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## IUCLID

## **Data Set**

**Existing Chemical** 

CAS No.

: ID: 108-08-7

: 108-08-7

EINECS Name

: 2,4-dimethylpentane

EC No.

TSCA Name

: 2,4-dimethylpentane

Molecular Formula

: C7H16

Producer related part

Company

: ExxonMobil Biomedical Sciences Inc.

Creation date

: 30.06.2008

Substance related part

Company

: ExxonMobil Biomedical Sciences Inc.

Creation date

: 30.06.2008

Status

Memo

: U.S. EPA - HPV Challenge Program

Printing date

: 30.06.2008

Revision date

Date of last update

: 30.06.2008

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Chapter (profile)

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Reliability (profile)

: Reliability: without reliability, 1, 2, 3, 4

Flags (profile)

: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

### ld 108-08-7 1. General Information Date 1.0.1 APPLICANT AND COMPANY INFORMATION 1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR 1.0.3 IDENTITY OF RECIPIENTS 1.0.4 DETAILS ON CATEGORY/TEMPLATE 1.1.0 SUBSTANCE IDENTIFICATION 1.1.1 GENERAL SUBSTANCE INFORMATION **Purity type** Substance type : petroleum product Physical status liquid Purity Colour Odour 03.07.2008 1.1.2 SPECTRA 1.2 SYNONYMS AND TRADENAMES 1.3 **IMPURITIES ADDITIVES** 1.5 TOTAL QUANTITY 1.6.1 LABELLING 1.6.2 CLASSIFICATION 1.6.3 PACKAGING

## **Date** 30.06.2008 1.7 USE PATTERN 1.7.1 DETAILED USE PATTERN 1.7.2 METHODS OF MANUFACTURE 1.8 REGULATORY MEASURES 1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES 1.8.2 ACCEPTABLE RESIDUES LEVELS 1.8.3 WATER POLLUTION 1.8.4 MAJOR ACCIDENT HAZARDS 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

1. General Information

**Id** 108-08-7

#### 2. Physico-Chemical Data

ld 108-08-7

Date

#### 2.1 MELTING POINT

Value : =-119.5 °C

Sublimation

Method : other: not specified

Year

GLP : no data

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.

**Reliability** : (2) valid with restrictions

The CRC Handbook of Chemistry and Physics is a peer reviewed

publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.

Flag : Critical study for SIDS endpoint

03.07.2008 (13)

#### 2.2 BOILING POINT

**Value** : = 80.4 °C at 1013 hPa

Decomposition :

Method : other: not specified

Year

GLP : no data

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.

**Reliability** : (2) valid with restrictions

The CRC Handbook of Chemistry and Physics is a peer reviewed

publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.

Flag : Critical study for SIDS endpoint

03.07.2008 (13)

#### 2.3 DENSITY

Type : density

Value : = .668 g/cm³ at 20 °C Method : other: not specified

Year

GLP : no data

**Test substance**: CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.

**Reliability** : (2) valid with restrictions

Data supplied by the experimental database associated with EPISuite.

This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-

reviewed publication.

Flag : Critical study for SIDS endpoint

03.07.2008 (16)

#### 2.3.1 GRANULOMETRY

#### 2.4 VAPOUR PRESSURE

Value : = 10.59 hPa at 25 °C

Decomposition

Method :

#### 2. Physico-Chemical Data

**Id** 108-08-7

Date

Year :

GLP : no data

**Method** : Method not specified.

**Test substance**: CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data were not

reviewed for quality, however, the reference is from a peer-reviewed

handbook.

Flag : Critical study for SIDS endpoint

03.07.2008 (4)

#### 2.5 PARTITION COEFFICIENT

Partition coefficient : octanol-water Log pow : = 3.63 at 25 °C

pH value

Method : other (calculated)

Year GLP

Method : Calculated values using KOWWIN version 1.67, a subroutine of the

computer program EPIWIN version 3.20

Test condition : Octanol / Water Partition Coefficient is calculated by the KOWWIN

subroutine, which is based on an atom/fragment contribution method of W.

Meylan and P. Howard in "Atom/fragment contribution method for

estimating octanol-water partition coefficients". 1995. J. Pharm. Sci. 84:83-

92.

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPI SuiteTM. This robust summary has a reliability rating of 2 because the

data are calculated and not measured.

Flag : Critical study for SIDS endpoint

03.07.2008 (17)(19)

#### 2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water

Value : = 5.5 mg/l at 25 °C

pH value

concentration : at °C

Temperature effects

Examine different pol.

**pKa** : at 25 °C

Description : Stable :

Deg. product

**Method** : other: no data

Year

GLP : no data

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions

Data supplied by the experimental database associated with EPISuite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-

reviewed publication.

Flag : Critical study for SIDS endpoint

03.07.2008 (20)

# 2. Physico-Chemical Data **Id** 108-08-7 Date 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY 2.10 EXPLOSIVE PROPERTIES 2.11 OXIDIZING PROPERTIES 2.12 DISSOCIATION CONSTANT 2.13 VISCOSITY 2.14 ADDITIONAL REMARKS

**Id** 108-08-7

Date

#### 3.1.1 PHOTODEGRADATION

Type : air Light source :

**Light spectrum** : nm

**Relative intensity**: based on intensity of sunlight

Conc. of substance : at 25 °C

**INDIRECT PHOTOLYSIS** 

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm<sup>3</sup>

Rate constant : =  $.00000000000685 \text{ cm}^3/(\text{molecule*sec})$ 

**Degradation** : = 50 % after 18.7 hour(s)

Deg. product

Method : other (calculated): Calculated values using AOPWIN version 1.92, a

subroutine of the computer program EPI SuiteTM version 3.20

Year :

Method : Calculated values using AOPWIN version 1.92, a subroutine of the

computer program EPI SuiteTM version 3.20

Indirect photodegradation, or atmospheric oxidation potential, is based on the structure-activity relationship methods developed by R. Atkinson under

the following conditions: Temperature: 25°C Sensitizer: OH- radical

Concentration of Sensitizer: 1.5E6 OH- radicals/cm3

**Remark**: 2,4-dimethylpentane has the potential to volatilize to air, based on a

relatively high vapor pressure, where it is subject to atmospheric oxidation. In air, 2,4-dimethylpentane can react with photosensitized oxygen in the form of hydroxyl radicals (OH-). The computer program AOPWIN (atmospheric oxidation program for Microsoft Windows) (EPI SuiteTM, 2000) calculates a chemical half-life for a 12-hour day (the 12-hour day half-life value normalizes degradation to a standard day light period during which hydroxyl radicals needed for degradation are generated), based on

an OH- reaction rate constant and a defined OH- concentration.

Based on a 12-hour day, a rate constant of 6.85 E-12 cm3/molecule\*sec, and an OH- concentration of 1.5 E6 OH-/cm3, 2,4-dimethylpentane has a

calculated half-life in air of 1. 6 days or 18.7 hours of daylight.

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown.

**Reliability** : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

03.07.2008 (19)

Deg. product : Method : Year : GLP :

Method : Technical discussion

**Remark**: Direct photochemical degradation occurs through the absorbance of solar

radiation by a chemical substance in aqueous solution. If the absorbed energy is high enough, then the resultant excited state of the chemical may undergo a transformation. A prerequisite for direct photodegradation is the

ability of one or more bonds within a chemical to absorb ultraviolet

(UV)/visible light in the 290 to 750 nm range. Light wavelengths longer than 750 nm do not contain sufficient energy to break chemical bonds, and wavelengths below 290 nm are shielded from the earth by the stratospheric

**Id** 108-08-7

Date

ozone layer (Harris, 1982).

An approach to assessing the potential for a substance to undergo photochemical degradation is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by constituent molecules (Zepp and Cline, 1977). Saturated and unsaturated hydrocarbons do not absorb light above 290 nm. Consequently, 2,4-dimethylpentane is not subject to photolytic processes in the aqueous

environment.

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability of 2 because it is a technical

discussion and not a study.

Flag : Critical study for SIDS endpoint

03.07.2008 (9)(21)

#### 3.1.2 STABILITY IN WATER

 Type
 : abiotic

 t1/2 pH4
 : at °C

 t1/2 pH7
 : at °C

 t1/2 pH9
 : at °C

Deg. product

Method : other: Technical discussion

Year

GLP : no data

**Result**: Hydrolysis of an organic chemical is the transformation process in which a

water molecule or hydroxide ion reacts to form a new carbon-oxygen bond. Chemicals with leaving groups that have a potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Gould, 1959). The lack of a suitable leaving group renders a compound resistant to hydrolysis. Heptane is resistant to hydrolysis because it lacks a functional group that is

hydrolytically reactive and Harris (1982) identifies hydrocarbons as

generally resistant to hydrolysis. Therefore, hydrolysis will not contribute to

the removal of 2,4-dimethylpentane from the environment.

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability of 2 because it is a technical

discussion and not a study.

Flag : Critical study for SIDS endpoint

03.07.2008 (7)(10)

#### 3.1.3 STABILITY IN SOIL

#### 3.2.1 MONITORING DATA

#### 3.2.2 FIELD STUDIES

#### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)
Water : % (Fugacity Model Level I)

ld 108-08-7

Date

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

Year

**Remark**: The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Physicochemical data used in the calculation:

Parameter Value w/ Units

Molecular Weight
Temperature
Log Kow
Water Solubility
Vapor Pressure
Melting Point

100.21
25° C
3.63
5.5 g/m3
10,586 Pa
-119.5° C

Result : Using the Mackay Level I calculation, the following

distribution is predicted for heptane:

%Distribution Compartment

99.98 Air <0.01 Water 0.01 Soil <0.01 Sediment

< 0.01 Suspended Sediment

<0.01 Biota

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated.

Flag : Critical study for SIDS endpoint

03.07.2008 (14)

Type : fugacity model level III

Media : other

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: Level III simulation using the Mackay Multimedia Environmental

Model (Mackay, 2001)

Year

Method : The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also

ld 108-08-7

**Date** 

used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 100.21 g/mol Water solubility = 5.5 g/m<sup>3</sup> Vapour pressure = 10,586 Pa

log Kow = 3.63

Melting point = -119.5 deg C

Degradation half-lives:

Air - 18.7 hrs Water - 360 hrs Soil - 720 hrs Sediment - 7200 hrs

Environmental Properties (EQC standard environment)

Dimensions (all defaults) Densities (all defaults)

Organic carbon & Advection (all defaults)

Transport Velocities (all defaults)

Emission and Inflows (defaults used)

Air 1000 kg/hr Water 1000 kg/hr Soil 1000 kg/hr Sediment 0 kg/hr

Result : Output:

Mass% Emissions(kg/hr)

 Air
 20.0
 1000

 Water
 70.6
 1000

 Soil
 2.8
 1000

 Sediment
 6.6
 0

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Conclusion**: The majority of 2,4-dimethylpentane is calculated to partition into the water

phase, with smaller but significant amounts into air, soil, and sediment based on the modeling parameters used in this calculation. 2,4-dimethylpentane is considered to be a Type 1 chemical with potential to

partition into all environmental compartments.

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated.

Flag : Critical study for SIDS endpoint

03.07.2008 (15)

#### 3.3.2 DISTRIBUTION

#### 3.4 MODE OF DEGRADATION IN ACTUAL USE

#### 3.5 BIODEGRADATION

Type : aerobic

Inoculum : activated sludge
Contact time : 28 day(s)

**Id** 108-08-7

**Date** 

**Degradation** : 74 (±) % after 28 day(s) **Result** : other: readily biodegradable

Deg. product

Method : OECD Guide-line 301 F "Ready Biodegradability: Manometric

Respirometry Test"

**Year** : 1996 **GLP** : yes

**Result**: Test material was readily biodegradable. Half-life was reached by day 10.

By day 28, 74% degradation of the test material was observed. 10%

biodegradation was achieved on day 4.

By day 10, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were

noted.

Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an

elemental analysis of the test material.

% Degradation\* Mean % Degradation
(day 28)
(day 28)
73.5.74.0.76.8

 Sample
 (day 28)
 (day 28)

 Test Material
 72.5, 74.0, 76.8
 74.4

 Na Benzoate
 88.7, 88.9
 88.8

#### **Test condition**

: Non acclimated activated sludge and test medium were combined prior to test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate, Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically monitored for oxygen consumption.

Test material was tested in triplicate, controls and blanks were tested in duplicate.

Test material concentration was approximately 45 mg/L. Sodium benzoate (positive control) concentration was 50mg/L.

Test temperature was 22 +/- 1 Deg C.

All test vessels were stirred constantly for 28 days using magnetic stir bars

and plates.

Test substance : CAS #142-82-5; heptane; 99% pure.
Conclusion : Heptane is readily biodegradable.
Reliability : (1) valid without restrictions

30.06.2008 (8)

Type : aerobic

**Inoculum** : other: soil, non-adapted

Contact time : 20 day(s)

**Degradation** : 70 (±) % after 20 day(s) **Result** : other: readily biodegradable

Deg. product

Method : other: Standard Methods for the Examination of Water and Waste Water

**Year** : 1971 **GLP** : no

**Result** : 70% degradation was measured after 20 days incubation with an

unacclimated inoculum.

% Biodegradation of test substance after days:

2 days = 28 % 5 days = 63 %

10 days = 70 % 20 days = 70 %

**Test condition**: American Public Health Association, Standard Methods for the

Examination of Water and Waste Water, using 1.0 mg/l of test substance. Biodegradation was determined by measuring biological oxygen demand

<sup>\*</sup> replicate data

**Id** 108-08-7

**Date** 

(BOD). Each 300 ml BOD bottle received 1.0 mg of heptane, 1.0 ml of a 1:10 suspension Hudson-Collamer silt loam soil in distilled water, and a mineral salts solution prepared as described in the test method. Bottles were incubated in the dark at 25C. The test substance was obtained from Aldrich Chemical Co.

**Test substance** : CAS #142-82-5; heptane; 99% pure. **Conclusion** : Heptane is readily biodegradable.

**Reliability** : (2) valid with restrictions

A standard test method was used. The study was conducted prior to GLP.

30.06.2008

#### 3.6 BOD5, COD OR BOD5/COD RATIO

#### 3.7 BIOACCUMULATION

**Species**: other: see remark

**Exposure period** : at 25 °C

Concentration

**BCF** : = 125

Elimination :

Method : other: calculation

Year

GLP : nc

**Remark**: A log bioconcentration factor (BCF) of 2.097 is calculated (BCF = 125).

With respect to a log Kow = 3.63, which was used to calculate the BCF, 2,4-dimethylpentane in the aquatic environment is expected to have a low

potential to bioaccumulate.

Test substance : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

03.07.2008 (19)

#### 3.8 ADDITIONAL REMARKS

Date

#### 4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type :

 Species
 : other: fish

 Exposure period
 : 96 hour(s)

 Unit
 : mg/l

 LC50
 : = 2.2

Method : other: ECOSAR version 0.99h, US EPA

Year GLP

Method : ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships

(SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

Result : Calculated 96-hr LC50 for fish = 2.2 mg/L

**Test condition** : Experimental water solubility = 5.5 mg/l @ 25°C (Yalkowsky and

Dannenfelser, 1992), log Kow = 3.63 @ 25°C (USEPA, 2000), and melting point = -119.5°C (Lide et al., 1997-1998) were entered into the program.

Class: Neutral organics

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

03.07.2008 (5)

#### 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type :

 Species
 : other: Daphnia

 Exposure period
 : 48 hour(s)

 Unit
 : mg/l

 LC50
 : = 2.6

Method : other: ECOSAR version 0.99h, US EPA

**Id** 108-08-7 4. Ecotoxicity

Date

Year **GLP** 

Method ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships

(SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow), SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

Result Calculated 48-hr LC50 for a daphnid = 2.6 mg/L

Experimental water solubility = 5.5 mg/l @ 25°C (Yalkowsky and **Test condition** 

Dannenfelser, 1992), log Kow = 3.63 @ 25°C (USEPA, 2000), and melting point = -119.5°C (Lide et al., 1997-1998) were entered into the program.

Class: Neutral organics

CAS #108-08-7; 2,4-dimethylpentane; purity is unknown **Test substance** 

Reliability (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

03.07.2008 (5)

**Type** static

Species other: Daphnia **Exposure period** 48 hour(s) mg/l Unit **EC50** = 1.5

Method other: based on discussions in GESAMP/MARPOL meetings held in 1973

Year

**GLP** 

Individual treatment concentrations were prepared by mixing the test Method

substance in freshwater for 24 hours in a conical flask. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were 250 ml conical flasks with 25 daphnids per flask. Two replicates of each

treatment level and control were evaluated.

Organisms supplied by testing lab; age = <24 hours old; parents age = approximately 21 days old.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

Date

119.).

Result : 48-hr EC50 for a daphnid = 0.423 mg/L

**Test condition**: Dissolved oxygen was >50% saturation during the study. The pH was 7.5

to 8.3. Temperature was 20 Deg C.

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L. Only the following analytical data were reported:

Nominal	Initial Measured48-hr Measured			
Conc. (mg/L)	Conc. (mg/L)	Conc. (mg/L)		
0.32	0.04	Not Determined		
1.0	0.04	Not Determined		
3.2	0.5	Not Determined		
5.6	2.1	1.7		
10	22	Not Determined		

Test substance Reliability CAS #142-82-5; heptane (2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline, OECD 202, and used acceptable

methods to prepare exposure solutions.

Flag : Critical study for SIDS endpoint

03.07.2008 (18)

Type : semistatic

**Species**: other: Gammarid (Chaetogammarus marinus)

**Exposure period** : 96 hour(s) **Unit** : mg/l **LC50** : = .2

Method : other: Static Gammarid Acute Toxicity Test

Year

GLP : no data

**Method**: Individual treatment solutions were prepared by mixing the test substance

in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24

hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8%; length = 5 mm.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

119.).

