

# Measured SF<sub>6</sub> Emissions From Magnesium Die Casting Operations

International Conference on SF<sub>6</sub> and the  
Environment: Emission Reduction Strategies

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Ravi Kantamaneni, ICF Consulting

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# Outline

- Background
- Past Studies
- Experimental Setup
- Results
- Summary

# Background

- EPA's Voluntary Partnership
  - 17 Partners: 100% Primary Production; 80% Casting
  - Partners Track and Report Annual SF<sub>6</sub> Usage
- Industry Emission Estimates based on IPCC Methodology
  - 100% SF<sub>6</sub> Consumption = 100% SF<sub>6</sub> Emissions
- Extent of SF<sub>6</sub> Destruction May Vary
  - Industry Opinion Speculates Between 10 to 50%
- Implement Measurement Study to Quantify Degree of SF<sub>6</sub> Destruction

# Past Studies

- Hanawalt, J.D. (1972)
  - At approx. 2” above the melt surface, potential SF<sub>6</sub> byproducts, including SO<sub>2</sub> and MgF<sub>2</sub>, were identified
- Couling S., et al. (1977)
  - Reported potential SF<sub>6</sub> destruction
- Couling, S., Leontis, T. (1980)
  - Identified CO and HF decomposition products at melt surface
- Tranell et al. (2001)
  - Proposed mechanism for SF<sub>6</sub> melt protection: SF<sub>6</sub> dissociates to highly reactive fluorine species. These species form a dense protective film containing MgF<sub>2</sub>

# Experimental Setup

- Measurements were Conducted on 2 Hot Chambered Die Casting Machines
  - Furnace temp. 650°C; Furnace volumes the same
- Cover Gas Mixture
  - 0.3-0.4% SF<sub>6</sub>; 17% CO<sub>2</sub>; 83% Dry Air
- Cover Gas Supply
  - Approx. 5 l/min
- Magnesium Alloy Ingot Feeding
  - Every 7-9 minutes

# Die Casting Machine



# Experimental Design

- Real-Time SF<sub>6</sub> Monitoring using FTIR Systems
- Monel Sample Probe Placed in Furnace Enclosure at Approx. 1-2" above Melt Surface
- FTIR Sample Flow Maintained at 2-3 l/min
- Signal Averaging over Approx. 2 Minute Periods
- Sample Cell/PFA-Grade Teflon Extraction Lines Maintained at 150°C to Preclude Formation of HF and Prevent Condensation Losses

