

Preliminary Evaluation of Remedial Alternatives Using a Reactive Transport Model: Cement Creek/Animas River

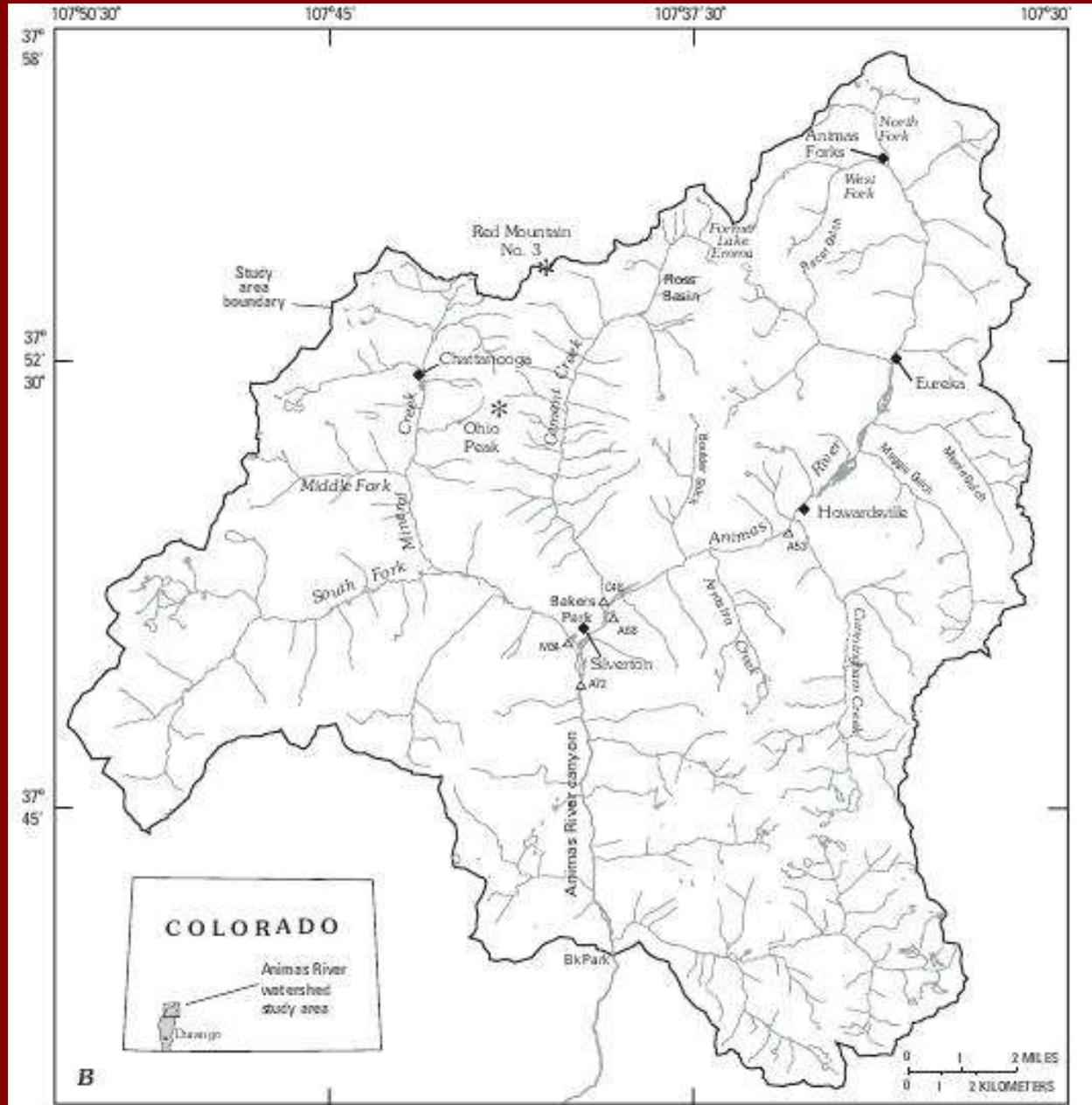


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Overview

- Part I: Study Objectives & Approach
- Part II: Background Info
 - Water Chemistry
 - Reactive Transport Model (OTEQ)
- Part III: Results
 - Concentrations and WQ Standards
 - Loads and Sources
 - OTEQ calibration and remedial scenarios
- Part IV: What next?

Part I: Study Objectives & Approach



Study Objective

- **Main Objective:**

Use a reactive transport model to evaluate remedial options for Cement Creek

- **Data Requirements:**

- Synoptic data set representing “steady-state” conditions
- Spatial profiles of streamflow and concentration

- **Existing data sets:**

- USGS (1996, 1999); diff. conditions: Gladstone treatment
- EPA/ARSG (recent); lacks spatial resolution

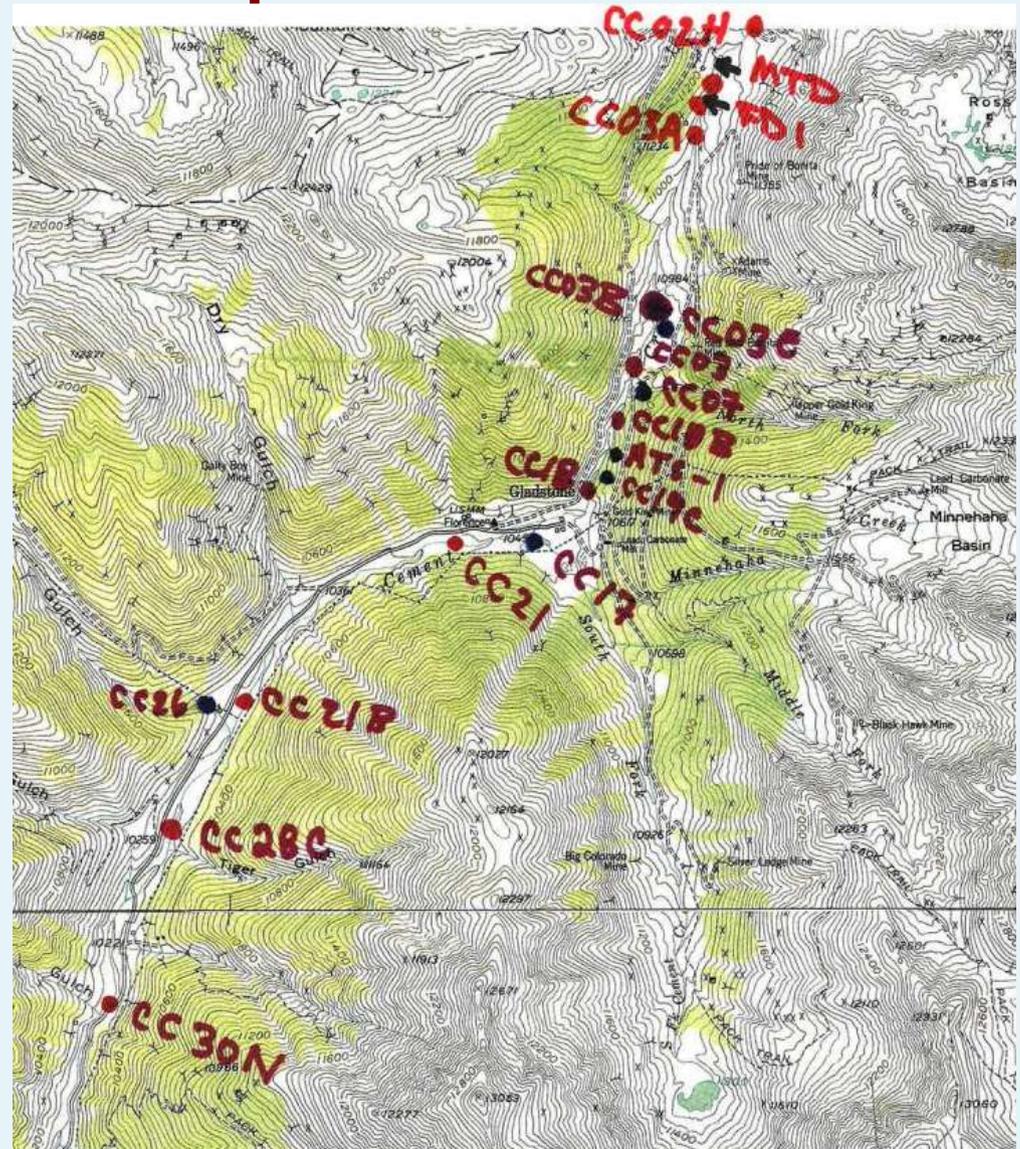
- **October 2012 Synoptic**

Approach

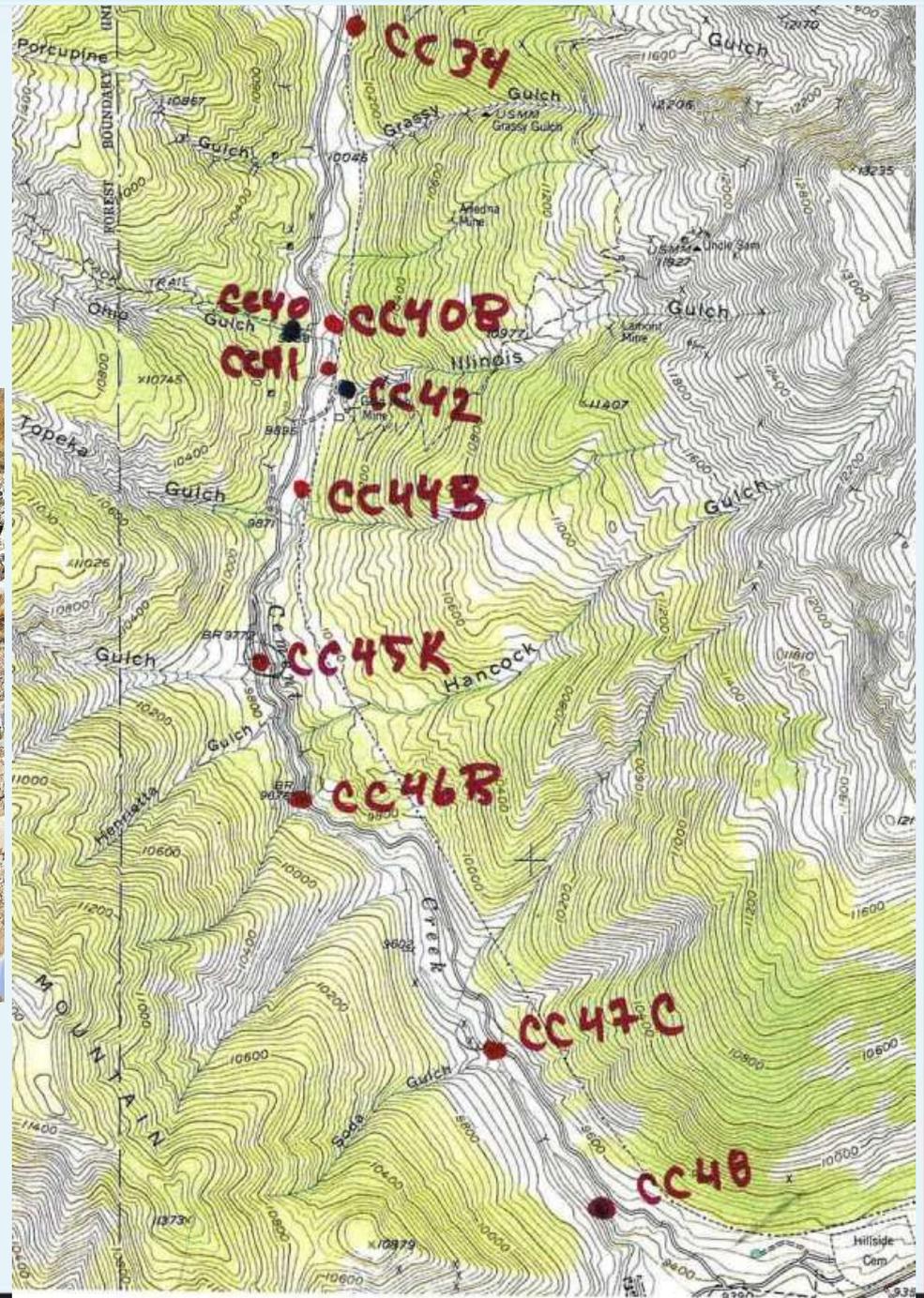
- Develop spatial profiles of streamflow & concentration
- Study Reach: Ross Basin to A72 (11.5 miles)
- Previous approach:
 - Tracer Injection to estimate streamflow
 - adv.: provides spatial detail, accurate flow
 - disadv.: time requirements → unlikely to capture “steady state”
- October 2012:
 - Acoustic Doppler Velocimeter (ADV) to estimate streamflow
 - 2 sampling teams & 3 ADVs
 - leapfrog sampling and day-to-day overlap (replication)

October 2012 Sample Locations

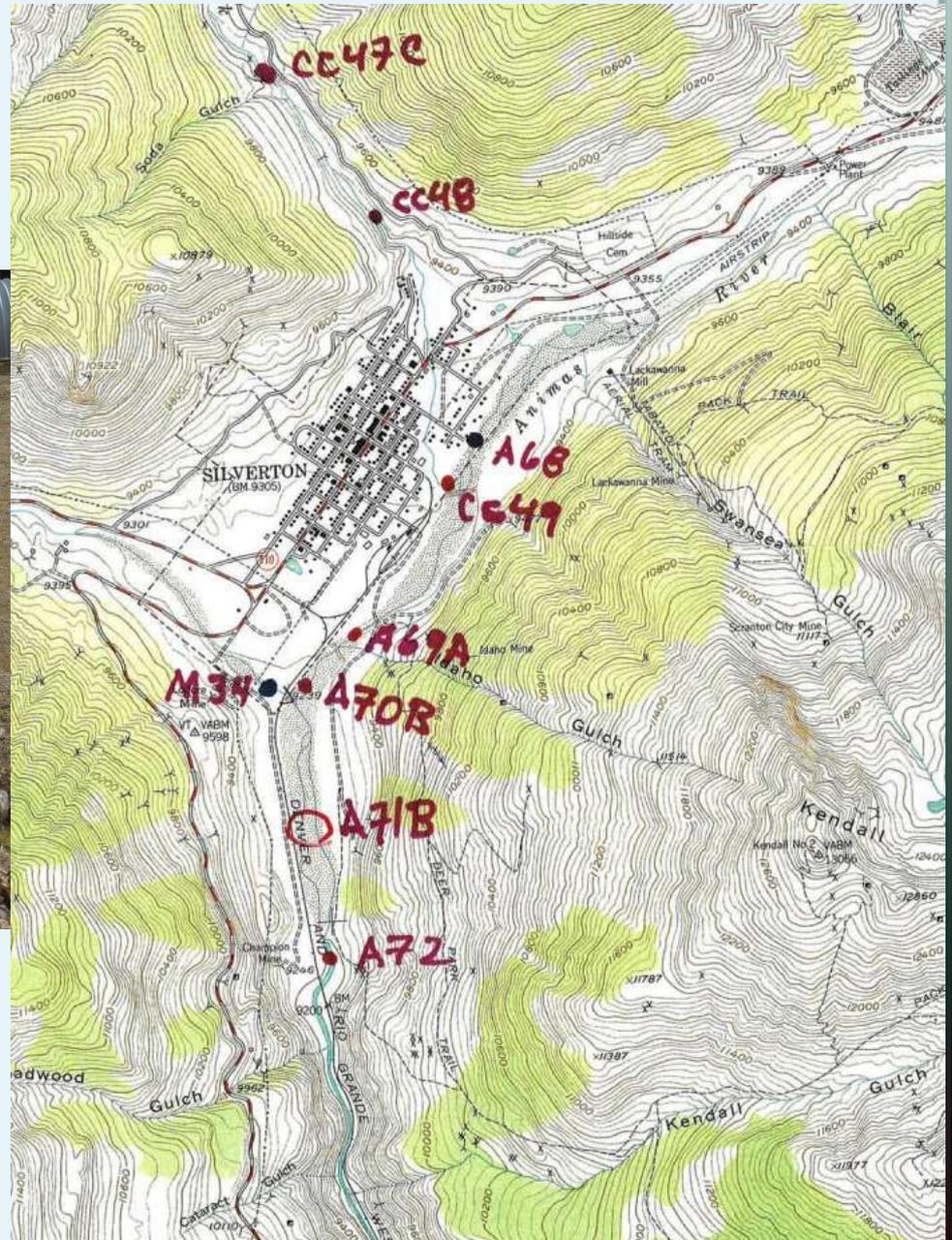
- Chemical “snapshot”
 - 30 stream sites
 - 15 inflows
- Sampled for:
 - pH, Alkalinity
 - cations (Fe, Al, Zn...)
 - anions (SO₄, Cl...)
- leapfrog sampling
- day-to-day replication
- flow from ADV



