
Final



Development Document for Effluent Limitations Guidelines and Standards for the

Electrical and Electronic Components

Point Source Category (Phase I)



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DEVELOPMENT DOCUMENT
for
EFFLUENT LIMITATIONS GUIDELINES
for the
ELECTRICAL AND ELECTRONIC COMPONENTS
POINT SOURCE CATEGORY
PHASE I

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EXECUTIVE SUMMARY

CONCLUSIONS

A study of the Electrical and Electronic Components Industrial Point Source Category was undertaken to establish discharge limitations guidelines and standards. The industry was subcategorized into 21 segments based on product type. Of the 21 subcategories, 17 have been excluded under Paragraph 8 of the NRDC Consent Decree, two are the subject of the Phase II Electrical and Electronic Components Proposed Rule, electron tubes and luminescent materials, and for the remaining two subcategories regulations are being promulgated. The last two subcategories are Semiconductors and Electronic Crystals.

In the Semiconductor and Electronic Crystals subcategories, pollutants of concern are fluoride, toxic organics, arsenic, and total suspended solids. The major source of fluoride is the use of hydrofluoric acid as an etchant or cleaning agent. Toxic organics are associated with the use of solvents in cleaning and degreasing operations and solvent based process chemicals. Arsenic is only found in significant concentrations at facilities that manufacture gallium or indium arsenide crystals; it is present in the wastewater as a result of the manufacturing process. Suspended solids are only found in significant concentrations at facilities that manufacture crystals where the solids come from cutting and grinding operations.

Several treatment and control technologies applicable to the reduction of pollutants generated by the manufacture of semiconductors and electronic crystals were evaluated, and the costs of these technologies were estimated. Pollutant concentrations achievable through the implementation of these technologies were based on industry data and transfer of performance assessments from industries with similar waste characteristics. These concentrations are presented below as limitations and standards for the semiconductor and electronic crystals subcategories.

EFFLUENT LIMITATIONS AND STANDARDS

For both subcategories, Tables 1 through 10 present regulations for Best Practicable Control Technology (BPT), Best Available Control Technology (BAT), Best Conventional Pollutant Control Technology (BCT), New Source Performance Standards (NSPS), and Pretreatment Standards for New and Existing Sources (PSNS and PSES). All limitations and standards are expressed as milligrams per liter.

TABLE 1: BPT REGULATIONS FOR SEMICONDUCTORS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>	<u>pH Range</u>
Total Toxic Organics *	1.37	**	
pH			6-9

TABLE 2: BAT REGULATIONS FOR SEMICONDUCTORS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>	
Total Toxic Organics *	1.37	**	
Fluoride	32	17.4	

TABLE 3: BCT REGULATIONS FOR SEMICONDUCTORS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>	<u>pH Range</u>
pH			6-9

TABLE 4: NSPS REGULATIONS FOR SEMICONDUCTORS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>	<u>pH Range</u>
Total Toxic Organics *	1.37	**	
Fluoride	32	17.4	
pH			6-9

* Total Toxic Organics is explained in Section 6.

** The Agency is not providing 30-day average limits for total toxic organics for reasons explained in Section 8.

TABLE 5: PSES and PSNS REGULATIONS FOR SEMICONDUCTORS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>
Total Toxic Organics *	1.37	**

TABLE 6: BPT REGULATIONS FOR ELECTRONIC CRYSTALS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>	<u>pH Range</u>
Total Toxic Organics *	1.37	**	
Fluoride	32	17.4	
Arsenic ***	2.09	0.83	
TSS	61	23	
pH			6-9

TABLE 7: BAT REGULATIONS FOR ELECTRONIC CRYSTALS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>
Total Toxic Organics *	1.37	**
Fluoride	32	17.4
Arsenic ***	2.09	0.83

* Total Toxic Organics is explained in Section 6.

** The Agency is not providing 30-day average limits for total toxic organics for reasons explained in Section 8.

*** The arsenic limitation applies only to discharges from gallium or indium arsenide crystals manufacturing operations.

TABLE 8. BCT REGULATIONS FOR
ELECTRONIC CRYSTALS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>	<u>pH Range</u>
TSS	61.0	23	
pH			6-9

TABLE 9. NSPS REGULATIONS FOR
ELECTRONIC CRYSTALS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>	<u>pH Range</u>
Total Toxic Organics *	1.37	**	
Fluoride	32	17.4	
Arsenic ***	2.09	0.83	
TSS	61.0	23	
pH			6-9

TABLE 10: PSNS AND PSES REGULATIONS FOR
ELECTRONIC CRYSTALS

<u>Pollutant</u>	<u>24-hour Maximum (mg/l)</u>	<u>30-day Average (mg/l)</u>
Total Toxic Organics *	1.37	**
Arsenic ***	2.09	0.83

* Total Toxic Organics is explained in Section 6.

** The Agency is not providing 30-day average limits for total toxic organics for reasons explained in Section 8.

*** The arsenic limitation applies only to discharges from gallium or indium arsenide crystals manufacturing operations.

SECTION 1

INTRODUCTION

The purpose of this document is to present the findings of the EPA study of the Electrical and Electronic Components (E&EC) Point Source Category, Phase I. The document (1) explains which segments of the industry are regulated and which are not; (2) discusses the reasons; and (3) explains how the actual limitations were developed. Section 1 describes the organization of the document and reviews the sources of industry data that were used to provide technical background for the limitations.

1.1 ORGANIZATION AND CONTENT OF THIS DOCUMENT

Industry data are used throughout this report in support of regulating subcategories or excluding subcategories from regulation under Paragraph 8 of the NRDC Consent Decree. Telephone contacts, the literature, and plant visits provided the information used to subcategorize the industry in Section 3. These data were also considered in subcategorizing the industry in Section 4, Description of the Industry.

Water use and wastewater characteristics in each subcategory are described in Section 5 in terms of flow, pollutant concentration, and load. Subcategories to be regulated, or excluded, are found in Section 6. The discussion in that section identifies and describes the pollutants to be regulated or presents the rationale for subcategory exclusion. Section 7 describes the technology options available. The regulatory limits and the bases for these limitations are presented in Section 8. Section 9 estimates the capital and operating costs for the treatment technologies used as the basis for limitations.

1.2 SOURCES OF INDUSTRY DATA

Data on the E&EC category were gathered from literature studies, contacts with EPA regional offices, from plant surveys and evaluations, and through contacting waste treatment equipment manufacturers. These data sources are discussed below.

Published literature in the form of books, reports, papers, periodicals, promotional materials, Dunn and Bradstreet surveys, and Department of Commerce Statistics was examined; the most informative sources are listed in Section 11, References. The researched material included product descriptions and uses, manufacturing

processes, raw materials consumed, waste treatment technology, and the general characteristics of plants in the E&EC category, including number of plants, employment levels, and production.

All 10 EPA offices were telephoned for assistance in identifying E&EC plants in their respective regions.

Three types of data collection were used to supplement available information pertaining to facilities in the E&EC category. First, more than 250 plants were contacted by phone or letter to obtain basic information regarding products, manufacturing processes, wastewater generation, and waste treatment. Second, based on this information, 78 plants were visited to view their operations and discuss their products, manufacturing processes, water use, and wastewater treatment. Third, 38 plants were selected for sampling visits to determine the pollutant characteristics of their wastewater.

The sampling program at each plant consisted of up to three days of sampling. Prior to any sampling visit, all available data, such as layouts and diagrams of the selected plant's production processes and waste treatment facilities, were reviewed. In most cases, a visit to the plant was made prior to the actual sampling visit to finalize the sampling approach.

Representative sample points were then selected. Finally, before the visit was conducted, a detailed sampling plan showing the selected sample points and all pertinent sample data to be obtained was presented and reviewed.

To more completely characterize each product by the number of producers, production levels, production processes, in-plant controls, waste sources and volumes, waste treatment, and waste disposition, a major survey of each industry was necessary.

Following literature surveys, telephone contacts, and plant visits, questionnaires for obtaining the above information were prepared for each product. After review and comments by selected industry personnel, the questionnaires were mailed to all known product manufacturers. The results of these surveys provided the major sources of industrial data presented in this document.

Various manufacturers of wastewater treatment equipment were contacted by phone or were visited to obtain cost and performance data on specific technologies. Information collected was based both on manufacturers' research and on actual operation.

SECTION 2

LEGAL BACKGROUND

2.1 PURPOSE AND AUTHORITY

The Federal Water Pollution Control Act Amendments of 1972 established a comprehensive program to "restore and maintain the chemical, physical, and biological integrity of the Nation's waters," Section 101(a). Section 301(b)(1)(A) set a deadline of July 1, 1977, for existing industrial dischargers to achieve "effluent limitations requiring the application of the best practicable control technology currently available" (BPT). Section 301(b)(2)(A) set a deadline of July 1, 1983, for these dischargers to achieve "effluent limitations requiring the application of the best available technology economically achievable (BAT), which will result in reasonable further progress toward the national goal of eliminating the discharge of all pollutants."

Section 306 required that new industrial direct dischargers comply with new source performance standards (NSPS), based on best available demonstrated technology. Sections 307(b) and (c) of the Act required pretreatment standards for new and existing dischargers to publicly owned treatment works (POTW). While the requirements for direct dischargers were to be incorporated into National Pollutants Discharge Elimination System (NPDES) permits issued under Section 402, the Act made pretreatment standards enforceable directly against dischargers to POTWs (indirect dischargers).

Section 402(a)(1) of the 1972 Act does allow requirements to be set case-by-case. However, Congress intended control requirements to be based, for the most part, on regulations promulgated by the Administrator of EPA. Section 304(b) required regulations that establish effluent limitations reflecting the ability of BPT and BAT to reduce effluent discharge. Sections 304(c) and 306 of the Act required promulgation of regulations for NSPS. Sections 304(f), 307(b), and 307(c) required regulations for pretreatment standards. In addition to these regulations for designated industry categories, Section 307(a) required the Administrator to promulgate effluent standards applicable to all dischargers of toxic pollutants.

Finally, Section 501(a) authorized the Administrator to prescribe any additional regulations "necessary to carry out his functions" under the Act.

The EPA was unable to promulgate many of these regulations by the deadlines contained in the Act, and as a result, in 1976, EPA was sued by several environmental groups. In settling this lawsuit, EPA and the plaintiffs executed a "Settlement Agreement" which was approved by the Court. This agreement required EPA to develop a program and meet a schedule for controlling 65 "priority" pollutants and classes of pollutants. In carrying out this program, EPA must promulgate BAT effluent limitations guidelines, pretreatment standards, and new source performance standards for 21 major industries. (See Natural Resources Defense Council, Inc. v. Train, 8 ERC 2120 (D.D.C. 1976), modified, 12 ERC 1833(D.D.C. 1979), modified by Order dated October 26, 1982.

Several of the basic elements of the Settlement Agreement program were incorporated into the Clean Water Act of 1977. This law made several important changes in the Federal water pollution control program. Sections 301(b)(2)(A) and 301(b)(2)(C) of the Act now set July 1, 1984, as the deadline for industries to achieve effluent limitations requiring application of BAT for "toxic" pollutants. "Toxic" pollutants here included the 65 "priority" pollutants and classes of pollutants that Congress declared "toxic" under Section 307(a) of the Act.

Likewise, EPA's programs for new source performance standards and pretreatment standards are now aimed principally at controlling toxic pollutants. To strengthen the toxics control program, Section 304(e) of the Act authorizes the Administrator to prescribe "best management practices" (BMPs). These BMPs are to prevent the release of toxic and hazardous pollutants from: (1) plant site runoff, (2) spillage or leaks, (3) sludge or waste disposal, and (4) drainage from raw material storage if any of these events are associated with, or ancillary to, the manufacturing or treatment process.

In keeping with its emphasis on toxic pollutants, the Clean Water Act of 1977 also revises the control program for non-toxic pollutants. For "conventional" pollutants identified under Section 304(a)(4) (including biochemical oxygen demand, suspended solids, fecal coliform, and pH), the new Section 301(b)(2)(E) requires "effluent limitations requiring the application of the best conventional pollutant control technology" (BCT) -- instead of BAT -- to be achieved by July 1, 1984. The factors considered in assessing BCT for an industry include the relationship between the cost of attaining a reduction in effluents and the effluent reduction benefits attained, and a comparison of the cost and level of reduction of such pollutants by publicly owned treatment works and industrial sources. For those pollutants that are neither "toxic" pollutants nor "conventional" pollutants, Sections 301(b)(2)(A) and (b)(2)(F) require achievement of BAT effluent

limitations within three years after their establishment or July 1, 1984, whichever is later, but not later than July 1, 1987.

The purpose of this proposed regulation is to establish BPT, BAT, and BCT effluent limitations and NSPS, PSES, and PSNS for the Electrical and Electronic Components Point Source Category.

2.2 GENERAL CRITERIA FOR EFFLUENT LIMITATIONS

2.2.1 BPT Effluent Limitations

The factors considered in defining best practicable control technology currently available (BPT) include: (1) the total cost of applying the technology relative to the effluent reductions that result, (2) the age of equipment and facilities involved, (3) the processes used, (4) engineering aspects of the control technology, (5) process changes, (6) non-water quality environmental impacts (including energy requirements), (7) and other factors as the Administrator considers appropriate. In general, the BPT level represents the average of the best existing performances of plants within the industry of various ages, sizes, processes, or other common characteristics. When existing performance is uniformly inadequate, BPT may be transferred from a different subcategory or category. BPT focuses on end-of-process treatment rather than process changes or internal controls, except when these technologies are common industry practice.

The cost/benefit inquiry for BPT is a limited balancing, committed to EPA's discretion, which does not require the Agency to quantify benefits in monetary terms. See, e.g., American Iron and Steel Institute v. EPA, 526 F.2d 1027 (3rd Cir. 1975). In balancing costs against the benefits of effluent reduction, EPA considers the volume and nature of existing discharges, the volume and nature of discharges expected after application of BPT, the general environmental effects of the pollutants, and the cost and economic impacts of the required level of pollution control. The Act does not require or permit consideration of water quality problems attributable to particular point sources or water quality improvements in particular bodies of water. See Weyerhaeuser Company v. Costle, 590 F.2d 1011 (D.C.Cir. 1978); Appalachian Power Company et al. v. U.S.E.P.A. (D.C. Cir., Feb. 8, 1972).

2.2.2 BAT Effluent Limitations

The factors considered in defining best available technology economically achievable (BAT) include the age of equipment and

facilities involved, the processes used, process changes, and engineering aspects of the technology process changes, non-water quality environmental impacts (including energy requirements) and the costs of applying such technology [(Section 304(b) (2)(B)]. At a minimum, the BAT level represents the best economically achievable performance of plants of various ages, sizes, processes, or other shared characteristics. As with BPT, uniformly inadequate performance within a category or subcategory may require transfer of BAT from a different subcategory or category. Unlike BPT, however, BAT may include process changes or internal controls, even when these technologies are not common industry practice.

The statutory assessment of BAT "considers" costs, but does not require a balancing of costs against effluent reduction benefits (see Weyerhaeuser v. Costle, supra). In developing BAT, however, EPA has given substantial weight to the reasonableness of costs. The Agency has considered the volume and nature of discharges, the volume and nature of discharges expected after application of BAT, the general environmental effects of the pollutants, and the costs and economic impacts of the required pollution control levels. Despite this expanded consideration of costs, the primary factor for determining BAT is the effluent reduction capability of the control technology. The Clean Water Act of 1977 establishes the achievement of BAT as the principal national means of controlling toxic water pollution from direct discharging plants.

2.2.3 BCT Effluent Limitations

The 1977 Amendments added Section 301(b)(2)(E) to the Act establishing "best conventional pollutant control technology" (BCT) for discharges of conventional pollutants from existing industrial point sources. Conventional pollutants are those defined in Section 304(a)(4) [biological oxygen demanding pollutants (BOD), total suspended solids (TSS), fecal coliform, and pH], and any additional pollutants defined by the Administrator as "conventional" [oil and grease, 44 FR 44501, July 30, 1979].

BCT is not an additional limitation but replaces BAT for the control of conventional pollutants. In addition to other factors specified in Section 304(b)(4)(B), the Act requires that BCT limitations be assessed in light of a two-part "cost reasonableness" test. American Paper Institute v. EPA, 660 F.2d 954 (4th Cir. 1981). The first test compares the costs for private industry to reduce its conventional pollutants with the costs to publicly owned treatment works for similar levels of reduction in their discharge of these pollutants. The second test examines the cost-effectiveness of additional industrial treatment beyond BPT. EPA must find that

limitations are "reasonable" under both tests before establishing them as BCT. In no case may BCT be less stringent than BPT.

EPA published its methodology for carrying out the BCT analysis on August 29, 1979 (44 FR 50732). In the case mentioned above, the Court of Appeals ordered EPA to correct data errors underlying EPA's calculation of the first test, and to apply the second cost test (EPA had argued that a second cost test was not required).

On October 29, 1982, the Agency proposed a revised BCT methodology. See 47 FR 49176. Although the Agency has not yet promulgated its revised BCT cost test methodology, we are promulgating BCT limitations as proposed for the semiconductor and electronic crystal industries. Application of the BCT cost test is not necessary for these industries for reasons presented in Section 8 of this document.

2.2.4 New Source Performance Standards

The basis for new source performance standards (NSPS) under Section 306 of the Act is the best available demonstrated technology. New plants have the opportunity to design the best and most efficient processes and wastewater treatment technologies. Therefore, Congress directed EPA to consider the best demonstrated process changes, in-plant controls, and end-of-process treatment technologies that reduce pollution to the maximum extent feasible.

2.2.5 Pretreatment Standards for Existing Sources

Section 307(b) of the Act requires EPA to promulgate pretreatment standards for existing sources (PSES) which industry must achieve within three years of promulgation. PSES are designed to prevent the discharge of pollutants that pass through, interfere with, or are otherwise incompatible with the operation of POTWs.

The legislative history of the 1977 Act indicates that pretreatment standards are to be technology-based, analogous to the best available technology for removal of toxic pollutants. The General Pretreatment Regulations which serve as the framework for the pretreatment standards are in 40 CFR Part 403, 46 FR 9404 (January 28, 1981).

EPA has generally determined that there is passthrough of pollutants if the percent of pollutants removed by a well-operated POTW achieving secondary treatment is less than the percent removed by the BAT model treatment system. A study of 40 well-operated POTWs with biological treatment and meeting secondary treatment criteria showed that metals are typically removed at rates varying from 20

percent to 70 percent. POTWs with only primary treatment have even lower rates of removal. In contrast, BAT level treatment by the industrial facility can achieve removal in the area of 97 percent or more. Thus, it is evident that metals do pass through POTWs. As for toxic organics, data from the same POTWs illustrate a wide range of removal, from 0 to greater than 99 percent. Overall, POTWs have removal rates of toxic organics which are less effective than BAT.

2.2.6 Pretreatment Standards for New Sources

Section 307(c) of the Act requires EPA to promulgate pretreatment standards for new sources (PSNS) at the same time that it promulgates NSPS. These standards are intended to prevent the discharge of pollutants which pass through, interfere with, or are otherwise incompatible with a POTW. New indirect dischargers, like new direct dischargers, have the opportunity to incorporate the best available demonstrated technologies -- including process changes, in-plant controls, and end-of-process treatment technologies -- and to select plant sites that ensure the treatment system will be adequately installed. Therefore, the Agency sets PSNS after considering the same criteria considered for NSPS. PSNS will have environmental benefits similar to those from NSPS.

SECTION 3

INDUSTRY SUBCATEGORIZATION

This section explains how the E&EC category was developed, discusses the rationale for subcategorization, and finally provides a listing of the E&EC subcategories.

3.1 E&EC CATEGORY DEVELOPMENT

The E&EC category is derived from industries found in the Standard Industrial Classification (SIC) major group 36, Electrical and Electronic Machinery, Equipment, and Supplies. Many of the industries listed under this SIC Code were never evaluated as part of the E&EC category because EPA initially concluded that the wastewater discharges from these industries were primarily associated with the Electroplating or Metal Finishing Category.

3.2 RATIONALE FOR INDUSTRY SUBCATEGORIZATION

After the Agency has obtained analyses of wastewater data and process information from facilities within a category, the Clean Water Act requires EPA to consider a number of factors to determine if subcategorization is appropriate for the purpose of establishing effluent limitations and standards. These factors include: raw materials, final products, manufacturing processes, geographical location, plant size and age, waste-water characteristics, non-water quality environmental impacts, treatment costs, energy costs, and solid waste generation.

A review of each of these factors revealed that product type is the principal factor affecting the wastewater characteristics of plants within the E&EC category. Product type determines both the raw and process material requirements, and the number and type of manufacturing processes used. Plants manufacturing the same product were found to use the same wet processes and produce wastewater with similar characteristics. Other factors affected the wastewater characteristics, but were not adequate in themselves to be used as bases for subcategorization.

3.3 SUBCATEGORY LISTING

Based on product type (discussed above), EPA established the following twenty-one (21) subcategories for the E&EC category:

Semiconductors
Electronic Crystals
Electron Tubes (Phase II)
Phosphorescent Coatings (Phase II)*
Capacitors, Fixed
Capacitors, Fluid Filled
Carbon and Graphite Products
Mica Paper
Incandescent Lamps
Fluorescent Lamps
Fuel Cells
Magnetic Coatings
Resistors
Transformers, Dry
Transformers, Fluid Filled
Insulated Devices, Plastic and Plastic Laminated
Insulated Wire and Cable, Nonferrous
Ferrite Electronic Parts
Motors, Generators, and Alternators
Resistance Heaters
Switchgear

* Phosphorescent coatings named as luminescent materials in Phase II proposal.

SECTION 4

DESCRIPTION OF THE INDUSTRY

This section provides a general description of the subcategories presented in the previous section. It includes a discussion of the number of plants and production capacity, product lines, and manufacturing processes including raw materials used. Industry descriptions for the regulated subcategories (Semiconductors and Electronic Crystals) are presented in considerable detail, while industry descriptions are abbreviated for subcategories which have been excluded or proposed under Phase II.

4.1 SEMICONDUCTORS

4.1.1 Number of Plants and Production Capacity

It is estimated that approximately 257 plants are involved in the production of semiconductor products. This estimate comes from an August 1979 listing of plant locations compiled by the Semiconductor Industry Association. Seventy-seven of the plants are direct dischargers and one hundred and eighty are indirect dischargers. The U.S. Department of Commerce 1977 Census of Manufacturers estimates that 62,000 production employees are engaged in the manufacture of semiconductor products. Plants surveyed or visited during this study employ between 30 and 2500 production employees. The majority of plants employ between 150 and 500 production employees, with a typical plant having about 350 employees. Only 9 of the 52 plants in the data base have more than 500 production employees.

The total number of semiconductor products for the year 1978 was obtained from the Semiconductor Industry Association. During that year, 8.844 billion units were produced for a total revenue of \$3.123 billion.

4.1.2 Products

Semiconductors are solid state electrical devices which perform a variety of functions in electronic circuits. These functions include information processing and display, power handling, data storage, signal conditioning, and the interconversion between light energy and electrical energy. The semiconductors range from the simple diode, commonly used as an alternating current rectifier, to the integrated circuit which may have the equivalent of 250,000 active components in a 0.635 cm (1/4 inch) square.

Semiconductors are used throughout the electronics industry. The major semiconductor products are:

- o Silicon based integrated circuits which include bipolar, MOS (metal oxide silicon), and digital and analog devices. Integrated circuits are used in a wide variety of commercial and consumer electronic equipment, calculators, electronic games and toys, and medical equipment.
- o Light emitting diodes (LED) which are produced from gallium arsenide and gallium phosphide wafers. These devices are commonly used as information displays in electronic games, watches, and calculators.
- o Diodes and transistors which are produced from silicon or germanium wafers. These devices are used as active components in electronic circuits which rectify, amplify, or condition electrical signals.
- o Liquid crystal display (LCD) devices which are produced from liquid crystals. These devices are primarily used for information displays as an alternative to LEDs.

4.1.3 Manufacturing Processes and Materials

The manufacturing processes and materials used for semiconductor production are described in the following paragraphs. Each type of semiconductor with its associated manufacturing operations is discussed separately because production processes differ depending on the basis material.

Silicon-Based Integrated Circuits -- (Figure 4-1 on page 4-3). These circuits require high purity crystal silicon as a basis material. Most of the companies involved in silicon-based integrated circuit production purchase crystal silicon ingots (cylindrical crystals which can be sliced into wafers), slices, or wafers from outside sources rather than grow their own crystals.

In cases where the ingot is received it is sliced into round wafers approximately 0.76mm (0.030 inches) thick. These slices are then lapped or polished by means of a mechanical grinding machine or are chemically etched to provide a smooth surface and remove surface oxides and contaminants. Commonly used etch solutions are hydrofluoric acid or hydrofluoric-nitric acid mixtures. The presence of hydrofluoric acid is generally necessary because of the solubility characteristics of silicon and silicon oxide. Other acids such as sulfuric or nitric may be

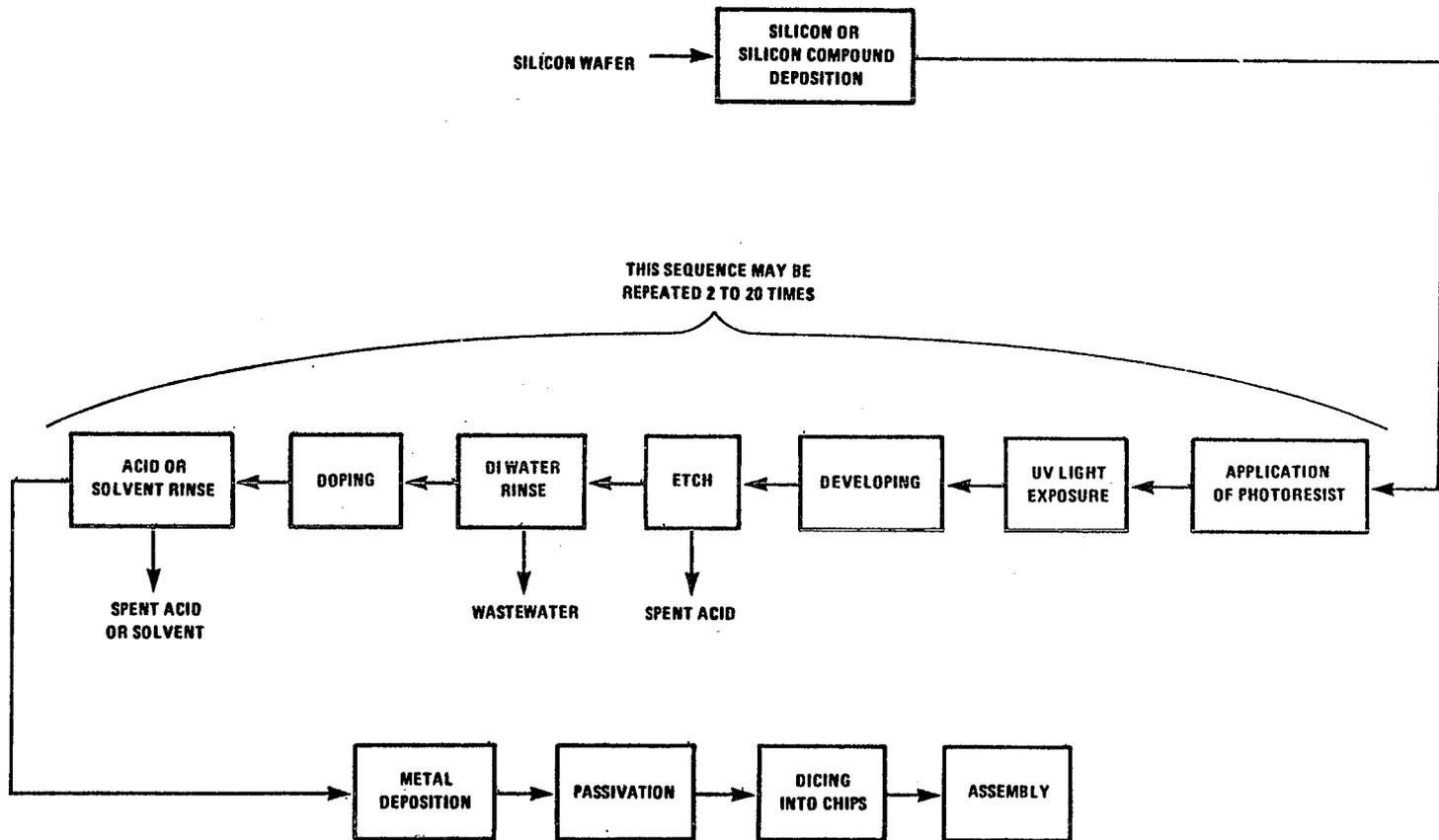


FIGURE 4-1. SILICON INTEGRATED CIRCUIT PRODUCTION

used depending on the nature of the material to be removed. Wastewater results from cooling the diamond tipped saws used for slicing, from spent etch solution, and from deionized (DI) water rinses following chemical etching and mechanical finishing operations.

The next step in the process depends on the type of integrated circuit device being produced, but commonly involves the deposition or growth of a layer or layers of silicon dioxide, silicon nitride, or epitaxial silicon. For example, a silicon dioxide layer is commonly applied to bipolar devices, and an initial layer of silicon dioxide with the subsequent deposition of a silicon nitride layer is commonly applied to MOS devices.

The wafer is then coated with a photoresist, a photosensitive emulsion. The wafer is next exposed to ultraviolet light using metal or glass photomasks that allow the light to strike only selected areas. After exposure to ultraviolet light, unexposed resist is removed from the wafer, usually in a DI water rinse. This allows selective etching of the wafer. The wafer is then visually inspected under a microscope and etched in a solution containing hydrofluoric acid (HF). The etchant produces depressions, called holes or windows, where the diffusion of dopants later occurs. Dopants are impurities such as boron, phosphorus and other specific metals. These impurities eventually form circuits through which electrical impulses can be transmitted. The wafer is then rinsed in an acid or solvent solution to remove the remainder of the hardened photoresist material.

Diffusion of dopants is generally a vapor phase process in which the dopant, in the form of a gas, is injected into a furnace containing the wafers. Gaseous phosphine and boron trifluoride are common sources for phosphorus and boron dopants, respectively. The gaseous compound breaks down into elemental phosphorus or boron on the hot wafer surface. Continued heating of the wafer allows diffusion of the dopant into the surface through the windows at controlled depths to form the electrical pathways within the wafer. Solid forms of the dopant may also be used. For example, boron oxide wafers can be introduced into the furnace in close proximity to the silicon wafers. The boron oxide sublimates and deposits boron on the surface of the wafer by condensation and then diffuses into the wafer upon continued heating.

Then a second oxide layer is grown on the wafer, and the process is repeated. This photolithographic-etching-diffusion-oxide process sequence may occur a number of times depending upon the application of the semiconductor.

During the photolithographic-etching-diffusion-oxide processes, the wafer may be cleaned many times in mild acid or alkali

solutions followed by DI water rinses and solvent drying with acetone or isopropyl alcohol. This is necessary to maintain wafer cleanliness.

After the diffusion processes are completed, a layer of metal is deposited onto the surface of the wafer to provide contact points for final assembly. The metals used for this purpose include aluminum, copper, chromium, gold, nickel, platinum, and silver. The processes associated with the application of the metal layer are covered by the electroplating or metal finishing effluent limitations and standards. One of the following three processes is used to deposit this metal layer:

- o Sputtering --
In this process the source metal and the target wafer are electrically charged, as the cathode and anode, respectively, in a partially evacuated chamber. The electric field ionizes the gas in the chamber and these ions bombard the source metal cathode, ejecting metal which deposits on the wafer surface.
- o Vacuum Deposition --
In this process the source metal is heated in a high vacuum chamber by resistance or electron beam heating to the vaporization temperature. The vaporized metal condenses on the surface of the silicon wafer.
- o Electroplating --
In this process the source metal is electrochemically deposited on the target wafer by immersion in an electroplating solution and the application of an electrical current.

Finally, the wafer receives a protective oxide layer (passivation) coating before being back lapped to produce a wafer of the desired thickness. Then the individual chips are diced from the wafer and are assembled in lead frames for use. Many companies involved in semiconductor production send completed wafers to overseas facilities where dicing and assembly operations are less costly as a result of the amount of hand labor necessary to inspect and assemble finished products.

Light Emitting Diodes (LEDs) -- LEDs are produced from single crystal gallium arsenide or gallium phosphide wafers. These wafers are purchased from crystal growers and upon receipt are placed in a furnace where a silicon nitride layer is grown on the wafer. The wafer then receives a thin layer of photoresist, is exposed through a photomask, and is developed with a xylene-based developer. Following this, the wafer is etched using hydrofluoric acid or a plasma-gaseous-etch process, rinsed in DI

water, and then stripped of resist. The wafer is again rinsed in DI water before a dopant is diffused into the surface of the wafer. A metal oxide covering is applied next, and then a photoresist is applied. The wafer is then masked, etched in a solution of aurostrip (a cyanide-containing chemical commonly used in gold stripping), and rinsed in DI water. The desired thickness is produced by backlapping and a layer of metal, usually gold, is sputtered onto the back of the wafer to provide electrical contacts. Testing and assembly complete the production process.

Diodes and Transistors -- Diodes and transistors are produced from single crystal silicon or germanium wafers. These devices, called discrete devices, are manufactured on a large scale, and their use is mainly in older or less sophisticated equipment designs, although discrete devices still play an important role in high power switching and amplification.

The single crystal wafer is cleaned in an acid or alkali solution, rinsed in DI water, and coated with a layer of photoresist. The wafer is then exposed and etched in a hydrofluoric acid solution. This is followed by rinsing in DI water, drying, and doping in diffusion furnaces where boron or phosphorus are diffused into specific areas on the surface of the wafer. The wafers are then diced into individual chips and sent to the assembly area. In the assembly area electrical contacts are attached to the appropriate areas and the device is sealed in rubber, glass, plastic, or ceramic material. Extra wires are attached and the device is inspected and prepared for shipment.

Liquid Crystal Display (LCD) Production -- A typical LCD production line begins with optically flat glass that is cut into four-inch squares. The squares are then cleaned in a solution containing ammonium hydroxide, immersed in a mild alkaline stripping solution, and rinsed in DI water. The plates are spun dry and sent to the photolithography area for further processing.

In the photolithographic process a photoresist mask is applied with a roller, and the square is exposed and developed. This square then goes through deionized water rinses and is dried, inspected, etched in an acid solution, and rinsed in DI water. A solvent drying step is followed by another alkaline stripping solution. The square then goes through DI water rinses, is spun dry, and is inspected.

The next step of the LCD production process is passivation. A silicon oxide layer is deposited on the glass by using liquid silicon dioxide, or by using silane and oxygen gas with phosphine gas as a dopant. This layer is used to keep harmful sodium ions

on the glass away from the surface where they could alter the electronic characteristics of the device. Several production steps may occur here if it is necessary to rework the piece. These include immersion in an ammonium bifluoride bath to strip silicon oxide from a defective piece followed by DI water rinses and a spin dry step. The glass is then returned to the passivation area for reprocessing.

After passivation, the glass is screen printed with devitrified liquid glass in a matrix. Subsequent baking causes the devitrified glass to become vitrified, and the squares are cut into the patterns outlined by the vitrified glass boundaries. The saws used to cut the glass employ contact cooling water which is filtered and discharged to the waste treatment system.

The glass is then cleaned in an alkaline solution and rinsed in deionized water. Following inspection, a layer of silicon oxide is evaporated onto the surface to provide alignment for the liquid crystal. The two mirror-image pieces of glass are aligned and heated in a furnace, bonding the vitrified glass and creating a space between the two pieces of glass. This glass assembly is immersed in the liquid crystal solution in a vacuum chamber, air is evacuated, and the liquid crystal is forced into the space between the glass pieces. The glass is then sealed with epoxy, vapor-degreased in a solvent, shaped on a diamond wheel, inspected, and sent to assembly.

4.2 ELECTRONIC CRYSTALS

4.2.1 Number of Plants

Table 4-1 on page 4-8 presents an estimate of the number of producers of each type of crystal. Of plants manufacturing crystals at seventy sites, six are direct dischargers and sixty-four are indirect dischargers. The last fifteen years have seen an extremely rapid evolution of electronic technology. A major part of that evolution has been the development of single crystals with unique structural and electronic properties which serve as essential parts of most microelectronic devices. The production and use of gallium based crystals are expected to have a particularly rapid growth over the next decade. Gallium based crystals have certain advantages over silicon based crystals for semiconductor applications with respect to circuit speed, power consumption, and higher temperature capabilities. Consequently the crystals industry has served an expanding market with an ever-increasing list of products. Companies comprising the industry include not only those long-established, but also a large proportion founded comparatively recently by entrepreneurs. Of this latter group some companies have grown considerably, while others are very small. This growth in the number of companies is expected to continue.