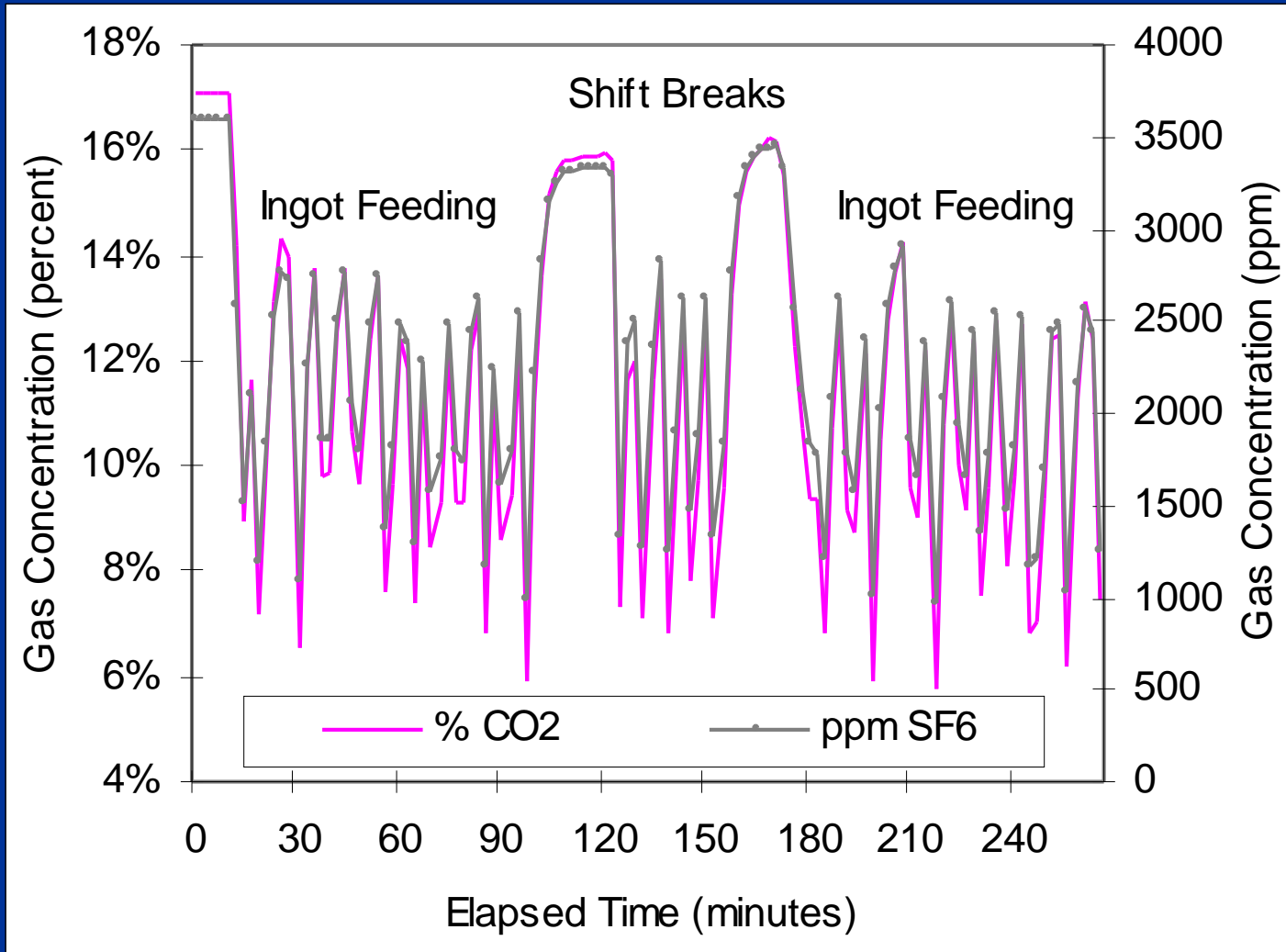




# Experimental Design.....contd.

- Test for Homogeneity of Cover Gas in Furnace Head Space
  - Used 2 FTIR systems of 20.1 and 5.1 meter fixed cell path lengths
  - Confirmed uniform distribution of compound concentrations throughout the furnace enclosure
- Gas grab samples taken for GC analysis
  - For CO<sub>2</sub>, GC/FTIR samples agree within  $\pm 9$  percent
  - For SF<sub>6</sub>, GC/FTIR samples agree within  $\pm 30$  percent