October 2012 Sample Locations



October 2012 Sample Locations



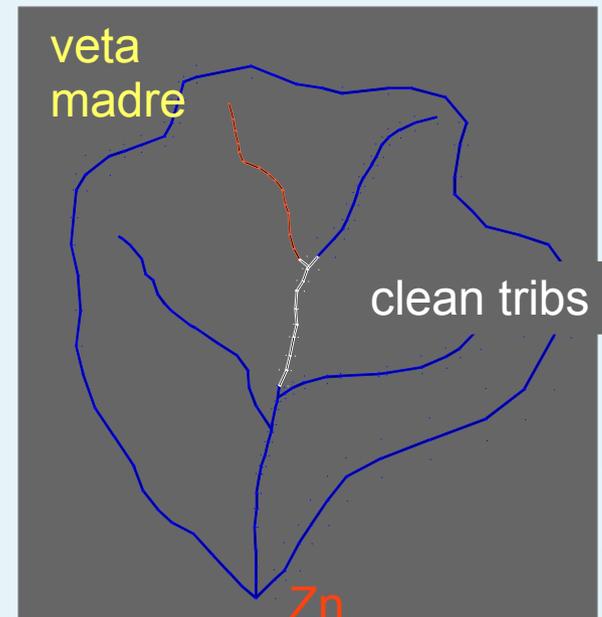
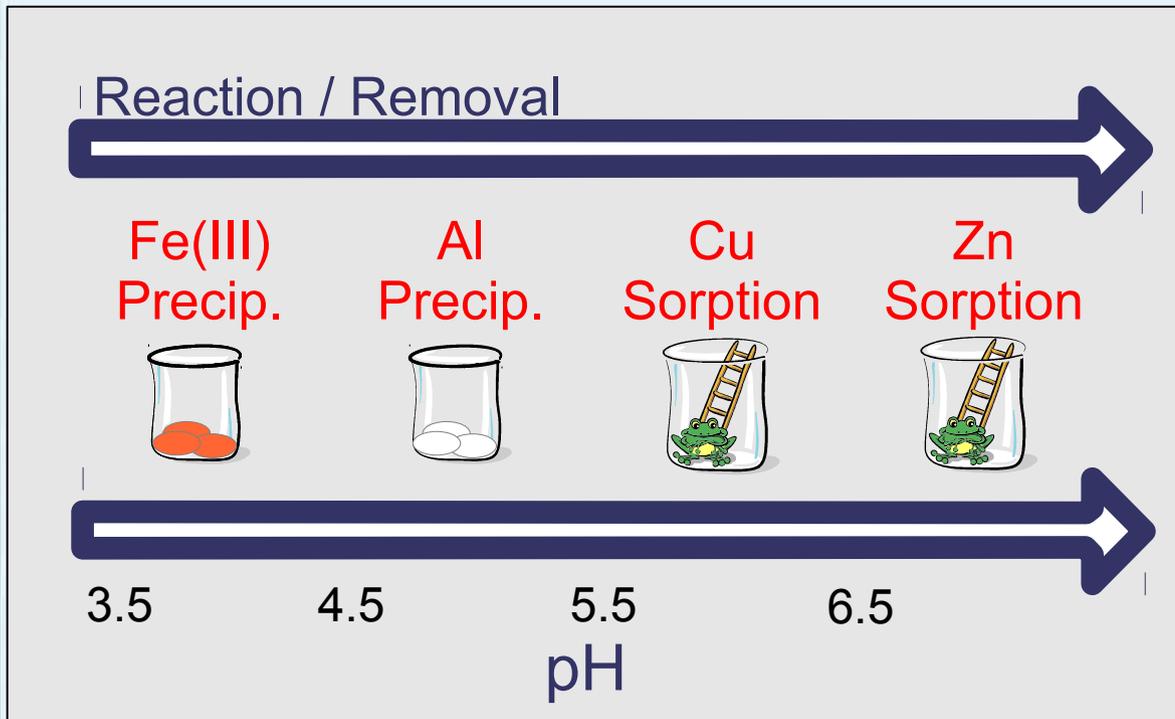
Part II: Background Info: Water Chemistry & Reactive Transport Model (OTEQ)



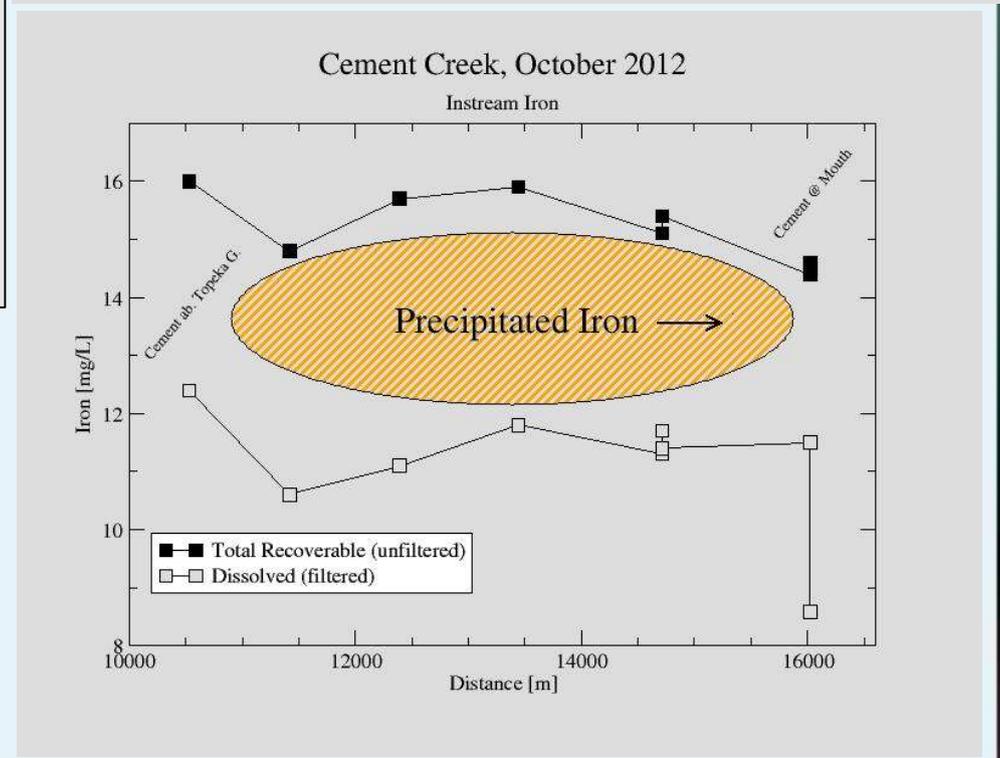
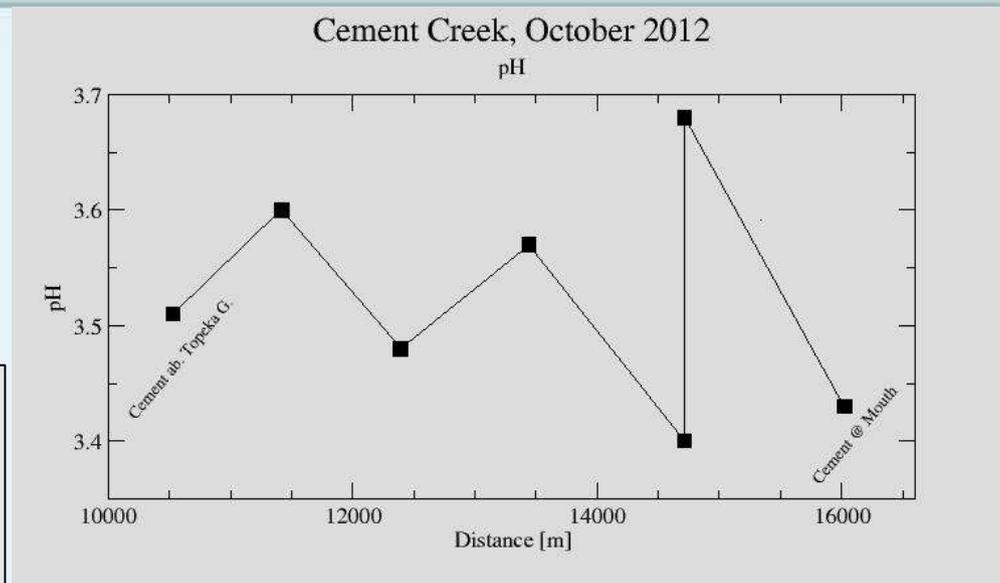
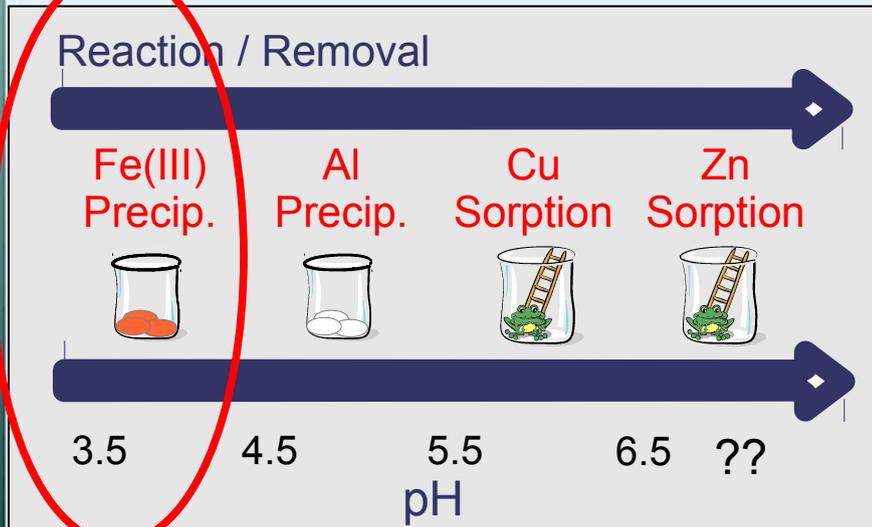
Background: Water Chemistry

- Mineralized Ore + (Water & Oxygen)
 - Sulfuric Acid → lowers pH → elevates metals
- “Total Recoverable”
 - unfiltered sample: dissolved + solids
 - solids: Fe and Al hydroxides precipitate
Cd, Cu, Zn, etc sorb onto Fe & Al
- “Dissolved”
 - filtered sample
 - pH-dependent solubility (low pH→high conc)

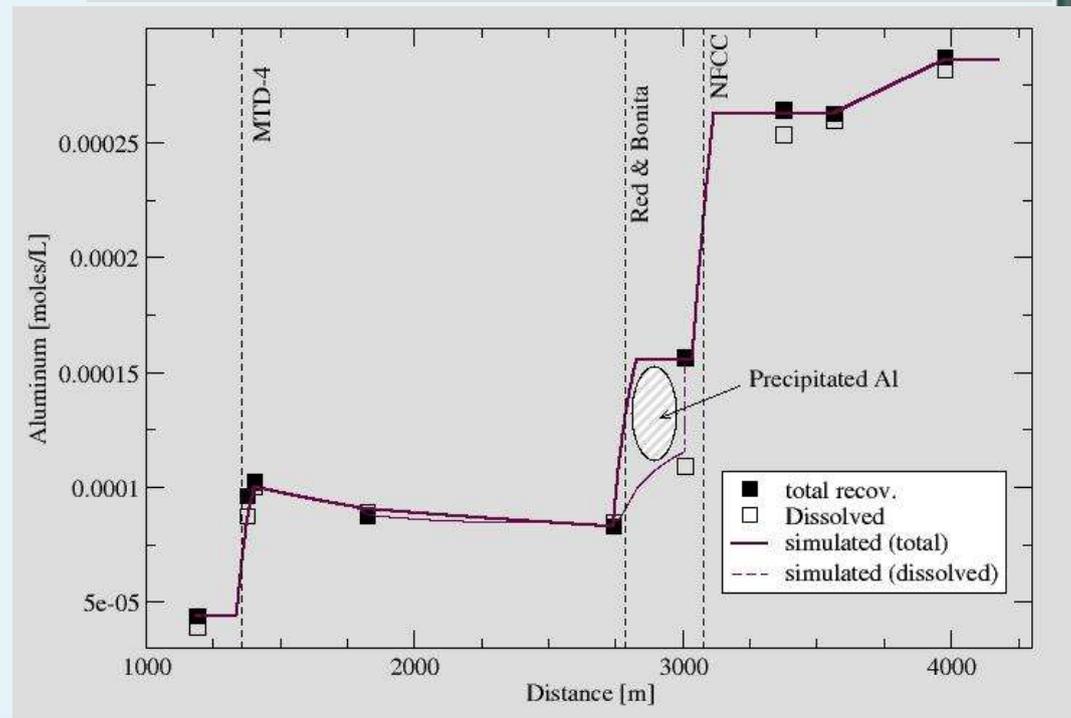
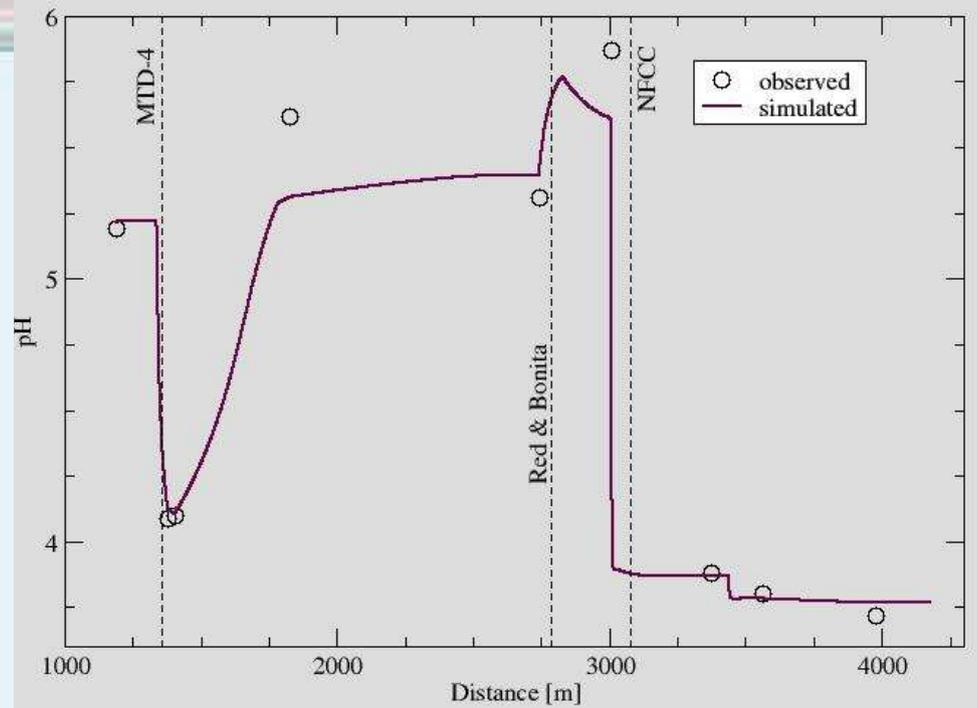
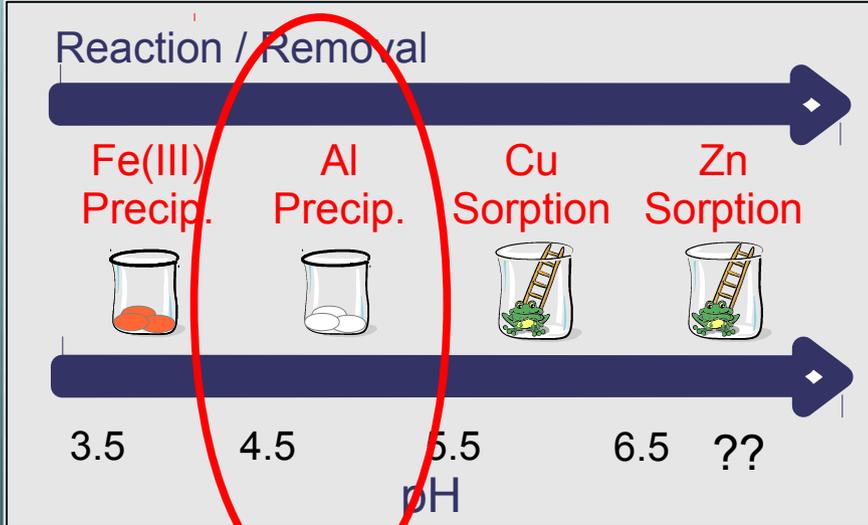
Reactive Transport