TABLE 4-1

PROFILE OF ELECTRONIC CRYSTALS INDUSTRY

Product	Estimated No. of Producers(1)	Product	Estimated No. of Producers(1)
Piezoelectric Crystals:		Semi-conducting Crystals:	
Quartz	40	Silicon	8
Ceramics(2)	8	Gallium arsenide	8
YIG	3	Gallium phosphide	8
YAG	2	Sapphire	1
Lithium Niobate	3	GGG	3
Liquid Crystals	2	Indium arsenide	1
		Indium antimonide	1
		Bismuth telluride	1

(1) Several producers manufacture more than one product.

(2) Ceramics include lead zirconate, ammonium hydrogen phosphate, potassium hydrogen phosphate and lead zirconium titanate.

4.2.2 Products

Based on their properties and thus their uses in the industry, electronic crystals can be divided into three types: piezoelectric, semiconducting, and liquid crystals.

Piezoelectric Crystals -- Piezoelectric crystals are transducers which interconvert electrical voltage and mechanical force. There are three principal types: quartz, ceramic, and yttrium-iron-garnet (YIG), and some other less common types.

Quartz crystals are the most widely used of the piezoelectric crystals, with applications as timing devices in watches, clocks, and record players; frequency controllers, modulators, and demodulators in oscillators; and filters. Some quartz is mined, but the main supply comes from synthesized material produced by about forty companies in the United States.

Ceramic crystals are basically fired mixtures of the oxides of lead, zirconium, and titanium. They are used in transducers, oscillators, ultrasonic cleaners, phonograph cartridges, gas igniters, audible alarms, keyboard switches, and medical electronic equipment.

YIG crystals are made by the slow crystal growth of a melt of yttrium oxide, iron oxide, and lead oxide. Their primary use is in the microwave industry for low frequency applications as in sonar. Their incorporation into microwave circuits makes wide-band tuning possible.

Other potentially useful piezoelectric crystals being developed or manufactured on a small scale include lithium niobate, bismuth germanium oxide, and yttrium-aluminum-garnet (YAG).

Semiconducting Crystals -- Semiconducting crystals have properties intermediate between a conductor and an insulator, thus allowing for a wide range of applications in the field of microelectronics. In conductors, current is carried by electrons that travel freely throughout the atomic lattice of the substance. In insulators the electrons are tightly bound and are therefore unavailable to serve as carriers of electric current. Semiconductors do not ordinarily contain free charge carriers but generate them with a modest expenditure of energy.

Silicon crystals are widely used in the manufacture of micro electronic chips: transistors, diodes, rectifiers, other circuit elements, and solar cells. Crystals of pure silicon are poor conductors of electricity. In order to make them better conductors, controlled amounts of impurity atoms are introduced into the crystal by a process called doping.

When silicon is doped with an element whose atoms contain more or fewer valence electrons than silicon, free electrons or electron "holes" are thus available to be mobilized when a voltage is applied to the crystal. Phosphorus and boron are common dopants used in silicon crystals.

Gallium arsenide and gallium phosphide crystals were developed under the need for a transistor material with good high temperature properties. These crystals exhibit low field electron mobility, and are therefore useful at high frequencies, in such devices as the field effect transistor (FET). The technology of manufacturing high performance gallium arsenide FET's is maturing at a rapid rate and the devices are experiencing a greatly expanding role in oscillators, power amplifiers, and low noise/high gain applications.

Most gallium arsenide/phosphide is presently being used for production of light emitting diodes (LEDs) which can convert electric energy into visible electromagnetic radiation. The interconversion of light energy and voltage in gallium arsenide is reversible. Hence this material is also undergoing intensive development as a solar cell, in which sunlight is converted directly to electricity.

Indium arsenide and indium antimonide crystals, formed by direct combination of the elements, are used as components of power measuring devices. These crystals are uniquely suited to this function because they demonstrate a phenomenon known as the Hall Effect, the development of a transverse electric field in a current-carrying conductor placed in a magnetic field.

Bismuth telluride crystals demonstrate a phenomenon known as thermoelectric cooling because of the Peltier Effect. When a current passes across a junction of dissimilar metals, one side is cooled and the other side heated. If the cold side of the junction is attached to a heat source, heat will be carried away to a place where it can be conveniently dissipated. Devices utilizing this effect are used to cool small components of electrical circuits.

Sapphire crystals are used by the semiconductor industry as single crystal wafers which act as inactive substrates for an epitaxial film of silicon, that is, substrates upon which a thin layer of silicon is deposited in a single-crystal configuration. This is referred to as silicon on sapphire (SOS). In addition to being a dielectric material, single crystal sapphire exhibits a combination of optical and physical properties which make it ideal for a variety of demanding optical applications. Sapphire, the hardest of the oxide crystals, maintains its strength at high temperatures, has good thermal and excellent electrical

properties and is chemically inert. Therefore, it can be used in hostile environments when optical transmission ranging from vacuum ultraviolet to near infrared is required. Sapphire crystals have found application in semiconductor substrates, infrared detector cell windows, UV windows and optics, high power laser optics, and ultracentrifuge cell windows.

Gallium Gadolinium Garnet (GGG) is the most suitable substrate for magnetic garnet films because of its excellent chemical, mechanical, and thermal stability, nearly perfect material and surface quality, crystalline structure, and the commercial availability of large diameter substrates. GGG is the standard substrate material used for epitaxial growth of single crystal iron garnet films which are used in magnetic bubble domain technology.

Liquid Crystals -- Liquid crystals are organic compounds or mixtures of two or more organic compounds which exhibit properties of fluidity and molecular order simultaneously over a small temperature range. An electric field can disrupt the orderly arrangement of liquid crystal molecules, changing the refractive properties. This darkens the liquid enough to form visible characters in a display assembly, even though no light is generated. This effect is achieved by application of a voltage and does not require a current flow. Therefore minimal use of power is required, allowing the display in battery operated devices to be activated continuously. Liquid crystals are used in liquid crystal display (LCD) devices for wrist watches, calculators and other consumer products requiring a low power display.

4.2.3 Manufacturing Processes and Materials

Piezoelectric Crystals -- The following is a description of the manufacturing processes used for growth and fabrication of the three major piezoelectric crystal types: quartz, ceramic, and yttrium-iron-garnet (YIG).

Quartz Crystals:

The growth of quartz crystals is a hydrothermal process carried out in an autoclave under high temperature and pressure. The vessel is typically filled to 80 percent of the free volume with a solution of sodium hydroxide or sodium carbonate. Particles of -quartz nutrient are placed in the lower portion of the vessel where they are dissolved. The quartz is then transferred by convection currents through the solution and deposited on seed crystals which are suspended in the upper portion of the vessel. Seeds are thin wafers or spears of quartz about six inches long. A vessel normally contains 20 seeds. Nutrient quartz will

dissolve and deposit onto the seed crystals because a small temperature gradient exists between the lower and upper portion of the autoclave, promoting the migration of quartz to the upper portion of the vessel. Upon completion of the growth cycle (45 to 60 days), crystals are removed and cleaned for the fabrication process.

The quartz crystals are cut or sliced using diamond blade saws or slurry saws. Diamond blade saws are used when one wafer at a time is cut. Slurry saws are utilized in mass production lines for cutting many wafers at a time. The crystal wafers are then lapped to the desired thickness. After lapping, the crystal is usually etched with hydrofluoric acid or ammonium bifluoride and subsequently rinsed with water. Crystal edges are then beveled using either a dry grinding grit or a water slurry. Following this, metals are deposited on the crystal by vacuum deposition. The crystal wafers are mounted on a masking plate and placed in an evacuated bell jar. Metal strips in the jar are vaporized, coating the unmasked area of the wafer. The metal coating (gold, silver, or aluminum are often used) functions as the crystal's conducting base. The metal coating operation is covered by regulations for the Metal Finishing Category. During fine tune deposition, the crystal is allowed to resonate at a specified frequency and another thin layer of metal is deposited on it. Wire leads are attached to the crystal and it is sealed in a nitrogen atmosphere. At this point the crystal is ready for sale or insertion into an electronic circuit. Figure 4-2 on page 4-13 presents a diagram of the process indicating major waste generating operations.

Ceramic Crystals:

Ceramic crystal production begins by mixing lead oxide, zirconium oxide and titanium oxide powders plus small amounts of dopants to achieve desired specifications in the final product. The powders are mixed with water to obtain uniform blending, then filtration takes place and the waste slurry is sent to disposal. This mixture is roasted, ground wet, and blended with a binder (polyvinyl alcohol) in a tank called a blundger. The mixture is then spray dried, pressed, and fired to drive off the binder, which is not recovered. Formed crystals are enclosed in alumina and refired. After this final firing crystals are polished, lapped, and sliced as in quartz production. Electrodes, usually made of silver, are then attached to the crystals. Approximately ten percent of the crystals have electrodes deposited by electroless nickel plating. This plating operation is covered by regulations for the Metal Finishing Category. Poling, the final process step, gives the crystal its piezoelectric properties. This step is performed with the crystal immersed in a mineral oil bath. Some companies sell the used mineral oil to reclaimers. After poling the crystal is ready for sale and use. Ceramic crystal production is very small.

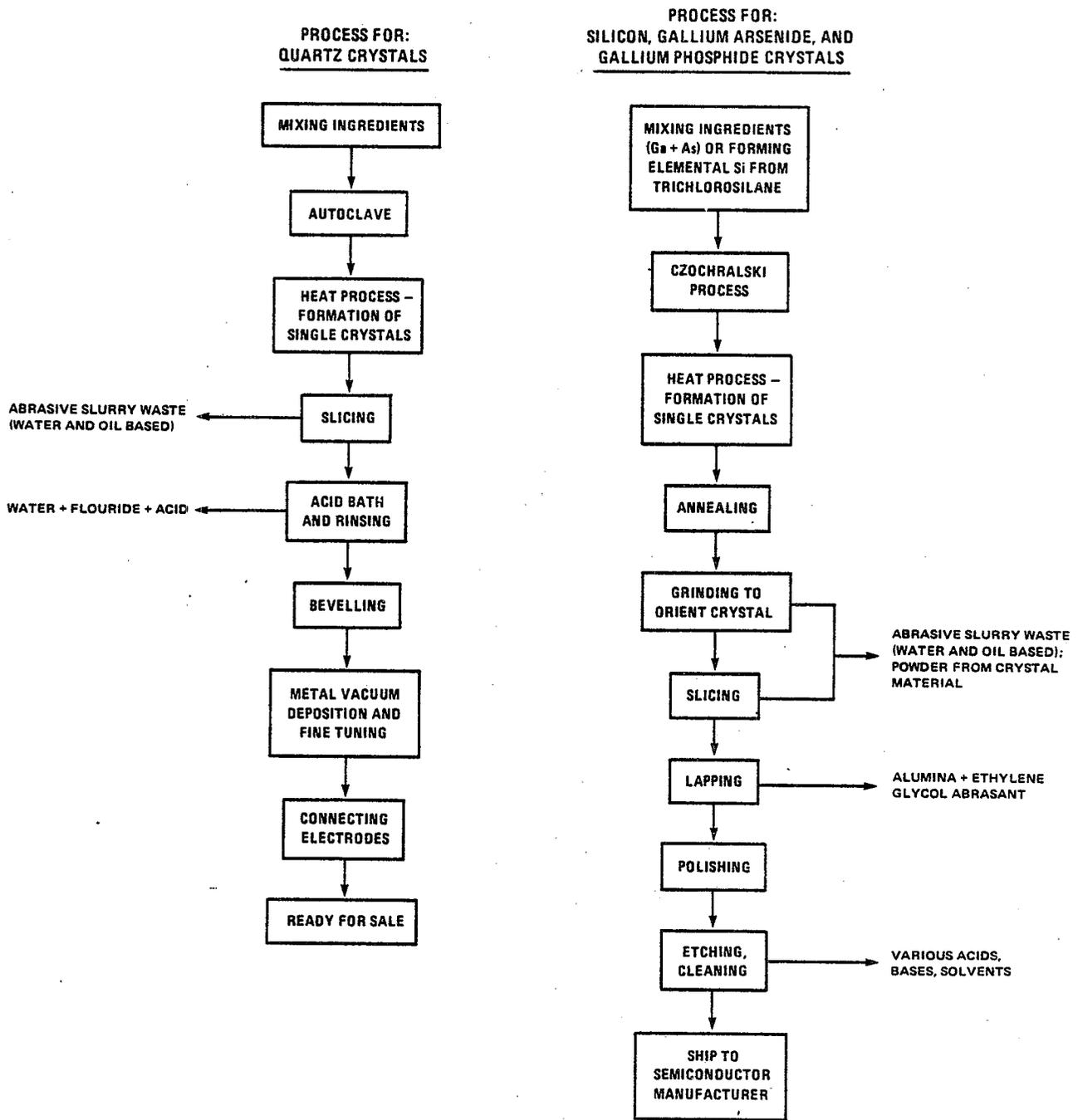


FIGURE 4-2. BASIC MANUFACTURING PROCESSES FOR ELECTRONIC CRYSTALS

Yttrium-Iron-Garnet (YIG) Crystals:

The production of YIG crystals involves the melting of metal compounds to form large single crystals which are processed to yield minute YIG spheres for use in microwave devices. Yttrium oxide, iron oxide and lead oxide powders are mixed, placed in a platinum crucible and melted in a furnace. After the melt equilibrates at this temperature the furnace is cooled, the slag is poured off, leaving the YIG crystals attached to the crucible. This growth process takes approximately 28 days. The crucible is soaked in hydrochloric and nitric acid to remove the crystals which are then sliced by a diamond blade saw to form cubes 0.04 inches on a side. These cubes are placed in a rounding machine, and the rounding process is followed by polishing to obtain perfectly spherical crystals for use in a microwave device.

The production of YIG and ceramic crystals with piezoelectric properties constitutes a minor portion of the piezoelectric crystal industry. The entire YIG production for the USA is less than fifteen pounds per year.

Semiconducting Crystals -- Several methods are currently in use for the production of semiconducting single crystals. An important method, the Czochralski, functions by lowering a seed crystal (a small single crystal) into a molten pool of the crystal material and raising the seed slowly (over a period of days) with constant slow rotation. Because the temperature of the melt is just above the melting point, material solidifies onto the seed crystal, maintaining the same crystal lattice. Crystals up to 6 inches in diameter and 4 feet long can be grown by this method. The Czochralski method is used to grow silicon, sapphire, GGG, and gallium arsenide.

Another method, called the Chalmers method, is used by some manufacturers to grow gallium arsenide crystals. If the molten material is contained in a horizontal boat and cooled slowly from one end, a solid/liquid interface will pass through the melt. Under controlled conditions or with the use of a seed crystal the solid will form as a single crystal.

Silicon Crystals:

The raw material used to produce silicon crystals is polycrystalline silicon. Reduction of purified trichlorosilane with hydrogen is the usual method for producing the high purity polycrystalline ("poly") silicon. Single crystals of silicon are then grown by the Czochralski method, the most common crystal growing technique for semiconductor crystals.

After a crystal has been grown, the outside diameter is ground to produce a crystalline rod of constant diameter. The ends are cut off and used to evaluate the quality of the crystal. At the same time, its orientation is determined and a flat is ground the length of the rod to fix its position. Rods are then sliced into wafers. Silicon dust and cutting oils mixed with water are waste products of the grinding and cutting operations.

Lapping is a machining operation using an alumina and ethylene glycol abrasive medium which produces a flat polished surface and reduces the thickness of the wafers. After lapping, the wafers are polished using a hydrated silica medium. The final cleaning is done with various acids, bases and solvents.

Sapphire and GGG Crystals:

To produce sapphire and gallium gadolinium garnet (GGG) crystals a raw material called crackle, (high purity alumina waste from a European gem crystal growing process) is melted in an iridium crucible. Sapphire is pure alumina. Gadolinium oxide and gallium oxide powders are added to the crucible if GGG is the desired product. These are melted using an induction furnace under a nitrogen atmosphere with a trace of oxygen added. Crystals are pulled from the melt using the Czochralski method.

These crystals are annealed in oxygen-gas furnaces after growth in order to remove internal stress and make the crystalline rods less brittle. Sapphire and GGG rods are ground and sliced using diamond abrasives and a coolant consisting of a mixture of oil and water. Wafers are lapped using a diamond abrasive compound and lubricants, and are polished with a colloidal silica slurry. GGG wafers are coated with a thin film using liquid-phase epitaxy. The film has small permanent magnetic domains, which make it useful for "magnetic bubble" memory devices. The sapphire wafers are coated with a layer of epitaxial silicon to produce the SOS substrates for microelectronic chip manufacture.

Other Semiconducting Crystals:

The formation of gallium arsenide, gallium phosphide, and indium bismuth telluride takes place by a chemical reaction which occurs in an enclosed capsule. When gallium arsenide or phosphide crystals are produced, the gallium, on one side of the capsule, is heated to more than 1200C. The arsenic or phosphorus on the other side of the capsule is heated separately until it vaporizes. The vapor and hot metal react to form a molten compound. (In the case of phosphorus, high pressure is required.) The molten compound can then be crystallized in situ by the Chalmers technique or cooled and crystallized by the Czochralski method. These crystals undergo the fabrication operations mentioned earlier.

To produce indium antimonide, indium arsenide and bismuth telluride, the elements are mixed together, melted to form the compound and frozen into a polycrystalline ingot. These materials are used in a polycrystalline state so no crystal growing step occurs. The ingot is fabricated into wafers by normal machining operations. Because these materials are relatively soft, carbide abrasives with water cooling are sufficient for machining the ingots. The wafers are milled into small pieces and incorporated into electronic components.

Liquid Crystals -- Liquid crystals are produced by organic synthesis. Precursor organic compounds are mixed together and heated until the reaction is complete. The reacted mass is dissolved in an organic solvent such as toluene, and is crystallized and recrystallized several times to obtain a product of the desired purity. Several of these organic compounds are then mixed to form a eutectic mixture with the correct balance of properties for LCD application.

4.3 ELECTRON TUBES (Proposed Under Phase II)

Electron tubes are devices in which electrons or ions are conducted between electrodes through a vacuum or ionized gas within a gas-tight envelope which may be glass, quartz, ceramic, or metal. A large variety of electron tubes are manufactured, including klystrons, magnetrons, cross field amplifiers, and modulators. These products are used in aircraft and missile guidance systems, weather radar, and specialized industrial applications. The Electron Tube subcategory also includes cathode-ray tubes and T.V. picture tubes that transform electrical current into visual images. Cathode-ray tubes generate images by focusing electrons onto a luminescent screen in a pattern controlled by the electrical field applied to the tube. In T.V. picture tubes, a stream of high-velocity electrons scans a luminescent screen. Variations in the electrical impulses applied to the tube cause changes in the intensity of the electron stream and generate the image on the screen.

Processes involved in the manufacture of electron tubes include degreasing of components; application of photoresist, graphite, and phosphors to glass panels; and sometimes electroplating operations including etching and machining. The application of phosphors is unique to T.V. picture tubes and other cathode-ray tubes. The phosphor materials may include sulphides of cadmium and zinc and yttrium and europium oxides. The electroplating operations are covered under the Metal Finishing Category. Raw materials can include copper and steel as basis materials, and copper, nickel, silver, gold, rhodium and chromium to be electroplated. Phosphors, graphite, and protective coatings

containing toluene or silicates and solders of lead oxide may also be used. Process chemicals may include hydrofluoric, hydrochloric, sulfuric, and nitric acids for cleaning and conditioning of metal parts; and solvents such as methylene chloride, trichloroethylene, methanol, acetone, and polyvinyl alcohol.

4.4 PHOSPHORESCENT COATINGS (Proposed Under Phase II As Luminescent Materials)

Phosphorescent coatings are coatings of certain chemicals, such as calcium halophosphate and activated zinc sulfide, which emit light. Phosphorescent coatings are used for a variety of applications, including fluorescent lamps, high-pressure mercury vapor lamps, cathode ray and television tubes, lasers, instrument panels, postage stamps, laundry whiteners, and specialty paints. This study is restricted to those coatings which are applicable to the E&EC category, specifically to those used in fluorescent lamps and television picture tubes. The most important fluorescent lamp coating is calcium halophosphate phosphor. The intermediate powders are calcium phosphate and calcium fluoride. There are three T.V. powders: red, blue, and green. The red phosphor is yttrium oxide activated with europium; the blue phosphor is zinc sulfide activated with silver, and the green phosphor is zinc-cadmium sulfide activated with copper. The major process steps in producing phosphorescent coatings are reacting, milling, and firing the raw materials; recrystallizing raw materials, if necessary; and washing, filtering, and drying the intermediate and final products.

4.5 CAPACITORS, FIXED

The primary function of capacitors is to store electrical energy. Fixed capacitors are layered structures of conductive and dielectric materials. The layering of fixed capacitors is either in the form of rigid plates or in the form of thin sheets of flexible material which are rolled. Typical capacitor applications are energy storage elements, protective devices, filtering devices, and bypass devices. Some typical processes in manufacturing fixed capacitors are anode fabrication, formation reactions, dipping, layering, cathode preparation, welding, and electrical evaluation. All manufacturing processes are covered under the Metal Finishing category by unit operation. Fixed capacitor types are distinguished from each other by type of conducting material, dielectric material, and encapsulating material.

4.6 CAPACITORS, FLUID FILLED

As with fixed capacitors, the primary function of fluid-filled capacitors is to store electrical energy. Wet capacitors

contain a fluid dielectric that separates the anode (in the center of the device) from the cathode (the capacitor shell), which also serves to contain the fluid. Fluid-filled capacitors are used for industrial applications as electrical storage, filtering, and circuit protection devices. Some typical processes in manufacturing fluid-filled capacitors are anode fabrication, formation reactions, metal can preparation, dielectric addition, soldering, and electrical evaluation. All manufacturing processes are covered under the Metal Finishing category by unit operation.

4.7 CARBON AND GRAPHITE PRODUCTS

Carbon and graphite (elemental carbon in amorphous crystalline form) products exhibit unique electrical, thermal, physical, and nuclear properties. The major carbon and graphite product areas are (1) carbon electrodes for aluminum smelting and graphite furnace electrodes for steel production, (2) graphite molds and crucibles for metallurgical applications, (3) graphite anodes for electrolytic cells used for production of such materials as caustic soda, chlorine, potash, and sodium chlorate, (4) non-electrical uses such as structural, refractory, and nuclear applications, (5) carbon and graphite brushes, contacts, and other products for electrical applications, and (6) carbon and graphite specialties such as jigs, fixtures, battery carbons, seals, rings, and rods for electric arc lighting, welding, and metal coating. The production process starts with weighing the required quantities of calcined carbon filler, binders, and additives; combining them as a batch in a heated mixer; and then forming the resulting "green" mixture by compression molding or by extrusion. Green bodies are carefully packed and baked for several weeks. After baking, the items are machined into final shape.

4.8 MICA PAPER

Mica paper is a dielectric (non-conducting) material used in the manufacture of fixed capacitors. Mica paper is manufactured in the following manner: Mica is heated in a kiln and then placed in a grinder where water is added. The resulting slurry is passed to a double screen separator where undersized and oversized particles are separated. The screened slurry flows to a mixing pit and then to a vortex cleaner. The properly-sized slurry is processed in a paper-making machine where excess water is drained or evaporated. The resulting cast sheet of mica paper is fed on a continuous roller to a radiant heat drying oven, where it is cured. From there, the mica paper is wound onto rolls, inspected, and shipped.

4.9 INCANDESCENT LAMPS

An incandescent lamp is an electrical device that emits light. Incandescent tungsten filament lamps operate by passage of an electric current through a conductor (the filament). Heat is produced in this process, and light is emitted if the temperature reaches approximately 500C. Most lamp-making operations are highly automated. The mount machine assembles a glass flare, an exhaust tube, lead-in wires, and molybdenum filament support. A glass bulb is electrostatically coated with silica and the bulb and mount are connected at the exhaust and seal machine. The bulb assembly is annealed, exhausted, filled with an inert gas, and sealed with a natural gas flame. The finishing machine solders the lead wires to the metallic base which is then attached to the bulb assembly by a phenolic resin cement or by a mechanical crimping operation. The finished lamp is aged and tested by illuminating it with excess current for a period of time to stabilize its electrical characteristics.

4.10 FLUORESCENT LAMPS

A fluorescent lamp is an electrical device that emits light by electrical excitation of phosphors that are coated on the inside surface of the lamp. Fluorescent lamps utilize a low pressure mercury arc in argon. Through this process, the lowest excited state of mercury efficiently produces short wave ultraviolet radiation at 2,537 Angstroms. Phosphor materials that are commonly used are calcium halophosphate and magnesium tungstate, which absorb the ultraviolet photons into their crystalline structure and re-emit them as visible white light.

There are two types of fluorescent lamps: hot cathode and cold cathode. Cold cathode manufacture is primarily an electroplating operation. Hot cathode fluorescent lamp manufacturing is a highly automated process. Glass tubing is rinsed with deionized water and gravity-coated with phosphor. Coiled tungsten filaments are assembled together with lead wires, an exhaust tube, a glass flare, and a starting device to produce a mount assembly. The mount assemblies are heat pressed to the two ends of the glass tubing. The glass tubes are exhausted and filled with an inert gas. The lead wires are soldered to the base and the base is attached to the tube ends. The finished lamp receives a silicone coating solution. The lamp is then aged and tested before shipment.

4.11 FUEL CELLS

Fuel cells are electrochemical generators in which the chemical energy from a reaction of air (oxygen) and a conventional fuel is converted directly into electricity. The major fuel cell

products, basically in research and development stages, are: (1) fuel cells for military applications, (2) fuel cells for power supply to vehicles, (3) fuel cells used as high power sources, and (4) low temperature and low pressure fuel cells with carbon electrodes. Some typical processes in the manufacture of fuel cells are extrusion or machining, heat treating, sintering, molding, testing, and assembling. Some typical raw materials are base carbon or graphite, plastics, resins, and Teflon.

4.12 MAGNETIC COATINGS

Magnetic coatings are applied to tapes to allow the recording of information. Magnetic tapes are used primarily for audio, video, computer, and instrument recording. The process begins with milling to create sub-micron magnetic particles. Ferric oxide particles are used almost exclusively with trace additions of other particles or alloys for specific applications. The particles are mixed, through several steps, with a variety of solvents, resins, and other additives. The coating mix is then applied to a flexible tape or film material (for example, cellulose acetate). After the coating mix is applied, particles are magnetically oriented by passing the tape through a magnetic field, and the tape is dried and slit for testing and sale.

4.13 RESISTORS

Resistors are devices commonly used as components of electric circuits to limit current flow or to provide a voltage drop. Resistors are used for television, radios, and other applications. Resistors can be made from various materials. Nickel-chrome alloys, titanium, and other resistive materials can be vacuum-deposited for thin film resistors. Glass resistors are also available for many resistor applications. Two examples of glass resistors are the precision resistor and the low power resistor.

4.14 TRANSFORMERS, DRY

A transformer is a stationary apparatus for converting electrical energy at one alternating voltage into electrical energy at another (usually different) alternating voltage by means of magnetic coupling (without change of frequency). Dry transformers use standard metal working and metal finishing processes (covered by the Metal Finishing category). The main operations in manufacturing a power transformer are the manufacture of a steel core, the winding of coils, and the assembly of the coil/core on some kind of frame or support.

4.15 TRANSFORMERS, FLUID FILLED

Wet transformers perform the same functions as dry transformers, but the former are filled with dielectric fluid. Wet transformers use standard metal working and metal finishing processes which are covered by the Metal Finishing category. The only wet process unique to E&EC are the cleanup and management of residual dielectric fluid. The main operations in manufacturing a power transformer are the manufacture of a steel core, the winding of coils, and the assembly of the coil/core on some kind of frame or support. In the manufacture of wet transformers there is the need for a container or tank to contain the dielectric fluid.

4.16 INSULATED DEVICES, PLASTIC AND PLASTIC LAMINATED

An insulated device is a device that prevents the conductance of electricity (dielectric). Plastic and plastic laminates are types of insulators. Plastics are used in electronic applications as connectors and terminal boards. Other uses include switch bases, gears, cams, lenses, connectors, plugs, stand-off insulators, knobs, handles, and wire ties. Thermo-setting plastics are melted and injected into a closed mold where they solidify. These insulating moldings include polyethylene, polyphenylene, and poly vinyl chloride. Laminates are used in transformer terminal boards, switchgear arc chutes, motor and generator slot wedges, motor bearings, structural support, and spacers. Laminates are made by bonding layers of a reinforcing web. The reinforcements consist of fiberglass, paper, fabrics, or synthetic fibers. The bonding resins are usually phenolic, melamine, polyester, epoxy, and silicone. Laminates are made by impregnating the reinforcing webs in treating towers, partially polymerizing, pressing and finally polymerizing them to shape under heat and pressure. Manufacturing processes associated with these products are studied as part of the Plastics Molding and Forming category.

4.17 INSULATED WIRE AND CABLE, NON-FERROUS

Insulated wires and cables are products containing a conductor covered with a non-conductive material to eliminate shock hazard. The major products in this segment are: (1) insulated non-ferrous wire, (2) auto wiring systems, (3) magnetic wire, (4) bulk cable appliances, and (5) camouflage netting. Typical processes used in the manufacture of insulated wire and cable are drawing, spot welding, heat treating, forming, and assembling. All manufacturing processes are included in the Metal Finishing category. Some of the basis materials are copper, carbon, stainless steel, steel, brass-bronze, and aluminum.

4.18 FERRITE ELECTRONIC PARTS

Ferrite electronic parts are electronic products utilizing metallic oxides. The metallic oxides have ferromagnetic properties that offer high resistance, making current losses extremely low at high frequencies. Ferrite electronic products include: (1) magnetic recording tape, (2) magnetic tape transport heads, (3) electronic and aircraft instruments, (4) microwave connectors and components, and (5) electronic digital equipment. Some typical processes to manufacture ferrite electronic parts are shearing, slitting, fabrication and machining. All production processes in this segment are included in the Metal Finishing category. Some typical raw materials are aluminum, magnesium, bronze, and brass.

4.19 MOTORS, GENERATORS, AND ALTERNATORS

Motors are devices that convert electric energy into mechanical energy. Generators are devices which convert an input mechanical energy into electrical energy. Alternators are devices that convert mechanical energy into electrical energy in the form of an alternating current. The major motor, generator, and alternator products are: (1) variable speed drives and gear motors, (2) fractional horsepower motors, (3) hermetic motor parts, (4) appliance motors, (5) special purpose electric motors, (6) electrical equipment for internal combustion engines, and (7) automobile electrical parts. Some typical processes are casting, stamping, blanking, drawing, welding, heat treating, assembling and machining. All production processes are included in the Metal Finishing category. Some basis materials are carbon steel, copper, aluminum and iron. These materials are used as sheet metal, rods, bars, strips, coils, casting, and tubing.

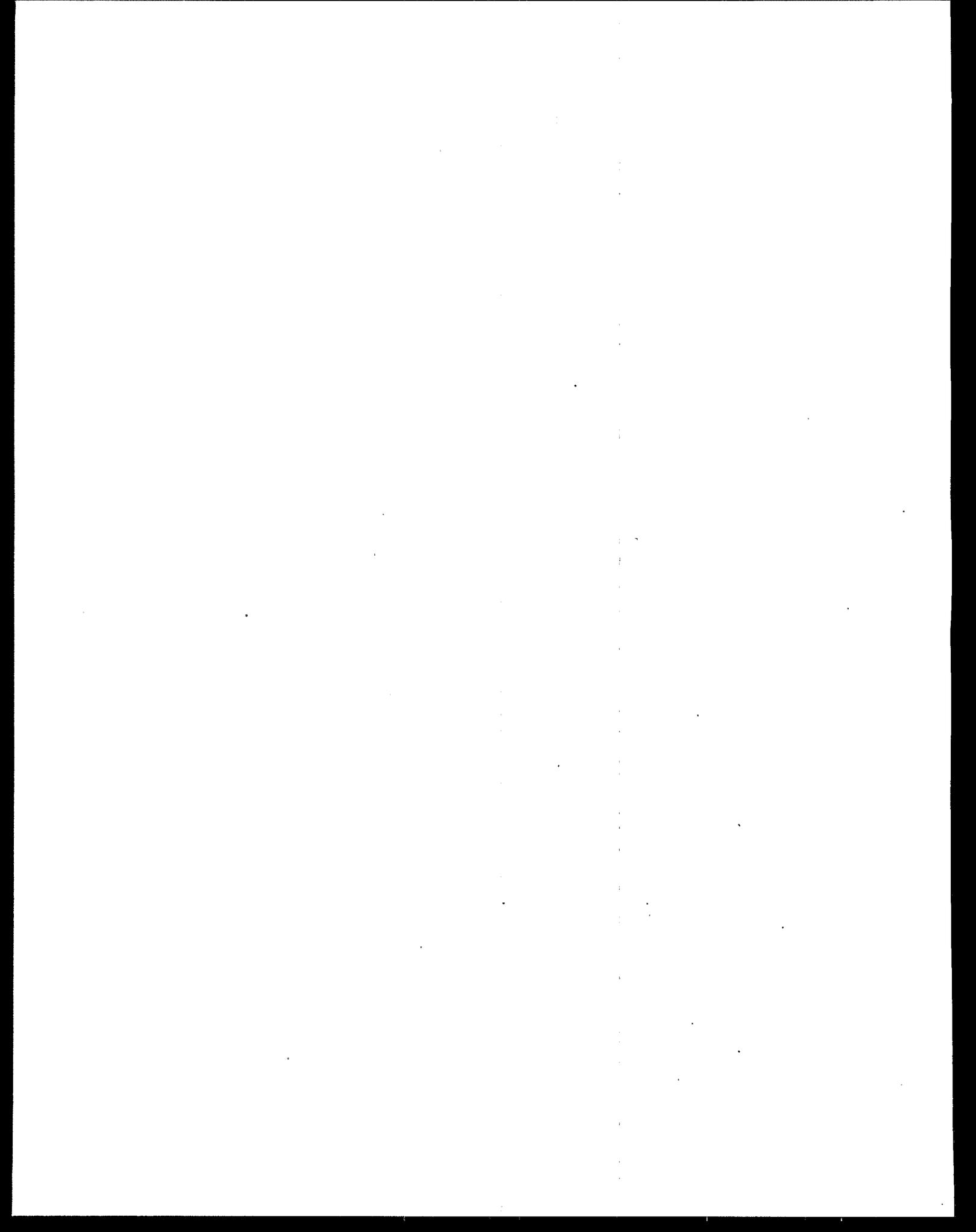
4.20 RESISTANCE HEATERS

Resistance heaters convert electrical energy into usable heat energy. Three types of resistance heaters are made; rigid encased elements used for electric stoves and ovens, bare wire heaters used in toasters and hair dryers, and insulated flexible heater wire that is incorporated into blankets and heating pads. Some typical processes used in the manufacture of resistance heaters are plating, welding or soldering, molding, and machining. These processes are included in the Metal Finishing category. Some raw materials used are steel, nickel, copper, plastic, and rubber.

4.21 SWITCHGEAR

Switchgear are products used to control electrical flow and to protect equipment from electrical power surges and short

circuits. The major switchgear products are: (1) electrical power distribution controls and metering panel assemblies, (2) circuit breakers, (3) relays, (4) switches, and (5) fuses. Some typical manufacturing processes are: chemical milling, grinding, electroplating, soldering or welding, machining and assembly. All processes are included in the Metal Finishing and Plastics Processing categories. Some typical basis materials are plastic, steel, copper, brass, and aluminum.



SECTION 5

WASTEWATER CHARACTERISTICS

This section presents information related to wastewater flows, wastewater sources, pollutants found, and the sources of these pollutants. For subcategories which are excluded or proposed under Phase II, the discussion of wastewater characteristics is abbreviated. A general discussion of sampling techniques and wastewater analysis is also provided.

5.1 SAMPLING AND ANALYTICAL PROGRAM

More than 250 plants were contacted to obtain data on the E&EC Category. Seventy-eight of these plants were visited for an on-site study of their manufacturing processes, water used and wastewater treatment. In addition, wastewater samples were collected at thirty-eight of the plants visited in order to quantitate the level of pollutants in the waste streams. Sampling was utilized to determine the source and quantity of pollutants in the raw process wastewater and the treated effluent from a cross-section of plants in the E&EC Category.

5.1.1 Pollutants Analyzed

The chemical pollutants sought in analytical procedures fall into three groups: Conventional, non-conventional, and toxics. The latter group comprises the 126 chemicals found in the priority pollutant list shown in Table 5-1 (p. 5-11).

Conventional pollutants are those generally treatable by secondary municipal wastewater treatment. The conventional pollutants examined for this study are:

- pH
- Biochemical Oxygen Demand (BOD)
- Oil and Grease (O&G)
- Total Suspended Solids (TSS)

Non-conventional pollutants are simply those which are neither conventional nor on the list of toxic pollutants. The non-conventional pollutants listed on page 5-2 were examined in one or more subcategories of the E&EC industry.