**Result** : 96-hr LC50 for a gammarid = 0.2 mg/L

**Test condition**: Dissolved oxygen was >50% saturation during the study. The pH was 7.5

to 8.3. Temperature was 15 Deg C. Natural seawater was used with a

salinity of 2.8%

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L.

Test solutions were analyzed only upon test initiation.

Only the following analytical data were reported:

Date

Nominal	Initial Measured		
Conc. (mg/L)	Conc. (mg/L)		
0.32	0.003		
1.0	0.07		
3.2	0.2		
5.6	Not Determined		
10	Not Determined		

Test substance Reliability

CAS #142-82-5; heptane(2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to

prepare exposure solutions.

Flag : Critical study for SIDS endpoint

03.07.2008 (18)

Type : semistatic

Species : other: mysid shrimp (Mysidopsis bahia)

 Exposure period
 : 96 hour(s)

 Unit
 : mg/l

 LC50
 : = .1

Method : other: Static Gammarid Acute Toxicity Test

Year

GLP : no data

**Method** : Individual treatment solutions were prepared by mixing the test substance

in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24

hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8%; test organisms were approximately 4 weeks old, with lengths of approximately 6 mm.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

119.).

**Result** : 96-hr LC50 for a gammarid = 0.1 mg/L

**Test condition** : Dissolved oxygen was >50% saturation during the study. The pH was 7.5

to 8.3. Temperature was 20 Deg C. Natural seawater was used with a

salinity of 2.8%

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L.

Test solutions were analyzed only upon test initiation.

Only the following analytical data were reported:

Nominal	Initial Measured		
Conc. (mg/L)	Conc. (mg/L)		
0.32	0.003		
1.0	0.07		
3.2	0.2		
5.6	Not Dete		

5.6 Not Determined10 Not Determined

**Test substance** : CAS #142-82-5; heptane

Date

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to

prepare exposure solutions.

Flag : Critical study for SIDS endpoint

03.07.2008 (18)

#### 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

**Species**: other algae: Green Alga

Endpoint

Exposure period : 96 hour(s)
Unit : mg/l
EC50 : = 2.0
ChV : = 0.5

Method : other: ECOSAR version 0.99h, US EPA

Year

GLP :

Method

ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

Result : Calculated 96-hr EC50 for a green alga = 2.0 mg/L Calculated 96-hr ChV for a green alga = 0.5 mg/L

**Test condition** : Experimental water solubility = 5.5 mg/l @ 25°C (Yalkowsky and

Dannenfelser, 1992), log Kow = 3.63 @  $25^{\circ}$ C (USEPA, 2000), and melting point = -119.5°C (Lide et al., 1997-1998) were entered into the program.

Class: Neutral organics

**Test substance** : CAS #108-08-7; 2,4-dimethylpentane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

03.07.2008 (5)

**Id** 108-08-7 4. Ecotoxicity

Date

#### TOXICITY TO MICROORGANISMS E.G. BACTERIA

#### 4.5.1 CHRONIC TOXICITY TO FISH

#### 4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

#### 4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

#### 4.6.2 TOXICITY TO TERRESTRIAL PLANTS

**Species** other terrestrial plant: Lactuca sativa Ravel R2

**Endpoint** other: growth **Exposure period** day(s) Unit mg/l Method other

Year **GLP** 

n-Heptane; CAS #142-82-5; >/= 95% pure Test substance

no

Soil Test Results Result

> Test 1 (lab 1): Days 7 and 14 EC50 = >1000 mg/kg soil (dry weight), based on nominal concentrations. No effects at the highest concentration tested. Test 2 (lab 2): Days 7 and 14 EC50 = >1000 mg/kg soil (dry weight), based on nominal concentrations. No effects at the highest concentration tested.

**Nutrient Solution Test Results** 

Test 1 (lab 1): Day 16 or 21 EC50 = 1.7 (1.4-2.0) mg/L, based on measured concentrations (value is below water solubility).

Test 2 (lab 2): Day 16 or 21 EC50 = 47 (38-65) mg/L, based on measured

concentrations (value exceeds water solubility).

**Test condition** Testing occurred in two labs. Two tests were conducted in soil (one at each

lab) and two tests in a nutrient solution (one at each lab). In both soil tests, lettuce seeds (Lactuca sativa) were germinated in soil obtained from an orchard. The characteristics of the soil collected for the first study were: pH

= 7.5; organic matter content = 1.4%; clay content = 12%. The

characteristics of the soil collected for the second study were: pH = 7.5; organic matter content = 1.8%; clay content = 24%. Test soil was sieved, 4 mm, prior to use. The nutrient solution composition was as follows: 3.73 mM Ca(NO3); 4.40 mM KNO3; 0.97 mM KH2PO4; 1.92 mM MgSO4; 0.89 mM K2SO4; trace elements and FeEDTA (per Steiner A, 1968. Soilless culture. Proceedings, Sixth Colloquium of the International Potash Institute,

Florence, Italy, pp. 342-341.).

#### Soil Experiments

Test substances were dissolved in acetone and/or mixed with a small amount of quartz sand, which was then mixed through the test soil to obtain a uniform distribution of test substance through the soil sample. A nominal concentration of 1000 mg/kg soil was tested.

Test substance concentrations in dry soil were spaced by a factor of 3.2. Controls and at least 3 test concentrations were run in duplicate. Soil was brought to 25 to 30% moisture content, which was equivalent to approximately 80% water holding capacity. Test systems were covered

Date

with glass plates until germination occurred to prevent moisture loss. After germination, the plates were removed and moisture loss was control by adding demineralized water back to the soil.

Five seedlings from each replicate test system were selected for the test and allowed to grow. Seedlings were sampled on days 7 and 14 by cutting them at soil level and measuring fresh weight. Soil water content and pH were measured at experimental initiation and termination.

Heptane was analyzed in soil at test initiation and termination by solvent extraction and GC/FID and/or GC/ECD. Measured concentrations in soil for test substances ranged from 70 to 150% of nominal concentrations. Specific results for heptane were not given.

#### **Nutrient Solution Experiment**

Test substances were dissolved in tributylalcohol and added to the nutrient solution. Seeds were placed on plastic trays, which contained perlite that was saturated with nutrient solution. The trays were covered with glass plates to reduce volatile loss during germination. After germination, seedlings were transferred to containers filled with nutrient solution and test substance. Solutions were renewed 3 times a week. Duplicate containers were used for each test concentration. Test concentrations were spaced by a factor of 3.2. Shoots were harvested on either day 16 or 21 and fresh weights determined. Oxygen content and pH were measured at solution renewal.

Heptane was analyzed in nutrient solution at test initiation and termination by solvent extraction and GC/FID and/or GC/ECD. Measured concentrations in solution for heptane were <50% of nominal concentrations. Specific results for heptane were not given. The volatility of heptane was suspected to have contributed to the loss in solution.

#### **Statistics**

EC50 values were determined based only on harvested shoots and calculated by applying a logistic model (per Haanstra et al., 1985. The use of sigmoidal dose response curves in soil ecotoxicological research. Plant Soil 82, 293-297.)

**Reliability** : (2) valid with restrictions

Although a specific standard guideline was not used, the test method generally followed standard test procedures, however, less specific

analytical information was provided than is desirable.

07.07.2008 (12)

Species : other terrestrial plant: Daucus carota L. cv. Early Horn

**Endpoint** : other: leaf damage, electrolyte loss

no

 Exposure period
 : 0 day(s)

 Unit
 : mg/l

 Method
 : other

 Year
 :

Year :

**Test substance**: n-Heptane; CAS #142-82-5; >99.5% pure

**Result**: The rate at which electrolytes were lost from carrot leaves exposed to n-

heptane was 34 min-1 x E4. The rate at which electrolytes were lost from untreated leaves was 12 min-1 x E4. The effects on cell membrane permeability as indicated by conductance agreed with the known susceptibility of different plant species towards hydrocarbons applied in the liquid state. The effect on the final conductance values and electrolyte

concentration was similar for the hydrocarbons tested.

Electrolyte loss from leaves exposed to a liquid hydrocarbon solvent

#### 4. Ecotoxicity

**Id** 108-08-7 Date 30.06.2008

#### **Test condition**

coating is explained as resulting from the destruction of plant cell membrane permeability and the release of internal cell electrolytes. Carrot plants, Daucus carota, were grown in a nutrient solution under controlled temperatures of 23 deg C during the day and 13 deg C during the night. The plants used were 3 to 5 weeks old. 5 g, fresh weight, of leaves were selected at random from the plants. 2 ml of test material was added to abaxial leaf surfaces by dropwise application so as to cover them as evenly as possible. This quantity represented an excess when compared to the lipid content of the plant cells. The leaves were immediately immersed in 100 ml of deionized water at 25 deg C in a glass flask. The leakage of salts from the leaves was inferred from changes in conductance across a dip-type bright platinum electrode inserted into the water over a period of 6 to 7 hours. The flasks were sealed with parafilm and the contents shaken before each reading. Data are based on 3

A final conductance value representing the total electrolytes released from the leaves was taken after the suspensions were boiled.

Reliability

(4) not assignable

replicates.

There is insufficient documentation on the samples and results to rate this study for reliability. The relevance of the method of exposure and adding the test substance to the abaxial surface of the leaf to mimic the effect of applying agrochemical sprays is not clear.

07.07.2008 (2)

**Species** other terrestrial plant: Helianthus annuus **Endpoint** other: leaf damage, electrolyte loss

**Exposure period** 0 day(s) Unit mg/l Method other Year **GLP** 

Test substance

n-Heptane; CAS #142-82-5; >99.5% pure Result

replicates.

The rate at which electrolytes were lost from sunflower leaves exposed to n-heptane was 160 min-1 x E4. The rate at which electrolytes were lost from untreated leaves was 16 min-1 x E4. A second study with n-heptane resulted in a rate of 170 min-1 x E4. The rate at which electrolytes were lost from untreated leaves in the second study was 17 min-1 x E4. The effects on cell membrane permeability as indicated by conductance agreed with the known susceptibility of different plant species towards hydrocarbons applied in the liquid state. The effect on the final conductance values and electrolyte concentration was similar for the hydrocarbons tested.

Electrolyte loss from leaves exposed to a liquid hydrocarbon solvent coating is explained as resulting from the destruction of plant cell membrane permeability and the release of internal cell electrolytes.

**Test condition** 

Sunflower plants. Helianthus annuus, were grown in a nutrient solution under controlled temperatures of 23 deg C during the day and 13 deg C during the night. The plants used were 3 to 5 weeks old. 5 g, fresh weight, of leaves were selected at random from the plants. 2 ml of test material was added to abaxial leaf surfaces by dropwise application so as to cover them as evenly as possible. This quantity represented an excess when compared to the lipid content of the plant cells. The leaves were immediately immersed in 100 ml of deionized water at 25 deg C in a glass flask. The leakage of salts from the leaves was inferred from changes in conductance across a dip-type bright platinum electrode inserted into the water over a period of 6 to 7 hours. The flasks were sealed with parafilm

A final conductance value representing the total electrolytes released from

and the contents shaken before each reading. Data are based on 3

4. Ecotoxicity

ld 108-08-7

**Date** 

Reliability

the leaves was taken after the suspensions were boiled.

: (4) not assignable

There is insufficient documentation on the samples and results to rate this study for reliability. The relevance of the method of exposure and adding the test substance to the abaxial surface of the leaf to mimic the effect of

applying agrochemical sprays is not clear.

07.07.2008 (2)

- 4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS
- 4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES
- 4.7 **BIOLOGICAL EFFECTS MONITORING**
- 4.8 **BIOTRANSFORMATION AND KINETICS**
- 4.9 **ADDITIONAL REMARKS**

Date

#### 5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

#### 5.1.1 ACUTE ORAL TOXICITY

#### 5.1.2 ACUTE INHALATION TOXICITY

Type : LC50

**Value** : > 29.29 mg/l

Species : rat

Strain : Sprague-Dawley
Sex : male/female

Number of animals : 10

Vehicle : other: none

**Doses** : 29.29 mg/L (17940 ppm)

**Exposure time** : 4 hour(s)

Method : other: Similar to OECD guideline 403

**Year** : 1982

GLP

**Test substance**: other TS: n-Heptane (CAS # 142-82-5)

**Remark**: Animals were exposed to n-heptane vapor for 4 hours at a concentration of

29.29 mg/L (nominal) or 17937.5 ppm (mean analytical).

**Result**: There was no mortality during the course of the study. A slight reduction of

mean male body weights was noted on day 2 post exposure but males recovered by day 4. All animals appeared normal throughout the study and at terminal necropsy with the exception of one female observed with

enlarged mandibular lymph nodes on the right side.

**Conclusion**: n-Heptane has a low order of toxicity by the inhalation route of exposure.

**Reliability** : (2) valid with restrictions

07.07.2008 (11)

#### 5.1.3 ACUTE DERMAL TOXICITY

#### 5.1.4 ACUTE TOXICITY, OTHER ROUTES

#### 5.2.1 SKIN IRRITATION

#### **5.2.2 EYE IRRITATION**

#### 5.3 SENSITIZATION

#### 5.4 REPEATED DOSE TOXICITY

Type : Species : rat

Sex : male/female
Strain : Sprague-Dawley
Route of admin. : inhalation

Date

**Exposure period** : 6 hours/day

Frequency of treatm. : 5 days/week for 26 weeks

Post exposure period : 2-week post exposure recovery period

**Doses** : 0, 500, 2000 and 4000 ppm

Control group : yes

**NOAEL** : = 2970 ppm

Method : other: similar to OECD guideline 413

**Year** : 1980

GLP

**Test substance** : other TS: n-Heptane (CAS # 142-82-5)

**Remark**: Animals were exposed to 0, 398 or 2970 ppm n-heptane.

Type: 26-week inhalation toxicity study Number of animals: 15/sex/dose group

**Result**: There were no treatment-related deaths during the study. The only

treatment-related observations were labored breathing or rapid breathing and slight prostration during the first week of study during exposure only, and anogenital fur and dry rales during weekly observations. The in chamber signs were generally more numerous and severe in the higher dose group and appeared to abate by the second week of the study.

No treatment-related effects were observed for body weight, hematology or urinalysis. Serum alkaline phosphatase was significantly elevated in female high dose rats and slightly elevated in low dose females. All other clinical chemistry values appeared normal with the exception of one male high level rat whose serum glutamic pyruvic transaminase and serum alkaline phosphatase levels were markedly elevated when compared to all other male rats on test. Proteinuria, elevated specific gravity and ketones

were observed but do not appear to be related to treatment.

Conclusion : The effects observed are consistent with acute CNS depression and

generally abated by the second week of study. Under the conditions of this study, the LOAEL for acute CNS depression is 2,970 ppm and the NOAEL

for systemic toxicity is 2,970 ppm.

**Reliability** : (2) valid with restrictions

08.07.2008 (1)

Type : rat Sex : male

Strain : Sprague-Dawley
Route of admin. : inhalation
Exposure period : 9 hours/day

Frequency of treatm. : 5 days/week for 7, 14 or 30 weeks

**Post exposure period**: None. Animals were sacrificed at 7, 14 or 30 weeks.

**Doses** : 0, 1500 ppm

Control group : other: yes, omitted the second air flow

NOAEL : > 1500 - ppm Method : other: none specified

**Year** : 1981

GLP

**Test substance**: other TS: n-Heptane (CAS # 142-82-5)

**Remark** : Only males and one dose group were used. The primary objective of this

study was to assess the appearance of polyneuropathy and urinary metabolites in rats following exposure to analytical grade solvents frequently used in Italian shoe factories. Nerve tissue was examined

microscopically.

Body weights were analyzed by a two-way analysis of variance and

Student's t test for the comparison of two slopes. Type: 30-week inhalation neurotoxicity study Number of animals: 6-9 males/dose group

**Result**: None of the animals developed signs of neuropathy. There were no

differences in weight gain of rats (30 weeks) compared to controls.

Date

Differences between mean values for hindlimb spreads observed in treated animals and controls were not statistically significant. However, authors note that in their hands, the test employed turned out to be scarcely effective due to high individual variability. No histological signs of giant axonal degeneration were noted in rats treated at 1500 ppm (30 weeks).

**Conclusion** : Under the conditions of this test, inhalation of n-heptane at 1500 ppm did

not induce neuropathy in rats.

**Reliability** : (2) valid with restrictions

08.07.2008 (6)

#### 5.5 GENETIC TOXICITY 'IN VITRO'

**Type**: Bacterial reverse mutation assay

System of testing : Salmonella typhimurium and Escherichia coli Test concentration : Doses ranging from 3.91 to 250 ug/ml

Cycotoxic concentr.: 500 μg/mlMetabolic activation: with and withoutResult: negative

**Method** : other: No specific method or guideline was noted.

**Year** : 1982

GLP :

Test substance : other TS: heptane (CAS # 142-82-5)
Remark : GLP: Quality assurance statement

Strains tested: Salmonella typhimurium tester strains TA98, TA100, TA1535, TA1537, TA1538; Escherichia coli strains WP2, WP uvr A

Test substance concentrations: 3.91, 7.81, 15.6, 31.3, 62.5, 125, 250

mg/ml

Metabolic activation: With and without (S9 fraction mix of livers from Aroclor 1254 pretreated rats)

Vehicle: Tween 80/ethanol

Positive Controls: benzo[a]pyrene, 4-nitroquinoline-N-oxide, sodium azide, neutral red, potassium dichromate.

Cytotoxicity study: A toxicity screening test conducted prior to the full assay indicated cytotoxicity at 500 mg/ml with and without metabolic activation.

The cultures were incubated at 37°C for 48-72 hours in a sealed container before the revertant colonies were counted. Pre-incubation method was

used to limit evaporation of test material.