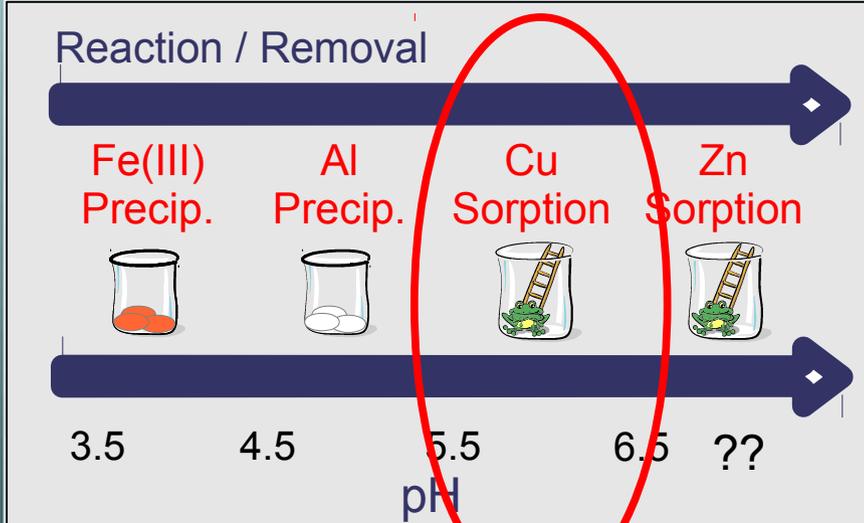
Reactive Transport



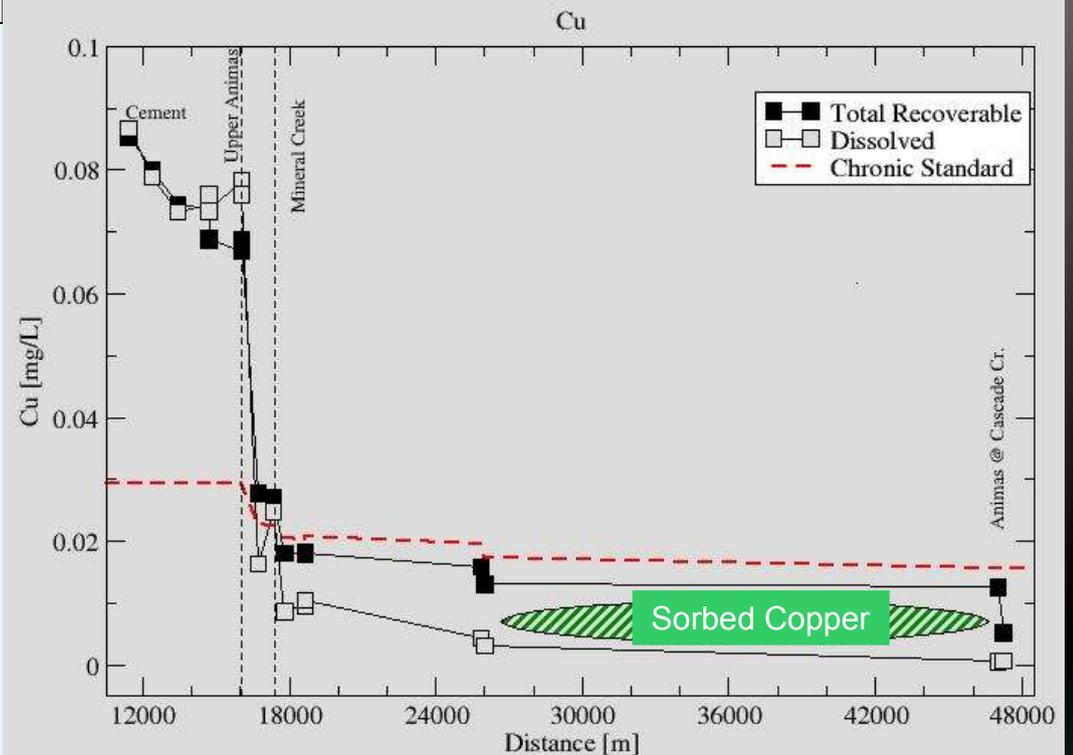
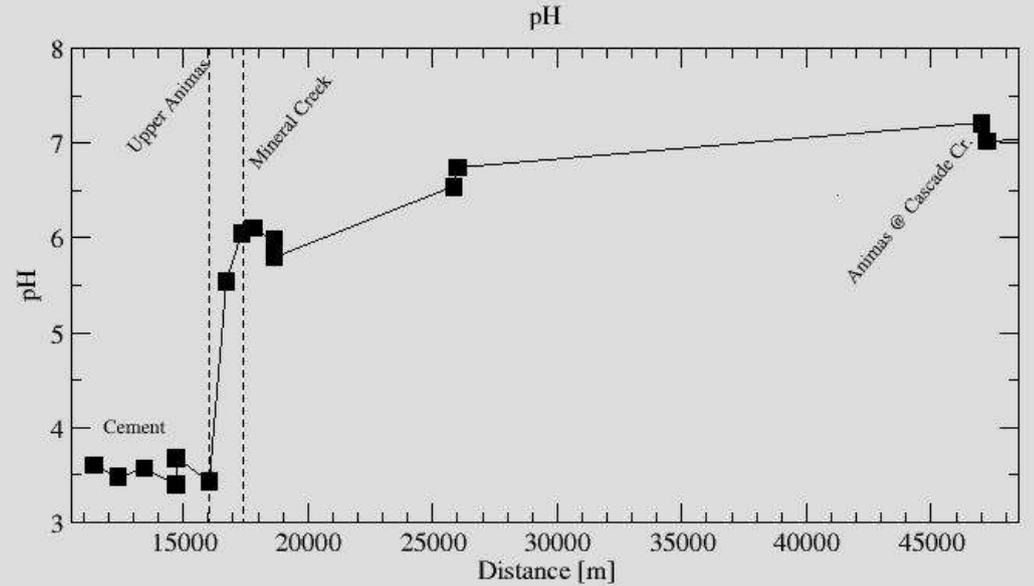
Reactive Transport



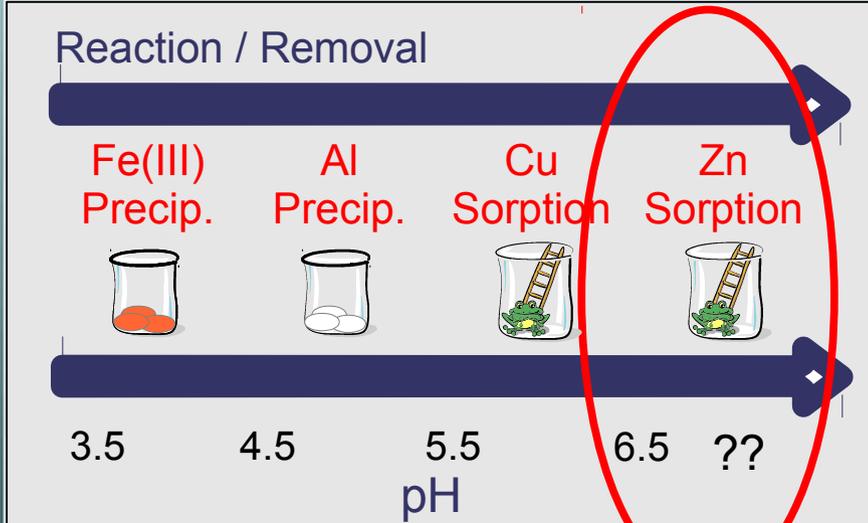
Reactive Transport



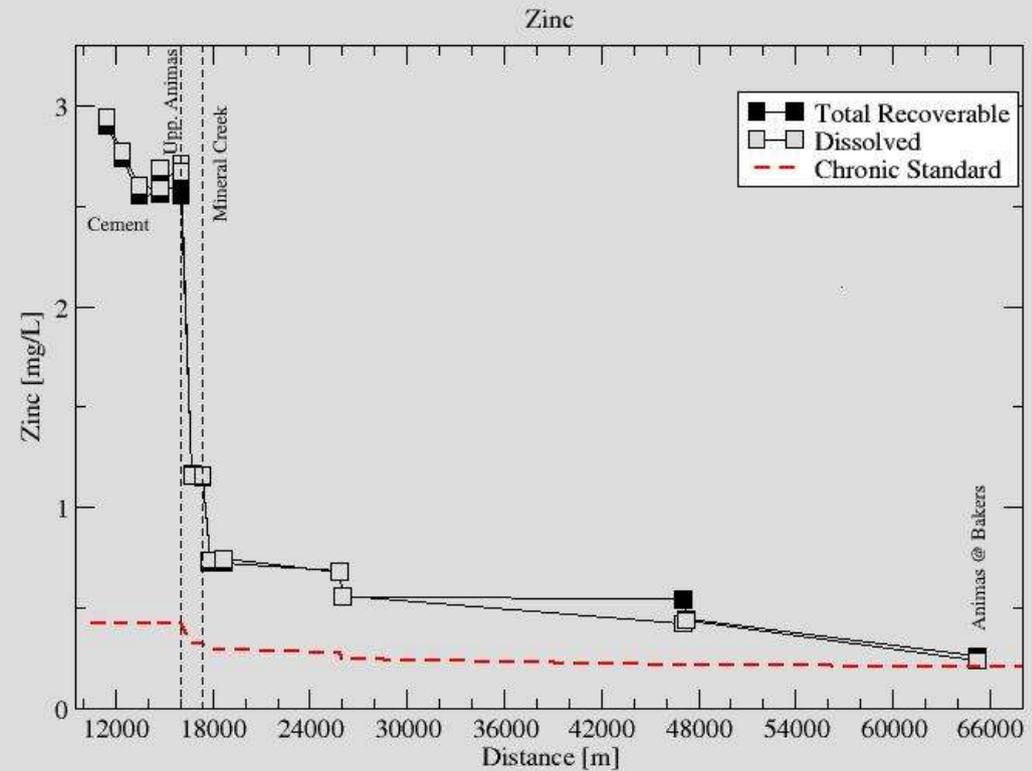
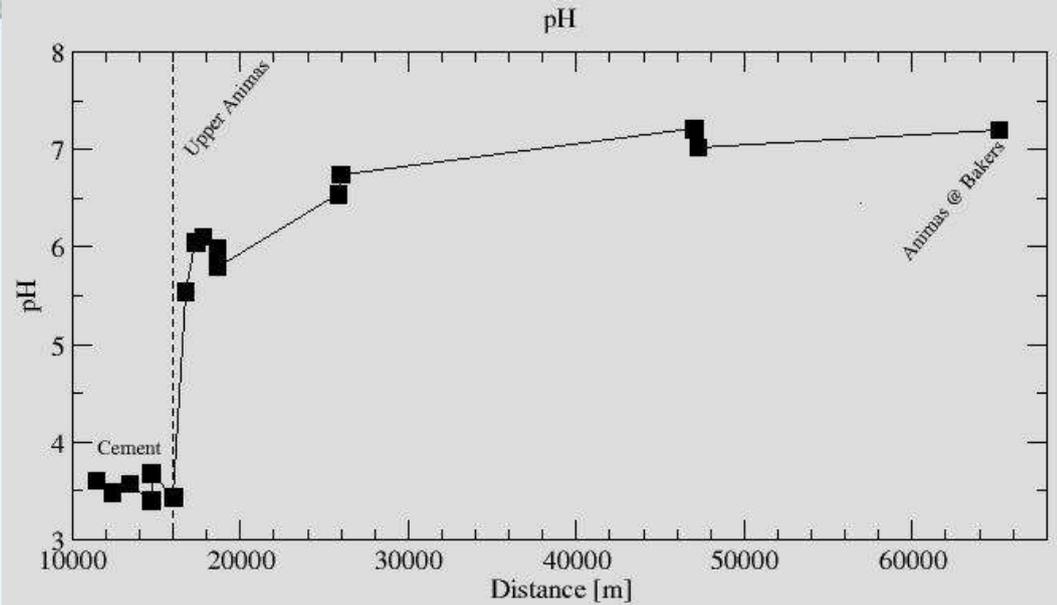
Cement Creek/Animas R., October 2012



Reactive Transport



Cement Creek/Animas R., October 2012



Reactive Transport



→ Quantify interplay between:

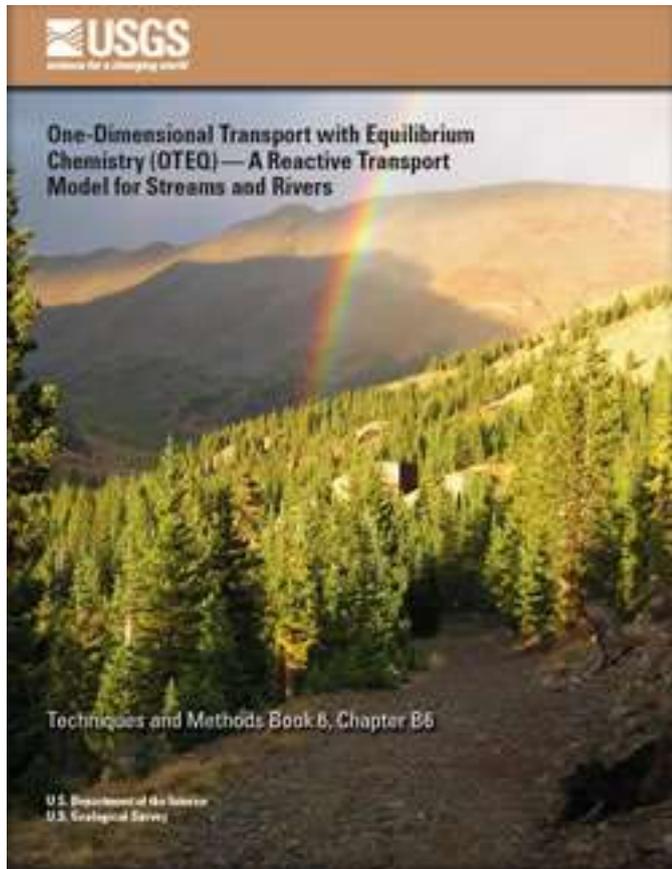
Hydrology: Advection, Dispersion, Storage, Inflow

Geochemistry: Precip./Dissolution, Sorption, pH

→ OTEQ: One-dimensional Transport w/ Equilibrium Chemistry

OTEQ:

One-dimensional Transport w/ **E**quilibrium Chemistry



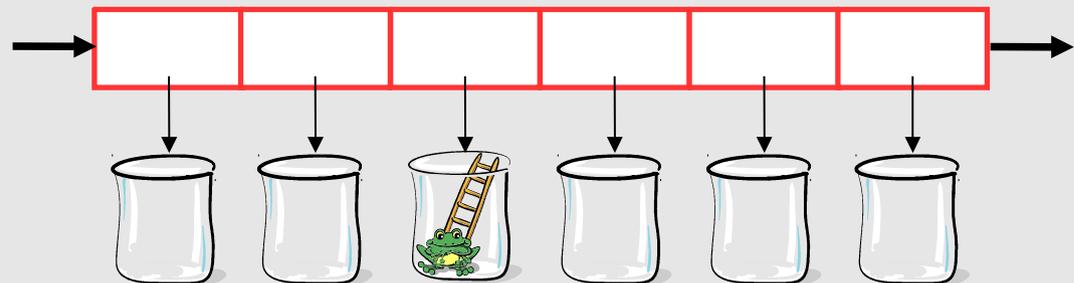
Transport (OTIS)

$$\frac{\partial C}{\partial t} = -\frac{Q\partial C}{A\partial X} + \frac{1}{A}\frac{\partial}{\partial X}(AD\frac{\partial C}{\partial X}) + \frac{q_{LIN}}{A}(C_L - C) + \alpha(C_S - C)$$

