

Processing of Biogenic Organics from the Surface to the Clouds

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SOAS Aircraft Effort:

- Characterize CO₂, CH₄, O₃, H₂O vertical profiles
- **Compare atmospheric particle & cloud water organic molecular composition to gain insights into in-cloud aqueous-phase reaction pathways**

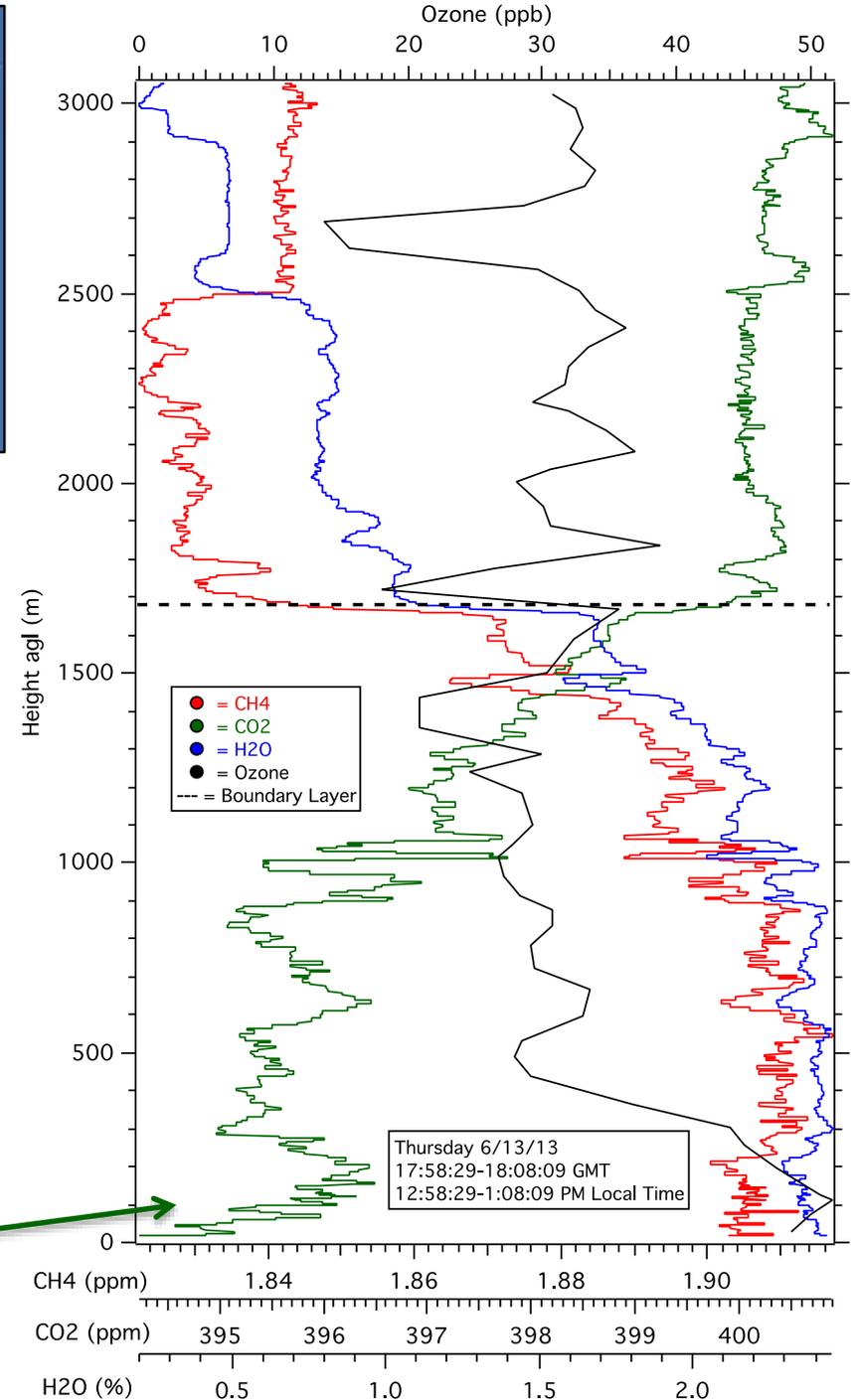
SOAS Centerville Ground Efforts:

- Isoprene nitrate (including lab studies, **Xiong et al., 2015, *ACP***), PAN and MPAN, speciated monoterpene concentrations
- Single-particle morphology & chemical composition (CCSEM-EDX and Raman microspectroscopy)
- Organic molecular composition (nano-DESI-MS)
- Vegetation survey