Bismuth	Magnanese
Europium	Vanadium
Fluoride	Boron
Gadolinium	Barium
Gallium	Molybdenum
Indium	Tin
Lithium	Cobalt
Niobium	Iron
Tellurium	Titanium
Total Organic Carbon	Xylenes
Total Phenols	Alkyl Epoxides
Yttrium	Platinum
Calcium	Palladium
Magnesium	Gold
Aluminum	

5.1.2 Sampling Methodology

During the initial visit to a facility, a selection was made of sampling points so as to best characterize process wastes and evaluate the efficiency of any wastewater treatment. The nature of the wastewater flow at each selected sampling point then determined the method of sampling, i.e., automatic composite or grab composite. The sampling points were of individual raw or treated process waste streams, or treated effluent.

Each sample was collected whenever possible by an automatic time series compositor over a single 24-hour sampling period. When automatic compositing was not possible or appropriate, as for volatile organics, grab samples were taken at intervals over the same period, and were composited manually. When a sample was taken for analysis of toxic organics, a blank was also taken to determine the level of contamination inherent to the sampling and transportation procedures.

Each sample was divided into several portions and preserved, when necessary, in accordance with established procedures for the measurement of toxic and classical pollutants. Samples were shipped in ice-cooled containers by the best available route to EPA-contracted laboratories for analysis. Chain of custody for the samples was maintained through the EPA Sample Control Center tracking forms.

5.1.3 Analytical Methods

The analytical techniques for the identification and quantitation of toxic pollutants were those described in Sampling and Analysis Procedures for Screening of Industrial Effluents for Priority Pollutants, revised in April 1977.

In the laboratory, samples for organic pollutant analysis were separated by specific extraction procedures into acid (A), base/neutral (B/N), and pesticide (P) fractions. Volatile organic samples (V) were taken separately as a series of grab samples at four-hour intervals and composited in the laboratory. The analysis of these fractions included the application of strict quality control techniques including the use of standards, blanks, and spikes. Gas chromatography and gas chromatography/mass spectrometry were the analytical procedures used for the organic pollutants. Two other analytical methods were used for the measurement of toxic metals: Flameless atomic absorption and inductively coupled argon plasma spectrometric analysis (ICAP). The metals determined by each method were:

Flameless AA

Antimony
Arsenic
Selenium
Silver
Thallium

ICAP

Beryllium
Cadmium
Chromium
Copper
Lead
Nickel
Zinc

Mercury was analyzed by a special manual cold-vapor atomic absorption technique.

For the analysis of conventional and non-conventional pollutants, procedures described by EPA were followed. The following conventions were used in quantifying the levels determined by analysis:

- o Pollutants detected at levels below the quantitation limit are reported as "less than" (<) the quantitation limit. All other pollutants are reported as the measured value.
- o Sample Blanks - Blank samples of organic-free distilled water were placed adjacent to sampling points to detect airborne contamination of water samples. These sample blank data are not subtracted from the analysis results, but, rather, are shown as a (B) next to the pollutant found in both the sample and the blank. The tables show data for total toxic organics, toxic and non-toxic metals, and other pollutants.
- o Blank Entries - Entries were left blank when the parameter was not detected.

5.2 SEMICONDUCTORS

5.2.1 Wastewater Flows

Table 5-2 presents a summary of the quantities of wastewater generated by the Semiconductor subcategory.

TABLE 5-2
SEMICONDUCTOR SUBCATEGORY

PROCESS WASTEWATER FLOW:

<u>Maximum</u> <u>l/day (gal/day)</u>	<u>Minimum</u> <u>l/day (gal/day)</u>	<u>Average</u> <u>l/day (gal/day)</u>
11,100,000 (2,940,000)	212,000 (56,000)	594,000 (157,000)

CONCENTRATED FLUORIDE WASTEWATER FLOW:

5,450	(1,440)	95	(25)	678	(179)
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Total Subcategory Process Water Use = 193,000,000 liters/day
(51,000,000 gal/day)

5.2.2 Wastewater Sources

Contact water is used throughout the production of semiconductors. Plant incoming water is first pretreated by deionization to provide ultrapure water for processing steps. This ultrapure water or deionized (DI) water is used to formulate acids; to rinse wafers after processing steps; to provide a medium for collecting exhaust gases from diffusion furnaces, solvents, and acid baths; and to clean equipment and materials used in semiconductor production. Water also cools and lubricates the diamond saws and grinding machines used to slice, lap, and dice wafers during processing.

5.2.3 Pollutants Found and Sources of These Pollutants

The major pollutants found at facilities in the Semiconductor subcategory are as follows:

Fluoride
Toxic Organics
pH

The process steps associated with the sources of these pollutants are described in Section 4.1.3 (p. 4-2). Table 5-3 (p. 5-13) summarizes pollutant concentration data for the sampled raw waste streams. Tables 5-4 through 5-15 (pages 5-15 through 5-73) present the analytical data for twelve sampled plants in the Semiconductor subcategory.⁽¹⁾

Fluoride -- The source of fluoride is hydrofluoric acid, which is used as an etchant and a cleaner. Certain areas of the basis material are etched to provide surfaces receptive to the entry of dopants that are subsequently added to the wafer. The major source of fluoride comes from the discharge of spent hydrofluoric acid after its use in etching. (The flows of this waste stream are shown in Table 5-2.) Minor quantities of fluoride enter the plant wastewater from rinses of etched or cleaned wafers.

Toxic organics -- The sources of toxic organics are solvents used for drying the wafer after rinsing, developing of photoresist, stripping of photoresist, and cleaning. A further discussion of the sources of toxic organics is presented in Section 7.

pH -- This parameter may be very high or very low. High pH results from the use of alkalis for caustic cleaning. Low pH results from the use of acids for etching and cleaning.

Several toxic metals were found in the wastewater because of electroplating operations associated with semiconductor manufacture. These metals are chromium, copper, nickel and lead, and are covered under the final electroplating or proposed metal finishing effluent limitations and standards.

5.3 ELECTRONIC CRYSTALS

5.3.1 Wastewater Flows

The following table (5-16) contains a summary of the wastewater flows generated in the Electronic Crystals subcategory.

- (1) Several corrections have been made to the data tables presented in the proposed development document of July 1982 (EPA 440/1-82/075-b). Data were either not transcribed or were incorrectly transcribed from laboratory analytical reports. In addition, stream descriptions have been changed to provide consistency among plants (i.e., frequently plants title their streams differently).

TABLE 5-16

SUMMARY OF WASTEWATER QUANTITIES GENERATED
IN THE ELECTRONIC CRYSTALS SUBCATEGORY

	<u>No. of Plants</u>	<u>Wastewater Discharge</u>		<u>Liters/day</u>
		<u>Min</u>	<u>Max</u>	<u>Mean</u>
All Plants	49	95	1,839,800	112,400

5.3.2 Wastewater Sources

The major source of wastewater from the manufacture of electronic crystals is from rinses associated with crystal fabrication, although some wastewater may be generated from crystal growing operations. Fabrication steps generating wastewater are slicing, lapping, grinding, polishing, etching, and cleaning of grown crystals. Certain growth processes generate a large volume of wastewater from the discharge of spent solutions of sodium hydroxide and sodium carbonate after each crystal growth cycle.

5.3.3 Pollutants Found and the Sources of These Pollutants

The major pollutants of concern from the Electronic Crystals subcategory are:

Toxic Organics
Fluoride
Arsenic
TSS
pH

The process steps associated with the sources of these pollutants are described in Section 4.2.3 on page 4-10. Table 5-17 (p. 5-74) summarizes the occurrence and levels at which these pollutants are found based on the sampling and analysis of raw wastes from eight crystals facilities. Concentrations represent total raw wastes after flow-proportioning individual discharge streams. Tables 5-18 through 5-25 (p. 5-75 through p. 5-94), summarize the analytical data obtained from each of the plants sampled and identify products produced and wastewater flows.⁽¹⁾

Toxic organics -- found in wastewater from the manufacture of electronic crystals as a result of the use of solvents such as

(1) Several corrections have been made to the data tables presented in the proposed development document of July 1982 (EPA 440/1-82/075-b). Data were either not transcribed or were incorrectly transcribed from laboratory analytical reports. In addition, stream descriptions have been changed to provide consistency among plants (i.e., frequently plants title their streams differently).

isopropyl alcohol, 1,1,1-trichloroethane, Freon, and acetone. These materials are used for cleaning and drying of crystals. Another source of toxic organics could be contaminants in oils used as lubricants in slicing and grinding operations. A further discussion of the sources of toxic organics is presented in Section 7.

Fluoride -- has as its source the use of hydrofluoric acid or ammonium bifluoride for etching electronic crystals. A minor source of fluoride is from the etch rinse process.

Arsenic -- originates from the gallium arsenide and indium arsenide used as raw material for crystals. Process steps generating wastewater containing arsenic are cleaning of the crystal-growing equipment, slicing and grinding operations, and etching and rinsing steps.

Total Suspended Solids -- common in crystals manufacturing waste streams as crystal grit from slicing and grinding operations. Grit and abrasives wastes are also generated by grinding and lapping operations.

pH -- may be very high or very low. High pH results from the presence of excess alkali such as sodium hydroxide or sodium carbonate. The alkali may come from crystal growth processes or from caustic cleaning and rinsing. Low pH results from the use of acid for etching and cleaning operations.

Several toxic metals were found in the wastewater because of electroplating operations associated with electronic crystals manufacture. These metals are chromium, copper, lead, nickel, and zinc, and are regulated under the Metal Finishing Category.

5.4 CARBON AND GRAPHITE PRODUCTS

The average flow of wastewater from these plants is 24.2 x 10⁶ l/day (6,388,400 gal/day). The major pollutants found and their concentrations are presented below:

Toxic Pollutants

<u>Pollutant</u>	<u>Raw Waste Load Concentration (mg/l)</u>	<u>Raw Waste Load kg/day (lbs/day)</u>	
Total Toxic Inorganics	0.080	1.93	(4.26)
Bis(2-ethylhexyl)phthalate	0.042	1.02	(2.24)
Methylene Chloride	0.013	0.31	(0.69)
Total Toxic Organics	0.080	1.93	(4.26)

Raw waste concentrations are based on flow weighted means from four plants. For toxic inorganics only flow weighted mean concentrations greater than or equal to 0.1 mg/l are shown. For toxic organics only flow weighted mean concentrations greater or equal to 0.01 mg/l are shown.

5.5 MICA PAPER

The average flow of wastewater from these plants is 3.50×10^6 l/day (926,000 gal/day). The major pollutants found and their concentrations are presented below:

Toxic Pollutants

<u>Pollutant</u>	<u>Raw Waste Load Concentration (mg/l)</u>	<u>Raw Waste Load kg/day (lbs/day)</u>	
Total Toxic Inorganics	0.055	0.20	(0.44)
1,1,1-Trichloroethane	0.180*	0.63	(1.39)
Methylene Chloride	0.029*	0.10	(0.22)
Total Toxic Organics	0.209	0.73	(1.61)

*Not confirmed by process or raw material usage.

Raw waste concentrations are based on raw waste data from one plant. For toxic organics only concentrations greater than or equal to 0.01 mg/l are shown.

5.6 INCANDESCENT LAMPS

The average flow of wastewater from these plants is 7.74×10^6 l/day (540,100 gal/day). The major pollutants found and their concentrations are described below:

Toxic Pollutants

<u>Pollutant</u>	<u>Raw Waste Load Concentration (mg/l)</u>	<u>Raw Waste Load kg/day (lbs/day)</u>	
Chromium	0.714	1.46	(3.22)
Copper	0.420	0.86	(1.89)
Lead	0.11	0.23	(0.50)
Total Toxic Inorganics	1.377	2.82	(6.21)
Methylene Chloride	0.048	0.05	(0.11)
Chloroform	0.024	0.10	(0.22)
Dichlorobromomethane	0.010	0.03	(0.05)
Total Toxic Organics	0.082	0.17	(0.38)

Raw waste concentrations are based on flow weighted means from three plants. For toxic inorganics only flow weighted mean concentrations greater than or equal to 0.1 mg/l are shown. For toxic organics only flow weighted mean concentrations greater than or equal to 0.01 mg/l are shown.

5.7 FLUORESCENT LAMPS

The major pollutants found in wastewaters from these plants and their concentrations or mass loadings are presented below:

Toxic Pollutants

<u>Pollutant</u>	<u>Raw Waste Load Concentration (mg/l)</u>	<u>Raw Waste Load kg/day (lbs/day)</u>	
Antimony	0.458	--	--
Cadmium	0.307	--	--
Total Toxic Inorganics	--	0.80	(1.76)
Methylene Chloride	0.063	--	--
Toluene	0.011	--	--
Total Toxic Organics	--	0.07	(0.16)

5.8 FUEL CELLS

Only a few plants manufacture fuel cells and these do not do so on a regular basis. In addition, all pollutants found were at quantities too low to be effectively treated.

5.9 MAGNETIC COATINGS

This subcategory discharges only a small amount of pollutants to water. The average wastewater discharge from this subcategory is 19,000 l/day (5,000 gal/day). The total toxic metals discharge for the subcategory is 0.045 kg/day (0.099 lbs/day), total toxic organics is 0.018 kg/day (0.040 lbs/day).

5.10 RESISTORS

No wastewaters result from the manufacture of resistors.

5.11 DRY TRANSFORMERS

No wastewaters result from the manufacture of dry transformers.

5.12 ELECTRON TUBES (Phase II)

The agency has proposed regulations for this subcategory under Electrical and Electronic Components, Phase II.

5.13 PHOSPHORESCENT COATINGS (Phase II)

The Agency has proposed regulations for this subcategory as luminescent materials under Electrical and Electronic Components, Phase II.

5.14 ALL OTHER SUBCATEGORIES

Information obtained from plant visits showed that wastewater discharges in the following subcategories result primarily from processes associated with metal finishing and, in the case of insulated plastic and plastic-laminated devices, from processes associated with the EPA study on plastics molding and forming. Because these processes are studied elsewhere, the E&EC project limited its sampling effort in these areas.

- Switchgear and Fuses
- Resistance Heaters
- Ferrite Electronic Parts
- Insulated Wire and Cable
- Fluid-filled Capacitors
- Fluid-filled Transformers
- Insulated Devices -- Plastics and Plastic Laminated
- Motors, Generators, and Alternators
- Fixed Capacitors

TABLE 5-1
THE PRIORITY POLLUTANTS

TOXIC POLLUTANT

1. Acenaphthene	46. Methyl Bromide (Bromomethane)
2. Acrolein	47. Bromoform (Tribromomethane)
3. Acrylonitrile	48. Dichlorobromomethane
4. Benzene	51. Chlorodibromomethane
5. Benzidine	52. Hexachlorobutadiene
6. Carbon Tetrachloride (Tetrachloromethane)	53. Hexachlorocyclopentadiene
7. Chlorobenzene	54. Isophorone
8. 1,2,4-Trichlorobenzene	55. Naphthalene
9. Hexachlorobenzene	56. Nitrobenzene
10. 1,2-Dichloroethane	57. 2-Nitrophenol
11. 1,1,1-Trichloroethane	58. 4-Nitrophenol
12. Hexachloroethane	59. 2,4-Dinitrophenol
13. 1,1-Dichloroethane	60. 4,6-Dinitro- <i>o</i> -cresol
14. 1,1,2-Trichloroethane	61. <i>N</i> -Nitrosodimethylamine
15. 1,1,2,2-Tetrachloroethane	62. <i>N</i> -Nitrosodiphenylamine
16. Chloroethane	63. <i>N</i> -Nitrosodi- <i>n</i> -propylamine
18. Bis(2-chloroethyl)ether	64. Pentachlorophenol
19. 2-Chloroethyl Vinyl Ether (Mixed)	65. Phenol
20. 2-Chloronaphthalene	66. Bis(2-ethylhexyl) Phthalate
21. 2,4,6-Trichlorophenol	67. Butyl Benzyl Phthalate
22. <i>p</i> -Chloro- <i>m</i> -cresol	68. Di- <i>n</i> -butyl Phthalate
23. Chloroform (Trichloromethane)	69. Di- <i>n</i> -octyl Phthalate
24. 2-Chlorophenol	70. Diethyl Phthalate
25. 1,2-Dichlorobenzene	71. Dimethyl Phthalate
26. 1,3-Dichlorobenzene	72. 1,2-Benzanthracene [Benzo(a)anthracene]
27. 1,4-Dichlorobenzene	73. Benzo(a)Pyrene (3,4-Benzopyrene)
28. 3,3'-Dichlorobenzidine	74. 3,4-Benzofluoranthene [Benzo(b)fluoranthene]
29. 1,1-Dichloroethylene	75. 1,1,12-Benzofluoranthene [Benzo(k)fluoranthene]
30. 1,2-trans-Dichloroethylene	76. Chrysene
31. 2,4-Dichlorophenol	77. Acenaphthylene
32. 1,2-Dichloropropane	78. Anthracene
33. 1,3-Dichloropropylene(1,3-Dichloropropene)	79. 1,12-Benzoperylene [Benzo(ghi)perylene]
34. 2,4-Dimethyl Phenol	80. Fluorene
35. 2,4-Dinitrotoluene	81. Phenanthrene
36. 2,6-Dinitrotoluene	82. 1,2,5,6-Dibenzanthracene [Dibenzo(a,h)anthracene]
37. 1,2-Diphenylhydrazine	83. Indeno(1,2,3-cd)pyrene (2,3- <i>o</i> -Phenylenepyrene)
38. Ethylbenzene	84. Pyrene
39. Fluoranthene	85. Tetrachloroethylene
40. 4-Chlorophenyl Phenyl Ether	86. Toluene
41. 4-Bromophenyl Phenyl Ether	87. Trichloroethylene
42. Bis(2-chloroisopropyl)ether	88. Vinyl Chloride (Chloroethylene)
43. Bis(2-chloroethoxy)methane	89. Aldrin
44. Methylene Chloride(Dichloromethane)	90. Dieldrin
45. Methyl Chloride(Chloromethane)	

TABLE 5-1
THE PRIORITY POLLUTANTS
(Continued)

TOXIC POLLUTANT

91. Chlordane (Technical Mixture and Metabolites)	110. PCB-1248 (Aroclor 1248)
92. 4,4'-DDT	111. PCB-1260 (Aroclor 1260)
93. 4,4'-DDE(P,P'-DDX)	112. PCB-1016 (Aroclor 1016)
94. 4,4'-DDD(P,P'-TDE)	113. Toxaphene
95. Alpha-Endosulfan	114. Antimony
96. Beta-Endosulfan	115. Arsenic
97. Endosulfan Sulfate	116. Asbestos
98. Endrin	117. Beryllium
99. Endrin Aldehyde	118. Cadmium
100. Heptachlor	119. Chromium
101. Heptachlor Epoxide(BHC-Hexachloro- cyclohexane)	120. Copper
102. Alpha-BHC	121. Cyanide
103. Beta-BHC	122. Lead
104. Gamma-BHC(Lindane)	123. Mercury
105. Delta-BHC	124. Nickel
106. PCB-1242 (Aroclor 1242)	125. Selenium
107. PCB-1254 (Aroclor 1254)	126. Silver
108. PCB-1221 (Aroclor 1221)	127. Thallium
109. PCB-1232 (Aroclor 1232)	128. Zinc
	129. 2,3,7,8-Tetrachlorodibenzo p dioxin(TCDD)

TABLE 5-3
SEMICONDUCTOR
SUMMARY OF THE RAW WASTE DATA

<u>Toxic Organics</u> Parameter	<u>Plant 42044</u> mg/l	<u>Plant 04294</u> mg/l
8 1,2,4-Trichlorobenzene	ND**	27.1
21 2,4,6-Trichlorophenol	ND	0.013
23 Chloroform	<0.01	0.012
25 1,2-Dichlorobenzene	0.04	186.0
26 1,3-Dichlorobenzene	<0.01	7.4
27 1,4-Dichlorobenzene	<0.01	7.4
38 Ethylbenzene	ND	0.107
44 Methylene chloride	0.044	0.101
55 Naphthalene	ND	1.504
57 2-Nitrophenol	ND	0.039
64 Pentachlorophenol	ND	0.250
65 Phenol	0.180	0.170
66 Bis(2-ethylhexyl) phthalate	0.010	0.012
68 Di-n-butyl phthalate	<0.01	0.017
85 Tetrachloroethylene	0.015	0.143
87 Trichloroethylene	<u>ND</u>	<u>0.204</u>
 TOTAL TOXIC ORGANICS	 0.279	 230.472

* This table shows the range of toxic organics observed.
** Not detected

TABLE 5-3
SEMICONDUCTOR
SUMMARY OF RAW WASTE DATA

TOXIC METALS

<u>Parameter</u>	<u>Min. Con.</u> mg/l	<u>Max. Conc.</u> mg/l	<u>Mean Conc.</u> mg/l
114 Antimony	<0.001	0.187	0.013
115 Arsenic	<0.003	0.067	0.015
117 Beryllium	<0.001	<0.015	<0.001
118 Cadmium	<0.001	0.008	0.003
119 Chromium †	<0.001	1.150	0.146
120 Copper †	<0.005	2.588	0.570
122 Lead †	<0.04	1.459	0.135
123 Mercury	<0.001	0.051	0.003
124 Nickel †	0.005	4.964	0.500
125 Selenium	<0.002	0.045	0.015
126 Silver	<0.001	0.013	0.002
127 Thallium	<0.001	0.012	<0.001
128 Zinc	0.001	0.289	0.092
Total Toxic Inorganics	0.063	10.848	1.496

CONVENTIONAL POLLUTANTS

Oil & Grease	ND	6.8	3.9
Total Suspended Solids*	ND	14	6.9
Biochemical Oxygen Demand	ND	30	21.3

NON-CONVENTIONAL POLLUTANTS

Total Organic Carbon	ND	80	55.7
Fluoride	26.6	146.5	65.5

† - These metals are associated with metal finishing operations.

ND - Not detected.

* - Data for TSS is from plants producing semiconductors only.

TABLE 5-4

SEMICONDUCTOR PROCESS WASTES
PLANT 02040

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes†		Equipment Cleaning Wastes†		Wafer Finishing Wastes		Effluent		
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	
	5437		31.5		2178		463505		
	24		24		24		24		
	3480		3481		3477		3478		
TOXIC ORGANICS									
4 Benzene					<0.01		<0.01		
7 Chlorobenzene							<0.01		
8 1,2,4-Trichlorobenzene									
11 1,1,1-Trichloroethane									
13 1,1-Dichloroethane					0.01		1.10	12.24	
23 Chloroform							<0.01		
24 2-Chlorophenol					0.047	0.0025	0.05	0.56	
25 1,2-Dichlorobenzene							<0.01		
27 1,4-Dichlorobenzene					0.012	0.0006	0.068	0.76	
29 1,1-Dichloroethylene					<0.01		0.410	4.56	
38 Ethylbenzene							<0.01		
44 Methylene chloride							<0.01		
48 Dichlorobromomethane					0.046	0.002	0.095		
51 Chlorodibromomethane					<0.01				
57 2-Nitrophenol					<0.01	0.0005	<0.01		
58 4-Nitrophenol							<0.01		
65 Phenol							<0.01		
66 Bis(2-ethylhexyl)phthalate					<0.01		0.270	3.0	
67 Butyl benzyl phthalate					0.010	0.0005	0.019	0.21	
68 Di-N-butyl phthalate							<0.01		
69 Di-N-octyl phthalate					<0.01		<0.01		
70 Diethyl phthalate							<0.01		
71 Dimethyl phthalate							<0.01		
86 Toluene									
87 Trichloroethylene					<0.01		0.14	1.56	
121 Cyanide*					<0.01		<0.01		
							<0.005		
Total Toxic Organics					0.105	0.0055	2.152	23.94	
TOXIC INORGANICS									
114 Antimony	<0.005		<0.005		<0.005		<0.005		
115 Arsenic	0.006	0.0008	0.074	0.00006	0.004	0.0002	0.01	0.11	
117 Beryllium	<0.001		<0.001		<0.001		<0.001		
118 Cadmium	<0.001		0.05	0.00004	<0.001		0.002	0.02	
119 Chromium	0.009	0.001	<0.001		<0.001		0.341	3.79	
120 Copper	0.002	0.0003	<0.001		0.056	0.003	0.413	4.59	
122 Lead	<0.001		0.25	0.0002	0.034	0.002	0.025	0.28	
123 Mercury	<0.001		<0.001		0.001	0.00005	<0.001		
124 Nickel	<0.001		0.90	0.0007	<0.001		4.964	55.2	
125 Selenium	<0.003		<0.003		0.003		<0.003		

* Not included in Total Toxic Organics summation.

† Organics not analyzed.

TABLE 5-4 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 02040

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes 5437 24 3480		Equipment Cleaning Wastes 31.5 24 3481		Wafer Finishing Wastes 2178 24 3477		Effluent 463505 24 3478	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	<0.005		<0.005		<0.005		<0.005	
127 Thallium	<0.025		<0.025		<0.025		<0.025	
128 Zinc	0.04	0.005	0.80	0.0006	0.070	0.0037	0.111	1.23
Total Toxic Inorganics	0.057	0.007	2.076	0.0016	0.165	0.0086	5.866	65.25
NON-CONVENTIONAL POLLUTANTS								
Aluminum	<0.001		16.31	0.012	0.155	0.008	0.323	3.59
Barium	0.026	0.003	0.05	0.00004	0.003	0.0002	0.024	0.27
Boron	0.267	0.035	60.66	0.046	0.251	0.013	0.690	7.68
Calcium	36.36		45.92		1.710		46.1	
Cobalt	0.002	0.0003	0.48	0.0004	<0.001		0.147	1.64
Gold	<0.02		<0.02		<0.02		<0.02	
Iron	0.012	0.0016	0.46	0.0003	0.109	0.0057	0.813	9.04
Magnesium	19.34		23.78		0.319		17.12	
Manganese	0.009	0.0012	<0.001		0.001	0.00005	0.014	0.16
Molybdenum	0.005	0.0007	0.57	0.0004	0.008	0.0042	0.006	0.067
Palladium	<0.08		<0.08		<0.08		<0.08	
Platinum	<0.05		<0.05		<0.05		<0.05	
Sodium	50.52		161.57		73.021		192.501	
Tellurium	<0.02		<0.02		<0.02		<0.02	
Tin	0.016	0.0021	1.01	0.0008	0.047	0.0025	0.297	3.30
Titanium	0.001		0.03	0.00002	0.022	0.001	0.003	0.03
Vanadium	0.130	0.017	0.16	0.0001	0.003	0.00016	0.123	1.37
Yttrium	<0.001		<0.001		<0.001		<0.001	
Phenols	0.009	0.0012			0.039	0.002	6.1	67.9
Total Organic Carbon	6 ##	0.783			26	1.36	37	411.6
Fluoride	0.48 ##	0.063	290	0.22	0.27	0.014	52.0	578.5
CONVENTIONAL POLLUTANTS								
Oil & Grease			NA		7.0	0.37	4	44.5
Total Suspended Solids	<2	0.26	NA		5.0	0.26	62	689.7
Biochemical Oxygen Demand	<5	0.65	NA		15	0.78	52	578.5
pH								

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-4 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 02040

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Wafer Finishing Wastes 10402 24 03476		Scrubber Wastes† 2580 24 03479	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS				
4 Benzene				
7 Chlorobenzene				
8 1,2,4-Trichlorobenzene		NA		
11 1,1,1-Trichloroethane	<0.01			
13 1,1-Dichloroethane	0.01	0.002		
23 Chloroform	0.02	0.005		
24 2-Chlorophenol				
25 1,2-Dichlorobenzene	<0.01			
27 1,4-Dichlorobenzene				
29 1,1-Dichloroethylene				
38 Ethylbenzene				
44 Methylene chloride	0.035	0.009		
51 Chlorodibromomethane				
57 2-Nitrophenol				
58 4-Nitrophenol				
65 Phenol	0.031	0.008		
66 Bis(2-ethylhexyl)phthalate	<0.01			
67 Butyl benzyl phthalate				
68 Di-N-butyl phthalate	<0.01			
69 Di-N-octyl phthalate				
70 Diethyl phthalate	<0.01			
71 Dimethyl phthalate				
86 Toluene				
87 Trichloroethylene				
121 Cyanide*				
Total Toxic Organics	0.086	0.021		
TOXIC INORGANICS				
114 Antimony	0.007	0.002	0.017	0.001
115 Arsenic	0.003	0.001	0.007	0.0004
117 Beryllium	<0.001		<0.001	
118 Cadmium	<0.001		<0.001	
119 Chromium	<0.001		0.011	0.0007
120 Copper	0.046	0.012	0.007	0.0004
122 Lead	0.001	0.0002	<0.001	
123 Mercury	<0.001		<0.001	
124 Nickel	<0.001		<0.001	
125 Selenium	<0.003		<0.003	

* Not included in Total Toxic Organics summation.

† Organics not analyzed

TABLE 5-4 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 02040

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Wafer Finishing Wastes		Scrubber Wastes	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	10402		2580	
	24		24	
	03476		03479	
TOXIC INORGANICS (Continued)				
126 Silver	<0.005		<0.005	
127 Thallium	<0.025		<0.025	
128 Zinc	1.113	0.278	0.059	0.0037
Total Toxic Inorganics				
NON-CONVENTIONAL POLLUTANTS				
Aluminum	0.015	0.004	<0.001	
Barium	0.024	0.006	0.026	0.002
Boron	0.222	0.055	0.164	0.010
Calcium	28.040		35.830	
Cobalt	<0.001		0.003	0.0002
Gold	<0.020		<0.020	
Iron	0.169	0.042	0.047	0.003
Magnesium	13.500		19.080	
Manganese	0.006	0.002	<0.001	
Molybdenum	0.001	0.0002	0.004	0.0002
Palladium	<0.080		<0.080	
Platinum	<0.050		<0.050	
Sodium	111.601		49.711	
Tellurium	<0.020		<0.020	
Tin	0.023	0.006	0.011	0.001
Titanium	0.006	0.002	<0.001	
Vanadium	0.091	0.023	0.130	0.008
Yttrium	<0.001		<0.001	
Phenols	0.032	0.008	0.01 #	
Total Organic Carbon	137 #	34.2	8 ##	0.495
Fluoride	0.48 #	0.12	0.46 #	0.028
CONVENTIONAL POLLUTANTS				
Oil & Grease	9.0	2.247		
Total Suspended Solids	885	220.9	<2 #	
Biochemical Oxygen Demand	310	77.39	<5 #	
pH				

Data incorrectly transcribed at proposal. (See note on page 5-5.)

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

TABLE 5-5

SEMICONDUCTOR PROCESS WASTES
PLANT 02347

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes 6099 24 03474		Effluent 130,688 24 03475	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS				
4 Benzene	0.190	0.128		
7 Chlorobenzene			<0.01 #	
8 1,2,4-Trichlorobenzene			0.089	0.279
11 1,1,1-Trichloroethane	0.170	0.025	3.5 ##	10.98
23 Chloroform	2.6	0.38	0.022	0.069
24 2-Chlorophenol	0.011	0.0016	<0.01	
25 1,2-Dichlorobenzene	<0.01		0.860	
27 1,4-Dichlorobenzene			0.170	0.53
29 1,1-Dichloroethylene				
31 2,4-Dichlorophenol	<0.01		0.017	0.053
37 1,2-Diphenylhydrazine			<0.01	
38 Ethylbenzene	<0.01		<0.01	
39 Fluoranthene			<0.01	
44 Methylene chloride	1.9	0.278	2.4	7.53
55 Naphthalene			<0.01	
57 2-Nitrophenol			<0.01	
65 Phenol	0.220	0.032	0.810	2.54
66 Bis(2-ethylhexyl)phthalate	<0.01		0.013	0.04
67 Butyl benzyl phthalate				
68 Di-N-butyl phthalate	<0.01		<0.01	
69 Di-N-octyl phthalate			<0.01	
70 Diethyl phthalate			<0.01	
85 Tetrachloroethylene			20.0 #	62.73
86 Toluene	<0.01		<0.01	
87 Trichloroethylene ##				
121 Cyanide*	NA		0.075 #	0.235
Total Toxic Organics	5.091	0.745	27.881	87.45
TOXIC INORGANICS				
114 Antimony	<0.005		<0.005	
115 Arsenic	0.003	0.0004	0.002	0.0063
117 Beryllium	<0.001		<0.001	
118 Cadmium	<0.001		<0.001	
119 Chromium	<0.001		0.110	0.345
120 Copper	<0.001		1.182	3.71
122 Lead	<0.001		0.042	0.132
123 Mercury	0.001	0.00015	0.001	0.003
124 Nickel	<0.001		<0.001	
125 Selenium	<0.003		<0.003	

* Not included in Total Toxic Organics summation.

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-5 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 02347

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes 6099 24 03474		Effluent 130,688 24 03475	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)				
126 Silver	<0.005		<0.005	
127 Thallium	<0.025		<0.025	
128 Zinc	0.052	0.0076	0.089	0.28
Total Toxic Inorganics	0.056	0.008	1.426	4.473
NON-CONVENTIONAL POLLUTANTS				
Aluminum	0.009	0.0013	0.02	0.063
Barium	0.039 #	0.006	0.015	0.05
Boron	0.121	0.018	0.76	2.38
Calcium	42.31		14.31	
Cobalt	<0.001		<0.001	
Gold	<0.02		<0.02	
Iron	0.016 ##	0.0023	0.106	0.38
Magnesium	11.02		3.542	
Manganese	<0.001		0.001	0.003
Molybdenum	0.008	0.001	<0.001	
Palladium	<0.08		<0.08	
Platinum	<0.05		<0.05	
Sodium	43.321		116.2	
Tellurium	<0.02		<0.02	
Tin	0.011 ##	0.002	0.029	0.091
Titanium	<0.001 ##	0.003	<0.001	
Vanadium	0.068	0.01	0.015	0.047
Yttrium	0.001		<0.001	
Phenols	0.88 #	0.129	14 #	43.9
Total Organic Carbon	10	1.46	38	119.2
Fluoride	1.7	0.25	50	156.8
CONVENTIONAL POLLUTANTS				
Oil & Grease	<5 #		<5 #	
Total Suspended Solids	<2 #		<2 #	
Biochemical Oxygen Demand	NA		NA	
pH				

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)
Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-6
SEMICONDUCTOR PROCESS WASTES
PLANT 04294

Stream Description Flow (l/hr) Duration (hrs) Sample ID No.	Developer Quench Rinse		Acid Wastes		Stripper Quench Rinse 110** 24		Etching Solution†	
	3647 Concentration mg/l	Mass Load kg/day	3643 Concentration mg/l	Mass Load kg/day	3645 Concentration mg/l	Mass Load kg/day	3648 Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene			<0.01		<0.01			
7 Chlorobenzene								
8 1,2,4-Trichlorobenzene								
11 1,1,1-Trichloroethane								
13 1,1-Dichloroethane								
21 2,4,6-Trichlorophenol								
23 Chloroform	0.026	<0.01			0.021	0.00006		
24 2-Chlorophenol								
25 1,2-Dichlorobenzene								
26 1,3-Dichlorobenzene								
27 1,4-Dichlorobenzene								
34 2,4-Dimethylphenol	<0.01							
37 1,2-Diphenylhydrazine								
38 Ethylbenzene					<0.01			
39 Fluoranthene								
44 Methylene chloride	0.042				0.01			
48 Dichlorobromomethane	<0.01		<0.01		<0.01			
51 Chlorodibromomethane	<0.01		<0.01		<0.01			
54 Isophorone	<0.01							
55 Naphthalene								
57 2-Nitrophenol								
58 4-Nitrophenol								
64 Pentachlorophenol	<0.01		<0.01 #		<0.01			
65 Phenol	<0.01 #							
66 Bis(2-ethylhexyl)phthalate	<0.01		<0.01		<0.01		<0.01 #	
67 Butyl benzyl phthalate								
68 Di-N-butyl phthalate			<0.01		<0.01			
69 Di-N-octyl phthalate								
70 Diethyl phthalate	<0.01		<0.01		<0.01		<0.01 #	
71 Dimethyl phthalate							<0.01 #	
85 Tetrachloroethylene	<0.01		<0.01					
86 Toluene	0.017		<0.01		<0.01			
87 Trichloroethylene	<0.01		<0.01		<0.01			
97 Endosulfan sulfate							<0.005 #	
103 Beta BHC					<0.005 ##			
104 Gamma BHC			<0.005					
121 Cyanide*	<0.005 #		<0.005 #		<0.005 #			
Total Toxic Organics	0.085				0.021	0.00006		

* Not included in Total Toxic Organics summation.