= Advection + Dispersion + Inflow + Storage

&

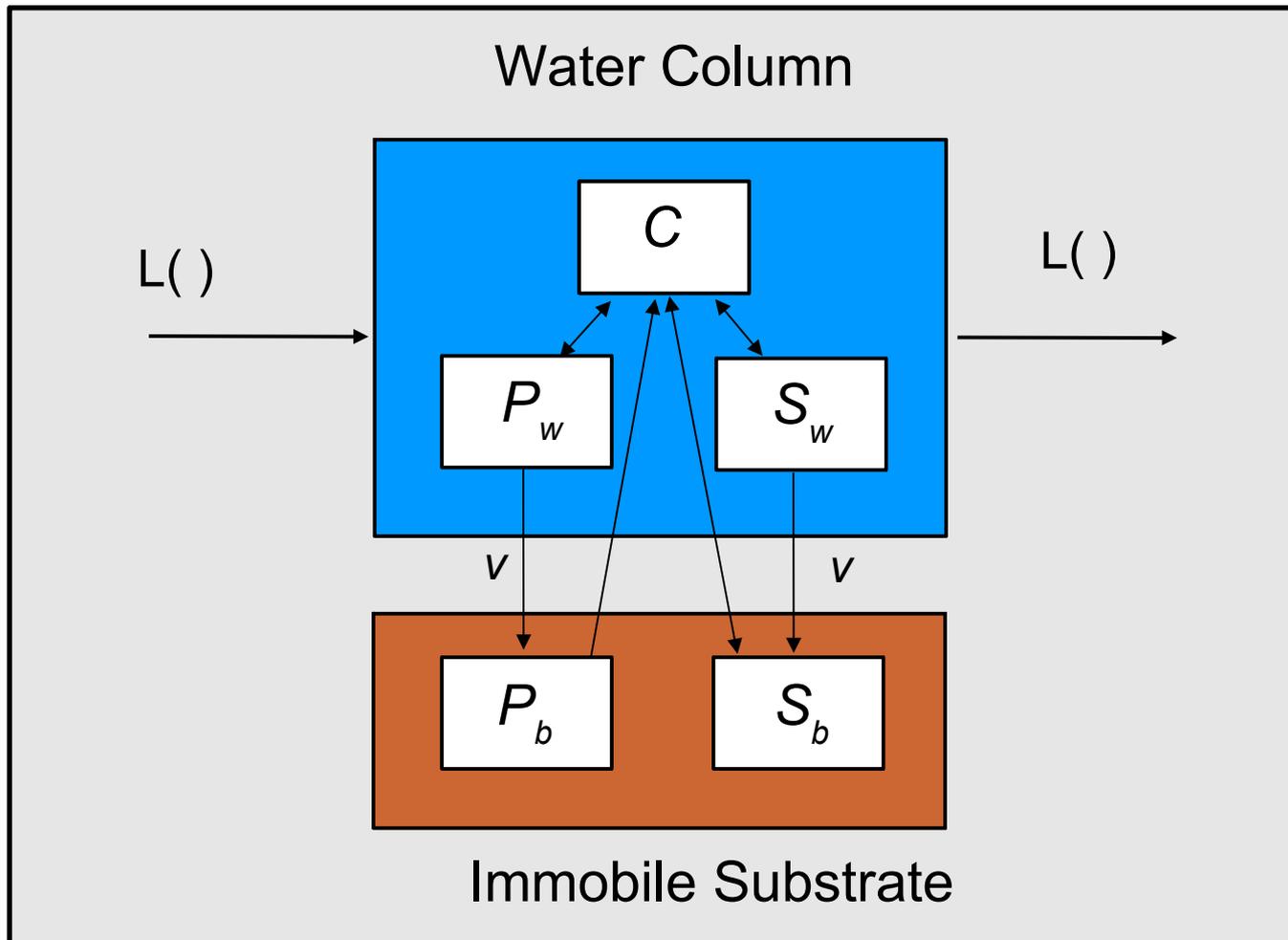
Equilibrium Chemistry (MINTEQ)



<http://water.usgs.gov/software/OTEQ>

Conceptual Surface Water System

$$T = C + P_w + P_b + S_w + S_b$$



Part III: Results



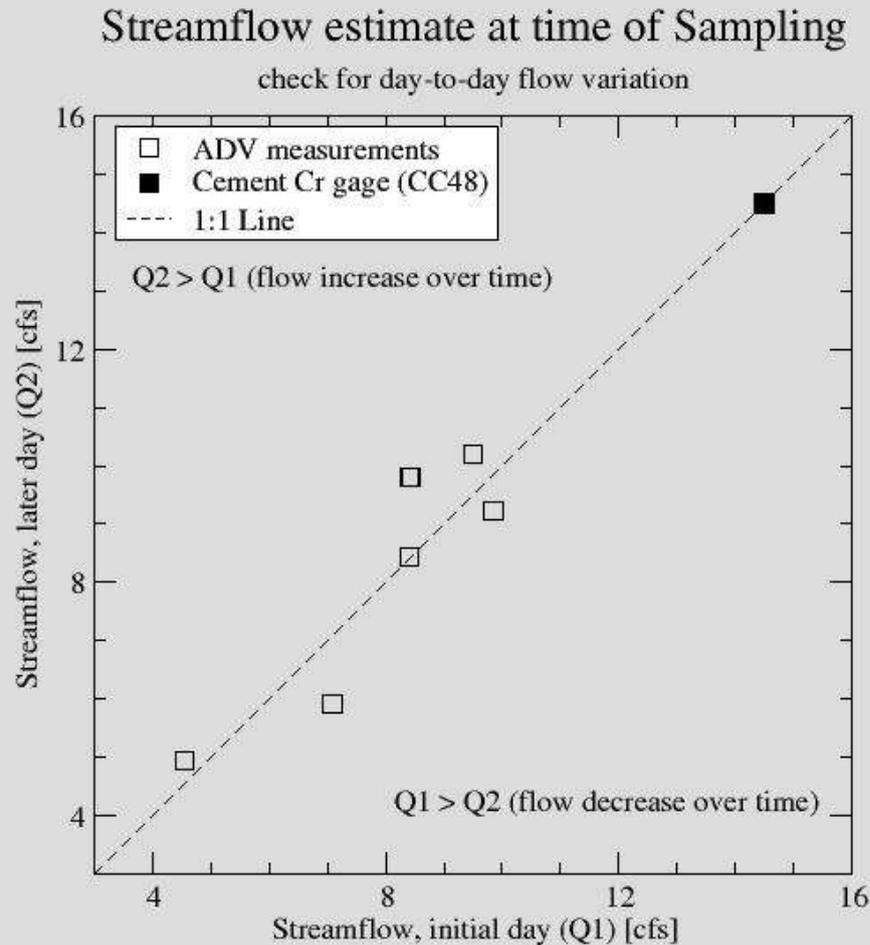
“It’s the largest outhouse in the
San Juan Mountains – an 8-holer”
– Bill Simon, August 2005

Part III: Data Set Quality Assurance

- Data “Backbone” – Spatial Streamflow Profile
 - Essential starting point for loading analysis & modeling
- Multiple means of estimating streamflow provide redundancy:
 - ADV (primary technique);
multiple measurements due to sample overlap
 - stream gages
 - flume measurements
 - slug injections
- Mass balance calculations to develop spatial profile of streamflow.

Do synoptic data represent “steady-state” conditions?

Loading analysis & modeling assume streamflow and concentration do not vary with time.



Constant Streamflow?

Cement Cr gage shows no change

ADV measurements on both sides of 1:1 line:

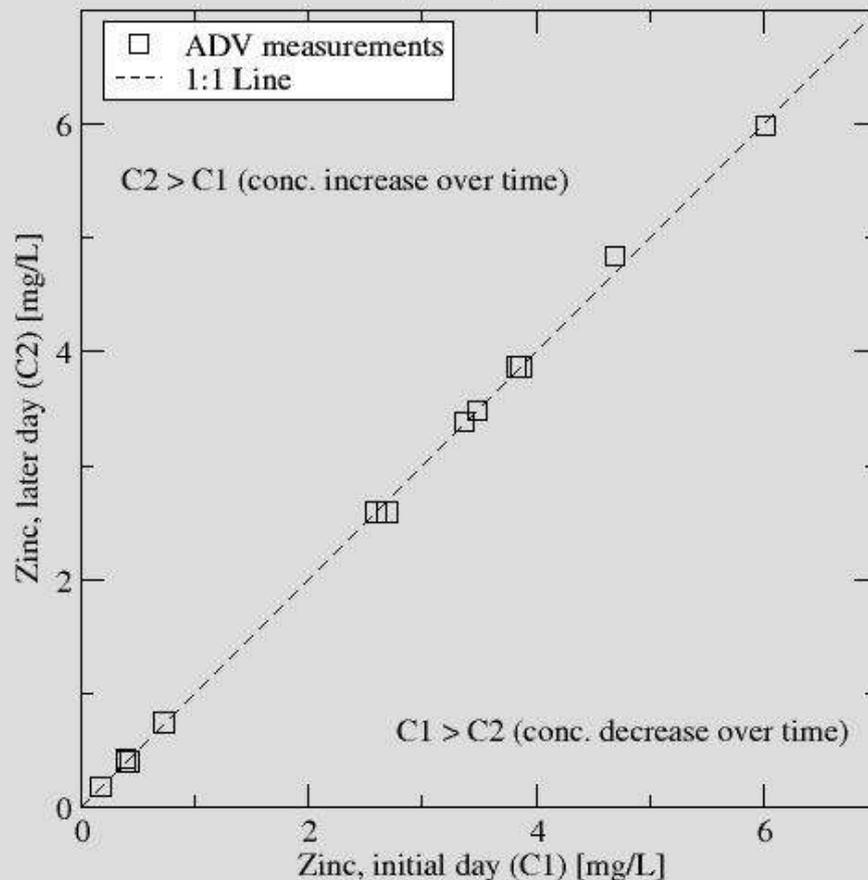
→ variability in ADV estimates > change in streamflow

Do synoptic data represent “steady-state” conditions?

Loading analysis & modeling assume streamflow and concentration do not vary with time.

Zn Concentration on different days

check for day-to-day conc. variation



Constant loading & rxn?

C1 & C2, good/great agreement:

Al, Ca, Cd, Co, Cu, F, Mg, Mn, Na, Sr, SO₄, Zn

more variable:

pH, Fe, K, Ni, Pb

→ data represent approximate steady state conditions

Mineral Cr: Al on 10/2 and 10/4 v. diff.

A68: diel variation of pH, Zn

Part III: Results

Concentrations and Water Quality Standards

Above std, entire study reach:

Al, Cd, Zn

Above std, top to Cascade Cr:

Fe

Above std, Mogul to Mineral:

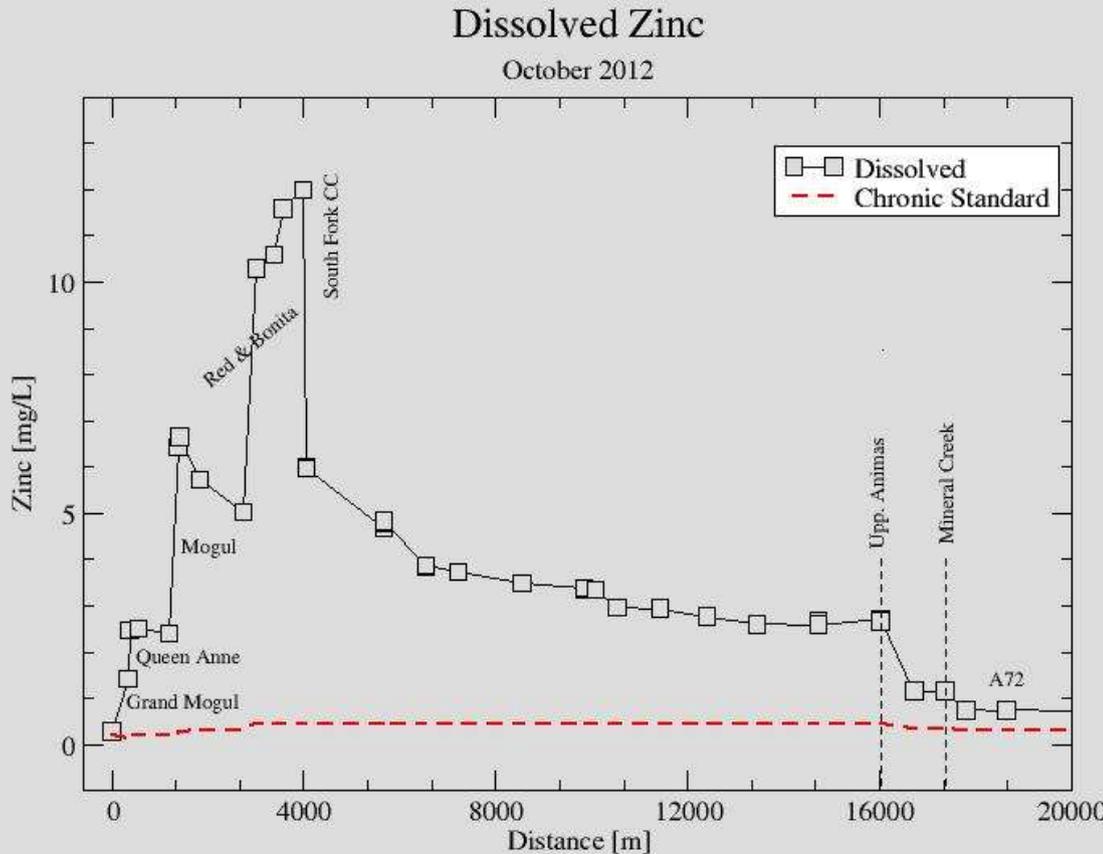
Mn

Above std, top to cement mouth:

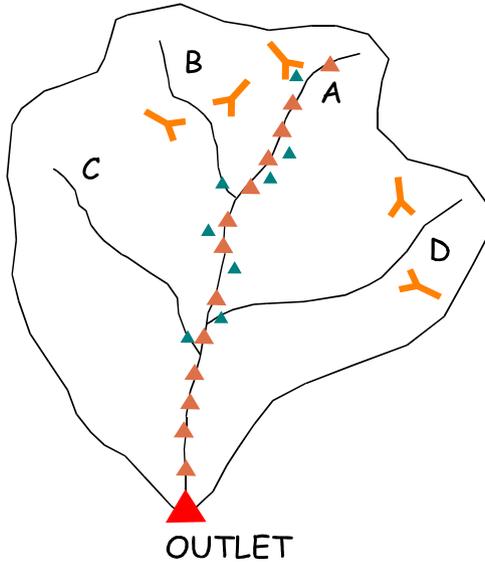
Cu

Above std, sub1 to cement mouth:

Pb



Part III: Results – Loads & Sources



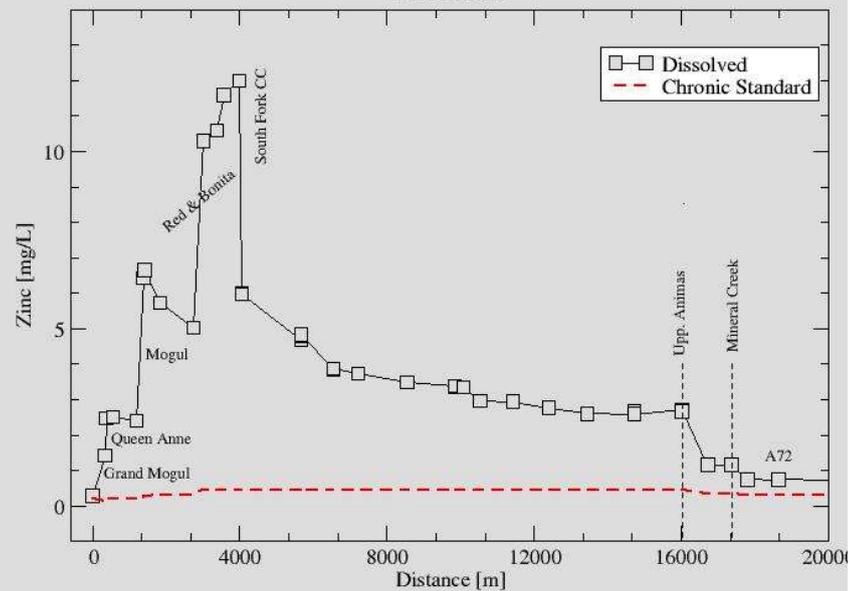
Synoptic Sampling,
October 2-4, 2012

- **Synoptic Study:**
Spatial profiles of Streamflow & Conc.
- **Load (mass / time) = flow * concentration**
- **→ Spatial profiles of mass load**
- **+ changes in mass load used to identify source areas**