**Result**: The addition of heptane at amounts up to 250 mg per ml to cultures of

Escherichia coli WP2 and WP2 uvr A, Salmonella typhimurium TA 1535, TA 1537, TA 1538, TA 98, and TA 100 did not lead to an increase in the reverse gene mutation frequency in any of these strains, either in the presence or in the absence of rat liver S9 fraction. In one assay with Escherichia coli WP2 in the presence of S9 fraction a greater than 2.5 fold increase over control values was seen at 15.6 and 31.3 mg per ml. This increase was not dose-related nor repeated in replicate assays and was

therefore not considered to be a compound-related effect.

**Conclusion**: Under the conditions of this study, the test material was not mutagenic.

**Reliability** : (1) valid without restriction

08.07.2008 (3)

**Type** : other: Mitotic gene conversion assay

System of testing : Yeast

**Test concentration**: Doses ranging from 0.01 to 5.0 mg/ml

Date

Cycotoxic concentr.

**Metabolic activation**: with and without

**Result** : negative

Method : other: No specific method or guideline was noted

**Year** : 1982

GLP

Test substance : other TS: heptane (CAS # 142-82-5)
Remark : GLP: Quality assurance statement

Strains tested: Saccharomyces cerevisiae JD1

Test substance concentrations: 0.01, 0.1, 0.5, 1.0, 5.0 mg/ml

Metabolic activation: With and without (S9 fraction mix of livers from

Aroclor 1254 pretreated rats)

Vehicle: Tween 80/ethanol

Positive Controls: 4-nitroquinoline-N-oxide, cyclophosphamide

After 18-hour incubation at 30°C the cultures were placed onto the appropriate culture media for the selection of prototrophic colonies. After three days incubation at 30°C the numbers of prototrophic colonies were

counted.

**Result** : Exposure of Saccharomyces cerevisiae JD1 to heptane at concentrations

up to 5.0 mg/ml did not result in a consistent increase in the rate of mitotic gene conversion, either in the presence or in the absence of rat liver S9

fraction.

**Conclusion**: Under the conditions of this study, the test material was not genotoxic.

**Reliability** : (1) valid without restriction

08.07.2008 (3)

**Type** : other: Chromosome aberration assay

System of testing : Rat Liver (RL4) cells Test concentration : 2.5, 5, 10 ug/ml

Cycotoxic concentr. : 20 ug/ml (100% cytotoxicity), 10 ug/ml (0% cytotoxicity)

Metabolic activation

**Result** : negative

Method : other: No specific method or guideline was noted

**Year** : 1982

GLP

Test substance : other TS: heptane (CAS # 142-82-5)
Remark : GLP: Quality assurance statement

Vehicle: Tween 80/ethanol

Positive Controls: 7,12-Dimethylbenzanthracene (DMBA)

Cultured rat liver cells were grown and treated on glass microscopic slides contained in 100-ml volume glass Leighton tubes. After 22-hour exposure to test compound or vehicle, colcemid was added to each culture. After further 2 hours, the slides were removed, subjected to hypotonic treatment followed by fixation (methanol:acetic acid, 3:1) and stained with Giemsa. The preparations were randomly coded and 100 cells from each culture

were analyzed microscopically.

Result : In one culture exposed to 10 mg/ml of heptane a total of 7 chromatid gaps

were seen; this increased the frequency to 0.024 gaps per cell which, although greater than the vehicle control frequency, was not accompanied by an increase in any other type of aberration and is not considered to be a compound-related effect. Thus there was no significant or dose-related increase of chromosome damage in any of the culture exposed to heptane. Cultures exposed to the positive control material, DMBA, showed a marked

increase in the frequency of chromosome damage.

**Conclusion**: Under the conditions of this study, the test material was not clastogenc.

5. To	oxicity	ld 108-08-7 Date	
<b>Rel</b> 08.0	iability : (2) valid with restrictions 07.2008		(3)
5.6	GENETIC TOXICITY 'IN VIVO'		
5.7	CARCINOGENICITY		
5.8.1	TOXICITY TO FERTILITY		
5.8.2	DEVELOPMENTAL TOXICITY/TERATOGENICITY		
5.8.3	TOXICITY TO REPRODUCTION, OTHER STUDIES		
5.9	SPECIFIC INVESTIGATIONS		
5.10	EXPOSURE EXPERIENCE		
5.11	ADDITIONAL REMARKS		

6. Analyt. Meth. for Detection and Identification	ld 108-08-7 Date
6.1 ANALYTICAL METHODS	
6.2 DETECTION AND IDENTIFICATION	
27 / 32	

7. Eff	. Against Target Org. and Intend	ed Uses	ld Date	108-08-7	
7.1	FUNCTION				
7.2	EFFECTS ON ORGANISMS TO BE CONTROL	LLED			
7.3	ORGANISMS TO BE PROTECTED				
7.4	USER				
7.5	RESISTANCE				
	28 /	32			

8. M	eas. Nec. to Prot. Man, Animals, Environment	ld Date	108-08-7
8.1	METHODS HANDLING AND STORING		
8.2	FIRE GUIDANCE		
8.3	EMERGENCY MEASURES		
8.4	POSSIB. OF RENDERING SUBST. HARMLESS		
8.5	WASTE MANAGEMENT		
8.6	SIDE-EFFECTS DETECTION		
8.7	SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER		
8.8	REACTIVITY TOWARDS CONTAINER MATERIAL		
	29 / 32		

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# **Id** 108-08-7 9. References **Date** 30.06.2008 (20)Yalkowsky S and Dannenfelser R (1992). Aquasol Database of Aqueous Solubility. Version 5. College of Pharmacy, University of Arizona, AZ, USA. Zepp R and Cline D (1977). Rates of direct photolysis in the aqueous environment. Environ (21) Sci Technol 11, 359-366. 31 / 32

10. S	Summary and Evaluation		ld	108-08-7	
	•		Date		
10.1	END POINT SUMMARY				
10.2	HAZARD SUMMARY				
10.3	RISK ASSESSMENT				
10.0	Non Accessing to				
		32 / 32			

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2008 SEP 26 AN 7:41

## IUCLID

## **Data Set**

**Existing Chemical** 

CAS No. EINECS Name

EC No. TSCA Name Molecular Formula : ID: 110-54-3

: 110-54-3

: hexane : 203-777-6 : Hexane

: C6H14

Producer related part

Company Creation date : ExxonMobil Biomedical Sciences Inc.

: 30.06.2008

Substance related part

Company

Creation date

: ExxonMobil Biomedical Sciences Inc.

: 30.06.2008

Status

Memo

: U.S. EPA - HPV Challenge Program

Printing date

: 12.08.2008

Revision date

Date of last update

: 12.08.2008

Number of pages

: 36

Chapter (profile)

Reliability (profile)

: Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10

: Reliability: without reliability, 1, 2, 3, 4 Flags (profile)

: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

### **Id** 110-54-3 1. General Information Date 1.0.1 APPLICANT AND COMPANY INFORMATION 1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR 1.0.3 IDENTITY OF RECIPIENTS 1.0.4 DETAILS ON CATEGORY/TEMPLATE 1.1.0 SUBSTANCE IDENTIFICATION 1.1.1 GENERAL SUBSTANCE INFORMATION **Purity type** Substance type petroleum product Physical status liquid Purity Colour Odour 30.06.2008 1.1.2 SPECTRA 1.2 SYNONYMS AND TRADENAMES 1.3 **IMPURITIES ADDITIVES** 1.5 TOTAL QUANTITY 1.6.1 LABELLING 1.6.2 CLASSIFICATION 1.6.3 PACKAGING

## **Date** 30.06.2008 1.7 USE PATTERN 1.7.1 DETAILED USE PATTERN 1.7.2 METHODS OF MANUFACTURE 1.8 REGULATORY MEASURES 1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES 1.8.2 ACCEPTABLE RESIDUES LEVELS 1.8.3 WATER POLLUTION 1.8.4 MAJOR ACCIDENT HAZARDS 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

1. General Information

**Id** 110-54-3

#### 2. Physico-Chemical Data

ld 110-54-3

Date

#### 2.1 MELTING POINT

Value : = -95.3 °C

Sublimation

Method : other: not specified

Year

GLP : no data

Test substance : other TS: hexane; (CAS #110-54-3)
Test substance : CAS #110-54-3; hexane; purity is unknown.

**Reliability** : (2) valid with restrictions

The CRC Handbook of Chemistry and Physics is a peer reviewed

publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.

Flag : Critical study for SIDS endpoint

30.06.2008 (20)

#### 2.2 BOILING POINT

**Value** : = 68.7 °C at 1013 hPa

Decomposition

**Method** : other: not specified

Year

GLP : no data

**Test substance**: other TS: hexane; (CAS #110-54-3)

**Test substance** : CAS #110-54-3; hexane; purity is unknown.

**Reliability** : (2) valid with restrictions

The CRC Handbook of Chemistry and Physics is a peer reviewed

publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.

Flag : Critical study for SIDS endpoint

30.06.2008 (20)

#### 2.3 DENSITY

Type : density

Value : = .659 g/cm³ at 20 °C Method : other: not specified

Year :

GLP : no data

**Test substance** : other TS: hexane; (CAS #110-54-3)

**Test substance** : CAS #110-54-3; hexane; purity is unknown.

**Reliability** : (2) valid with restrictions

Data supplied by the experimental database associated with EPISuite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-

reviewed publication.

Flag : Critical study for SIDS endpoint

30.06.2008 (31)

#### 2.3.1 GRANULOMETRY

#### 2. Physico-Chemical Data

**Id** 110-54-3 Date 30.06.2008

#### 2.4 **VAPOUR PRESSURE**

Value = 20.13 hPa at 25 °C

**Decomposition** Method Year

**GLP** 

: no data

Test substance other TS: hexane; (CAS #110-54-3)

Method Method not specified.

Test substance CAS #110-54-3; hexane; purity is unknown.

Reliability (2) valid with restrictions

> This robust summary has a reliability rating of 2 because the data were not reviewed for quality, however, the reference is from a peer-reviewed

handbook.

: Critical study for SIDS endpoint Flag

30.06.2008 (6)

#### 2.5 **PARTITION COEFFICIENT**

Partition coefficient : octanol-water Log pow = 3.9 at 20 °C

pH value Method Year

**GLP** : no data

Test substance other TS: hexane; (CAS #110-54-3) Test substance CAS #110-54-3; hexane; purity is unknown.

Reliability (2) valid with restrictions

The value cited by the author is a recommended value based on a review of data retrieved from the literature. This robust summary has a reliability rating of 2 because there is insufficient information available on the

method.

: Critical study for SIDS endpoint Flag

30.06.2008 (11)

#### 2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in Water

Value = 9.5 mg/l at 20 °C

pH value

concentration at °C

**Temperature effects** 

Examine different pol.

at 20 °C pKa

Description

Stable

Deg. product

Method other: no data

Year

**GLP** no data

Test substance other TS: hexane; (CAS #110-54-3) Test substance : CAS #110-54-3; hexane; purity is unknown.

Reliability (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are from

a peer-reviewed journal.

: Critical study for SIDS endpoint Flag

30.06.2008 (38)

# 2. Physico-Chemical Data **Id** 110-54-3 Date 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY 2.10 EXPLOSIVE PROPERTIES 2.11 OXIDIZING PROPERTIES 2.12 DISSOCIATION CONSTANT 2.13 VISCOSITY 2.14 ADDITIONAL REMARKS

Date

#### 3.1.1 PHOTODEGRADATION

Type : air Light source :

**Light spectrum** : nm

Relative intensity : based on intensity of sunlight

Conc. of substance : at 25 °C

INDIRECT PHOTOLYSIS

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm³

**Rate constant** : = .0000000000546 cm³/(molecule\*sec)

**Degradation** : = 50 % after 23.4 hour(s)

Deg. product

Method : other (calculated): Calculated values using AOPWIN version 1.92, a

subroutine of the computer program EPI SuiteTM version 3.20

Year GLP

**Test substance**: other TS: hexane; (CAS #110-54-3)

Method : Calculated values using AOPWIN version 1.92, a subroutine of the

computer program EPI SuiteTM version 3.20

Indirect photodegradation, or atmospheric oxidation potential, is based on the structure-activity relationship methods developed by R. Atkinson under

the following conditions: Temperature: 25°C Sensitizer: OH- radical

Concentration of Sensitizer: 1.5E6 OH- radicals/cm3

**Remark**: Hexane has the potential to volatilize to air, based on a relatively high

vapor pressure, where it is subject to atmospheric oxidation. In air, hexane can react with photosensitized oxygen in the form of hydroxyl radicals (OH-). The computer program AOPWIN (atmospheric oxidation program for Microsoft Windows) (EPI SuiteTM, 2000) calculates a chemical half-life for a 12-hour day (the 12-hour day half-life value normalizes degradation to a standard day light period during which hydroxyl radicals needed for

degradation are generated), based on an OH- reaction rate constant and a

defined OH- concentration.

Based on a 12-hour day, a rate constant of 5.46 E-12 cm3/molecule\*sec, and an OH- concentration of 1.5 E6 OH-/cm3, hexane has a calculated

half-life in air of 1.96 days or 23.4 hours of daylight.

**Test substance** : CAS #110-54-3; hexane; purity is unknown.

**Reliability** : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

30.06.2008 (37)

Deg. product : Method : Year : GLP :

Method : Technical discussion

**Remark** : Direct photochemical degradation occurs through the absorbance of solar

radiation by a chemical substance in aqueous solution. If the absorbed energy is high enough, then the resultant excited state of the chemical may undergo a transformation. A prerequisite for direct photodegradation is the ability of one or more bonds within a chemical to absorb ultraviolet

(UV)/visible light in the 290 to 750 nm range. Light wavelengths longer than

750 nm do not contain sufficient energy to break chemical bonds, and

**Id** 110-54-3 **Date** 30.06.2008

wavelengths below 290 nm are shielded from the earth by the stratospheric

ozone layer (Harris, 1982).

An approach to assessing the potential for a substance to undergo photochemical degradation is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by constituent molecules (Zepp and Cline, 1977). Saturated and unsaturated hydrocarbons do not absorb light above 290 nm. Consequently, hexane is not subject to photolytic processes in the aqueous environment.

: CAS #110-54-3: hexane Test substance (2) valid with restrictions Reliability

This robust summary has a reliability of 2 because it is a technical

discussion and not a study.

: Critical study for SIDS endpoint Flag

30.06.2008 (13)(39)

#### 3.1.2 STABILITY IN WATER

**Type** abiotic at °C t1/2 pH4 at °C t1/2 pH7 at °C t1/2 pH9

Deg. product

other: Technical discussion Method

Year

**GLP** : no data

other TS: hexane; (CAS #110-54-3) Test substance

Result Hydrolysis of an organic chemical is the transformation process in which a

water molecule or hydroxide ion reacts to form a new carbon-oxygen bond. Chemicals with leaving groups that have a potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Gould, 1959). The lack of a suitable leaving group renders a compound resistant to hydrolysis. Hexane is resistant to hydrolysis because it lacks a functional group that is hydrolytically reactive and Harris (1982) identifies hydrocarbons as

generally resistant to hydrolysis. Therefore, hydrolysis will not contribute to

the removal of hexane from the environment.

: CAS #110-54-3: hexane Test substance (2) valid with restrictions Reliability

This robust summary has a reliability of 2 because it is a technical

discussion and not a study.

Flag : Critical study for SIDS endpoint

30.06.2008 (11)(14)

#### 3.1.3 STABILITY IN SOIL

#### 3.2.1 MONITORING DATA

#### 3.2.2 FIELD STUDIES

#### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type

Media other: air - biota - sediment(s) - soil - water

Air % (Fugacity Model Level I)

ld 110-54-3

Date

Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)
Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

Year

Remark : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Physicochemical data used in the calculation:

Parameter Value w/ Units

Molecular Weight 86.18
Temperature 25° C
Log Kow 3.90
Water Solubility 9.5 g/m3
Vapor Pressure 20,131 Pa
Melting Point -95.3° C

Result : Using the Mackay Level I calculation, the following

distribution is predicted for heptane:

%Distribution Compartment

99.97 Air 0.02 Water 0.01 Soil <0.01 Sediment

< 0.01 Suspended Sediment

<0.01 Biota</li>: CAS #110-54-3; hexane: (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated.

Flag : Critical study for SIDS endpoint

30.06.2008 (22)

Type : fugacity model level III

Media : other

Air : % (Fugacity Model Level I)
Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)
Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)

Method : other: Level III simulation using the Mackay Multimedia Environmental

Model (Mackay, 2001)

Year :

Test substance

Reliability

Method : Level III simulation using the Mackay Multimedia Environmental Model

(Mackay, 2001). Mass balances are calculated for the four bulk media of air (gas + aerosol), water (solution + suspended sediment + biota), soil, (solids + air + water), and sediment (solids + pore water). Equilibrium exists within, but not between media. Physical-chemical properties are used to quantify a

chemical's behavior in an evaluative environment. Three types of

chemicals are treated in this model: chemicals that partition into all media

Date

(Type 1), non volatile chemicals (Type 2), and chemicals with zero, or near-zero, solubility (Type 3). The model cannot treat ionizing or speciating substances. The Level III model assumes a simple, evaluative environment with user-defined volumes and densities for the following homogeneous environmental media (or compartments): air, water, soil, sediment, suspended sediment, fish and aerosols.

This model provides a description of a chemical's fate including the important degradation and advection losses and the intermedia transport processes. The distribution of the chemical between media depends on how the chemical enters the system, e.g. to air, to water, or to both. This mode of entry also affects persistence or residence time.

The rates of intermedia transport are controlled by a series of 12 transport velocities. Reaction half-lives are requested for all 7 media. The advective residence time selected for air also applies to aerosols and the residence time for water applies to suspended sediment and fish. The advective residence time of aerosols, suspended sediment and fish cannot be specified independently of the air and water residence times.