** Estimated flow

† Volatile organics were not analyzed

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-6 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 04294

Stream Description Flow (l/hr) Duration (hrs) Sample ID No.	Developer Quench Rinse		Acid Wastes		Stripper Quench Rinse 110** 24		Etching Solution†	
	3647 Concentration mg/l	Mass Load kg/day	3643 Concentration mg/l	Mass Load kg/day	3645 Concentration mg/l	Mass Load kg/day	3648 Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS								
114 Antimony	<0.005		0.005		<0.005			
115 Arsenic	<0.003		<0.003		<0.003			
117 Beryllium	<0.001		<0.001		<0.001			
118 Cadmium	0.003		0.003		0.001	0.000003		
119 Chromium	0.004		0.003		0.001	0.000003		
120 Copper	0.015		0.046		0.019	0.000005		
122 Lead	0.019		0.161		0.012	0.00003		
123 Mercury	<0.001		<0.001		<0.001			
124 Nickel	0.057		0.07		0.005	0.00001		
125 Selenium	<0.003		<0.003		<0.003			
126 Silver	<0.003		<0.003		<0.003			
127 Thallium	<0.025		<0.025		<0.025			
128 Zinc	0.022		0.048		0.032	0.00008		
Total Toxic Inorganics	0.120		0.331		0.07	0.0002		
NON-CONVENTIONAL POLLUTANTS								
Aluminum	0.046		5.781		0.031	0.00008		
Barium	0.004		0.011		0.006	0.00002		
Boron	0.026 #		19.961 #		0.028 #	0.00007		
Calcium	1.718		2.371		0.258	0.0007		
Cobalt	<0.001		<0.001		<0.001			
Gold								
Iron	0.055		0.149		0.026	0.00007		
Magnesium	0.077		0.142		0.034	0.00009		
Manganese	0.001		0.006		0.001	0.000003		
Molybdenum	0.004		0.019		<0.001			
Palladium								
Platinum								
Sodium	0.071		18.315		0.143	0.0004		
Tellurium								
Tin	0.023		0.203		0.006	0.00002		
Titanium	0.002		0.036		0.001	0.000003		
Vanadium	0.001		0.081		0.001	0.000003		
Yttrium	0.005		<0.001		0.001	0.000003		
Phenols	0.014		0.016		0.001	0.000003		
Total Organic Carbon	30		<1.0		0.007	0.00002		
Fluoride	0.15		875		<1.0			
					0.24	0.0006		

** Estimated flow

† Inorganics and non-conventionals were not analyzed

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

TABLE 5-6 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 04294

Stream Description Flow (l/hr) Duration (hrs) Sample ID No.	Developer Quench Rinse		Acid Wastes		Stripper Quench Rinse		Etching Solution†	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	3647		3643		110** 24 3645		3648	
CONVENTIONAL POLLUTANTS								
Oil & Grease	3.0		<1.0		1.0	0.003		
Total Suspended Solids	<5.0		31.0		<5.0			
Biochemical Oxygen Demand	<4.0		<4.0		<4.0			
pH								

** Estimated flow

† Conventional pollutants were not analyzed

TABLE 5-6 (Continued);
SEMICONDUCTOR PROCESS WASTES
PLANT 04294

Stream Description Flow (L/hr) Duration (hrs) Sample ID No.	Acid Wastes		Effluent 6273 24 3652		Wafer Finishing Wastes 3641		Stripper Quench Rinse Batch 3644†		Photoresist Developer Batch 3646†	
	Concentration mg/L	Mass Load mg/day	Concentration mg/L	Mass Load mg/day	Concentration mg/L	Mass Load kg/day	Concentration mg/L	Mass Load kg/day	Concentration mg/L	Mass Load kg/day
TOXIC ORGANICS										
4 Benzene										
7 Chlorobenzene										
8 1,2,4-Trichlorobenzene			27.1	4.08						
11 1,1,1-Trichloroethane										
13 1,1-Dichloroethane										
18 Bis(2-chloroethyl)ether										
21 2,4,6-Trichlorophenol			0.013	0.002					<0.01 #	
22 P-chloro-m-cresol									0.043 #	
23 Chloroform			0.012	0.0018	0.046 #					
24 2-Chlorophenol			<0.01							
25 1,2-Dichlorobenzene			186.0	28.0						
26 1,3-Dichlorobenzene			7.4 ##	2.23						
27 1,4-Dichlorobenzene			7.4 ##	2.23						
34 2,4 -Dimethylphenol										
37 1,2-Diphenylhydrazine										
38 Ethylbenzene			0.107	0.016						
39 Fluoranthene										
44 Methylene chloride			0.101	0.015						
47 Bromoform					<0.01 #					
48 Dichlorobromomethane			<0.006		0.030 #					
51 Chlorodibromomethane					0.019 #					
54 Isophorone										
55 Naphthalene			1.504	0.226					<0.01 #	
57 2-Nitrophenol			0.039	0.006						
58 4-Nitrophenol										
62 N-Nitrosodiphenylamine									<0.01 #	
64 Pentachlorophenol			0.250	0.038	<0.01 #		0.032 #		0.925 #	
65 Phenol			0.170	0.026					0.052 #	
66 Bis(2-ethylhexyl) phthalate	<0.01 #		0.012	0.0018						
67 Butyl benzyl phthalate	<0.01 #									
68 Di-N-butyl phthalate	<0.01 #		0.017	0.0026	<0.01 #		0.013 #			
69 Di-N-octyl phthalate										
70 Diethyl phthalate	<0.01 #				<0.01 #				<0.01	
85 Tetrachloroethylene			0.143	0.022	<0.01 #					
86 Toluene			<0.003		<0.01 #					
87 Trichloroethylene			0.204	0.031	<0.01 #					
102 Alpha BHC									<0.005 #	
103 Beta BHC	<0.01 #						<0.005 #			
104 Gamma BHC										
121 Cyanide*			<0.005		<0.005 #					
Total Toxic Organics			230.472 ##	36.938	0.095 #		0.045 #		1.02 #	

* Not included in Total Toxic Organics summation.

† Volatile organics were not analyzed

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-6 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 04294

Stream Description Flow (l/hr) Duration (hrs) Sample ID No.	Acid Wastes†		Effluent		Wafer Finishing Wastes ‡		Stripper Quench Rinset		Photoresist Developer†	
	3650		3652		3641		Batch 3644		Batch 3646	
	Concentration mg/l	Mass Load mg/day	Concentration mg/l	Mass Load mg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS										
114 Antimony			<0.005		<0.005					
115 Arsenic			<0.003		0.003					
117 Beryllium			<0.001		<0.001					
118 Cadmium			0.003	0.0005	0.004					
119 Chromium			0.036	0.005	0.033					
120 Copper			0.103	0.016	0.078					
122 Lead			0.21	0.032	0.044					
123 Mercury			<0.001		<0.001					
124 Nickel			0.399	0.06	0.150					
125 Selenium			<0.003		<0.003					
126 Silver			0.013	0.002	<0.003					
127 Thallium			<0.025		<0.025					
128 Zinc			0.216	0.033	0.051					
Total Toxic Inorganics			0.980	0.1485	0.036					
NON-CONVENTIONAL POLLUTANTS										
Aluminum			0.247	0.037	0.359					
Barium			0.09	0.014	0.080					
Boron			0.253 ‡		0.224					
Calcium			72.448	--	55.968					
Cobalt			0.004	0.0006	0.028					
Gold										
Iron			0.477	0.072	8.472					
Magnesium			30.06	--	23.030					
Manganese			0.025	0.0038	0.055					
Molybdenum			0.016	0.002	0.017					
Palladium										
Platinum										
Sodium			115.147	--	92.237					
Tellurium										
Tin			0.078	0.012	0.075					
Titanium			0.006	0.0009	0.004					
Vanadium			0.214	0.032	0.170					
Yttrium			0.008	0.001	0.009					
Phenols			1.80	0.27	0.017					
Total Organic Carbon			20	3.01	11.0					
Fluoride			6.9	1.04	0.53					

† Toxic inorganics and non-conventionals were not analyzed.

‡ Data available but not transcribed from analytical sheets at proposal. (See note on page 5-5.)

TABLE 5-6 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 04294

Stream Description Flow (l/hr) Duration (hrs) Sample ID No.	Acid Wastes†		Effluent		Wafer Finishing Wastes #		Stripper Quench Rinse†		Photoresist Develop†	
	Concentration mg/l	Mass Load mg/day	Concentration mg/l	Mass Load mg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	3650		6273							
			24							
			3652		3641		Batch 3644		Batch 3646	

CONVENTIONAL POLLUTANTS

Oil & Grease		4.0	0.6	7	
Total Suspended Solids		14	2.11	135	
Biochemical Oxygen Demand		30	4.52	6.0	
pH					

† Conventional pollutants were not analyzed

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

TABLE 5-7

SEMICONDUCTOR PROCESS WASTES
PLANT 04296

Stream Description Flow (L/hr) Duration (hrs) Sample ID No.	Supply Water		Effluent		Scrubber Wastes	
	Concentration mg/L	Mass Load mg/day	Concentration mg/L	Mass Load mg/day	Concentration mg/L	Mass Load kg/day
	1798		1798		10	
	24		24		24	
	M16-0-0		M16-1-1		M16-2-1	
TOXIC ORGANICS						
4 Benzene						
8 1,2,4-Trichlorobenzene			4.5	0.194		
23 Chloroform						
24 2-Chlorophenol			0.09	0.0039		
25 1,2-Dichlorobenzene			4.5	0.194		
26 1,3-Dichlorobenzene			0.235	0.01		
27 1,4-Dichlorobenzene			0.235	0.01		
31 2,4-Dichlorophenol			0.01	0.0004		
37 1,2-Diphenylhydrazine						
38 Ethylbenzene						
44 Methylene chloride						
48 Dichlorobromomethane						
55 Naphthalene			0.190	0.008		
57 2-Nitrophenol			0.035	0.0015	0.70	0.00017
65 Phenol			3.5	0.151	0.045	0.00001
66 Bis(2-ethylhexyl) phthalate	0.290	0.013	0.05 ^B	0.002	0.750	0.00018
67 Butyl benzyl phthalate					0.013	0.000003
68 Di-N-butyl phthalate					0.280	0.00007
70 Diethyl phthalate					0.080	0.000019
85 Tetrachloroethylene						
86 Toluene						
87 Trichloroethylene						
121 Cyanide*	0.011	0.0005	0.002	0.0001	0.91	0.0002
Total Toxic Organics	0.290	0.013	13.335	0.575	1.868	0.00045
TOXIC INORGANICS						
114 Antimony	<0.0005		0.0007	0.00003	0.088	0.00002
115 Arsenic	<0.005		0.0068	0.00029	6.25	0.0015
117 Beryllium	<0.005		<0.005		<0.005	
118 Cadmium	<0.001		0.0003	0.00001	0.006	0.000001
119 Chromium	<0.025		1.15	0.05	1.14	0.00027
120 Copper	0.04	0.0017	0.005	0.0002	0.38	0.00009
122 Lead	0.24	0.01	0.0035	0.00015	0.42	0.0001
123 Mercury	<0.001		<0.001		<0.001	
124 Nickel	<0.025		<0.025		0.34	0.00008
125 Selenium	<0.005		<0.005		<0.005	

* Not included in Total Toxic Organics summation.

B = Present in sample blank

TABLE 5-7 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 04296

Stream Description Flow (l/hr) Duration (hrs) Sample ID No.	Supply Water 1798		Effluent 1798		Scrubber 10	
	M16-0-0		M16-1-1		M16-2-1	
	Concentration mg/l	Mass Load mg/day	Concentration mg/l	Mass Load mg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)						
126 Silver	<0.015		<0.015		<0.015	
127 Thallium	<0.0005		0.007	0.0003	0.0065	0.000002
128 Zinc	0.009	0.00039	0.029	0.0013	25.6	0.006
Total Toxic Inorganics	0.289	0.012	1.202	0.052	34.23	0.0081
NON-CONVENTIONAL POLLUTANTS						
Aluminum	NA		NA		NA	
Barium	NA		NA		NA	
Boron	NA		NA		NA	
Calcium	NA		NA		NA	
Cobalt	NA		NA		NA	
Gold	NA		NA		NA	
Iron	NA		NA		NA	
Magnesium	NA		NA		NA	
Manganese	NA		NA		NA	
Molybdenum	NA		NA		NA	
Palladium	NA		NA		NA	
Platinum	NA		NA		NA	
Sodium	NA		NA		NA	
Tellurium	NA		NA		NA	
Tin	NA		NA		NA	
Titanium	NA		NA		NA	
Vanadium	NA		NA		NA	
Yttrium	NA		NA		NA	
Phenols	<0.002		0.093	0.004	<0.002	
Total Organic Carbon	2.3	0.10	13.6	0.59	52	0.012
Fluoride	NA		NA		NA	
CONVENTIONAL POLLUTANTS						
Oil & Grease	8.7	0.38	6.8	0.29	7.7	0.0018
Total Suspended Solids	0.4	0.017	2.4	0.104	14	0.003
Biochemical Oxygen Demand	<3.0		30.0	1.295	<3.0	
pH	8.7		2.6		1.5	

NA = Not Analyzed

TABLE 5-8 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 06143

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes 2,509		Dilute Rinses 43,214		Effluent 42,496		Scrubber Wastes 2,509	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	24		24		24		24	
	3482		3483		3484		3485	
TOXIC INORGANICS								
114 Antimony	0.002		0.002	0.0021	0.002	0.002	0.001	0.00006
115 Arsenic	0.004		0.003	0.0031	0.006	0.006	0.002	0.00012
117 Beryllium	<0.001		<0.001		<0.001		<0.001	
118 Cadmium	<0.002		<0.002		<0.002		<0.002	
119 Chromium	0.189	0.011	<0.001		<0.001		<0.001	
120 Copper	0.055	0.003	0.049	0.051	1.76	1.80	0.012	0.0007
122 Lead	<0.041		<0.038		0.362	0.369	<0.039	
123 Mercury	<0.001		<0.001		<0.001		<0.001	
124 Nickel	<0.015	0.0009	<0.005		<0.005		<0.005	
125 Selenium	0.002	0.0001	0.002	0.002	0.006	0.006	<0.001	
126 Silver	<0.001		<0.001		<0.001		<0.001	
127 Thallium	<0.001		<0.001		<0.001		<0.001	
128 Zinc	0.062	0.0037	<0.001		0.225	0.229	0.022	0.0013
Total Toxic Inorganics	0.329	0.0187	0.056	0.582	2.361	2.412	0.037	0.0022
NON-CONVENTIONAL POLLUTANTS								
Aluminum	0.136	0.008	0.048	0.05	0.218	0.222	0.144	0.0087
Barium	0.016	0.00096	0.001		0.006	0.006	0.012	0.0007
Boron	0.069	0.004	<0.002		0.234	0.239	0.007	0.0004
Calcium	18.4		0.125		4.98		18.4	
Cobalt	<0.051		<0.048		<0.051		<0.048	
Gold	<0.001		<0.001		0.003	0.003	<0.001	
Iron	0.546	0.033	<0.001		<0.001		<0.001	
Magnesium	5.11		<0.024		1.44		5.05	
Manganese	0.025	0.0015	<0.001		0.104	0.106	0.012	0.0007
Molybdenum	<0.035		<0.033		<0.036		<0.034	
Palladium	<0.003		<0.003		<0.003		<0.003	
Platinum	<0.01		<0.01		<0.01		<0.01	
Sodium	13.4		1.5		147.0		14	
Tellurium	<0.002		0.003	0.003	0.002	0.002	<0.002	
Tin	0.027	0.002	<0.024		0.03	0.03	0.034	0.002
Titanium	<0.002		<0.002		<0.002		<0.002	
Vanadium	<0.002		<0.001		<0.002		<0.001	
Yttrium	<0.004		<0.003		<0.004		<0.003	
Phenols	18.125	1.09	0.041	0.043	2.438	2.49	0.114	0.0069
Total Organic Carbon	50.9	3.06	7.6	7.88	38	38.8	26.3	1.58
Fluoride	14.5	0.87	0.9	0.93	80	81.6	24.5	1.48

TABLE 5-8 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 06143

Stream Description	Scrubber Wastes		Dilute Rinses		Effluent		Scrubber Wastes	
	Concentration mg/l	Mass Load kg/day						
Flow (l /hr)	2,509		43,214		42,496		2,509	
Duration (hrs)	24		24		24		24	
Sample ID No.	3482		3483		3484		3485	
CONVENTIONAL POLLUTANTS								
Oil & Grease	1.57	0.09	3.41	3.54	5.46	5.57	12.67	0.76
Total Suspended Solids	0.3	0.018	0.3	0.31	3.3	3.37	1.4	0.08
Biochemical Oxygen Demand	22	1.32			16.8	17.1	12.6	0.76
pH								

TABLE 5-8 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 06143

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Dilute Rinses 43,214 24 3486		Effluent 47,701 24 3487		Scrubber Wastes 2,509 24 3488		Dilute Rinses 43,214 24 3489	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene								
5 Benzidine	<0.01 #		<0.01 #		<0.01 #		<0.01	
6 Carbon Tetrachloride					0.025	0.002		
7 Chlorobenzene			<0.01					
8 1,2,4-Trichlorobenzene			<0.01					
10 1,2-Dichloroethane			<0.01					
11 1,1,1-Trichloroethane	0.014	0.145	3.2	3.66	0.011	0.0007		
14 1,1,2-Trichloroethane	<0.01				0.033	0.002	0.019	0.020
23 Chloroform	<0.01				<0.01		<0.01 #	
24 2-Chlorophenol					0.018 ^B	0.001	<0.01	
25 1,2-Dichlorobenzene	<0.01		0.270 #	0.309	<0.01			
26 1,3-Dichlorobenzene			<0.01		0.018 #	0.001		
27 1,4-Dichlorobenzene			0.044	0.05	<0.01			
29 1,1-Dichloroethylene			0.02	0.023				
30 1,2-Transdichloroethylene					0.013	0.0008		
34 2,4-Dimethylphenol					<0.01			
37 1,2-Diphenylhydrazine					<0.01			
38 Ethylbenzene	<0.01		<0.01		0.015 ^B ##	<0.01	<0.01	
39 Fluoranthene			<0.01 ^B		<0.01 ^B		<0.01	
44 Methylene chloride	<0.01							
45 Methyl Chloride			<0.01					
46 Methyl Bromide								
48 Dichlorobromomethane			<0.01					
49 Trichlorofluoromethane								
51 Chlorodibromomethane					0.016	0.0010		
55 Naphthalene								
56 Nitrobenzene								
57 2-Nitrophenol			0.011	0.01	<0.01			
58 4-Nitrophenol					0.13	0.008		
65 Phenol	0.31	0.32	0.61	0.70	0.97	0.058	0.011	0.01
66 Bis(2-ethylhexyl)phthalate	<0.01		<0.01		<0.01		<0.01	
67 Butyl benzyl phthalate	<0.01				<0.01			
68 Di-N-butyl phthalate	<0.01		<0.01		<0.01		<0.01	
70 Diethyl phthalate					<0.01			
78 Anthracene								
81 Phenathrene								
84 Pyrene								
85 Tetrachloroethylene	<0.01		<0.01		0.074	0.0045		
86 Toluene	<0.01		<0.01 ^B		0.012 ^B	<0.01	<0.01	
87 Trichloroethylene					0.08 ^B	0.0048		
121 Cyanide*	0.01	0.01	0.01	0.01	0.01	0.001	0.01	0.01
Total Toxic Organics	0.324 ##	0.336	4.155 ##	4.757	1.415 ##	0.0852	0.03	0.031

* Not included in Total Toxic Organics summation.

B = Present in sample blank

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-8 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 06143

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Dilute Rinses 43,214 24 3486		Effluent 47,701 24 3487		Scrubber Wastes 2,509 24 3488		Dilute Rinses 43,214 24 3489	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS								
114 Antimony	0.002	0.0021	<0.001		0.002	0.0001	0.001	0.001
115 Arsenic	<0.001		0.003	0.003	0.001	0.00006	0.002	0.0021
117 Beryllium	<0.001		<0.001		<0.001		<0.001	
118 Cadmium	<0.002		<0.002		<0.002		<0.002	
119 Chromium	0.310	0.322	<0.001		<0.001		<0.001	
120 Copper	0.046	0.048	0.904	1.03	0.005	0.0003	<0.002	
122 Lead	<0.039		<0.039		<0.039		<0.039	
123 Mercury	<0.001		<0.001		<0.001		<0.001	
124 Nickel	0.135	0.14	<0.005		<0.005		<0.005	
125 Selenium	0.003	0.003	0.007	0.008	0.001	0.00006	<0.001	
126 Silver	<0.001		0.001	0.001	<0.001		0.001	0.001
127 Thallium	0.001	0.001	<0.001		<0.001		<0.001	
128 Zinc	1.84	1.91	0.05	0.057	<0.001		<0.001	
Total Toxic Inorganics	2.337	2.426	0.965	1.099	0.009	0.0005	0.004	0.0041
NON-CONVENTIONAL POLLUTANTS								
Aluminum	0.041	0.043	0.572	0.655	0.148	0.0089	0.024	0.025
Barium	0.001	0.001	0.007	0.008	0.013	0.0008	<0.001	
Boron	0.058	0.06	0.908	1.04	0.009	0.0005	0.022	0.023
Calcium	0.546		7.0		18.2		0.032	
Cobalt	<0.048		<0.049		<0.049		<0.048	
Gold	<0.001		0.002	0.0023	<0.001		<0.001	
Iron	1.23	1.28	<0.001		<0.001		<0.001	
Magnesium	0.147		2.11		5.14		<0.024	
Manganese	0.024	0.025	0.029	0.045	0.031	0.002	<0.001	
Molybdenum	<0.034		<0.034		<0.034		<0.034	
Palladium	<0.003		<0.003		<0.003		<0.003	
Platinum	<0.01		<0.01		<0.01		<0.01	
Sodium	<1.5		344		13.5		<1.5	
Tellurium	0.005	0.005	<0.002		<0.002	0.005	0.0052	
Tin	<0.024		<0.025		<0.024		<0.024	
Titanium	<0.002		0.012	0.014	<0.002		0.002	
Vanadium	<0.001		<0.001		<0.001		0.001	
Yttrium	<0.003		<0.003		<0.003		0.003	
Phenols	0.036	0.037	0.040	0.046	4.4	0.26	0.019	0.19
Total Organic Carbon	5.3	5.5	49.8	57.0	18.81	1.13	5.3	5.5
Fluoride	1.1	1.14	125	143.1	27.5	1.66	1.5	1.56

TABLE 5-8 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 06143

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Dilute Rinses		Effluent		Scrubber Wastes		Dilute Rinses	
	Concentration mg/l	Mass Load kg/day						
	43,214		47,701		2,509		43,214	
	24		24		24		24	
	3486		3487		3488		3489	
CONVENTIONAL POLLUTANTS								
Oil & Grease			11.67	13.4	0.24	0.01		
Total Suspended Solids	1.6	1.66	3.0	3.43	1.6	0.096	0.8	0.83
Biochemical Oxygen Demand	22	22.8	1.2	1.47	30	1.81		
pH								

TABLE 5-8 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 06143

Stream Description	Effluent	
Flow (l /hr)	46,002	
Duration (hrs)	24	
Sample ID No.	3490	
	Concentration	Mass Load
	mg/l	kg/day
TOXIC ORGANICS		
4 Benzene	<0.01 ^B	
5 Benzidine	<0.01	
6 Carbon Tetrachloride		
7 Chlorobenzene		
8 1,2,4-Trichlorobenzene	<0.01	
10 1,2-Dichloroethane		
11 1,1,1-Trichloroethane	7.7	8.5
14 1,1,2-Trichloroethane		
23 Chloroform	<0.01	
24 2-Chlorophenol	<0.01	
25 1,2-Dichlorobenzene	0.091	0.10
26 1,3-Dichlorobenzene	<0.01	
27 1,4-Dichlorobenzene	0.015	0.017
29 1,1-Dichloroethylene	0.071	0.08
30 1,2-Transdichloroethylene		
34 2,4-Dimethylphenol		
37 1,2-Diphenylhydrazine		
38 Ethylbenzene	<0.01 ^B	
39 Fluoranthene		
44 Methylene chloride	<0.01 ^B	
45 Methyl Chloride		
46 Methyl Bromide		
48 Dichlorobromomethane		
49 Trichlorofluoromethane		
51 Chlorodibromomethane		
55 Naphthalene	<0.01	
56 Nitrobenzene		
57 2-Nitrophenol	<0.01	
58 4-Nitrophenol	0.043	0.047
65 Phenol	0.31	0.34
66 Bis(2-ethylhexyl)phthalate	<0.01	
67 Butyl benzyl phthalate	<0.01	
68 Di-N-butyl phthalate	<0.01	
70 Diethyl phthalate		
78 Anthracene		
81 Phenathrene		
85 Tetrachloroethylene		
86 Toluene	<0.01 ^B	
87 Trichloroethylene		
121 Cyanide*	0.01	0.01
Total Toxic Organics	8.23	9.084

* Not included in Total Toxic Organics summation.

B = Present in sample blank

TABLE 5-8 (Continued)
 SEMICONDUCTOR PROCESS WASTES
 PLANT 06143

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent 46,002 24 3490	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS			
114 Antimony		<0.001	
115 Arsenic		0.01	0.011
117 Beryllium		<0.001	
118 Cadmium		<0.002	
119 Chromium		<0.001	
120 Copper		1.31	1.45
122 Lead		0.282	0.311
123 Mercury		<0.001	
124 Nickel		<0.005	
125 Selenium		0.002	0.002
126 Silver		0.001	0.001
127 Thallium		<0.001	
128 Zinc		0.128	0.14
Total Toxic Inorganics		1.733	1.915
NON-CONVENTIONAL POLLUTANTS			
Aluminum		3.2	3.53
Barium		0.011	0.012
Boron		0.748	0.83
Calcium		7.62	
Cobalt		<0.05	
Gold		<0.012	
Iron		<0.001	
Magnesium		2.29	
Manganese		0.044	0.049
Molybdenum		<0.035	
Palladium		<0.003	
Platinum		<0.01	
Sodium		554	
Tellurium		<0.002	
Tin		0.057	0.063
Titanium		0.004	0.004
Vanadium		<0.001	
Yttrium		<0.003	
Phenols		1.05	1.16
Total Organic Carbon		45.1	49.8
Fluoride		213	235.2

TABLE 5-8 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 06143

Stream Description	Effluent	
Flow (l /hr)	46,002	
Duration (hrs)	24	
Sample ID No.	3490	
	Concentration	Mass Load
	mg/l	kg/day
CONVENTIONAL POLLUTANTS		
Oil & Grease	2.44	2.69
Total Suspended Solids	3.8	4.20
Biochemical Oxygen Demand	24.4	26.9
pH		

TABLE 5-9
SEMICONDUCTOR PROCESS WASTES
PLANT 30167

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Supply Water 205020 24 M19-0		Acid Wastes 24529 24 M19-2		Treated Acid Wastes 24529 24 M19-3		Influent to Treatment 205043 24 M19-4	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene								
7 Chlorobenzene								
8 1,2,4-Trichlorobenzene								
11 1,1,1-Trichloroethane								
13 1,1-Dichloroethane								
23 Chloroform					0.013	0.008	0.011	0.054
24 2-Chlorophenol					0.006	0.004		
25 1,2-Dichlorobenzene								
44 Methylene chloride			0.01	0.006				
51 Chlorodibromomethane					0.005	0.003		
55 Naphthalene								
57 2-Nitrophenol			0.147	0.087	0.140	0.082		
58 4-Nitrophenol								
65 Phenol								
66 Bis(2-ethylhexyl)phthalate	0.01	0.05	0.018	0.011	0.034	0.020	0.536 ^B	2.638
67 Butyl benzyl phthalate								
68 Di-N-butyl phthalate								
69 Di-N-octyl phthalate								
70 Diethyl phthalate							0.01	0.049
85 Tetrachloroethylene	0.03	0.15	0.007	0.004	0.085	0.050	0.290	1.427
86 Toluene							0.01	0.049
87 Trichloroethylene	0.009	0.04					0.0365	0.180
121 Cyanide*	0.002	0.01	0.35	0.206	0.110	0.065	<0.001	
Total Toxic Organics	0.030 #	0.24	0.165 #	0.097	0.272 #	0.160	0.874 #	4.301
TOXIC INORGANICS								
114 Antimony	<0.001		<0.002		<0.001			
115 Arsenic	<0.01		<0.01		<0.01		0.001	0.005
117 Beryllium	<0.01		<0.01		<0.01		<0.01	
118 Cadmium	<0.001		<0.01		<0.01		<0.01	
119 Chromium	<0.005		0.004	0.002	<0.001		<0.001	
120 Copper	<0.01		22.8	13.42	0.055	0.032	0.025	0.123
122 Lead	<0.001		2.2	1.295	0.145	0.085	0.035	0.172
123 Mercury	<0.001		5.35	3.15	0.005	0.003	0.008	0.039
124 Nickel	<0.025		<0.001		<0.001		<0.001	
125 Selenium	<0.005		0.69	0.406	0.065	0.038	0.035	0.172
			<0.005		<0.005		<0.005	

*Not Included in Total Toxic Organics summation.

#Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

B = Present in sample blank

TABLE 5-9 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 30167

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Supply Water 205020 24 M19-0		Acid Wastes 24529 24 M19-2		Treated Acid Wastes 24529 24 M19-3		Influent to Treatment 205043 24 M19-4	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	<0.01		0.025	0.014	<0.01		<0.01	
127 Thallium	0.001	0.005	0.005	0.0029	0.012	0.007	0.012	0.059
128 Zinc	<0.01		<0.01		<0.01		<0.01	
Total Toxic Inorganics	0.001	0.541	31.07	18.29	0.282	0.166	0.116	0.571
NON-CONVENTIONAL POLLUTANTS								
Aluminum	NA		NA		NA		NA	
Barium	NA		NA		NA		NA	
Boron	NA		NA		NA		NA	
Calcium	NA		NA		NA		NA	
Cobalt	NA		NA		NA		NA	
Gold	NA		NA		NA		NA	
Iron	NA		NA		NA		NA	
Magnesium	NA		NA		NA		NA	
Manganese	NA		NA		NA		NA	
Molybdenum	NA		NA		NA		NA	
Palladium	NA		NA		NA		NA	
Platinum	NA		NA		NA		NA	
Sodium	NA		NA		NA		NA	
Tellurium	NA		NA		NA		NA	
Tin	NA		NA		NA		NA	
Titanium	NA		NA		NA		NA	
Vanadium	NA		NA		NA		NA	
Yttrium	NA		NA		NA		NA	
Phenols	<0.002		0.004	0.002	0.004	0.0024	<0.002	
Total Organic Carbon	56	275.5	414	243.7	255	150.1	47	231.3
Fluoride	4.2	20.67	760	447.4	37	7.418		
CONVENTIONAL POLLUTANTS								
Oil & Grease	2.0	9.84	2.8	1.648	3.1	1.825	1.0	4.921
Total Suspended Solids	1.2	5.9	5.6	3.297	71	41.80	203	999.0
Biochemical Oxygen Demand	3	14.8	<3		550	323.8	11	54.13
pH	7.8		1.2		11.9		9.4	

NA=Non-toxic metals not analyzed

TABLE 5-9 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 30167

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent 205043 24 M19-5		Wafer Finishing Wastes 3950 24 M19-1		Equipment Cleaning 1577 24 M19-6		Display Panel Production Effluent 5520 24 M19-7	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene								
7 Chlorobenzene								
8 1,2,4-Trichlorobenzene								
11 1,1,1-Trichloroethane	0.006	0.03						
13 1,1-Dichloroethane								
23 Chloroform								
24 2-Chlorophenol								
44 Methylene chloride	0.021	0.10						
51 Chlorodibromomethane					0.023 #	0.0009	0.023 #	
55 Naphthalene	0.006	0.03			0.030 #	0.001		
57 2-Nitrophenol								
58 4-Nitrophenol								
65 Phenol								
66 Bis(2-ethylhexyl)phthalate	0.08	0.039	0.250 ^B #	0.024	0.130 #	0.005	0.045 #	
67 Butyl benzyl phthalate								
68 Di-N-butyl phthalate			0.006 #	0.006				
69 Di-N-octyl phthalate								
70 Diethyl phthalate								
85 Tetrachloroethylene	0.0505	0.25			0.005 #	0.0002		
86 Toluene	0.057	0.28						
87 Trichloroethylene	0.01	0.05					0.023 #	
121 Cyanide*	0.011	0.05	0.002 #	0.0002				
Total Toxic Organics	0.2085 ##	1.14	0.250 #	0.024	0.183#	0.007	0.091#	
TOXIC INORGANICS								
114 Antimony	<0.001		<0.001 #		0.001 #	0.00004	<0.001 #	
115 Arsenic	<0.01		<0.01 #		<0.01 #		0.011 #	
117 Beryllium	<0.01		<0.01 #		<0.01 #		<0.01 #	
118 Cadmium	<0.001		0.005 #	0.0005	0.016 #	0.0006	<0.001 #	
119 Chromium	0.05	0.25	<0.005 #		1.125 #	0.043	0.040 #	
120 Copper	0.035	0.17	0.020 #	0.002	5.360 #	0.203	0.098 #	
122 Lead	0.005	0.02	0.002 #	0.0002	12.200 #	0.462	<0.005 #	
123 Mercury	<0.001		<0.001 #		<0.001 #		<0.001 #	
124 Nickel	<0.025		<0.025 #		0.135 #	0.005	0.035 #	
125 Selenium	<0.005		<0.005 #		<0.005 #		<0.005 #	

* Not Included in Total Toxic Organics summation.

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-9 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 30167

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent 205043 24 M19-5		Wafer Finishing Wastes 3950 24 M19-1		Equipment Cleaning 1577 24 M19-6		Display Panel Production Effluent 5520 24 M19-7	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	<0.01		<0.010 #		<0.010 #		<0.010 #	
127 Thallium	0.003	0.01	<0.001 #		0.002 #	0.00008	<0.001 #	
128 Zinc	<0.01		<0.01 #		0.173 #	0.007	0.013 #	
Total Toxic Inorganics	0.093	0.46	0.020 #	0.002	19.009 #	0.719	0.197 #	
NON-CONVENTIONAL POLLUTANTS								
Aluminum	NA		NA		NA		NA	
Barium	NA		NA		NA		NA	
Boron	NA		NA		NA		NA	
Calcium	NA		NA		NA		NA	
Cobalt	NA		NA		NA		NA	
Gold	NA		NA		NA		NA	
Iron	NA		NA		NA		NA	
Magnesium	NA		NA		NA		NA	
Manganese	NA		NA		NA		NA	
Molybdenum	NA		NA		NA		NA	
Palladium	NA		NA		NA		NA	
Platinum	NA		NA		NA		NA	
Sodium	NA		NA		NA		NA	
Tellurium	NA		NA		NA		NA	
Tin	NA		NA		NA		NA	
Titanium	NA		NA		NA		NA	
Vanadium	NA		NA		NA		NA	
Yttrium	NA		NA		NA		NA	
Phenols	0.008 #		<0.002 #		0.006 #	0.0002	<0.002 #	
Total Organic Carbon	97 #		132 #	12.5	107 #	40.8	68 #	
Fluoride					60 #	2.27		
CONVENTIONAL POLLUTANTS								
Oil & Grease	17.4	85.62	6.0 #	0.57	2.8 #	0.106	2.1 #	
Total Suspended Solids	350	1722.2	325 #	30.8	0.6 #	0.023	3 #	
Biochemical Oxygen Demand	70	344.4	56 #	5.31	<3 #			
pH	8.8		8.2 #		1.0 #		8.1 #	

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)
NA = Not analyzed.

TABLE 5-9 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 30167

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent 189250 24 3314		Influent to Treatment 189250 24 3315		Acid Wastes 20187 24 3316		Treated Acid Wastes 20187 24 3317	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene								
7 Chlorobenzene								
8 1,2,4-Trichlorobenzene								
11 1,1,1-Trichloroethane	<0.01							
13 1,1-Dichloroethane								
23 Chloroform								
24 2-Chlorophenol								
44 Methylene chloride	0.016	0.073	0.001	0.005	0.016	0.008	0.006	0.003
51 Chlorodibromomethane								
55 Naphthalene								
57 2-Nitrophenol								
58 4-Nitrophenol								
65 Phenol								
66 Bis(2-ethylhexyl)phthalate								
67 Butyl benzyl phthalate								
68 Di-N-butyl phthalate								
69 Di-N-octyl phthalate					0.001			
70 Diethyl phthalate								
85 Tetrachloroethylene	0.013	0.059	0.012	0.055	0.047	0.023	0.042	0.020
86 Toluene		<0.01						
87 Trichloroethylene	0.006	0.027	0.005	0.023	0.002	0.001	0.001	0.001
121 cyanide*	<0.04		<0.04		<0.04		<0.04	
Total Toxic Organics	0.029 **	0.159	0.012 **	0.083	0.063 **	0.032	0.042 **	0.024
TOXIC INORGANICS								
114 Antimony	<0.003		<0.003		<0.003		<0.003	
115 Arsenic	0.014	0.064	0.010	0.045	0.004	0.002	<0.003	
117 Beryllium	0.002	0.009	0.002	0.009	0.002	0.001	<0.001	
118 Cadmium	0.015	0.068	0.018	0.082	0.030	0.015	<0.001	
119 Chromium	0.115	0.522	0.027	0.123	19.00	9.205	0.128	0.062
120 Copper	0.158	0.718	0.045	0.204	1.742	0.844	0.050	0.024
122 Lead	0.040	0.182	<0.010		3.675	1.780	0.18	0.009
123 Mercury	<0.003		0.003	0.014	0.002	0.001	0.001	0.001
124 Nickel	0.108	0.491	0.054	0.245	1.956	0.948	0.121	0.059
125 Selenium	<0.003		<0.003		<0.003		<0.003	

*Not Included in Total Toxic Organics summation.

**Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-9 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 30167

Stream Description Flow (l/hr) Duration (hrs) Sample ID No.	Effluent		Influent to Treatment		Acid Wastes		Treated Acid Wastes	
	189250 24 3314	189250 24 3314	189250 24 3315	189250 24 3315	20187 24 3316	20187 24 3316	20187 24 3317	20187 24 3317
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	0.025	0.114	0.015	0.068	0.011	0.005	0.020	<0.01
127 Thallium	0.120	0.545	0.05	0.227	0.40	0.019	0.19	0.092
128 Zinc	0.358	1.626	0.162	0.736	0.197	0.095	0.033	0.016
Total Toxic Inorganics	0.955	4.334	0.386	1.753	26.659	12.916	13.039	6.317
NON-CONVENTIONAL POLLUTANTS								
Aluminum	1.352	6.141	0.986	4.478	4.440	2.151	<0.001	
Barium	0.089	0.404	0.053	0.241	0.018	0.009	<0.001	
Boron	0.353	1.603	0.306	1.390	12.145	5.884	0.571	0.277
Calcium	618.62		313.02		4.155		1090.0	
Cobalt	0.050	0.227	0.042	0.191	0.041	0.202	<0.001	
Gold								
Iron	7.571	34.387	5.404	24.545	1.025	0.50	0.071	0.034
Magnesium	55.39		46.810		3.325		0.783	
Manganese	0.217	0.986	0.059	0.268	22.37	10.840	0.133	0.064
Molybdenum	0.065	0.295	0.052	0.236	0.198	0.10	0.158	0.077
Palladium								
Platinum								
Sodium	488.93		504.23		1400.0		231.73	
Tellurium								
Tin	0.121	0.550	0.106	0.481	0.270	0.131	<0.001	
Titanium	<0.030		<0.03		<0.03		0.024	0.012
Vanadium	0.385	1.75	0.339	1.540	0.134	0.065	<0.001	
Yttrium	0.064	0.291	0.056	0.254	0.069	0.033	0.005	0.002
Phenols	<0.004		<0.004		<0.004		<0.004	
Total Organic Carbon	70.0	317.94	90.0	408.78	400.0	193.80	250.0	121.12
Fluoride	1.5	6.813	1.9	8.63	306.0	148.25	9.5	4.603
CONVENTIONAL POLLUTANTS								
Oil & Grease	1.3	5.91	1.2	5.450	1.3	0.630	0.3	0.145
Total Suspended Solids	12.27	5573.03	145.0	658.59	2.0	0.970	66.2	32.073
Biochemical Oxygen Demand	91.8	477.41	116.0	526.87	704.0	341.10	452.0	219.0
pH								

TABLE 5-9 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 30167

Stream Description	Wafer Finishing Wastes	
Flow (l /hr)	2059	
Duration (hrs)	24	
Sample ID No.	3318	
	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS		
4 Benzene		
7 Chlorobenzene		
8 1,2,4-Trichlorobenzene		
11 1,1,1-Trichloroethane		
13 1,1-Dichloroethane		
23 Chloroform		
24 2-Chlorophenol		
44 Methylene chloride	0.009	<0.001
51 Chlorodibromomethane		
55 Naphthalene		
57 2-Nitrophenol		
58 4-Nitrophenol		
65 Phenol		
66 Bis(2-ethylhexyl)phthalate		
67 Butyl benzyl phthalate		
68 Di-N-butyl phthalate		
69 Di-N-octyl phthalate		
70 Diethyl phthalate		
85 Tetrachloroethylene	0.002	<0.001
86 Toluene		
87 Trichloroethylene	0.018	0.001
121 Cyanide*	<0.004	
Total Toxic Organics	0.018 ##	0.001
TOXIC INORGANICS		
114 Antimony	<0.003	
115 Arsenic	<0.003	
117 Beryllium	<0.001	
118 Cadmium	<0.001	
119 Chromium	<0.020	
120 Copper	0.092	0.005
122 Lead	0.015	<0.001
123 Mercury	0.002	<0.001
124 Nickel	<0.028	
125 Selenium	<0.003	

*Not Included in Total Toxic Organics summation.

##Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-9 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 30167

Stream Description	Wafer Finishing Wastes	
Flow (t /hr)	2059	
Duration (hrs)	24	
Sample ID No.	3318	
	Concentration mg/t	Mass Load kg/day
TOXIC INORGANICS (Continued)		
126 Silver	<0.002	
127 Thallium	<0.020	
128 Zinc	0.047	0.002
Total Toxic Inorganics	0.156	0.008
NON-CONVENTIONAL POLLUTANTS		
Aluminum	<0.001	
Barium	<0.001	
Boron	1.194	0.059
Calcium	8.156	
Cobalt	<0.001	
Gold		
Iron	<0.001	
Magnesium	6.457	
Manganese	<0.001	
Molybdenum	<0.025	
Palladium		
Platinum		
Sodium	148.224	
Tellurium		
Tin	0.037	0.002
Titanium	<0.03	
Vanadium	<0.001	
Yttrium	<0.001	
Phenols	0.11	0.001
Total Organic Carbon	70.0	3.46
Fluoride	<0.10	
CONVENTIONAL POLLUTANTS		
Oil & Grease	14.1	0.697
Total Suspended Solids	344.0	17.0
Biochemical Oxygen Demand	69.0	3.41
pH		

TABLE 5-10 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 35035

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes		Dilute Rinses		Effluent		Dilute Rinses	
	Concentration mg/l	Mass Load kg/day						
	50		6865		4778		9469	
	24		24		24		24	
	3718		3719		3720		3721	
TOXIC INORGANICS (Continued)								
126 Silver	<0.001		<0.001		0.002	0.0002	<0.001	
127 Thallium	<0.001		<0.001		<0.001		<0.001	
128 Zinc	<0.001		<0.001		0.035	0.004	0.014	0.003
Total Toxic Inorganics	0.019	0.000026	0.004	0.0006	0.367	0.041	0.02	0.004
NON-CONVENTIONAL POLLUTANTS								
Aluminum	0.253	0.0003	0.022	0.004	0.21	0.024	0.041	0.009
Barium	<0.001		<0.001		0.001	0.0001	<0.001	
Boron	0.372	0.0004	0.215	0.03	0.639	0.07	0.186	0.04
Calcium	5.80		<0.005		10.1		<0.005	
Cobalt	<0.05		<0.05		<0.054		<0.05	
Gold	<0.002		<0.002		I		<0.002	
Iron	<0.001		<0.001		0.04	0.005	<0.001	
Magnesium	8.33		0.077		1.82		0.121	
Manganese	0.033	0.00004	<0.001		<0.001		<0.001	
Molybdenum	<0.035		<0.035		<0.038		<0.035	
Palladium	<0.003		<0.003		0.006	0.0007	<0.003	
Platinum	<0.003		<0.003		<0.003		<0.003	
Sodium	27.20		<1.50		1860		<1.5	
Tellurium	0.01	0.00001	<0.006		<0.015		<0.006	
Tin	<0.025		<0.025		<0.027		<0.025	
Titanium	0.004	0.000004	<0.002		0.002	0.0002	<0.002	
Vanadium	0.013	0.00002	0.002	0.0003	0.006	0.0007	0.004	0.0009
Yttrium	<0.003		<0.003		<0.004		<0.003	
Lithium	0.006	0.000007	0.001	0.0016	0.063	0.007	0.001	0.0002
Phenols	135.0	0.162	<0.001		0.31	0.36	<0.001	
Total Organic Carbon	177.0	0.21	1.2	0.20	102	11.7	0.8	0.18
Fluoride	119	0.143	0.35	0.058	16.3	1.87	0.21	0.05
CONVENTIONAL POLLUTANTS								
Oil & Grease	5.0	0.006	1.4	0.23	1.2	0.138		
Total Suspended Solids	24.8	0.03			1.3	0.15		
Biochemical Oxygen Demand	471	0.57						
pH								

I = interference present

TABLE 5-10 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 35035

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent 8740 24 3722		Dilute Rinses 7904 24 3723		Effluent 7681 24 3724	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS						
4 Benzene						
8 1,2,4-Trichlorobenzene	5.200	1.091	0.0096	0.0018	5.300	0.977
11 1,1,1-Trichloroethane			0.0009	0.0002		
13 1,1-Dichloroethane						
23 Chloroform	0.0055	0.0012	0.0054	0.001	0.0092	0.0017
24 2-Chlorophenol	0.0015	0.0003			0.0083	0.0015
25 1,2-Dichlorobenzene						
26 1,3-Dichlorobenzene	0.0027	0.0006			0.0032	0.0006
27 1,4-Dichlorobenzene					0.018	0.003
38 Ethylbenzene			0.0002	0.00004	0.0005	0.00009
44 Methylene chloride	0.0075	0.0016	0.013	0.0025	0.011	0.002
55 Naphthalene	0.086	0.018	0.046	0.0087	0.130	0.024
57 2-Nitrophenol	0.018	0.0038			0.024	0.004
58 4-Nitrophenol						
65 Phenol	0.263	0.055	0.0011	0.0002	0.44	0.081
66 Bis(2-ethylhexyl)phthalate	0.013	0.003	0.003	0.00057	0.0057	0.001
68 Di-N-butyl phthalate	0.0022	0.0005	0.0009	0.0002	0.0012	0.0002
70 Diethyl phthalate						
85 Tetrachloroethylene	0.0002	0.00004	0.0003	0.00006	0.0002	0.00004
86 Toluene	0.013	0.003	0.0072	0.0014	0.0085	0.0016
87 Trichloroethylene	0.0087	0.0018	0.0049	0.00093	0.0066	0.0012
121 Cyanide*	<0.005		<0.005		<0.005	
Total Toxic Organics	5.593 **	1.173	0.059 **	0.0112	5.923 **	1.092
TOXIC INORGANICS						
114 Antimony	0.10	0.02	0.002	0.0004	I	
115 Arsenic	I		I		I	
117 Beryllium	<0.001		<0.001		<0.001	
118 Cadmium	0.004	0.0008	<0.002		0.002	0.00037
119 Chromium	0.005	0.001	<0.001		<0.001	
120 Copper	0.049	0.01	<0.002		0.059	0.11
122 Lead	<0.04		<0.04		<0.04	
123 Mercury	<0.001		<0.001		<0.001	
124 Nickel	0.022	0.005	<0.005		0.015	0.0028
125 Selenium	0.044	0.009	0.003	0.0006	I	

B = present sample blank

I = interferences present

##Data incorrectly transcribed at proposal. (See note on page 5-5.)

*Not Included in Total Toxic Organics summation.

TABLE 5-10 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 35035

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent 8740 24 3722		Dilute Rinses 7904 24 3723		Effluent 7681 24 3724	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)						
126 Silver	0.001	0.0002	0.002	0.0004	0.002	0.00037
127 Thallium	<0.001		<0.001		<0.001	
128 Zinc	0.184	0.039	<0.001		0.067	0.012
Total Toxic Inorganics	0.409	0.085	0.007	0.0014	0.145	0.027
NON-CONVENTIONAL POLLUTANTS						
Aluminum	0.263	0.055	<0.01		0.16	0.03
Barium	0.004	0.0008	<0.001		0.002	0.0004
Boron	0.372	0.078	0.015	0.003	0.43	0.079
Calcium	21.4		<0.005		16.10	
Cobalt	<0.05		<0.05		<0.05	
Gold	I		<0.002		I	
Iron	0.483	0.101	<0.001		0.068	0.013
Magnesium	3.79		0.089		2.76	
Manganese	0.002	0.0004	<0.001		0.002	0.0004
Molybdenum	0.046	0.0096	<0.035		<0.035	
Palladium	0.004	0.0008	<0.003		0.007	0.0013
Platinum	0.003	0.0006	<0.003		<0.003	
Sodium	1130		<1.50		1400	
Tellurium	I		<0.006		<0.006	
Tin	<0.025		<0.025		<0.025	
Titanium	0.006	0.0013	<0.002		0.005	0.0009
Vanadium	0.013	0.0027	0.003	0.00063	0.007	0.0013
Yttrium	<0.003		<0.003		<0.003	
Lithium	0.018	0.0038	0.001	0.0002	0.04	0.0074
Phenols	0.53	0.111	<0.001		0.32	0.059
Total Organic Carbon	78	16.36	0.5	0.10	36	6.64
Fluoride	8.6	1.80	0.25	0.05	11.2	2.065
CONVENTIONAL POLLUTANTS						
Oil & Grease						
Total Suspended Solids	2.4	0.503		0.20		
Biochemical Oxygen Demand		0	1.0			
pH						

I = interferences present

TABLE 5-11
SEMICONDUCTOR PROCESS WASTES
PLANT 36133

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Treated Acid Wastes 337 24 3779		Effluent 272,353 24 3780		Acid Wastes† 189 24 3781		Treated Acid Wastes† 481 24 3782	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene	<0.01							
7 Chlorobenzene	<0.01							
8 1,2,4-Trichlorobenzene			<0.01					
23 Chloroform	<0.01 ^B		0.530 #	3.46				
25 1,2-Dichlorobenzene			0.02 ^B	0.13				
26 1,3-Dichlorobenzene	<0.01		0.053 #	0.346				
27 1,4-Dichlorobenzene								
29 1,1-Dichloroethylene			<0.01					
32 1,2-Dichloropropane	<0.01		<0.01					
37 1,2-Diphenylhydrazine	0.022	0.00018						
38 Ethylbenzene	<0.01							
44 Methylene chloride	<0.01 ^B		0.180 ^B #	1.177				
45 Methyl chloride	<0.01							
48 Dichlorobromomethane	<0.01							
51 Chlorodibromomethane								
62 N-nitrosodiphenylamine	0.039	0.00032	<0.01					
65 Phenol			<0.01					
66 Bis(2-ethylhexyl)phthalate								
85 Tetrachloroethylene	<0.01		<0.01					
86 Toluene	<0.01 ^B		<0.01 ^B					
87 Trichloroethylene	0.03 ^B	0.00024	0.052 ^B	0.340				
89 Aldrin	<0.005							
90 Dieldrin	<0.005							
101 Heptachlor epoxide	<0.005							
102 Alpha BHC	<0.005		<0.005 #					
103 Beta BHC	<0.005							
104 Gamma BHC	<0.005							
105 Delta BHC	<0.005		<0.005 #					
121 Cyanide	0.005	0.00004	<0.005					
Xylene	<0.01		<0.005		<0.005		<0.005	
Total Toxic Organics	0.091	0.0007	0.835 ##	5.458				
TOXIC INORGANICS								
114 Antimony	<0.005		<0.005		<0.005		<0.005	
115 Arsenic	0.002	0.00002	0.001	0.007	0.119	0.0003	0.003	0.00003
117 Beryllium	<0.003		0.001	0.007	<0.003		<0.003	
118 Cadmium	<0.003		0.007	0.046	<0.003		<0.003	
119 Chromium	<0.02		0.059	0.39	408.000	1.097	1.07	0.012

†Organics not analyzed.

B = Present in sample blank

#Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

##Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-11

SEMICONDUCTOR PROCESS WASTES
PLANT 36133

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Treated Acid Wastes		Effluent		Acid Wastes		Treated Acid Wastes	
	Concentration mg/l	Mass Load kg/day						
	337		272,353		189		481	
	24		24		24		24	
	3779		3780		3781		3782	
TOXIC INORGANICS (Continued)								
120 Copper	0.16	0.0013	0.115	0.75	3.746	0.010	0.09	0.001
122 Lead	0.045	0.00036	0.085	0.56	0.150	0.0004	0.04	0.0005
123 Mercury	0.011	0.00009	<0.001		<0.001		<0.001	
124 Nickel	0.22	0.0018	0.531	3.47	0.20	0.0005	0.20	0.002
125 Selenium	<0.005		<0.005		0.007	0.00002	<0.005	
126 Silver	0.015	0.00012	0.005		0.03	0.00008	0.020	0.002
127 Thallium	<0.03	0	<0.03		<0.03		<0.03	
128 Zinc	0.087	0.0007	0.04	0.26	0.429	0.001	0.432	0.005
Total Toxic Inorganics	0.54	0.0044	0.844	5.49	412.28	1.109	1.855	0.023
NON-CONVENTIONAL POLLUTANTS								
Aluminum	0.411	0.003	0.231	1.51	320.06	0.86	0.411	0.005
Barium			0.023	0.15				
Boron	<3.0		0.248	1.62	697	1.87	<3.0	
Calcium	425.23		153.4	-	825.18	-	332.94	
Cobalt	<0.02		0.01	0.065	0.14	0.0004	0.02	0.0002
Gold	0.029	0.0002	0.051	0.33	<0.02		<0.02	
Iron			0.092	0.60				
Magnesium			12.6					
Manganese			0.011	0.072				
Molybdenum	0.042	0.0003	0.035	0.229			0.044	0.0005
Palladium	0.04	0.0003	<0.04		<0.04		<0.04	
Platinum	<0.05		<0.05		<0.05		<0.05	
Sodium			199.5					
Tellurium	<0.02		<0.02		<0.02		<0.02	
Tin			0.006	0.039				
Titanium	<0.02		0.105	0.686	11.32	0.03	<0.02	
Vanadium			0.105	0.686				
Yttrium			0.023	0.150				
Phenols	<0.001		0.021	0.137	0.103	0.00028	0.004	0.00005
Total Organic Carbon	1537	12.4	10	65.4	2777	7.465	957	11.05
Fluoride	20.1	0.16	5.42	35.4	50,000	134.4	24	0.28
CONVENTIONAL POLLUTANTS								
Oil & Grease	2.0	0.016	2.4	15.7	5.1	0.014	9.8	0.113
Total Suspended Solids	176	1.42	2	1307	5760	15.483	1930	22.28
Biochemical Oxygen Demand	3700	29.9	18	117.7	241	0.653	2275	26.26
pH								

TABLE 5-11 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 36133

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent [†]		Acid Wastes [†]		Treated Acid Wastes [†]		Effluent [†]	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	280,020		189		281		285,800	
	24		24		24		24	
	3783		3785		3786		3787	
TOXIC ORGANICS								
4 Benzene								
7 Chlorobenzene								
23 Chloroform								
25 1,2-Dichlorobenzene								
26 1,3-Dichlorobenzene								
27 1,4-Dichlorobenzene								
32 1,2-Dichloropropane								
37 1,2-Diphenylhydrazine								
38 Ethylbenzene								
44 Methylene chloride								
45 Methyl chloride								
48 Dichlorobromomethane								
51 Chlorodibromomethane								
62 N-nitrosodiphenylamine								
65 Phenol								
66 Bis(2-ethylhexyl)phthalate								
85 Tetrachloroethylene								
86 Toluene								
87 Trichloroethylene								
89 Aldrin								
90 Dieldrin								
101 Heptachlor epoxide								
102 Alpha BHC								
103 Beta BHC								
104 Gamma BHC								
105 Delta BHC								
121 Cyanide*	<0.005		<0.005		0.477	0.0032	<0.005	
Total Toxic Organics								
TOXIC INORGANICS								
114 Antimony	<0.005		<0.005		<0.005		<0.005	
115 Arsenic	0.002	0.013	0.055	0.0001	0.002	0.00003	0.002	0.014
117 Beryllium	0.001	0.007	<0.003		<0.003		0.001	0.007
118 Cadmium	0.006	0.04	<0.003		<0.003		0.007	0.05
119 Chromium	0.058	0.39	26.31	0.059	0.09	0.0005	0.054	0.037

*Not included in Total Organics summation.

†Organics not analyzed.

TABLE 5-11

SEMICONDUCTOR PROCESS WASTES
PLANT 36133

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent		Acid Wastes		Treated Acid Wastes		Effluent	
	280,020		189		281		285,800	
	24		24		24		24	
	3783		3785		3786		3787	
	Concentration mg/l	Mass Load kg/day						
TOXIC INORGANICS (Continued)								
120 Copper	0.12	0.81	1.07	0.002	0.06	0.0004	0.134	0.92
122 Lead	0.083	0.56	<0.02		<0.02		0.10	0.69
123 Mercury	0.012	0.08	0.005	0.00001	0.01		0.011	0.07
124 Nickel	0.523	3.51	0.09	0.00002	0.18	0.001	0.596	4.09
125 Selenium	<0.005		0.007	0.00002	<0.005		0.009	0.06
126 Silver	0.005	0.03	0.01	0.00002	0.02	0.0001	0.005	0.03
127 Thallium	<0.03		<0.003		<0.03		<0.03	
128 Zinc	0.03	0.20	0.179	0.0004	0.136	0.0008	0.038	0.26
Total Toxic Inorganics	0.84	5.64	27.726	0.062	0.518	0.0029	0.957	6.561
NON-CONVENTIONAL POLLUTANTS								
Aluminum	0.215	1.44	173.83	0.39	0.793	0.004	0.231	1.58
Barium	0.022	0.15					0.023	0.16
Boron	0.289	1.94	41.0	0.09	<3.00		0.226	1.53
Calcium	154.8		215.29		578.83		174.10	
Cobalt	0.011	0.07	<0.02		<0.02		0.009	0.06
Gold	0.056	0.38	<0.02		<0.02		0.04	0.27
Iron	0.081	0.54					0.089	0.61
Magnesium	13.22						13.55	
Manganese	0.011	0.07					0.011	0.075
Molybdenum	0.037	0.25	0.11	0.0002	0.032	0.0002	0.043	0.295
Palladium	<0.04		<0.04		<0.04		<0.04	
Platinum	<0.05		<0.05		<0.05		<0.05	
Sodium	225.62						257.12	
Tellurium	<0.02		<0.02		<0.02		<0.02	
Tin	0.002	0.013					0.0	0.0
Titanium	0.008	0.054	10.83	0.024	<0.02		0.007	0.048
Vanadium	0.105	0.71					0.109	0.75
Yttrium	0.022	0.15					0.028	0.19
Phenols	0.014	0.094	0.105	0.0002	0.015	0.0001	0.006	0.041
Total Organic Carbon	11.4	76.61	967	2.16	655	4.42	1.8	12.35
Fluoride	12	80.65	27,500	61.38	28.8	0.19	9.0	61.73
CONVENTIONAL POLLUTANTS								
Oil & Grease	4.2	28.23	3.6	0.008	5.0	0.033	3.39	23.25
Total Suspended Solids	1.0	6.72	2540	5.67	136	0.917	2.7	18.52
Biochemical Oxygen Demand	17	114.25	87	0.19	1475	9.95	12	82.3
pH								

TABLE 5-12
SEMICONDUCTOR PROCESS WASTES
PLANT 36135

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Dilute Rinses†		Effluent†		Dilute Rinses		Effluent	
	Concentration mg/l	Mass Load kg/day						
	38035		76,070		38276		76,551	
	24		24		24		24	
	3763		3764		3765		3766	
TOXIC ORGANICS								
4 Benzene								
7 Chlorobenzene					<0.01 ^B		<0.01	
11 1,1,1-Trichloroethane								
13 1,1-Dichloroethane								
23 Chloroform								
24 2-Chlorophenol					0.015	0.014	0.01	0.018
25 1,2-Dichlorobenzene								
27 1,4-Dichlorobenzene								
38 Ethylbenzene								
44 Methylene chloride								
57 2-Nitrophenol					<0.01 ^B		0.037	0.068
65 Phenol								
66 Bis(2-ethylhexyl)phthalate								
67 Butyl benzyl phthalate					0.070	0.64		
68 Di-N-butyl phthalate								
69 Di-N-octyl phthalate							<0.01	
70 Diethyl phthalate								
85 Tetrachloroethylene					<0.01			
86 Toluene								
87 Trichloroethylene					0.025 ^B	0.023	<0.01	
121 Cyanide*	<0.005		0.013	0.04	<0.01		0.011	0.020
					<0.005		0.009	0.017
Toxic Organics					0.11	0.101	0.048	0.088
TOXIC INORGANICS								
114 Antimony	<0.001		<0.001		<0.001		<0.001	
115 Arsenic	<0.005		<0.005		<0.005		<0.005	
117 Beryllium	0.001	0.0009	0.001	0.002	0.001	0.001	0.001	0.002
118 Cadmium	0.008	0.007	0.007	0.013	0.008	0.007	0.007	0.013
119 Chromium	0.024	0.022	0.048	0.088	0.028	0.026	0.05	0.092
120 Copper	0.232	0.212	0.051	0.093	0.347	0.319	0.05	0.092
122 Lead	0.09	0.082	0.098	0.179	0.096	0.088	0.102	0.187
123 Mercury	<0.001		<0.001		0.01	0.009	0.01	0.018
124 Nickel	1.659	1.514	0.531	0.969	0.815	0.749	0.52	0.955
125 Selenium	<0.005		<0.005		<0.005		0.01	0.018

B = present in sample blank

* Not included in Total Toxic Organics summation.

† Organics not analyzed.

TABLE 5-13

SEMICONDUCTOR PROCESS WASTES
PLANT 36136

Stream Description Flow (l /hr) Duration (hrs) Sample-ID No.	Influent to Treatment		Effluent		Influent to Treatment †		Effluent †	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	55760		59141		53412		57963	
	24		24		24		24	
	3595		3596		3598		3599	
TOXIC ORGANICS								
4 Benzene	<0.01		<0.01					
7 Chlorobenzene	<0.01							
11 1,1,1-Trichloroethane	<0.01		0.013	0.018				
23 Chloroform	<0.01 ^B		<0.01					
24 2-Chlorophenol			<0.01					
25 1,2-Dichlorobenzene	<0.01		<0.01					
31 2,4-Dichlorophenol	<0.01							
39 Fluoranthene	<0.01							
44 Methylene chloride	0.051 ^B	0.068	0.049	0.070				
48 Dichlorobromomethane	<0.01 #							
55 Naphthalene	<0.01		<0.01					
57 2-Nitrophenol			<0.01					
58 4-Nitrophenol	0.01							
64 Pentachlorophenol	0.039 #	0.052						
65 Phenol	0.014	0.019	<0.01					
66 Bis(2-ethylhexyl)phthalate	<0.01		<0.01					
67 Butyl benzyl phthalate	<0.01		<0.01					
68 Di-N-butyl phthalate	<0.01		<0.01					
70 Diethyl phthalate	<0.01		<0.01					
71 Dimethyl phthalate	<0.01 #							
78 Anthracene	<0.01 #							
81 Phenanthrene	<0.01 #							
85 Tetrachloroethylene	<0.01							
87 Trichloroethylene	0.027 ^B	0.036	<0.01					
93 4,4'-DDE	<0.01 #		<0.01 #					
102 Alpha-BHC			<0.01 #					
121 Cyanide	<0.005		<0.005		0.001	0.0013	<0.005	
Total Toxic Organics	0.131	0.175	0.062	0.088				
TOXIC INORGANICS								
114 Antimony	<0.005		0.005		0.009	0.012	<0.005	
115 Arsenic	<0.003		0.003		<0.003		<0.003	
117 Beryllium	<0.001		0.001		<0.001		<0.001	
118 Cadmium	0.006	0.008	0.003	0.004	0.006	0.0077	0.004	0.0056
119 Chromium	0.042	0.056	0.019	0.027	0.035	0.045	0.019	0.026
120 Copper	0.855	1.14	0.041	0.058	2.588	3.318	0.030	0.042
122 Lead	1.459	1.95	0.082	0.116	0.313	0.40	0.083	0.115
123 Mercury	<0.001		<0.001		<0.001		<0.001	
124 Nickel	0.323	0.432	0.844	1.2	1.03	1.32	0.703	0.978
125 Selenium	<0.003		<0.003		<0.003		<0.003	

† Organics not analyzed.

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

TABLE 5-13 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 36136

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Influent to Treatment 55760		Effluent 59141		Influent to Treatment 53412		Effluent 57963	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	<0.005		<0.003		<0.005		0.006	0.008
127 Thallium	<0.025		0.065	0.092	<0.025		0.035	0.049
128 Zinc	0.130	0.174	0.027	0.038	0.289	0.370	0.025	0.035
Total Toxic Inorganics	2.815	3.76	1.081	1.54	4.27	5.47	0.905	1.26
NON-CONVENTIONAL POLLUTANTS								
Aluminum	3.177	4.25	0.227	0.322	5.749	7.37	0.292	0.406
Barium	0.027	0.036	0.012	0.017	0.016	0.02	0.01	0.0139
Boron	0.132	0.177	0.102	0.145	0.431	0.552	0.198	0.275
Calcium	5.196		243.708		3.544		171.508	
Cobalt	0.013	0.017	0.014	0.02	0.016	0.02	0.007	0.0097
Iron	3.725	4.985	0.088	0.125	3.760	4.82	0.106	0.147
Magnesium	2.132		6.794		1.5		4.93	
Manganese	0.144	0.193	0.021	0.03	0.209	0.267	0.025	0.035
Molybdenum	0.024	0.032	0.018	0.026	0.026	0.033	0.018	0.025
Palladium								
Platinum								
Sodium	140.516		38.906		21.732		98.066	
Tin	0.200	0.268	0.012	0.017	0.168	0.215	0.028	0.039
Titanium	0.027	0.036	0.007	0.01	0.033	0.042	0.006	0.008
Vanadium	0.072	0.096	0.064	0.091	0.109	0.14	0.054	0.075
Yttrium	<0.001		0.002	0.003	<0.001		0.033	0.046
Phenols	0.179	0.24	0.112	0.16	0.038	0.049	0.115	0.16
Total Organic Carbon	202	270.3	191	271.1	193	247.4	130	180.8
Fluoride	99.38	133	10.50	14.9	148.75	190.68	12	16.7
CONVENTIONAL POLLUTANTS								
Oil & Grease	20.1	26.90	5.2	7.38	7.3	9.36	6.9	9.6
Total Suspended Solids	72	96.35	56	79.5	80	102.55	44	61.21
Biochemical Oxygen Demand	330	441.62	300	425.8	290	371.75	250	347.8
pH								

TABLE 5-13 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 36136

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Influent to Treatment†		Effluent†	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	61225		61211	
	24		24	
	85110		85111	
TOXIC ORGANICS				
4 Benzene				
7 Chlorobenzene				
11 1,1,1-Trichloroethane				
13 1,1-Dichloroethane				
23 Chloroform				
24 2-Chlorophenol				
25 1,2-Dichlorobenzene				
27 1,4-Dichlorobenzene				
29 1,1-Dichloroethylene				
31 2,4-Dichlorophenol				
38 Ethylbenzene				
44 Methylene chloride				
51 Chlorodibromomethane				
55 Naphthalene				
57 2-Nitrophenol				
65 Phenol				
66 Bis(2-ethylhexyl)phthalate				
67 Butyl benzyl phthalate				
68 Di-N-butyl phthalate				
69 Di-N-octyl phthalate				
70 Diethyl phthalate				
71 Dimethyl phthalate				
85 Tetrachloroethylene				
86 Toluene				
121 Cyanide*	0.005		<0.005	
Total Toxic Organics				
TOXIC INORGANICS				
114 Antimony	<0.005		<0.005	
115 Arsenic	<0.003		<0.003	
117 Beryllium	<0.001		<0.001	
118 Cadmium	0.007	0.010	0.002	0.003
119 Chromium	0.038	0.056	0.019	0.028
120 Copper	0.691	1.02	0.033	0.048
122 Lead	0.175	0.257	0.06	0.088
123 Mercury	<0.001		0.003	0.004
124 Nickel	1.039	1.527	0.576	0.846
125 Selenium	<0.003		<0.003	

* Not included in Total Toxic Organics summation.

† Organics were not analyzed

TABLE 5-13 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 36136

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Influent to Treatment		Effluent	
	61225		61211	
	24		24	
	85110		85111	
	Concentration	Mass Load	Concentration	Mass Load
	mg/l	kg/day	mg/l	kg/day
TOXIC INORGANICS (Continued)				
126 Silver	<0.005		0.006	0.0088
127 Thallium	<0.025		0.065	0.095
128 Zinc	0.183	0.269	0.031	0.046
Total Toxic Inorganics	2.133	3.14	0.795	1.17
NON-CONVENTIONAL POLLUTANTS				
Aluminum	2.838	4.17	0.253	0.37
Barium	0.047	0.069	0.013	0.019
Boron	0.233	0.34	0.144	0.212
Calcium	7.6		253.408	
Cobalt	0.008	0.012	0.012	0.018
Gold				
Iron	2.065	3.03	0.146	0.214
Magnesium	2.507		6.462	
Manganese	0.126	0.185	0.023	0.034
Molybdenum	0.026	0.038	0.015	0.022
Palladium	<0.025		<0.025	
Platinum	<0.03		<0.03	
Sodium	125.816		52.456	
Tellurium	<0.02		<0.02	
Tin	0.076	0.112	0.02	0.029
Titanium	0.020	0.029	0.007	0.01
Vanadium	0.071	0.10	0.06	0.088
Yttrium	<0.001		0.028	0.041
Phenols	0.114	0.168	0.181	0.266
Total Organic Carbon	76	111.67	136	199.8
Fluoride	83.75	123.1	17.50	25.7
CONVENTIONAL POLLUTANTS				
Oil & Grease	7.1	10.4	7.8	11.46
Total Suspended Solids	72	105.8	60	88.14
Biochemical Oxygen Demand	140	205.7	330	484.8
pH				

TABLE 5-14

SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Stripper Quench Rinse†		Acid Wastes		Stripper Quench Rinse†		Acid Wastes†	
	3265 ##		3262		3260		3264	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene	<0.01 #		<0.01		<0.01			
11 1,1,1-Trichloroethane	<0.01 #							
23 Chloroform	0.044 #		0.034					
25 1,2-Dichlorobenzene	0.800 #				<0.01		0.066	
26 1,3-Dichlorobenzene			<0.01					
27 1,4-Dichlorobenzene	<0.01 #							
31 2,4-Dichlorophenol								
37 1,2-Diphenylhydrazine								
38 Ethylbenzene	5.40 #							
44 Methylene chloride	<0.01 #		<0.01		<0.01			
48 Dichlorobromomethane	<0.01 #		<0.01		<0.01		<0.01	
51 Chlorodibromomethane	<0.01 #							
58 4-Nitrophenol					<0.01			
65 Phenol	0.520 #		<0.01		<0.01			
66 Bis(2-ethylhexyl)phthalate	<0.01 #		<0.01		<0.01			
67 Butyl benzyl phthalate	<0.01 #		<0.01		<0.01		<0.01	
68 Di-N-butyl phthalate	<0.01 #		<0.01		<0.01		<0.01	
69 Di-N-octyl phthalate	<0.01 #		<0.01		<0.01		<0.01	
70 Diethyl phthalate	<0.01 #		<0.01				<0.01	
78 Anthracene	<0.01 #				<0.01		<0.01	
81 Phenanthrene	<0.01 #						<0.01 #	
85 Tetrachloroethylene	0.096 #						<0.01 #	
86 Toluene			<0.01		<0.01			
87 Trichloroethylene	<0.01						<0.01	
121 Cyanide*	<0.006		<0.005		<0.005			
Total Toxic Organics	6.86		0.034				<0.005	0.066
TOXIC INORGANICS								
114 Antimony	<0.002 #		<0.002		<0.002		<0.002	
115 Arsenic	<0.003 #		<0.003		<0.003		<0.003	
118 Cadmium	<0.003 #		<0.003		<0.003		<0.003	
119 Chromium	<0.020 #		<0.02		<0.02		<0.02	
120 Copper	<0.003 #		<0.003		<0.003		<0.003	
122 Lead	<0.01 #		<0.01		<0.01		<0.01	
123 Mercury	<0.001 #		<0.001		<0.001		<0.001	
124 Nickel	<0.025 #		<0.025		<0.025		<0.025	

† Pesticides not analyzed.

* Not included in Total Toxic Organics summation.