Part IV: Results – Loads & Sources

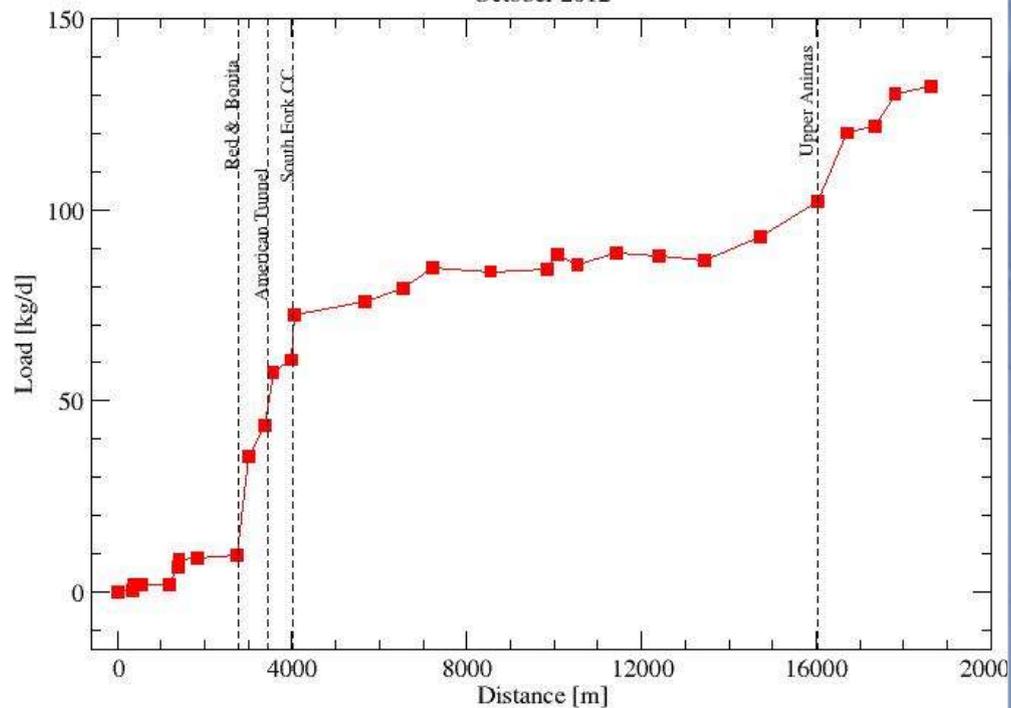
Dissolved Zinc

October 2012



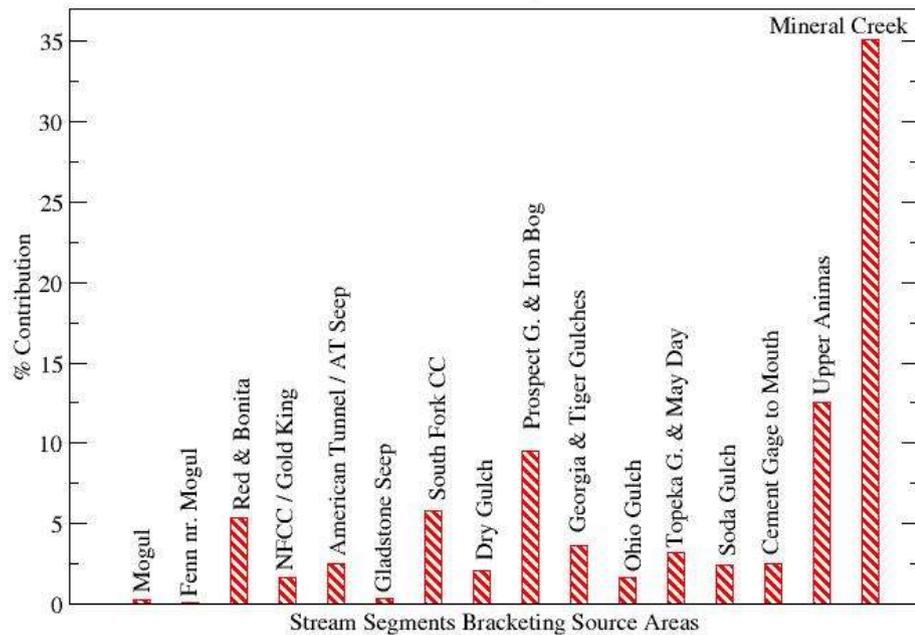
Zn Load

October 2012

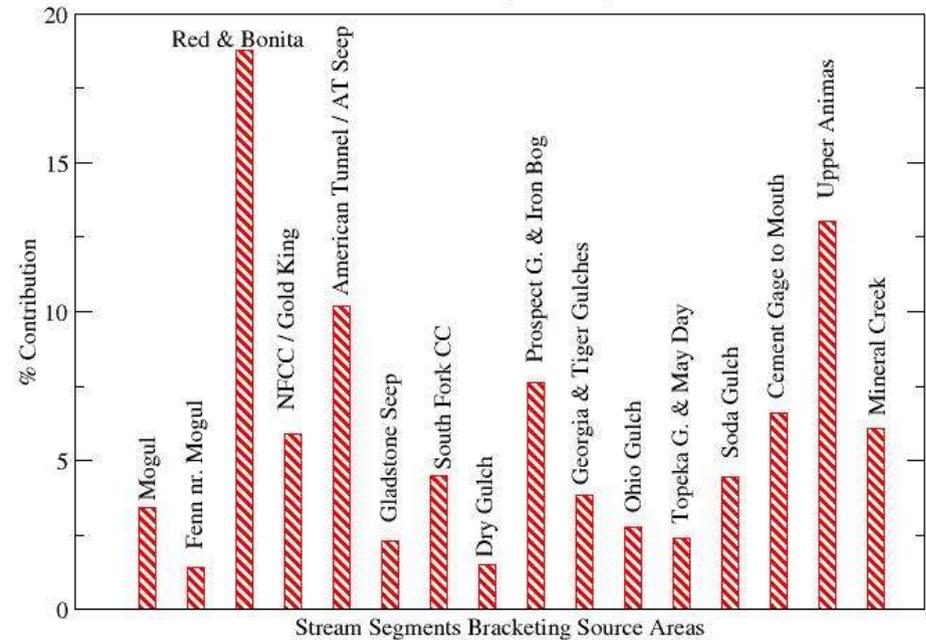


Part III: Results – Loads & Sources

Percent Contribution to SO4 load at A72
October 2013 (19 April 2013 Q)



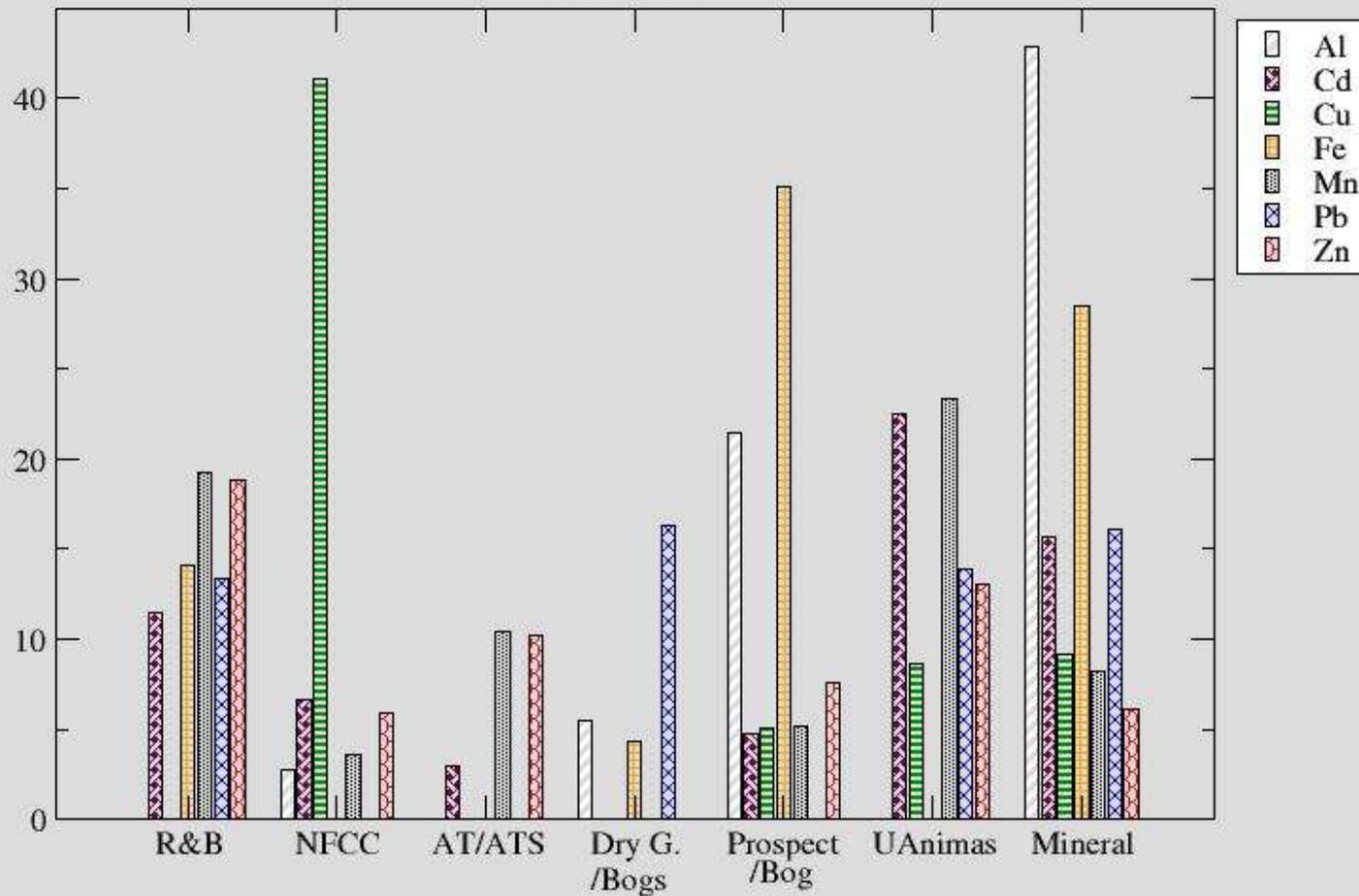
Percent Contribution to Zn load at A72
October 2012 (19Apr2013 Q profile)



Part III: Results – Loads & Sources

Percent Contribution to Metal Loads at A72

October 2012



Part III Results:

Reactive transport modeling

"Making predictions is difficult,
especially when it's about the future"

– Casey Stengel or Yogi Bera or some physicist dude

Study Objective

- **Main Objective:**

Use a reactive transport model to evaluate remedial options for Cement Creek

- **Data Requirements:**

- Synoptic data set representing “steady-state” conditions
- Spatial profiles of streamflow and concentration

- **Doing the 2-Step:**

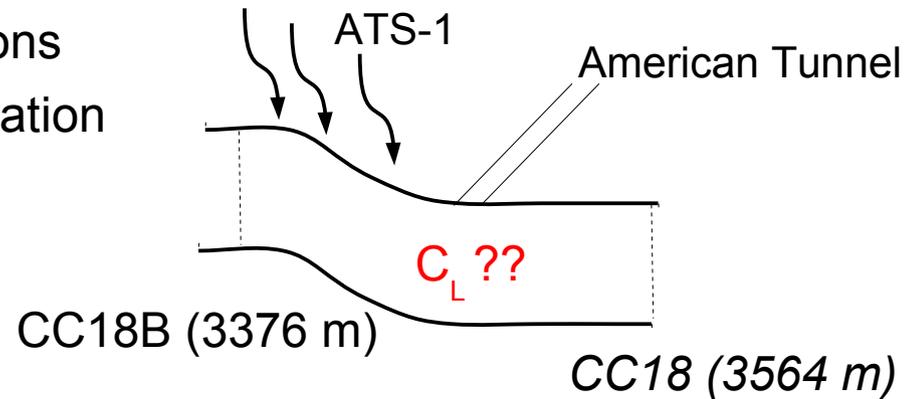
- Model Calibration: Reproduce existing conditions
- Prediction: Modify calibrated model to reflect remediation

Model Calibration: Reproduce existing conditions

- **Components:**
 - Al, As, Cd, Cu, Fe(II), Fe(III), H⁺, Ni, Pb, SO₄, Zn, Ca, CO₃
- **Reactions:**
 - precipitation: Fe(OH)₃, Al(OH)₃
 - sorption of As, Cd, Cu, Pb, and Zn onto HFO
 - oxidation of Fe(II)
 - degassing of CO₂
- **Model Input:**
 - spatial streamflow profile (1 flow / site based on QA)
 - inflow chemistry (synoptic sampling)
 - equilibrium constants (MINTEQ database)

Model Calibration: Reproduce existing conditions

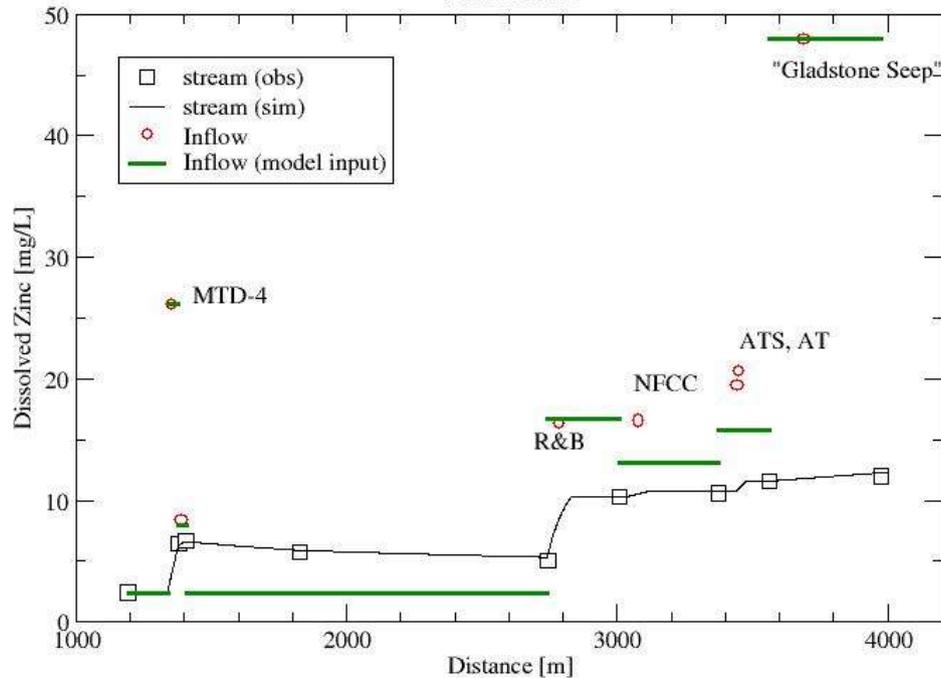
- Model input: inflow chemistry (synoptic sampling)
 - Reaches w/ multiple inflow observations
 - Reaches w/ a single inflow observation, but multiple inflows
 - Reaches w/o inflow observations
 - historical observations
 - no historical observation



Model Calibration: Inflow Chemistry (C_L)

Zinc Concentrations -- Stream & Inflow

October 2012



Set CL to reproduce stream data

Zinc – input values realistic

check other elements →
Fe CL > observed data, for example

When calibrating: Error on the side of caution

Possible outcomes of Predictions vs. Reality

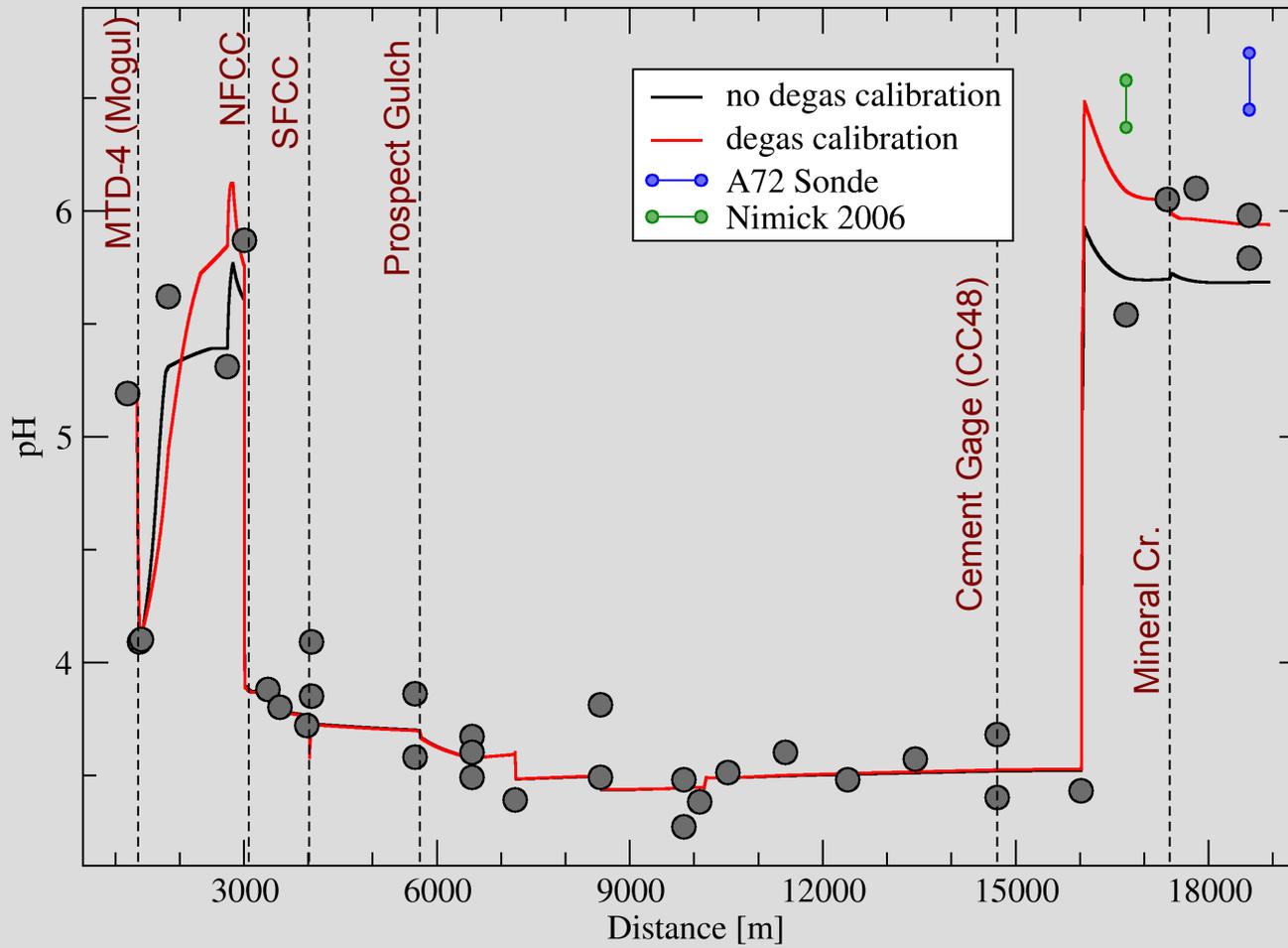
Table 1. Water Quality Standard Assessment (WQSA).