Clear Sky Vertical Profiles

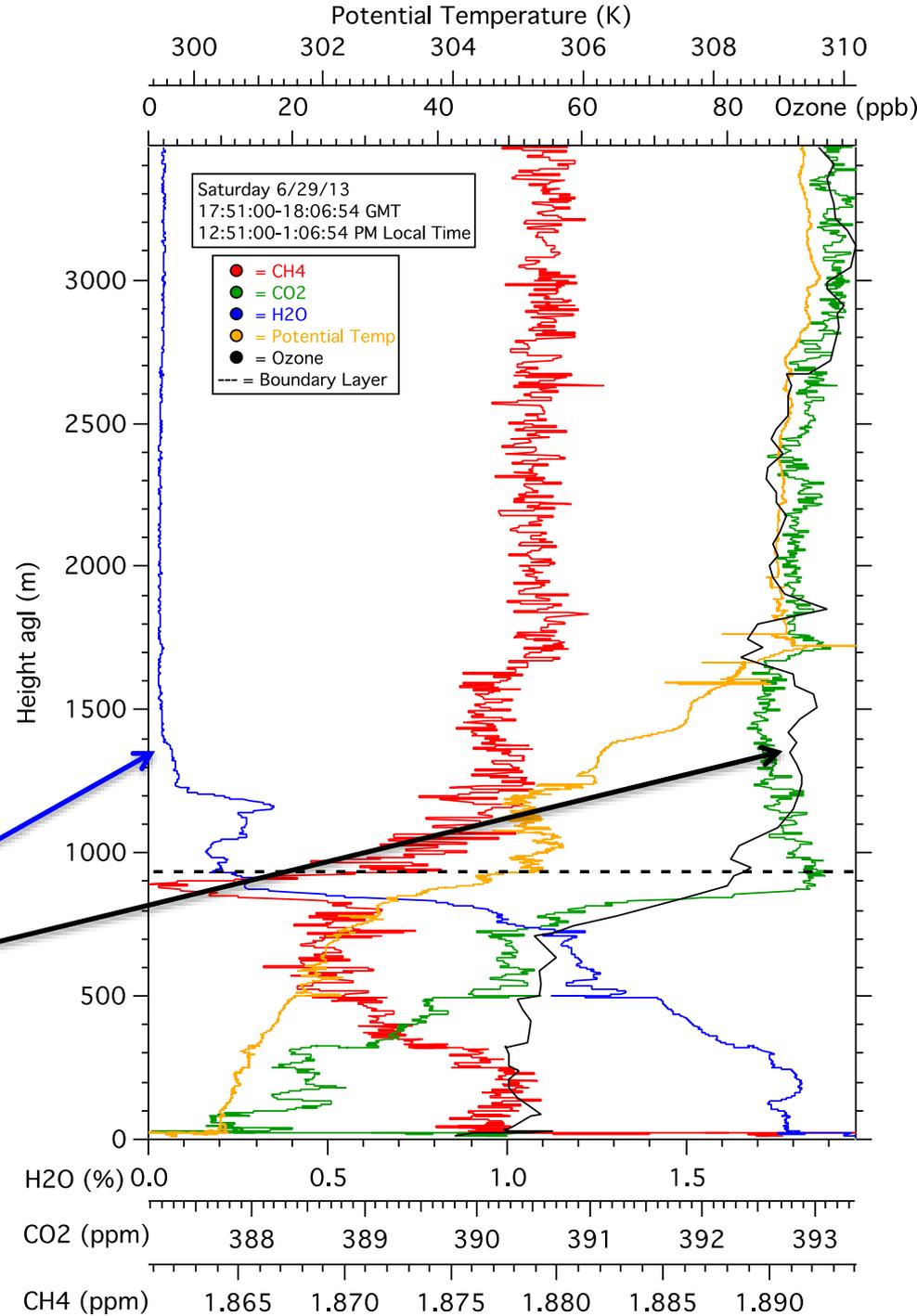
- CO₂, CH₄, H₂O
- O₃
- Temperature, turbulence parameters