Data not transcribed from analytical sheet at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-14 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Stripper Quench Rinse†		Acid Wastes†		Stripper Quench Rinse†		Acid Wastes†	
	3265 ##		3262		3260		3264	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	<0.002 #		<0.002		<0.002		<0.002	
127 Thallium	<0.020 #		<0.02		<0.02		<0.02	
128 Zinc	<0.002 #		0.005		0.002		0.002	
Total Toxic Inorganics			0.094		0.091		0.091	
NON-CONVENTIONAL POLLUTANTS								
Aluminum								
Barium								
Boron								
Calcium								
Cobalt								
Gold								
Iron								
Magnesium								
Manganese								
Molybdenum								
Palladium								
Platinum								
Sodium								
Tellurium								
Tin								
Titanium								
Vanadium								
Yttrium								
Phenols	0.02		<0.01		0.026		<0.01	
Total Organic Carbon	4 #		<2.0 #		<2 #		<2 #	
Fluoride	0.07 #		134 #		0.14 #		0.20 #	
CONVENTIONAL POLLUTANTS								
Oil & Grease	27 #		27 #		14 #		31 #	
Total Suspended Solids	9 #		7 #		5 #		7 #	
Biochemical Oxygen Demand	*		*		*		*	
pH	5.1 #		3.6 #		3.4 #		3.7 #	

† Non-toxic metals not analyzed.

* Blank out of control range. Results not reliable.

Data not transcribed from analytical sheet at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-14 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Semiconductor Effluent† 476962 ##		Scrubber Wastes† 2271 ##		Effluent† 953923 ##		Semiconductor Effluent†† 476962 ##	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
1 Acenaphthene								
4 Benzene			<0.01		<0.01			
6 Carbon Tetrachloride								
8 1,2,4-Trichlorobenzene	<0.01				0.150 #	3.434		
11 1,1,1-Trichloroethane					<0.01			
23 Chloroform	0.020	0.229	<0.01		0.63	14.42		
25 1,2-Dichlorobenzene			<0.01		0.019	0.435		
26 1,3-Dichlorobenzene					0.078	1.786		
27 1,4-Dichlorobenzene					<0.01			
29 1,1-Dichloroethylene					<0.01			
30 1,2-Trans-Dichloroethylene					0.016 #	0.366		
38 Ethylbenzene					0.019 #	0.435		
39 Fluoranthene	<0.01		<0.01		<0.01			
44 Methylene chloride	<0.01		0.013	0.0007	0.051	1.168		
45 Methyl Chloride					<0.01 #			
48 Dichlorobromomethane					<0.01 #			
51 Chlorodibromomethane					<0.01			
55 Naphthalene	<0.01		<0.01		<0.01			
57 2-Nitrophenol			0.02	0.001	<0.01 #			
65 Phenol			0.025	0.0014	0.053	1.213		
66 Bis(2-ethylhexyl)phthalate	<0.01		<0.01 B		<0.01			
67 Butyl benzyl phthalate	<0.01		<0.01		<0.01 ^B			
68 Di-N-butyl phthalate	<0.01		<0.01 B		<0.01 ^B			
69 Di-N-octyl phthalate	<0.01		<0.01 B		<0.01			
70 Diethyl phthalate	<0.01		<0.01		<0.01 B			
85 Tetrachloroethylene			<0.01		0.760	17.40		
86 Toluene	<0.01		<0.01		<0.01			
87 Trichloroethylene					0.022	0.504		
121 Cyanide*	0.013	0.149	<0.005		<0.005		<0.005	
Total Toxic Organics	0.020	0.229	0.058	0.003	1.798	41.164		
TOXIC INORGANICS								
114 Antimony	<0.002		0.025	0.0014	<0.002		<0.002	
115 Arsenic	<0.003		<0.003		0.011	0.252	<0.003	
118 Cadmium	<0.003		<0.003		0.003	0.069	<0.003	
119 Chromium	<0.02	0.114	<0.02		2.953	<0.02		
120 Copper	0.01	0.114	0.024	0.0013	24.27	<0.003		
122 Lead	0.018	0.206	<0.01		0.116	2.66	<0.01	
123 Mercury	<0.025		<0.001		0.006	0.137	0.001	0.011
124 Nickel	<0.025		<0.025		0.575	13.16	<0.025	
125 Selenium	<0.003						<0.003	

* Not included in Total Toxic Organics summation.

† Pesticides not analyzed

†† Organics not analyzed

Data not transcribed from analytical sheet at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-14 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Semiconductor Effluent† 476962 ##		Scrubber Wastest 2271 ##		Effluent† 953923 ##		Semiconductor Effluent† 476962 ##	
	24 3251		24 3250		24 3252		24 3255	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	<0.002		<0.002		0.008	0.183	<0.002	
127 Thallium	<0.02		<0.02		<0.02		<0.02	
128 Zinc	0.006	0.069	0.021	0.0011	0.088	2.015	0.004	0.046
Total Toxic Inorganics	0.034	0.389	0.07	0.004	2.0	45.79	0.005	0.057
NON-CONVENTIONAL POLLUTANTS								
Aluminum								
Barium								
Boron								
Calcium								
Cobalt								
Gold								
Iron								
Magnesium								
Manganese								
Molybdenum								
Palladium								
Platinum								
Sodium								
Tellurium								
Tin								
Titanium								
Vanadium								
Yttrium								
Phenols	0.01	0.114	<0.3		<0.013		<0.01	
Total Organic Carbon	3	34.34	34	1.836	11	251.8	<2 #	
Fluoride	215	2461	39	2.106	34	778.4	102 #	1168
CONVENTIONAL POLLUTANTS								
Oil & Grease	<1.0		<1		1.24	28.39	20 #	228.9
Total Suspended Solids	1	11.45	15	0.81	52	1190	5 #	57.24
Biochemical Oxygen Demand							<2 #	
pH							1.8 #	20.6

† Non-toxic metals not analyzed.

Data not transcribed from analytical sheet at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-14 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes†		Effluent†		Semiconductor Effluent†		Scrubber Wastes†	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	2271 ##		953923 ##		476962 ##		2271 ##	
	24		24		24		24	
	3254		3256		3259		3258	
TOXIC ORGANICS								
4 Benzene								
7 Chlorobenzene								
11 1,1,1-Trichloroethane								
13 1,1-Dichloroethane								
23 Chloroform								
24 2-Chlorophenol								
25 1,2-Dichlorobenzene								
27 1,4-Dichlorobenzene								
29 1,1-Dichloroethylene								
31 2,4-Dichlorophenol								
38 Ethylbenzene								
44 Methylene chloride								
51 Chlorodibromomethane								
55 Naphthalene								
57 2-Nitrophenol								
65 Phenol								
66 Bis(2-ethylhexyl)phthalate								
67 Butyl benzyl phthalate								
68 Di-N-butyl phthalate								
69 Di-N-octyl phthalate								
70 Diethyl phthalate								
85 Tetrachloroethylene								
86 Toluene								
121 Cyanide*	0.006	0.0003	<0.005		<0.005		<0.005	
Total Toxic Organics								
TOXIC INORGANICS								
114 Antimony	0.028	0.002	<0.02		<0.02		0.02	0.001
115 Arsenic	<0.003		0.017	0.389	<0.003		<0.003	
118 Cadmium	<0.003		<0.003		<0.003		<0.003	
119 Chromium	<0.02		0.116	2.656	0.02	0.229	<0.02	
120 Copper	0.026	0.0014	1.333	30.52	<0.003		0.024	0.0013
122 Lead	<0.01		0.04	0.916			<0.01	
123 Mercury	0.003	0.0002	0.001	0.023	<0.01		<0.001	
124 Nickel	<0.025		0.355	8.13	<0.025		<0.025	
125 Selenium					<0.003			

* Not included in Total Organics summation.

Data incorrectly transcribed at proposal. (See note on page 5-5.)

† Organics were not analyzed

TABLE 5-14 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes† 2271 ## 24 3254		Effluent† 953923 ## 24 3256		Semiconductor Effluent 476962 ## 24 3259		Scrubber Wastes† 2271 ## 24 3258	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	<0.002		0.002	0.046	<0.002		<0.002	
127 Thallium	<0.02		<0.02		<0.02		<0.02	
128 Zinc	0.012	0.0006	0.016	0.366	<0.001		0.01	0.0005
Total Toxic Inorganics	0.07	0.004	1.88	43.04	0.02	0.229	0.054	0.003
NON-CONVENTIONAL POLLUTANTS								
Aluminum					0.717			
Barium					<2.500			
Boron								
Calcium								
Cobalt								
Gold								
Iron								
Magnesium								
Manganese								
Molybdenum								
Palladium					<0.04			
Platinum								
Sodium								
Tellurium					<0.05			
Tin					<0.06			
Titanium								
Vanadium								
Yttrium								
Phenols	0.428	0.023	0.012	0.275	0.026	0.298	0.436	0.024
Total Organic Carbon	86 #	4.644	22 #	503.7	2 #	22.89		
Fluoride	110 #	5.94	52 #	1190	123 #	1408		
OTHER POLLUTANTS								
Oil & Grease	25 #	1.35	3 #	68.68	12 #	137.4		
Total Suspended Solids	11 #	0.594	46 #	1053	3 #	34.34		
Biochemical Oxygen Demand	35 #	1.89	33 #	755.6	*			
pH	2.8 #		7.6 #		2.1 #	24.04	2.9 #	

† Non-toxic metals not analyzed.

* Sample blank out of control range. Results not reliable.

Data not transcribed from analytical sheet at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-14 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	GAs Crystal Effluent 24 3267		EFFluent† 439110* 24 41-33-FE1		EFFluent† † 439110* 24 41-33-PE2		City Water† 439110* 24 41-33-CW1	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene	<0.01							
6 Carbon tetrachloride							0.015	0.158
8 1,2,4-Trichlorobenzene			0.600	6.32				
11 1,1,1-Trichloroethane	<0.01		0.025	0.26	0.180	1.897		
13 1,1-Dichloroethane			3.0	31.62				
23 Chloroform	0.012							
24 2-Chlorophenol			0.015	0.158				
25 1,2-Dichlorobenzene							0.025	0.263
26 1,3-Dichlorobenzene	<0.01		0.185	1.95	0.605	6.376		
27 1,4-Dichlorobenzene	<0.01							
29 1,1-Dichloroethylene			0.015	0.158				
37 1,2-Diphenylhydrazine								
38 Ethylbenzene	0.019							
39 Fluoranthene			0.005	0.053				
44 Methylene chloride	0.220		1.00	10.54				
51 Chlorodibromomethane			0.005	0.053			0.005	0.053
57 2-Nitrophenol							0.005	0.053
58 4-Nitrophenol					0.105	1.107		
65 Phenol			0.225	2.37	0.605	6.376		
66 Bis(2-ethylhexyl)phthalate	<0.01		0.008	0.084	0.009	0.095		
67 Butyl benzyl phthalate	<0.01						0.475	5.006
68 Di-N-butyl phthalate	<0.01							
69 Di-N-octyl phthalate	<0.01		0.006	0.063				
70 Diethyl phthalate	<0.01							
85 Tetrachloroethylene	<0.01							
86 Toluene	<0.01		0.80	8.43				
87 Trichloroethylene			0.01	0.105				
113 Toxaphene			0.08	0.84				
121 Cyanide**					0.126	1.328		
Total Toxic Organics	0.251		5.945	62.65	1.621	17.08	0.515	5.48
TOXIC INORGANICS								
114 Antimony	<0.002		<0.10					
115 Arsenic	<0.003		0.067	0.706			<0.02	
117 Beryllium			<0.015				<0.01	
118 Cadmium	<0.003		0.004	0.042			<0.015	
119 Chromium	<0.02		0.265	2.79			0.002	0.021
120 Copper	0.003		1.230	12.96			<0.05	
122 Lead	<0.01	0.095	1.001				0.28	2.95
123 Mercury	<0.001		0.051	0.537			0.038	0.40
124 Nickel	<0.025		<0.01	2.16			<0.05	
125 Selenium							<0.01	

*Estimated Flow Rate

**Not Included in Total Toxic Organics summation

†Not used in data base because these streams are associated with metal finishing.

††Volatile organics were not analyzed.

TABLE 5-14 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	GaAs		Effluent ‡ †		Effluent ‡ †		City Water ‡ †			
	Crystal Effluent ‡		439110*		439110*		439110*			
	24 3267	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	24 41-33-FE1	Mass Load kg/day	24 41-33-FE2	Mass Load kg/day	24 41-33-CW1	Mass Load kg/day
TOXIC INORGANICS (Continued)										
126 Silver	<0.002		<0.015				<0.015			
127 Thallium	<0.02		<0.002				<0.002			
128 Zinc	0.002		0.093	0.98			0.755	7.96		
Total Toxic Inorganics	0.005		2.01	21.18			1.075	11.33		
NON-CONVENTIONAL POLLUTANTS										
Aluminum										
Barium										
Boron										
Calcium										
Cobalt										
Gold										
Iron										
Magnesium										
Manganese										
Molybdenum										
Palladium										
Platinum										
Sodium										
Tellurium										
Tin										
Titanium										
Vanadium										
Yttrium										
Phenols										
Total Organic Carbon	25 #		14.7	154.9			3.70	38.99		
Fluoride	72 #									
CONVENTIONAL POLLUTANTS										
Oil & Grease	12.5 #									
Total Suspended Solids	11 #		39.0	411.0			<0.01			
Biochemical Oxygen Demand	*		51.0	537.5			41.0	432.1		
pH	1.7 #		9.6				8.2			

\$Non-toxic metals not analyzed.

† Not used in data base because these metals are associated with metal finishing.

* Estimated Flow Rate

Data not transcribed from analytical sheets at proposal. (See note on page 5-5.)

TABLE 5-14 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent†		Silicon Crystal Effluent†† #	
	953923 24 3266	##	21577 24 3249	*
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS				
4 Benzene				
11 1,1,1-Trichloroethane				
13 1,1-Dichloroethane				
23 Chloroform			0.020	0.0104
25 1,2-Dichlorobenzene				
27 1,4-Dichlorobenzene				
29 1,1-Dichloroethylene				
38 Ethylbenzene				
39 Fluoranthene			<0.01	
44 Methylene chloride			<0.01	
51 Chlorodibromomethane			<0.01	
57 2-Nitrophenol			0.033	0.017
65 Phenol			0.035	0.018
66 Bis(2-ethylhexyl)phthalate			<0.01	
67 Butyl benzyl phthalate			<0.01	
68 Di-N-butyl phthalate			<0.01	
69 Di-N-octyl phthalate			<0.01	
70 Diethyl phthalate			<0.01	
85 Tetrachloroethylene				
86 Toluene			<0.01	
121 Cyanide*	0.009	0.206	0.106	0.055
Total Toxic Organics			0.088	0.046
TOXIC INORGANICS				
114 Antimony	<0.002		<0.002	
115 Arsenic	0.018	0.412	<0.003	
117 Beryllium			<0.004	
118 Cadmium	<0.003		<0.003	
119 Chromium	0.098	2.244	<0.020	
120 Copper	0.558	12.77	0.003	0.002
122 Lead	0.048	1.099	<0.010	
123 Mercury	0.001	0.023	<0.001	
124 Nickel	0.03	0.687	<0.025	
125 Selenium			<0.003	

†Organics were not analyzed.

*Not included in Total Organics figure.

††Pesticides were not analyzed.

Data not transcribed from analytical sheet at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-14 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 41061

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent†		Silicon Crystal Effluent‡ #	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	953923 ##		21577	
	24		24	
	3266		3249	
TOXIC INORGANICS (Continued)				
126 Silver	0.002	0.046	<0.002	
127 Thallium	<0.02		<0.020	
128 Zinc	0.012	0.275	0.003	0.002
Total Toxic Inorganics	0.767	17.56		
NON-CONVENTIONAL POLLUTANTS				
Aluminum				
Barium				
Boron				
Calcium				
Cobalt				
Gold				
Iron				
Magnesium				
Manganese				
Molybdenum				
Palladium				
Platinum				
Sodium				
Tellurium				
Tin				
Titanium				
Vanadium				
Yttrium				
Phenols			<0.013	
Total Organic Carbon	22 #	503.7	2	1.036
Fluoride	67 #	1534	0.65	0.337
OTHER POLLUTANTS				
Oil & Grease	10 #	228.9	1.4	0.725
Total Suspended Solids	56 #	1282	2	1.036
Biochemical Oxygen Demand	*		1.7	
pH	9.7 #			

† Non-toxic metals were not analyzed.

* Blank out of control range. Results not reliable.

Data not transcribed from analytical sheet at proposal. (See note on page 5-5.)

Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-15 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 42044

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Dilute Rinses 34505		Effluent 40504		Dilute Rinses 33774		Effluent 36907	
	24		24		24		24	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)								
126 Silver	<0.001		0.007	0.007	<0.001		0.002	0.0018
127 Thallium	<0.001		0.001	0.001	<0.001		<0.001	
128 Zinc	0.157	0.13	0.25	0.024	0.006	0.005	0.019	0.017
	0.218		0.499		0.07	0.057	0.283	
Total Toxic Inorganics	0.163	0.135	0.492	0.481	0.018	0.014	0.24	0.221
NON-CONVENTIONAL POLLUTANTS								
Aluminum	0.066	0.055	0.744	0.72	0.10	0.08	0.603	0.53
Barium	0.003	0.0025	0.057	0.055	0.004	0.003	0.048	0.04
Boron	0.264	0.219	0.922	0.90	0.046	0.037	0.695	0.62
Calcium	<0.005		38.6		0.013		33	
Cobalt	<0.05		<0.052		0.062	0.05	0.081	0.72
Gold	0.002	0.0017	0.02 I	0.019				
Iron	0.138	0.114	0.382	0.37	0.04	0.038	0.207	0.18
Magnesium	<0.025		10.3		0.036		9.54	
Manganese	<0.001		0.007	0.0068	0.001	0.0008	0.004	0.0035
Molybdenum	<0.035		<0.037		<0.035		0.062	0.055
Palladium	<0.003		<0.003					
Platinum	<0.01		<0.01					
Sodium	<1.5		1860		3.35		1090	
Tellurium	0.008	0.0066	0.004	0.0039				
Tin	<0.025		0.036	0.035	<0.025		<0.025	
Titanium	<0.002		<0.002		<0.002		0.005	0.004
Vanadium	<0.001		<0.002		0.002	0.0016	0.005	0.004
Yttrium	<0.003		<0.004		<0.003		<0.003	
Lithium	<0.001		0.075	0.073				
Phenols	<0.001		0.023	0.022	<0.001		0.004	0.0035
Total Organic Carbon	3.0	2.48	33	32.1	12.0	9.73	53.0	46.95
Fluoride	4.80	3.97	46.0	44.72	4.10	3.32	46.0	40.75
CONVENTIONAL POLLUTANTS								
Oil & Grease	<1.0		<1.0		<1.0		<1.0	
Total Suspended Solids	4.0	3.3	17	16.5	3.0	2.43	14.0	12.4
Biochemical Oxygen Demand	<1.0		<1.0		<1.0		12.4	10.98
pH								

I Interference.

TABLE 5-15 (Continued)
SEMICONDUCTOR PROCESS WASTES
PLANT 42044

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Dilute Rinses		LCD Effluent		Effluent	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	30001		7319		34533	
	24		24		24	
	3674		3669		3675	
TOXIC ORGANICS						
1 Acenaphthene			<0.01 #			
4 Benzene			<0.01			
7 Chlorobenzene						
11 1,1,1-Trichloroethane						
23 Chloroform	0.004	0.003	<0.01		0.130	0.108
24 2-Chlorophenol			<0.01		0.010	0.008
25 1,2-Dichlorobenzene	0.002	0.001			0.012	0.010
26 1,3-Dichlorobenzene			<0.01		0.033 #	0.027
27 1,4-Dichlorobenzene	<0.01		<0.01		0.005	0.004
44 Methylene chloride	0.067	0.048	<0.01		<0.01	
55 Naphthalene			0.040	0.007	0.070	0.058
57 2-Nitrophenol					0.033	0.027
65 Phenol	0.001	0.0004			0.011	0.009
66 Bis(2-ethylhexyl)phthala	0.012	0.0086	0.010	0.0018	0.180	0.149
67 Butyl benzyl phthalate					0.020	0.0166
68 Di-N-butyl phthalate	0.006	0.004	<0.01			
69 Di-N-octyl phthalate					0.004	0.003
70 Diethyl phthalate			<0.01			
85 Tetrachloroethylene					0.001	0.0008
86 Toluene	0.002	0.0014	<0.01			
87 Trichloroethylene						
121 Cyanide*	<0.001		0.017	0.003	0.004	0.0033
Total Toxic Organics	0.079 **	0.057	0.040 **	0.007	0.489 **	0.405
TOXIC INORGANICS						
114 Antimony	0.001	0.007	<0.001		<0.001	
115 Arsenic	<0.01 I		0.004	0.0007	0.12	0.10
117 Beryllium	<0.001		<0.001		<0.001	
118 Cadmium	<0.002		<0.002		0.003	0.0025
119 Chromium	<0.001		0.029	0.005	0.205	0.170
120 Copper	<0.002		0.003	0.0005	0.012	0.01
122 Lead	<0.04		<0.04		0.049	0.041
123 Mercury	<0.001		<0.001		<0.001	
124 Nickel	<0.005		<0.005		0.009	0.0075
125 Selenium	<0.001		<0.001		0.046	0.0038

*Not Included in Total Toxic Organics summation.

Data not transcribed from analytical sheet at proposal. (See note on page 5-5.)

** Data incorrectly transcribed at proposal. (See note on page 5-5.)

TABLE 5-15 (Continued)

SEMICONDUCTOR PROCESS WASTES
PLANT 42044

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Dilute Rinses 30001 24 3674		LCD Effluent 7319 24 3669		Effluent 34533 24 3675	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC INORGANICS (Continued)						
126 Silver	<0.001		0.001	0.0002	<0.001	
127 Thallium	<0.001		<0.001		<0.001	
128 Zinc	<0.001		0.008	0.0014	0.01	0.008
Total Toxic Inorganics	0.001	0.0007	0.044	0.0078	0.445	0.287
NON-CONVENTIONAL POLLUTANTS						
Aluminum	0.053	0.038	0.038	0.0067	0.895	0.74
Barium	0.002	0.0014	0.006	0.001	0.048	0.0398
Boron	0.172	0.12	0.499	0.088	0.753	0.62
Calcium	<0.005		0.124		35	
Cobalt	<0.05		<0.05		0.058	0.048
Gold						
Iron	0.023	0.017	0.028	0.0049	0.352	0.29
Magnesium	<0.025		<0.025		9.98	
Manganese	<0.001		0.002	0.0004	0.005	0.004
Molybdenum	<0.035		<0.035		0.042	0.035
Palladium						
Platinum						
Sodium	1.5		3.24		1030	
Tellurium						
Tin	<0.025		<0.025		0.027	0.022
Titanium	<0.002		<0.002		0.005	0.004
Vanadium	<0.001		<0.001		0.006	0.005
Yttrium	<0.003		<0.003		<0.003	
Lithium						
Phenols	0.005	0.0036	0.006	0.001	0.002	0.0017
Total Organic Carbon	2.0	1.44	109.0	19.1	46.0	38.12
Fluoride	5.8	4.18	0.17	0.03	64.5	53.46
CONVENTIONAL POLLUTANTS						
Oil & Grease	1.2	0.86	4.0	0.70	1.0	0.83
Total Suspended Solids	1.0	0.72	5.0	0.88	11.0	9.117
Biochemical Oxygen Demand	1.4	1.01	15.0	2.6	10.2	8.45
pH						

TABLE 5-17
ELECTRONIC CRYSTALS
SUMMARY OF THE RAW WASTE DATA††

<u>Toxic Organics</u> Parameter	<u>Plant 301</u> mg/l	<u>Plant 380</u> mg/l
8 1,2,4-Trichlorobenzene	ND**	3.66
11 1,1,1-trichloroethane	0.170	ND
25 1,2-Dichlorobenzene	ND	132.6
26 1,3-Dichlorobenzene	ND	1.96
27 1,4-Dichlorobenzene	ND	52.6
37 1,2-diphenylhydrazine	0.014	ND
44 Methylene chloride	0.032	0.010
55 Naphthalene	0.038	ND
68 Di-n-butyl phthalate	ND	0.046
78 Anthracene	0.015	ND
85 Tetrachloroethylene	ND	1.4
87 Trichloroethylene	ND	0.02
TOTAL TOXIC ORGANICS	0.269	192.286

ND - Not detected.

<u>Toxic Metals</u>	<u>Min. Conc.</u> mg/l	<u>Max. Conc.</u> mg/l	<u>Mean Conc.</u> mg/l
Antimony	<0.001	0.91	0.122
Arsenic*	1.75	3.03	2.39
Beryllium	<0.001	0.001	<0.001
Cadmium	<0.005	0.040	0.009
Chromium †	0.008	6.95	0.948
Copper †	0.024	7.92	1.23
Lead	0.004	0.308	0.085
Mercury	<0.001	0.001	<0.001
Nickel †	<0.025	2.74	0.454
Selenium	<0.002	0.129	0.016
Silver	<0.005	0.025	0.005
Thallium	<0.001	0.050	0.008
Zinc †	0.040	4.23	0.654

Conventional Pollutants

Oil and Grease	8.0	94	31.5
Total Suspended Solids	7.0	2900	616
Biochemical Oxygen Demand	4	27	19

Non-Conventional Pollutants

Fluoride	28	378	129.7
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* This table shows the range of toxic organics observed.

** Not detected.

†† This table shows the range of toxic organic concentrations observed.

TABLE 5-18

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 301

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent		Wafer Finishing Wastes	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	788		8	
	24		24	
	3469		3470	
TOXIC ORGANICS				
1 Acenaphthene	<0.01		<0.01	
4 Benzene			0.029	
11 1,1,1-Trichloroethane	0.170		0.035	
23 Chloroform	<0.01		0.190	
37 1,2-Diphenylhydrazine	0.014		<0.01	
39 Fluoranthene	<0.01		<0.01	
44 Methylene chloride	0.032 ^B #		0.360	
48 Dichlorobromomethane	<0.01			
54 Isophorone			<0.01	
55 Naphthalene	0.038		<0.01	
59 Dinitrophenol			0.187	
60 Dinitro-o-cresol			0.070	
62 N-Nitrosodiphenylamine			0.051	
64 Pentachlorophenol			0.016	
66 Bis(2-ethylhexyl)phthalate	<0.01		0.011	
67 Butyl benzyl phthalate	<0.01			
68 Di-N-butyl phthalate	<0.01			
69 Di-N-octyl phthalate	<0.01			
70 Diethyl phthalate	<0.01			
72 1,2 benzanthracene	<0.01		<0.01	
77 Acenaphthylene	<0.01		<0.01	
78 Anthracene	0.015		0.048	
80 Fluorene	<0.01		0.013	
84 Pyrene	<0.01		<0.01	
100 Heptachlor	<0.01			
Total Toxic Organics	0.269		1.010	
TOXIC INORGANICS				
114 Antimony	<0.005		0.200	
115 Arsenic	<0.002		0.004 #	
117 Beryllium	<0.001		<0.001	
118 Cadmium	0.006 #		0.009 #	
119 Chromium	0.008 #		0.013 #	
120 Copper	0.055 #		0.633	
122 Lead	0.037 #		0.080 #	
123 Mercury	<0.001		<0.001	
124 Nickel	0.062 #		0.139	
125 Selenium	<0.003		<0.003	

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-18 (Continued)
 ELECTRONIC CRYSTALS PROCESS WASTES
 PLANT 301

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent		Wafer Finishing Wastes	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	788		8	
	24		24	
	3469		3470	
126 Silver	0.005 #		0.006 #	
127 Thallium	<0.025		<0.025	
128 Zinc	0.643		0.091 #	
Total Toxic Inorganics				
NON-CONVENTIONAL POLLUTANTS #				
Aluminum	0.482		0.115	
Barium	0.016		0.194	
Boron	0.650		12.822	
Calcium	30.764		35.054	
Cobalt	0.004		0.125	
Gold				
Iron	1.092		29.230	
Magnesium	6.879		12.029	
Manganese	0.021		0.375	
Molybdenum	0.021		0.023	
Palladium				
Platinum				
Sodium	258.244		91.694	
Tellurium				
Tin	0.019		0.068	
Titanium	0.010		0.002	
Vanadium	0.054		0.083	
Yttrium	<0.001		0.014	
Phenols	<0.002		<0.002	
Total Organic Carbon	2.6		7600	
Fluoride	44		3.3	
CONVENTIONAL POLLUTANTS				
Oil & Grease	94		20%	
Total Suspended Solids	36		320	
Biochemical Oxygen Demand	28		25	
pH	9.6		7.8	

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-19
ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 304

Stream Description	Influent to Treatment		Wafer Finishing Wastes*†		Effluent	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
Flow (l /hr)	2,183		182		2,365	
Duration (hrs)	24		24		24	
Sample ID No.	3841		3842		3844	
TOXIC ORGANICS						
1 Acenaphthene					<0.01	
4 Benzene					<0.01	
10 1,2 Dichloroethane					4.0 #	
11 1,1,1-Trichloroethane	1.400				140.0 ##	
14 1,1,2 Trichloroethane					0.085 ##	
23 Chloroform					<0.01	
29 1,1-Dichloroethylene	<0.01				2.2	
31 2,4-Dichlorophenol	<0.01				<0.01	
44 Methylene chloride	0.015				0.060	
55 Naphthalene					<0.01	
56 Nitrobenzene					<0.01	
68 Di-n-butyl phthalate					<0.01	
70 Diethyl phthalate					<0.01	
78 Anthracene (or phenanthrene)		(Rest of sample lost before it was analyzed)			0.014	
85 Tetrachloroethylene					0.015	
86 Toluene	0.016				0.025	
87 Trichloroethylene					<0.01	
Total Toxic Organics	1.431				146.399	
TOXIC INORGANICS						
114 Antimony	0.018 #		0.018 #		0.011 #	
115 Arsenic	<0.005 #		0.074 #		0.023 #	
117 Beryllium	<0.001		0.004 #		<0.001	
118 Cadmium	0.013 #		0.060 #		0.014 #	
119 Chromium	1.148		1.122 #		0.516	
120 Copper	18.981 #		34.947 #		7.918	
122 Lead	0.605		11.817 #		0.308	
123 Mercury	<0.001		<0.001		<0.001	
124 Nickel	6.065		0.972 #		2.739	
125 Selenium	<0.005		<0.005		<0.005	

* Separate settling system; oil is collected and contract hauled.

† Organics were not analyzed.

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

Data incorrectly transcribed at proposal. (See note on page 5-6.)

TABLE 5-19 (Continued)
ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 304

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Influent to Treatment		Wafer Finishing Wastes		Effluent	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	2,183		182		2,367	
	24		24		24	
	3841		3842		3844	
126 Silver	0.034 #		0.025 #		0.025 #	
127 Thallium	<0.050		<0.050		<0.050	
128 Zinc	1.727		32.960 #		4.231	
Total Toxic Inorganics						
NON-CONVENTIONAL POLLUTANTS						
Aluminum	4.381 #		22.4 #		2.141 #	
Barium						
Boron					0.855 #	
Calcium						
Cobalt					0.682 #	
Gold	0.050 #				0.050 #	
Iron	11.661 #				20.931 #	
Magnesium						
Manganese						
Molybdenum						
Palladium						
Platinum						
Sodium						
Tellurium						
Tin	0.190 #					
Titanium						
Vanadium						
Yttrium						
Phenols						
Total Organic Carbon	460 #					
Fluoride	30		1.2		120	
CONVENTIONAL POLLUTANTS						
Oil & Grease	41		76# #		280 #	
Total Suspended Solids	2000		3400		2900	
Biochemical Oxygen Demand	3		<3		6	
pH	6.3		6.3		5.9	

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-20

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 380

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent		Wafer Finishing Wastes	
	658		227	
	24		24	
	M18-1		M18-2	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS				
8 1,2,4-Trichlorobenzene	3.660			
10 1,2 Dichloroethane			0.400	
11 1,1,1-Trichloroethane			0.320	
13 1,1-Dichloroethane			0.005	
25 1,2-Dichlorobenzene	132.600		1.440	
26 1,3-Dichlorobenzene	1.960		0.014	
27 1,4-Dichlorobenzene	52.600		0.049	
44 Methylene chloride	0.010		0.026	
66 Bis(2-ethylhexyl) phthalate			0.077	
68 Di-n-butyl phthalate	0.046		0.010	
85 Tetrachloroethylene	1.400		0.040	
87 Trichloroethylene	0.020			
Total Toxic Organics	192.286		2.366	
TOXIC INORGANICS				
114 Antimony	<0.0005		<0.0005	
115 Arsenic	<0.005		<0.005	
117 Beryllium	<0.005		<0.005	
118 Cadmium	0.0003 #		0.0013 #	
119 Chromium	<0.025		<0.025	
120 Copper	0.185		<0.005	
122 Lead	0.002 #		0.026 #	
123 Mercury	<0.001		<0.001	
124 Nickel	<0.025		<0.025	
125 Selenium	<0.005		<0.005	

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-20 (Continued)
ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 380

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Effluent 658 24 M18-1		Wafer Finishing Wastes 227 24 M18-2	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
126 Silver	<0.015		<0.015	
127 Thallium	0.038 #		0.0025 #	
128 Zinc	0.038 #		0.041 #	

Total Toxic Inorganics

NON-CONVENTIONAL POLLUTANTS †

Aluminum		
Barium		
Boron		
Calcium		
Cobalt		
Gold		
Iron		
Magnesium		
Manganese		
Molybdenum		
Palladium		
Platinum		
Sodium		
Tellurium		
Tin		
Titanium		
Vanadium		
Yttrium		
Phenols		0.103
Total Organic Carbon	5.4	47 ##
Fluoride		

CONVENTIONAL POLLUTANTS

Oil & Grease	8.4	9.6
Total Suspended Solids	1.2	577
Biochemical Oxygen Demand	--	26
pH	3.0	7.6

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

Data incorrectly transcribed at proposal. (See note on page 5-6.)

† Non-toxic metals were not analyzed.

TABLE 5-21
ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 401

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes†		Acid Wastes		Wafer Finishing Wastes*		Wafer Finishing Wastes*	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	681		3.80		0.80		1.75	
	24		24		24		24	
	3799		3800		3801		3802	
TOXIC ORGANICS								
4 Benzene	<0.01				<0.01		<0.01	
6 Carbon tetrachloride			<0.01					
10 1,2-Dichloroethane			0.01					
11 1,1,1-Trichloroethane			0.089 #		0.089		<0.01	
21 2,4,6-Trichlorophenol							<0.01	
23 Chloroform			0.024 #		0.013			
30 1,2-Trans-dichloroethylene			<0.01					
31 2,4-Dichlorophenol					0.230		<0.01	
37 1,2-Diphenylhydrazine							<0.01	
38 Ethylbenzene					0.010			
44 Methylene chloride	0.78 #		0.092 #		0.020		0.039	
48 Dichlorobromomethane			<0.01					
51 Chlorodibromomethane			<0.01					
54 Isophorone					0.130			
65 Phenol							<0.01	
66 Bis(2-ethylhexyl)phthalate					0.030			
68 Di-N-butyl phthalate			<0.01		0.140		0.023	
70 Diethyl phthalate			<0.01		<0.01		0.011 #	
78 Anthracene			<0.01		0.018			
81 Phenanthrene			<0.01					
85 Tetrachloroethylene	<0.01				<0.01			
86 Toluene	<0.01		<0.01		0.178		0.035	
87 Trichloroethylene			<0.01					
97 Endosulfan sulfate							**	
Total Toxic Organics	0.78		0.205		0.848		0.108	
TOXIC INORGANICS								
114 Antimony	<0.005		<0.005		0.040 #		<0.005	
115 Arsenic	<0.005		<0.005		<0.005		<0.005	
117 Beryllium	<0.001		<0.001		<0.001		<0.001	
118 Cadmium	0.001 #		0.004 #		0.023 #		0.002 #	
119 Chromium	0.012 #		0.064 #		0.029 #		0.008 #	
120 Copper	0.011 #		0.203		11.321		0.017 #	
122 Lead	0.035 #		0.137		0.409		0.172	
123 Mercury	<0.001		<0.001		<0.001		0.001 #	
124 Nickel	0.125		0.267		0.281		0.055 #	
125 Selenium	<0.005		<0.005		<0.005		<0.005	

* Stream is contract hauled.

** 0.5 µg/l reported by GC/ECD, but could not be confirmed by GC/MS because of low concentration.

† Extractable organics and pesticides were not analyzed.

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-21 (Continued)
ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 401

Stream Description Flow (L /hr) Duration (hrs) Sample ID No.	Scrubber Wastes†		Acid Wastes		Wafer Finishing Wastes*		Wafer Finishing Wastes*	
	Concentration mg/L	Mass Load kg/day	Concentration mg/L	Mass Load kg/day	Concentration mg/L	Mass Load kg/day	Concentration mg/L	Mass Load kg/day
	681		3.80		0.80		1.75	
	24		24		24		24	
	3799		3800		3801		3802	
TOXIC INORGANICS (Continued)								
126 Silver	<0.006		0.006		<0.050		<0.005	
127 Thallium	<0.050		<0.050		<0.050		<0.030	
128 Zinc	0.058 #		0.076 #		0.777		0.023 #	
Total Toxic Inorganics								
NON-CONVENTIONAL POLLUTANTS								
Aluminum								
Barium								
Boron								
Calcium								
Cobalt								
Gold								
Iron								
Magnesium								
Manganese								
Molybdenum								
Palladium								
Platinum								
Sodium								
Tellurium	<0.02		<0.02		12 #		<0.02	
Gallium	1.65 #		3.4 #		10 #		<6	
Gadolinium	1.4 #		2.8 #		5 #		45 #	
Lithium	0.02 #		0.04 #		0.09 #		4.8 #	
Phenols	<0.002		<0.002					
Total Organic Carbon	9.3		56					
Fluoride	0.6		33		0.7		0.8	
CONVENTIONAL POLLUTANTS								
Oil & Grease								
Total Suspended Solids	0.4		110		990		14	
Biochemical Oxygen Demand					1200		2100	
pH								

* Stream is contract hauled.