Result : Output:

Mass% Emissions(kg/hr)

 Air
 21.3
 1000

 Water
 63.3
 1000

 Soil
 4.3
 1000

 Sediment
 11.1
 0

**Test condition**: Physicochemical data used in the calculation:

Parameter Value w/ Units

Molecular Weight 86.18
Temperature 25° C
Log Kow 3.90
Water Solubility 9.5 g/m3
Vapor Pressure 20,131 Pa
Melting Point -95.3° C

Reaction Half Lives in hours as predicted using EPI SuiteTM:

Air (gaseous) 23.5 Water (no susp. part.) 360

Bulk Soil 720 Bulk Sediment 7200

Environmental Properties (EQC standard environment)

Dimensions (all defaults) Densities (all defaults)

Organic carbon & Advection (all defaults)

Transport Velocities (all defaults)

Emission and Inflows (defaults used)

Air 1000 kg/hr Water 1000 kg/hr Soil 1000 kg/hr Sediment 0 kg/hr

Test substance Conclusion

: CAS #110-54-3; hexane

The majority of hexane is calculated to partition into the water phase, with smaller but significant amounts into air, soil, and sediment based on the modeling parameters used in this calculation. Hexane is considered to be a

Type 1 chemical with potential to partition into all environmental

compartments.

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated.

ld 110-54-3 **Date** 30.06.2008

Flag : Critical study for SIDS endpoint

30.06.2008 (23)

#### 3.3.2 DISTRIBUTION

#### 3.4 MODE OF DEGRADATION IN ACTUAL USE

#### 3.5 BIODEGRADATION

Type : aerobic

Inoculum : activated sludge

Contact time : 28 day(s)

Degradation : 74 (±) % after 28 day(s)

Result : other: readily biodegradable

Deg. product

Method : OECD Guide-line 301 F "Ready Biodegradability: Manometric

Respirometry Test"

Year : 1996 GLP : yes

Test substance : other TS: heptane; (CAS #142-82-5)

**Result**: Test material was readily biodegradable. Half-life was reached by day 10.

By day 28, 74% degradation of the test material was observed. 10%

biodegradation was achieved on day 4.

By day 10, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were

noted.

Sample

Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an

elemental analysis of the test material.

% Degradation\* Mean % Degradation (day 28) (day 28)

 Test Material
 72.5, 74.0, 76.8
 74.4

 Na Benzoate
 88.7, 88.9
 88.8

**Test condition**: Non acclimated activated sludge and test medium were combined prior to

test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate,

Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically monitored for oxygen consumption.

To the state of th

Test material was tested in triplicate, controls and blanks were tested in

duplicate.

Test material concentration was approximately 45 mg/L. Sodium benzoate (positive control) concentration was 50mg/L.

To at the rest are there was a CO at 1 De a CO

Test temperature was 22 +/- 1 Deg C.

All test vessels were stirred constantly for 28 days using magnetic stir bars

and plates.

Test substance : CAS #142-82-5; heptane; 99% pure.
Conclusion : Heptane is readily biodegradable.
Reliability : (1) valid without restrictions

30.06.2008 (10)

Type : aerobic

<sup>\*</sup> replicate data

ld 110-54-3 **Date** 30.06.2008

**Inoculum** : other: soil, non-adapted

Contact time : 20 day(s)

**Degradation** : 70 (±) % after 20 day(s) **Result** : other: readily biodegradable

Deg. product

Method : other: Standard Methods for the Examination of Water and Waste Water

**Year** : 1971 **GLP** : no

**Test substance**: other TS: heptane; (CAS #142-82-5)

Result : 70% degradation was measured after 20 days incubation with an

unacclimated inoculum.

% Biodegradation of test substance after days:

2 days = 28 % 5 days = 63 %

10 days = 70 % 20 days = 70 %

**Test condition**: American Public Health Association, Standard Methods for the

Examination of Water and Waste Water, using 1.0 mg/l of test substance. Biodegradation was determined by measuring biological oxygen demand (BOD). Each 300 ml BOD bottle received 1.0 mg of heptane, 1.0 ml of a 1:10 suspension Hudson-Collamer silt loam soil in distilled water, and a mineral salts solution prepared as described in the test method. Bottles were incubated in the dark at 25C. The test substance was obtained from

Aldrich Chemical Co.

**Test substance** : CAS #142-82-5; heptane; 99% pure. **Conclusion** : Heptane is readily biodegradable.

**Reliability** : (2) valid with restrictions

A standard test method was used. The study was conducted prior to GLP.

30.06.2008 (12)

Type : aerobic

**Inoculum** : activated sludge

Contact time : 28 day(s)

**Degradation** : =  $100 (\pm) \%$  after 28 day(s) **Result** : Readily biodegradable

Deg. product

Method : other: Modified MITI test (Comparable to OECD 301C)

Year : 1992 GLP : no data

**Test substance** : CAS No. 110-54-3; hexane

**Test condition** : Concentration of the test substance was 100 mg/l, with a concentration of

inoculum of 30 mg/l. The source of the inoculum was non-acclimated

activated sludge. Results of the study were based on BOD.

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

**Reliability** : (2) valid with restrictions

The study was performed following acceptable guidlines, however, the data

were not retrieved an reviewed for quality.

Flag : Critical study for SIDS endpoint

30.06.2008 (5)

#### 3.6 BOD5, COD OR BOD5/COD RATIO

#### 3.7 BIOACCUMULATION

**Species** : other: see remark

**Exposure period** : at 25 °C

Concentration

**BCF** : = 200

ld 110-54-3 **Date** 30.06.2008

Elimination

Method : other: calculation

Year :

GLP : no

**Test substance** : other TS: hexane; (CAS #110-54-3)

**Remark** : A log bioconcentration factor (BCF) of 2.30 is calculated (BCF = 200). With

respect to a log Kow = 3.90, which was used to calculate the BCF, hexane

in the aquatic environment is expected to have a low potential to

bioaccumulate.

**Test substance** : CAS #110-54-3; hexane **Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

30.06.2008 (37)

#### 3.8 ADDITIONAL REMARKS

**Id** 110-54-3 4. Ecotoxicity **Date** 30.06.2008

#### 4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type

**Species** other: fish **Exposure period** 96 hour(s) Unit mg/l = 1.0 LC50

Method other: ECOSAR version 0.99h, US EPA

Year **GLP** 

other TS: hexane; (CAS #110-54-3) Test substance

ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships Method

(SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

Result Calculated 96-hr LC50 for fish = 1.0 mg/L

Experimental water solubility = 9.5 mg/l @ 25°C (McAuliffe, 1996), log Kow **Test condition** 

= 3.90 @ 20°C (Hansch, et.al, 1995), and melting point = -95.3°C (Lide et

al., 2003) were entered into the program.

Class: Neutral organics

Test substance CAS #110-54-3: hexane (2) valid with restrictions Reliability

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

30.06.2008 (8)

#### 4.2 **ACUTE TOXICITY TO AQUATIC INVERTEBRATES**

**Type** : static

Species : other: Daphnia **Exposure period** : 48 hour(s) Unit : mg/l **EC50** = 2.1

Method : other: based on discussions in GESAMP/MARPOL meetings held in 1973

Year

GLP : no data

Method : Individual treatment concentrations were prepared by mixing the test

substance in freshwater for 24 hours in a conical flask. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were 250 ml conical flasks with 25 daphnids per flask. Two replicates of each

treatment level and control were evaluated.

Organisms supplied by testing lab; age = <24 hours old; parents age =

approximately 21 days old.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

119.).

Result : 48-hr EC50 for a daphnid = 2.1 mg/L

**Test condition** : Dissolved oxygen was >50% saturation during the study. The pH was 7.5

to 8.3. Temperature was 20 Deg C.

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 1.0 to 10 mg/L.

Test substance : C Reliability : (2

CAS #110-54-3; hexane(2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline, OECD 202, and used acceptable

methods to prepare exposure solutions.

Flag : Critical study for SIDS endpoint

30.06.2008 (36)

Type : semistatic

**Species**: other: Gammarid (Chaetogammarus marinus)

**Exposure period** : 96 hour(s) **Unit** : mg/l **LC50** : = .4

Method : other: Static Gammarid Acute Toxicity Test

Year

GLP : no data

Method : Individual treatment solutions were prepared by mixing the test substance

in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24

hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity

of 2.8%; length = 5 mm.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

119.).

Result : 96-hr LC50 for a gammarid = 0.4 mg/L

**Test condition**: Dissolved oxygen was >50% saturation during the study. The pH was 7.5

to 8.3. Temperature was 15 Deg C. Natural seawater was used with a

salinity of 2.8%

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 1.0 to 10 mg/L. Test solutions were analyzed only upon test initiation.

Test substance Reliability

CAS #110-54-3; hexane(2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to

prepare exposure solutions.

Flag : Critical study for SIDS endpoint

30.06.2008 (36)

Type : semistatic

Species : other: mysid shrimp (Mysidopsis bahia)

**Exposure period** : 96 hour(s) **Unit** : mg/l **LC50** : = .4

Method : other: Static Gammarid Acute Toxicity Test

Year

GLP : no data

**Method** : Individual treatment solutions were prepared by mixing the test substance

in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24

hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8%; test organisms were approximately 4 weeks old, with lengths of approximately 6 mm.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

19.).

Result : 96-hr LC50 for a mysid = 0.4 mg/L

**Test condition**: Dissolved oxygen was >50% saturation during the study. The pH was 7.5

to 8.3. Temperature was 20 Deg C. Natural seawater was used with a

salinity of 2.8%

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L.

Test solutions were analyzed only upon test initiation.

Test substance Reliability

: CAS #110-54-3; hexane : (2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to

prepare exposure solutions.

Flag : Critical study for SIDS endpoint

4. Ecotoxicity Id 110-54-3

Date

30.06.2008 (36)

Type :

Species : other: Daphnia
Exposure period : 48 hour(s)
Unit : mg/l
LC50 : = 1.3

Method : other: ECOSAR version 0.99h, US EPA

Year

GLP

**Test substance**: other TS: hexane; (CAS #110-54-3)

Method : ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships

(SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

Result : Calculated 48-hr LC50 for a dahpnid = 1.3 mg/L

**Test condition** : Experimental water solubility = 9.5 mg/l @ 25°C (McAuliffe, 1996), log Kow

= 3.90 @ 20°C (Hansch, et.al, 1995), and melting point = -95.3°C (Lide et

al., 2003) were entered into the program.

Class: Neutral organics : CAS #110-54-3; hexane

**Test substance** : CAS #110-54-3; hexane Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

30.06.2008 (8)

#### 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species : other algae: Green Alga

Endpoint

Exposure period : 96 hour(s)
Unit : mg/l
EC50 : = .9
ChV : = .3

Method : other: ECOSAR version 0.99h, US EPA

Year :

#### 4. Ecotoxicity

ld 110-54-3 **Date** 30.06.2008

GLP Method

:

ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

Result : Calculated 96-hr EC50 for a green alga = 0.9 mg/L

Calculated 96-hr ChV for a green alga = 0.3 mg/L

**Test condition** : Experimental water solubility = 9.5 mg/l @ 25°C (McAuliffe, 1996), log Kow

= 3.90 @ 20°C (Hansch, et.al, 1995), and melting point = -95.3°C (Lide et

al., 2003) were entered into the program.

Class: Neutral organics CAS #110-54-3; hexane

**Test substance** : CAS #110-54-3; hexane **Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

30.06.2008 (8)

#### 4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

#### 4.5.1 CHRONIC TOXICITY TO FISH

#### 4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

#### 4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

#### 4.6.2 TOXICITY TO TERRESTRIAL PLANTS

ld 110-54-3 Date

#### 5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

#### 5.1.1 ACUTE ORAL TOXICITY

Type : LD50

**Value** : 28,720 mg/kg

Species : rat
Strain : no data
Sex : no data

Number of animals

Vehicle : no data

Year

GLP

Remark

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

**Conclusion**: Hexane has a low order of toxicity by the oral route of exposure

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

21.07.2008 (17)

Type : LD50

**Value** : =15,840 mg/kg

Species: ratStrain: no dataSex: no data

Number of animals

Vehicle : no data

Doses

Method : no data

Year GLP

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

**Conclusion**: Hexane has a low order of toxicity by the oral route of exposure

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

21.07.2008 (17)

Type : LD50

**Value** : 29,700 mg/kg

Species : rat
Strain : no data
Sex : no data

Number of animals

Vehicle : no data

Doses : Method :

Year : 1970 GLP : pre-GLP

Remark : The oral LD50 in a 14-day old rat was 29,700 mg/kg. Symptoms included

depressive effect on the central nervous system, salivation and soft faeces.

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

**Conclusion** : Hexane has a low order of toxicity by the oral route of exposure

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

21.07.2008 (19)

#### 5.1.2 ACUTE INHALATION TOXICITY

**Type** : LC50 **Value** : 48,000 ppm

Species : rat
Strain : no data
Sex : no data
Number of animals : no data
Vehicle : other: none

**Doses** : 1000 to 64000 ppm

**Exposure time** : 4 hour(s)

Method : other: Similar to OECD guideline 403

Year :

**Remark**: Animals were exposed to n-heptane vapor for 4 hours at concentrations of

1000 to 64000 ppm.

**Conclusion**: n-Hexane has a low order of toxicity by the inhalation route of exposure.

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

21.07.2008 (16)

#### 5.1.3 ACUTE DERMAL TOXICITY

#### 5.1.4 ACUTE TOXICITY, OTHER ROUTES

#### 5.2.1 SKIN IRRITATION

**Species** : humans **Sex** : no data

Number of animals : Duration of study : Doses :

Method: no dataYear: 1973GLP: pre-GLP

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

Result : Mild irritant

**Remark**: Dermal exposure can lead to peripheral neuropathy in humans.

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

22.07.2008 (29)(34)(35)

#### 5.2.2 EYE IRRITATION

Date

#### 5.3 SENSITIZATION

#### 5.4 REPEATED DOSE TOXICITY

Type :

Species : rat

Sex : male/female
Number of animals : 15/sex/group
Strain : F344
Route of admin : inhalation

Route of admin. : inhalation Exposure period : 6 hours/day

Frequency of treatm. : 5 days/week for 13 weeks

Doses : 0, 3000, 6500 and 10000 ppm

Control group : yes

Control group : yes
Method : no data
Year : 1984

GLP

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

**Result**: There were no n-hexane-related clinical signs of toxicity, effects on food

consumption, ophthalmological findings, or changes in neurological function. However, there was a lowering of the urinary pH in high-dose males. There were increased organ/body weight ratios for liver, kidney,

and testis in high-dose males and kidney in mid-dose males.

Histopathological examination of the tibial nerves revealed paranodal

axonal swelling in mid- and high-dose males.

**Remark** : n-Hexane is metabolized to 5-hydroxy-2-hexanone and 2,5-hexanedione *in* 

*vivo* (DiVincenzo *et al.*, 1976). These two metabolites are believed to be responsible for the neurotoxicities associated with n-hexane exposure. There is evidence to show that 2,5-hexanedione is more persistent in peripheral nerve tissue than the parent n-hexane (Bus *et al.*, 1981).

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

08.08.2008 (3)(4)

Type :
Species : rat
Sex : male
Strain : Wistar
Route of admin. : inhalation
Exposure period : 12 hours/day

Frequency of treatm. : 7 days/week for 16 weeks Doses : 0, 500, 1200, 3000 ppm

Control group : yes, NOAEL : 500 ppm

Method : other: none specified

Year : 1989 GLP : no data

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

**Remark** : Motor nerve conduction velocity was measured in the tail nerve along with

body weight before exposure and after 4, 8, 12, and 16 weeks of exposure. One animal from each group was sacrificed after 16 weeks of exposure for histopathological evaluation of the nerve fibers in the tail. In addition, nerve-specific proteins (i.e. enolase and  $\beta$ -S100), involved in processes such as cell-cell communication, cell growth, intracellular signal

transduction, and development and maintenance of the central nervous

system, were measured.

Date

Result

: Statistically significant reductions in body weight gain were observed in a for the mid-dose (at 12 weeks) and high-dose (at 8 weeks) rats. Neurological deficits (i.e. reduction in grip strength and comparative slowness of motion) in mid- and high-dose rats were noted from 12 weeks of exposure. No hind-limb paralysis was observed by the time of sacrifice. A reduction in motor nerve conduction velocity, statistically significant during weeks 8-16, was seen with mid and high-dose rats. In addition increased incidence of paranodal swelling, along with some evidence of demyelination and remyelination was present in the peripheral nerves with 1,200 and 3,000 ppm. These histopathologic findings were most sever in the high-dose group. Dose-dependent biochemical changes included reductions in nervous system-specific proteins, particularly the β-S100 protein in tail nerve fibers which was reduced by approximately 75% at all dose levels.

**Conclusion**: Under the conditions of this study, the NOAEL was 500 ppm based on the

neurophysiologic deficits and histopathologic effects seen with 1,200 and

3,000 ppm.

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

08.08.2008 (18)

Type :

Species: mouseSex: male /femaleNumber of animals: 10/sex/groupStrain: B6C3F1Route of admin.: inhalationExposure period: 6 hours/day

**Frequency of treatm.** : 5 days/week for 13 weeks **Doses** : 0, 500, 1000, 4000, 10000 ppm

Control group : yes, NOAEL : 500 - ppm

**Method** : other: none specified

Year : 1991 GLP : no data

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

Remark: Groups of 10 mice/sex/group were exposed to 0, 500, 1,000, 4,000, or

10,000 ppm n-hexane 6 hours/day, 5 days per week for the duration of the study. A second group of 10 mice was exposed at 1,000 ppm for 22 hours/day, 5 days/week for the duration of the study. Separate groups of 8 mice/sex/group received identical treatments but were subjected to neurobehavioral tests before the start of dosing then again after 6 and 13 weeks of exposure. Four males and four females were randomly selected from the 0, 1,000 ppm extended duration, and 10,000 ppm exposure groups for histopathological examination of the spinal cord and tibial nerves. Animals were observed daily for signs of clinical toxicity and

weighed weekly.