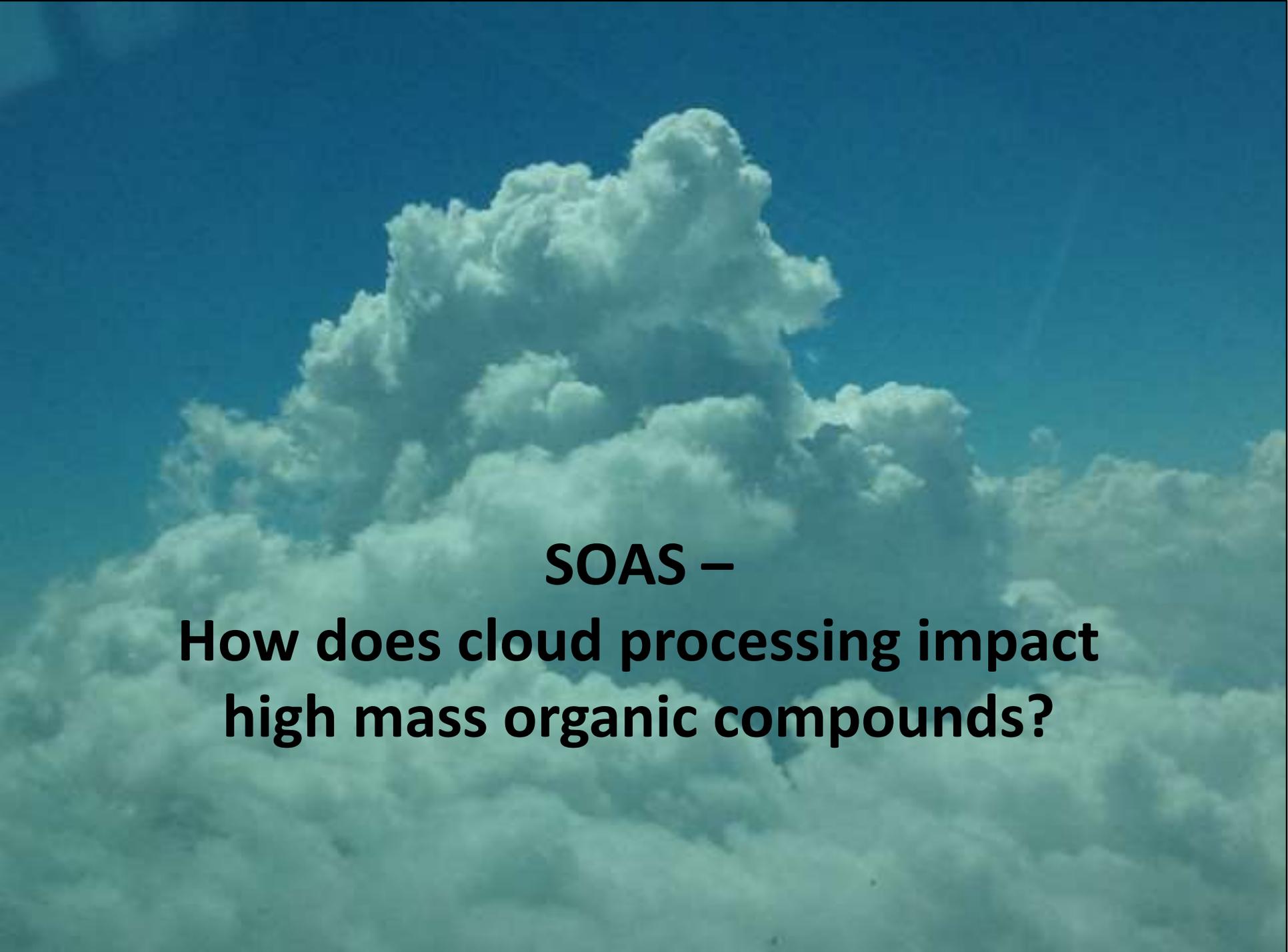


Decreased CO₂
over forest

June 29, 2013 - Case of Stratospheric Intrusion

Very low H₂O
High O₃

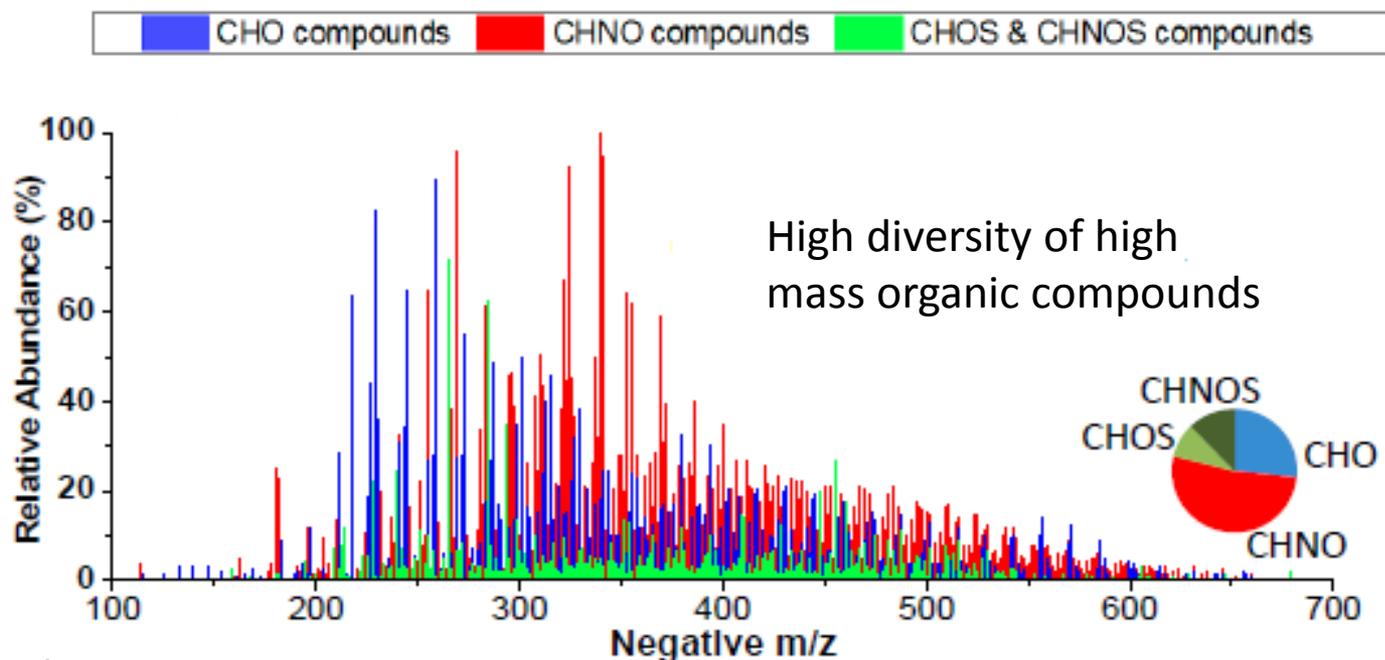


A large, fluffy white cumulus cloud dominates the center of the frame, set against a clear, vibrant blue sky. The cloud has a soft, textured appearance with various shades of white and light blue, suggesting depth and volume. The overall scene is bright and clear, typical of a sunny day.

SOAS –
How does cloud processing impact
high mass organic compounds?

Few Cloud Water Measurements of Individual High Mass Organic Compounds Exist

- Fall, Germany – Feng and Möller 2004, *J. Atmos. Chem.*
- Summer/Fall, Germany – Van Pinxteren and Herrman 2007, *J. Chromat. A*
- Winter, Colorado – Samy et al. 2010, *Atmos. Environ.*
- **Summer, Missouri – Pratt et al. 2013, *Atmos. Environ.***
- Winter, Colorado – Zhao et al. 2013, *ACP*



- Summer/Fall, New York – Sagona et al 2014, *J. Chromatogr. A*

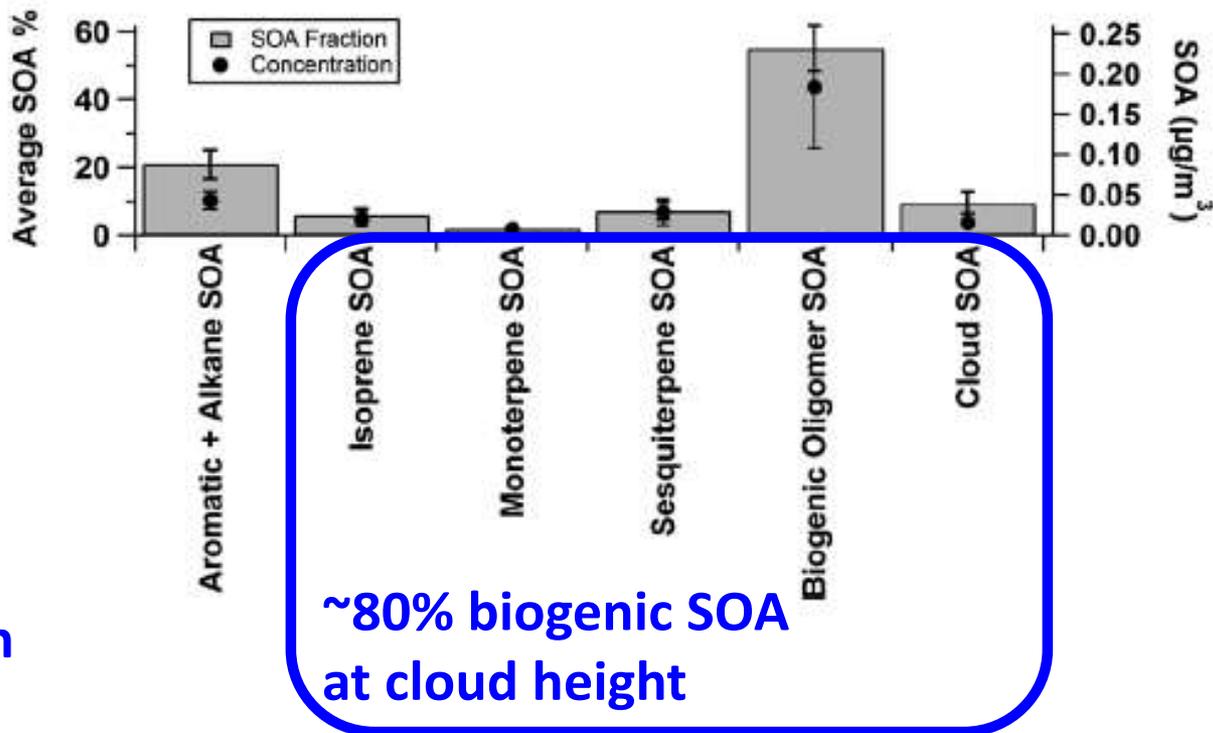
Previous EPA STAR Grants:

Pratt et al. 2013, *Atmos. Environ.*

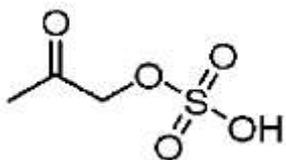
*EPA STAR: G2007-STAR-E1 & 83504101

Biogenic VOC Influence on Cloud Water

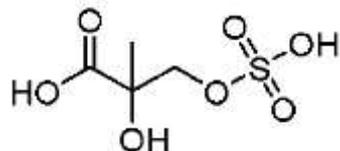
- Cloud water collection & ESI-MS analysis
- CMAQ simulations for cloud sampling altitude in July over the Missouri Ozarks (high isoprene emissions)



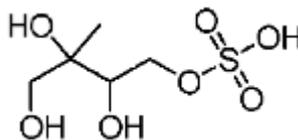
- **Detection of isoprene-derived organosulfates in cloud water:**



m/z 153



m/z 199



m/z 215

- All proposed structures (Surratt et al. 2007, *ES&T*) are primary organosulfates that are resistant to hydrolysis (Darer et al. 2011, *ES&T*).

Focus: Aerosol & Cloud Sampling

Boone et al. 2015, *ES&T*



Step 2: Cloud water sampling at ~ 1.8 - 2.0 km asl (middle/upper sections of developing cumulus clouds)



Step 1: Below-cloud particle sampling at ~ 1 - 1.2 km asl

- June 10, 13, & 27 flights with particle + cloud water sampling
- Moderate \rightarrow very high convective instability (boundary layer influence)

Aircraft-based Sample Collection

- [Below-cloud particle collection](#)
 - ~30-45 min of sampling onto Al foil using a rotating drum impactor (Bateman et al. 2009, *PCCP*)
 - Analyzed particles from Stage 2 (0.34-1.2 μm) – expected to have activated to form cloud droplets
 - Estimated ~20 ng of organic material collected
- [Cloud water collection](#)
 - ~30-45 min of sampling using a modified Mohnen slotted rod cloud water collector (Huebert et al. 1988, *J. Atmos. Chem.*)
 - Cloud droplet 50% cut-point of ~5.5 μm (Kim & Boatman 1992, *J. Atmos. Ocean Technol.*)
 - Typically ~5 mL of cloud water collected

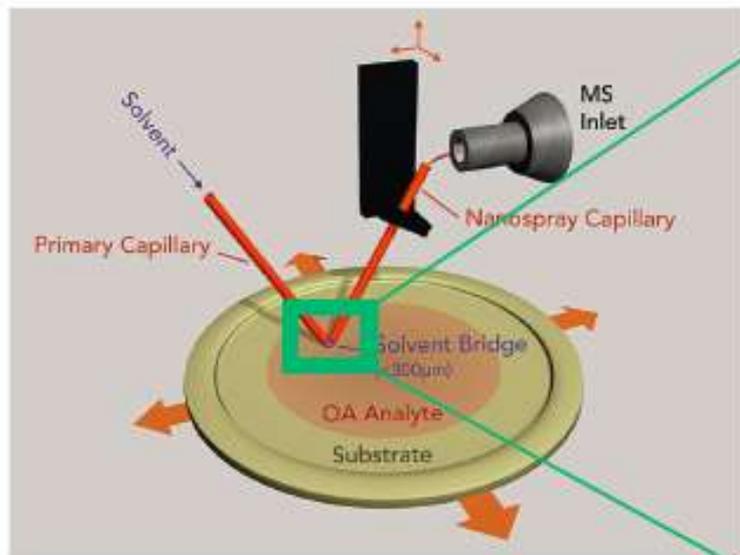


cloud water collector

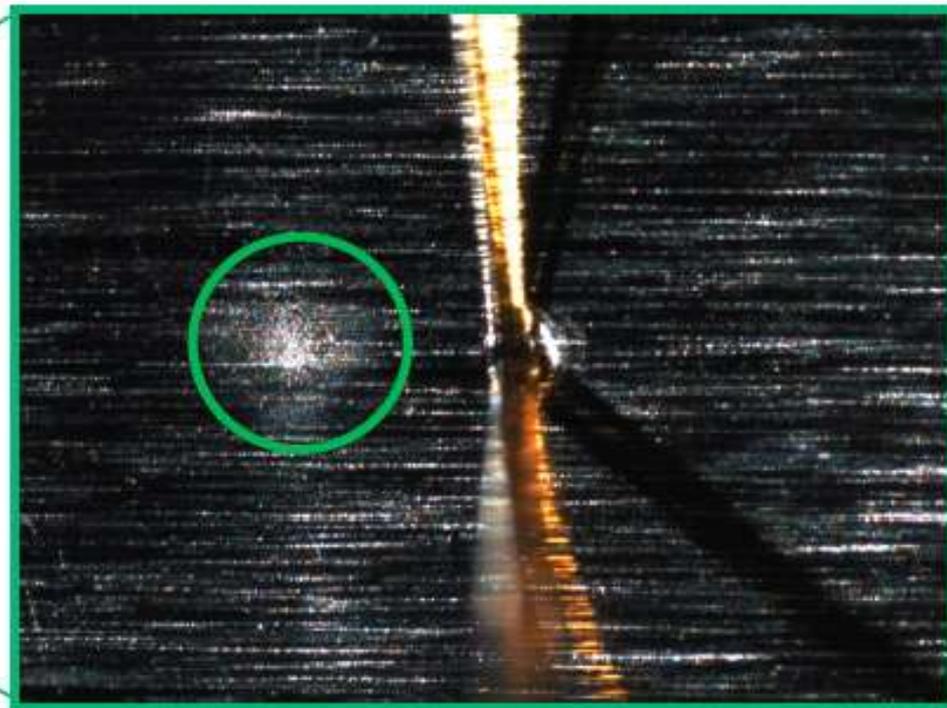


Mass Spectrometry Analysis

- Cloud water: Direct injection electrospray ionization (ESI), following acetonitrile addition (70% by volume)
- Atmospheric particles: Nanospray desorption electrospray ionization (nano-DESI) using acetonitrile/water (70/30, by volume) as the solvent