		Observed Data (MC2)	
		Attainment	Non-attainment
Best Estimate Simulation	Attainment	Correct Assessment	Type I Error: False Positive
	Non-attainment	Type II Error: False Negative	Correct Assessment

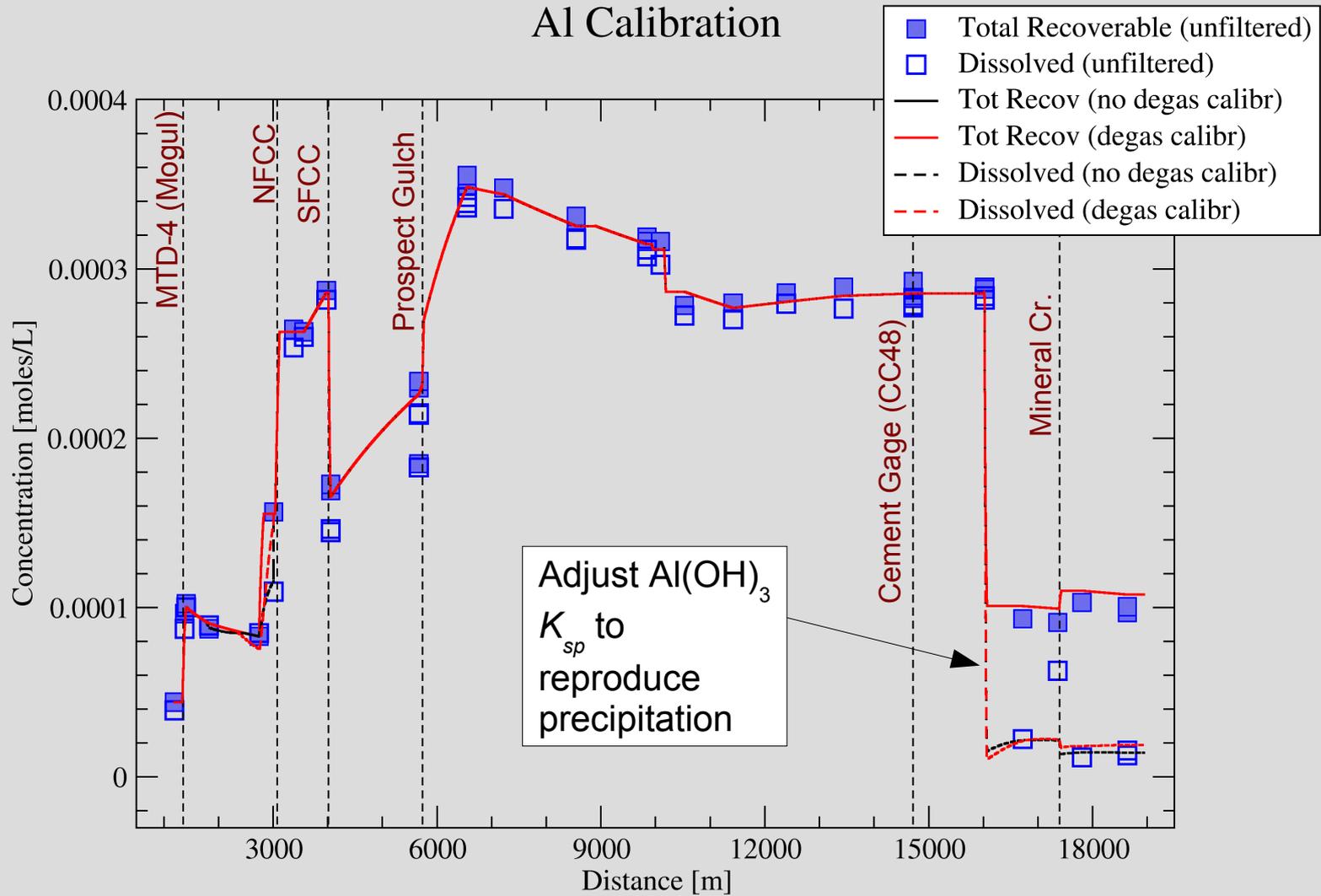
Runkel et al., *Envir. Sci. & Tech.*, 2012

To avoid Type I errors:
Overestimate concentrations/Underestimate attenuation

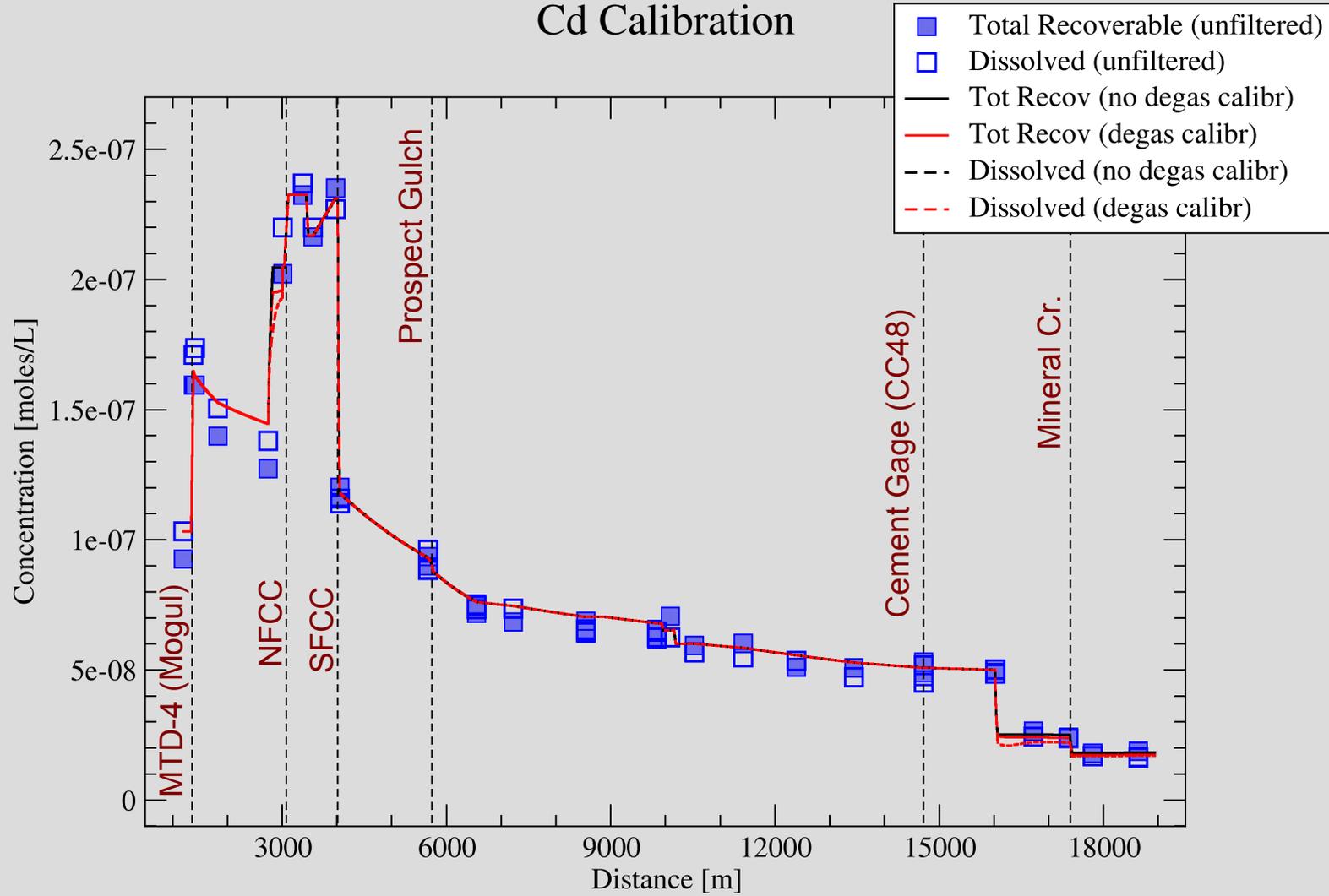
pH Calibration



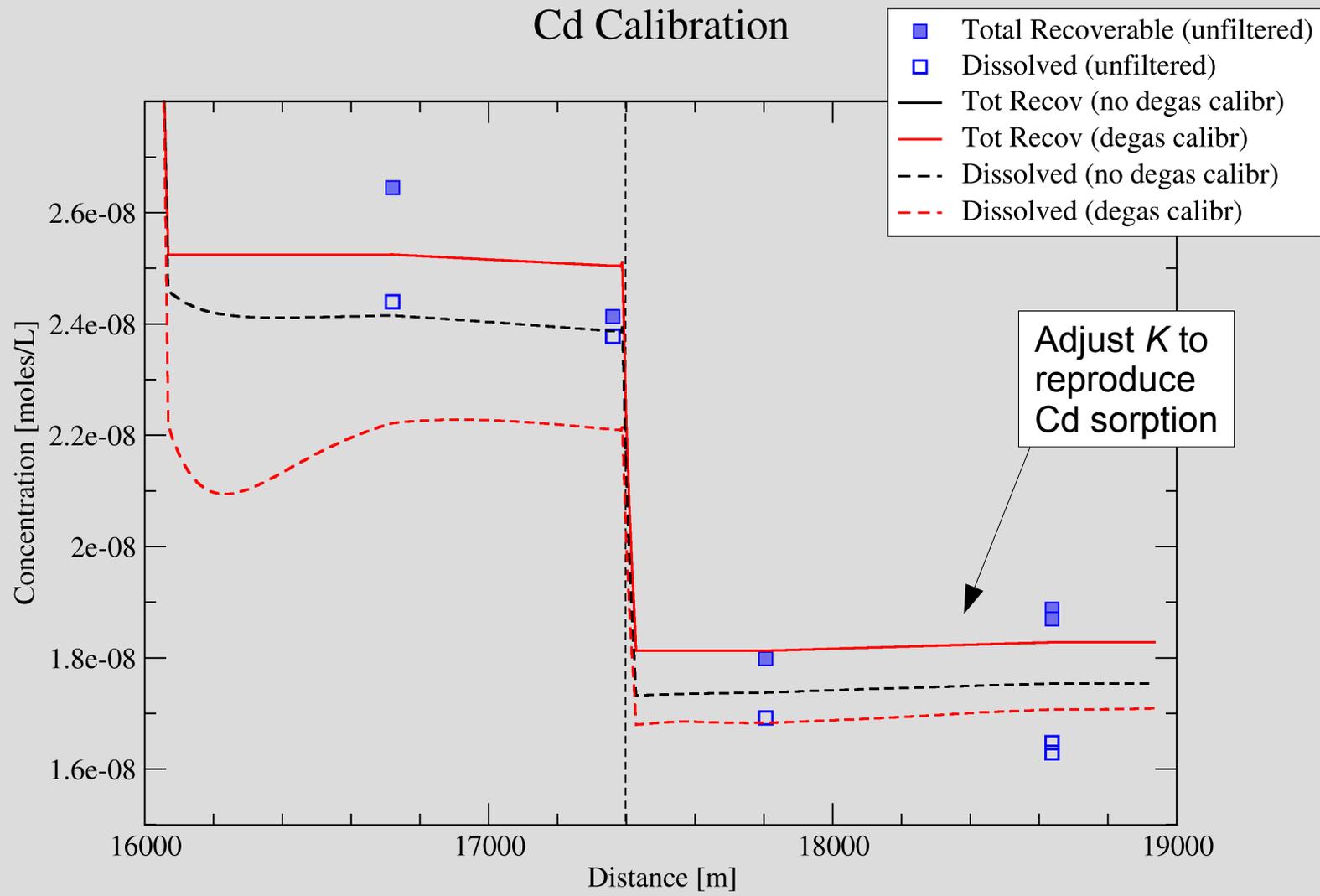
Al Calibration



Cd Calibration



Cd Calibration



Predicting Effects of Remediation

- Modify calibrated model to reflect proposed remedial action
- Remedial Scenarios:
 - Plug Red & Bonita Adit (100% reduction)
 - 1996 Treatment Plant Conditions

Predicting Effects of Remediation

Role of Uncertainty

- Estimates of post-remediation water quality are uncertain
- Calibration (low pH) → Estimation (increased pH)
- Sources of Uncertainty:
 - form of precipitates and solubility products
 - surface complexation (sorption) constants
 - uncalibrated rxns (e.g. Zn sorption)
 - effect of degassing
 - variation in low flow water quality

Role of Uncertainty

Factor #1: Equilibrium Constants

- Solubility Products for Precipitation of Al & Fe
 - known to vary over a wide range
 - default K_{sp} values modified during calibration
- Surface complexation constants for sorption of As, Cd, Cu, Ni, Pb, Zn
 - defaults based on best estimates of Dzombak & Morel
 - default K values modified during calibration (As, Cd, Cu)
- Dealing w/ Uncertainty:
 - consider predictive simulations w/ both calibrated values and defaults

Role of Uncertainty

Factor #2: Gas Exchange

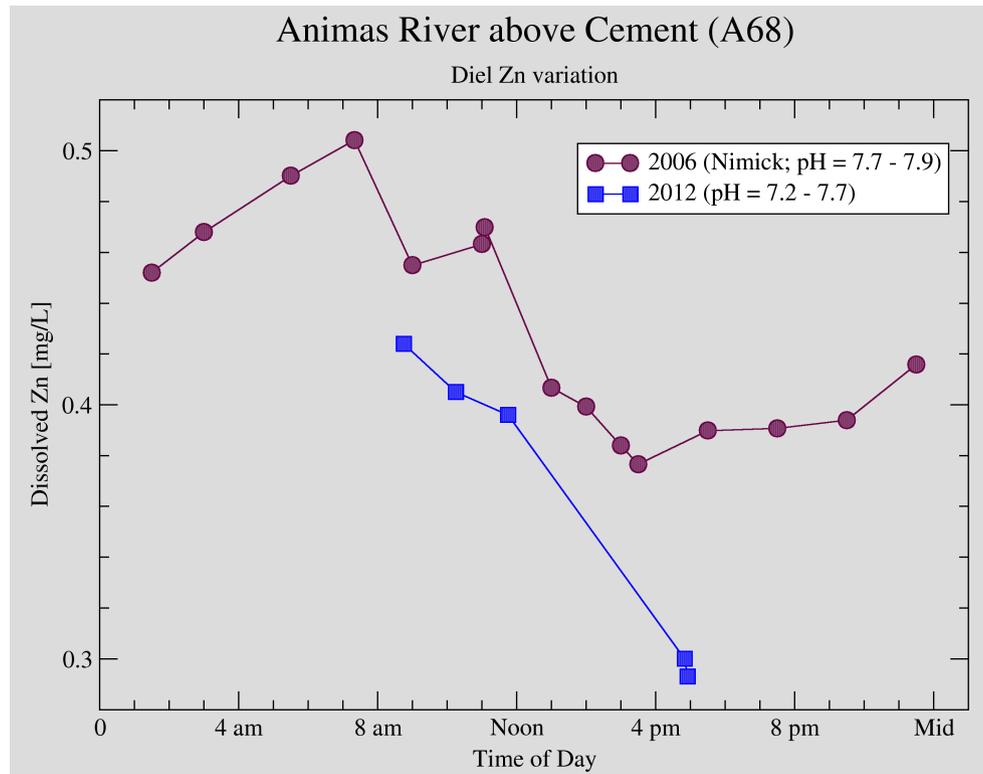
- Calibrated model w/o degassing errs on the side of caution (under-predicts pH)
- Calibrated model w/ degassing reproduces observed data
- Dealing w/ Uncertainty:
 - consider predictive simulations w/ and w/o degassing

Role of Uncertainty

Factor #3: Variation in low-flow water quality

Inflow concentrations are set during calibration

Inflow concentrations are known to vary:



Dealing w/ Uncertainty:

Conduct “worst case” simulations where Mineral Creek and Upper Animas inflow concentrations are set equal to the max concentrations (2006/2012)

Dealing w/ Uncertainty

- Sensitivity Analysis
 - Factors
 - Equilibrium constants: calibrated vs. defaults
 - Gas Exchange: with vs. without
 - Inflow concentrations: calibrated vs. worst case
 - Factorial design: $N_{\text{levels}}^{M_{\text{factors}}} = 2^3 = 8$ simulations
- 8 simulations for each remedial scenario
 - provides range of possible outcomes
 - “best estimate” = calibrated degas model
 - compare results based on median