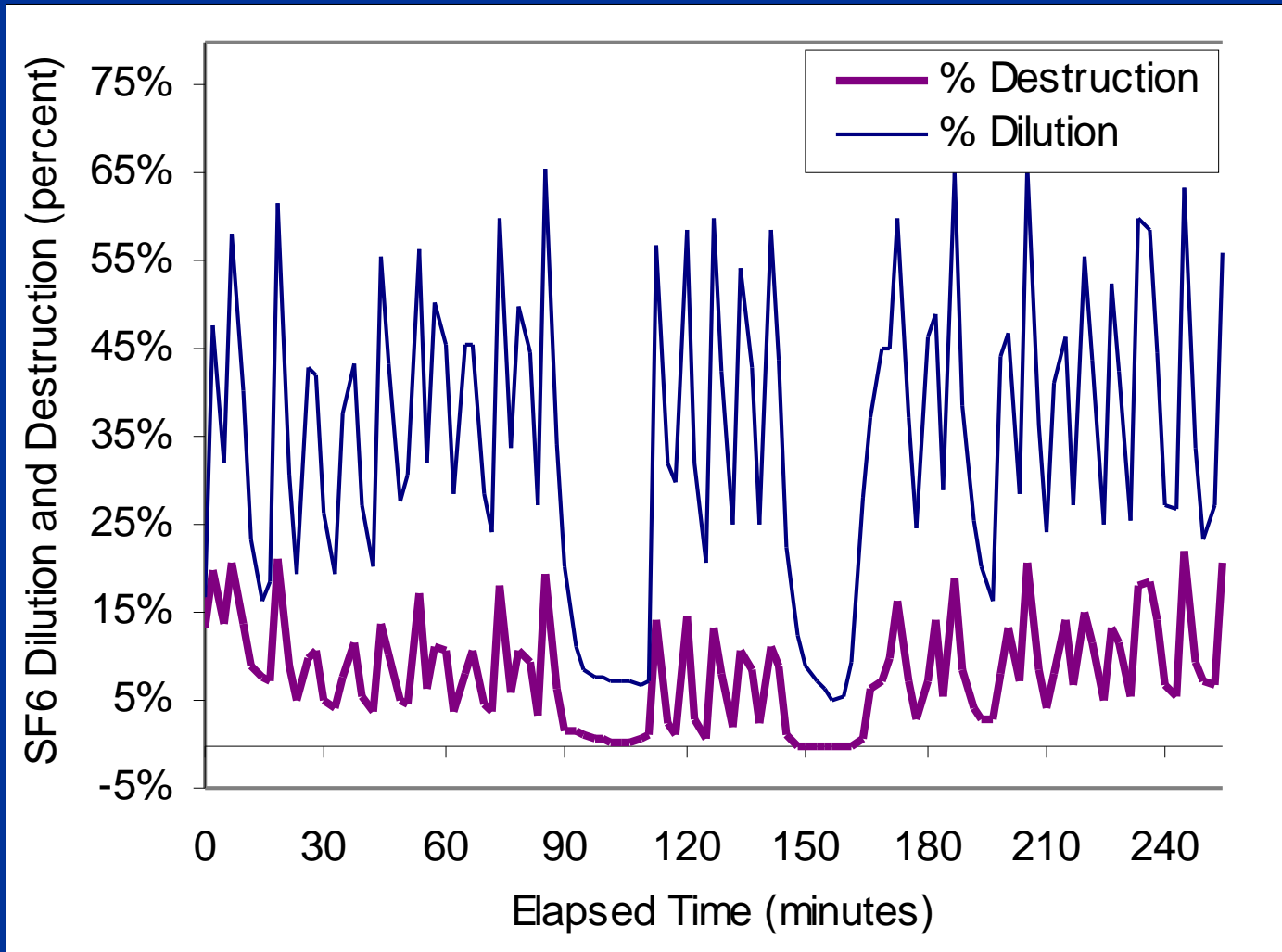
# Time Series Plots of SF<sub>6</sub>/CO<sub>2</sub> – Machine 1