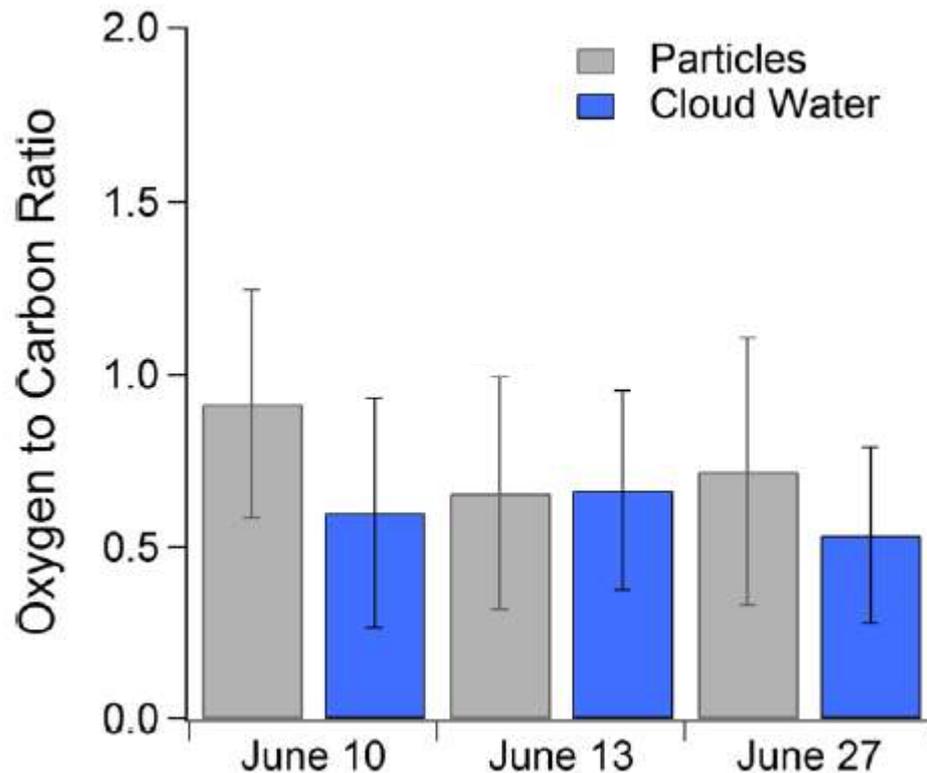


Roach et al. 2010, *Analytical Chem.*



- Orbitrap mass spectrometry ($R = 100,000$)
 - Negative ion mode; m/z 150-1000 (focus on $m/z < 400$); Kendrick mass defect analysis used to assign peaks

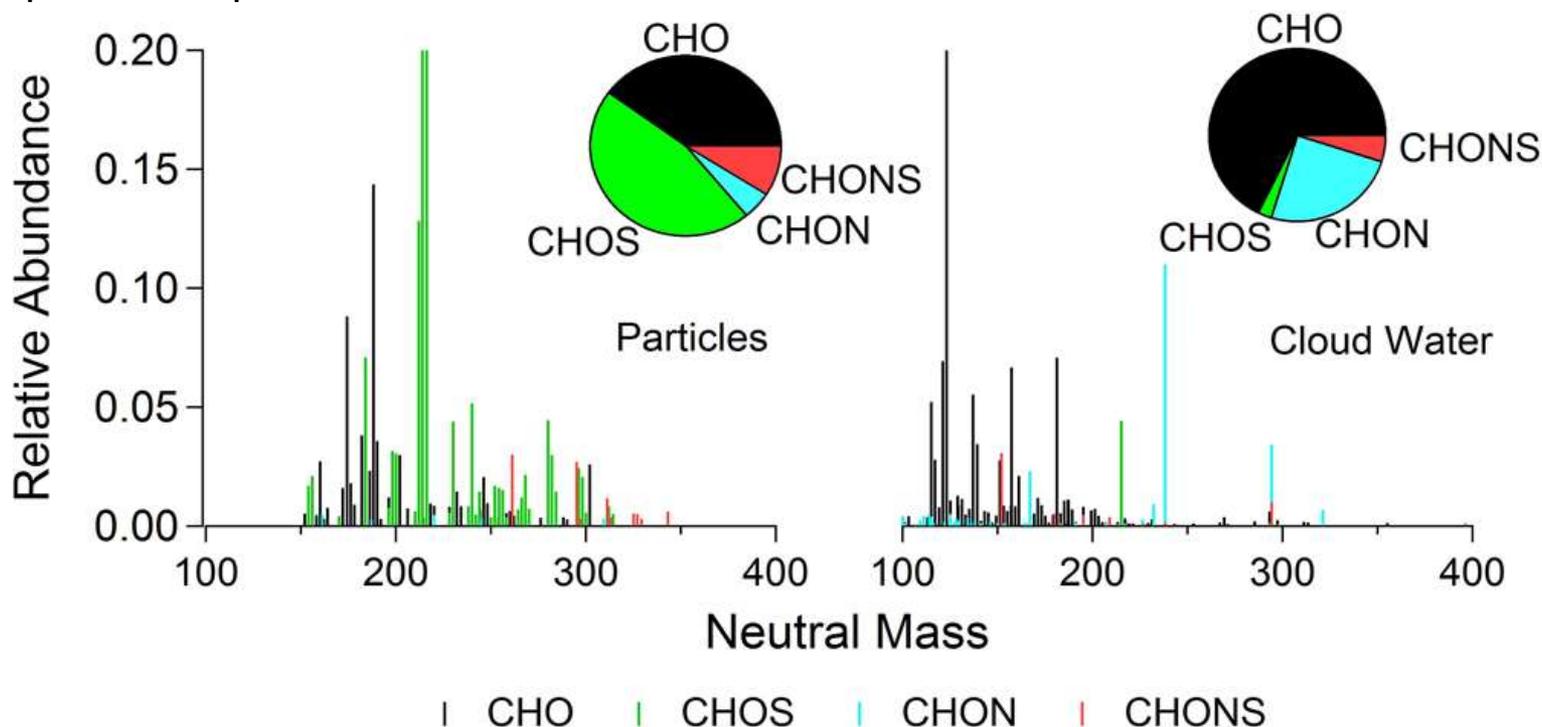
No Change in Bulk O:C of High Mass Organics Observed with Cloud Processing



- Expected organic compounds in cloud water to have a higher O:C ratio based on many bulk aqueous photochemistry lab studies (Ervens et al. 2011, *ACP*)
- However, no O:C ratio change in recent biogenic SOA aqueous photochemistry studies (Nguyen et al. 2012, *ACP*; Romonosky et al. 2014, *JPCA*)