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-22

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 402

Stream Description Flow (L/hr) Duration (hrs) Sample ID No.	LCD Effluent		Equipment Cleaning	
	Concentration mg/L	Mass Load kg/day	Concentration mg/L	Mass Load kg/day
	6308		946	
	24		24	
	3838		3837	
TOXIC ORGANICS #				
4 Benzene	<0.01		<0.01	
11 1,1,1-Trichloroethane			<0.01	
23 Chloroform	<0.01		<0.01	
26 1,3-Dichlorobenzene			<0.01	
37 1,2-Diphenylhydrazine			<0.01	
38 Ethylbenzene			<0.01	
39 Fluoranthene			<0.01	
44 Methylene chloride	<0.01		<0.01	
48 Dichlorobromomethane			<0.01	
54 Isophorone				
59 Dinitrophenol				
60 Dinitro-o-cresol				
62 N-Nitrosodiphenylamine				
64 Pentachlorophenol				
65 Phenol			<0.01	
66 Bis(2-ethylhexyl)phthalate			<0.01	
67 Butyl benzyl phthalate			<0.01	
68 Di-N-butyl phthalate			<0.01	
69 Di-N-octyl phthalate			<0.01	
70 Diethyl phthalate			<0.01	
77 Acenaphthylene				
78 Anthracene				
80 Fluorene				
86 Toluene			<0.01	
87 Trichloroethylene	<0.01		<0.01	
96 Beta-endosulfan			<0.01	
100 Heptachlor				
Total Toxic Organics				
TOXIC INORGANICS #				
114 Antimony	0.042		0.045	
115 Arsenic	<0.005		<0.005	
117 Beryllium	<0.001		<0.001	
118 Cadmium	0.002		0.005	
119 Chromium	0.017		0.052	
120 Copper	0.024		0.048	
122 Lead	0.044		0.105	
123 Mercury	<0.001		<0.001	
124 Nickel	0.087		0.304	
125 Selenium	<0.005		<0.005	

* Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-22 (Continued)
ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 402

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	LCD Effluent		Equipment Cleaning	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	6308		946	
	24		24	
	3838		3837	
126 Silver	<0.006		0.017	
127 Thallium	<0.05		0.070	
128 Zinc	0.048		0.184	
Total Toxic Inorganics				
NON-CONVENTIONAL POLLUTANTS #				
Aluminum	0.081		0.290	
Barium	0.019		0.093	
Boron	0.052		0.098	
Calcium	21.33		89.326	
Cobalt	0.006		0.018	
Gold				
Iron	0.122		0.408	
Magnesium	21.69		65.300	
Manganese	0.009		.033	
Molybdenum	0.014		0.041	
Palladium				
Platinum				
Sodium	83.038		757.776	
Tellurium				
Tin	0.074		0.104	
Titanium	0.003		0.007	
Indium	0.6			
Phenols	<0.002		<0.002	
Vanadium	0.165		0.482	
Yttrium	0.002		0.019	
Total Organic Carbon	820		58	
Fluoride	1.2		1.2	
CONVENTIONAL POLLUTANTS				
Oil & Grease	9.8		5.1	
Total Suspended Solids				
Biochemical Oxygen Demand				
pH				

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-23

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 403

Stream Description	Wafer Finishing and Acid Wastes†		Wafer Finishing Wastes		Wafer Finishing Wastes		Wafer Finishing Wastes	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
Flow (l /hr)	61.92		4.73		4.73		2.37	
Duration (hrs)	24		24		24		24	
Sample ID No.	3834-1		3834-2		3834-3		3834-4	
TOXIC ORGANICS								
4 Benzene	<0.01		NA		NA		NA	
6 Carbon tetrachloride	<0.01		NA		NA		NA	
11 1,1,1-Trichloroethane	<0.01		NA		NA		NA	
13 1,1-Dichloroethane	<0.01		NA		NA		NA	
23 Chloroform	0.040		NA		NA		NA	
38 Ethylbenzene	<0.01		NA		NA		NA	
44 Methylene chloride	0.050		NA		NA		NA	
66 Bis(2-ethylhexyl)phthalate	<0.01		NA		NA		NA	
70 Diethyl phthalate	<0.01		NA		NA		NA	
86 Toluene	0.010		NA		NA		NA	
87 Trichloroethylene	<0.01		NA		NA		NA	
Total Toxic Organics	0.09							
TOXIC INORGANICS								
114 Antimony	65.0 #		1.180		187.5**		<0.050	
115 Arsenic	0.0179 #		0.270		0.034		0.225	
117 Beryllium	0.001 #		NA		NA		NA	
118 Cadmium	0.002 #		NA		NA		NA	
119 Chromium	0.014 #		NA		NA		NA	
120 Copper	0.143		NA		NA		NA	
122 Lead	0.035 #		NA		NA		NA	
123 Mercury	0.001 #		NA		NA		NA	
124 Nickel	0.114		NA		NA		NA	
125 Selenium	0.129		NA		NA		NA	

** The high levels of antimony occur in the slicing machine coolant, which is recirculated, and then hauled for disposal.

† Composite of streams -2, -3, -4, -5, -6.

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-23 (Continued)

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 403

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Wafer Finishing and Acid Wastes†		Wafer Finishing Wastes		Wafer Finishing Wastes		Wafer Finishing Wastes	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	61.92		4.73		4.73		2.37	
	24		24		24		24	
	3834-1		3834-2		3834-3		3834-4	
TOXIC INORGANICS (Continued)								
126 Silver	<0.006		NA		NA		NA	
127 Thallium	<0.050		NA		NA		NA	
128 Zinc	0.060		NA		NA		NA	
Total Toxic Inorganics								
NON-CONVENTIONAL POLLUTANTS								
Aluminum	0.490		NA		NA		NA	
Barium	0.023		NA		NA		NA	
Boron	6.724		NA		NA		NA	
Calcium	25.550		NA		NA		NA	
Cobalt	0.001		NA		NA		NA	
Gold			NA		NA		NA	
Iron	0.124		NA		NA		NA	
Magnesium	5.886		NA		NA		NA	
Manganese	0.021		NA		NA		NA	
Molybdenum	0.022		NA		NA		NA	
Palladium			NA		NA		NA	
Platinum			NA		NA		NA	
Sodium	28.635		NA		NA		NA	
Bismuth			NA		NA		NA	
Indium			0.360		0.230		0.030	
Tellurium			0.570		0.720		9.0	
Tin	0.171		3.200		17.70		0.120	
Titanium	0.006		NA		NA		NA	
Vanadium	0.058		NA		NA		NA	
Yttrium	0.001		NA		NA		NA	
Total Organic Carbon	440							
Fluoride			0.4		0.9		0.3	
CONVENTIONAL POLLUTANTS								
Oil & Grease			12		160		27	
Total Suspended Solids			14		400		49	
Biochemical Oxygen Demand								
pH			7.5		8.8		6.7	

† Composite of streams -2, -3, -4, -5, -6

TABLE 5-23 (Continued)

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 403

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Wafer Finishing Wastes		Acid Wastes	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
		2.37		47.32
		24		24
		3834-5		3834-6
TOXIC ORGANICS				
4 Benzene	NA		NA	
6 Carbon tetrachloride	NA		NA	
11 1,1,1-Trichloroethane	NA		NA	
13 1,1-Dichloroethane	NA		NA	
23 Chloroform	NA		NA	
38 Ethylbenzene	NA		NA	
44 Methylene chloride	NA		NA	
66 Bis(2-ethylhexyl)phthalate	NA		NA	
70 Diethyl phthalate	NA		NA	
86 Toluene	NA		NA	
87 Trichloroethylene	NA		NA	
Total Toxic Organics				
TOXIC INORGANICS				
114 Antimony	3.3		<0.080	
115 Arsenic	0.112		0.325	
117 Beryllium	NA		NA	
118 Cadmium	NA		NA	
119 Chromium	NA		NA	
120 Copper	NA		NA	
122 Lead	NA		NA	
123 Mercury	NA		NA	
124 Nickel	NA		NA	
125 Selenium	NA		NA	

TABLE 5-23 (Continued)

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 403

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Wafer Finishing Wastes 2.37 24 3834-5		Acid Wastes 47.32 24 3834-6	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
126 Silver	NA		NA	
127 Thallium	NA		NA	
128 Zinc	NA		NA	
Total Toxic Inorganics	NA		NA	
NON-CONVENTIONAL POLLUTANTS				
Aluminum	NA		NA	
Barium	NA		NA	
Boron	NA		NA	
Calcium	NA		NA	
Cobalt	NA		NA	
Gold	NA		NA	
Iron	NA		NA	
Magnesium	NA		NA	
Manganese	NA		NA	
Molybdenum	NA		NA	
Palladium	NA		NA	
Platinum	NA		NA	
Sodium	NA		NA	
Tin	NA		NA	
Bismuth	0.020		0.040	
Indium	0.340		0.570	
Tellurium	0.120		0.170	
Total Organic Carbon Fluoride	0.6		36	
CONVENTIONAL POLLUTANTS				
Oil & Grease	50		12	
Total Suspended Solids	18		4.0	
Biochemical Oxygen Demand				
pH	7.4		3.0	

TABLE 5-24
ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 404

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	GaAs Crystal Effluent 20500 24 3729		Acid Wastes** 56.7 24 3730		Wafer Slicing** 157.7 24 3731		Scrubber Wastes 850 24 3732	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
TOXIC ORGANICS								
4 Benzene								
6 Carbon tetrachloride								
7 Chlorobenzene					0.031			
8 1,2,4-Trichlorobenzene					<0.01			
11 1,1,1-Trichloroethane	0.238		0.458					
23 Chloroforml			0.013					
24 2-Chlorophenol			<0.01		0.168		<0.01	
25 1,2-Dichlorobenzene								
29 1,1-Dichloroethylene	0.039							
30 1,2-Trans-dichloroethylene	<0.01							
31 2,4-Dichlorophenol			<0.01					
44 Methylene chloride	0.126		0.026		0.038		<0.01	
45 Methyl chloride			<0.01		<0.01		0.054	
57 2-Nitrophenol					<0.01		<0.01	
58 4-Nitrophenol								
65 Phenol			<0.01				<0.01	
66 Bis(2-ethylhexyl)phthalate								
67 Butyl benzyl phthalate	0.031							
68 Di-N-butyl phthalate	<0.01		<0.01		<0.01		<0.01	
70 Diethyl phthalate	<0.01						<0.01	
85 Tetrachloroethylene	<0.01		0.011				<0.01	
86 Toluene	<0.01		<0.01					
87 Trichloroethylene	3.100		1.700		<0.01		0.660	
121 Cyanide*	0.017		0.013		<0.004		<0.004	
Total Toxic Organics	3.534		2.208		0.237		0.714	
TOXIC INORGANICS								
114 Antimony	0.003 #		0.100		0.260		0.001 #	
115 Arsenic	3.03		62.500		80.300		0.043 #	
117 Beryllium	<0.010		0.010 #		<0.010		<0.010	
118 Cadmium	0.040 #		0.002 #		0.005 #		<0.010	
119 Chromium	0.030 #		6.060		0.720		0.030 #	
120 Copper	0.040 #		2.200		2.300		0.020 #	
122 Lead	0.040 #		0.040 #		0.065 #		0.030 #	
123 Mercury	0.001		0.009 #		0.010 #		0.001 #	
124 Nickel	0.100		0.120		0.120		0.110	
125 Selenium	<0.005		0.006 #		0.100 #		<0.005 #	

* Not included in TTO summation.

** Waste contract hauled.

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-24 (Continued)

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 404

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	GaAs Crystal Effluent		Acid Wastes * *		Wafer Slicing * *		Scrubber Wastes	
	Concentration mg/l	Mass Load kg/day						
	20500		56.7		157.7		850	
	24		24		24		24	
	3729		3730		3731		3732	
TOXIC INORGANICS (Continued)								
126 Silver	<0.006		0.003 #		0.010 #		0.012 #	
127 Thallium	<0.050		0.016 #		0.050 #		<0.050	
128 Zinc	<0.090		0.060 #		0.030 #		0.040 #	
Total Toxic Inorganics	0.057	0.007	2.076	0.0014	0.165	0.0086	5.866	65.25
NON-CONVENTIONAL POLLUTANTS #								
Aluminum	0.19						0.29	
Barium	0.01						0.01	
Boron	21.38						0.27	
Calcium	7.29						9.31	
Cobalt							<0.01	
Gold	<0.05		0.020		0.050		<0.05	
Iron	0.23						0.24	
Magnesium	0.92						1.27	
Manganese	<0.01		0.130		0.032		<0.01	
Molybdenum	0.13		<0.010		<0.010		0.09	
Palladium	<0.025		0.022		0.063		<0.025	
Platinum	<0.03		<0.200		<0.200		<0.030	
Sodium	56.91						10.24	
Tellurium	<0.02		0.830		1.160		<0.02	
Tin	0.03						<0.01	
Titanium	<0.01		0.100		0.031		<0.01	
Tungsten			<1.000		<1.000		<0.01	
Vanadium	0.02						0.02	
Yttrium	0.01						0.01	
Phenols	<0.002		0.004		0.004		0.021	
Total Organic Carbon	110		1,800		40		30	
Fluoride	3.8		3,150		300		2.7	
CONVENTIONAL POLLUTANTS								
Oil & Grease	3.2		6.1		9.4		9.0	
Total Suspended Solids	7.0		29		32		5.0	
Biochemical Oxygen Demand							<3.0	
pH	9.7		1.7		1.1		5.2	

** Waste contract hauled.

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-24 (Continued)

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 404

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Scrubber Wastes # 863		Scrubber Wastes # 863		Semiconductor Effluent # 45740	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
126 Silver	<0.004		<0.006		<0.006	
127 Thallium	<0.030		<0.050		<0.050	
128 Zinc	0.083		0.050		0.140	
Total Toxic Inorganics						
NON-CONVENTIONAL POLLUTANTS						
Aluminum	0.477		1.52		0.49	
Barium	0.015		0.01		0.01	
Boron	0.14		0.13		0.13	
Calcium	10.371		10.36		9.61	
Cobalt	0.011		<0.01		<0.01	
Gold	<0.02		<0.05		<0.05	
Iron	0.194		0.27		0.2	
Magnesium	1.439		1.31		1.25	
Manganese	0.01		0.01		<0.01	
Molybdenum	0.028		0.07		0.06	
Palladium	<0.08		<0.025		<0.025	
Platinum	<0.05		<0.03		<0.03	
Sodium	13.224		16.43		15.3	
Tellurium	<0.02		<0.02		<0.02	
Tin	0.039		0.01		0.01	
Titanium	0.007		<0.01		0.01	
Vanadium	0.029		0.04		0.02	
Yttrium	0.023		<0.01		0.01	
Phenols	3.1		0.13		0.13	
Total Organic Carbon	47		21		38	
Fluoride	9.1		17		7.7	
CONVENTIONAL POLLUTANTS						
Oil & Grease	0.1		0.9		1.1	
Total Suspended Solids	4.0		2.0		3.5	
Biochemical Oxygen Demand	<3.0		<3		<3	
pH	3.3		2.4		6.3	

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-25
ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 405

Stream Description Flow (l/hr) Duration (hrs) Sample ID No.	Influent to Treatment		Silicon Crystal Effluent		GaAs Crystals Effluent		Effluent	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	1135		1135		42250		70958	
	24		24		24		24	
	4033		4035		4038		4036	
TOXIC ORGANICS								
4 Benzene	NA		NA		<0.01			
23 Chloroform	NA		NA		<0.01			
24 2-Chlorophenol	NA		NA		0.029 #			
27 1,4-Dichlorobenzene	NA		NA		0.086 #		0.081	
30 1,2-trans-dichloroethylene	NA		NA		<0.01			
44 Methylene chloride	NA		NA		0.620 #		0.490	
55 Naphthalene	NA		NA		<0.01			
57 2-Nitrophenol	NA		NA		1.200 #			
58 4-Nitrophenol	NA		NA		0.065 #			
65 Phenol	NA		NA		0.480 #			
66 Bis(2-ethylhexyl)phthalate	NA		NA		<0.01			
68 Di-N-butyl phthalate	NA		NA		<0.01		<0.01	
69 Di-N-octyl phthalate	NA		NA				0.180	
70 Diethyl phthalate	NA		NA		<0.01		0.025	
78 Anthracene	NA		NA		<0.01		<0.01	
80 Fluorene	NA		NA		<0.01		<0.01	
81 Phenanthrene	NA		NA		<0.01		<0.01	
85 Tetrachloroethylene	NA		NA		0.120 #			
86 Toluene	NA		NA		0.023 #		0.020	
87 Trichloroethylene	NA		NA		3.800 #		0.110	
121 Cyanide*	NA		NA		<0.002		<0.002	
Total Toxic Organics	NA		NA		6.423		0.905	
TOXIC INORGANICS								
114 Antimony	0.150		0.019 #		0.002 #		0.002 #	
115 Arsenic	<0.002		0.200		1.80		0.180	
117 Beryllium	<0.005		<0.005		<0.005		<0.005	
118 Cadmium	0.005 #		0.006 #		<0.005		<0.005	
119 Chromium	251		54.3		0.401		0.070 #	
120 Copper	0.139 #		0.275 #		0.058 #		0.012 #	
122 Lead	0.153		0.069 #		<0.050		<0.050	
123 Mercury	<0.001		<0.001		<0.001		<0.001	
124 Nickel	2.580		0.327		0.314		0.084 #	
125 Selenium	<0.002		<0.050		<0.002		<0.002	

* Not include in TTO summation.

Data not transcribed from analytical sheets at proposal. (See note on page 5-6.)

TABLE 5-25 (Continued)

ELECTRONIC CRYSTALS PROCESS WASTES
PLANT 405

Stream Description Flow (l /hr) Duration (hrs) Sample ID No.	Influent to Treatment		Silicon Crystal Effluent		GaAs Crystals Effluent		Effluent	
	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day	Concentration mg/l	Mass Load kg/day
	1135		1135		42250		70958	
	24		24		24		24	
	4033		4035		4038		4036	
TOXIC INORGANICS (Continued)								
126 Silver	0.002 #		0.003 #		0.002 #		<0.001	
127 Thallium	<0.001		<0.025		<0.001		0.002 #	
128 Zinc	0.668		0.628		0.107		0.048 #	
Total Toxic Inorganics								
NON-CONVENTIONAL POLLUTANTS								
Aluminum	NA		NA		NA		NA	
Barium	NA		NA		NA		NA	
Boron	NA		NA		NA		NA	
Calcium	NA		NA		NA		NA	
Cobalt	NA		NA		NA		NA	
Gold	NA		NA		NA		NA	
Iron	NA		NA		NA		NA	
Magnesium	NA		NA		NA		NA	
Manganese	NA		NA		NA		NA	
Molybdenum	NA		NA		NA		NA	
Palladium	NA		NA		NA		NA	
Platinum	NA		NA		NA		NA	
Sodium	NA		NA		NA		NA	
Tellurium	NA		NA		NA		NA	
Tin	NA		NA		NA		NA	
Titanium	NA		NA		NA		NA	
Vanadium	NA		NA		NA		NA	
Yttrium	NA		NA		NA		NA	
Lithium	NA		NA		NA		NA	
Phenols	NA		NA		NA		NA	
Total Organic Carbon	NA		NA		0.10		<0.01	
Fluoride	10,400		2.1		160		40	
					66		20	
CONVENTIONAL POLLUTANTS								
Oil & Grease					52		28	
Total Suspended Solids	1550		2700		560		60	
Biochemical Oxygen Demand								
pH								

SECTION 6

SUBCATEGORIES AND POLLUTANTS TO BE REGULATED, EXCLUDED OR DEFERRED

This section cites the E&EC subcategories which are being (1) regulated or (2) excluded from regulation. In addition, this section explains, for those subcategories being regulated, which pollutants are being regulated and which pollutants are being excluded from regulation.

6.1 SUBCATEGORIES TO BE REGULATED

Based on wastewater characteristics presented in Section 5, discharge effluent regulations are being proposed for the Semiconductor and the Electronic Crystals subcategories.

6.1.1 Pollutants To Be Regulated

The specific pollutants selected for regulation in these subcategories are pH, total suspended solids, fluoride, total toxic organics, and arsenic. Arsenic is to be regulated only in the Electronic Crystals subcategory and only at facilities that produce gallium arsenide or indium arsenide crystals. Total suspended solids are also only to be regulated in the Electronic Crystals subcategory. The rationale for regulating these pollutants is presented below.

(pH) Acidity or Alkalinity

During semiconductor manufacture, alkaline wastes result from alkaline cleaning solutions; and during electronic crystal manufacture, alkaline wastes result from the use of hydroxides and carbonates from crystal growth and cleaning and rinsing operations. Acid wastes occur in both subcategories from the use of acids for cleaning and etching operations. The pH in the raw waste can range from 1.1 to 11.9 from these operations.

Although not a specific pollutant, pH is a measure of acidity or alkalinity of a wastewater stream. The term pH is used to describe the hydronium ion balance in water. Technically, pH is the negative logarithm of the hydrogen ion concentration. A pH of 7 indicates neutrality, a balance between free hydrogen and free hydroxyl ions. A pH above 7 indicates that the solution is alkaline, while a pH below 7 indicates that the solution is acidic.

Waters with a pH below 6.0 are corrosive to water works structures, distribution lines, and household plumbing fixtures and such corrosion can add constituents to drinking water such as iron, copper, zinc, cadmium, and lead. Low pH waters not only tend to dissolve metals from structures and fixtures, but also tend to redissolve or leach metals from sludges and bottom sediments. Waters with a pH above 9.9 can corrode certain metals, are detrimental to most natural organic materials, and are toxic to living organisms.

Total Suspended Solids

Suspended solids are found in wastewaters from electronic crystals manufacturers at an average concentration of 616 milligrams per liter. Suspended solids result from slicing, lapping, and grinding operations performed on the crystal. Some abrasives used for these operations may also enter the wastewaters.

Suspended solids increase the turbidity of water, reduce light penetration, and impair the photosynthetic activity of aquatic plants. Solids, when transformed to sludge deposit, may blanket the stream or lake bed and destroy the living spaces for those benthic organisms that would otherwise occupy the habitat.

Fluoride

Hydrofluoric acid is commonly used as an etchant in providing proper surface texture for application of other materials and creating depressions for dopants in device manufacture. Fluoride concentrations have been observed as high as 147 milligrams per liter in raw wastes from semiconductor manufacture, and as high as 378 milligrams per liter in raw wastes from electronic crystals manufacture.

Although fluoride is not listed as a priority pollutant, it can be toxic to livestock and plants, and can cause tooth mottling in humans. The National Academy of Sciences recommends: (1) two milligrams per liter as an upper limit for watering livestock and, (2) one milligram per liter for continuous use as irrigation water on acid soils to prevent plant toxicity and reduced crop yield. Although some fluoride in drinking water helps to prevent tooth decay, EPA's National Interim Primary Drinking Water Regulations set limits of 1.4 to 2.4 milligrams per liter in drinking water to protect against tooth mottling.

Arsenic

Arsenic is being regulated only in the Electronic Crystals subcategory and only at facilities that produce gallium arsenide or indium arsenide crystals. The manufacture of gallium arsenide and indium arsenide crystals generates arsenic wastes from slicing, grinding, lapping, etching, and cleaning operations. Concentrations in raw wastes from crystals manufacture have been observed as high as 80 milligrams per liter.

Certain compounds of arsenic are toxic to man both as poisons and as carcinogenic agents. The carcinogenic effects have only recently been discovered and little is known about the mechanism. Arsenic can be ingested, inhaled, or absorbed through the skin. The EPA 1980 water quality criteria for protection of aquatic life is 0.44 milligrams per liter.

Total Toxic Organics

Toxic organic pollutants were frequently found in wastewaters from semiconductor and electronic crystal facilities. The sources of these organics are operations such as solvent cleaning, developing of photoresist, and stripping of photoresist.

Because of the wide variety of solvents used in the manufacture of semiconductors and electronic crystals, and the subsequent large number of toxic organics found in process wastewaters, the Agency has decided that total toxic organics (TTO) be used as the pollutant parameter for discharge limitations. TTO is the sum of the concentrations of toxic organics listed in Table 6-1 (which is found on page 6-4) and found at concentrations greater than 0.01 milligrams per liter.

6.2 TOXIC POLLUTANTS AND SUBCATEGORIES NOT REGULATED

The settlement agreement explained in Section 2 contained provisions authorizing the exclusion from regulation, in certain circumstances, of toxic pollutants and industry categories and subcategories. These provisions have been rewritten in a Revised Settlement Agreement which was approved by the District Court for the District of Columbia on March 9, 1979, NRDC v. Costle, 12 ERC 1833.

6.2.1 Exclusion of Pollutants

Ninety-five toxic pollutants are being excluded from regulation for both the Semiconductor and Electronic Crystals subcategories. The basis for exclusion for eighty-two of these pollutants is Paragraph

8(a)(iii) which allows exclusion for pollutants which are not detectable with state-of-the-art analytical methods. The basis of exclusion for another nine of these pollutants is also provided by Paragraph 8(a)(iii) which allows exclusion of pollutants which are present in amounts too small to be effectively reduced. Four toxic pollutants are being excluded from regulation because these pollutants are already subject to effluent limitations and standards being promulgated under the Metal Finishing Category. This is permitted by Paragraph 8(a)(iii).

TABLE 6-1
POLLUTANTS COMPRISING TOTAL TOXIC ORGANICS

<u>Toxic Pol- lutant No.</u>		<u>Toxic Pol- lutant No.</u>	
6	carbon tetrachloride	44	methylene chloride
8	1,2,4-trichlorobenzene	48	dichlorobromoethane
10	1,2-dichloroethane	54	isophorone
11	1,1,1-trichloroethane	55	naphthalene
14	1,1,2 trichloroethane	57	2-nitrophenol
21	2,4,6-trichlorophenol	58	4-nitrophenol
23	chloroform	64	pentachlorophenol
24	2-chlorophenol	65	phenol
25	1,2-dichlorobenzene	66	bis(2-ethylhexyl)phthalate
26	1,3-dichlorobenzene	67	butyl benzyl phthalate
27	1,4-dichlorobenzene	68	di-n-butyl phthalate
29	1,1-dichloroethylene	78	anthracene
31	2,4-dichlorophenol	85	tetrachloroethylene
37	1,2-diphenylhydrazine	86	toluene
38	ethylbenzene	87	trichloroethylene

In addition to the exclusion of the ninety-five pollutants for both subcategories, another toxic pollutant is being excluded for the Semiconductor subcategory only. This pollutant is arsenic and is being excluded under Paragraph 8(a)(iii) because it was found in amounts too small to be effectively treated.

The nine toxic pollutants that are being excluded under Paragraph 8(a)(iii) because they were found in amounts too small to be effectively treated are: antimony, beryllium, cadmium, mercury, selenium, silver, thallium, zinc, and cyanide.

The four toxic pollutants which are being excluded under Paragraph 8(a)(iii) because they are subject to effluent limitations being promulgated under the metal finishing category are as follows: nickel, copper, chromium, and lead.

The eighty-two pollutants which are being excluded under 8 (a)(iii) because they were not detected are presented in Table 6-2 on page 6-7.

6.2.2 Exclusion of Subcategories

Seventeen subcategories are being excluded from this regulation based on either paragraph 8(a)(iii) or paragraph 8(a)(iv) of the Revised Settlement Agreement. Five subcategories are being excluded under Paragraph 8(a)(iii) because pollutants are found only in trace amounts and in quantities too small to be effectively reduced by treatment. These subcategories are magnetic coatings, mica paper, carbon and graphite products, and fluorescent lamps. Incandescent lamps are being excluded on the same grounds, with the exception of chromium which is excluded under paragraph 8(a)(iii) because the sulfuric-chromium acid cleaning process will be regulated under the metal finishing category. Eight subcategories are being excluded under Paragraph 8(a)(iii) because the pollutants will be effectively controlled by technologies upon which are based other effluent limitations and pretreatment standards. Six of the eight subcategories generate wastewater from unit operations which will be covered by metal finishing: these are switchgear, resistance heaters, ferrite devices, capacitors (fluid-filled), transformers (fluid-filled), and the subcategory of motors, generators, and alternators. Another subcategory, insulated devices-plastic and plastic laminated, will be covered by the the plastic molding and forming regulation. The last subcategory, insulated wire and cable, will be covered by a number of other categories which include aluminum and aluminum alloys, copper and copper alloys, iron and steel, plastics processing, and metal finishing.

Two subcategories are being excluded from regulation under Paragraph 8(a)(iv) because no water is used in the manufacturing process: these are resistors and dry transformers. Another subcategory, fuel cells, is also being excluded under Paragraph 8(a)(iv) because there are only two or three plants in this subcategory and fuel cells are not manufactured on a regular basis.

Finally, one subcategory is being excluded under both 8(a)(iii) and 8(a)(iv). All pollutants except copper and lead are being excluded under 8(a)(iii) because these pollutants are present only in trace amounts and are not found in treatable quantities. Copper generated by this subcategory is being excluded from regulation under Paragraph 8(a)(iii) because the unit operation which generates copper will be covered by metal finishing. Lead found in the subcategory is being excluded from regulation under Paragraph 8(a)(iv) because it is unique to two plants.

6.3 CONVENTIONAL POLLUTANTS NOT REGULATED

BOD, fecal coliform, and oil and grease are not being regulated for either subcategory because they were found at concentrations below treatability. Total suspended solids (TSS) is not being regulated in the case of semiconductors because it was found at an average concentration of 10 milligrams per liter which is below treatability.

6.4 SUBCATEGORIES DEFERRED

Two subcategories of the Electrical and Electronic Components category were proposed for regulation on March 11, 1983, under Phase II of Electrical and Electronic Components. These subcategories are electron tubes and luminescent materials (referred to as phosphorescent coatings in this document).

TABLE 6-2
TOXIC POLLUTANTS NOT DETECTED

TOXIC POLLUTANT

- | | |
|--|--|
| 1. Acenaphthene | 47. Bromoform (Tribromomethane) |
| 2. Acrolein | 51. Chlorodibromomethane |
| 3. Acrylonitrile | 52. Hexachlorobutadiene |
| 4. Benzene | 53. Hexachlorocyclopentadiene |
| 5. Benzidine | 56. Nitrobenzene |
| 7. Chlorobenzene | 59. 2,4-Dinitrophenol |
| 9. Hexachlorobenzene | 60. 4,6-Dinitro-o-cresol |
| 12. Hexachloroethane | 61. N-Nitrosodimethylamine |
| 13. 1,1-Dichloroethane | 62. N-Nitrosodiphenylamine |
| 15. 1,1,2,2-Tetrachloroethane | 63. N-Nitrosodi-n-propylamine |
| 16. Chloroethane | 69. Di-n-octyl Phthalate |
| 18. Bis(2-chloroethyl)ether | 70. Diethyl Phthalate |
| 19. 2-Chloroethyl Vinyl Ether (Mixed) | 71. Dimethyl Phthalate |
| 20. 2-Chloronaphthalene | 72. 1,2-Benzanthracene [Benzo(a)anthracene] |
| 22. p-Chloro-m-cresol | 73. Benzo(a)Pyrene (3,4-Benzopyrene) |
| 28. 3,3'-Dichlorobenzidine | 74. 3,4-Benzofluoranthene [Benzo(b)fluoranthene] |
| 30. 1,2-trans-Dichloroethylene | 75. 11,12-Benzofluoranthene [Benzo(k)fluoranthene] |
| 32. 1,2-Dichloropropane | 76. Chrysene |
| 33. 1,3-Dichloropropylene(1,3-Dichloropropene) | 77. Acenaphthylene |
| 34. 2,4-Dimethyl Phenol | 79. 1,12-Benzoperylene [Benzo(ghi)perylene] |
| 35. 2,4-Dinitrotoluene | 80. Fluorene |
| 36. 2,6-Dinitrotoluene | 81. Phenanthrene |
| 39. Fluoranthene | 82. 1,2,5,6-Dibenzathracene [Dibenzo(a,h)anthracene] |
| 40. 4-Chlorophenyl Phenyl Ether | 83. Indeno(1,2,3-cd)pyrene (2,3-O-Phenylene-pyrene) |
| 41. 4-Bromophenyl Phenyl Ether | 84. Pyrene |
| 42. Bis(2-chloroisopropyl)ether | 88. Vinyl Chloride (Chloroethylene) |
| 43. Bis(2-chloroethoxy)methane | 89. Aldrin |
| 45. Methyl Chloride(Chloromethane) | 90. Dieldrin |
| 46. Methyl Bromide (Bromomethane) | |

TABLE 6-2
TOXIC POLLUTANTS NOT DETECTED
(Continued)

TOXIC POLLUTANT

- | | |
|---|--|
| 91. Chlordane
(Technical Mixture and Metabolites) | 104. Gamma-BHC(Lindane) |
| 92. 4,4'-DDT | 105. Delta-BHC |
| 93. 4,4'-DDE(P,P'-DDX) | 106. PCB-1242 (Aroclor 1242) |
| 94. 4,4'-DDD(P,P'-TDE) | 107. PCB-1254 (Aroclor 1254) |
| 95. Alpha-Endosulfan | 108. PCB-1221 (Aroclor 1221) |
| 96. Beta-Endosulfan | 109. PCB-1232 (Aroclor 1232) |
| 97. Endosulfan Sulfate | 110. PCB-1248 (Aroclor 1248) |
| 98. Endrin | 111. PCB-1260 (Aroclor 1260) |
| 99. Endrin Aldehyde | 112. PCB-1016 (Aroclor 1016) |
| 100. Heptachlor | 113. Toxaphene |
| 101. Heptachlor Epoxide(BHC-Hexachloro-
cyclohexane) | 116. Asbestos |
| 102. Alpha-BHC | 129. 2,3,7,8-Tetrachlorodibenzo-p-dioxin(TCDD) |
| 103. Beta-BHC | |

SECTION 7

CONTROL AND TREATMENT TECHNOLOGY

The wastewater pollutants and pollutant parameters of concern in the manufacture of semiconductors and electronic crystals, as identified in Section 6, are arsenic, total toxic organics, fluoride, suspended solids, and pH. A discussion of the treatment technologies currently practiced and other applicable technologies for the reduction of these pollutants is presented below, followed by an identification of six treatment system options.

7.1 CURRENT TREATMENT AND CONTROL PRACTICES

Wastewater treatment techniques currently used in the semiconductor and electronic crystal industries include both in-process and end-of-pipe waste treatment. In-process waste treatment is designed to remove pollutants from contaminated manufacturing process wastewater at some point in the manufacturing process. End-of-pipe treatment is wastewater treatment at the point of discharge.

7.1.1 Semiconductor Subcategory

In-process Control -- In-process control techniques with widespread use in this subcategory are collection of spent solvents for resale or contractor hauling, and treatment or contract hauling of the concentrated fluoride wastestream. Contract hauling, in this instance, refers to the industry practice of contracting with a firm to collect and transport wastes for off-site disposal.

Available information indicates that all semiconductor facilities collect spent solvents to some degree. Fifteen of 45 plants surveyed either treat or have contract-hauled the concentrated fluoride stream.

Rinse water recycle (as much as 85 percent) is practiced at three of the plants that were sampled. The pollutants present in the reused process wastewater are removed in the deionized water production area. Although reuse conserves water and decreases wastewater discharge, certain facilities have found recycle to result in frequent process upsets and subsequent product contamination. Because of these problems, this technology has limited applicability as the basis for national standards.

End-of-pipe treatment -- End-of-pipe controls consist primarily of neutralization which is practiced by all dischargers for pH control. One plant uses end-of-pipe precipitation/clarification for control of fluoride.

7.1.2 Electronic Crystals Subcategory

In-Process Control -- In-plant control techniques at electronic crystal manufacturers are similar to those in the semiconductor subcategory. Segregation and collection of spent solvents for resale or contract disposal is practiced to some degree at all plants. Of eight plants visited, two treat their concentrated fluoride stream; one has the fluoride waste contract hauled.

End-of-Pipe Treatment -- Treatment technologies currently being used at electronic crystals plants include neutralization and precipitation/clarification. All six direct dischargers treat to control pH, suspended solids and fluoride. One direct discharger also treats end-of-pipe to reduce arsenic.

