS endpoint																		
03.07.2008	(18)																		
Type	: semistatic																		
Species	: other: Gammarid (Chaetogammarus marinus)																		
Exposure period	: 96 hour(s)																		
Unit	: mg/l																		
LC50	: = .2																		
Method	: other: Static Gammarid Acute Toxicity Test																		
Year	:																		
GLP	: no data																		
Method	: Individual treatment solutions were prepared by mixing the test substance in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24 hours up to 96 hours.																		
Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8‰; length = 5 mm.																			
Statistical method:																			
Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-119.).																			
Result	: 96-hr LC50 for a gammarid = 0.2 mg/L																		
Test condition	: Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 15 Deg C. Natural seawater was used with a salinity of 2.8‰																		
Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L. Test solutions were analyzed only upon test initiation.																			
Only the following analytical data were reported:																			

## 4. Ecotoxicity

Id 108-08-7

Date

	<table><tr><th>Nominal Conc. (mg/L)</th><th>Initial Measured Conc. (mg/L)</th></tr><tr><td>0.32</td><td>0.003</td></tr><tr><td>1.0</td><td>0.07</td></tr><tr><td>3.2</td><td>0.2</td></tr><tr><td>5.6</td><td>Not Determined</td></tr><tr><td>10</td><td>Not Determined</td></tr></table>	Nominal Conc. (mg/L)	Initial Measured Conc. (mg/L)	0.32	0.003	1.0	0.07	3.2	0.2	5.6	Not Determined	10	Not Determined
Nominal Conc. (mg/L)	Initial Measured Conc. (mg/L)												
0.32	0.003												
1.0	0.07												
3.2	0.2												
5.6	Not Determined												
10	Not Determined												
<b>Test substance</b>	: CAS #142-82-5; heptane												
<b>Reliability</b>	: (2) valid with restrictions This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to prepare exposure solutions.												
<b>Flag</b> 03.07.2008	: Critical study for SIDS endpoint (18)												
<b>Type</b>	: semistatic												
<b>Species</b>	: other: mysid shrimp (Mysidopsis bahia)												
<b>Exposure period</b>	: 96 hour(s)												
<b>Unit</b>	: mg/l												
<b>LC50</b>	: = .1												
<b>Method</b>	: other: Static Gammarid Acute Toxicity Test												
<b>Year</b>	:												
<b>GLP</b>	: no data												
<b>Method</b>	: Individual treatment solutions were prepared by mixing the test substance in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24 hours up to 96 hours.  Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8%; test organisms were approximately 4 weeks old, with lengths of approximately 6 mm. Statistical method: Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-119.).												
<b>Result</b>	: 96-hr LC50 for a gammarid = 0.1 mg/L												
<b>Test condition</b>	: Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 20 Deg C. Natural seawater was used with a salinity of 2.8%  Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L. Test solutions were analyzed only upon test initiation.  Only the following analytical data were reported: <table><tr><th>Nominal Conc. (mg/L)</th><th>Initial Measured Conc. (mg/L)</th></tr><tr><td>0.32</td><td>0.003</td></tr><tr><td>1.0</td><td>0.07</td></tr><tr><td>3.2</td><td>0.2</td></tr><tr><td>5.6</td><td>Not Determined</td></tr><tr><td>10</td><td>Not Determined</td></tr></table>	Nominal Conc. (mg/L)	Initial Measured Conc. (mg/L)	0.32	0.003	1.0	0.07	3.2	0.2	5.6	Not Determined	10	Not Determined
Nominal Conc. (mg/L)	Initial Measured Conc. (mg/L)												
0.32	0.003												
1.0	0.07												
3.2	0.2												
5.6	Not Determined												
10	Not Determined												
<b>Test substance</b>	: CAS #142-82-5; heptane												

**Reliability** : (2) valid with restrictions  
 This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to prepare exposure solutions.

**Flag** : Critical study for SIDS endpoint  
 03.07.2008 (18)

#### 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

**Species** : other algae: Green Alga  
**Endpoint** :  
**Exposure period** : 96 hour(s)  
**Unit** : mg/l  
**EC50** : = 2.0  
**ChV** : = 0.5  
**Method** : other: ECOSAR version 0.99h, US EPA  
**Year** :  
**GLP** :  
**Method** : ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

**Result** : Calculated 96-hr EC50 for a green alga = 2.0 mg/L  
 Calculated 96-hr ChV for a green alga = 0.5 mg/L

**Test condition** : Experimental water solubility = 5.5 mg/l @ 25°C (Yalkowsky and Dannenfelser, 1992), log Kow = 3.63 @ 25°C (USEPA, 2000), and melting point = -119.5°C (Lide et al., 1997-1998) were entered into the program.  
 Class: Neutral organics

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions  
 This robust summary has a reliability rating of 2 because the data are calculated and not measured.

03.07.2008

(5)

## 4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

## 4.5.1 CHRONIC TOXICITY TO FISH

## 4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

## 4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

## 4.6.2 TOXICITY TO TERRESTRIAL PLANTS

Species	: other terrestrial plant: Lactuca sativa Ravel R2
Endpoint	: other: growth
Exposure period	: day(s)
Unit	: mg/l
Method	: other
Year	:
GLP	: no
Test substance	: n-Heptane; CAS #142-82-5; >= 95% pure
Result	: Soil Test Results Test 1 (lab 1): Days 7 and 14 EC50 = >1000 mg/kg soil (dry weight), based on nominal concentrations. No effects at the highest concentration tested. Test 2 (lab 2): Days 7 and 14 EC50 = >1000 mg/kg soil (dry weight), based on nominal concentrations. No effects at the highest concentration tested.  Nutrient Solution Test Results Test 1 (lab 1): Day 16 or 21 EC50 = 1.7 (1.4-2.0) mg/L, based on measured concentrations (value is below water solubility). Test 2 (lab 2): Day 16 or 21 EC50 = 47 (38-65) mg/L, based on measured concentrations (value exceeds water solubility).
Test condition	: Testing occurred in two labs. Two tests were conducted in soil (one at each lab) and two tests in a nutrient solution (one at each lab). In both soil tests, lettuce seeds (Lactuca sativa) were germinated in soil obtained from an orchard. The characteristics of the soil collected for the first study were: pH = 7.5; organic matter content = 1.4%; clay content = 12%. The characteristics of the soil collected for the second study were: pH = 7.5; organic matter content = 1.8%; clay content = 24%. Test soil was sieved, 4 mm, prior to use. The nutrient solution composition was as follows: 3.73 mM Ca(NO3); 4.40 mM KNO3; 0.97 mM KH2PO4; 1.92 mM MgSO4; 0.89 mM K2SO4; trace elements and FeEDTA (per Steiner A, 1968. Soilless culture. Proceedings, Sixth Colloquium of the International Potash Institute, Florence, Italy, pp. 342-341.).  Soil Experiments  Test substances were dissolved in acetone and/or mixed with a small amount of quartz sand, which was then mixed through the test soil to obtain a uniform distribution of test substance through the soil sample. A nominal concentration of 1000 mg/kg soil was tested.  Test substance concentrations in dry soil were spaced by a factor of 3.2. Controls and at least 3 test concentrations were run in duplicate. Soil was brought to 25 to 30% moisture content, which was equivalent to approximately 80% water holding capacity. Test systems were covered

with glass plates until germination occurred to prevent moisture loss. After germination, the plates were removed and moisture loss was control by adding demineralized water back to the soil.

Five seedlings from each replicate test system were selected for the test and allowed to grow. Seedlings were sampled on days 7 and 14 by cutting them at soil level and measuring fresh weight. Soil water content and pH were measured at experimental initiation and termination.

Heptane was analyzed in soil at test initiation and termination by solvent extraction and GC/FID and/or GC/ECD. Measured concentrations in soil for test substances ranged from 70 to 150% of nominal concentrations. Specific results for heptane were not given.

#### Nutrient Solution Experiment

Test substances were dissolved in tributylalcohol and added to the nutrient solution. Seeds were placed on plastic trays, which contained perlite that was saturated with nutrient solution. The trays were covered with glass plates to reduce volatile loss during germination. After germination, seedlings were transferred to containers filled with nutrient solution and test substance. Solutions were renewed 3 times a week. Duplicate containers were used for each test concentration. Test concentrations were spaced by a factor of 3.2. Shoots were harvested on either day 16 or 21 and fresh weights determined. Oxygen content and pH were measured at solution renewal.

Heptane was analyzed in nutrient solution at test initiation and termination by solvent extraction and GC/FID and/or GC/ECD. Measured concentrations in solution for heptane were <50% of nominal concentrations. Specific results for heptane were not given. The volatility of heptane was suspected to have contributed to the loss in solution.

#### Statistics

EC50 values were determined based only on harvested shoots and calculated by applying a logistic model (per Haanstra et al., 1985. The use of sigmoidal dose response curves in soil ecotoxicological research. Plant Soil 82, 293-297.)

#### Reliability

- : (2) valid with restrictions
- Although a specific standard guideline was not used, the test method generally followed standard test procedures, however, less specific analytical information was provided than is desirable.

07.07.2008

(12)

#### Species

- : other terrestrial plant: *Daucus carota* L. cv. Early Horn

#### Endpoint

- : other: leaf damage, electrolyte loss

#### Exposure period

- : 0 day(s)

#### Unit

- : mg/l

#### Method

- : other

#### Year

- :

#### GLP

- : no

#### Test substance

- : n-Heptane; CAS #142-82-5; >99.5% pure

#### Result

- : The rate at which electrolytes were lost from carrot leaves exposed to n-heptane was 34 min<sup>-1</sup> x E4. The rate at which electrolytes were lost from untreated leaves was 12 min<sup>-1</sup> x E4. The effects on cell membrane permeability as indicated by conductance agreed with the known susceptibility of different plant species towards hydrocarbons applied in the liquid state. The effect on the final conductance values and electrolyte concentration was similar for the hydrocarbons tested.

Electrolyte loss from leaves exposed to a liquid hydrocarbon solvent

**Test condition**

coating is explained as resulting from the destruction of plant cell membrane permeability and the release of internal cell electrolytes.

: Carrot plants, *Daucus carota*, were grown in a nutrient solution under controlled temperatures of 23 deg C during the day and 13 deg C during the night. The plants used were 3 to 5 weeks old. 5 g, fresh weight, of leaves were selected at random from the plants. 2 ml of test material was added to abaxial leaf surfaces by dropwise application so as to cover them as evenly as possible. This quantity represented an excess when compared to the lipid content of the plant cells. The leaves were immediately immersed in 100 ml of deionized water at 25 deg C in a glass flask. The leakage of salts from the leaves was inferred from changes in conductance across a dip-type bright platinum electrode inserted into the water over a period of 6 to 7 hours. The flasks were sealed with parafilm and the contents shaken before each reading. Data are based on 3 replicates.

**Reliability**

A final conductance value representing the total electrolytes released from the leaves was taken after the suspensions were boiled.

: (4) not assignable  
There is insufficient documentation on the samples and results to rate this study for reliability. The relevance of the method of exposure and adding the test substance to the abaxial surface of the leaf to mimic the effect of applying agrochemical sprays is not clear.

07.07.2008

(2)

**Species**: other terrestrial plant: *Helianthus annuus***Endpoint**

: other: leaf damage, electrolyte loss

**Exposure period**

: 0 day(s)

**Unit**

: mg/l

**Method**

: other

**Year**

:

**GLP**

: no

**Test substance**

: n-Heptane; CAS #142-82-5; &gt;99.5% pure

**Result**

: The rate at which electrolytes were lost from sunflower leaves exposed to n-heptane was 160 min<sup>-1</sup> x E4. The rate at which electrolytes were lost from untreated leaves was 16 min<sup>-1</sup> x E4. A second study with n-heptane resulted in a rate of 170 min<sup>-1</sup> x E4. The rate at which electrolytes were lost from untreated leaves in the second study was 17 min<sup>-1</sup> x E4. The effects on cell membrane permeability as indicated by conductance agreed with the known susceptibility of different plant species towards hydrocarbons applied in the liquid state. The effect on the final conductance values and electrolyte concentration was similar for the hydrocarbons tested.

**Test condition**

Electrolyte loss from leaves exposed to a liquid hydrocarbon solvent coating is explained as resulting from the destruction of plant cell membrane permeability and the release of internal cell electrolytes.

: Sunflower plants, *Helianthus annuus*, were grown in a nutrient solution under controlled temperatures of 23 deg C during the day and 13 deg C during the night. The plants used were 3 to 5 weeks old. 5 g, fresh weight, of leaves were selected at random from the plants. 2 ml of test material was added to abaxial leaf surfaces by dropwise application so as to cover them as evenly as possible. This quantity represented an excess when compared to the lipid content of the plant cells. The leaves were immediately immersed in 100 ml of deionized water at 25 deg C in a glass flask. The leakage of salts from the leaves was inferred from changes in conductance across a dip-type bright platinum electrode inserted into the water over a period of 6 to 7 hours. The flasks were sealed with parafilm and the contents shaken before each reading. Data are based on 3 replicates.

A final conductance value representing the total electrolytes released from

**Reliability**

: the leaves was taken after the suspensions were boiled.  
(4) not assignable  
There is insufficient documentation on the samples and results to rate this study for reliability. The relevance of the method of exposure and adding the test substance to the abaxial surface of the leaf to mimic the effect of applying agrochemical sprays is not clear.

07.07.2008

(2)

### 4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

### 4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

### 4.7 BIOLOGICAL EFFECTS MONITORING

### 4.8 BIOTRANSFORMATION AND KINETICS

### 4.9 ADDITIONAL REMARKS



## 5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

## 5.1.1 ACUTE ORAL TOXICITY

## 5.1.2 ACUTE INHALATION TOXICITY

Type : LC50  
Value : > 29.29 mg/l  
Species : rat  
Strain : Sprague-Dawley  
Sex : male/female  
Number of animals : 10  
Vehicle : other: none  
Doses : 29.29 mg/L (17940 ppm)  
Exposure time : 4 hour(s)  
Method : other: Similar to OECD guideline 403  
Year : 1982  
GLP :  
Test substance : other TS: n-Heptane (CAS # 142-82-5)  
Remark : Animals were exposed to n-heptane vapor for 4 hours at a concentration of 29.29 mg/L (nominal) or 17937.5 ppm (mean analytical).  
Result : There was no mortality during the course of the study. A slight reduction of mean male body weights was noted on day 2 post exposure but males recovered by day 4. All animals appeared normal throughout the study and at terminal necropsy with the exception of one female observed with enlarged mandibular lymph nodes on the right side.  
Conclusion : n-Heptane has a low order of toxicity by the inhalation route of exposure.  
Reliability : (2) valid with restrictions  
07.07.2008 (11)

## 5.1.3 ACUTE DERMAL TOXICITY

## 5.1.4 ACUTE TOXICITY, OTHER ROUTES

## 5.2.1 SKIN IRRITATION

## 5.2.2 EYE IRRITATION

## 5.3 SENSITIZATION

## 5.4 REPEATED DOSE TOXICITY

Type :  
Species : rat  
Sex : male/female  
Strain : Sprague-Dawley  
Route of admin. : inhalation

## 5. Toxicity

**Id** 108-08-7

**Date** 30.06.2008

<b>Exposure period</b>	: 6 hours/day
<b>Frequency of treatm.</b>	: 5 days/week for 26 weeks
<b>Post exposure period</b>	: 2-week post exposure recovery period
<b>Doses</b>	: 0, 500, 2000 and 4000 ppm
<b>Control group</b>	: yes
<b>NOAEL</b>	: = 2970 ppm
<b>Method</b>	: other: similar to OECD guideline 413
<b>Year</b>	: 1980
<b>GLP</b>	:
<b>Test substance</b>	: other TS: n-Heptane (CAS # 142-82-5)
<b>Remark</b>	: Animals were exposed to 0, 398 or 2970 ppm n-heptane. Type: 26-week inhalation toxicity study Number of animals: 15/sex/dose group
<b>Result</b>	: There were no treatment-related deaths during the study. The only treatment-related observations were labored breathing or rapid breathing and slight prostration during the first week of study during exposure only, and anogenital fur and dry rales during weekly observations. The in chamber signs were generally more numerous and severe in the higher dose group and appeared to abate by the second week of the study.  No treatment-related effects were observed for body weight, hematology or urinalysis. Serum alkaline phosphatase was significantly elevated in female high dose rats and slightly elevated in low dose females. All other clinical chemistry values appeared normal with the exception of one male high level rat whose serum glutamic pyruvic transaminase and serum alkaline phosphatase levels were markedly elevated when compared to all other male rats on test. Proteinuria, elevated specific gravity and ketones were observed but do not appear to be related to treatment.
<b>Conclusion</b>	: The effects observed are consistent with acute CNS depression and generally abated by the second week of study. Under the conditions of this study, the LOAEL for acute CNS depression is 2,970 ppm and the NOAEL for systemic toxicity is 2,970 ppm.
<b>Reliability</b> 08.07.2008	: (2) valid with restrictions (1)
<b>Type</b>	:
<b>Species</b>	: rat
<b>Sex</b>	: male
<b>Strain</b>	: Sprague-Dawley
<b>Route of admin.</b>	: inhalation
<b>Exposure period</b>	: 9 hours/day
<b>Frequency of treatm.</b>	: 5 days/week for 7, 14 or 30 weeks
<b>Post exposure period</b>	: None. Animals were sacrificed at 7, 14 or 30 weeks.
<b>Doses</b>	: 0, 1500 ppm
<b>Control group</b>	: other: yes, omitted the second air flow
<b>NOAEL</b>	: > 1500 - ppm
<b>Method</b>	: other: none specified
<b>Year</b>	: 1981
<b>GLP</b>	:
<b>Test substance</b>	: other TS: n-Heptane (CAS # 142-82-5)
<b>Remark</b>	: Only males and one dose group were used. The primary objective of this study was to assess the appearance of polyneuropathy and urinary metabolites in rats following exposure to analytical grade solvents frequently used in Italian shoe factories. Nerve tissue was examined microscopically.  Body weights were analyzed by a two-way analysis of variance and Student's t test for the comparison of two slopes. Type: 30-week inhalation neurotoxicity study Number of animals: 6-9 males/dose group
<b>Result</b>	: None of the animals developed signs of neuropathy. There were no differences in weight gain of rats (30 weeks) compared to controls.