**Result** : A full necropsy was performed at sacrifice, weights of the major organs

were recorded, and histopathological evaluations were carried out at term on a variety of excised organs and tissues. The liver was examined only in the males of all exposure groups. Animals exposed to 10,000 ppm n-hexane exhibited some signs of nasal irritation and all animals survived to term. Relative liver, kidney, and heart weights appeared to be increased compared with controls in exposed females. In addition, females exposed to 10,000 ppm 6 hours/day and 1,000 ppm for 22 hours/day exhibited neurobehavioral deficits with a reduction in locomotor activity. There was an increased incidence of paranodal axonal swelling in high-dose or

Date

extended exposure duration mice.

**Conclusion**: It was concluded that n-hexane caused minimal toxicity to the nervous

system and/or respiratory system at 1,000 ppm and above indicating a

NOAEL of 500 ppm.

Reliability : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

08.11.2008 (28)

#### 5.5 GENETIC TOXICITY 'IN VITRO'

Type : Ames test

System of testing: Test species/strain: Salmonella typhimurium TA100, TA1535, TA97, TA98,

TA1537

**Test concentration** : 0, 0.001, 0.0033, 0.010, 0.033, 0.10, and 0.333 mg/plate

Cytotoxic concentr.

Metabolic activation : with and without

Result : negative
Method : no data
Year : 1986
GLP : no data

**Test Method**: Salmonella typhimurium (TA1535, TA1537, TA97, TA98 and TA100) were

incubated with and without metabolic activation. Two metabolic activation systems were used, one with S9 rats livers and the other with Syrian hamster livers. Doses of hexane ranged from 0 to 0.333 mg/plate.

**Result**: The highest negative dose tested in any Salmonella typhimurium strain

was 0.333 mg/plate. Some cultures exhibited slight clearing of the

background bacterial lawn at the two highest doses tested.

Genotoxic effects:

With metabolic activation: negative Without metabolic activation: negative

**Test substance** : CAS No. 110-54-3; hexane **Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

22.07.2008 (27)(28)

Type : Mouse Lymphoma Assay

**System of testing**: L5178Y mouse lymphoma cells (TK locus)

**Test concentration**: 40 to 180 μg/ml

Cycotoxic concentr.

Metabolic activation : with and without

Result : negative

Method : Method equivalent to current guidelines

Year : 1980 GLP : no data

**Method** : The ability of n-hexane to induce specific locus mutations at the TK locus in

cultured L5178Y mouse lymphoma cells was evaluated in the presence and absence of Aroclor-induced rat liver S9 metabolic activation. Based on preliminary toxicity tests, 8 non-activated cultures were treated with 80, 90, 100, 110, 120, 130, 140, or 150  $\mu$ g/ml which produced a range of 0 to 140% total growth. Eight activated cultures were treated with 40, 60, 80, 100, 120, 140, 160, or 180  $\mu$ g/ml which produced a range of 0 to 22% total

growth. The method used was equivalent to the guidelines.

**Date** 

**Result**: None of the non-activated or activated cultures produced mutant

frequencies significantly greater than the solvent controls.

**Test substance** : CAS No. 110-54-3; n-hexane **Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

11.08.2008 (15)

Type : Sister Chromatid Exchange Test

**System of testing** : Mouse bone marrow cells **Test concentration** : 500, 1000, 2000 mg/kg

Cycotoxic concentr. : no data

**Metabolic activation**: with and without

**Result** : negative

Method : Method equivalent to current guidelines

Year : 1982 GLP : no data

**Remark**: No increase in the incidence of sister chromatid exchanges in *in vivo* 

mouse bone marrow cells was seen with intraperitoneal doses of 500,

1,000, or 2,000 mg/kg n-hexane.

**Result**: The dosed groups displayed slight increases in chromosomal aberrations,

but this increase was not considered to be significant.

**Test substance** : CAS No. 110-54-3; hexane **Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

12.08.2008 (28)

Type : Unscheduled DNA synthesis

**System of testing** : Human lymphocytes **Test concentration** : 0.1 to 10 mM cyclohexane

Cycotoxic concentr. : no data

**Metabolic activation**: with and without

Result : negative

**Method** : Method equivalent to current guidelines

Year : 1983 GLP : no data

**Remark** : Human lymphocytes ( + or – S9 mix) were cultured for 4 hours in the

presence or absence of hexane. The effects on the DNA synthesis were

measured through cellular [<sup>3</sup>H]TdR uptake.

**Result**: DNA synthesis was inhibited in human lymphocytes in the presence of

concentrations of n-hexane from  $10^{-4} - 10^{-2}$  M but only at cytotoxic

concentrations

**Test substance** : CAS No. 110-54-3; hexane **Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

22.07.2008 (30)

#### 5.6 GENETIC TOXICITY 'IN VIVO'

Type : Micronucleus

Species: mouseStrain: B6C3F1Sex: no data

Route of admin. : Intraperitoneal injection

Result : negative

Doses Method Year

GLP : no data

Remark

Result: n-Hexane did not induce chromosomal aberrations and micronuclei in bone

marrow cells of B6C3F1 mice injected intraperitoneally.

**Test substance**: CAS No. 110-54-3; hexane; purity is unknown

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

22.07.2008

: Cytogenetic Assay

Type : Cytog Species : rat

Strain : Sprague Dawley
Sex : Male/female
Number of animals : 10 per sex
Route of admin. : inhalation

Exposure period : 6 hours/day for 5 days
Doses : 0, 150, 300, 600 ppm
Method : other: not specified

Year

GLP : no data

Remark

**Result**: A slight, but significant, increase in the number of chromosomal mutations

induced by n-hexane in albino rat bone marrow cells was reported. In addition, an in vivo bone marrow cytogenetic assay found that male albino

rats exposed to 150, 300, and 600 ppm of n-hexane for 5-days experienced a significant increase in chromosomal aberrations at all

treatment levels compared with controls.

**Test substance** : CAS No. 110-54-3; hexane; purity is unknown

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

22.07.2008 (15)

#### 5.7 CARCINOGENICITY

#### 5.8.1 TOXICITY TO FERTILITY

#### 5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Type : Developmental toxicity

Species: ratSex: femaleNumber of animals: 30/group

Strain : Sprague-Dawley
Route of admin. : inhalation
Exposure period : 14 days
Frequency of treatm. : 20 hours/day

**Doses** : 0, 200, 1000, or 5000 ppm

Control group: yesMethod: no dataYear: 1987GLP: no data

Method : The developmental toxicity of n-hexane was assessed using timed-

pregnant (30 animals per group) and virgin (10 animals per group) Sprague-Dawley rats exposed to 0 (filtered air), 200, 1,000, and 5,000 ppm n-hexane vapor in inhalation chambers for 20 hours per day for a period of 14 consecutive days (Mast, 1987). Spermpositive females were exposed on gestation days (GD) 6-19 and virgins were exposed concurrently for 14

consecutive days.

Result

Maternal toxicity, manifested as a reduction in extra-gestational maternal weight gain, was observed at all exposure levels, and was statistically significant for the 5,000 ppm exposure group. Extra-gestational maternal weight gain (calculated from GD 0 to GD 20) relative to control animals was reduced for the 200, 1,000, and 5,000 ppm exposure groups. Cumulative weight gain (CWG) for dams in the 1,000 and 5,000 ppm exposure groups was significantly reduced with respect to controls by GD 20. The CWG for the 5,000 ppm was also significantly reduced with respect to controls by GD 13.

Comparison of n-hexane exposed groups with the control group (0 ppm) indicated that gestational exposure to n-hexane did not result in an increase in the incidence of intrauterine deaths or in the incidence of fetal malformations. A statistically significant reduction in fetal body weight relative to controls was observed for males at the 1,000 and 5,000 ppm exposure levels. Female weights were also reduced with respect to controls for these exposure levels, but the reduction was statistically significant for only the 5,000 ppm group. Gravid uterine weight was also significantly less than controls for the 5,000 ppm exposure groups. A statistically significant increase in the mean percent incidence per litter of reduced ossification of sternebrae 1-4 was observed for the 5,000 ppm group, and was positively correlated with exposure concentration. This increased incidence of reduced ossification in the sternebrae, and the reduction in fetal body weight at the 5,000 ppm level, may have been interrelated manifestations of slight growth retardation.

No major abnormalities were found in any of the fetuses. Variations observed included dilated ureter, renal pelvic cavitation, supernumerary ribs, and reduced skeletal ossifications at several sites. The increase in mean percent incidence per litter of reduced ossification of sternebrae 1-4 was statistically significant for the highest exposure concentration, and the increase was positively correlated with increasing exposure concentration.

**Conclusion**: The NOAEL for developmental toxicity was 200 ppm.

**Test substance**: CAS No. 110-54-3; hexane; purity is unknown

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

18.07.2008 (24)

Type : rat Sex : male

Number of animals : 12 to 39/group
Strain : Sprague-Dawley
Route of admin. : inhalation
Exposure period : 24 hours

Frequency of treatm. : single
Doses : 5000 ppm
Control group : yes
Method : no data
Year : 1987
GLP : no data

**Method**: The effect of n-hexane on the male reproductive system when

administered via the inhalation route was examined by exposing male Sprague-Dawley rats (12-39/group) to 5,000 ppm n-hexane in either a single 24 hour exposure, repeated 16 hour/day exposures for up to 8 days,

or repeated 16 hour/day exposures, 6 hours/day for up to 6 weeks.

**Result**: Rats exposed to 5,000 ppm n-hexane displayed some evidence of

neuropathy such as paralysis. Focal degeneration of spermatocytes and exfoliation of elongated spermatids was also observed in response to treatment. Early meiotic prophase spermatocytes and transitional spermatocytes as well as those undergoing meiotic metaphase appeared

to be more susceptible to the action of n-hexane than pachytene

spermatocytes. Rats exposed repeatedly to 5,000 ppm n-hexane over a 6-week period showed complete atrophy of the seminiferous tubules. In

addition, a reduction in food consumption and body weight gain accompanied by signs of incipient neuropathy were seen with repeated n-hexane exposure. A wide range of testicular lesion did not complete resolve during a recovery period even though body weights and clinical

symptoms improved.

**Test substance**: CAS No. 110-54-3; hexane; purity is unknown

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag : Critical study for SIDS endpoint

11.08.2008 (7)

Type :

Species: mouseSex: femaleNumber of animals: 33/groupStrain: CD-1Route of admin.: inhalationExposure period: 14 daysFrequency of treatm.: 20 hours/day

**Doses** : 0, 200, 1000, or 5000 ppm

Control group : yes
Method : no data
Year : 1988
GLP : no data

Method : Timed-pregnant (~33 females per group) and virgin (10 females per group)

Swiss (CD-1) mice were exposed to 0, 200, 1,000, and 5,000 ppm n-hexane (99.2% purity) vapor in inhalation chambers, 20 h/day, for a period of 12 consecutive days. Plug-positive females were exposed on GD 6-17.

**Result**: Maternal body weight at sacrifice (GD 18) and total cumulative weight gain

for dams in the 5,000 ppm exposure group were significantly reduced with respect to controls; however, this was due to an exposure correlated reduction in gravid uterine weight, not to a decrease in extragestational gain. An exposure-correlated decrease in the gravid uterine weight to extragestational weight gain ratio (significant for the 5,000 ppm group)

occurred in the absence of an effect on placental weight.

Gestational exposure to n-hexane resulted in an increase in the number of resorbed fetuses for all exposure groups relative to the control group; however, the increases were not directly correlated to exposure

concentration. The differences were statistically significant for the 200-ppm group with respect to total intrauterine death (early plus late resorptions), and with respect to late resorptions for the 5,000 ppm group. A small, but statistically significant, reduction in female (but not male) fetal body weight relative to the control group was observed at the 5,000 ppm exposure level. There were no exposure-related increases in any individual fetal malformation or variation, nor was there any increase in the incidence of combined malformations or variations.

Gestational exposure of CD-1 mice to n-hexane vapors appeared to cause a degree of concentration-related developmental toxicity in the absence of overt maternal toxicity, but the test material was not found to be teratogenic. This developmental toxicity was manifested as an increase in the number of resorptions per litter for all exposure levels, and as a decrease in the uterine: extra-gestational weight gain ratio at the 5,000 ppm exposure level. Because of the significant increase in the number of resorptions at the 200-ppm exposure level, a NOEL for developmental toxicity was not established for exposure of mice to 200, 1,000, or 5,000 ppm n-hexane vapors.

**Conclusion**: The LOAEL for developmental toxicity (in mice) was 200 ppm.

Test substance

: CAS No. 110-54-3; hexane; purity 99.2%

Reliability

: (2) valid with restrictions

Original study not retrieved and reviewed. Data from reliable peer-

reviewed source.

Flag

: Critical study for SIDS endpoint

11.08.2008 (26)

#### 5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

#### 5.9 SPECIFIC INVESTIGATIONS

#### 5.10 EXPOSURE EXPERIENCE

#### 5.11 ADDITIONAL REMARKS

6. Analyt. Meth. for Detection and Identification	ld 110-54-3  Date 30.06.2008
6.1 ANALYTICAL METHODS	
6.2 DETECTION AND IDENTIFICATION	
30 / 36	

7. Ef	f. Against Target Org. and Intended Uses	ld Date	110-54-3
7.1	FUNCTION		
7.2	EFFECTS ON ORGANISMS TO BE CONTROLLED		
7.3	ORGANISMS TO BE PROTECTED		
7.4	USER		
7.5	RESISTANCE		
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## **Id** 110-54-3 8. Meas. Nec. to Prot. Man, Animals, Environment **Date** 30.06.2008 8.1 METHODS HANDLING AND STORING 8.2 FIRE GUIDANCE **EMERGENCY MEASURES** 8.3 8.4 POSSIB. OF RENDERING SUBST. HARMLESS **WASTE MANAGEMENT** 8.6 **SIDE-EFFECTS DETECTION** 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER 8.8 REACTIVITY TOWARDS CONTAINER MATERIAL 32 / 36

### 9. References Id 110-54-3

**Date** 30.06.2008

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### 9. References Id 110-54-3 Date 30.06.2008

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10. Summary and Evaluation		110-54-3
	Date	30.06.2008
10.1 END POINT SUMMARY		
10.2 HAZARD SUMMARY		
10.3 RISK ASSESSMENT		
	36 / 36	

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## IUCLID

## **Data Set**

**Existing Chemical** 

CAS No.

EINECS Name

EC No. TSCA Name

Molecular Formula

IUPAC Name

: ID: 110-82-7

: 110-82-7

: Cyclohexane

: 203-806-2

: Cyclohexane

: C6H12

: Cyclohexane

Producer related part

Company

: ExxonMobil Biomedical Sciences Inc.

Creation date

: 30.06.2008

Substance related part

Creation date

Company

: ExxonMobil Biomedical Sciences Inc.

: 30.06.2008

Status

Memo

: U.S. EPA - HPV Challenge Program

Printing date

: 30.06.2008

Revision date

Date of last update

: 30.06.2008

Number of pages

: 41

Chapter (profile) Reliability (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10

: Reliability: without reliability, 1, 2, 3, 4

Flags (profile)

: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE),

Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

### ld 110-82-7 1. General Information Date 1.0.1 APPLICANT AND COMPANY INFORMATION 1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR 1.0.3 IDENTITY OF RECIPIENTS 1.0.4 DETAILS ON CATEGORY/TEMPLATE 1.1.0 SUBSTANCE IDENTIFICATION 1.1.1 GENERAL SUBSTANCE INFORMATION **Purity type** Substance type petroleum product Physical status liquid Purity Colour Odour 01.07.2008 1.1.2 SPECTRA 1.2 SYNONYMS AND TRADENAMES 1.3 **IMPURITIES ADDITIVES** 1.5 TOTAL QUANTITY 1.6.1 LABELLING 1.6.2 CLASSIFICATION 1.6.3 PACKAGING

## **Date** 30.06.2008 1.7 USE PATTERN 1.7.1 DETAILED USE PATTERN 1.7.2 METHODS OF MANUFACTURE 1.8 REGULATORY MEASURES 1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES 1.8.2 ACCEPTABLE RESIDUES LEVELS 1.8.3 WATER POLLUTION 1.8.4 MAJOR ACCIDENT HAZARDS 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

1. General Information

**Id** 110-82-7

#### 2. Physico-Chemical Data

ld 110-82-7

Date

#### 2.1 MELTING POINT

Value : = 6.6 °C

Sublimation

**Method** : other: not specified

Year :

GLP : no data

Test substance

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

The CRC Handbook of Chemistry and Physics is a peer reviewed

publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.

Flag : Critical study for SIDS endpoint

01.07.2008 (27)

#### 2.2 BOILING POINT

**Value** : = 80.7 °C at 1013 hPa

Decomposition

Method : other: not specified

Year

GLP : no data

Test substance :

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

The CRC Handbook of Chemistry and Physics is a peer reviewed

publication. This robust summary has a reliability rating of 2 because there is insufficient information available on the method and analytical procedure.

Flag : Critical study for SIDS endpoint

01.07.2008 (27)

#### 2.3 DENSITY

Type : density

Value : = .774 g/cm³ at 20 °C Method : other: not specified

Year

GLP : no data

Test substance :

**Test substance**: CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

Data supplied by the experimental database associated with EPISuite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-

reviewed publication.

Flag : Critical study for SIDS endpoint

01.07.2008 (43)

#### 2.3.1 GRANULOMETRY

# 2. Physico-Chemical Data

**Id** 110-82-7 **Date** 30.06.2008

#### 2.4 **VAPOUR PRESSURE**

 $: = 12.9 \text{ hPa at } 25 \,^{\circ}\text{C}$ **Value** 

**Decomposition** 

Method other (measured): not specified

Year

**GLP** no data

Test substance

Test substance CAS No. 110-82-7; cyclohexane; purity is unknown

Reliability (2) valid with restrictions

> Data supplied by the experimental database associated with EPISuite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-

reviewed publication.