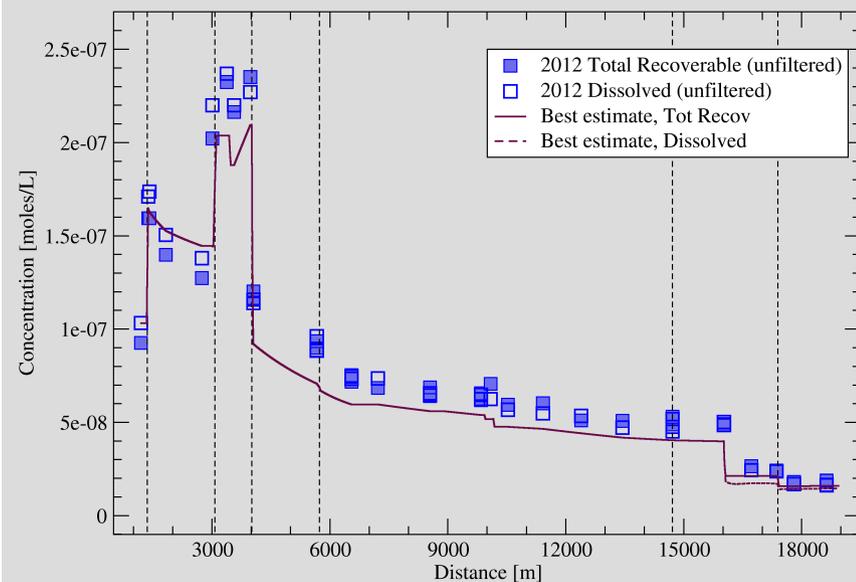
Remedial Scenario #1: 100% removal of R&B load



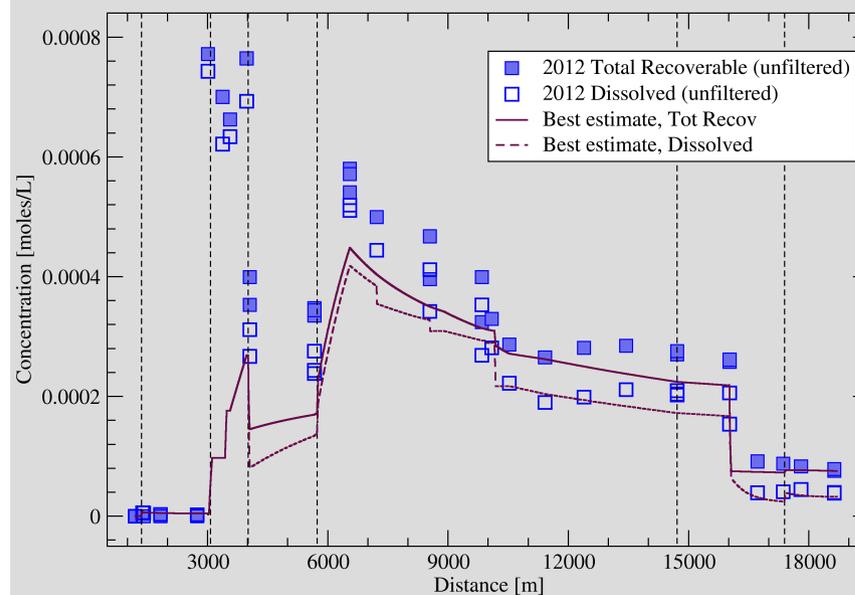
Modify the calibrated model such that no water enters between the two stream sites that bracket the Red & Bonita

Results: 100% removal of Red & Bonita load

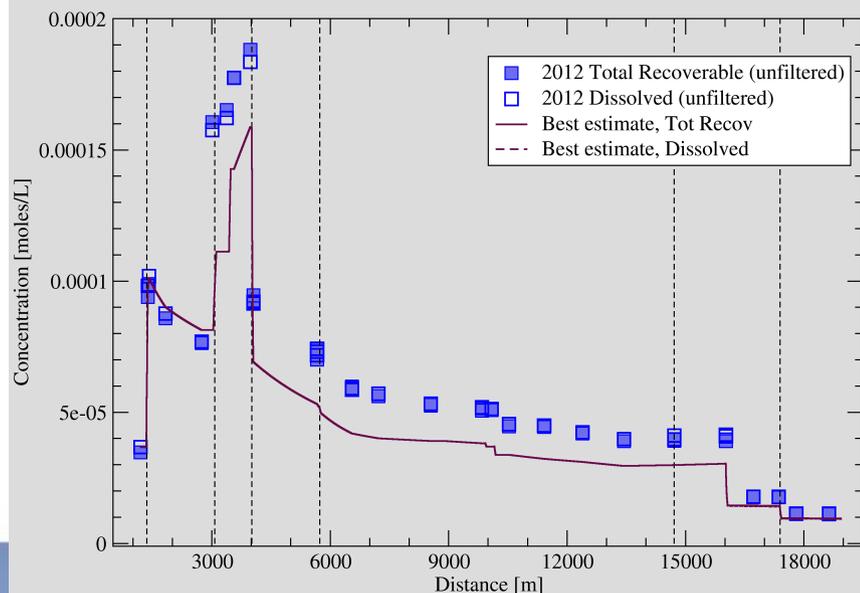
Cd, 100% Removal of Red & Bonita load



Fe, 100% Removal of Red & Bonita load

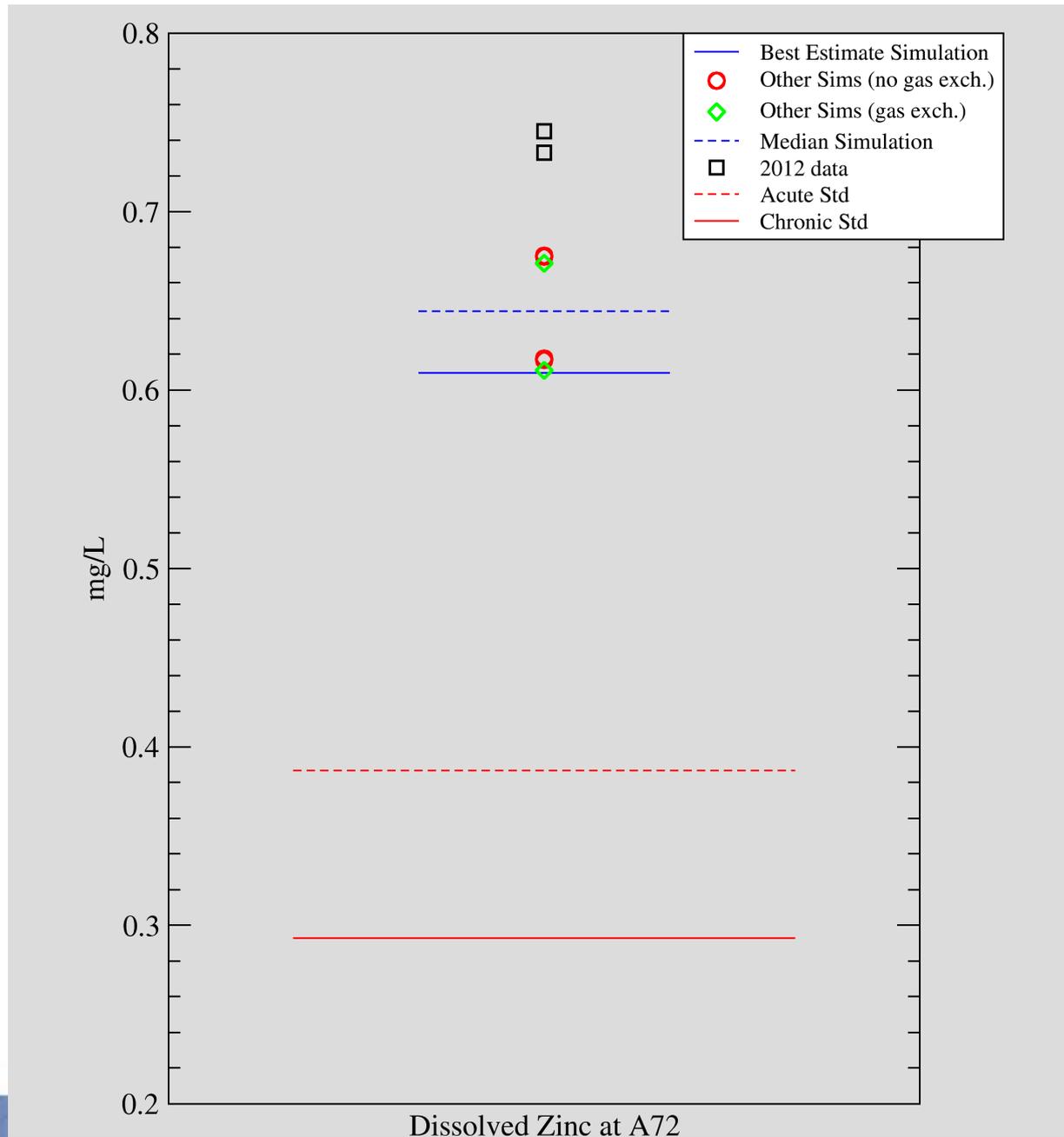


Zn, 100% Removal of Red & Bonita load



pH & Al – v. similar to existing conditions

Results: 100% removal of Red & Bonita load



Normalized Results

- Concentrations vary between constituents
– normalize to allow for comparison

$$C_{norm} = 100 * (C_{pred} - wqstd) / (C_{2012} - wqstd)$$

$C_{norm} = 100 \rightarrow$ Prediction = 2012 conditions

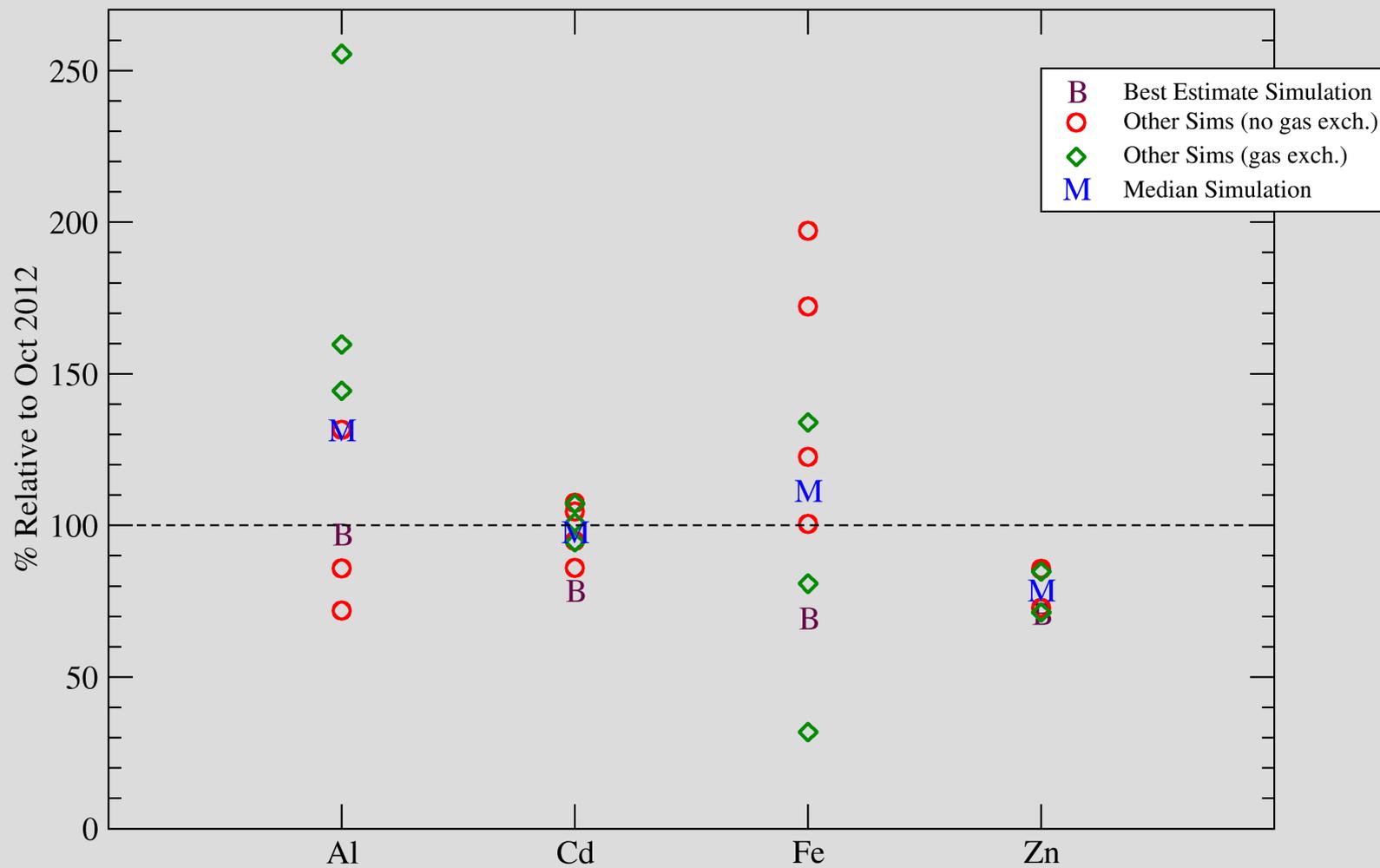
$C_{norm} = 0 \rightarrow$ Prediction = chronic water quality std