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Date

### Conclusion

Reliability  
08.07.2008

- Differences between mean values for hindlimb spreads observed in treated animals and controls were not statistically significant. However, authors note that in their hands, the test employed turned out to be scarcely effective due to high individual variability. No histological signs of giant axonal degeneration were noted in rats treated at 1500 ppm (30 weeks).
- : Under the conditions of this test, inhalation of n-heptane at 1500 ppm did not induce neuropathy in rats.
- : (2) valid with restrictions

(6)

### 5.5 GENETIC TOXICITY 'IN VITRO'

- Type : Bacterial reverse mutation assay
- System of testing : Salmonella typhimurium and Escherichia coli
- Test concentration : Doses ranging from 3.91 to 250 ug/ml
- Cycotoxic concentr. : 500 µg/ml
- Metabolic activation : with and without
- Result : negative
- Method : other: No specific method or guideline was noted.
- Year : 1982
- GLP :
- Test substance : other TS: heptane (CAS # 142-82-5)
- Remark : GLP: Quality assurance statement

Strains tested: Salmonella typhimurium tester strains TA98, TA100, TA1535, TA1537, TA1538; Escherichia coli strains WP2, WP uvr A

Test substance concentrations: 3.91, 7.81, 15.6, 31.3, 62.5, 125, 250 mg/ml

Metabolic activation: With and without (S9 fraction mix of livers from Aroclor 1254 pretreated rats)

Vehicle: Tween 80/ethanol

Positive Controls: benzo[a]pyrene, 4-nitroquinoline-N-oxide, sodium azide, neutral red, potassium dichromate.

Cytotoxicity study: A toxicity screening test conducted prior to the full assay indicated cytotoxicity at 500 mg/ml with and without metabolic activation.

### Result

- The cultures were incubated at 37°C for 48-72 hours in a sealed container before the revertant colonies were counted. Pre-incubation method was used to limit evaporation of test material.
- : The addition of heptane at amounts up to 250 mg per ml to cultures of Escherichia coli WP2 and WP2 uvr A, Salmonella typhimurium TA 1535, TA 1537, TA 1538, TA 98, and TA 100 did not lead to an increase in the reverse gene mutation frequency in any of these strains, either in the presence or in the absence of rat liver S9 fraction. In one assay with Escherichia coli WP2 in the presence of S9 fraction a greater than 2.5 fold increase over control values was seen at 15.6 and 31.3 mg per ml. This increase was not dose-related nor repeated in replicate assays and was therefore not considered to be a compound-related effect.

Conclusion  
Reliability  
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- : Under the conditions of this study, the test material was not mutagenic.
- : (1) valid without restriction

(3)

- Type : other: Mitotic gene conversion assay
- System of testing : Yeast
- Test concentration : Doses ranging from 0.01 to 5.0 mg/ml

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**Cycotoxic concentr.** :  
**Metabolic activation** : with and without  
**Result** : negative  
**Method** : other: No specific method or guideline was noted  
**Year** : 1982  
**GLP** :  
**Test substance** : other TS: heptane (CAS # 142-82-5)  
**Remark** : GLP: Quality assurance statement  
Strains tested: *Saccharomyces cerevisiae* JD1

Test substance concentrations: 0.01, 0.1, 0.5, 1.0, 5.0 mg/ml

Metabolic activation: With and without (S9 fraction mix of livers from Aroclor 1254 pretreated rats)

Vehicle: Tween 80/ethanol

Positive Controls: 4-nitroquinoline-N-oxide, cyclophosphamide

After 18-hour incubation at 30°C the cultures were placed onto the appropriate culture media for the selection of prototrophic colonies. After three days incubation at 30°C the numbers of prototrophic colonies were counted.

**Result** : Exposure of *Saccharomyces cerevisiae* JD1 to heptane at concentrations up to 5.0 mg/ml did not result in a consistent increase in the rate of mitotic gene conversion, either in the presence or in the absence of rat liver S9 fraction.

**Conclusion** : Under the conditions of this study, the test material was not genotoxic.  
**Reliability** : (1) valid without restriction

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(3)

**Type** : other: Chromosome aberration assay  
**System of testing** : Rat Liver (RL4) cells  
**Test concentration** : 2.5, 5, 10 ug/ml  
**Cycotoxic concentr.** : 20 ug/ml (100% cytotoxicity), 10 ug/ml (0% cytotoxicity)  
**Metabolic activation** :  
**Result** : negative  
**Method** : other: No specific method or guideline was noted  
**Year** : 1982  
**GLP** :  
**Test substance** : other TS: heptane (CAS # 142-82-5)  
**Remark** : GLP: Quality assurance statement  
Vehicle: Tween 80/ethanol

Positive Controls: 7,12-Dimethylbenzanthracene (DMBA)

Cultured rat liver cells were grown and treated on glass microscopic slides contained in 100-ml volume glass Leighton tubes. After 22-hour exposure to test compound or vehicle, colcemid was added to each culture. After further 2 hours, the slides were removed, subjected to hypotonic treatment followed by fixation (methanol:acetic acid, 3:1) and stained with Giemsa. The preparations were randomly coded and 100 cells from each culture were analyzed microscopically.

**Result** : In one culture exposed to 10 mg/ml of heptane a total of 7 chromatid gaps were seen; this increased the frequency to 0.024 gaps per cell which, although greater than the vehicle control frequency, was not accompanied by an increase in any other type of aberration and is not considered to be a compound-related effect. Thus there was no significant or dose-related increase of chromosome damage in any of the culture exposed to heptane. Cultures exposed to the positive control material, DMBA, showed a marked increase in the frequency of chromosome damage.

**Conclusion** : Under the conditions of this study, the test material was not clastogenic.

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**Reliability** : (2) valid with restrictions  
08.07.2008

(3)

### 5.6 GENETIC TOXICITY 'IN VIVO'

### 5.7 CARCINOGENICITY

#### 5.8.1 TOXICITY TO FERTILITY

#### 5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

#### 5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

### 5.9 SPECIFIC INVESTIGATIONS

### 5.10 EXPOSURE EXPERIENCE

### 5.11 ADDITIONAL REMARKS

### 6.1 ANALYTICAL METHODS

### 6.2 DETECTION AND IDENTIFICATION

7.1 FUNCTION

7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED

7.3 ORGANISMS TO BE PROTECTED

7.4 USER

7.5 RESISTANCE

**8.1 METHODS HANDLING AND STORING**

**8.2 FIRE GUIDANCE**

**8.3 EMERGENCY MEASURES**

**8.4 POSSIB. OF RENDERING SUBST. HARMLESS**

**8.5 WASTE MANAGEMENT**

**8.6 SIDE-EFFECTS DETECTION**

**8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER**

**8.8 REACTIVITY TOWARDS CONTAINER MATERIAL**



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**Date** 30.06.2008

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### 10.1 END POINT SUMMARY

### 10.2 HAZARD SUMMARY

### 10.3 RISK ASSESSMENT

RECEIVED  
10 JUL 30 AM 3:46

## I U C L I D

## Data Set

Existing Chemical : ID: 110-54-3  
CAS No. : 110-54-3  
EINECS Name : hexane  
EC No. : 203-777-6  
TSCA Name : Hexane  
Molecular Formula : C<sub>6</sub>H<sub>14</sub>

Producer related part  
Company : ExxonMobil Biomedical Sciences Inc.  
Creation date : 30.06.2008

Substance related part  
Company : ExxonMobil Biomedical Sciences Inc.  
Creation date : 30.06.2008

Status :  
Memo : U.S. EPA - HPV Challenge Program

Printing date : 12.08.2008  
Revision date :  
Date of last update : 12.08.2008

Number of pages : 37

Chapter (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10  
Reliability (profile) : Reliability: without reliability, 1, 2, 3, 4  
Flags (profile) : Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE),  
Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

## 1. General Information

**Id** 110-54-3  
**Date** 30.06.2008

### 1.0.1 APPLICANT AND COMPANY INFORMATION

### 1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

### 1.0.3 IDENTITY OF RECIPIENTS

### 1.0.4 DETAILS ON CATEGORY/TEMPLATE

### 1.1.0 SUBSTANCE IDENTIFICATION

#### 1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type	:	
Substance type	:	petroleum product
Physical status	:	liquid
Purity	:	
Colour	:	
Odour	:	

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#### 1.1.2 SPECTRA

### 1.2 SYNONYMS AND TRADENAMES

### 1.3 IMPURITIES

### 1.4 ADDITIVES

### 1.5 TOTAL QUANTITY

#### 1.6.1 LABELLING

#### 1.6.2 CLASSIFICATION

#### 1.6.3 PACKAGING

## 1. General Information

**Id** 110-54-3  
**Date** 30.06.2008

### 1.7 USE PATTERN

#### 1.7.1 DETAILED USE PATTERN

#### 1.7.2 METHODS OF MANUFACTURE

### 1.8 REGULATORY MEASURES

#### 1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

#### 1.8.2 ACCEPTABLE RESIDUES LEVELS

#### 1.8.3 WATER POLLUTION

#### 1.8.4 MAJOR ACCIDENT HAZARDS

#### 1.8.5 AIR POLLUTION

#### 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES

#### 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS

#### 1.9.2 COMPONENTS

#### 1.10 SOURCE OF EXPOSURE

#### 1.11 ADDITIONAL REMARKS

#### 1.12 LAST LITERATURE SEARCH

#### 1.13 REVIEWS

## 2. Physico-Chemical Data

Id 110-54-3

Date

### 2.1 MELTING POINT

**Value** : = -95.3 °C  
**Sublimation** :  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** : other TS: hexane; (CAS #110-54-3)  
**Test substance** : CAS #110-54-3; hexane; purity is unknown.  
**Reliability** : (2) valid with restrictions  
The CRC Handbook of Chemistry and Physics is a peer reviewed publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.  
**Flag** : Critical study for SIDS endpoint  
30.06.2008 (20)

### 2.2 BOILING POINT

**Value** : = 68.7 °C at 1013 hPa  
**Decomposition** :  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** : other TS: hexane; (CAS #110-54-3)  
**Test substance** : CAS #110-54-3; hexane; purity is unknown.  
**Reliability** : (2) valid with restrictions  
The CRC Handbook of Chemistry and Physics is a peer reviewed publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.  
**Flag** : Critical study for SIDS endpoint  
30.06.2008 (20)

### 2.3 DENSITY

**Type** : density  
**Value** : = .659 g/cm³ at 20 °C  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** : other TS: hexane; (CAS #110-54-3)  
**Test substance** : CAS #110-54-3; hexane; purity is unknown.  
**Reliability** : (2) valid with restrictions  
Data supplied by the experimental database associated with EPISuite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-reviewed publication.  
**Flag** : Critical study for SIDS endpoint  
30.06.2008 (31)

#### 2.3.1 GRANULOMETRY

## 2.4 VAPOUR PRESSURE

Value : = 20.13 hPa at 25 °C  
 Decomposition :  
 Method :  
 Year :  
 GLP : no data  
 Test substance : other TS: hexane; (CAS #110-54-3)  
 Method : Method not specified.  
 Test substance : CAS #110-54-3; hexane; purity is unknown.  
 Reliability : (2) valid with restrictions  
 This robust summary has a reliability rating of 2 because the data were not reviewed for quality, however, the reference is from a peer-reviewed handbook.  
 Flag : Critical study for SIDS endpoint  
 30.06.2008 (6)

## 2.5 PARTITION COEFFICIENT

Partition coefficient : octanol-water  
 Log pow : = 3.9 at 20 °C  
 pH value :  
 Method :  
 Year :  
 GLP : no data  
 Test substance : other TS: hexane; (CAS #110-54-3)  
 Test substance : CAS #110-54-3; hexane; purity is unknown.  
 Reliability : (2) valid with restrictions  
 The value cited by the author is a recommended value based on a review of data retrieved from the literature. This robust summary has a reliability rating of 2 because there is insufficient information available on the method.  
 Flag : Critical study for SIDS endpoint  
 30.06.2008 (11)

## 2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water  
 Value : = 9.5 mg/l at 20 °C  
 pH value :  
 concentration : at °C  
 Temperature effects :  
 Examine different pol. :  
 pKa : at 20 °C  
 Description :  
 Stable :  
 Deg. product :  
 Method : other: no data  
 Year :  
 GLP : no data  
 Test substance : other TS: hexane; (CAS #110-54-3)  
 Test substance : CAS #110-54-3; hexane; purity is unknown.  
 Reliability : (2) valid with restrictions  
 This robust summary has a reliability rating of 2 because the data are from a peer-reviewed journal.  
 Flag : Critical study for SIDS endpoint  
 30.06.2008 (38)



## 2. Physico-Chemical Data

Id 110-54-3  
Date

2.6.2 SURFACE TENSION

2.7 FLASH POINT

2.8 AUTO FLAMMABILITY

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

2.13 VISCOSITY

2.14 ADDITIONAL REMARKS

## 3.1.1 PHOTODEGRADATION

Type	: air
Light source	:
Light spectrum	: nm
Relative intensity	: based on intensity of sunlight
Conc. of substance	: at 25 °C
<b>INDIRECT PHOTOLYSIS</b>	
Sensitizer	: OH
Conc. of sensitizer	: 1500000 molecule/cm <sup>3</sup>
Rate constant	: = .0000000000546 cm <sup>3</sup> /(molecule*sec)
Degradation	: = 50 % after 23.4 hour(s)
Deg. product	:
Method	: other (calculated): Calculated values using AOPWIN version 1.92, a subroutine of the computer program EPI Suite <sup>TM</sup> version 3.20
Year	:
GLP	:
Test substance	: other TS: hexane; (CAS #110-54-3)
Method	: Calculated values using AOPWIN version 1.92, a subroutine of the computer program EPI Suite <sup>TM</sup> version 3.20
<p>Indirect photodegradation, or atmospheric oxidation potential, is based on the structure-activity relationship methods developed by R. Atkinson under the following conditions:</p> <p>Temperature: 25°C</p> <p>Sensitizer: OH- radical</p> <p>Concentration of Sensitizer: 1.5E6 OH- radicals/cm<sup>3</sup></p>	
Remark	: Hexane has the potential to volatilize to air, based on a relatively high vapor pressure, where it is subject to atmospheric oxidation. In air, hexane can react with photosensitized oxygen in the form of hydroxyl radicals (OH-). The computer program AOPWIN (atmospheric oxidation program for Microsoft Windows) (EPI Suite <sup>TM</sup> , 2000) calculates a chemical half-life for a 12-hour day (the 12-hour day half-life value normalizes degradation to a standard day light period during which hydroxyl radicals needed for degradation are generated), based on an OH- reaction rate constant and a defined OH- concentration.
<p>Based on a 12-hour day, a rate constant of 5.46 E-12 cm<sup>3</sup>/molecule*sec, and an OH- concentration of 1.5 E6 OH-/cm<sup>3</sup>, hexane has a calculated half-life in air of 1.96 days or 23.4 hours of daylight.</p>	
Test substance	: CAS #110-54-3; hexane; purity is unknown.
Reliability	: (2) valid with restrictions
<p>The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data are calculated and not measured.</p>	
Flag	: Critical study for SIDS endpoint
30.06.2008	(37)
Deg. product	:
Method	:
Year	:
GLP	:
Method	: Technical discussion
Remark	: Direct photochemical degradation occurs through the absorbance of solar radiation by a chemical substance in aqueous solution. If the absorbed energy is high enough, then the resultant excited state of the chemical may undergo a transformation. A prerequisite for direct photodegradation is the ability of one or more bonds within a chemical to absorb ultraviolet (UV)/visible light in the 290 to 750 nm range. Light wavelengths longer than 750 nm do not contain sufficient energy to break chemical bonds, and

### 3. Environmental Fate and Pathways

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wavelengths below 290 nm are shielded from the earth by the stratospheric ozone layer (Harris, 1982).