Below-Cloud Aerosol vs. Cloud Water

Example Mass Spectra



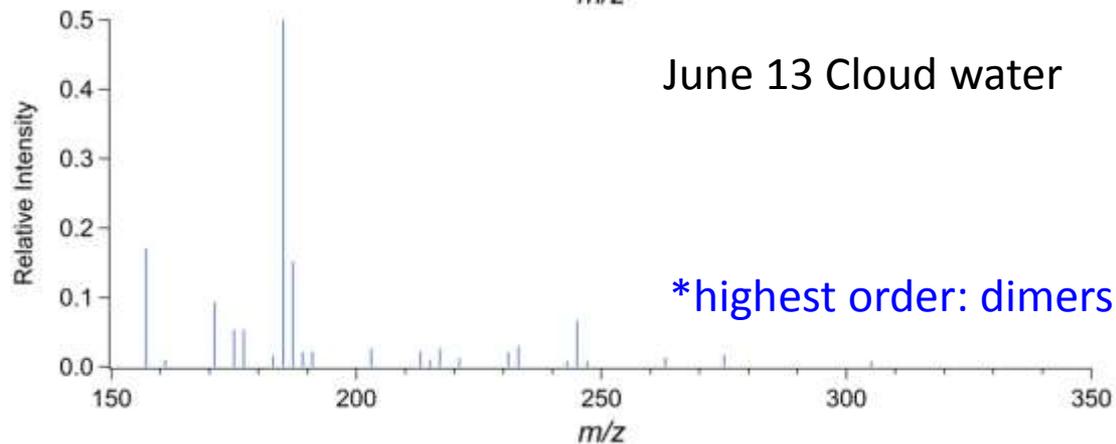
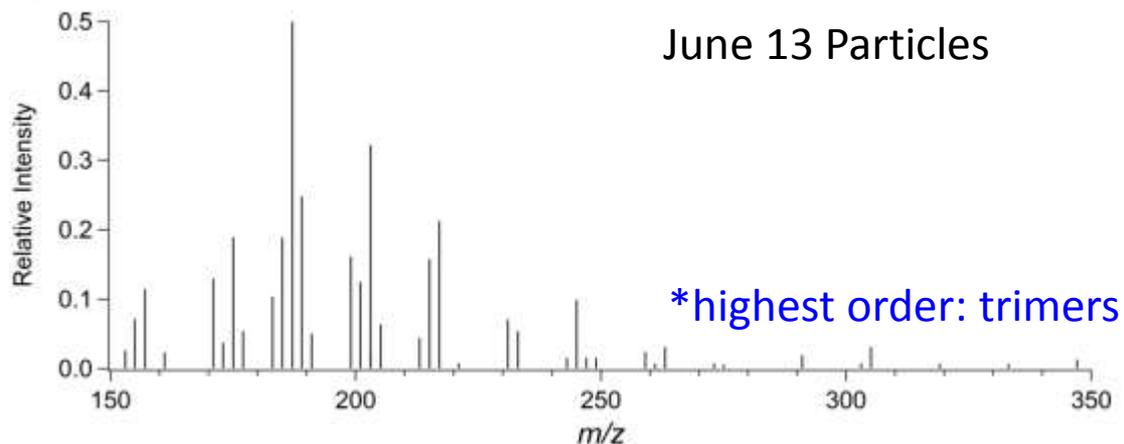
General Trends:

- Significant decreases in CHOS & increases in CHON (number) from particles → cloud water
- Decreasing overall molecular masses from particles → cloud water

Fragmentation of Isoprene SOA Oligomers in Cloud Water

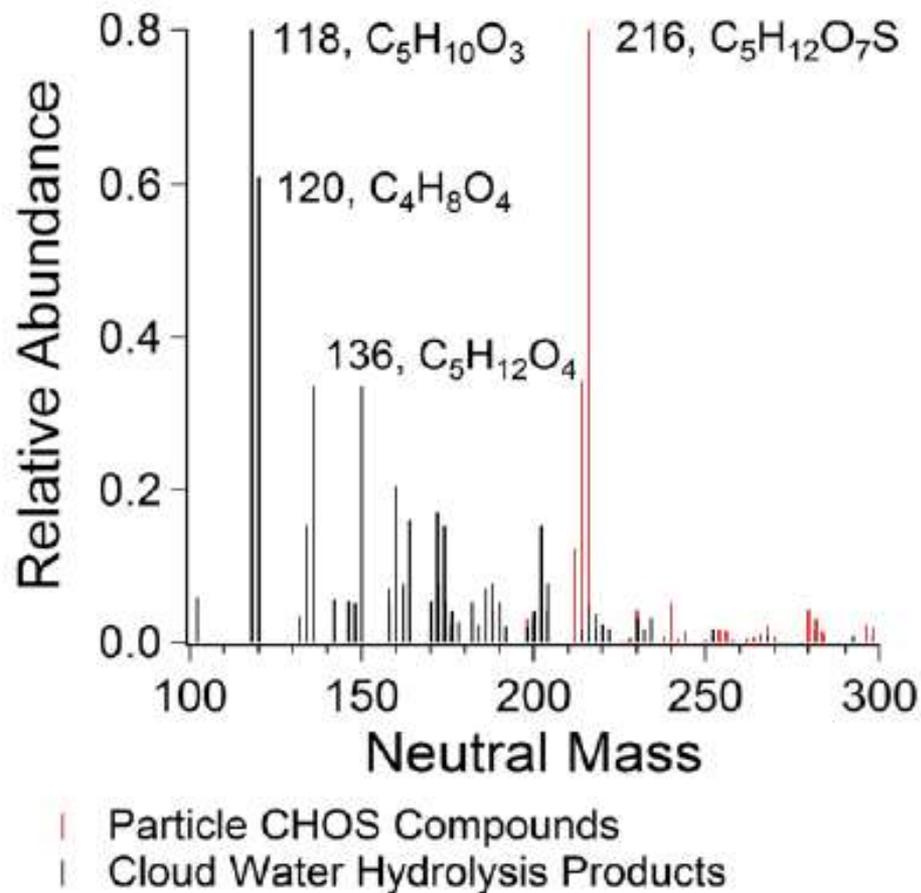
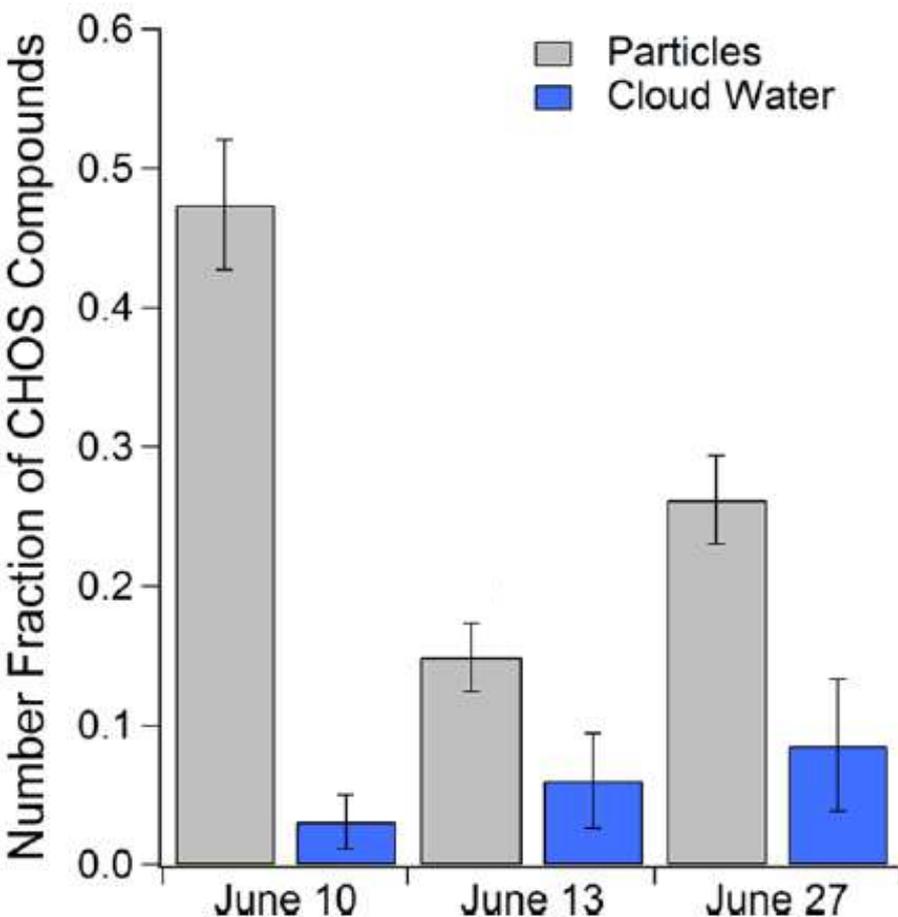
2-methylglyceric acid (2MGA) oligomers (Surratt et al. 2006, *JPCA*)
& glycoaldehyde oligomers (Nguyen et al. 2011, *ACP*)