: Critical study for SIDS endpoint Flag

01.07.2008 (4)

#### 2.5 **PARTITION COEFFICIENT**

**Partition coefficient** octanol-water = 3.44 at 20 °C Log pow

pH value

Method other (measured): not specified

Year

**GLP** no data

Test substance

CAS No. 110-82-7; cyclohexane; purity is unknown Test substance

(2) valid with restrictions Reliability

Data supplied by the experimental database associated with EPISuite. This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-

reviewed publication.

: Critical study for SIDS endpoint Flag

01.07.2008 (14)

# 2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in Water

**Value** = 55 mg/l at 25 °C

pH value

at °C concentration

Temperature effects

Examine different pol.

at 25 °C pKa

Description

**Stable** 

Deg. product

Method other: not specified

Year

**GLP** no data

**Test substance** 

Test substance CAS No. 110-82-7; cyclohexane; purity is unknown

Reliability (2) valid with restrictions

Data supplied by the experimental database associated with EPISuite.

# 2. Physico-Chemical Data

2.14 ADDITIONAL REMARKS

ld 110-82-7

**Date** 

This robust summary has a reliability rating of 2 because the data was not reviewed for quality, however, the reference is associated with a peer-

reviewed publication.

Flag : Critical study for SIDS endpoint (30)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

2.8 AUTO FLAMMABILITY

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

2.13 VISCOSITY

Date

# 3.1.1 PHOTODEGRADATION

Type : air
Light source : Sun light
Light spectrum : nm

Relative intensity : based on intensity of sunlight

**INDIRECT PHOTOLYSIS** 

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm³

**Rate constant** : = .0000000000848 cm³/(molecule\*sec)

**Degradation** : = 50 % after 15.1 hour(s)

Deg. product

Method : other (calculated): Calculated values using AOPWIN version 1.92, a

subroutine of the computer program EPI SuiteTM version 3.20

Year :

GLP : Test substance :

Method : Calculated values using AOPWIN version 1.92, a subroutine of the

computer program EPI SuiteTM version 3.20

Indirect photodegradation, or atmospheric oxidation potential, is based on the structure-activity relationship methods developed by R. Atkinson under

the following conditions: Temperature: 25°C Sensitizer: OH- radical

Concentration of Sensitizer: 1.5E6 OH- radicals/cm3

**Remark**: Cyclohexane has the potential to volatilize to air, based on a relatively high

vapor pressure, where it is subject to atmospheric oxidation.

In air, cyclohexane can react with photosensitized oxygen in the form of hydroxyl radicals (OH-). The computer program AOPWIN (atmospheric oxidation program for Microsoft Windows) (EPI SuiteTM, 2000) calculates a

chemical half-life for a 12-hour day (the 12-hour day half-life value normalizes degradation to a standard day light period during which

hydroxyl radicals needed for degradation are generated), based on an OH-

reaction rate constant and a defined OH- concentration.

Based on a 12-hour day, a rate constant of 8.48 E-12 cm3/molecule\*sec, and an OH- concentration of 1.5 E6 OH-/cm3, cyclohexane has a calculated half-life in air of 1.3 days or 15.1 hours of daylight.

**Test substance**: CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

01.07.2008 (46)

Type : water

Light source :

**Light spectrum** : nn

**Relative intensity**: based on intensity of sunlight

**Test condition**: Direct photochemical degradation occurs through the absorbance of solar

radiation by a chemical substance in aqueous solution. If the absorbed energy is high enough, then the resultant excited state of the chemical may undergo a transformation. A prerequisite for direct photodegradation is the

ability of one or more bonds within a chemical to absorb ultraviolet

(UV)/visible light in the 290 to 750 nm range.

ld 110-82-7 **Date** 30.06.2008

Light wavelengths longer than 750 nm do not contain sufficient energy to break chemical bonds, and wavelengths below 290 nm are shielded from the earth by the stratospheric ozone layer (Harris, 1982).

An approach to assessing the potential for a substance to undergo photochemical degradation is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by constituent molecules (Zepp and Cline, 1977). Saturated and unsaturated hydrocarbons do not absorb light above 290 nm. Consequently, cyclohexane is not subject to photolytic processes in the aqueous

environment.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability of 2 because it is a technical

discussion and not a study.

Flag : Critical study for SIDS endpoint

01.07.2008 (15)(48)

#### 3.1.2 STABILITY IN WATER

Deg. product

Method : other: Technical Discussion

Year

GLP : no data

Test substance :

**Result** : Hydrolysis of an organic chemical is the transformation process in which a

water molecule or hydroxide ion reacts to form a new carbon-oxygen bond. Chemicals with leaving groups that have a potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Gould, 1959). The lack of a suitable leaving group renders a compound resistant to hydrolysis. Cyclohexane is resistant to hydrolysis because it lacks a functional group that is hydrolytically reactive and Harris (1982) identifies hydrocarbons as generally resistant to hydrolysis. Therefore, hydrolysis will not contribute to

the removal of cyclohexane from the environment.

: CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability of 2 because it is a technical

discussion and not a study.

Flag : Critical study for SIDS endpoint

01.07.2008 (13)(16)

#### 3.1.3 STABILITY IN SOIL

Test substance

# 3.2.1 MONITORING DATA

# 3.2.2 FIELD STUDIES

# 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level I

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)

ld 110-82-7

Date

Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)
Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

**Year** : 2003

Method : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:

Molecular mass = 84.16 g/mol Water solubility = 55 g/m<sup>3</sup> Vapour pressure = 12,919 Pa

log Kow = 3.44

Melting point = 6.6 deg C

Result

Air - 99.91% Water - 0.03% Soil - 0.06% Sediment - <0.01% Suspended Sed - <0.01%

Biota - < 0.01%

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

01.07.2008 (28)

Type : fugacity model level III

Media : other: air - sediment(s) - soil - water

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level III

Year : 2003

Method : The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.20 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, and sediment).

Input values used:

**Id** 110-82-7

**Date** 

Molecular mass = 84.16 g/mol Water solubility = 55 g/m<sup>3</sup> Vapour pressure = 12,919 Pa

log Kow = 3.44

Melting point = 6.6 deg C

Degradation half-lives:

Air - 15.1 hrs Water - 360 hrs Soil - 720 hrs Sediment - 7200 hrs

Environmental Properties (EQC standard environment)

Dimensions (all defaults)
Densities (all defaults)

Organic carbon & Advection (all defaults)

Transport Velocities (all defaults)

Emission and Inflows (defaults used)

Air 1000 kg/hr Water 1000 kg/hr Soil 1000 kg/hr Sediment 0 kg/hr

Result : Output:

Mass% Emissions(kg/hr)

 Air
 15.8
 1000

 Water
 67.3
 1000

 Soil
 12.9
 1000

 Sediment
 4.0
 0

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

01.07.2008 (28)

# 3.3.2 DISTRIBUTION

# 3.4 MODE OF DEGRADATION IN ACTUAL USE

# 3.5 BIODEGRADATION

Type : aerobic

Inoculum : activated sludge Contact time : 28 day(s)

**Degradation** : 77 (±) % after 28 day(s) **Result** : other: readily biodegradable

Deg. product

Method : OECD Guide-line 301 F "Ready Biodegradability: Manometric

Respirometry Test"

**Year** : 1995 **GLP** : yes

**Test substance**: CAS No. 110-82-7; cyclohexane; purity is unknown

**Id** 110-82-7 **Date** 30.06.2008

Result : Test material was readily biodegradable. Half-life was reached by day 20.

By day 28, 77% degradation of the test material was observed. 10%

biodegradation was achieved on day 13.

By day 2, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were noted. Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an elemental analysis of the test material.

Mean % Degradation % Degradation\*

<u>Sa</u>mple (day 28) (day 28) Test Material 79.8, 70.7, 80.2 76.9 Na Benzoate 91.0.90.4 90.7

\* replicate data

: Non acclimated activated sludge and test medium were combined prior to **Test condition** 

> test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate,

Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically

monitored for oxygen consumption.

Test material was tested in triplicate, controls and blanks were tested in

duplicate.

Test material concentration was approximately 34 mg/L. Sodium benzoate

(positive control) concentration was 50mg/L.

Test temperature was 22 +/- 1 Deg C.

All test vessels were stirred constantly for 28 days using magnetic stir bars

Test substance : CAS No. 110-82-7; cyclohexane; purity is unknown.

: Cyclohexane is readily biodegradable. Conclusion

: (1) valid without restrictions Reliability

01.07.2008 (9)

# **BOD5, COD OR BOD5/COD RATIO**

#### 3.7 **BIOACCUMULATION**

**Species** other: see remark

Exposure period at 25 °C

Concentration

**BCF** = 89

Elimination

other: calculation Method

Year

**GLP** 

**Test substance** CAS No. 110-82-7; cyclohexane; purity is unknown

Remark A log bioconcentration factor (BCF) of 1.95 is calculated (BCF = 89). With

respect to a log Kow = 3.44, which was used to calculate the BCF,

cyclohexane in the aquatic environment is expected to have a low potential

to bioaccumulate.

Test substance : CAS No. 110-82-7; cyclohexane; purity is unknown

(2) valid with restrictions Reliability

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

: Critical study for SIDS endpoint Flag

01.07.2008 (43)

3. Environmental Fate and Pathways	110-82-7 30.06.2008
3.8 ADDITIONAL REMARKS	
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Date

# 4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : Flow Through Acute Fish Toxicity Test Species : Pimephales promelas (Fish, fresh water)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : 4.53 measured/nominal

Limit test

Analytical monitoring : no data

Method :

Year : 1987 GLP : no data

Statistical Method : Trimmed Spearrman Karber Method

**Test Conditions**: Treatment solutions were prepared by diluting a 37.9mg/L stock solution. Nominal cyclohexane treatment levels were 1.62, 3.24, 4.86, 6.48, 8.10

mg/L, which measured 2.00, 3.53, 4.84, 6.96 and 8.86mg/L, respectively.

Control/dilution water was EPA Duluth laboratory water.

Ten fish were tested per treatment. Treatment volume = 250ml.

Test parameters were as follows: temperature = 25.2 Deg C (s.d. 0.14); dissolved oxygen (DO) = 7.2 mg/L (s.d. 0.38); pH = 7.5 (s.d. 0.10); fish age = 30 days old; fish mean wt. = 0.119 g; fish mean length = 20.5 mm; fish loading = 0.1190 g/L/day. Temperature, DO, and pH data from the treatment solutions were only provided as mean values with standard deviations.

Organism supplier was U.S. EPA Environmental Research Lab, Duluth, MN, USA.

**Results** : 96-hour LC50 = 4.53 mg/L (95% CI 3.96 to 5.18) based upon measured

concentrations

Analytical method used was Gas-Liquid Chromatography.

Measured	Fish Total Mortality
Conc. (mg/L)	(@ 24, 48, 72, 96 hrs)*
Control	0, 0, 0, 0
2.00	0, 0, 0, 0
3.52	1, 1, 1, 1
4.84	0, 2, 3, 6
6.96	10, 10, 10, 10
8.86	10, 10, 10, 10

\* 10 fish added at test initiation

**Test substance**: CAS No. 110-82-7; cyclohexane; purity is unknown

**Conclusion** : 96-hour LC50 = 4.53 mg/L (95% CI 3.96 to 5.18) based upon measured

concentrations

**Reliability** : (1) valid without restrictions

Although a standard method was not cited, the testing procedures followed generally accepted fish acute toxicity guideline methods and sufficient information on testing method and conditions was available to rate this

study as "reliable without restriction".

Flag : Critical study for SIDS endpoint

01.07.2008 (12)

Туре

**Species**: other: freshwater fish

**Exposure period** : 96 hour(s)

Date

Unit : mg/l

**LC50** : = 2.8 calculated

Method : other: ECOSAR Computer Model

Year : GLP :

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Method**: ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships

(SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

Result :

Calculated 96-hr LC50 for fish = 2.8 mg/L

Test condition : Experimental water solubility = 55.0 mg/l @ 25°C (McAuliffe, 1966), log

Kow = 3.44 @ 20°C (Hansch, et.al, 1995), and melting point = 6.6°C (Lide,

2003) were entered into the program.

Class: Neutral organics

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

01.07.2008 (3)

Type : Fish Acute Toxicity Test

**Species**: Oncorhynchus mykiss (Fish, fresh water, marine)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : 3.2 measured/nominal

**Limit test** : no **Analytical monitoring** : no

Method : OECD 203 Fish Acute Toxicity Test

Year : 2000 GLP : yes

Statistical Method : Binomial Method

**Test Conditions**: Individual Water Accomodated Fractions (WAF's) were prepared for each

test treatment. The test substance was added volumetrically, via a syringe, to 19L of dilution water in a 20L glass carboy. The solutions were mixed for

Date

24 hours at a vortex of </= 10% of the total depth. The test solutions were pumped from each mixing vessel into three replicates of 4.5L in 4.0L glass aspirator bottles (no headspace). Five fish were added to each test replicate and the replicates sealed. Daily renewals were performed by removing  $\sim 80\%$  of the test solution through the port at the bottom and refilling with fresh solution.

Test temperature was 14.5 Deg C., Lighting was 16 hours light: 8 hours dark with 562 to 728 Lux during full daylight periods.

Dissolved Oxygen at initiation ranged from 8.6 to 8.9 mg/L and from 7.1 to 8.1 mg/L in "old" solutions prior to renewals. The pH was ranged from 7.5 to 7.8 during the study. Fish were not fed during the study. Water hardness ranged from 92 to 96 mg/L as CaCO<sub>3</sub>.

Fish Mean Wt.= 0.213g. Mean Total length = 3.1cm, Test Loading = 0.24 g

: LL50 = 3.2mg/L (Cl 1.0 to 10.0), based upon nominal loading levels.

 Nominal Conc.
 % Mortality @ 96 hr.

 Control
 0

 1 mg/L
 0

 10 mg/L
 10

 100 mg/L
 10

Dissolved oxygen levels dropped below 60% of saturation in some of the treatments on Days 1 through 4 of the test. Since no mortality occurred in these treatments, the deviations are not believed to have affected the outcome of the study.

Test substance

: CAS No. 110-82-7; cyclohexane; purity is unknown

Reliability

Results

: (2) valid with restrictions

No analytical monitoring of test concentrations was performed.

01.07.2008 (11)

# 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Туре

Species : Daphnia sp. (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

**EC50** : = 3.3 calculated

Method : other: ECOSAR Computer Model

Year :

GLP :

**Test condition**: ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships

(SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be

**Id** 110-82-7 **Date** 30.06.2008

> calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

> To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

CAS No. 110-82-7; cyclohexane; purity is unknown Test substance Calculated 48-hr LC50 for a dahpnid = 3.3 mg/L Result

Experimental water solubility = 55.0 mg/l @ 25°C (McAuliffe, 1966), log **Test condition** 

Kow = 3.44 @ 20°C (Hansch, et.al, 1995), and melting point = 6.6°C (Lide,

2003) were entered into the program.

Class: Neutral organics (2) valid with restrictions Reliability

> The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag Critical study for SIDS endpoint

08.07.2008 (3)

Type static

other: Daphnia Species **Exposure period** 48 hour(s) Unit mg/l **EC50** = 0.9

Method other: based on discussions in GESAMP/MARPOL meetings held in 1973

Year

**GLP** 

Method Individual treatment concentrations were prepared by mixing the test

substance in freshwater for 24 hours in a conical flask. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were 250 ml conical flasks with 25 daphnids per flask. Two replicates of each

treatment level and control were evaluated.

Organisms supplied by testing lab; age = <24 hours old; parents age =

approximately 21 days old.

Statistical method:

Parametric model developed by Kooiiman (Kooiiman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

119.).

48-hr EC50 for a daphnid = 0.9 mg/L Result

Dissolved oxygen was >50% saturation during the study. The pH was 7.5 **Test condition** 

to 8.3. Temperature was 20 Deg C.

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 1.0 to 10 mg/L.

Test substance Reliability

CAS No. 110-82-7; cyclohexane; purity is unknown

(2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient

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information in the report to suggest that the testing procedure generally followed an acceptable test guideline, OECD 202, and used acceptable

methods to prepare exposure solutions.

Flag : Critical study for SIDS endpoint

30.06.2008 (44)

Type : semistatic

**Species**: other: Gammarid (Chaetogammarus marinus)

 Exposure period
 : 96 hour(s)

 Unit
 : mg/l

 LC50
 : = 2.2

Method : other: Static Gammarid Acute Toxicity Test

Year

GLP : no data

Method : Individual treatment solutions were prepared by mixing the test substance

in freshwater for 24 hours in conical flasks. The flask was almost completely filled with solution. After mixing, the treatment solutions were allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24

hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity

of 2.8%; length = 5 mm. Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

119.).

**Result** : 96-hr LC50 for a gammarid = 2.2 mg/L

**Test condition**: Dissolved oxygen was >50% saturation during the study. The pH was 7.5

to 8.3. Temperature was 15 Deg C. Natural seawater was used with a

salinity of 2.8%

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 1.0 to 10 mg/L.

Test solutions were analyzed only upon test initiation.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to

prepare exposure solutions.

Flag : Critical study for SIDS endpoint

30.06.2008 (44)

Type : semistatic

**Species**: other: mysid shrimp (Mysidopsis bahia)

**Exposure period** : 96 hour(s) **Unit** : mg/l **LC50** : = 2.2

Method : other: Static Gammarid Acute Toxicity Test

Year

GLP : no data

Method : Individual treatment solutions were prepared by mixing the test substance

in freshwater for 24 hours in conical flasks. The flask was almost

completely filled with solution. After mixing, the treatment solutions were

Date

allowed to settle for 24 hours. The aqueous solution was then drained through a stopcock at the base of the flask and tested. Test vessels were scintillation vials almost filled with approximately 20 ml of test solution and one organism per vial. Ten organisms were tested per treatment level. Organisms were transferred into fresh control and test solutions every 24 hours up to 96 hours.