An approach to assessing the potential for a substance to undergo photochemical degradation is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by constituent molecules (Zepp and Cline, 1977). Saturated and unsaturated hydrocarbons do not absorb light above 290 nm. Consequently, hexane is not subject to photolytic processes in the aqueous environment.

**Test substance** : CAS #110-54-3; hexane  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability of 2 because it is a technical discussion and not a study.  
**Flag** : Critical study for SIDS endpoint  
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#### 3.1.2 STABILITY IN WATER

**Type** : abiotic  
**t1/2 pH4** : at °C  
**t1/2 pH7** : at °C  
**t1/2 pH9** : at °C  
**Deg. product** :  
**Method** : other: Technical discussion  
**Year** :  
**GLP** : no data  
**Test substance** : other TS: hexane; (CAS #110-54-3)  
**Result** : Hydrolysis of an organic chemical is the transformation process in which a water molecule or hydroxide ion reacts to form a new carbon-oxygen bond. Chemicals with leaving groups that have a potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Gould, 1959). The lack of a suitable leaving group renders a compound resistant to hydrolysis. Hexane is resistant to hydrolysis because it lacks a functional group that is hydrolytically reactive and Harris (1982) identifies hydrocarbons as generally resistant to hydrolysis. Therefore, hydrolysis will not contribute to the removal of hexane from the environment.  
**Test substance** : CAS #110-54-3; hexane  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability of 2 because it is a technical discussion and not a study.  
**Flag** : Critical study for SIDS endpoint  
30.06.2008 (11) (14)

#### 3.1.3 STABILITY IN SOIL

#### 3.2.1 MONITORING DATA

#### 3.2.2 FIELD STUDIES

#### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

**Type** :  
**Media** : other: air - biota - sediment(s) - soil - water  
**Air** : % (Fugacity Model Level I)

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**Water** : % (Fugacity Model Level I)  
**Soil** : % (Fugacity Model Level I)  
**Biota** : % (Fugacity Model Level II/III)  
**Soil** : % (Fugacity Model Level II/III)  
**Method** : other: Calculation according Mackay, Level I  
**Year** :  
**Remark** : The EQC Level I is a steady state, equilibrium model that utilizes the input of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Physicochemical data used in the calculation:

Parameter	Value w/ Units
Molecular Weight	86.18
Temperature	25° C
Log Kow	3.90
Water Solubility	9.5 g/m3
Vapor Pressure	20,131 Pa
Melting Point	-95.3° C

**Result** : Using the Mackay Level I calculation, the following distribution is predicted for heptane:

%Distribution	Compartment
99.97	Air
0.02	Water
0.01	Soil
<0.01	Sediment
<0.01	Suspended Sediment
<0.01	Biota

**Test substance** : CAS #110-54-3; hexane  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data are calculated.

**Flag** : Critical study for SIDS endpoint

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(22)

**Type** : fugacity model level III  
**Media** : other  
**Air** : % (Fugacity Model Level I)  
**Water** : % (Fugacity Model Level I)  
**Soil** : % (Fugacity Model Level I)  
**Biota** : % (Fugacity Model Level II/III)  
**Soil** : % (Fugacity Model Level II/III)  
**Method** : other: Level III simulation using the Mackay Multimedia Environmental Model (Mackay, 2001)  
**Year** :  
**Method** : Level III simulation using the Mackay Multimedia Environmental Model (Mackay, 2001). Mass balances are calculated for the four bulk media of air (gas + aerosol), water (solution + suspended sediment + biota), soil, (solids + air + water), and sediment (solids + pore water). Equilibrium exists within, but not between media. Physical-chemical properties are used to quantify a chemical's behavior in an evaluative environment. Three types of chemicals are treated in this model: chemicals that partition into all media

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(Type 1), non volatile chemicals (Type 2), and chemicals with zero, or near-zero, solubility (Type 3). The model cannot treat ionizing or speciating substances. The Level III model assumes a simple, evaluative environment with user-defined volumes and densities for the following homogeneous environmental media (or compartments): air, water, soil, sediment, suspended sediment, fish and aerosols.

This model provides a description of a chemical's fate including the important degradation and advection losses and the intermedia transport processes. The distribution of the chemical between media depends on how the chemical enters the system, e.g. to air, to water, or to both. This mode of entry also affects persistence or residence time.

The rates of intermedia transport are controlled by a series of 12 transport velocities. Reaction half-lives are requested for all 7 media. The advective residence time selected for air also applies to aerosols and the residence time for water applies to suspended sediment and fish. The advective residence time of aerosols, suspended sediment and fish cannot be specified independently of the air and water residence times.

**Result**

: Output:

	Mass%	Emissions(kg/hr)
Air	21.3	1000
Water	63.3	1000
Soil	4.3	1000
Sediment	11.1	0

**Test condition**

: Physicochemical data used in the calculation:

Parameter	Value w/ Units
Molecular Weight	86.18
Temperature	25° C
Log Kow	3.90
Water Solubility	9.5 g/m3
Vapor Pressure	20,131 Pa
Melting Point	-95.3° C

Reaction Half Lives in hours as predicted using EPI Suite™:

Air (gaseous) 23.5  
Water (no susp. part.) 360  
Bulk Soil 720  
Bulk Sediment 7200

Environmental Properties (EQC standard environment)  
Dimensions (all defaults)  
Densities (all defaults)  
Organic carbon & Advection (all defaults)  
Transport Velocities (all defaults)

Emission and Inflows (defaults used)  
Air 1000 kg/hr  
Water 1000 kg/hr  
Soil 1000 kg/hr  
Sediment 0 kg/hr

**Test substance  
Conclusion**

: CAS #110-54-3; hexane

: The majority of hexane is calculated to partition into the water phase, with smaller but significant amounts into air, soil, and sediment based on the modeling parameters used in this calculation. Hexane is considered to be a Type 1 chemical with potential to partition into all environmental compartments.

**Reliability**

: (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are calculated.

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#### 3.3.2 DISTRIBUTION

#### 3.4 MODE OF DEGRADATION IN ACTUAL USE

#### 3.5 BIODEGRADATION

Type : aerobic  
Inoculum : activated sludge  
Contact time : 28 day(s)  
Degradation : 74 (±) % after 28 day(s)  
Result : other: readily biodegradable  
Deg. product :  
Method : OECD Guide-line 301 F "Ready Biodegradability: Manometric  
Respirometry Test"  
Year : 1996  
GLP : yes  
Test substance : other TS: heptane; (CAS #142-82-5)  
Result : Test material was readily biodegradable. Half-life was reached by day 10.  
By day 28, 74% degradation of the test material was observed. 10%  
biodegradation was achieved on day 4.  
By day 10, >60% biodegradation of positive control was observed, which  
met the guideline requirement. No excursions from the protocol were  
noted.  
Biodegradation was based on oxygen consumption and the theoretical  
oxygen demand of the test material as calculated using results of an  
elemental analysis of the test material.

Sample	% Degradation* (day 28)	Mean % Degradation (day 28)
Test Material	72.5, 74.0, 76.8	74.4
Na Benzoate	88.7, 88.9	88.8

\* replicate data

Test condition : Non acclimated activated sludge and test medium were combined prior to  
test material addition. Test medium consisted of glass distilled water and  
mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate,  
Calcium chloride).  
Test vessels were 1L glass flasks placed in a waterbath and electronically  
monitored for oxygen consumption.  
Test material was tested in triplicate, controls and blanks were tested in  
duplicate.  
Test material concentration was approximately 45 mg/L. Sodium benzoate  
(positive control) concentration was 50mg/L.  
Test temperature was 22 +/- 1 Deg C.  
All test vessels were stirred constantly for 28 days using magnetic stir bars  
and plates.  
Test substance : CAS #142-82-5; heptane; 99% pure.  
Conclusion : Heptane is readily biodegradable.  
Reliability : (1) valid without restrictions

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Type : aerobic

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<b>Inoculum</b>	: other: soil, non-adapted
<b>Contact time</b>	: 20 day(s)
<b>Degradation</b>	: 70 (±) % after 20 day(s)
<b>Result</b>	: other: readily biodegradable
<b>Deg. product</b>	:
<b>Method</b>	: other: Standard Methods for the Examination of Water and Waste Water
<b>Year</b>	: 1971
<b>GLP</b>	: no
<b>Test substance</b>	: other TS: heptane; (CAS #142-82-5)
<b>Result</b>	: 70% degradation was measured after 20 days incubation with an unacclimated inoculum. % Biodegradation of test substance after days: 2 days = 28 % 5 days = 63 % 10 days = 70 % 20 days = 70 %
<b>Test condition</b>	: American Public Health Association, Standard Methods for the Examination of Water and Waste Water, using 1.0 mg/l of test substance. Biodegradation was determined by measuring biological oxygen demand (BOD). Each 300 ml BOD bottle received 1.0 mg of heptane, 1.0 ml of a 1:10 suspension Hudson-Collamer silt loam soil in distilled water, and a mineral salts solution prepared as described in the test method. Bottles were incubated in the dark at 25C. The test substance was obtained from Aldrich Chemical Co.
<b>Test substance</b>	: CAS #142-82-5; heptane; 99% pure.
<b>Conclusion</b>	: Heptane is readily biodegradable.
<b>Reliability</b>	: (2) valid with restrictions A standard test method was used. The study was conducted prior to GLP.
30.06.2008	(12)
<b>Type</b>	: aerobic
<b>Inoculum</b>	: activated sludge
<b>Contact time</b>	: 28 day(s)
<b>Degradation</b>	: = 100 (±) % after 28 day(s)
<b>Result</b>	: Readily biodegradable
<b>Deg. product</b>	:
<b>Method</b>	: other: Modified MITI test (Comparable to OECD 301C)
<b>Year</b>	: 1992
<b>GLP</b>	: no data
<b>Test substance</b>	: CAS No. 110-54-3; hexane
<b>Test condition</b>	: Concentration of the test substance was 100 mg/l, with a concentration of inoculum of 30 mg/l. The source of the inoculum was non-acclimated activated sludge. Results of the study were based on BOD.
<b>Test substance</b>	: CAS No. 110-54-3; hexane; purity is unknown
<b>Reliability</b>	: (2) valid with restrictions The study was performed following acceptable guidelines, however, the data were not retrieved after a review for quality.
<b>Flag</b>	: Critical study for SIDS endpoint
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#### 3.6 BOD5, COD OR BOD5/COD RATIO

#### 3.7 BIOACCUMULATION

<b>Species</b>	: other: see remark
<b>Exposure period</b>	: at 25 °C
<b>Concentration</b>	:
<b>BCF</b>	: = 200

### 3. Environmental Fate and Pathways

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<b>Elimination</b>	:	
<b>Method</b>	:	other: calculation
<b>Year</b>	:	
<b>GLP</b>	:	no
<b>Test substance</b>	:	other TS: hexane; (CAS #110-54-3)
<b>Remark</b>	:	A log bioconcentration factor (BCF) of 2.30 is calculated (BCF = 200). With respect to a log Kow = 3.90, which was used to calculate the BCF, hexane in the aquatic environment is expected to have a low potential to bioaccumulate.
<b>Test substance</b>	:	CAS #110-54-3; hexane
<b>Reliability</b>	:	(2) valid with restrictions This robust summary has a reliability rating of 2 because the data are calculated and not measured.
<b>Flag</b>	:	Critical study for SIDS endpoint
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#### 3.8 ADDITIONAL REMARKS



## 4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type	:	
Species	:	other: fish
Exposure period	:	96 hour(s)
Unit	:	mg/l
LC50	:	= 1.0
Method	:	other: ECOSAR version 0.99h, US EPA
Year	:	
GLP	:	
Test substance	:	other TS: hexane; (CAS #110-54-3)
Method	:	ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.
		To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.
Result	:	Calculated 96-hr LC50 for fish = 1.0 mg/L
Test condition	:	Experimental water solubility = 9.5 mg/l @ 25°C (McAuliffe, 1996), log Kow = 3.90 @ 20°C (Hansch, et.al, 1995), and melting point = -95.3°C (Lide et al., 2003) were entered into the program.
Test substance	:	Class: Neutral organics
Reliability	:	CAS #110-54-3; hexane
	:	(2) valid with restrictions
		This robust summary has a reliability rating of 2 because the data are calculated and not measured.

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## 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type	:	static
Species	:	other: Daphnia
Exposure period	:	48 hour(s)
Unit	:	mg/l
EC50	:	= 2.1

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**Method** : other: based on discussions in GESAMP/MARPOL meetings held in 1973  
**Year** :  
**GLP** : no data  
**Method** : Individual treatment concentrations were prepared by mixing the test substance in freshwater for 24 hours in a conical flask. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were 250 ml conical flasks with 25 daphnids per flask. Two replicates of each treatment level and control were evaluated.

Organisms supplied by testing lab; age = <24 hours old; parents age = approximately 21 days old.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-119.).

**Result** : 48-hr EC50 for a daphnid = 2.1 mg/L  
**Test condition** : Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 20 Deg C.

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 1.0 to 10 mg/L.

**Test substance** : CAS #110-54-3; hexane  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline, OECD 202, and used acceptable methods to prepare exposure solutions.

**Flag** : Critical study for SIDS endpoint

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**Type** : semistatic  
**Species** : other: Gammarid (*Chaetogammarus marinus*)  
**Exposure period** : 96 hour(s)  
**Unit** : mg/l  
**LC50** : = .4

**Method** : other: Static Gammarid Acute Toxicity Test

**Year** :  
**GLP** : no data

**Method** : Individual treatment solutions were prepared by mixing the test substance in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24 hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8‰; length = 5 mm.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-119.).

**Result** : 96-hr LC50 for a gammarid = 0.4 mg/L  
**Test condition** : Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 15 Deg C. Natural seawater was used with a

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salinity of 2.8%

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 1.0 to 10 mg/L. Test solutions were analyzed only upon test initiation.

**Test substance** : CAS #110-54-3; hexane  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to prepare exposure solutions.

**Flag** : Critical study for SIDS endpoint  
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**Type** : semistatic  
**Species** : other: mysid shrimp (Mysidopsis bahia)  
**Exposure period** : 96 hour(s)  
**Unit** : mg/l  
**LC50** : = .4  
**Method** : other: Static Gammarid Acute Toxicity Test  
**Year** :  
**GLP** : no data  
**Method** : Individual treatment solutions were prepared by mixing the test substance in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24 hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8%; test organisms were approximately 4 weeks old, with lengths of approximately 6 mm.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-119.).

**Result** : 96-hr LC50 for a mysid = 0.4 mg/L  
**Test condition** : Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 20 Deg C. Natural seawater was used with a salinity of 2.8%

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L. Test solutions were analyzed only upon test initiation.

**Test substance** : CAS #110-54-3; hexane  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to prepare exposure solutions.

**Flag** : Critical study for SIDS endpoint

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**Type** :  
**Species** : other: Daphnia  
**Exposure period** : 48 hour(s)  
**Unit** : mg/l  
**LC50** : = 1.3  
**Method** : other: ECOSAR version 0.99h, US EPA  
**Year** :  
**GLP** :  
**Test substance** : other TS: hexane; (CAS #110-54-3)  
**Method** : ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

**Result** : Calculated 48-hr LC50 for a daphnid = 1.3 mg/L  
**Test condition** : Experimental water solubility = 9.5 mg/l @ 25°C (McAuliffe, 1996), log Kow = 3.90 @ 20°C (Hansch, et.al, 1995), and melting point = -95.3°C (Lide et al., 2003) were entered into the program.  
Class: Neutral organics  
**Test substance** : CAS #110-54-3; hexane  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data are calculated and not measured.