- 2MGA oligomers enhanced in particle-phase under lower RH (Nguyen et al. 2011, *ACP*; Zhang et al. 2011, *ACP*)
- 2MGA oligomers previously suggested to be observed in cloud water (Pratt et al. 2013, *AE*)



Reconstructed mass spectra

In-Cloud Hydrolysis of Organosulfates

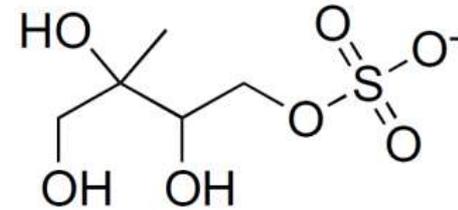


- $ROSO_3H + H_2O \leftrightarrow ROH + H_2SO_4$
- ~70% of CHOS compounds had corresponding hydrolysis product CHO formulas in the cloud water

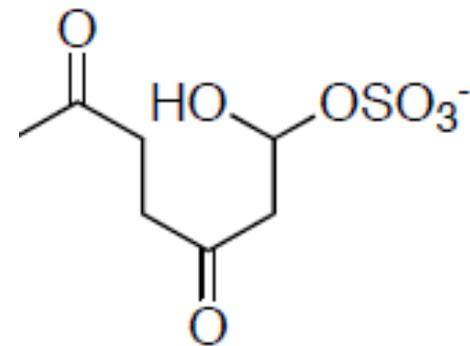
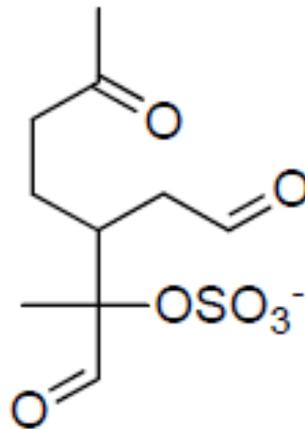
Most Abundant Organosulfates

Across all particle-phase samples:

- m/z 211 ($C_5H_7O_7S^-$), 213 ($C_5H_9O_7S^-$), 215 ($C_5H_{11}O_7S^-$)
- Attributed to isoprene oxidation
- Particle samples by number: 30-50% of identified organosulfates attributed to isoprene oxidation, with the remaining identified organosulfates corresponding to monoterpene oxidation.
- Particle samples by intensity: >90% from isoprene oxidation

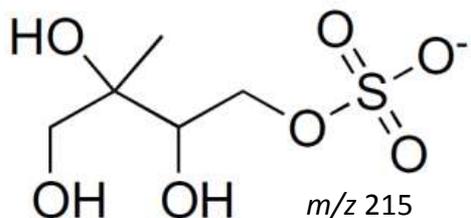


Other abundant organosulfates:
limonene oxidation products



Primary Organosulfates

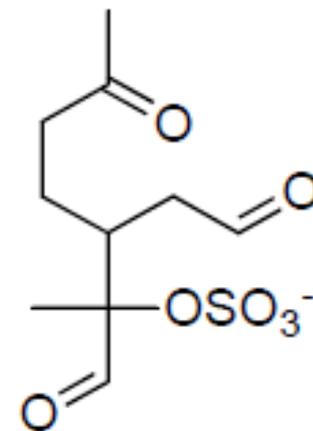
- Only abundant organosulfate observed in cloud water (also observed in particles):



- Primary isoprene-derived organosulfate
- Hydrolysis lifetime >2500 h (Darer et al. 2011, *ES&T*)
- Previously measured in cloud water (Pratt et al. 2013, *Atmos. Environ.*)