Organisms supplied by testing lab, grown in natural seawater with a salinity of 2.8%; test organisms were approximately 4 weeks old, with lengths of approximately 6 mm.

Statistical method:

Parametric model developed by Kooijman (Kooijman, S.A.L.M. 1981. Parametric analyses of mortality rate in bio-assays. Water Res., 15:107-

119.).

Result : 96-hr LC50 for a mysid = 2.2 mg/L

**Test condition**: Dissolved oxygen was >50% saturation during the study. The pH was 7.5

to 8.3. Temperature was 20 Deg C. Natural seawater was used with a

salinity of 2.8%

Analytical method used was Gas Chromatography with Flame Ionization Detection (GC-FID). Nominal treatment levels ranged from 0.32 to 10 mg/L.

Test solutions were analyzed only upon test initiation.

Test substance

: CAS No. 110-82-7; cyclohexane; purity is unknown

**Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because there is less raw data and information on the testing procedure than is desirable in order to rate this study for reliability at a level higher than 2. There is sufficient information in the report to suggest that the testing procedure generally followed an acceptable test guideline and used acceptable methods to

prepare exposure solutions.

Flag : Critical study for SIDS endpoint

30.06.2008 (44)

# 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species : other algae: Pseudokirchneriella subcapitata

Endpoint

Exposure period : 72 hour(s)
Unit : mg/l

**EC50** : = 3.4 measured

Method : OECD 201 - Alga, Growth Inhibition Test

**Year** : 1998 **GLP** : yes

Statistical Method
Test condition

Proc regression procedure of SAS, Anova procedure of SAS for NOEC

Individual test treatment solutions were prepared as Water Accommodated Fractions (WAFs). Test material was added to algal media via syringe in 2.0L aspirator bottles. The bottles were completely filled, no headspace. The mixing vessels were sealed with Teflon-covered stoppers and mixed on magnetic stir plates with teflon coated stir bars for 23.5 hours at room temperature. After mixing the solutions were allowed to settle for one hour and the WAF was removed from the bottom of the mixing vessel via the port and used for testing. Test vessels were 125ml glass Erlenmeyer flasks with approximately 140 ml of treatment solution and inoculated with algae. Test vessels were completely filled and sealed with glass stoppers. Samples were taken daily for cell counts. Six replicates were prepared for each treatment level. The initial algal concentration was 1.0 x 10<sup>4</sup> cells/ml. All test replicates were placed on a shaker table at 100 oscillations per minute

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during the study. Biomass was calculated as the area under the growth curve.

Nominal loading levels were 0.5, 1.4, 3.9, 11, and 31 mg/L

Test temperature was 24.6 Deg. C. Lighting was continuous at approximately 4200 Lux. The pH was 7.5 to 7.6 at test initiation and ranged from 8.4 to 8.7 at test termination.

Test treatments were analyzed by GC-FID. Measured values on Day 0 were 0.085, 0.719, 0.952, 2.940 and 4.425 mg/L. The test material was not detected in the control.

**Test substance**: CAS No. 110-82-7; cyclohexane; purity is unknown

**Result** : 72-hr EC50 for a green alga = 3.4 mg/L

72-hr NOEC for a green alga = 0.952 mg/L

**Reliability** : (1) valid without restrictions

Flag : Critical study for SIDS endpoint

02.07.2008 (10)

Species : other algae: Pseudokirchneriella subcapitata

Endpoint

**Exposure period** : 96 hour(s) **Unit** : mg/l

**EC50** : = 2.2 calculated **ChV** : = 0.5 calculated

**Method** : other: ECOSAR Computer Model

Year :

GLP Test condition

ECOSAR version 0.99h, U.S. EPA. The structure-activity relationships (SARs) presented in this program are used to predict the aquatic toxicity of chemicals based on their similarity of structure to chemicals for which the aquatic toxicity has been previously measured. Most SAR calculations in the ECOSAR Class Program are based upon the octanol/water partition coefficient (Kow). SARs have been used by the U.S. Environmental Protection Agency since 1981 to predict the aquatic toxicity of new industrial chemicals in the absence of test data. SARs are developed for chemical classes based on measured test data that have been submitted by industry or they are developed by other sources for chemicals with similar structures, e.g., phenols. Using the measured aquatic toxicity values and estimated Kow values, regression equations can be developed for a class of chemicals. Toxicity values for new chemicals may then be calculated by inserting the estimated Kow into the regression equation and correcting the resultant value for the molecular weight of the compound.

To date, over 150 SARs have been developed for more than 50 chemical classes. These chemical classes range from the very large, e.g., neutral organics, to the very small, e.g., aromatic diazoniums. Some chemical classes have only one SAR, such as acid chlorides, for which only a fish 96-hour LC50 has been developed. The class with the greatest number of SARs is the neutral organics, which has SARs ranging from acute and chronic SARs for fish to a 14-day LC50 for earthworms in artificial soil. The ECOSAR Class Program is a computerized version of the ECOSAR analysis procedures as currently practiced by the Office of Pollution Prevention and Toxics (OPPT). It has been developed within the regulatory constraints of the Toxic Substances Control Act (TSCA). It is a pragmatic approach to SAR as opposed to a theoretical approach.

Test substance Result CAS No. 110-82-7; cyclohexane; purity is unknown
 Calculated 96-hr EC50 for a green alga = 2.2 mg/L
 Calculated 96-hr ChV for a green alga = 0.5 mg/L

**Test condition** 

Experimental water solubility = 55.0 mg/l @ 25°C (McAuliffe, 1966), log Kow = 3.44 @ 20°C (Hansch, et.al, 1995), and melting point = 6.6°C (Lide,

4. Ecotoxicity

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2003) were entered into the program.

Class: Neutral organics

**Reliability** : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag

: Critical study for SIDS endpoint

02.07.2008 (3)

- 4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA
- 4.5.1 CHRONIC TOXICITY TO FISH
- 4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES
- 4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS
- 4.6.2 TOXICITY TO TERRESTRIAL PLANTS
- 4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS
- 4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES
- 4.7 BIOLOGICAL EFFECTS MONITORING
- 4.8 BIOTRANSFORMATION AND KINETICS
- 4.9 ADDITIONAL REMARKS

ld 110-82-7 5. Toxicity

**Date** 

# 5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

# 5.1.1 ACUTE ORAL TOXICITY

Type LD50

**Value** >5000 mg/kg

Species rat Strain no data Sex no data

Number of animals

Vehicle no data

**Doses** Method

Year **GLP** 

Remark

Test substance CAS No. 110-82-7; cyclohexane; purity is unknown

Conclusion Cyclohexane has a low order of toxicity by the oral route of exposure

Reliability (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Critical study for SIDS endpoint Flag

14.07.2008 (32)

LD50 Type

Value =29,800 mg/kg

Species Strain no data Sex no data

Number of animals

Vehicle no data

Doses Method Year **GLP** Remark

CAS No. 110-82-7; cyclohexane; purity is unknown Test substance

Cyclohexane has a low order of toxicity by the oral route of exposure Conclusion

Reliability (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Critical study for SIDS endpoint Flag

14.07.2008 (7)

Type LD50

Value 8,000 - 39,000 mg/kg

Species rat Strain no data Sex no data

Number of animals

Vehicle no data

**Doses** Method Year

1970 **GLP** : pre-GLP

Remark : The oral LD50 in a 14-day old rat, a young adult rat and an older rat was

8.0, 39.0 and 16.5 ml/kg, respectively (6,240, 30,420 and 12,870 mg/kg,

Date

respectively). Symptoms included depressive effect on the central nervous

system, salivation and soft faeces.

**Test substance**: CAS No. 110-82-7; cyclohexane; purity is unknown

**Conclusion** : Cyclohexane has a low order of toxicity by the oral route of exposure

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (25)

# 5.1.2 ACUTE INHALATION TOXICITY

Type : LC50

**Value** : > 32.88 mg/l

Species: ratStrain: no dataSex: no data

Number of animals

Vehicle : no data

Doses

**Exposure time** : 4 hour(s) **Method** : other

Year

GLP : no data

**Test substance**: CAS No. 110-82-7; cyclohexane; purity is unknown

Remark : Single 4-hour exposure at one dose level of 9500 ppm (32.28 mg/l). No

death occurred at level tested. Exposure-related symptoms noted during the exposure included tremors, hyperactivity, rapid respiration, and also

hypoactivity.

Result :

**Conclusion**: Cyclohexane has a low order of toxicity by the inhalation route of exposure

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (33)

Type : LC50 Value :

Species : rabbit Strain : no data Sex : no data

Number of animals

Vehicle : no data

Doses

Exposure time : 1 hour(s)

Method : other

Year :

**GLP**: pre-GLP

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Remark**: A 1 hour acute inhalation exposure of rabbits to cyclohexane vapor

produced effects on the central nervous system, with exposure-related observations including convulsions, tremors, hyperactivity, rapid

respiration, cyanosis and diarrhea; all the animals exposed to 89.6 mg/L

died (26,000 ppm).

Result

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

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Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (45)

# 5.1.3 ACUTE DERMAL TOXICITY

Type : LD50

**Value** : > 2000 mg/kg bw

Species : rabbit
Strain : no data
Sex : no data

Number of animals

Vehicle : no data

Doses

**Method** : other

Year :

GLP : no data

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

**Result**: No deaths or systemic symptoms were observed, a slight erythema and

oedema were noted in a few animals.

**Conclusion** : Cyclohexane has a low order of toxicity by the dermal route of exposure

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (34)

# 5.1.4 ACUTE TOXICITY, OTHER ROUTES

# 5.2.1 SKIN IRRITATION

Species: rabbitStrain: no dataSex: no dataNumber of animals: 6Duration of study: 7 days

Doses

Method : EEC Directive 83/467/EEC

Year

GLP : yes

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

Result : not irritating

**Remark**: Test substance applied under a semi-occlusive dressing. Single

application. Resulted in a 24-hr and 72-hr PI score of zero. Cyclohexane

had no corrosive properties

Reliability : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (35)

Species: rabbitStrain: no dataSex: no dataNumber of animals: 6

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5. Toxicity Id 110-82-7

Date 30.06.2008

**Duration of study** : 7 days

Doses

Method : EEC Directive 83/467/EEC

Year

GLP : yes

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

Result : not irritating

**Remark**: Test substance applied under a semi-occlusive dressing. Single

application. Resulted in a mean erythema score for 24-hr and 72-hr of

1.93.

A review of this study did however note that the erythematous reaction reached maximum severity at 5 days post-application (mean score 2.56). During this time, there was a gradual increase in dermal reaction for a further 144 h observation time (2.83). Overall, the irritation reactions were

important and still present at the end of the study.

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (24)

# 5.2.2 EYE IRRITATION

Species: rabbitStrain: no dataSex: no dataNumber of animals: 6Vehicle: none

Doses : Single undiluted Method : Draize Test

Year

GLP : ves

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

Result : slightly irritating

**Remark** : After application of the test substance to one eye of the rabbit the eye was

left unwashed.

At one hour post instillation, corneal opacity, involving up to 25% of the cornea, was noted in one rabbit and iritis was noted in another rabbit. Conjunctival redness was noted in five rabbits with conjunctival chemosis in one rabbit. All ocular lesions had cleared within 24 hours and no

conjunctival discharge was noted in any of the six animals.

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (36)

Species: rabbitStrain: no dataSex: no dataNumber of animals: 6Vehicle: none

Doses : Single undiluted Method : Draize Test

Year :

24 / 41

Date

GLP : yes

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

Result : slightly irritating

**Remark**: After application of the test substance to one eye of the rabbit the eye was

rinsed / washed.

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (37)

# 5.3 SENSITIZATION

**Type** : Dermal sensitization

Species : Guinea pig
Strain : no data
Sex : male / female
Number of animals : 9 male / 11 female
Vehicle : ethanol / acetone

Doses

Method : EC Annex V Method B6 - modified Buehler method

**Year** : 1996 **GLP** : yes

**Test conditions**: Twenty guinea pigs (9 males and 11 females) were induced dermally with

10% cyclohexane (purity 99.98%) in ethanol and challenged with 10% cyclohexane in acetone. Concurrent negative controls (no cyclohexane) and positive controls (DNCB-0.1% in 50% ethanol at induction and 0.07% in acetone at challenge) were tested. It should be noted that a maximisation test was not required because of the very poor tolerance to intra-dermal injection of solvents. During the induction phase, the response ranged from no redness (14/20 animals) to very faint redness on some tested animals (6/20 animals with a slight reaction). A very faint redness was observed 24 hours after the challenge application in 1/20 tested animals, no reactions were observed in other tested animals or negative controls. The incidence of sensitisation among cyclohexane induced and challenged animals was 0/20. The incidence of sensitisation among the DNCB induced and

challenged animals was 8/10.

A higher challenge concentration could have been chosen (15% in acetone did not produce any dermal irritation) and there were only a few animals with dermal reactions during the induction phase, these findings reduce the significance of this test.

**Test substance** : CAS No. 110-82-7; cyclohexane; (purity 99.98%)

Result : not a dermal sensitizer
Reliability : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

14.07.2008 (8)

#### 5.4 REPEATED DOSE TOXICITY

Type : Species : rat

Sex : male/female

5. Toxicity ld 110-82-7
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Number of animals : 5/sex/group Strain : other: Crl:CD.BR

Route of admin. : inhalation
Exposure period : 9 days
Frequency of treatm. : 6 hrs per day

**Doses** : 3000, 6000, 9000 ppm

Control group : yes

NOAEL : = 3000 ppm Method : EPA test method

**Year** : 1995 **GLP** : yes

Test Methods : In a two-week inhalation range finding study, Crl CD.BR rats (5/sex/group)

were exposed (whole body in an exposure chamber) to 0-3,000-6,000 and 9,000 ppm (0-10,500-21,000 and 31,500 mg/m3) of cyclohexane (purity 99.97%). Nine exposures, each lasting six hours, were performed in total. The animals were weighed before treatment, clinical signs were checked before, during and after exposure, and common biochemical parameters and histological examinations were conducted at the end of the study.

For neurotoxicity assessment, the animals were checked for alerting behaviour in response to a standardised auditory stimulus at least three times during each exposure. They were also submitted to an abbreviated Functional Observational Battery (FOB) before and after exposure on two separate days (test days 4 and 11). This assessment was also performed prior to the initiation of exposures to establish baseline measurements.

During the FOB, the following parameters were assessed:

- in home cage: posture and palpebral closure,
- in open field: righting reflex, convulsions, gait characteristics, vocalisations, labored breathing, coordination, arousal and palpebral closure.
- during manipulations: approach and touch response, auditory response (clicker) and tail pinch.

This study was performed according to EPA guidelines and following EPA and OECD GLPs.

Result

A slight but significant decrease in body weight gain was observed in males treated with 9,000 ppm. Except for a minimal increase in mitotic index figures detected in the hepatocytes of males at 6,000 ppm and higher and in females at 9,000 ppm, no other treatment related findings were observed for systemic toxicity. In particular, no modification in absolute and relative liver weights was noted in these studies. Based on these findings, a NO(A)EL of 3,000 ppm (10,320 mg/m3) can be assumed for systemic toxicity. For neurotoxic effects, diminished responses to stimulus were observed from day 2 at 9,000 ppm and from 7 exposures at 6,000 ppm. No effect was observed in FOB. A NOAEL of 3,000 ppm (10,320 mg/m3) can be assumed for neurotoxic effects in rats. This study served as a range-finding study for a 90-day inhalation toxicity study. It should be noted that this value is very conservative because the effects are very slight and may be of adaptive nature; this is taken into account in the risk characterisation.

**Test substance**: CAS No. 110-82-7; cyclohexane; (purity 99.97%)

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

17.07.2008 (17)

Type : Species : rat

Sex : male/female

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Number of animals : 20/sex/group (controls & hi dose); 10/sex/group (intermediary doses)

Strain : other: Crl:CD.BR

Route of admin. : inhalation Exposure period : 13 weeks

**Frequency of treatm.** : 6 hrs / day, 5 days / week (total 66 exposures)

**Doses** : 500, 2000, 7000 ppm

Control group : yes

NOAEL:

Method : EPA test method

**Year** : 1995 **GLP** : yes

Methods : In a 13-week repeated dose toxicity study conducted using CD BR rats

(20/sex/group for controls and high concentration and 10/sex/group for intermediary concentration groups) were exposed by whole body inhalation to cyclohexane at target concentrations of 0, 500, 2,000, and 7,000 ppm for 6 hours/day, 5 day/week for 13 weeks (66 exposures). Ten rats per month were allowed a one-month recovery period for control group and 7,000 ppm groups. After 45 and 90 days of exposure, blood and urine were collected for evaluation of clinical pathology parameters. Gross pathology, organ weight, macroscopic and microscopic examinations were performed

at the end of the study.

**Result**: No treatment-related effects were observed on body weight, body weight gain, food consumption, urine analysis and clinical examinations. A slight

decrease (not significant) in succinate dehydrogense and lactate dehydrogenase was observed in males and females at 7,000 and 2,000 ppm at both sampling times. In males exposed to 7,000 ppm a slight increase in adrenals weight was observed at the end of the recovery period. This finding was not observed at the end of the 90 day exposure so the relevance and significance is guestionable. In the 7,000 ppm group, a statistically significant increase in the relative liver weight with hepatic hypertrophy was observed in males (10/10), concurrent with an increase in the incidence of centrilobular hypertrophy in both sexes (9/10 males and 5/10 females). This finding was partially reversible in the one-month recovery period. For neurological effects, decreases in or absences of response to auditory stimulations were observed with a dose-response relationship from 500 ppm. In the 500 ppm group, there was a decrease in response on treatment days 61, 66, 67 and 68. In the 2,000 ppm group, there was decrease in the response during 16 exposures and no response during 50 exposures. In the 7,000 ppm group, a decreased response was observed in one exposure and no response was observed in the other 65 exposures. These effects were transient, and as no clinical observations of compromised neurological function were detected they were considered to be due to a reversible sedation caused by cyclohexane. The NOAEL for neurological effects was 500 ppm while the NOAEL for hepatic effects was 2,000 ppm. However, the partially reversible hepatic effects observed in males at 7,000 ppm were slight and may be considered of an adaptive

Remark : /

nature.