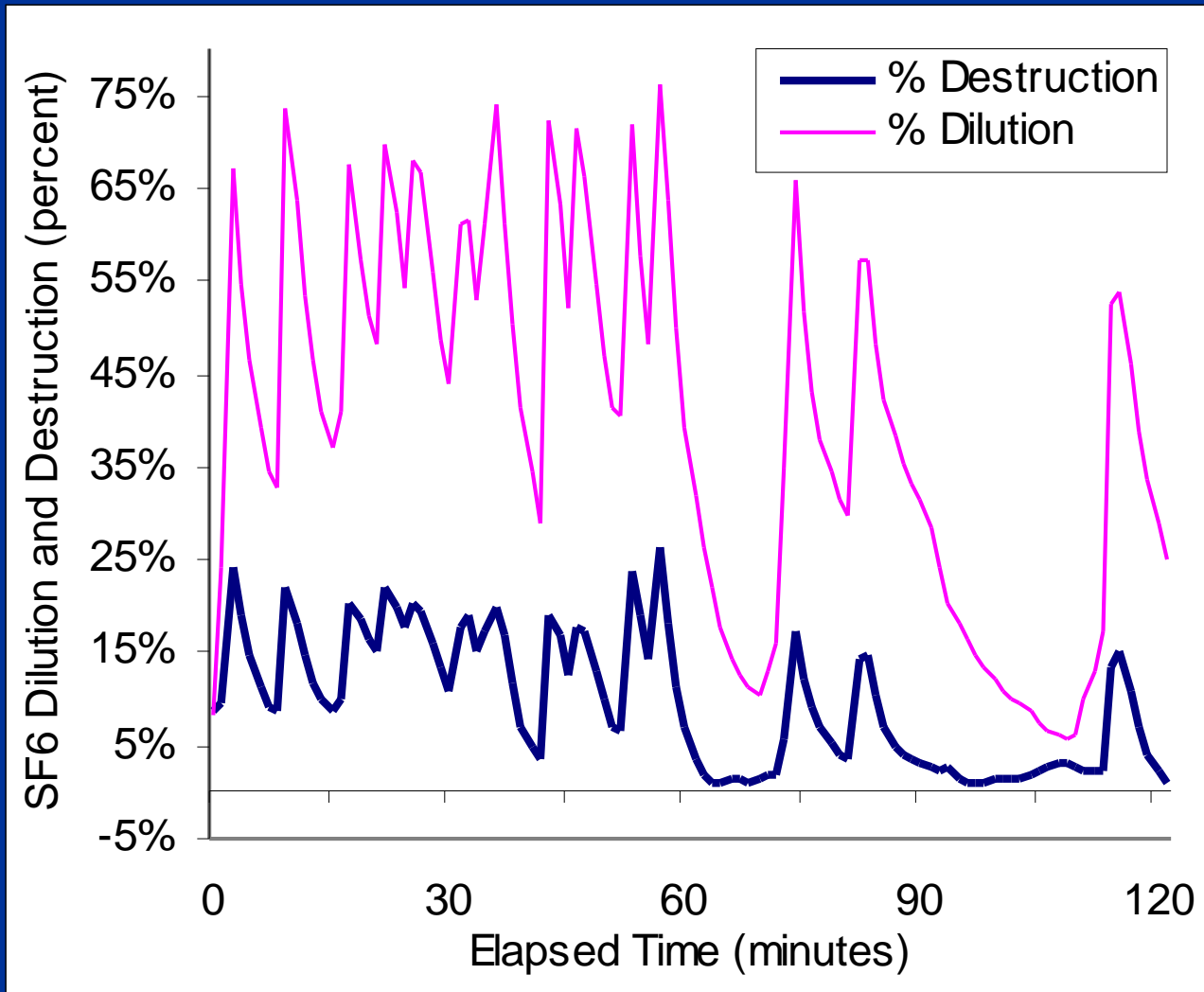
# Assumptions

- Reactivity of  $\text{CO}_2$  was estimated to be  $< 0.3\%$  conversion to  $\text{CO}$
- Consequently, it was assumed that the observed reduction in measured  $\text{CO}_2$ , compared to its feed cover gas concentration, is almost solely attributable to ambient air dilution and leakage
- Used  $\text{CO}_2$  measurements to estimate dilution/leakage effects, i.e., ratio of headspace concentration to average cover gas content
- Quantity of  $\text{SF}_6$  destroyed is defined as the difference between the concentration of  $\text{SF}_6$  that was expected to be in the furnace headspace, after factoring in dilution/leakage effects, and the  $\text{SF}_6$  that was present in the cover gas feed

# SF<sub>6</sub> Dilution and Destruction – Machine 1



# SF<sub>6</sub> Dilution and Destruction – Machine 2



# Observations

- There appears to be a correlation between  $SF_6$  decomposition and dilution spikes, with an increase in destruction occurring during ingot feeding.
- During ingot feeding, magnesium alloy ingots are added to the furnace via the access lid, which disrupts the  $SF_6$  protective film by creating surface turbulence.
- Also, since  $SF_6$  destruction drops to near zero during work stoppages, it can be concluded that the percent destruction is increased during die casting operations

# Results

Operation	Average SF <sub>6</sub> Destruction
Ingot Feeding Periods	15.7 – 20.1 %
Non-Feeding Periods	6.5 – 12.5 %
Staff Breaks/Shift Changes	0.6 – 1.4 %

# Measured Byproducts

- FTIR spectral analysis identified HF and SO<sub>2</sub> as the primary reaction byproducts
- No other byproducts were detected above the noise level of the FTIR
- HF and SO<sub>2</sub> concentrations remained low, on the order of 20 ppmv total
- Since there is little release of gaseous byproducts, most of the SF<sub>6</sub> decomposition may occur at the melt surface



# Summary

- Maximum SF<sub>6</sub> destruction was observed to occur during ingot feeding operations, and ranged from 16 to 20 percent
- During non-feeding periods the level of destruction decreased to approximately half that observed during feeding periods
- Low levels of byproducts detected – possibility that most of the SF<sub>6</sub> decomposition occurs at the molten metal surface

# Acknowledgements

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