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### 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

**Species** : other algae: Green Alga  
**Endpoint** :  
**Exposure period** : 96 hour(s)  
**Unit** : mg/l  
**EC50** : = .9  
**ChV** : = .3  
**Method** : other: ECOSAR version 0.99h, US EPA  
**Year** :

<b>GLP</b>	:	
<b>Method</b>	:	<p>ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.</p> <p>To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.</p>
<b>Result</b>	:	<p>Calculated 96-hr EC50 for a green alga = 0.9 mg/L  Calculated 96-hr ChV for a green alga = 0.3 mg/L</p>
<b>Test condition</b>	:	<p>Experimental water solubility = 9.5 mg/l @ 25°C (McAuliffe, 1996), log Kow = 3.90 @ 20°C (Hansch, et.al, 1995), and melting point = -95.3°C (Lide et al., 2003) were entered into the program.  Class: Neutral organics</p>
<b>Test substance</b>	:	CAS #110-54-3; hexane
<b>Reliability</b>	:	<p>(2) valid with restrictions  This robust summary has a reliability rating of 2 because the data are calculated and not measured.</p>

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#### 4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

##### 4.5.1 CHRONIC TOXICITY TO FISH

##### 4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

##### 4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

##### 4.6.2 TOXICITY TO TERRESTRIAL PLANTS

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4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

## 5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

## 5.1.1 ACUTE ORAL TOXICITY

Type	: LD50	
Value	: 28,720 mg/kg	
Species	: rat	
Strain	: no data	
Sex	: no data	
Number of animals	:	
Vehicle	: no data	
Year	:	
GLP	:	
Remark	:	
Test substance	: CAS No. 110-54-3; hexane; purity is unknown	
Conclusion	: Hexane has a low order of toxicity by the oral route of exposure	
Reliability	: (2) valid with restrictions Original study not retrieved and reviewed. Data from reliable peer-reviewed source.	
Flag	: Critical study for SIDS endpoint	(17)
21.07.2008		
Type	: LD50	
Value	: =15,840 mg/kg	
Species	: rat	
Strain	: no data	
Sex	: no data	
Number of animals	:	
Vehicle	: no data	
Doses	:	
Method	: no data	
Year	:	
GLP	:	
Test substance	: CAS No. 110-54-3; hexane; purity is unknown	
Conclusion	: Hexane has a low order of toxicity by the oral route of exposure	
Reliability	: (2) valid with restrictions Original study not retrieved and reviewed. Data from reliable peer-reviewed source.	
Flag	: Critical study for SIDS endpoint	(17)
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Type	: LD50	
Value	: 29,700 mg/kg	
Species	: rat	
Strain	: no data	
Sex	: no data	
Number of animals	:	
Vehicle	: no data	
Doses	:	
Method	:	
Year	: 1970	
GLP	: pre-GLP	
Remark	: The oral LD50 in a 14-day old rat was 29,700 mg/kg. Symptoms included depressive effect on the central nervous system, salivation and soft faeces.	
Test substance	: CAS No. 110-54-3; hexane; purity is unknown	
Conclusion	: Hexane has a low order of toxicity by the oral route of exposure	
Reliability	: (2) valid with restrictions	

## 5. Toxicity

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Flag  
21.07.2008

Original study not retrieved and reviewed. Data from reliable peer-reviewed source.  
: Critical study for SIDS endpoint

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### 5.1.2 ACUTE INHALATION TOXICITY

Type : LC50  
Value : 48,000 ppm  
Species : rat  
Strain : no data  
Sex : no data  
Number of animals : no data  
Vehicle : other: none  
Doses : 1000 to 64000 ppm  
Exposure time : 4 hour(s)  
Method : other: Similar to OECD guideline 403  
Year :  
GLP :  
Remark : Animals were exposed to n-heptane vapor for 4 hours at concentrations of 1000 to 64000 ppm.  
Conclusion : n-Hexane has a low order of toxicity by the inhalation route of exposure.  
Reliability : (2) valid with restrictions  
Original study not retrieved and reviewed. Data from reliable peer-reviewed source.  
21.07.2008 (16)

### 5.1.3 ACUTE DERMAL TOXICITY

### 5.1.4 ACUTE TOXICITY, OTHER ROUTES

### 5.2.1 SKIN IRRITATION

Species : humans  
Sex : no data  
Number of animals :  
Duration of study :  
Doses :  
Method : no data  
Year : 1973  
GLP : pre-GLP  
Test substance : CAS No. 110-54-3; hexane; purity is unknown  
Result : Mild irritant  
Remark : Dermal exposure can lead to peripheral neuropathy in humans.  
Reliability : (2) valid with restrictions  
Original study not retrieved and reviewed. Data from reliable peer-reviewed source.  
Flag : Critical study for SIDS endpoint  
22.07.2008 (29)(34)(35)

### 5.2.2 EYE IRRITATION



## 5. Toxicity

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### 5.3 SENSITIZATION

### 5.4 REPEATED DOSE TOXICITY

Type :  
Species : rat  
Sex : male/female  
Number of animals : 15/sex/group  
Strain : F344  
Route of admin. : inhalation  
Exposure period : 6 hours/day  
Frequency of treatm. : 5 days/week for 13 weeks  
Doses : 0, 3000, 6500 and 10000 ppm  
Control group : yes  
Method : no data  
Year : 1984  
GLP :  
Test substance : CAS No. 110-54-3; hexane; purity is unknown  
Result : There were no n-hexane-related clinical signs of toxicity, effects on food consumption, ophthalmological findings, or changes in neurological function. However, there was a lowering of the urinary pH in high-dose males. There were increased organ/body weight ratios for liver, kidney, and testis in high-dose males and kidney in mid-dose males. Histopathological examination of the tibial nerves revealed paranodal axonal swelling in mid- and high-dose males.

Remark : n-Hexane is metabolized to 5-hydroxy-2-hexanone and 2,5-hexanedione *in vivo* (DiVincenzo *et al.*, 1976). These two metabolites are believed to be responsible for the neurotoxicities associated with n-hexane exposure. There is evidence to show that 2,5-hexanedione is more persistent in peripheral nerve tissue than the parent n-hexane (Bus *et al.*, 1981).

Reliability : (2) valid with restrictions  
Original study not retrieved and reviewed. Data from reliable peer-reviewed source.

08.08.2008

(3)(4)

Type :  
Species : rat  
Sex : male  
Strain : Wistar  
Route of admin. : inhalation  
Exposure period : 12 hours/day  
Frequency of treatm. : 7 days/week for 16 weeks  
Doses : 0, 500, 1200, 3000 ppm  
Control group : yes,  
NOAEL : 500 ppm  
Method : other: none specified  
Year : 1989  
GLP : no data  
Test substance : CAS No. 110-54-3; hexane; purity is unknown  
Remark : Motor nerve conduction velocity was measured in the tail nerve along with body weight before exposure and after 4, 8, 12, and 16 weeks of exposure. One animal from each group was sacrificed after 16 weeks of exposure for histopathological evaluation of the nerve fibers in the tail. In addition, nerve-specific proteins (i.e. enolase and  $\beta$ -S100), involved in processes such as cell-cell communication, cell growth, intracellular signal transduction, and development and maintenance of the central nervous system, were measured.

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<b>Result</b>	: Statistically significant reductions in body weight gain were observed in a for the mid-dose (at 12 weeks) and high-dose (at 8 weeks) rats. Neurological deficits (i.e. reduction in grip strength and comparative slowness of motion) in mid- and high-dose rats were noted from 12 weeks of exposure. No hind-limb paralysis was observed by the time of sacrifice. A reduction in motor nerve conduction velocity, statistically significant during weeks 8-16, was seen with mid and high-dose rats. In addition increased incidence of paranodal swelling, along with some evidence of demyelination and remyelination was present in the peripheral nerves with 1,200 and 3,000 ppm. These histopathologic findings were most severe in the high-dose group. Dose-dependent biochemical changes included reductions in nervous system-specific proteins, particularly the $\beta$ -S100 protein in tail nerve fibers which was reduced by approximately 75% at all dose levels.
<b>Conclusion</b>	: Under the conditions of this study, the NOAEL was 500 ppm based on the neurophysiologic deficits and histopathologic effects seen with 1,200 and 3,000 ppm.
<b>Reliability</b>	: (2) valid with restrictions Original study not retrieved and reviewed. Data from reliable peer-reviewed source.
08.08.2008	(18)
<b>Type</b>	:
<b>Species</b>	: mouse
<b>Sex</b>	: male /female
<b>Number of animals</b>	: 10/sex/group
<b>Strain</b>	: B6C3F1
<b>Route of admin.</b>	: inhalation
<b>Exposure period</b>	: 6 hours/day
<b>Frequency of treatm.</b>	: 5 days/week for 13 weeks
<b>Doses</b>	: 0, 500, 1000, 4000, 10000 ppm
<b>Control group</b>	: yes,
<b>NOAEL</b>	: 500 - ppm
<b>Method</b>	: other: none specified
<b>Year</b>	: 1991
<b>GLP</b>	: no data
<b>Test substance</b>	: CAS No. 110-54-3; hexane; purity is unknown
<b>Remark</b>	: Groups of 10 mice/sex/group were exposed to 0, 500, 1,000, 4,000, or 10,000 ppm n-hexane 6 hours/day, 5 days per week for the duration of the study. A second group of 10 mice was exposed at 1,000 ppm for 22 hours/day, 5 days/week for the duration of the study. Separate groups of 8 mice/sex/group received identical treatments but were subjected to neurobehavioral tests before the start of dosing then again after 6 and 13 weeks of exposure. Four males and four females were randomly selected from the 0, 1,000 ppm extended duration, and 10,000 ppm exposure groups for histopathological examination of the spinal cord and tibial nerves. Animals were observed daily for signs of clinical toxicity and weighed weekly.
<b>Result</b>	: A full necropsy was performed at sacrifice, weights of the major organs were recorded, and histopathological evaluations were carried out at term on a variety of excised organs and tissues. The liver was examined only in the males of all exposure groups. Animals exposed to 10,000 ppm n-hexane exhibited some signs of nasal irritation and all animals survived to term. Relative liver, kidney, and heart weights appeared to be increased compared with controls in exposed females. In addition, females exposed to 10,000 ppm 6 hours/day and 1,000 ppm for 22 hours/day exhibited neurobehavioral deficits with a reduction in locomotor activity. There was an increased incidence of paranodal axonal swelling in high-dose or

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extended exposure duration mice.

**Conclusion** : It was concluded that n-hexane caused minimal toxicity to the nervous system and/or respiratory system at 1,000 ppm and above indicating a NOAEL of 500 ppm.

**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Data from reliable peer-reviewed source.

08.11.2008

(28)

### 5.5 GENETIC TOXICITY 'IN VITRO'

**Type** : Ames test  
**System of testing** : Test species/strain: Salmonella typhimurium TA100, TA1535, TA97, TA98, TA1537  
**Test concentration** : 0, 0.001, 0.0033, 0.010, 0.033, 0.10, and 0.333 mg/plate  
**Cytotoxic concentr.** :  
**Metabolic activation** : with and without  
**Result** : negative  
**Method** : no data  
**Year** : 1986  
**GLP** : no data  
**Test Method** : *Salmonella typhimurium* (TA1535, TA1537, TA97, TA98 and TA100) were incubated with and without metabolic activation. Two metabolic activation systems were used, one with S9 rats livers and the other with Syrian hamster livers. Doses of hexane ranged from 0 to 0.333 mg/plate.

**Result** : The highest negative dose tested in any Salmonella typhimurium strain was 0.333 mg/plate. Some cultures exhibited slight clearing of the background bacterial lawn at the two highest doses tested.

Genotoxic effects:  
With metabolic activation: negative  
Without metabolic activation: negative

**Test substance** : CAS No. 110-54-3; hexane  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Data from reliable peer-reviewed source.

**Flag** : Critical study for SIDS endpoint

22.07.2008

(27)(28)

**Type** : Mouse Lymphoma Assay  
**System of testing** : L5178Y mouse lymphoma cells (TK locus)  
**Test concentration** : 40 to 180 µg/ml  
**Cycotoxic concentr.** :  
**Metabolic activation** : with and without  
**Result** : negative  
**Method** : Method equivalent to current guidelines  
**Year** : 1980  
**GLP** : no data  
**Method** : The ability of n-hexane to induce specific locus mutations at the TK locus in cultured L5178Y mouse lymphoma cells was evaluated in the presence and absence of Aroclor-induced rat liver S9 metabolic activation. Based on preliminary toxicity tests, 8 non-activated cultures were treated with 80, 90, 100, 110, 120, 130, 140, or 150 µg/ml which produced a range of 0 to 140% total growth. Eight activated cultures were treated with 40, 60, 80, 100, 120, 140, 160, or 180 µg/ml which produced a range of 0 to 22% total growth. The method used was equivalent to the guidelines.

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<b>Result</b>	: None of the non-activated or activated cultures produced mutant frequencies significantly greater than the solvent controls.	
<b>Test substance</b>	: CAS No. 110-54-3; n-hexane	
<b>Reliability</b>	: (2) valid with restrictions Original study not retrieved and reviewed. Data from reliable peer-reviewed source.	
<b>Flag</b> 11.08.2008	: Critical study for SIDS endpoint	(15)
<b>Type</b>	: Sister Chromatid Exchange Test	
<b>System of testing</b>	: Mouse bone marrow cells	
<b>Test concentration</b>	: 500, 1000, 2000 mg/kg	
<b>Cycotoxic concentr.</b>	: no data	
<b>Metabolic activation</b>	: with and without	
<b>Result</b>	: negative	
<b>Method</b>	: Method equivalent to current guidelines	
<b>Year</b>	: 1982	
<b>GLP</b>	: no data	
<b>Remark</b>	: No increase in the incidence of sister chromatid exchanges in <i>in vivo</i> mouse bone marrow cells was seen with intraperitoneal doses of 500, 1,000, or 2,000 mg/kg n-hexane.	
<b>Result</b>	: The dosed groups displayed slight increases in chromosomal aberrations, but this increase was not considered to be significant.	
<b>Test substance</b>	: CAS No. 110-54-3; hexane	
<b>Reliability</b>	: (2) valid with restrictions Original study not retrieved and reviewed. Data from reliable peer-reviewed source.	
<b>Flag</b> 12.08.2008	: Critical study for SIDS endpoint	(28)
<b>Type</b>	: Unscheduled DNA synthesis	
<b>System of testing</b>	: Human lymphocytes	
<b>Test concentration</b>	: 0.1 to 10 mM cyclohexane	
<b>Cycotoxic concentr.</b>	: no data	
<b>Metabolic activation</b>	: with and without	
<b>Result</b>	: negative	
<b>Method</b>	: Method equivalent to current guidelines	
<b>Year</b>	: 1983	
<b>GLP</b>	: no data	
<b>Remark</b>	: Human lymphocytes ( + or – S9 mix) were cultured for 4 hours in the presence or absence of hexane. The effects on the DNA synthesis were measured through cellular [ <sup>3</sup> H]TdR uptake.	
<b>Result</b>	: DNA synthesis was inhibited in human lymphocytes in the presence of concentrations of n-hexane from 10 <sup>-4</sup> – 10 <sup>-2</sup> M but only at cytotoxic concentrations	
<b>Test substance</b>	: CAS No. 110-54-3; hexane	
<b>Reliability</b>	: (2) valid with restrictions Original study not retrieved and reviewed. Data from reliable peer-reviewed source.	
<b>Flag</b> 22.07.2008	: Critical study for SIDS endpoint	(30)

### 5.6 GENETIC TOXICITY 'IN VIVO'

<b>Type</b>	: Micronucleus
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**Id** 110-54-3  
**Date** 30.06.2008

**Species** : mouse  
**Strain** : B6C3F1  
**Sex** : no data  
**Route of admin.** : Intraperitoneal injection  
**Result** : negative  
**Doses** :  
**Method** :  
**Year** :  
**GLP** : no data  
**Remark** :  
**Result** : n-Hexane did not induce chromosomal aberrations and micronuclei in bone marrow cells of B6C3F1 mice injected intraperitoneally.

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Data from reliable peer-reviewed source.

**Flag** : Critical study for SIDS endpoint  
22.07.2008

(32)

**Type** : Bone Marrow Cytogenetic Assay  
**Species** : rat  
**Strain** : Sprague Dawley  
**Sex** : Male  
**Number of animals** : Not specified  
**Route of admin.** : inhalation  
**Exposure period** : 6 hours/day for 5 days  
**Doses** : 0, 150, 300, 600 ppm  
**Method** : other: not specified  
**Year** : 1979  
**GLP** : no data  
**Remark** :  
**Result** : n-hexane increased the number of chromosomal mutations in albino rat bone marrow cells at all dose levels.

**Test substance** : CAS No. 110-54-3; hexane; purity unknown  
**Reliability** : (4) Unassignable  
**Flag** :  
22.07.2008

(15)

**Comment [anb1]:** Hazelton laboratories America, Inc. January – July 1979 study. Report was never issued only referenced in a subsequent report.

**Type** : Bone Marrow Cytogenetic Assay  
**Species** : rat  
**Strain** : Sprague Dawley  
**Sex** : Male  
**Number of animals** : 25  
**Route of admin.** : inhalation  
**Exposure period** : 6 hours/day for 5 days  
**Doses** : 0, 150, 300, 600 ppm  
**Method** : other: not specified  
**Year** : 1981  
**GLP** : no data  
**Remark** :  
**Result** : A slight, but significant, increase in the number of chromosomal aberrations induced by n-hexane in albino rat bone marrow cells was reported at all dose levels tested. The most frequently observed classes of aberrations were chromatid breaks and markers. These types of aberrations are indicative of damage sustained at any stage in mitosis. No significant differences in mean mitotic indices were observed between the groups.