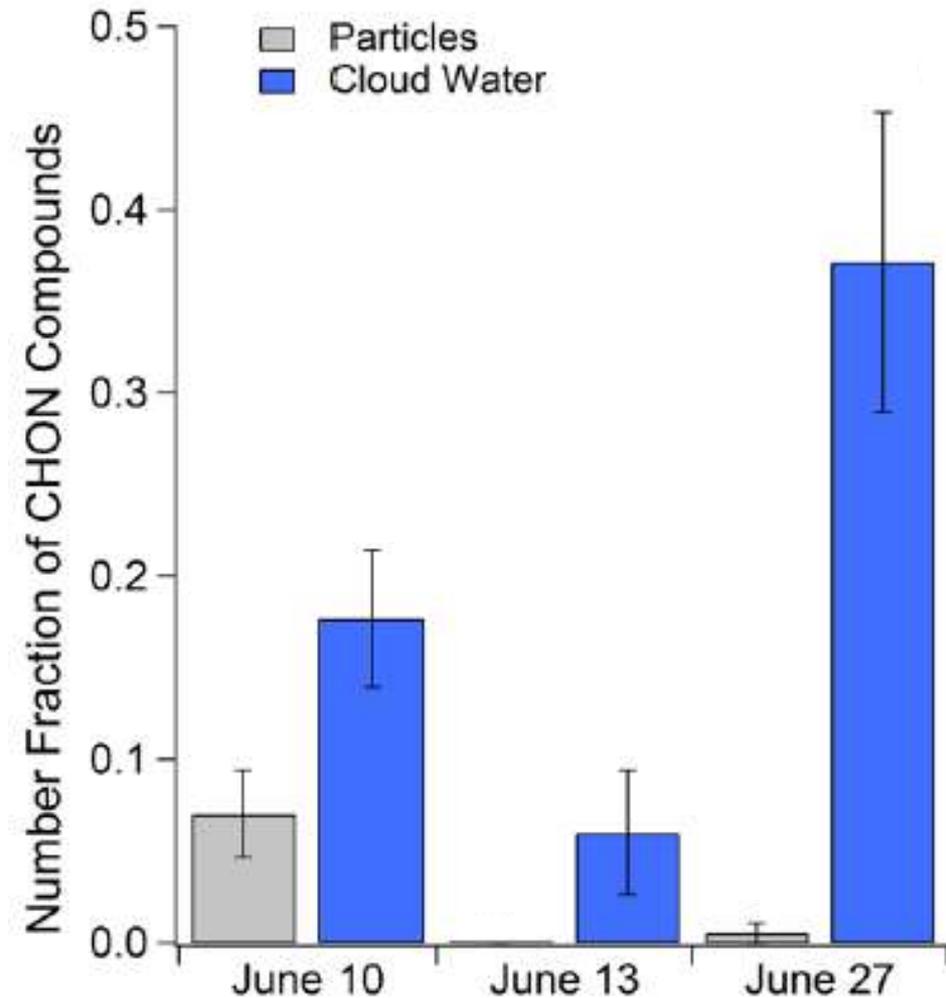
Tertiary Organosulfates

- Considering all proposed organosulfates with structures identified based on comparisons to lab studies:
 - 50-86% particle-phase organosulfates had tertiary functionality.
 - Shorter hydrolysis lifetimes of ~20-460 h at neutral pH, shorter in acidic cloudwater (Darer et al. 2011, *ES&T*)
 - Only 12% of these tertiary organosulfates were observed in cloud water.



Boone et al. 2015, *ES&T*

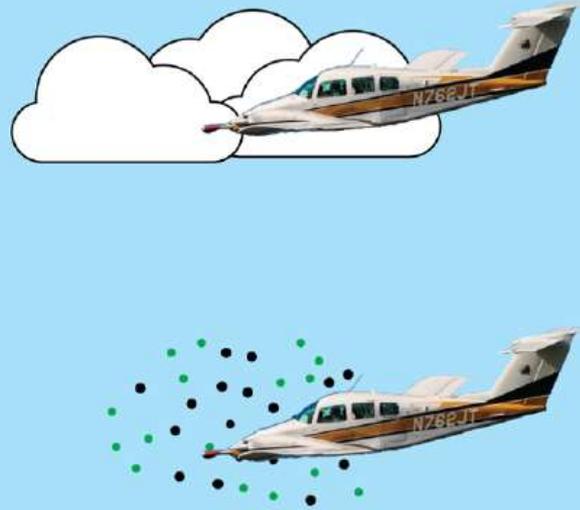
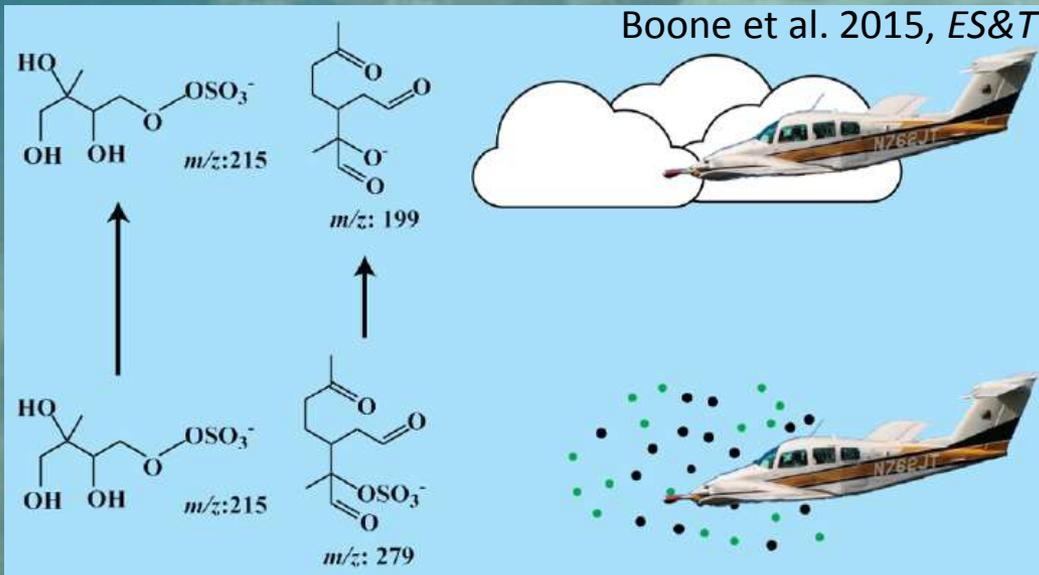
Formation of CHON Compounds



- On average, ~4x more CHON compounds detected in cloud water
- Average N:C ratio increased from 0.03 for particles to 0.14 for cloud water (0.3 for CHON compounds in cloud water)
- Not organonitrates, ~50% expected to have aromatic structures (chromophores)
- Formation mechanism? More lab studies needed!

Aqueous Processing of Atmospheric Organic Particles in Cloud Droplets

- Significant influence of BVOC oxidation products during convective cloud events during SOAS
- No overall trend in O:C ratio with cloud processing
- In-cloud fragmentation of isoprene oxidation oligomers
- Hydrolysis of tertiary organosulfates in cloud droplets



- Formation of CHON compounds
- Additional measurements of high molecular weight compounds in atmospheric particles and cloud droplets are needed to assess cloud processing pathways!