$C_{norm} < 0 \rightarrow$ Prediction < chronic water quality std

$C_{norm} > 100 \rightarrow$ Prediction > 2012 conditions

Normalized Results

100% Removal of Red & Bonita Load



Remedial Scenario #2: 1996 Treatment Conditions

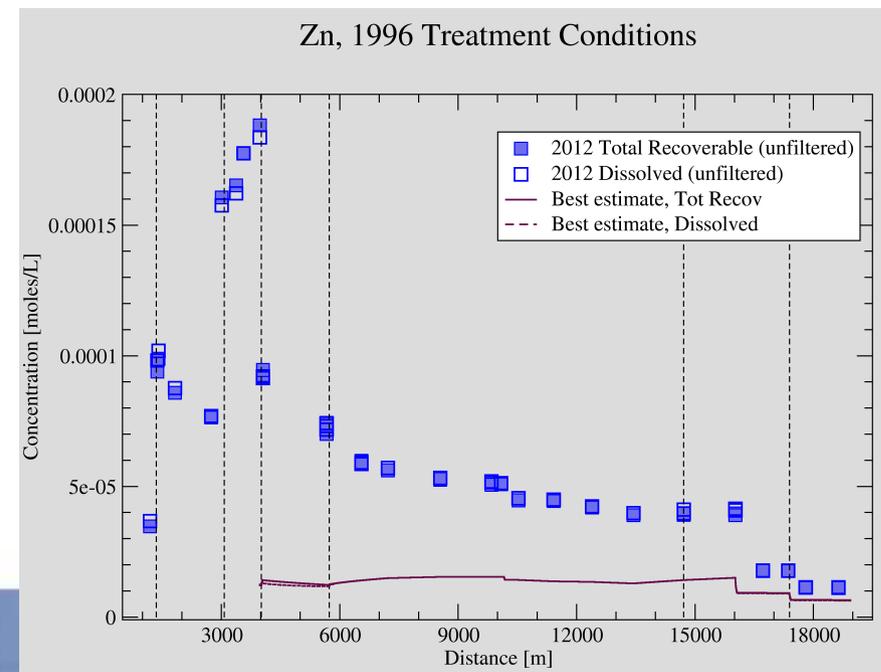
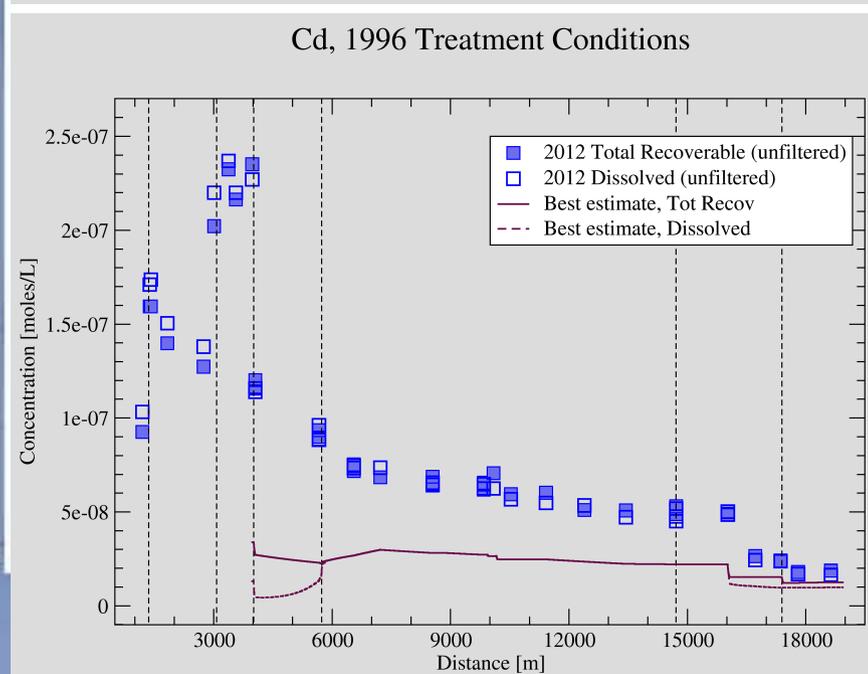
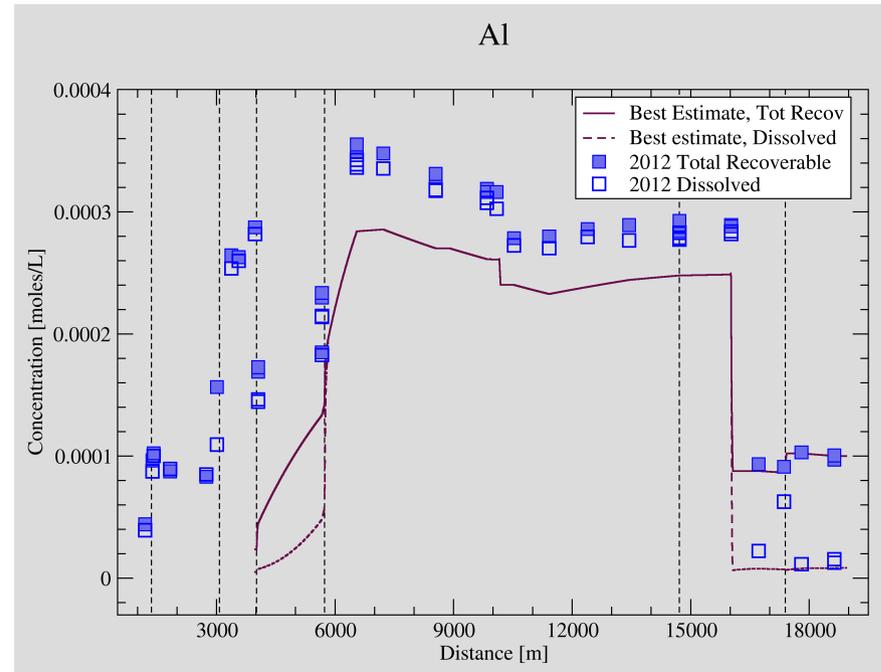
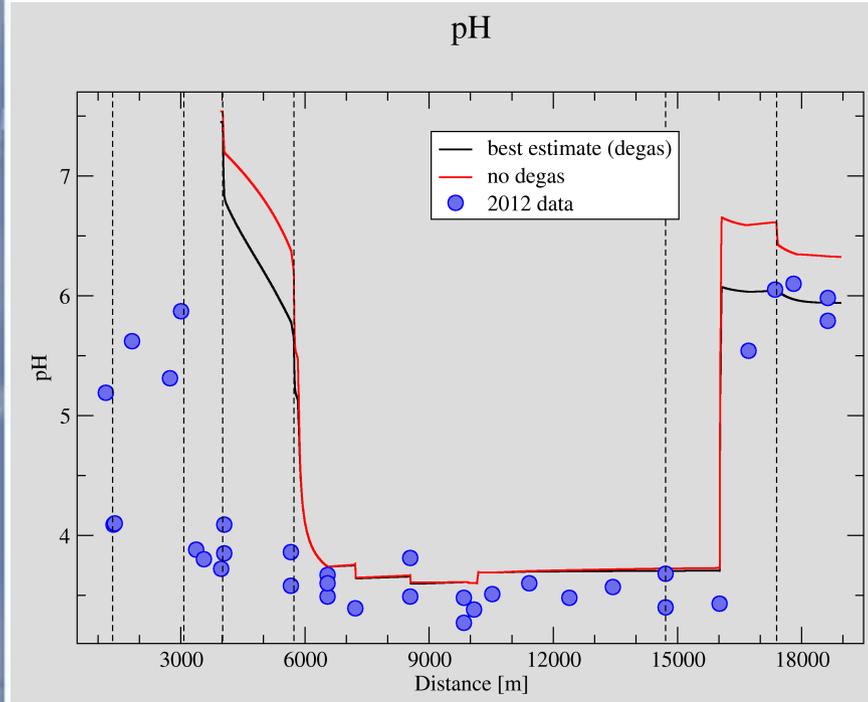


1996: Gladstone treatment system collected most of the water upstream of the South Fork

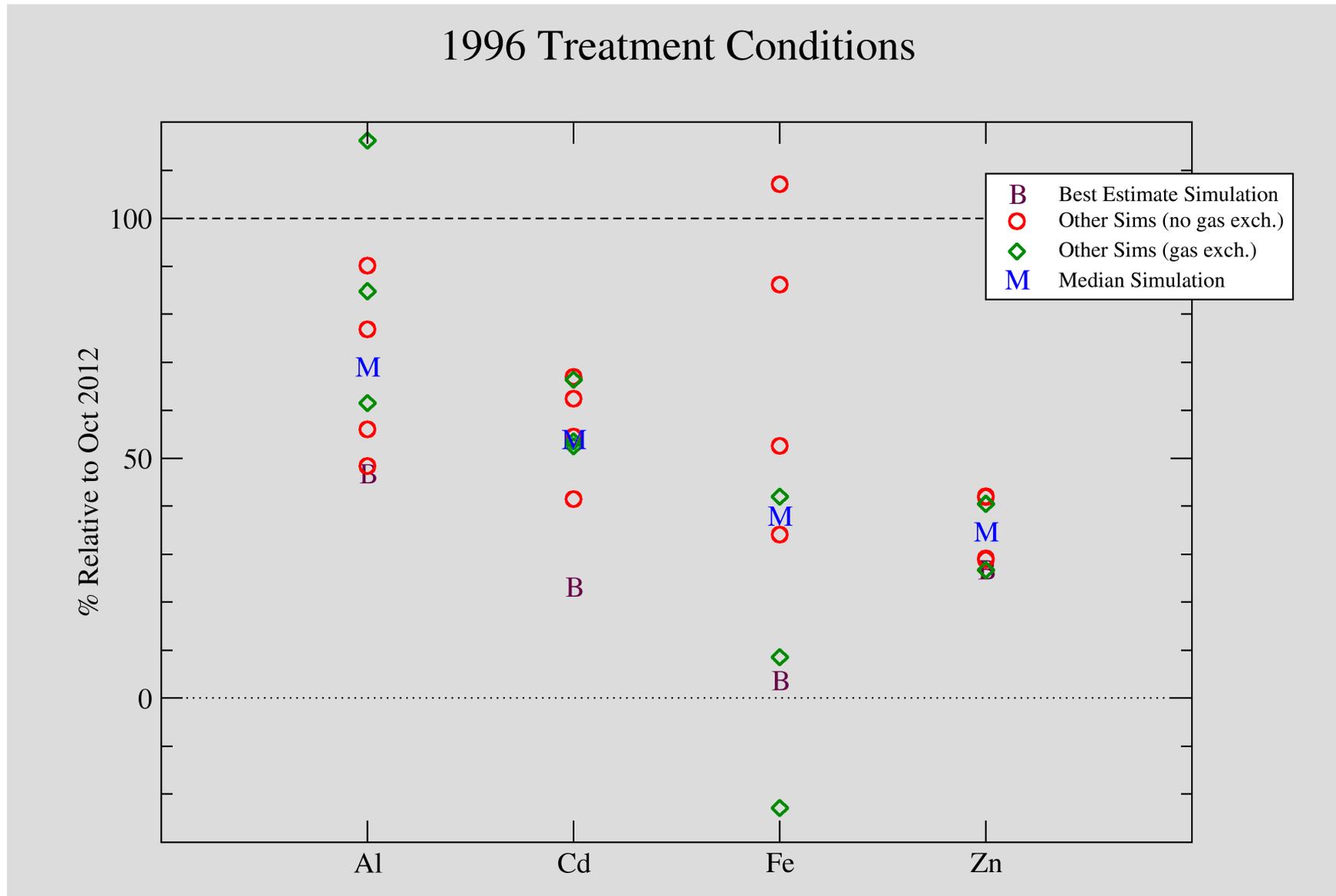
Move upstream boundary (above Mogul) to just above South Fork.

Specify upstream boundary chemistry based on 1996 data for Cement Creek, just downstream of the treatment system

Results: 1996 Treatment Conditions

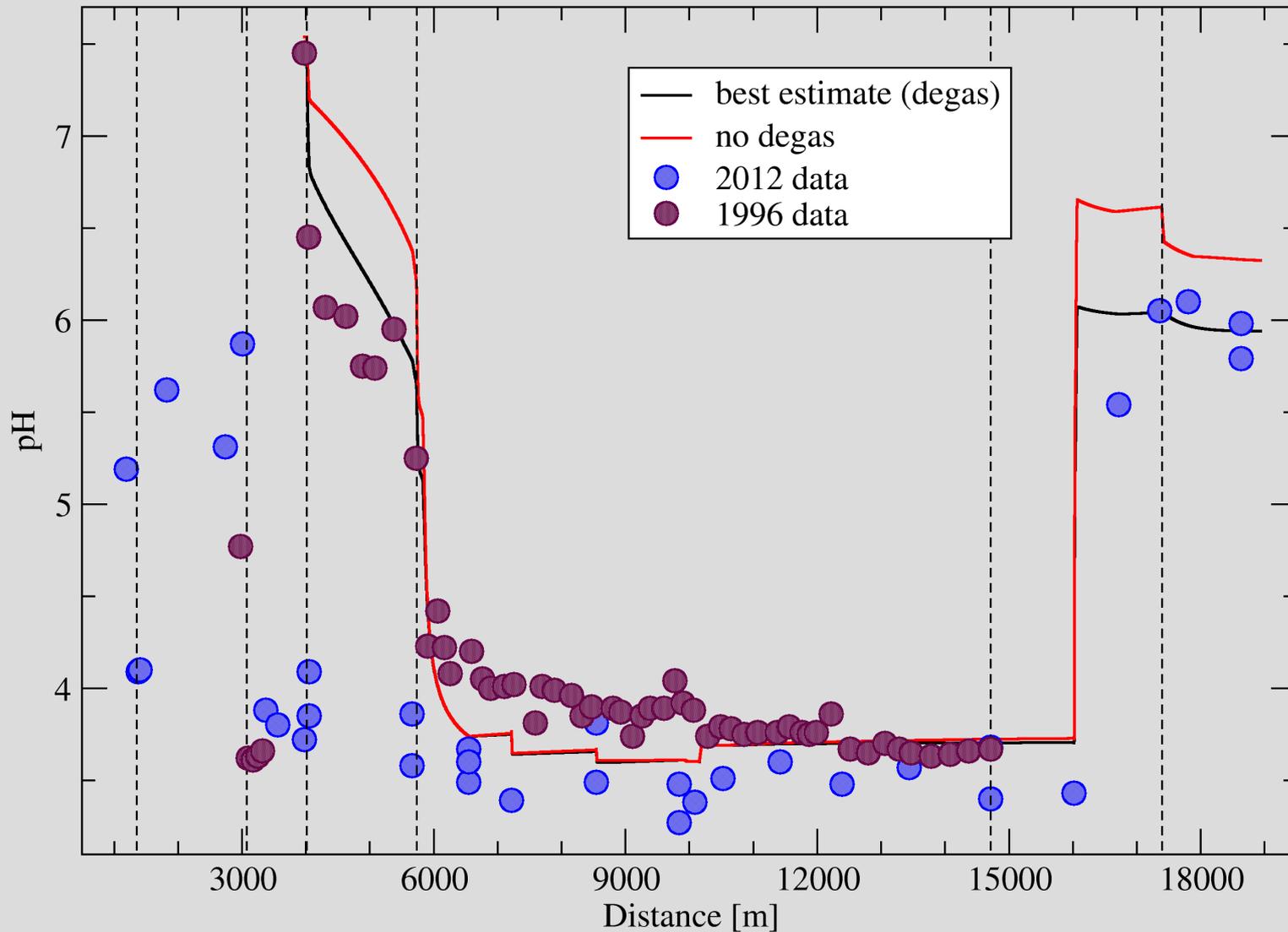


Results: 1996 Treatment Conditions



Model "Validation"

Comparison of Treatment Simulation w/1996 Data



Normalized Results, Revisited

Median Values (n=8)

	Al	Cd	Fe	Zn
R&B Plug	132	98	112	79
1996 Treatment	69	54	38	35

Best Estimate Simulation

	Al	Cd	Fe	Zn
R&B Plug	97	79	69	71
1996 Treatment	47	23	4	27

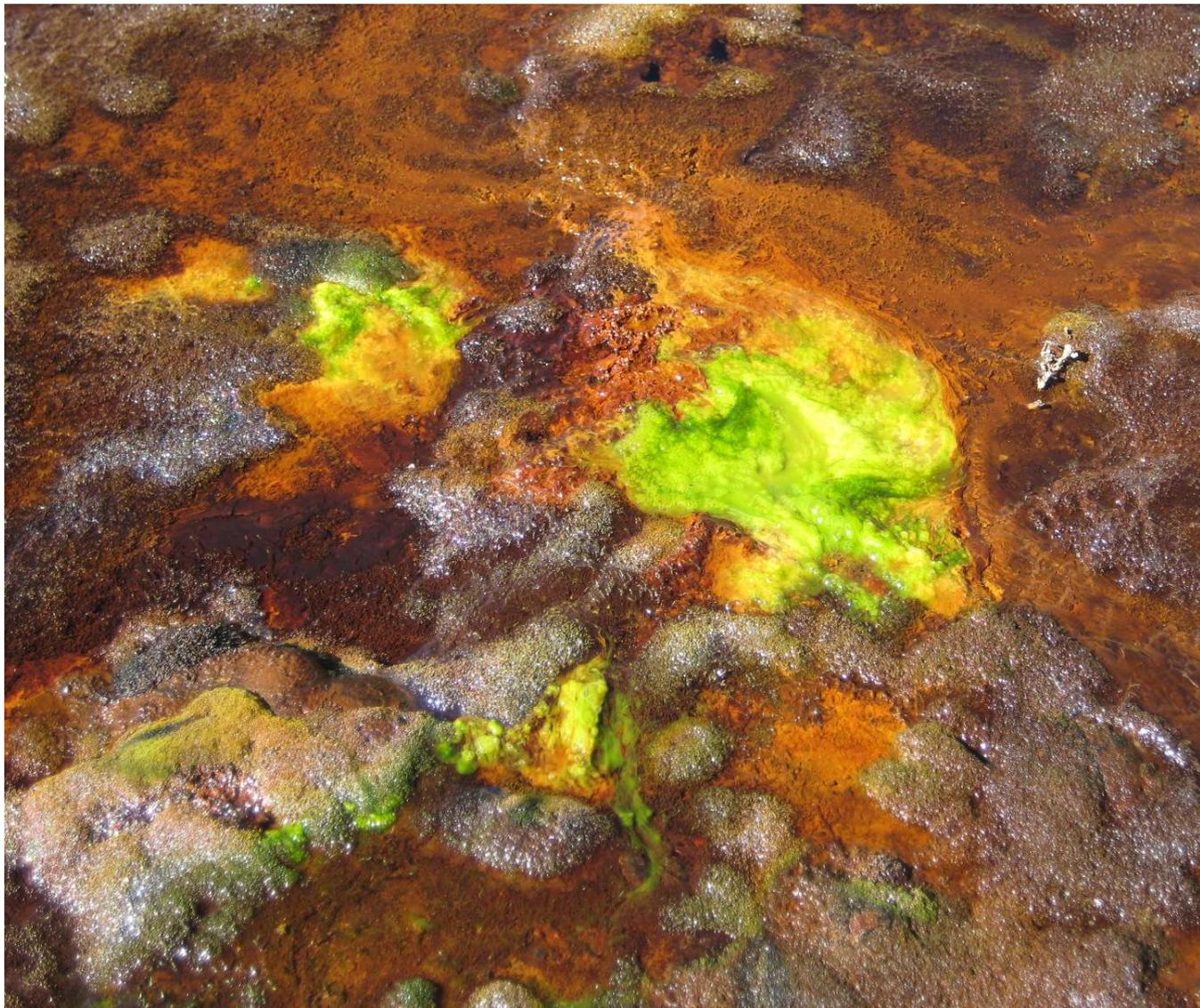
Summary

- Neither scenario results in the attainment of chronic water quality standards at A72
- Cement Creek is an appropriate focus for Fe & Zn
- Cd: need to consider U. Animas & Mineral
- Al: need to consider Mineral Creek

Middle Fork Mineral Creek “White Death”



Next Steps?



Thank you!