**Test substance** : CAS No. 110-54-3; hexane; purity: assume 100% active ingredients  
**Reliability** : (1) valid without restriction  
**Flag** : Critical study for SIDS endpoint

22.07.2008

(15)

**5.7 CARCINOGENICITY****5.8.1 TOXICITY TO FERTILITY****5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY**

Type	: Developmental toxicity
Species	: rat
Sex	: female
Number of animals	: 30/group
Strain	: Sprague-Dawley
Route of admin.	: inhalation
Exposure period	: 14 days
Frequency of treatm.	: 20 hours/day
Doses	: 0, 200, 1000, or 5000 ppm
Control group	: yes
Method	: no data
Year	: 1987
GLP	: no data
Method	: The developmental toxicity of n-hexane was assessed using timed-pregnant (30 animals per group) and virgin (10 animals per group) Sprague-Dawley rats exposed to 0 (filtered air), 200, 1,000, and 5,000 ppm n-hexane vapor in inhalation chambers for 20 hours per day for a period of 14 consecutive days (Mast, 1987). Spermpositive females were exposed on gestation days (GD) 6-19 and virgins were exposed concurrently for 14 consecutive days.
Result	<p>: Maternal toxicity, manifested as a reduction in extra-gestational maternal weight gain, was observed at all exposure levels, and was statistically significant for the 5,000 ppm exposure group. Extra-gestational maternal weight gain (calculated from GD 0 to GD 20) relative to control animals was reduced for the 200, 1,000, and 5,000 ppm exposure groups. Cumulative weight gain (CWG) for dams in the 1,000 and 5,000 ppm exposure groups was significantly reduced with respect to controls by GD 20. The CWG for the 5,000 ppm was also significantly reduced with respect to controls by GD 13.</p> <p>Comparison of n-hexane exposed groups with the control group (0 ppm) indicated that gestational exposure to n-hexane did not result in an increase in the incidence of intrauterine deaths or in the incidence of fetal malformations. A statistically significant reduction in fetal body weight relative to controls was observed for males at the 1,000 and 5,000 ppm exposure levels. Female weights were also reduced with respect to controls for these exposure levels, but the reduction was statistically significant for only the 5,000 ppm group. Gravid uterine weight was also significantly less than controls for the 5,000 ppm exposure groups. A statistically significant increase in the mean percent incidence per litter of reduced ossification of sternebrae 1-4 was observed for the 5,000 ppm group, and was positively correlated with exposure concentration. This increased incidence of reduced ossification in the sternebrae, and the reduction in fetal body weight at the 5,000 ppm level, may have been inter-related manifestations of slight growth retardation.</p> <p>No major abnormalities were found in any of the fetuses. Variations</p>

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observed included dilated ureter, renal pelvic cavitation, supernumerary ribs, and reduced skeletal ossifications at several sites. The increase in mean percent incidence per litter of reduced ossification of sternebrae 1-4 was statistically significant for the highest exposure concentration, and the increase was positively correlated with increasing exposure concentration.

**Conclusion** : The NOAEL for developmental toxicity was 200 ppm.

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Data from reliable peer-reviewed source.

**Flag** : Critical study for SIDS endpoint  
18.07.2008

(24)

**Type** :  
**Species** : rat  
**Sex** : male  
**Number of animals** : 12 to 39/group  
**Strain** : Sprague-Dawley  
**Route of admin.** : inhalation  
**Exposure period** : 24 hours  
**Frequency of treatm.** : single  
**Doses** : 5000 ppm  
**Control group** : yes  
**Method** : no data  
**Year** : 1987  
**GLP** : no data  
**Method** : The effect of n-hexane on the male reproductive system when administered via the inhalation route was examined by exposing male Sprague-Dawley rats (12-39/group) to 5,000 ppm n-hexane in either a single 24 hour exposure, repeated 16 hour/day exposures for up to 8 days, or repeated 16 hour/day exposures, 6 hours/day for up to 6 weeks.

**Result** : Rats exposed to 5,000 ppm n-hexane displayed some evidence of neuropathy such as paralysis. Focal degeneration of spermatocytes and exfoliation of elongated spermatids was also observed in response to treatment. Early meiotic prophase spermatocytes and transitional spermatocytes as well as those undergoing meiotic metaphase appeared to be more susceptible to the action of n-hexane than pachytene spermatocytes. Rats exposed repeatedly to 5,000 ppm n-hexane over a 6-week period showed complete atrophy of the seminiferous tubules. In addition, a reduction in food consumption and body weight gain accompanied by signs of incipient neuropathy were seen with repeated n-hexane exposure. A wide range of testicular lesion did not complete resolve during a recovery period even though body weights and clinical symptoms improved.

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Data from reliable peer-reviewed source.

**Flag** : Critical study for SIDS endpoint  
11.08.2008

(7)

**Type** :  
**Species** : mouse  
**Sex** : female  
**Number of animals** : 33/group  
**Strain** : CD-1  
**Route of admin.** : inhalation  
**Exposure period** : 14 days

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<b>Frequency of treatm.</b>	: 20 hours/day
<b>Doses</b>	: 0, 200, 1000, or 5000 ppm
<b>Control group</b>	: yes
<b>Method</b>	: no data
<b>Year</b>	: 1988
<b>GLP</b>	: no data
<b>Method</b>	: Timed-pregnant (~33 females per group) and virgin (10 females per group) Swiss (CD-1) mice were exposed to 0, 200, 1,000, and 5,000 ppm n-hexane (99.2% purity) vapor in inhalation chambers, 20 h/day, for a period of 12 consecutive days. Plug-positive females were exposed on GD 6-17.
<b>Result</b>	<p>: Maternal body weight at sacrifice (GD 18) and total cumulative weight gain for dams in the 5,000 ppm exposure group were significantly reduced with respect to controls; however, this was due to an exposure correlated reduction in gravid uterine weight, not to a decrease in extragestational gain. An exposure-correlated decrease in the gravid uterine weight to extragestational weight gain ratio (significant for the 5,000 ppm group) occurred in the absence of an effect on placental weight.</p> <p>Gestational exposure to n-hexane resulted in an increase in the number of resorbed fetuses for all exposure groups relative to the control group; however, the increases were not directly correlated to exposure concentration. The differences were statistically significant for the 200-ppm group with respect to total intrauterine death (early plus late resorptions), and with respect to late resorptions for the 5,000 ppm group. A small, but statistically significant, reduction in female (but not male) fetal body weight relative to the control group was observed at the 5,000 ppm exposure level. There were no exposure-related increases in any individual fetal malformation or variation, nor was there any increase in the incidence of combined malformations or variations.</p> <p>Gestational exposure of CD-1 mice to n-hexane vapors appeared to cause a degree of concentration-related developmental toxicity in the absence of overt maternal toxicity, but the test material was not found to be teratogenic. This developmental toxicity was manifested as an increase in the number of resorptions per litter for all exposure levels, and as a decrease in the uterine: extra-gestational weight gain ratio at the 5,000 ppm exposure level. Because of the significant increase in the number of resorptions at the 200-ppm exposure level, a NOEL for developmental toxicity was not established for exposure of mice to 200, 1,000, or 5,000 ppm n-hexane vapors.</p>
<b>Conclusion</b>	: The LOAEL for developmental toxicity (in mice) was 200 ppm.
<b>Test substance</b>	: CAS No. 110-54-3; hexane; purity 99.2%
<b>Reliability</b>	: (2) valid with restrictions Original study not retrieved and reviewed. Data from reliable peer-reviewed source.
<b>Flag</b>	: Critical study for SIDS endpoint
11.08.2008	(26)

### 5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

### 5.9 SPECIFIC INVESTIGATIONS

### 5.10 EXPOSURE EXPERIENCE



## 5. Toxicity

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**Date** 30.06.2008

### 5.11 ADDITIONAL REMARKS

**6.1 ANALYTICAL METHODS**

**6.2 DETECTION AND IDENTIFICATION**

- 7.1 FUNCTION
- 7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED
- 7.3 ORGANISMS TO BE PROTECTED
- 7.4 USER
- 7.5 RESISTANCE

**8.1 METHODS HANDLING AND STORING****8.2 FIRE GUIDANCE****8.3 EMERGENCY MEASURES****8.4 POSSIB. OF RENDERING SUBST. HARMLESS****8.5 WASTE MANAGEMENT****8.6 SIDE-EFFECTS DETECTION****8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER****8.8 REACTIVITY TOWARDS CONTAINER MATERIAL**

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## 9. References

Id 110-54-3

Date 30.06.2008

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## 10. Summary and Evaluation

**Id** 110-54-3  
**Date** 30.06.2008

### 10.1 END POINT SUMMARY

### 10.2 HAZARD SUMMARY

### 10.3 RISK ASSESSMENT



RECEIVED  
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## I U C L I D

## Data Set

Existing Chemical	: ID: 110-82-7
CAS No.	: 110-82-7
EINECS Name	: Cyclohexane
EC No.	: 203-806-2
TSCA Name	: Cyclohexane
Molecular Formula	: C6H12
IUPAC Name	: Cyclohexane
Producer related part	
Company	: ExxonMobil Biomedical Sciences Inc.
Creation date	: 30.06.2008
Substance related part	
Company	: ExxonMobil Biomedical Sciences Inc.
Creation date	: 30.06.2008
Status	:
Memo	: U.S. EPA - HPV Challenge Program
Printing date	: 30.06.2008
Revision date	:
Date of last update	: 30.06.2008
Number of pages	: 41
Chapter (profile)	: Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10
Reliability (profile)	: Reliability: without reliability, 1, 2, 3, 4
Flags (profile)	: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

**1.0.1 APPLICANT AND COMPANY INFORMATION****1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR****1.0.3 IDENTITY OF RECIPIENTS****1.0.4 DETAILS ON CATEGORY/TEMPLATE****1.1.0 SUBSTANCE IDENTIFICATION****1.1.1 GENERAL SUBSTANCE INFORMATION**

Purity type	:	
Substance type	:	petroleum product
Physical status	:	liquid
Purity	:	
Colour	:	
Odour	:	

01.07.2008

**1.1.2 SPECTRA****1.2 SYNONYMS AND TRADENAMES****1.3 IMPURITIES****1.4 ADDITIVES****1.5 TOTAL QUANTITY****1.6.1 LABELLING****1.6.2 CLASSIFICATION****1.6.3 PACKAGING**

## 1. General Information

**Id** 110-82-7  
**Date** 30.06.2008

### 1.7 USE PATTERN

#### 1.7.1 DETAILED USE PATTERN

#### 1.7.2 METHODS OF MANUFACTURE

### 1.8 REGULATORY MEASURES

#### 1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

#### 1.8.2 ACCEPTABLE RESIDUES LEVELS

#### 1.8.3 WATER POLLUTION

#### 1.8.4 MAJOR ACCIDENT HAZARDS

#### 1.8.5 AIR POLLUTION

#### 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES

#### 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS

#### 1.9.2 COMPONENTS

### 1.10 SOURCE OF EXPOSURE

### 1.11 ADDITIONAL REMARKS

### 1.12 LAST LITERATURE SEARCH

### 1.13 REVIEWS

## 2.1 MELTING POINT

**Value** : = 6.6 °C  
**Sublimation** :  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** :  
  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
The CRC Handbook of Chemistry and Physics is a peer reviewed publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.  
  
**Flag** : Critical study for SIDS endpoint  
01.07.2008 (27)

## 2.2 BOILING POINT

**Value** : = 80.7 °C at 1013 hPa  
**Decomposition** :  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** :  
  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
The CRC Handbook of Chemistry and Physics is a peer reviewed publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.  
  
**Flag** : Critical study for SIDS endpoint  
01.07.2008 (27)

## 2.3 DENSITY

**Type** : density  
**Value** : = .774 g/cm<sup>3</sup> at 20 °C  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** :  
  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
Data supplied by the experimental database associated with EPISuite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-reviewed publication.  
  
**Flag** : Critical study for SIDS endpoint  
01.07.2008 (43)

## 2.3.1 GRANULOMETRY

## 2.4 VAPOUR PRESSURE

**Value** : = 12.9 hPa at 25 °C  
**Decomposition** :  
**Method** : other (measured): not specified  
**Year** :  
**GLP** : no data  
**Test substance** :  
  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
 Data supplied by the experimental database associated with EPISuite.  
 This robust summary has a reliability rating of 2 because the data was not  
 reviewed for quality, however, the reference is associated with a peer-  
 reviewed publication.  
  
**Flag** : Critical study for SIDS endpoint  
 01.07.2008 (4)

## 2.5 PARTITION COEFFICIENT

**Partition coefficient** : octanol-water  
**Log pow** : = 3.44 at 20 °C  
**pH value** :  
**Method** : other (measured): not specified  
**Year** :  
**GLP** : no data  
**Test substance** :  
  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
 Data supplied by the experimental database associated with EPISuite.  
 This robust summary has a reliability rating of 2 because the data was not  
 reviewed for quality, however, the reference is associated with a peer-  
 reviewed publication.  
  
**Flag** : Critical study for SIDS endpoint  
 01.07.2008 (14)

## 2.6.1 SOLUBILITY IN DIFFERENT MEDIA

**Solubility in** : Water  
**Value** : = 55 mg/l at 25 °C  
**pH value** :  
**concentration** : at °C  
**Temperature effects** :  
**Examine different pol.** :  
**pKa** : at 25 °C  
**Description** :  
**Stable** :  
**Deg. product** :  
**Method** : other: not specified  
**Year** :  
**GLP** : no data  
**Test substance** :  
  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
 Data supplied by the experimental database associated with EPISuite.

## 2. Physico-Chemical Data

Id 110-82-7  
Date

**Flag**

01.07.2008

This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-reviewed publication.  
: Critical study for SIDS endpoint

(30)

### 2.6.2 SURFACE TENSION

### 2.7 FLASH POINT

### 2.8 AUTO FLAMMABILITY

### 2.9 FLAMMABILITY

### 2.10 EXPLOSIVE PROPERTIES

### 2.11 OXIDIZING PROPERTIES

### 2.12 DISSOCIATION CONSTANT

### 2.13 VISCOSITY

### 2.14 ADDITIONAL REMARKS

## 3.1.1 PHOTODEGRADATION

**Type** : air  
**Light source** : Sun light  
**Light spectrum** : nm  
**Relative intensity** : based on intensity of sunlight  
**INDIRECT PHOTOLYSIS**  
**Sensitizer** : OH  
**Conc. of sensitizer** : 1500000 molecule/cm<sup>3</sup>  
**Rate constant** : = .0000000000848 cm<sup>3</sup>/(molecule\*sec)  
**Degradation** : = 50 % after 15.1 hour(s)  
**Deg. product** :  
**Method** : other (calculated): Calculated values using AOPWIN version 1.92, a subroutine of the computer program EPI SuiteTM version 3.20  
**Year** :  
**GLP** :  
**Test substance** :

**Method** : Calculated values using AOPWIN version 1.92, a subroutine of the computer program EPI SuiteTM version 3.20

Indirect photodegradation, or atmospheric oxidation potential, is based on the structure-activity relationship methods developed by R. Atkinson under the following conditions:

Temperature: 25°C

Sensitizer: OH- radical

Concentration of Sensitizer: 1.5E6 OH- radicals/cm<sup>3</sup>

**Remark** : Cyclohexane has the potential to volatilize to air, based on a relatively high vapor pressure, where it is subject to atmospheric oxidation.

In air, cyclohexane can react with photosensitized oxygen in the form of hydroxyl radicals (OH-). The computer program AOPWIN (atmospheric oxidation program for Microsoft Windows) (EPI SuiteTM, 2000) calculates a chemical half-life for a 12-hour day (the 12-hour day half-life value normalizes degradation to a standard day light period during which hydroxyl radicals needed for degradation are generated), based on an OH- reaction rate constant and a defined OH- concentration.

Based on a 12-hour day, a rate constant of 8.48 E-12 cm<sup>3</sup>/molecule\*sec, and an OH- concentration of 1.5 E6 OH-/cm<sup>3</sup>, cyclohexane has a calculated half-life in air of 1.3 days or 15.1 hours of daylight.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
 The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data are calculated and not measured.

**Flag** : Critical study for SIDS endpoint

01.07.2008

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**Type** : water  
**Light source** :  
**Light spectrum** : nm  
**Relative intensity** : based on intensity of sunlight  
**Test condition** : Direct photochemical degradation occurs through the absorbance of solar radiation by a chemical substance in aqueous solution. If the absorbed energy is high enough, then the resultant excited state of the chemical may undergo a transformation. A prerequisite for direct photodegradation is the ability of one or more bonds within a chemical to absorb ultraviolet (UV)/visible light in the 290 to 750 nm range.

Light wavelengths longer than 750 nm do not contain sufficient energy to break chemical bonds, and wavelengths below 290 nm are shielded from the earth by the stratospheric ozone layer (Harris, 1982).