Additional groups of rats (12/sex/group) were treated in parallel with those of the main study in order to assess neurotoxicity of cyclohexane in FOB, motor activity and neuropathology tests (Haskell Laboratory 1996c). Neurobehavioral evaluations were conducted prior to exposure and at week 4, 8, and 13. During each evaluation period FOB was performed prior to the motor activity test. At the end of the study, 6 rats/sex/group were selected for neuropathology, the controls and 7,000 ppm tissues selected were examined, and the intermediate dose tissues were saved. Neurological lesions were also assessed by examining sections of the brain, spinal cord, sciatic nerve, gasserian ganglia, cervical and dorsal root fibers and ganglia, cervical and lumbar ventral root fibers and gastrocnemius muscle.

Similar to the main 90 day study, a sedative effect was observed at doses

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of 2,000 ppm and higher characterized by a decrease in the mean response to an alerting stimulus. This effect was transient since no effects were observed immediately after removal from the exposure chamber. No effects were observed during the FOB and motor activity assessment. Histologically, no treatment-related findings were observed, the only lesions observed being identical in character and severity to those observed in controls. These have already been described as occurring spontaneously in the rat. The NOAEL for neurotoxicity was 500 ppm based on the transient sedative effect observed at 2,000 ppm and higher.

**Test substance**: CAS No. 110-82-7; cyclohexane; (purity 99.98%)

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

17.07.2008 (18)(20)

Type

Species : mouse Sex : male/female

Number of animals : 20/sex/group (controls & hi dose); 10/sex/group (intermediary doses)

Strain : other: Crl:CD1 (ICR) BR

Route of admin. : inhalation Exposure period : 13 weeks

Frequency of treatm. : 6 hrs / day, 5 days / week (total 66 exposures)

**Doses** : 500, 2000, 7000 ppm

Control group : yes NOAEL :

**Method** : EPA test method

**Year** : 1996 **GLP** : yes

Methods : A 13 week inhalation toxicity study in mice was also performed following a

2-week range finding study in mice. The study was comparable in

experimental conditions to that performed on rats.

After a stimulus, the animals of the 500 ppm group reacted as controls. In the 2,000 ppm group, a decrease in or an absence of response was observed from the third exposure onwards, the effects appearing to get worst with time (more and more no-response with increasing numbers of exposures). In the 7,000 ppm group, there was an increase in the incidence of decreased response, absence of response and hyperactive state from test day 4 to test day 30. From test day 30 to the end of the study, the response to the stimulus was impossible to determine due to the hyperactive state of the animals. These symptoms were observed just after exposure but were reversible until the next exposure. The most frequently described symptoms were: abnormal gait or mobility, excessive grooming, hyperactivity, hyper reactivity, spasms, aggressivity, hypo-activity and ruffled fur.

In males, haematological abnormalities were observed from 500 ppm, these symptoms (increase in RBC - increase in Hb - increase in Ht and decrease in platelets) were not always statistically significant and not always dose-related. In females, increases in RBC, Hb and Ht were only observed at 7,000 ppm. Variation in the haematological parameters occured for all animals at the 7,000 ppm dose level, but could not be explained in relation to the lack of systemic symptoms of dehydration. They were considered to be of no toxicological importance.

An increase in absolute and relative liver weight was observed in males at 7,000 ppm (absolute: 1.504 g vs 1.275 g). Only the relative weight was increased in females. No concomitant histological findings were observed. These results were not in accordance with those found during the two-week range finding test (histological findings from 3,000 onwards in

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females and at 9,000 in males) but did not interfere with the determination

of a NOEL.

**Result**: For neurologic effects, a NOAEL of 500 ppm was determined based on

signs of sedation observed in animals at 2,000 ppm. The NOAEL for systemic effects was 2,000 ppm based on effects observed in the liver at

7,000 ppm (increase in absolute and relative liver weights).

**Test substance** : CAS No. 110-82-7; cyclohexane; (purity 99.98%)

Reliability : (1) valid without restrictions
Flag : Critical study for SIDS endpoint

17.07.2008 (19)(29)

# 5.5 GENETIC TOXICITY 'IN VITRO'

Type : Ames test

System of testing : Test species/strain: Salmonella typhimurium TA100, TA1535, TA98,

TA1537

**Test concentration**: 0 to 10000 nl/ml

Cytotoxic concentr. : 7800 µg/ml (10000 nl/ml)

**Metabolic activation**: with and without

Result : negative
Method : not data
Year : 1986
GLP : no data

**Test Method**: Salmonella typhimurium (TA1535, TA1537, TA98 and TA100) were

incubated with and without metabolic activation in DMSO. Two metabolic activation systems were used, one with SD rats livers and the other with Syrian hamster livers. Doses of cyclohexane ranged from 0 to 10,000

μg/plate.

Result : Signs of toxicity were noted in the 3,333 µg/plate for TA1537 and TA98 and

in the 10,000  $\mu$ g/plate for TA100 and TA1535.

There was no evidence of reverse mutation for any dose tested with and

without metabolic activation.

Genotoxic effects:

With metabolic activation: negative Without metabolic activation: negative CAS No. 110-82-7; cyclohexane

Test substance : CAS No. 110-82-7; cyclol

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

18.07.2008 (31)

Type : Mouse Lymphoma Assay

System of testing : L5178Y mouse lymphoma cells (TK locus) Test concentration : 313 to 10000 nl/ml (250 to 7800 μg/ml)

Cycotoxic concentr. : 7800 µg/ml (10000 nl/ml)

Metabolic activation : with and without

Result : negative

**Method** : Method equivalent to current guidelines

Year : 1986 GLP : no data

**Remark**: Cyclohexane, solubilised in desionised water, was tested at doses ranging

from 313 nl/ml to 10,000 nl/ml (250  $\mu$ g/ml to 7,800  $\mu$ g/ml) (API, 1986). The

method used was equivalent to the guidelines.

Date

Result

: Cytotoxicity was seen at 10,000 nl/ml (7,800 μg/ml) with and without metabolic activation. Without metabolic activation the percentage of relative growths ranged from 39 to 59 % without dose-effect relationship. There was no evidence of forward mutation at any dose. With metabolic activation, the percentage of relative growths ranged from 46 to 64% and was not dose related. There were very slight increases in mutant frequency in four treatments, but not dose related, and it was decided to confirm this result with another test.

Doses ranging from 3,000 to 8,000 nl/ml (2,340 to 6,240  $\mu$ g/ml) were tested in the second trial (cytotoxicity was found at 9,000 nl/ml (7,020  $\mu$ g/ml)). The percentage of relative growths ranged from 23 to 69%. In this trial the results were clearly negative. Overall, this test can be considered as negative with and without metabolic activation.

Test substance : CAS No. 110-82-7; cyclohexane Reliability : (1) valid without restriction Flag : Critical study for SIDS endpoint

18.07.2008 (2)(38)

Type : Sister Chromatid Exchange Test
System of testing : Chinese Hamster Ovary cells
Test concentration : 0.25 to 25 μg/ml in DMSO

Cycotoxic concentr. : no data

Metabolic activation : with and without Result : negative

**Method** : Method equivalent to current guidelines

Year : 1982 GLP : no data

**Remark**: Cyclohexane, solubilised in DMSO, was tested at doses ranging from 0.25

to 25 µgl/ml. The method used was equivalent to the guidelines.

**Result**: The higher dose tested corresponding to a complete growth inhibition of the

cell culture (25 µg/ml). No effect was seen within the range of the doses

tested and the test is considered negative.

**Test substance**: CAS No. 110-82-7; cyclohexane

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

18.07.2008 (39)

Type : Unscheduled DNA synthesis

**System of testing** : Human lymphocytes **Test concentration** : 0.1 to 10 mM cyclohexane

Cycotoxic concentr. : no data

**Metabolic activation**: with and without

Result : negative

**Method** : Method equivalent to current guidelines

Year : 1983 GLP : no data

**Remark**: Human lymphocytes ( + or – S9 mix) were cultured for 4 hours in the

presence or absence of cyclohexane. The effects on the DNA synthesis

were measured through cellular [3H]TdR uptake.

**Result**: Cyclohexane induced a marked inhibition of [<sup>3</sup>H]TdR uptake in the S9 mix-

lacking cultures while the corresponding cellular viabilities were unaffected. No effect was seen with metabolic activation. The effects seen without metabolic activation were not dose-dependent; solvent controls and negative controls were highly variable. Decrease of the uptake for the

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highest dose was within the values of the controls. No conclusion was

drawn from this study.

**Test substance**: CAS No. 110-82-7; cyclohexane; purity = 99.8%

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

18.07.2008 (40)

Type : DNA cell binding assay

System of testing : E. coli

**Test concentration**: 10 and 100 µM cyclohexane

Cycotoxic concentr. : no data

**Metabolic activation**: with and without

Result : negative
Method : no data
Year : 1981
GLP : no data

Remark : Cyclohexane was tested in a DNA cell binding assay at doses of 10 and

100  $\mu$ M. Cyclohexane was tested alone, mixed with lysozyme, mixed with liver extract and mixed with lysozyme and liver extract. Positive control was MMS and negative control was the culture middle only. The results are expressed as a "binding percentage". If this percentage is > 1%, the

substance was considered positive.

**Result**: Cyclohexane was found negative when tested alone in the groups treated

with liver extract and –with lysozyme and liver extract. A positive finding (1.6% only) was found in the group treated with cyclohexane + lysozyme at the highest dose (100  $\mu$ M). This result is considered doubtful because this is a very slight increase and also because this effects is not found in the

group - cyclohexane + lysozyme + liver extract.

**Test substance**: CAS No. 110-82-7; cyclohexane; purity = 99.8%

**Reliability** : (2) valid with restrictions

Original study not retrieved and reviewed. Study used for EU Risk

Assessment of cyclohexane.

Flag : Critical study for SIDS endpoint

18.07.2008 (26)

# 5.6 GENETIC TOXICITY 'IN VIVO'

Type : Cytogenetic Assay

Species : rat

Strain:Sprague DawleySex:male/femaleNumber of animals:10 per sexRoute of admin.:inhalation

Exposure period : 6 hours/day for 5 days

Doses : 0, 97, 307, 1042 ppm

Method : other: not specified

Year :

GLP : no data

Date

Remark

: Samples of bone marrow cells were taken for cytogenetic analysis 6 hours after completion of the final dose. A positive control, triethyleneamine, showed a significant increase in structural aberration frequency. For cyclohexane a small but statistically significant increase in numerical aberrations was recorded in low and medium dose females, and pooled data at the low dose groups of both sexes. There was no information on general toxicity; no decrease on mitotic index was seen at all the doses tested. However, the authors of the report concluded that the lack of a dose-related response indicated that these increases were not of biological importance. Moreover, the numerical aberrations parameter had often shown great variation in this laboratory, having no statistical significance even for positive controls (numerical data is not available). It can be considered that cyclohexane does not produce chromosomal aberrations under the conditions of this test.

Result : Negative result. Some increased aberrations at low and medium doses but

no dose-related effect.

**Test substance** : CAS No. 110-82-7; cyclohexane; purity is unknown

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

14.07.2008 (1)

# 5.7 CARCINOGENICITY

#### 5.8.1 TOXICITY TO FERTILITY

**Type** : Two generation reproductive study

Species : rat

Sex: male/femaleNumber of animals: 30/sex/groupStrain: other: CD BRRoute of admin.: inhalationExposure period: 90 days

Frequency of treatm. : 5 day/week and 6 h/day

Premating exposure period

Male : 10 weeks Female : 10 weeks

Duration of test :

No. Of generation : 2

studies

**Doses** : 0, 500, 2,000, or 7,000 ppm

Control group : yes

NOAEL parental

NOAEL F1 offspring

Method : other: US and OECD Test guideline

**Year** : 1997 **GLP** : yes

**Remark**: In this study, weanling F0 rats (30/sex/group) inhaled cyclohexane vapor at

0, 500, 2,000, or 7,000 ppm 5 day/week and 6 h/day. Exposure duration was 10 weeks before mating until sacrifice of the P1 generation and 11 weeks before mating until sacrifice of the F1 generation. Gravid females were not exposed from day 21 of gestation until day 4 of lactation. From day 5 of lactation until weaning the neonates were potentially exposed by maternal milk; no other exposure was administered. At post partum day 25, thirty F1 animals/sex/dose were chosen to produce the next generation, treatment was continued 11 weeks before mating and during gestation. Fertility parameters were calculated. From 500 ppm, there was an

increased incidence of diminished response to a stimulus during exposure,

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this finding being significant at 2,000 ppm and higher. At 7,000 ppm, major effects were observed on body weight, body weight gain and food efficiency. A decrease in mean body weight was seen with F1 male rats, P1 and F1 females during pre-mating, P1 females throughout gestation, and F1 females during lactation. A decrease in mean body weight gain was observed with F1 male rats and P1 and F1 females during pre-mating; however no reductions were seen for P1 females during destation. suggesting that the reduction in mean gestation body weight was probably due to pre-existing body weight deficits established during the pre-mating period. The same findings were also seen with the F1 generation. A decrease in mean food efficiency of P1 and F1 females during lactation and a decrease in food consumption of P1 females during lactation were also observed. Effects on reproduction were limited to a decrease in mean pup weight for both the F1 and F2 generations at the high dose which was significant between post partum day 7 and 25 during which time pups were fed only maternal milk indicating the effect is due to cyclohexane via lactation. There was a slight increase in the incidence of pro-static inflammation at 7,000 ppm in P1 and F1 adults, but this was considered incidental due to the lack of severity and the reported common occurrence in rats. There was a slight but significant decrease in the mean percentage of animals born alive in the F1 litters dosed with 7,000 ppm, but given that the value was still in the range of historical controls and that this effect was not dose-related, this was not considered biologically significant.

Result : The NOAEL for adult reproductive toxicity was 2,000 ppm based on

decreases in pup body weight observed at 7,000 ppm. The systemic NOAEL for this study was 500 ppm based on sedative effects observed at

2,000 ppm and higher.

Test substance : CAS No. 110-82-7; cyclohexane Reliability : (1) valid without restriction : Critical study for SIDS endpoint

18.07.2008 (23)

# 5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

**Type** : Developmental toxicity

Species: ratSex: femaleNumber of animals: 8/group

Strain : other: Crl:CD BR

Route of admin. : inhalation Exposure period : 9 days

Frequency of treatm.

**Duration of test** : 14 days

**Doses** : 0, 3000, 6,000, or 9,000 ppm

Control group : yes

Method : other: US and OECD Test guideline

**Year** : 1997 **GLP** : yes

**Remark**: As a pilot study, four groups of eight pregnant CDBR rats were exposed

whole-body to concentrations of 0, 3,000, 6,000, or 9,000 ppm cyclohexane from gestational day 7 to 16. Dams were sacrificed on day 22 and examined for gross pathologies; implantations and resorptions were counted and their relative positions recorded; fetuses were weighed and

examined externally for alterations.

**Result**: Maternal effects were limited to a reduction in overall maternal bodyweight

gain, overall food consumption and diminished response of animals to a

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sound stimulus during exposure to 6,000 ppm and higher. No effects were observed in the pups. The NOAEL was 3,000 ppm for the dams and 9,000 ppm for the pups.

Remark

This pilot study served as a range-finding study used to design a more complete study which was carried out during the 90 day inhalation study previously described (Haskell Laboratory, 1997b). Four groups of CD BR rats were exposed whole body to cyclohexane at concentrations of 0, 500, 2,000 or 7,000 ppm from gestational day 7 to 16. Animals were sacrificed on day 22 and examined. Findings were limited to the dams and included a diminished response of the animals to a sound stimulus while in the chamber during exposure and at 2,000 ppm or higher and reductions in overall body weight gain and food consumption throughout the treatment period. A slight but significant decrease in implantation number with the number of corpora lutea unchanged was seen when compared with controls. This finding was consistent with slight pre-implantation losses and can be considered as not treatment-related since there was no treatment during the pre-implantation period. The NOAEL was 500 ppm for maternal toxicity and 7,000 ppm for developmental toxicity considering the lack of toxic effects noted.

Test substance Reliability Flag 18.07.2008 : CAS No. 110-82-7; cyclohexane: (1) valid without restriction: Critical study for SIDS endpoint

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# 5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

# 5.9 SPECIFIC INVESTIGATIONS

# 5.10 EXPOSURE EXPERIENCE

# 5.11 ADDITIONAL REMARKS

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6.1 ANALYTICAL METHODS		
0.1 ARALI HOAL METHODS		
6.2 DETECTION AND IDENTIFICATION		
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7. Ef	f. Against Target Org. and Intended Uses	ld Date	110-82-7
7.1	FUNCTION		
7.2	EFFECTS ON ORGANISMS TO BE CONTROLLED		
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7.4	USER		
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# 8. Meas. Nec. to Prot. Man, Animals, Environment **Date** 30.06.2008 8.1 METHODS HANDLING AND STORING 8.2 FIRE GUIDANCE **EMERGENCY MEASURES** 8.3 POSSIB. OF RENDERING SUBST. HARMLESS 8.4 **WASTE MANAGEMENT SIDE-EFFECTS DETECTION** 8.6 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER 8.8 REACTIVITY TOWARDS CONTAINER MATERIAL

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(37)

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10.	END POINT SUMMARY				
10.2	2 HAZARD SUMMARY				
40.4					
10.3	RISK ASSESSMENT				
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