An approach to assessing the potential for a substance to undergo photochemical degradation is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by constituent molecules (Zepp and Cline, 1977). Saturated and unsaturated hydrocarbons do not absorb light above 290 nm. Consequently, cyclohexane is not subject to photolytic processes in the aqueous environment.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability of 2 because it is a technical discussion and not a study.  
**Flag** : Critical study for SIDS endpoint  
01.07.2008 (15)(48)

### 3.1.2 STABILITY IN WATER

**Deg. product** :  
**Method** : other: Technical Discussion  
**Year** :  
**GLP** : no data  
**Test substance** :

**Result** : Hydrolysis of an organic chemical is the transformation process in which a water molecule or hydroxide ion reacts to form a new carbon-oxygen bond. Chemicals with leaving groups that have a potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Gould, 1959). The lack of a suitable leaving group renders a compound resistant to hydrolysis. Cyclohexane is resistant to hydrolysis because it lacks a functional group that is hydrolytically reactive and Harris (1982) identifies hydrocarbons as generally resistant to hydrolysis. Therefore, hydrolysis will not contribute to the removal of cyclohexane from the environment.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability of 2 because it is a technical discussion and not a study.  
**Flag** : Critical study for SIDS endpoint  
01.07.2008 (13)(16)

### 3.1.3 STABILITY IN SOIL

### 3.2.1 MONITORING DATA

### 3.2.2 FIELD STUDIES

### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

**Type** : fugacity model level I  
**Media** : other: air - biota - sediment(s) - soil - water  
**Air** : % (Fugacity Model Level I)



### 3. Environmental Fate and Pathways

Id 110-82-7

Date

**Water** : % (Fugacity Model Level I)  
**Soil** : % (Fugacity Model Level I)  
**Biota** : % (Fugacity Model Level II/III)  
**Soil** : % (Fugacity Model Level II/III)  
**Method** : other: Calculation according Mackay, Level I  
**Year** : 2003

**Method** : The EQC Level I is a steady state, equilibrium model that utilizes the input of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:  
Molecular mass = 84.16 g/mol  
Water solubility = 55 g/m<sup>3</sup>  
Vapour pressure = 12,919 Pa  
log Kow = 3.44  
Melting point = 6.6 deg C

**Result** :  
Air - 99.91%  
Water - 0.03%  
Soil - 0.06%  
Sediment - <0.01%  
Suspended Sed - <0.01%  
Biota - <0.01%

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data are calculated and not measured.

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**Type** : fugacity model level III  
**Media** : other: air - sediment(s) - soil - water  
**Air** : % (Fugacity Model Level I)  
**Water** : % (Fugacity Model Level I)  
**Soil** : % (Fugacity Model Level I)  
**Biota** : % (Fugacity Model Level II/III)  
**Soil** : % (Fugacity Model Level II/III)  
**Method** : other: Calculation according Mackay, Level III  
**Year** : 2003

**Method** : The EQC Level III model is a steady state model that is useful for determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, and sediment).

Input values used:

### 3. Environmental Fate and Pathways

Id 110-82-7

Date

Molecular mass = 84.16 g/mol  
Water solubility = 55 g/m<sup>3</sup>  
Vapour pressure = 12,919 Pa  
log Kow = 3.44  
Melting point = 6.6 deg C

Degradation half-lives:

Air - 15.1 hrs  
Water - 360 hrs  
Soil - 720 hrs  
Sediment - 7200 hrs

Environmental Properties (EQC standard environment)  
Dimensions (all defaults)  
Densities (all defaults)  
Organic carbon & Advection (all defaults)  
Transport Velocities (all defaults)

Emission and Inflows (defaults used)

Air 1000 kg/hr  
Water 1000 kg/hr  
Soil 1000 kg/hr  
Sediment 0 kg/hr

**Result**

: Output:

	Mass%	Emissions(kg/hr)
Air	15.8	1000
Water	67.3	1000
Soil	12.9	1000
Sediment	4.0	0

**Test substance**

: CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability**

: (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data are calculated and not measured.

**Flag**

: Critical study for SIDS endpoint

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#### 3.3.2 DISTRIBUTION

#### 3.4 MODE OF DEGRADATION IN ACTUAL USE

#### 3.5 BIODEGRADATION

**Type** : aerobic  
**Inoculum** : activated sludge  
**Contact time** : 28 day(s)  
**Degradation** : 77 (±) % after 28 day(s)  
**Result** : other: readily biodegradable  
**Deg. product** :  
**Method** : OECD Guide-line 301 F "Ready Biodegradability: Manometric Respirometry Test"  
**Year** : 1995  
**GLP** : yes  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

### 3. Environmental Fate and Pathways

Id 110-82-7

Date 30.06.2008

**Result** : Test material was readily biodegradable. Half-life was reached by day 20. By day 28, 77% degradation of the test material was observed. 10% biodegradation was achieved on day 13. By day 2, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were noted. Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an elemental analysis of the test material.

Sample	% Degradation* (day 28)	Mean % Degradation (day 28)
Test Material	79.8, 70.7, 80.2	76.9
Na Benzoate	91.0, 90.4	90.7

**Test condition** : \* replicate data  
: Non acclimated activated sludge and test medium were combined prior to test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate, Calcium chloride).  
Test vessels were 1L glass flasks placed in a waterbath and electronically monitored for oxygen consumption.  
Test material was tested in triplicate, controls and blanks were tested in duplicate.  
Test material concentration was approximately 34 mg/L. Sodium benzoate (positive control) concentration was 50mg/L.  
Test temperature was 22 +/- 1 Deg C.  
All test vessels were stirred constantly for 28 days using magnetic stir bars and plates.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown.  
**Conclusion** : Cyclohexane is readily biodegradable.  
**Reliability** : (1) valid without restrictions

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(9)

#### 3.6 BOD5, COD OR BOD5/COD RATIO

#### 3.7 BIOACCUMULATION

**Species** : other: see remark  
**Exposure period** : at 25 °C  
**Concentration** :  
**BCF** : = 89  
**Elimination** :  
**Method** : other: calculation  
**Year** :  
**GLP** : no  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Remark** : A log bioconcentration factor (BCF) of 1.95 is calculated (BCF = 89). With respect to a log Kow = 3.44, which was used to calculate the BCF, cyclohexane in the aquatic environment is expected to have a low potential to bioaccumulate.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because the data are calculated and not measured.

**Flag** : Critical study for SIDS endpoint

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#### 3.8 ADDITIONAL REMARKS

## 4.1 ACUTE/PROLONGED TOXICITY TO FISH

**Type** : Flow Through Acute Fish Toxicity Test  
**Species** : Pimephales promelas (Fish, fresh water)  
**Exposure period** : 96 hour(s)  
**Unit** : mg/l  
**LC50** : 4.53 measured/nominal  
**Limit test** :  
**Analytical monitoring** : no data  
**Method** :  
**Year** : 1987  
**GLP** : no data  
**Statistical Method** : Trimmed Spearman Karber Method  
**Test Conditions** : Treatment solutions were prepared by diluting a 37.9mg/L stock solution. Nominal cyclohexane treatment levels were 1.62, 3.24, 4.86, 6.48, 8.10 mg/L, which measured 2.00, 3.53, 4.84, 6.96 and 8.86mg/L, respectively.

Control/dilution water was EPA Duluth laboratory water.  
Ten fish were tested per treatment. Treatment volume = 250ml.

Test parameters were as follows: temperature = 25.2 Deg C (s.d. 0.14); dissolved oxygen (DO) = 7.2 mg/L (s.d. 0.38); pH = 7.5 (s.d. 0.10); fish age = 30 days old; fish mean wt. = 0.119 g; fish mean length = 20.5 mm; fish loading = 0.1190 g/L/day. Temperature, DO, and pH data from the treatment solutions were only provided as mean values with standard deviations.

Organism supplier was U.S. EPA Environmental Research Lab, Duluth, MN, USA.

**Results** : 96-hour LC50 = 4.53 mg/L (95% CI 3.96 to 5.18) based upon measured concentrations

Analytical method used was Gas-Liquid Chromatography.

Measured Conc. (mg/L)	Fish Total Mortality (@ 24, 48, 72, 96 hrs)*
Control	0, 0, 0, 0
2.00	0, 0, 0, 0
3.52	1, 1, 1, 1
4.84	0, 2, 3, 6
6.96	10, 10, 10, 10
8.86	10, 10, 10, 10

\* 10 fish added at test initiation

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Conclusion** : 96-hour LC50 = 4.53 mg/L (95% CI 3.96 to 5.18) based upon measured concentrations

**Reliability** : (1) valid without restrictions  
 Although a standard method was not cited, the testing procedures followed generally accepted fish acute toxicity guideline methods and sufficient information on testing method and conditions was available to rate this study as "reliable without restriction".

**Flag** : Critical study for SIDS endpoint  
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(12)

**Type** :  
**Species** : other: freshwater fish  
**Exposure period** : 96 hour(s)

## 4. Ecotoxicity

Id 110-82-7

Date

<b>Unit</b>	:	mg/l
<b>LC50</b>	:	= 2.8 calculated
<b>Method</b>	:	other: ECOSAR Computer Model
<b>Year</b>	:	
<b>GLP</b>	:	
<b>Test substance</b>	:	CAS No. 110-82-7; cyclohexane; purity is unknown
<b>Method</b>	:	ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.  To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.
<b>Result</b>	:	Calculated 96-hr LC50 for fish = 2.8 mg/L
<b>Test condition</b>	:	Experimental water solubility = 55.0 mg/l @ 25°C (McAuliffe, 1966), log Kow = 3.44 @ 20°C (Hansch, et.al, 1995), and melting point = 6.6°C (Lide, 2003) were entered into the program. Class: Neutral organics
<b>Test substance</b>	:	CAS No. 110-82-7; cyclohexane; purity is unknown
<b>Reliability</b>	:	(2) valid with restrictions The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data are calculated and not measured.
<b>Flag</b>	:	Critical study for SIDS endpoint
01.07.2008		(3)
<b>Type</b>	:	Fish Acute Toxicity Test
<b>Species</b>	:	Oncorhynchus mykiss (Fish, fresh water, marine)
<b>Exposure period</b>	:	96 hour(s)
<b>Unit</b>	:	mg/l
<b>LC50</b>	:	3.2 measured/nominal
<b>Limit test</b>	:	no
<b>Analytical monitoring</b>	:	no
<b>Method</b>	:	OECD 203 Fish Acute Toxicity Test
<b>Year</b>	:	2000
<b>GLP</b>	:	yes
<b>Statistical Method</b>	:	Binomial Method
<b>Test Conditions</b>	:	Individual Water Accomodated Fractions (WAF's) were prepared for each test treatment. The test substance was added volumetrically, via a syringe, to 19L of dilution water in a 20L glass carboy. The solutions were mixed for

24 hours at a vortex of  $\leq 10\%$  of the total depth. The test solutions were pumped from each mixing vessel into three replicates of 4.5L in 4.0L glass aspirator bottles (no headspace). Five fish were added to each test replicate and the replicates sealed. Daily renewals were performed by removing  $\sim 80\%$  of the test solution through the port at the bottom and refilling with fresh solution.

Test temperature was 14.5 Deg C., Lighting was 16 hours light : 8 hours dark with 562 to 728 Lux during full daylight periods.

Dissolved Oxygen at initiation ranged from 8.6 to 8.9 mg/L and from 7.1 to 8.1 mg/L in "old" solutions prior to renewals. The pH was ranged from 7.5 to 7.8 during the study. Fish were not fed during the study. Water hardness ranged from 92 to 96 mg/L as  $\text{CaCO}_3$ .

Fish Mean Wt. = 0.213g. Mean Total length = 3.1cm, Test Loading = 0.24 g of fish/L.

**Results** : LL50 = 3.2mg/L (CI 1.0 to 10.0), based upon nominal loading levels.

<u>Nominal Conc.</u>	<u>% Mortality @ 96 hr.</u>
Control	0
1 mg/L	0
10 mg/L	10
100 mg/L	10

Dissolved oxygen levels dropped below 60% of saturation in some of the treatments on Days 1 through 4 of the test. Since no mortality occurred in these treatments, the deviations are not believed to have affected the outcome of the study.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

No analytical monitoring of test concentrations was performed.

01.07.2008

(11)

#### 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

<b>Type</b>	:	
<b>Species</b>	:	Daphnia sp. (Crustacea)
<b>Exposure period</b>	:	48 hour(s)
<b>Unit</b>	:	mg/l
<b>EC50</b>	:	= 3.3 calculated
<b>Method</b>	:	other: ECOSAR Computer Model
<b>Year</b>	:	
<b>GLP</b>	:	
<b>Test condition</b>	:	ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be

calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Result** : Calculated 48-hr LC50 for a daphnid = 3.3 mg/L  
**Test condition** : Experimental water solubility = 55.0 mg/l @ 25°C (McAuliffe, 1966), log Kow = 3.44 @ 20°C (Hansch, et.al, 1995), and melting point = 6.6°C (Lide, 2003) were entered into the program.  
 Class: Neutral organics  
**Reliability** : (2) valid with restrictions  
 The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data are calculated and not measured.  
**Flag** : Critical study for SIDS endpoint  
 08.07.2008 (3)

**Type** : static  
**Species** : other: Daphnia  
**Exposure period** : 48 hour(s)  
**Unit** : mg/l  
**EC50** : = 0.9  
**Method** : other: based on discussions in GESAMP/MARPOL meetings held in 1973  
**Year** :  
**GLP** : no data  
**Method** : Individual treatment concentrations were prepared by mixing the test substance in freshwater for 24 hours in a conical flask. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were 250 ml conical flasks with 25 daphnids per flask. Two replicates of each treatment level and control were evaluated.

Organisms supplied by testing lab; age = <24 hours old; parents age = approximately 21 days old.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981.

Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-119.).

**Result** : 48-hr EC50 for a daphnid = 0.9 mg/L  
**Test condition** : Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 20 Deg C.

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 1.0 to 10 mg/L.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
 This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient



## 4. Ecotoxicity

Id 110-82-7

Date 30.06.2008

**Flag**  
30.06.2008

: Critical study for SIDS endpoint (44)

**Type** : semistatic  
**Species** : other: Gammarid (*Chaetogammarus marinus*)  
**Exposure period** : 96 hour(s)  
**Unit** : mg/l  
**LC50** : = 2.2  
**Method** : other: Static Gammarid Acute Toxicity Test  
**Year** :  
**GLP** : no data  
**Method** : Individual treatment solutions were prepared by mixing the test substance in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24 hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8‰; length = 5 mm.  
Statistical method:  
Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. *Water Res.*, 15:107-119.).

**Result** : 96-hr LC50 for a gammarid = 2.2 mg/L  
**Test condition** : Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 15 Deg C. Natural seawater was used with a salinity of 2.8‰

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 1.0 to 10 mg/L. Test solutions were analyzed only upon test initiation.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to prepare exposure solutions.

**Flag**  
30.06.2008

: Critical study for SIDS endpoint (44)

**Type** : semistatic  
**Species** : other: mysid shrimp (*Mysidopsis bahia*)  
**Exposure period** : 96 hour(s)  
**Unit** : mg/l  
**LC50** : = 2.2  
**Method** : other: Static Gammarid Acute Toxicity Test  
**Year** :  
**GLP** : no data  
**Method** : Individual treatment solutions were prepared by mixing the test substance in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were

allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24 hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8‰; test organisms were approximately 4 weeks old, with lengths of approximately 6 mm.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981.

Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-119.).

**Result** : 96-hr LC50 for a mysid = 2.2 mg/L  
**Test condition** : Dissolved oxygen was >50% saturation during the study. The pH was 7.5 to 8.3. Temperature was 20 Deg C. Natural seawater was used with a salinity of 2.8‰

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L. Test solutions were analyzed only upon test initiation.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Reliability** : (2) valid with restrictions  
 This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to prepare exposure solutions.

**Flag** : Critical study for SIDS endpoint

30.06.2008

(44)

#### 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

**Species** : other algae: *Pseudokirchneriella subcapitata*  
**Endpoint** :  
**Exposure period** : 72 hour(s)  
**Unit** : mg/l  
**EC50** : = 3.4 measured  
**Method** : OECD 201 - Alga, Growth Inhibition Test  
**Year** : 1998  
**GLP** : yes  
**Statistical Method** : Proc regression procedure of SAS, Anova procedure of SAS for NOEC  
**Test condition** : Individual test treatment solutions were prepared as Water Accommodated Fractions (WAFs). Test material was added to algal media via syringe in 2.0L aspirator bottles. The bottles were completely filled, no headspace. The mixing vessels were sealed with Teflon-covered stoppers and mixed on magnetic stir plates with teflon coated stir bars for 23.5 hours at room temperature. After mixing the solutions were allowed to settle for one hour and the WAF was removed from the bottom of the mixing vessel via the port and used for testing. Test vessels were 125ml glass Erlenmeyer flasks with approximately 140 ml of treatment solution and inoculated with algae. Test vessels were completely filled and sealed with glass stoppers. Samples were taken daily for cell counts. Six replicates were prepared for each treatment level. The initial algal concentration was  $1.0 \times 10^4$  cells/ml. All test replicates were placed on a shaker table at 100 oscillations per minute

during the study. Biomass was calculated as the area under the growth curve.

Nominal loading levels were 0.5, 1.4, 3.9, 11, and 31 mg/L

Test temperature was 24.6 Deg. C. Lighting was continuous at approximately 4200 Lux. The pH was 7.5 to 7.6 at test initiation and ranged from 8.4 to 8.7 at test termination.

Test treatments were analyzed by GC-FID. Measured values on Day 0 were 0.085, 0.719, 0.952, 2.940 and 4.425 mg/L. The test material was not detected in the control.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Result** : 72-hr EC50 for a green alga = 3.4 mg/L  
 72-hr NOEC for a green alga = 0.952 mg/L  
**Reliability** : (1) valid without restrictions

**Flag** : Critical study for SIDS endpoint  
 02.07.2008

(10)

**Species** : other algae: Pseudokirchneriella subcapitata  
**Endpoint** :  
**Exposure period** : 96 hour(s)  
**Unit** : mg/l  
**EC50** : = 2.2 calculated  
**ChV** : = 0.5 calculated  
**Method** : other: ECOSAR Computer Model  
**Year** :  
**GLP** :

**Test condition** : ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Result** : Calculated 96-hr EC50 for a green alga = 2.2 mg/L  
 Calculated 96-hr ChV for a green alga = 0.5 mg/L  
**Test condition** : Experimental water solubility = 55.0 mg/l @ 25°C (McAuliffe, 1966), log Kow = 3.44 @ 20°C (Hansch, et.al, 1995), and melting point = 6.6°C (Lide,

**Reliability**

2003) were entered into the program.  
Class: Neutral organics  
: (2) valid with restrictions  
The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data are calculated and not measured.

**Flag**

02.07.2008

: Critical study for SIDS endpoint

(3)

**4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA****4.5.1 CHRONIC TOXICITY TO FISH****4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES****4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS****4.6.2 TOXICITY TO TERRESTRIAL PLANTS****4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS****4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES****4.7 BIOLOGICAL EFFECTS MONITORING****4.8 BIOTRANSFORMATION AND KINETICS****4.9 ADDITIONAL REMARKS**

## 5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

## 5.1.1 ACUTE ORAL TOXICITY

**Type** : LD50  
**Value** : >5000 mg/kg  
**Species** : rat  
**Strain** : no data  
**Sex** : no data  
**Number of animals** :  
**Vehicle** : no data  
**Doses** :  
**Method** :  
**Year** :  
**GLP** :  
**Remark** :  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Conclusion** : Cyclohexane has a low order of toxicity by the oral route of exposure  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.  
**Flag** : Critical study for SIDS endpoint  
14.07.2008 (32)

**Type** : LD50  
**Value** : =29,800 mg/kg  
**Species** : rat  
**Strain** : no data  
**Sex** : no data  
**Number of animals** :  
**Vehicle** : no data  
**Doses** :  
**Method** :  
**Year** :  
**GLP** :  
**Remark** :  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Conclusion** : Cyclohexane has a low order of toxicity by the oral route of exposure  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.  
**Flag** : Critical study for SIDS endpoint  
14.07.2008 (7)

**Type** : LD50  
**Value** : 8,000 - 39,000 mg/kg  
**Species** : rat  
**Strain** : no data  
**Sex** : no data  
**Number of animals** :  
**Vehicle** : no data  
**Doses** :  
**Method** :  
**Year** : 1970  
**GLP** : pre-GLP  
**Remark** : The oral LD50 in a 14-day old rat, a young adult rat and an older rat was 8.0, 39.0 and 16.5 ml/kg, respectively (6,240, 30,420 and 12,870 mg/kg,

## 5. Toxicity

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Date

respectively). Symptoms included depressive effect on the central nervous system, salivation and soft faeces.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Conclusion** : Cyclohexane has a low order of toxicity by the oral route of exposure  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.  
**Flag** : Critical study for SIDS endpoint  
14.07.2008 (25)

### 5.1.2 ACUTE INHALATION TOXICITY

**Type** : LC50  
**Value** : > 32.88 mg/l  
**Species** : rat  
**Strain** : no data  
**Sex** : no data  
**Number of animals** :  
**Vehicle** : no data  
**Doses** :  
**Exposure time** : 4 hour(s)  
**Method** : other  
**Year** :  
**GLP** : no data  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Remark** : Single 4-hour exposure at one dose level of 9500 ppm (32.28 mg/l). No death occurred at level tested. Exposure-related symptoms noted during the exposure included tremors, hyperactivity, rapid respiration, and also hypoactivity.

**Result** :  
**Conclusion** : Cyclohexane has a low order of toxicity by the inhalation route of exposure  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.  
**Flag** : Critical study for SIDS endpoint  
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**Type** : LC50  
**Value** :  
**Species** : rabbit  
**Strain** : no data  
**Sex** : no data  
**Number of animals** :  
**Vehicle** : no data  
**Doses** :  
**Exposure time** : 1 hour(s)  
**Method** : other  
**Year** :  
**GLP** : pre-GLP  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Remark** : A 1 hour acute inhalation exposure of rabbits to cyclohexane vapor produced effects on the central nervous system, with exposure-related observations including convulsions, tremors, hyperactivity, rapid respiration, cyanosis and diarrhea; all the animals exposed to 89.6 mg/L died (26,000 ppm).  
**Result** :  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk

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Critical study for SIDS endpoint (45)

### 5.1.3 ACUTE DERMAL TOXICITY

Type : LD50  
Value : > 2000 mg/kg bw  
Species : rabbit  
Strain : no data  
Sex : no data  
Number of animals :  
Vehicle : no data  
Doses :  
Method : other  
Year :  
GLP : no data  
Test substance : CAS No. 110-82-7; cyclohexane; purity is unknown  
Result : No deaths or systemic symptoms were observed, a slight erythema and oedema were noted in a few animals.

Conclusion : Cyclohexane has a low order of toxicity by the dermal route of exposure  
Reliability : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

Flag 14.07.2008 : Critical study for SIDS endpoint (34)

### 5.1.4 ACUTE TOXICITY, OTHER ROUTES

### 5.2.1 SKIN IRRITATION

Species : rabbit  
Strain : no data  
Sex : no data  
Number of animals : 6  
Duration of study : 7 days  
Doses :  
Method : EEC Directive 83/467/EEC  
Year :  
GLP : yes  
Test substance : CAS No. 110-82-7; cyclohexane; purity is unknown  
Result : not irritating  
Remark : Test substance applied under a semi-occlusive dressing. Single application. Resulted in a 24-hr and 72-hr PI score of zero. Cyclohexane had no corrosive properties

Reliability : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

Flag 14.07.2008 : Critical study for SIDS endpoint (35)

Species : rabbit  
Strain : no data  
Sex : no data  
Number of animals : 6

## 5. Toxicity

**Id** 110-82-7  
**Date** 30.06.2008

**Duration of study** : 7 days  
**Doses** :  
**Method** : EEC Directive 83/467/EEC  
**Year** :  
**GLP** : yes  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Result** : not irritating  
**Remark** : Test substance applied under a semi-occlusive dressing. Single application. Resulted in a mean erythema score for 24-hr and 72-hr of 1.93.

A review of this study did however note that the erythematous reaction reached maximum severity at 5 days post-application (mean score 2.56). During this time, there was a gradual increase in dermal reaction for a further 144 h observation time (2.83). Overall, the irritation reactions were important and still present at the end of the study.

**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

**Flag** : Critical study for SIDS endpoint

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### 5.2.2 EYE IRRITATION

**Species** : rabbit  
**Strain** : no data  
**Sex** : no data  
**Number of animals** : 6  
**Vehicle** : none  
**Doses** : Single undiluted  
**Method** : Draize Test  
**Year** :  
**GLP** : yes  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Result** : slightly irritating  
**Remark** : After application of the test substance to one eye of the rabbit the eye was left unwashed.

At one hour post instillation, corneal opacity, involving up to 25% of the cornea, was noted in one rabbit and iritis was noted in another rabbit. Conjunctival redness was noted in five rabbits with conjunctival chemosis in one rabbit. All ocular lesions had cleared within 24 hours and no conjunctival discharge was noted in any of the six animals.

**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

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**Species** : rabbit  
**Strain** : no data  
**Sex** : no data  
**Number of animals** : 6  
**Vehicle** : none  
**Doses** : Single undiluted  
**Method** : Draize Test  
**Year** :



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**GLP** : yes  
**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown  
**Result** : slightly irritating  
**Remark** : After application of the test substance to one eye of the rabbit the eye was rinsed / washed.

**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

**Flag** : Critical study for SIDS endpoint  
14.07.2008 (37)

### 5.3 SENSITIZATION

**Type** : Dermal sensitization  
**Species** : Guinea pig  
**Strain** : no data  
**Sex** : male / female  
**Number of animals** : 9 male / 11 female  
**Vehicle** : ethanol / acetone  
**Doses** :  
**Method** : EC Annex V Method B6 - modified Buehler method  
**Year** : 1996  
**GLP** : yes  
**Test conditions** : Twenty guinea pigs (9 males and 11 females) were induced dermally with 10% cyclohexane (purity 99.98%) in ethanol and challenged with 10% cyclohexane in acetone. Concurrent negative controls (no cyclohexane) and positive controls (DNCB-0.1% in 50% ethanol at induction and 0.07% in acetone at challenge) were tested. It should be noted that a maximisation test was not required because of the very poor tolerance to intra-dermal injection of solvents. During the induction phase, the response ranged from no redness (14/20 animals) to very faint redness on some tested animals (6/20 animals with a slight reaction). A very faint redness was observed 24 hours after the challenge application in 1/20 tested animals, no reactions were observed in other tested animals or negative controls. The incidence of sensitisation among cyclohexane induced and challenged animals was 0/20. The incidence of sensitisation among the DNCB induced and challenged animals was 8/10.

A higher challenge concentration could have been chosen (15% in acetone did not produce any dermal irritation) and there were only a few animals with dermal reactions during the induction phase, these findings reduce the significance of this test.

**Test substance** : CAS No. 110-82-7; cyclohexane; (purity 99.98%)  
**Result** : not a dermal sensitizer  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

**Flag** : Critical study for SIDS endpoint  
14.07.2008 (8)

### 5.4 REPEATED DOSE TOXICITY

**Type** :  
**Species** : rat  
**Sex** : male/female

## 5. Toxicity

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**Number of animals** : 5/sex/group  
**Strain** : other: Crl:CD.BR  
**Route of admin.** : inhalation  
**Exposure period** : 9 days  
**Frequency of treatm.** : 6 hrs per day  
**Doses** : 3000, 6000, 9000 ppm  
**Control group** : yes  
**NOAEL** : = 3000 ppm  
**Method** : EPA test method  
**Year** : 1995  
**GLP** : yes  
**Test Methods** : In a two-week inhalation range finding study, Crl CD.BR rats (5/sex/group) were exposed (whole body in an exposure chamber) to 0-3,000-6,000 and 9,000 ppm (0-10,500-21,000 and 31,500 mg/m<sup>3</sup>) of cyclohexane (purity 99.97%). Nine exposures, each lasting six hours, were performed in total. The animals were weighed before treatment, clinical signs were checked before, during and after exposure, and common biochemical parameters and histological examinations were conducted at the end of the study.

For neurotoxicity assessment, the animals were checked for alerting behaviour in response to a standardised auditory stimulus at least three times during each exposure. They were also submitted to an abbreviated Functional Observational Battery (FOB) before and after exposure on two separate days (test days 4 and 11). This assessment was also performed prior to the initiation of exposures to establish baseline measurements.

During the FOB, the following parameters were assessed:

- in home cage: posture and palpebral closure,
- in open field: righting reflex, convulsions, gait characteristics, vocalisations, labored breathing, coordination, arousal and palpebral closure,
- during manipulations: approach and touch response, auditory response (clicker) and tail pinch.

This study was performed according to EPA guidelines and following EPA and OECD GLPs.

**Result** : A slight but significant decrease in body weight gain was observed in males treated with 9,000 ppm. Except for a minimal increase in mitotic index figures detected in the hepatocytes of males at 6,000 ppm and higher and in females at 9,000 ppm, no other treatment related findings were observed for systemic toxicity. In particular, no modification in absolute and relative liver weights was noted in these studies. Based on these findings, a NO(A)EL of 3,000 ppm (10,320 mg/m<sup>3</sup>) can be assumed for systemic toxicity. For neurotoxic effects, diminished responses to stimulus were observed from day 2 at 9,000 ppm and from 7 exposures at 6,000 ppm. No effect was observed in FOB. A NOAEL of 3,000 ppm (10,320 mg/m<sup>3</sup>) can be assumed for neurotoxic effects in rats. This study served as a range-finding study for a 90-day inhalation toxicity study. It should be noted that this value is very conservative because the effects are very slight and may be of adaptive nature; this is taken into account in the risk characterisation.

**Test substance** : CAS No. 110-82-7; cyclohexane; (purity 99.97%)  
**Reliability** : (1) valid without restriction  
**Flag** : Critical study for SIDS endpoint  
17.07.2008

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**Type** :  
**Species** : rat  
**Sex** : male/female

## 5. Toxicity

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- Number of animals** : 20/sex/group (controls & hi dose); 10/sex/group (intermediary doses)  
**Strain** : other: Crl:CD.BR  
**Route of admin.** : inhalation  
**Exposure period** : 13 weeks  
**Frequency of treatm.** : 6 hrs / day, 5 days / week (total 66 exposures)  
**Doses** : 500, 2000, 7000 ppm  
**Control group** : yes  
**NOAEL** :  
**Method** : EPA test method  
**Year** : 1995  
**GLP** : yes  
**Methods** : In a 13-week repeated dose toxicity study conducted using CD BR rats (20/sex/group for controls and high concentration and 10/sex/group for intermediary concentration groups) were exposed by whole body inhalation to cyclohexane at target concentrations of 0, 500, 2,000, and 7,000 ppm for 6 hours/day, 5 day/week for 13 weeks (66 exposures). Ten rats per month were allowed a one-month recovery period for control group and 7,000 ppm groups. After 45 and 90 days of exposure, blood and urine were collected for evaluation of clinical pathology parameters. Gross pathology, organ weight, macroscopic and microscopic examinations were performed at the end of the study.
- Result** : No treatment-related effects were observed on body weight, body weight gain, food consumption, urine analysis and clinical examinations. A slight decrease (not significant) in succinate dehydrogenase and lactate dehydrogenase was observed in males and females at 7,000 and 2,000 ppm at both sampling times. In males exposed to 7,000 ppm a slight increase in adrenals weight was observed at the end of the recovery period. This finding was not observed at the end of the 90 day exposure so the relevance and significance is questionable. In the 7,000 ppm group, a statistically significant increase in the relative liver weight with hepatic hypertrophy was observed in males (10/10), concurrent with an increase in the incidence of centrilobular hypertrophy in both sexes (9/10 males and 5/10 females). This finding was partially reversible in the one-month recovery period. For neurological effects, decreases in or absences of response to auditory stimulations were observed with a dose-response relationship from 500 ppm. In the 500 ppm group, there was a decrease in response on treatment days 61, 66, 67 and 68. In the 2,000 ppm group, there was decrease in the response during 16 exposures and no response during 50 exposures. In the 7,000 ppm group, a decreased response was observed in one exposure and no response was observed in the other 65 exposures. These effects were transient, and as no clinical observations of compromised neurological function were detected they were considered to be due to a reversible sedation caused by cyclohexane. The NOAEL for neurological effects was 500 ppm while the NOAEL for hepatic effects was 2,000 ppm. However, the partially reversible hepatic effects observed in males at 7,000 ppm were slight and may be considered of an adaptive nature.
- Remark** : Additional groups of rats (12/sex/group) were treated in parallel with those of the main study in order to assess neurotoxicity of cyclohexane in FOB, motor activity and neuropathology tests (Haskell Laboratory 1996c). Neurobehavioral evaluations were conducted prior to exposure and at week 4, 8, and 13. During each evaluation period FOB was performed prior to the motor activity test. At the end of the study, 6 rats/sex/group were selected for neuropathology, the controls and 7,000 ppm tissues selected were examined, and the intermediate dose tissues were saved. Neurological lesions were also assessed by examining sections of the brain, spinal cord, sciatic nerve, gasserian ganglia, cervical and dorsal root fibers and ganglia, cervical and lumbar ventral root fibers and gastrocnemius muscle.
- Similar to the main 90 day study, a sedative effect was observed at doses

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of 2,000 ppm and higher characterized by a decrease in the mean response to an alerting stimulus. This effect was transient since no effects were observed immediately after removal from the exposure chamber. No effects were observed during the FOB and motor activity assessment. Histologically, no treatment-related findings were observed, the only lesions observed being identical in character and severity to those observed in controls. These have already been described as occurring spontaneously in the rat. The NOAEL for neurotoxicity was 500 ppm based on the transient sedative effect observed at 2,000 ppm and higher.

**Test substance** : CAS No. 110-82-7; cyclohexane; (purity 99.98%)  
**Reliability** : (1) valid without restriction  
**Flag** : Critical study for SIDS endpoint  
17.07.2008

(18)(20)

**Type** :  
**Species** : mouse  
**Sex** : male/female  
**Number of animals** : 20/sex/group (controls & hi dose); 10/sex/group (intermediary doses)  
**Strain** : other: Crl:CD1 (ICR) BR  
**Route of admin.** : inhalation  
**Exposure period** : 13 weeks  
**Frequency of treatm.** : 6 hrs / day, 5 days / week (total 66 exposures)  
**Doses** : 500, 2000, 7000 ppm  
**Control group** : yes  
**NOAEL** :  
**Method** : EPA test method  
**Year** : 1996  
**GLP** : yes  
**Methods** : A 13 week inhalation toxicity study in mice was also performed following a 2-week range finding study in mice. The study was comparable in experimental conditions to that performed on rats.

After a stimulus, the animals of the 500 ppm group reacted as controls. In the 2,000 ppm group, a decrease in or an absence of response was observed from the third exposure onwards, the effects appearing to get worst with time (more and more no-response with increasing numbers of exposures). In the 7,000 ppm group, there was an increase in the incidence of decreased response, absence of response and hyperactive state from test day 4 to test day 30. From test day 30 to the end of the study, the response to the stimulus was impossible to determine due to the hyperactive state of the animals. These symptoms were observed just after exposure but were reversible until the next exposure. The most frequently described symptoms were: abnormal gait or mobility, excessive grooming, hyperactivity, hyper reactivity, spasms, aggressivity, hypo-activity and ruffled fur.

In males, haematological abnormalities were observed from 500 ppm, these symptoms (increase in RBC - increase in Hb - increase in Ht and decrease in platelets) were not always statistically significant and not always dose-related. In females, increases in RBC, Hb and Ht were only observed at 7,000 ppm. Variation in the haematological parameters occurred for all animals at the 7,000 ppm dose level, but could not be explained in relation to the lack of systemic symptoms of dehydration. They were considered to be of no toxicological importance.

An increase in absolute and relative liver weight was observed in males at 7,000 ppm (absolute: 1.504 g vs 1.275 g). Only the relative weight was increased in females. No concomitant histological findings were observed. These results were not in accordance with those found during the two-week range finding test (histological findings from 3,000 onwards in

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females and at 9,000 in males) but did not interfere with the determination of a NOEL.

**Result** : For neurologic effects, a NOAEL of 500 ppm was determined based on signs of sedation observed in animals at 2,000 ppm. The NOAEL for systemic effects was 2,000 ppm based on effects observed in the liver at 7,000 ppm (increase in absolute and relative liver weights).

**Test substance** : CAS No. 110-82-7; cyclohexane; (purity 99.98%)

**Reliability** : (1) valid without restrictions

**Flag** : Critical study for SIDS endpoint

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(19)(29)

### 5.5 GENETIC TOXICITY 'IN VITRO'

**Type** : Ames test

**System of testing** : Test species/strain: Salmonella typhimurium TA100, TA1535, TA98, TA1537

**Test concentration** : 0 to 10000 nl/ml

**Cytotoxic concentr.** : 7800 µg/ml (10000 nl/ml)

**Metabolic activation** : with and without

**Result** : negative

**Method** : not data

**Year** : 1986

**GLP** : no data

**Test Method** : *Salmonella typhimurium* (TA1535, TA1537, TA98 and TA100) were incubated with and without metabolic activation in DMSO. Two metabolic activation systems were used, one with SD rats livers and the other with Syrian hamster livers. Doses of cyclohexane ranged from 0 to 10,000 µg/plate. The following mutagens were used as concurrent positive controls: sodium azide for TA1535 and TA100, 4-nitro-o-phenylenediamine for TA98, and 9-aminoacridine for TA97 and TA1537; 2-aminoanthracene was used with all strains with hamster and rat liver metabolic activation systems. The dose levels of the positive controls were determined by the individual laboratory generation of a dose-response curve. Positive controls confirmed the validity of the assay.

**Result** : Signs of toxicity were noted in the 3,333 µg/plate for TA1537 and TA98 and in the 10,000 µg/plate for TA100 and TA1535.

There was no evidence of reverse mutation for any dose tested with and without metabolic activation.

Genotoxic effects:

With metabolic activation: negative

Without metabolic activation: negative

**Test substance** : CAS No. 110-82-7; cyclohexane

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

**Flag** : Critical study for SIDS endpoint

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(31)

**Type** : Mouse Lymphoma Assay

**System of testing** : L5178Y mouse lymphoma cells (TK locus)

**Test concentration** : 313 to 10000 nl/ml (250 to 7800 µg/ml)

**Cycotoxic concentr.** : 7800 µg/ml (10000 nl/ml)

**Metabolic activation** : with and without

**Result** : negative

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<b>Method</b>	: Method equivalent to current guidelines
<b>Year</b>	: 1986
<b>GLP</b>	: no data
<b>Remark</b>	: Cyclohexane, solubilised in desionised water, was tested at doses ranging from 313 nl/ml to 10,000 nl/ml (250 µg/ml to 7,800 µg/ml) (API, 1986). The method used was equivalent to the guidelines.
<b>Result</b>	: Cytotoxicity was seen at 10,000 nl/ml (7,800 µg/ml) with and without metabolic activation. Without metabolic activation the percentage of relative growths ranged from 39 to 59 % without dose-effect relationship. There was no evidence of forward mutation at any dose. With metabolic activation, the percentage of relative growths ranged from 46 to 64% and was not dose related. There were very slight increases in mutant frequency in four treatments, but not dose related, and it was decided to confirm this result with another test.  Doses ranging from 3,000 to 8,000 nl/ml (2,340 to 6,240 µg/ml) were tested in the second trial (cytotoxicity was found at 9,000 nl/ml (7,020 µg/ml)). The percentage of relative growths ranged from 23 to 69%. In this trial the results were clearly negative. Overall, this test can be considered as negative with and without metabolic activation.
<b>Test substance</b>	: CAS No. 110-82-7; cyclohexane
<b>Reliability</b>	: (1) valid without restriction
<b>Flag</b>	: Critical study for SIDS endpoint
18.07.2008	(2)(38)
<b>Type</b>	: Sister Chromatid Exchange Test
<b>System of testing</b>	: Chinese Hamster Ovary cells
<b>Test concentration</b>	: 0.25 to 25 µg/ml in DMSO
<b>Cycotoxic concentr.</b>	: no data
<b>Metabolic activation</b>	: with and without
<b>Result</b>	: negative
<b>Method</b>	: Method equivalent to current guidelines
<b>Year</b>	: 1982
<b>GLP</b>	: no data
<b>Remark</b>	: Cyclohexane, solubilised in DMSO, was tested at doses ranging from 0.25 to 25 µg/ml. The method used was equivalent to the guidelines.
<b>Result</b>	: The higher dose tested corresponding to a complete growth inhibition of the cell culture (25 µg/ml). No effect was seen within the range of the doses tested and the test is considered negative.
<b>Test substance</b>	: CAS No. 110-82-7; cyclohexane
<b>Reliability</b>	: (2) valid with restrictions Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.
<b>Flag</b>	: Critical study for SIDS endpoint
18.07.2008	(39)
<b>Type</b>	: Unscheduled DNA synthesis
<b>System of testing</b>	: Human lymphocytes
<b>Test concentration</b>	: 0.1 to 10 mM cyclohexane
<b>Cycotoxic concentr.</b>	: no data
<b>Metabolic activation</b>	: with and without
<b>Result</b>	: negative
<b>Method</b>	: Method equivalent to current guidelines
<b>Year</b>	: 1983
<b>GLP</b>	: no data
<b>Remark</b>	: Human lymphocytes ( + or – S9 mix) were cultured for 4 hours in the presence or absence of cyclohexane. The effects on the DNA synthesis were measured through cellular [ <sup>3</sup> H]TdR uptake.

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**Result** : Cyclohexane induced a marked inhibition of [<sup>3</sup>H]TdR uptake in the S9 mix-lacking cultures while the corresponding cellular viabilities were unaffected. No effect was seen with metabolic activation. The effects seen without metabolic activation were not dose-dependent; solvent controls and negative controls were highly variable. Decrease of the uptake for the highest dose was within the values of the controls. No conclusion was drawn from this study.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity = 99.8%  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

**Flag** : Critical study for SIDS endpoint  
18.07.2008

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**Type** : DNA cell binding assay  
**System of testing** : *E. coli*  
**Test concentration** : 10 and 100 µM cyclohexane  
**Cycotoxic concentr.** : no data  
**Metabolic activation** : with and without  
**Result** : negative  
**Method** : no data  
**Year** : 1981  
**GLP** : no data  
**Remark** : Cyclohexane was tested in a DNA cell binding assay at doses of 10 and 100 µM. Cyclohexane was tested alone, mixed with lysozyme, mixed with liver extract and mixed with lysozyme and liver extract. Positive control was MMS and negative control was the culture middle only. The results are expressed as a "binding percentage". If this percentage is > 1%, the substance was considered positive.

**Result** : Cyclohexane was found negative when tested alone in the groups treated with liver extract and –with lysozyme and liver extract. A positive finding (1.6% only) was found in the group treated with cyclohexane + lysozyme at the highest dose (100 µM). This result is considered doubtful because this is a very slight increase and also because this effects is not found in the group - cyclohexane + lysozyme + liver extract.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity = 99.8%  
**Reliability** : (2) valid with restrictions  
Original study not retrieved and reviewed. Study used for EU Risk Assessment of cyclohexane.

**Flag** : Critical study for SIDS endpoint  
18.07.2008

(26)

### 5.6 GENETIC TOXICITY 'IN VIVO'

**Type** : Cytogenetic Assay  
**Species** : rat  
**Strain** : Sprague Dawley  
**Sex** : male/female  
**Number of animals** : 10 per sex  
**Route of admin.** : inhalation  
**Exposure period** : 6 hours/day for 5 days  
**Doses** : 0, 97, 307, 1042 ppm  
**Method** : other: not specified  
**Year** :  
**GLP** : no data

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<b>Remark</b>	: Samples of bone marrow cells were taken for cytogenetic analysis 6 hours after completion of the final dose. A positive control, triethylenamine, showed a significant increase in structural aberration frequency. For cyclohexane a small but statistically significant increase in numerical aberrations was recorded in low and medium dose females, and pooled data at the low dose groups of both sexes. There was no information on general toxicity; no decrease on mitotic index was seen at all the doses tested. However, the authors of the report concluded that the lack of a dose-related response indicated that these increases were not of biological importance. Moreover, the numerical aberrations parameter had often shown great variation in this laboratory, having no statistical significance even for positive controls (numerical data is not available). It can be considered that cyclohexane does not produce chromosomal aberrations under the conditions of this test.
<b>Result</b>	: Negative result. Some increased aberrations at low and medium doses but no dose-related effect.
<b>Test substance</b>	: CAS No. 110-82-7; cyclohexane; purity is unknown
<b>Reliability</b>	: (1) valid without restriction
<b>Flag</b>	: Critical study for SIDS endpoint
14.07.2008	(1)

### 5.7 CARCINOGENICITY

#### 5.8.1 TOXICITY TO FERTILITY

<b>Type</b>	: Two generation reproductive study
<b>Species</b>	: rat
<b>Sex</b>	: male/female
<b>Number of animals</b>	: 30/sex/group
<b>Strain</b>	: other: CD BR
<b>Route of admin.</b>	: inhalation
<b>Exposure period</b>	: 90 days
<b>Frequency of treatm.</b>	: 5 day/week and 6 h/day
<b>Premating exposure period</b>	
<b>Male</b>	: 10 weeks
<b>Female</b>	: 10 weeks
<b>Duration of test</b>	:
<b>No. Of generation studies</b>	: 2
<b>Doses</b>	: 0, 500, 2,000, or 7,000 ppm
<b>Control group</b>	: yes
<b>NOAEL parental</b>	:
<b>NOAEL F1 offspring</b>	:
<b>Method</b>	: other: US and OECD Test guideline
<b>Year</b>	: 1997
<b>GLP</b>	: yes
<b>Remark</b>	: In this study, weanling F0 rats (30/sex/group) inhaled cyclohexane vapor at 0, 500, 2,000, or 7,000 ppm 5 day/week and 6 h/day. Exposure duration was 10 weeks before mating until sacrifice of the P1 generation and 11 weeks before mating until sacrifice of the F1 generation. Gravid females were not exposed from day 21 of gestation until day 4 of lactation. From day 5 of lactation until weaning the neonates were potentially exposed by maternal milk; no other exposure was administered. At post partum day 25, thirty F1 animals/sex/dose were chosen to produce the next generation, treatment was continued 11 weeks before mating and during gestation. Fertility parameters were calculated. From 500 ppm, there was an increased incidence of diminished response to a stimulus during exposure,



this finding being significant at 2,000 ppm and higher. At 7,000 ppm, major effects were observed on body weight, body weight gain and food efficiency. A decrease in mean body weight was seen with F1 male rats, P1 and F1 females during pre-mating, P1 females throughout gestation, and F1 females during lactation. A decrease in mean body weight gain was observed with F1 male rats and P1 and F1 females during pre-mating; however no reductions were seen for P1 females during gestation, suggesting that the reduction in mean gestation body weight was probably due to pre-existing body weight deficits established during the pre-mating period. The same findings were also seen with the F1 generation. A decrease in mean food efficiency of P1 and F1 females during lactation and a decrease in food consumption of P1 females during lactation were also observed. Effects on reproduction were limited to a decrease in mean pup weight for both the F1 and F2 generations at the high dose which was significant between post partum day 7 and 25 during which time pups were fed only maternal milk indicating the effect is due to cyclohexane via inflammation at 7,000 ppm in P1 and F1 adults, but this was considered incidental due to the lack of severity and the reported common occurrence in rats. There was a slight but significant decrease in the mean percentage of animals born alive in the F1 litters dosed with 7,000 ppm, but given that the value was still in the range of historical controls and that this effect was not dose-related, this was not considered biologically significant.

**Result** : The NOAEL for adult reproductive toxicity was 2,000 ppm based on decreases in pup body weight observed at 7,000 ppm. The systemic NOAEL for this study was 500 ppm based on sedative effects observed at 2,000 ppm and higher.

**Test substance** : CAS No. 110-82-7; cyclohexane  
**Reliability** : (1) valid without restriction  
**Flag** : Critical study for SIDS endpoint  
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### 5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

**Type** : Developmental toxicity  
**Species** : rat  
**Sex** : female  
**Number of animals** : 8/group  
**Strain** : other: CrI:CD BR  
**Route of admin.** : inhalation  
**Exposure period** : 9 days  
**Frequency of treatm.** :  
**Duration of test** : 14 days  
**Doses** : 0, 3,000, 6,000, or 9,000 ppm  
**Control group** : yes  
**Method** : other: US and OECD Test guideline  
**Year** : 1997  
**GLP** : yes  
**Remark** : As a pilot study, four groups of eight pregnant CDBR rats were exposed whole-body to concentrations of 0, 3,000, 6,000, or 9,000 ppm cyclohexane from gestational day 7 to 16. Dams were sacrificed on day 22 and examined for gross pathologies; implantations and resorptions were counted and their relative positions recorded; fetuses were weighed and examined externally for alterations.

**Result** : Maternal effects were limited to a reduction in overall maternal bodyweight gain, overall food consumption and diminished response of animals to a

sound stimulus during exposure to 6,000 ppm and higher. No effects were observed in the pups. The NOAEL was 3,000 ppm for the dams and 9,000 ppm for the pups.

**Remark**

: This pilot study served as a range-finding study used to design a more complete study which was carried out during the 90 day inhalation study previously described (Haskell Laboratory, 1997b). Four groups of CD BR rats were exposed whole body to cyclohexane at concentrations of 0, 500, 2,000 or 7,000 ppm from gestational day 7 to 16. Animals were sacrificed on day 22 and examined. Findings were limited to the dams and included a diminished response of the animals to a sound stimulus while in the chamber during exposure and at 2,000 ppm or higher and reductions in overall body weight gain and food consumption throughout the treatment period. A slight but significant decrease in implantation number with the number of corpora lutea unchanged was seen when compared with controls. This finding was consistent with slight pre-implantation losses and can be considered as not treatment-related since there was no treatment during the pre-implantation period. The NOAEL was 500 ppm for maternal toxicity and 7,000 ppm for developmental toxicity considering the lack of toxic effects noted.

**Test substance**

: CAS No. 110-82-7; cyclohexane

**Reliability**

: (1) valid without restriction

**Flag**

: Critical study for SIDS endpoint

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**5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES****5.9 SPECIFIC INVESTIGATIONS****5.10 EXPOSURE EXPERIENCE****5.11 ADDITIONAL REMARKS**

### 6.1 ANALYTICAL METHODS

### 6.2 DETECTION AND IDENTIFICATION

7.1 FUNCTION

7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED

7.3 ORGANISMS TO BE PROTECTED

7.4 USER

7.5 RESISTANCE

**8.1 METHODS HANDLING AND STORING**

**8.2 FIRE GUIDANCE**

**8.3 EMERGENCY MEASURES**

**8.4 POSSIB. OF RENDERING SUBST. HARMLESS**

**8.5 WASTE MANAGEMENT**

**8.6 SIDE-EFFECTS DETECTION**

**8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER**

**8.8 REACTIVITY TOWARDS CONTAINER MATERIAL**

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### 10.1 END POINT SUMMARY

### 10.2 HAZARD SUMMARY

### 10.3 RISK ASSESSMENT