7.2 APPLICABLE TREATMENT TECHNOLOGIES

7.2.1 pH Control

Acids and bases are commonly used in the manufacture of semiconductors and electronic crystals and result in process waste streams exhibiting high or low pH values. Sodium hydroxide and sodium carbonate are used in some crystal growth processes and for caustic cleaning. Sulfuric, nitric and hydrofluoric acids are used for etching and acid cleaning operations.

Several methods can be used to treat acidic or basic wastes. Treatment is based upon chemical neutralization usually to pH 6-9. Methods include: mixing acidic and basic wastes, neutralizing high pH streams with acid or low pH streams with bases. The method of neutralization used is selected on a basis of overall cost. Process water can be treated continuously or on a batch basis. When neutralization is used in conjunction with precipitation of metals it may be necessary to use a batch method regardless of flowrate.

Hydrochloric or sulfuric acid may be used to neutralize alkaline wastewaters; sulfuric acid is most often chosen because of its lower cost.

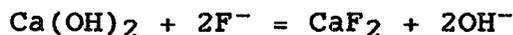
Sodium hydroxide (caustic soda), sodium carbonate (soda ash), or calcium hydroxide (lime) may be used to neutralize acidic wastewater. The factors considered in selection include price, neutralization rate, storage and equipment costs, and neutralization end products. Sodium hydroxide is more expensive than

many other alkalis but is often selected due to the ease of storage, rapid reaction rate and the general solubility of its end product.

7.2.2 Fluoride Treatment

Fluoride appears in semiconductor and electronic crystals wastewater because of the use of hydrofluoric acid and ammonium bifluoride as etching and cleaning agents. Basically two options are available to reduce fluoride in wastewaters from these facilities: chemical precipitation of fluoride followed by solids removal, or isolation for contract hauling of strong fluoride wastes.

The most usual treatment procedure practiced today in the United States for reducing the fluoride concentration in wastewater is precipitation by the addition of lime followed by clarification. Calcium fluoride is formed:



The theoretical solubility of calcium fluoride in water is 7.8 mg fluoride ion per liter at 18°C. The treatability of fluoride in industrial wastewaters however is higher and is dependent on the characteristics of the specific wastewater. Data from the semiconductor subcategory show that plants using precipitation and clarification treatment technologies are achieving an average effluent concentration of 14 milligrams per liter fluoride.

Hydroxide precipitation has proven to be an effective technique for removing many pollutants from industrial wastewater. Metal ions are precipitated as hydroxides and fluoride is precipitated as insoluble calcium fluoride. The system operates at ambient conditions and is well suited to automatic control. Lime is usually added as a slurry when used in hydroxide precipitation. The slurry must be kept well mixed and the addition lines periodically checked to prevent blocking, which may result from a buildup of solids. The use of hydroxide precipitation does produce sludge requiring disposal following precipitation. The performance of a precipitation system depends on several variables. The most important factors affecting precipitation effectiveness are:

1. Addition of sufficient excess chemicals to drive the precipitation reaction to completion. If treatment chemicals are not present in slight excess concentrations, some pollutants will remain dissolved in the waste stream.
2. Maintenance of an alkaline pH throughout the precipitation reaction and subsequent settling.

3. Effective removal of precipitated solids.

Removal of suspended solids or precipitates by gravitational forces may be conducted in a settling tank, clarifier, or lagoon, but the performance of the unit is a function of the retention time, particle size and density, and the surface area of the sedimentation chamber. Accumulated sludge can then be removed either periodically or continuously as in the case of a clarifier.

The effectiveness of a solids settling unit can often be enhanced by the addition of chemical coagulants or flocculants which reduce the repulsive forces between ions or particles and allow them to form larger flocs which are then removed more easily. Commonly used coagulants include ferric sulfate and chloride; commonly used flocculants are organic polyelectrolytes.

An applicable technology for further reduction of fluoride is filtration of the waste stream following precipitation and clarification. Filtration is commonly used in water and wastewater treatment for the removal of finely suspended particles not removed by gravity separation.

A filtration unit commonly consists of a container holding a filter medium or combination of media such as sand or anthracite coal, through which is passed the liquid stream. The unit can operate by gravity flow or under pressure. Periodic backwashing or scraping of the media is necessary to remove particles filtered from the liquid stream and prevent clogging of the filter. The proper design of a filtration unit considers such criteria as filter flow rate (gpm/sq. ft.), media grain size, and density.

For the electrical and electronic components category, the usefulness of filtration technology is questionable. An evaluation of the effectiveness of precipitation and clarification in this industry has shown that this technology can achieve an effluent concentration of approximately 14 mg/l. Addition of a filtration unit would not further reduce the fluoride concentration significantly (approximately three percent) since 14 mg/l of fluoride is approximately equal to the dissolved calcium fluoride concentration soon after formation of the precipitate. Insoluble filterable calcium fluoride would probably constitute only a small fraction of the 14 mg/l fluoride.

7.2.3 Arsenic Treatment

Arsenic is found in the wastewaters of plants fabricating crystals of gallium arsenide and indium arsenide. These wastes are found in the form of powdered gallium arsenide or indium arsenide and result from slicing, lapping, and polishing of crystals. Dissolved arsenides result from crystal etching. The aim of wastewater treatment for arsenic is to remove arsenic from

the water in the form of an insoluble sludge, which may then be disposed of in a manner which keeps it permanently segregated from the environment.

Probably the most common technique used today for arsenic treatment, as discussed in the wastewater treatment literature, is alkaline precipitation with lime followed by clarification. This has been reported to reduce arsenic concentrations to the 1-10 milligrams per liter range. The addition of coagulants such as ferric sulfate or ferric chloride can further reduce the concentration of arsenic; levels of 0.05 milligrams per liter have been reported in the literature. Some additional removal can then be achieved using a filtration polishing step. Precipitation/clarification technology for arsenic reduction has been demonstrated in the industry (see page 7-2).

A general discussion of the technologies of precipitation, clarification and filtration was presented in the previous subsection dealing with the treatment of fluoride in wastewater. Filtration technology has not been demonstrated at any plant in this industry and, as with fluoride, the technology would be expected to provide only minimal further reduction of arsenic in plant effluents.

7.2.4 Total Toxic Organics Control and Treatment

Toxic organics are found in the wastewaters of semiconductor and electronic crystal facilities as a result of contamination from various process streams and as a result of dumping spent solvent baths. The two most applicable control or treatment technologies for limiting toxic organic discharges from semiconductor and electronic crystal plants are solvent management and carbon adsorption. Both of these control technologies are discussed below.

Solvent Management -- Solvent management refers to the practice of preventing spent solvent baths, containing toxic organics, from entering the plant wastewater streams. While a small amount of the solvent baths will enter the wastewaters through process contamination (e.g., drag out), plants substantially reduce toxic organic discharges by transferring the used solvent baths to tanks or drums for disposal. Transfer is done both manually and mechanically through minor piping modifications.

Available data and information show that the above practice of collecting solvents is done at all plants to some degree. The effectiveness of solvent management (i.e., the effluent reduction of toxic organics achieved) depends upon the extent to which plants collect the spent solvents and the extent to which they are handled properly in transferring the spent solvents to tanks

and drums for disposal. Plants with the best solvent management programs use well designed segregation controls or practices to minimize solvent bath spills into rinse or other process streams, have some type of system for collecting routine spills and leaks during handling, and have implemented rigorous employee training programs.

A substantial number of plants in the semiconductor and electronic crystal industries have demonstrated that solvent management will reduce toxic organic discharges to low concentrations. This in-process control is effective because the only other source of toxic organics in the effluent is from the contamination of process wastewater streams (e.g., drag out). Available data show that process streams contribute a very small amount of toxic organics to the effluent and this amount of toxic organics is difficult to reduce or eliminate because the concentrations approximate the level of treatability.

In addition to being relatively inexpensive, especially when compared to more sophisticated end-of-pipe treatment such as carbon adsorption, solvent management has another advantage. After plants have collected the spent solvents in tanks or drums for disposal, they are able to sell the solvents to companies which purify the used solvents in bulk and then resell these solvents. (Note: Names of some companies which provide this reclaim service can be found in the public record for the electrical and electronic components regulation.) The revenue obtained from the sale of these solvents generally offsets the costs of collecting the solvents.

A method of determining the effluent level of toxic organics achievable using this in-plant control is to identify and sum the concentration of toxic organics from each source⁽¹⁾. Below we have described all the process wastewater sources of toxic organics from plants in the semiconductor and electronic crystal industries. These sources are based on data and information collected from plant personnel and confirmed by observation during plant visits. Table 7-1 on page 7-10 summarizes the toxic organic effluent contribution from each source.

(1) At proposal, a slightly different method of determining the TTO limit was used. The method consisted of graphing all the effluent TTO data and then examining the graph to locate a point at which a distinct separation occurred in the magnitude of the TTO effluent concentrations. This breakpoint was 0.47 mg/l. Concentrations falling below the breakpoint reflected the solvent management practices of the best performing plants, whereas those above the breakpoint reflected poor solvent management practices. We assumed that the concentrations below 0.47 mg/l reflected the total of all process wastewater contamination but did not specifically add up the total contribution of all process wastestreams.

Acid Wastes -- Acid wastes consist primarily of hydrofluoric acid and at some plants consist of lesser amounts of hydrochloric and nitric acids. These are generated from etching and cleaning steps. This waste stream includes the spent hydrofluoric acid used directly in etching and may also include the strong or quench rinses after etching. Some plants segregate this waste for contractor disposal while other plants treat this in-process waste stream to control pH and fluoride prior to discharge.

Developer Quench Rinse -- Developer quench rinse is the water rinse that follows the photoresist development step.

Dilute Rinses -- Dilute rinses consist of those water rinses following the first or quench rinse after a process operation, such as acid etching, photoresist stripping, or solvent cleaning. These rinses are usually very low in pollutant concentrations and as a result are recycled to the process at some plants. Included are rinses following the application of both negative and positive photoresist.

Equipment Cleaning Wastes -- Cleaning of process equipment and related items takes place at all facilities. The process equipment cleaned includes glassware, bell jars, the stainless steel or molybdenum masks used in photoresist exposure operations, shipping containers, and general laboratory equipment. The cleaning solutions used include acetic, hydrochloric and fluoboric acids, freon, hydrogen peroxide, and various proprietary mixtures.

Scrubber Wastes -- Wet air scrubbers are used at many electronic crystals and semiconductor facilities to clean the air from process operations utilizing acids and solvents, laminar benches, diffusion ovens and from epitaxial growth operations. The discharges from these scrubbers are frequently high in toxic organics although the flows generally represent only a small portion of the total facility effluent.

Stripper Quench Rinses -- The stripper quench rinse is a deionized water rinse that immediately follows the photoresist stripping operation. This quench may contain residual concentrations of sulfuric acid and hydrogen peroxide (common constituents of inorganic strippers) and such organics as tetrachloroethylene and phenol (common constituents of organic strippers).

Wafer Slicing Wastes -- Although wafer slicing wastes are usually hauled for disposal due to their high concentrations of solids and oils, one waste stream generated solely from wafer slicing operations was analyzed for TTO.

Wafer Finishing Wastes -- Plants which produce crystal wafers generate wastes from, grinding, lapping, and polishing the wafers. Wafer finishing wastes include associated wafer washes and rinses.

Carbon Adsorption -- Another applicable technology for the control of toxic organic discharges is end-of-pipe treatment using carbon adsorption. Frequently used in advanced wastewater treatment, adsorption is a process in which soluble substances become chemically or physically bonded to a solid surface. In operation, wastewater relatively free of suspended matter is passed through a chamber containing activated carbon which has a high capacity for adsorbing organic substances from the stream. Once the capacity of the carbon is exhausted, it must be replaced or regenerated.

The effectiveness of carbon in removing specific organics varies and is dependent on molecular weight and polarity of the molecules, and on operating conditions such as contact time, temperature and carbon surface area. EPA isotherm tests have indicated that activated carbon is very effective in adsorbing 65 percent of the toxic organic pollutants and is reasonably effective for another 22 percent. Table 7-2 presents the theoretical treatability using activated carbon for the 30 toxic organics found in semiconductor and electronic crystals wastewater.

Most of the 30 toxic organics are theoretically treatable by activated carbon to 0.05 milligrams per liter. Eight of these organics have estimated treatabilities of between 0.10 and 1.0 milligrams per liter.

In order to assess the effectiveness of using activated carbon for removal of toxic organics, the Agency used a model plant approach. Data from wastewater sampling in these subcategories have shown that between five and 15 toxic organics occur in any particular plant effluent. The estimated lower limit would consist of a plant having one of the three most difficult pollutants to treat and four of the organics that can be reduced to 0.05 mg/l. An estimated upper limit could be approximated from a plant having all three of the most difficult pollutants to treat and the remaining 12 reducible to 0.05 mg/l. The TTO effluent concentrations based on these occurrences would range from 0.7 mg/l to 2.1 mg/l.

Because this range approximates the TTO effluent level achievable by solvent management, the use of carbon adsorption would result in minimal, if any, additional removal of toxic organics beyond solvent management. While plants could use carbon adsorption to

achieve approximately the same effluent concentration of toxic organics as they could using solvent management, carbon adsorption is unlikely to be used since plants have found solvent management to be much less expensive, relatively simple to institute, and approximately as effective in controlling toxic organic discharges.

7.3 TREATMENT AND CONTROL OPTIONS

For the purpose of establishing effluent limitations and evaluating the costs of wastewater treatment and control for the semiconductor and electronic crystal industries the Agency identified the following six treatment and control options:

- Option 1: Neutralization for pH control and solvent management for control of toxic organics.
- Option 2: Option 1 plus end-of-pipe precipitation/clarification for treatment of arsenic, fluoride, and total suspended solids (TSS).
- Option 3: Option 1 plus in-plant treatment (precipitation/clarification) of the concentrated fluoride stream.
- Option 4: Option 2 plus recycle of the treated effluent stream to further reduce fluoride.
- Option 5: Option 2 plus filtration for reduction of fluoride, arsenic, and suspended solids.
- Option 6: Option 5 plus carbon adsorption to reduce toxic organic concentrations.

Table 7-1
Process Stream Contribution to Effluent TTO

<u>Waste Source</u>	<u>Stream ID</u>	<u>Plant ID</u>	<u>TTO mg/l</u>	<u>Total Plant Flow %</u>	<u>TTO Contribution*</u>
<u>Acid Wastes</u>					
Acid Wastes	3730	404	2.208	0.30	0.006
Acid Wastes	M19-2	30167	0.165	12.0	0.020
Acid Wastes	M19-3	30167	0.272	12.0	0.033
Acid Wastes	3316	30167	0.063	10.7	0.007
Acid Wastes	3317	30167	0.042	10.7	0.004
Acid Wastes	3779	36133	0.091	0.12	0.0001
Acid Wastes	3262	41061	0.034	12.0(1)	0.004
Acid Wastes	3264	41061	0.066	12.0(1)	0.008
<u>Developer Quench Rinse</u>					
Developer Rinse	3647	04294	0.085	1.8(1)	0.002
<u>Dilute Rinses</u>					
Dilute Rinses	3719	35035	0.024	59.0	0.014
Dilute Rinses	3721	35035	0.059	52.0	0.031
Dilute Rinses	3723	35035	0.059	50.7	0.030
Dilute Rinses	3668	42044	0.112	46.0	0.052
Dilute Rinses	3672	42044	0.060	47.8	0.029
Dilute Rinses	3674	42044	0.079	46.5	0.037
Dilute Rinses	3483	06143	0.014	50.4	0.007
Dilute Rinses	3486	06143	0.324	47.5	0.154
Dilute Rinses	3489	06143	0.030	48.4	0.015
Dilute Rinses	3765	36135	0.110	50.0	0.055
<u>Equipment Cleaning Wastes</u>					
Equipment Cleaning	3837	402	<0.010	15.0	<0.0015
Equipment Cleaning	M19-6	30167	0.183	0.8	0.001

* The total toxic organic contribution of each process wastewater stream to the effluent is obtained by multiplying the measured concentration of TTO by the ratio of the plant reported flow for that stream to the total plant effluent flow. The units of mg/l refer to the effluent concentration of TTO which could be attributed to a particular wastestream.

Table 7-1 (Continued)
Process Stream Contribution to Effluent TTO

<u>Waste Source</u>	<u>Stream ID</u>	<u>Plant ID</u>	<u>TTO mg/l</u>	<u>Total Plant Flow %</u>	<u>TTO Contribution mg/l</u>
<u>Scrubber Wastes</u>					
Scrubber Wastes	M16-2	04296	1.868	2.2(2)	0.041
Scrubber Wastes	3474	02347	5.091	4.7	0.239
Scrubber Wastes	3250	41061	0.058	0.5	0.0003
Scrubber Wastes	3718	35035	9.086	1.0	0.091
Scrubber Wastes	3482	06143	2.615	5.9	0.154
Scrubber Wastes	3485	06143	2.115	5.3	0.112
Scrubber Wastes	3488	06143	1.415	5.5	0.078
Scrubber Wastes(3)	3733,3734	404	2.116	3.8	0.080
Scrubber Wastes	3732	404	0.714	4.1	0.029
<u>Stripper Quench Rinse</u>					
Resist strip rinse	3645	04294	0.021	1.8	0.0004
Resist strip rinse	3260	41061	0.010	1.8(1)	0.0002
Resist strip rinse	3265	41061	6.86	1.8(1)	0.123
Resist strip rinse	(4)	06143	200	0.06	0.120
<u>Wafer Finishing Wastes</u>					
Wafer Finishing	M18-2	380	2.366	34.5	0.816
Wafer Finishing	M19-1	30167	0.250	1.9	0.005
Wafer Finishing	3318	30167	0.018	1.1	0.0002
Wafer Finishing	3470	301	1.010	1.0	0.010
Wafer Finishing	3641	04294	0.095	34.5(1)	0.033
Wafer Finishing	3477	02040	0.105	0.5	0.005
Wafer Finishing	3476	02040	0.086	2.2	0.002
<u>Wafer Slicing Wastes</u>					
Wafer Slicing	3731	404	0.237	0.8	0.002

- (1) Plants unable to furnish flow data; flow percent assumed equal to maximum observed at other plants.
- (2) Because data were available for only one of four scrubbers at the plant, its TTO concentration was used and its flow was multiplied by four.
- (3) Flow-proportionally combined discharge from two scrubbers.
- (4) Industry data submitted during comment period.

TABLE 7-2

TREATABILITY OF TOXIC ORGANICS
USING ACTIVATED CARBON

<u>Toxic Pollutant</u>	<u>Treatability mg/l</u>
6 carbon tetrachloride	0.050
8 1,2,4-trichlorobenzene	0.01
10 1,2-dichloroethane	0.1-1.0
11 2,4,6-trichlorophenol	0.1-1.0
14 1,1,2-trichloroethane	0.1-1.0
21 2,4,6-trichlorophenol	0.025
23 chloroform	0.1-1.0
24 2-chlorophenol	0.050
25 1,2-dichlorobenzene	0.050
26 1,3-dichlorobenzene	0.050
27 1,4-dichlorobenzene	0.025
29 1,1-dichloroethylene	0.1-1.0
31 2,4-dichlorophenol	0.050
37 1,2-diphenylhydrazine	0.050
38 ethylbenzene	0.050
44 methylene chloride	0.1-1.0
48 dichlorobromomethane	0.1-1.0
54 isophorone	0.050
55 naphthalene	0.050
57 2-nitrophenol	0.050
58 4-nitrophenol	0.050
64 pentachlorophenol	0.010
65 phenol	0.050
66 bis(2-ethylhexyl)phthalate	0.010
67 butyl benzyl phthalate	0.001-0.010
78 di-n-butyl phthalate	0.025
78 anthracene	0.010
85 tetrachloroethylene	0.050
86 toluene	0.050
87 trichloroethylene	0.1-1.0

SECTION 8

SELECTION OF APPROPRIATE CONTROL AND
TREATMENT TECHNOLOGIES AND BASES FOR
LIMITATIONS

Effluent limitations for the semiconductor subcategory and the electronic crystals subcategory are presented in this section. The technology basis and the numerical basis are also presented for each regulation, in addition to the statistical methodology used to develop limitations.

8.1 SEMICONDUCTOR SUBCATEGORY

8.1.1 Best Practicable Control Technology Currently Available (BPT)

TABLE 8-1

BPT EFFLUENT LIMITATIONS
SEMICONDUCTORS

Pollutant	Long-term Average (LTA) (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit (mg/l)
pH in range 6-9					
Total Toxic Organics			*		1.37

* The Agency is not establishing 30-day limitations for reasons presented below.

BPT limitations are based on Option 1 which consists of neutralization and solvent management. Solvent management is widely practiced and will reduce the amount of toxic organics presently being discharged by approximately 80,000 kilograms per year. For the facilities which do not practice effective solvent management already, compliance costs should be minimal as discussed in Section 9.2. Neutralization is practiced by all facilities subject to BPT and therefore facilities will not incur additional costs for compliance as discussed in Sections 7.2.4 and 9.2.

Option 2 was not selected because, in the semiconductor subcategory, Option 3 can be substituted for and is also less expensive than Option 2. Fluoride in this industry is primarily generated from a particular process stream, hydrofluoric acid etching, and in-plant treatment eliminates the need for end-of-pipe treatment of all process wastewater as in Option 2. Option 3 was not selected because it is more appropriately reserved for consideration under BAT. Options 4, 5, and 6 were not selected for the reasons provided under the BAT discussion.

pH -- Properly operated end-of-pipe neutralization of wastewater will ensure discharges in the pH range of 6 to 9.

Total Toxic Organics (TTO) -- As explained below, the Agency is regulating total toxic organics rather than individual toxic organics. Section 7 presents the effluent contribution of toxic organics from each process wastewater stream in the semiconductor and electronic crystal subcategories. In order to explicitly account for the contribution of each stream, the TTO limit is derived by summing the TTO contribution from each stream as shown in Table 8-2. In cases where we have more than one value for the TTO contribution, we have used the maximum or worst case contribution. No single plant exhibited the TTO maximum for each process stream. Thus, the summation of maximum stream contributions provides a theoretical "worst case" that does not actually exist.

By basing the limit on the total contribution of all process wastestreams, EPA has determined that solvent management can reduce the discharges from other than process contamination to zero or close to zero. Such a limit is feasible for two reasons: First, as stated above, the TTO limit is a "worst case" limit. Therefore, there is some margin for minor releases of TTO from solvent baths, particularly when commingling with other process wastestreams is taken into account. Second, EPA has concluded that solvent management can and does reduce any discharges that would cause the plant to come into noncompliance. As stated previously, 53 percent of the plants already meet the limit. Further, solvent management practices are specifically designed not only to control deliberate dumping, but also to control spills and leaks and poor employee work habits that would lead to violations.

Toxic organics are being regulated as the sum of 30 individual toxic organics; the sum of this total is referred to as total toxic organics (TTO). Compounds included in TTO are listed in Table 6-1. Each of the 30 toxic organic pollutants on the TTO list was found in the effluent at concentrations greater than 0.01 milligrams per liter from plants in the semiconductor and

electronic crystal subcategories. The Agency is using 0.01 milligrams per liter as the basis for inclusion because this level is consistent with the Agency's level of detection for these pollutants as presented in EPA's proposed Guidelines Establishing Test Procedures for the Analysis of Pollutants (December 3, 1979). We are not regulating individual compounds because of the wide range of solvents used and associated concentration ranges.

A 30 day average was not established for TTO primarily because it is not a treatment system which exhibits occasional wide variations in performance. In cases where both a daily maximum and monthly averages exist, the Agency recognizes that the performance of the treatment system can periodically fluctuate as a result of variations in the process flow, pollutant loading, mixing effectiveness, and combinations of these and other reasons. Thus, monthly averages are often less than the daily maximum because better performance can be achieved over a longer period of time. Here, however, the daily and monthly limits would not be expected to significantly differ because solvent management does not rely on the operation of a treatment system, but rather on no dumping and other housekeeping practices as previously described.

TABLE 8-2

CONTRIBUTION OF TTO FROM
PROCESS WASTEWATER STREAMS TO PLANT EFFLUENT

<u>Wastewater Source</u>	<u>Stream ID</u>	<u>TTO Effluent Contribution*</u>
Acid Wastes	M19-3	0.033
Developer Quench Rinse	3647	0.002
Dilute Rinses	3486	0.154
Equipment Cleaning Wastes	M19-6	0.001
Stripper Quench Rinse	3265	0.123
Scrubber Wastes	3474	0.239
Wafer Finishing Wastes	M18-2	0.816
Wafer Slicing Wastes	3731	<u>0.002</u>
Total Toxic Organics		1.370 mg/l

*The total toxic organic contribution of each process wastewater stream to the effluent is obtained by multiplying the measured stream concentration of TTO by the ratio of the plant reported flow for that stream to the total plant effluent flow. The units of mg/l refer to the effluent concentration of TTO which would be attributed to a particular wastestream.

8.1.2 Best Available Technology Economically Achievable (BAT)

TABLE 8-3

BAT EFFLUENT LIMITATIONS
SEMICONDUCTORS

Pollutant	(LTA) (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit (mg/l)
Total Toxic Organic Fluoride	14.5	1.2	17.4	2.2	1.37 32

BAT limitations are based on Option 3. This technology consists of neutralization and solvent management (Option 1) plus in-plant precipitation/clarification of the concentrated fluoride stream. Contract hauling of the concentrated fluoride stream is an acceptable alternative to treatment as a means of achieving compliance.

Option 4 (Option 1 plus end-of-pipe precipitation/clarification followed by a recycle of the treated effluent) was not selected because very few facilities have been able to solve serious operational problems associated with recycling. Therefore Option 4 is not demonstrated in this industry on a national basis. However, facilities located in areas which experience water shortages are encouraged to investigate this technology option. Option 5 (Option 1 plus end-of-pipe precipitation/clarification followed by filtration) was not selected because it will only achieve a three (3) percent increase in fluoride reduction while at the same time significantly increasing treatment costs to the facilities. Option 6 (Option 5 plus carbon adsorption) was not selected because the vast majority of facilities practicing effective solvent management would not discharge concentrations of toxic organics which could be further reduced.

The basis for the total toxic organics (TTO) limitation was presented in Section 8.1.1. This limit does not change for BAT. The basis for fluoride limits is presented below.

Fluoride -- The long term treated fluoride data on which EPA based its BAT fluoride limitation was obtained from a plant with fluoride raw wastes similar in all major respects to fluoride raw wastes from all plants in the semiconductor and electronic crystal subcategories. EPA confirmed the similarity of the fluoride raw

waste from this plant with other plants in these industries by comparing trip reports from 20 visited plants. A statistical analysis of daily concentrations of fluoride in the effluent was conducted to derive the long term average concentration and variability factors for use in establishing proposed limitations. The statistical methodology is presented in Section 8.3. Table 8-4 summarizes the analysis of the historical performance data.

TABLE 8-4
HISTORICAL PERFORMANCE DATA ANALYSIS OF
EFFLUENT FLUORIDE WITH HYDROXIDE
PRECIPITATION/CLARIFICATION SYSTEM

<u>Number of Data Points</u>	<u>Average Concentration mg/l</u>	<u>Variability Factors</u>	
		<u>Daily</u>	<u>30-day</u>
281	14.5	2.2	1.2

8.1.3 Best Conventional Pollutant Control Technology (BCT)

TABLE 8-5
BCT EFFLUENT LIMITATIONS
SEMICONDUCTORS

<u>Pollutant (mg/l)</u>	<u>LTA (mg/l)</u>	<u>30-day Average</u>		<u>Daily Maximum</u>	
		<u>VF</u>	<u>Limit (mg/l)</u>	<u>VF</u>	<u>Limit</u>
<hr/>					
pH in range 6-9					
<hr/>					

For BCT the pH limitation is based on the BPT technology, because BPT achieves the maximum feasible control for pH. Since BPT is also the minimal level of control required, no possible application of the BCT cost test could result in BCT limitations more stringent than those promulgated here. There are no other conventional pollutants of concern in the semiconductor subcategory as discussed in Section 6.

8.1.4 New Source Performance Standards (NSPS)

TABLE 8-6

NSPS EFFLUENT LIMITATIONS
SEMICONDUCTORS

Pollutant (mg/l)	LTA (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit
pH in range 6-9					
Total Toxic Organics					1.37
Fluoride	14.5	1.2	17.4	2.2	32

NSPS limitations are based on solvent management, neutralization, and precipitation/clarification of the concentrated fluoride stream (Option 3). These technologies are equivalent to BAT for control of toxic organics and fluoride, and BCT for control of pH. Other options were not selected for reasons previously presented under BAT.

NSPS limitations are the same as those for BAT and BPT for pH. The bases for those limitations were presented in Section 8.1.2.

8.1.5 Pretreatment Standards for New and Existing Sources (PSES and PSNS)

TABLE 8-7

PSES AND PSNS EFFLUENT LIMITATIONS
SEMICONDUCTORS

Pollutant	LTA (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit (mg/l)
Total Toxic Organics					1.37

For PSES and PSNS, the Agency is promulgating TTO (total toxic organics) limitations based on solvent management. Since biological treatment at well operated POTWs achieving secondary treatment does not achieve removal equivalent to BAT for TTO, pass through occurs. Effective solvent management can reduce TTO by over 99 percent while a POTW will only remove 13 to 97 percent of these same pollutants. Accordingly, EPA is promulgating PSES and PSNS based on technology equivalent to BPT/BAT/NSPS for reduction of TTO.

The Agency is not promulgating pretreatment standards for fluoride. Fluoride is not a toxic pollutant under the Act and EPA has more discretion concerning the establishment of pretreatment standards for such pollutants. In this particular instance fluoride is not a pollutant of concern for indirect dischargers, although fluoride does pass through POTWs. The flow for the semiconductor category is 157,000 gallons per day and the concentration of fluoride in the wastewater entering the POTW is 65.5 mg/l. EPA's environmental assessment, based on a substantial body of scientific literature, shows that there is little likelihood of health or environmental effects from the introduction of fluoride at these flows and concentrations into a POTW. For these reasons, EPA believes it is not appropriate to establish nationally applicable categorical pretreatment standards.

PSES and PSNS limitations are the same as those for BPT/BAT except that pH is not regulated for pretreatment. The basis for TTO limitations was presented in Section 8.1.1.

8.2 ELECTRONIC CRYSTALS SUBCATEGORY

8.2.1 Best Practicable Control Technology Currently Available (BPT)

TABLE 8-8

BPT EFFLUENT LIMITATIONS
ELECTRONIC CRYSTALS

Pollutant	LTA (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit (mg/l)
pH in range 6-9					
Total Toxic Organics					1.37
Arsenic*	0.51	1.62	0.83	4.09	2.09
Total Suspended Solids	18.2	1.26	23	3.35	61.0
Fluoride	14.2	1.2	17.4	2.2	32

* Arsenic limitations are applicable only to discharges from gallium arsenide and indium arsenide crystal manufacturing operations.

BPT limitations are based on Option 2. This technology consists of Option 1 (solvent management and end-of-pipe neutralization) plus end-of-pipe precipitation/clarification. These technologies control pH, toxic organics, total suspended solids (TSS), fluoride, and arsenic. With the exception of solvent management, these treatment technologies have already been installed at all electronic crystal facilities subject to BPT. For facilities which do not practice effective solvent management, compliance costs will be minimal as discussed in Sections 7.2.4 and 9.2.

Arsenic is only being regulated at facilities which manufacture gallium or indium arsenide crystals. Total toxic organic limitations, rather than limitations on each toxic organic pollutant, are set for the same reasons explained under BPT for the semiconductor subcategory.

Option 3 was not selected because this technology is an in-plant control for only one process stream, hydrofluoric acid etching, and as such, will not control all wastewater sources of arsenic and TSS.

Option 4 (Option 1 plus end-of-pipe precipitation/clarification followed by a recycle of the treated effluent) was not selected because very few facilities have been able to solve serious operational problems associated with recycling. Therefore Option 4 is not demonstrated in this industry on a nationwide basis. However, facilities located in areas which experience water shortages are encouraged to investigate this technology option. Option 5 (Option 1 plus end-of-pipe precipitation/clarification followed by filtration) was not selected for arsenic because the Agency has no data available to demonstrate or reason to believe that filtration will further reduce arsenic discharges. This option was also not selected for fluoride because, as previously stated under BAT for semiconductors, filtration would only reduce fluoride by three percent while significantly increasing treatment costs to the facilities. Option 6 (Option 5 plus carbon adsorption) was not selected because the vast majority of facilities practicing solvent management would not discharge treatable concentrations of toxic organics.

The bases of pH, total toxic organics (TTO) and fluoride limitations were presented in Section 8.1 for the semiconductor subcategory. The bases for arsenic and suspended solids limitations are presented below.

Arsenic -- Only limited data are available from the electronic crystals subcategory for the treatment of arsenic-bearing wastes. Therefore, transfer of performance from the non-ferrous metals industrial category is being used for arsenic limitations.

The rationale for transferring performance from this industry are (1) the treatment technology used in the non-ferrous metals industry for reduction of arsenic is the same as that for

electronic crystals, and (2) the raw waste arsenic concentrations (1-10 milligrams per liter) found in non-ferrous metals wastewater compare reasonably with those found in electronic crystals wastes. Based on engineering judgment, the Agency has determined that electronic crystals effluent limitations based on non-ferrous data are achievable.

Monitoring data were submitted from one non-ferrous metals plant using a lime precipitation/clarification treatment system to control arsenic discharge, the same technology as Option 2.

Excluded from the data base were data where pH was less than 7.0 or TSS was greater than 50 milligrams per liter; data points where the treated value was greater than the raw value; and data points where the raw value was too low to ensure pollutant removal. A statistical analysis of daily concentrations of arsenic in the treated effluent was conducted to derive long-term average concentration and variability factors for use in proposing limitations. Table 8-9 summarizes the analysis of the monitoring data.

TABLE 8-9

HISTORICAL PERFORMANCE DATA ANALYSIS OF EFFLUENT ARSENIC WITH HYDROXIDE PRECIPITATION/CLARIFICATION

<u>Number of Data Points</u>	<u>Long-Term Average</u>	<u>Variability Daily</u>	<u>Factors 30-Day</u>
111	0.51	4.09	1.62

Total Suspended Solids -- TSS limitations in Table 8-8 represent a transfer of performance data from the metal finishing industrial category. The rationale for transferring performance data from this industry are (1) the raw waste TSS concentrations are similar to those found in electronic crystals wastes, (2) the treatment technology used for solids reduction in the metal finishing industry is the same as that proposed for electronic crystals, (3) several electronic crystals facilities also conduct metal finishing operations, and (4) the use of metal finishing treatment data provided us with substantially more data as the basis of the TSS limit. Based on engineering judgment, the Agency has determined that electronic crystals effluent limitations based on metal finishing data are achievable.

The average effluent concentration of 18.2 milligrams per liter was derived from EPA sampling data from numerous metal finishing plants practicing solids removal by clarification technology. Excluded from the data base were effluent TSS concentrations greater than 50 milligrams per liter, since this represents a level above which no well-operated treatment plant should be operating. The variability factors of 1.26 and 3.35 each represent the median of variability factors from 17 metal finishing plants with long-term data.

8.2.2 Best Available Technology Economically Achievable (BAT)

TABLE 8-10

BAT EFFLUENT LIMITATIONS
ELECTRONIC CRYSTALS

Pollutant	LTA (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit (mg/l)
Total Toxic Organics					1.37
Arsenic*	0.51	1.62	0.83	4.09	2.09
Fluoride	14.5	1.2	17.4	2.2	32

* Arsenic limitations are applicable only to discharges from gallium arsenide and indium arsenide crystal manufacturing operations.

BAT limitations are based on the BPT technology (Option 2). Option 3 was not selected for the same reason presented above. Options 4, 5, and 6 were not chosen for reasons explained under BPT (Section 8.2.1).

The bases for arsenic, fluoride, and total toxic organics (TTO) limitations were presented in Section 8.2.1 under BPT. These limitations do not change for BAT.

8.2.3 Best Conventional Pollutant Control Technology (BCT)

TABLE 8-11

BCT EFFLUENT LIMITATIONS
ELECTRONIC CRYSTALS

Pollutant	LTA (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit (mg/l)
pH in range 6-9					
Total Suspended Solids	18.2	1.26	23	3.35	61.0

For BCT, pH, and TSS limitations are based on BPT technology. For pH, BPT is equal to BCT for the same reason discussed under the Semiconductor subcategory. For TSS, the Agency considered the addition of filtration to BPT (Option 5), but rejected this technology option because of the minimal additional reduction of total suspended solids. Based on BPT, the average removal of TSS for each of the six (6) direct dischargers will be approximately 5,400 kilograms per year. Filtration would only increase this amount by 100 kilograms per year (0.4 kgs/day) or by less than two percent (2%). Since there is no other technology option which would remove significant amounts of TSS, the Agency is setting BCT equal to BPT. Accordingly, there is no need to conduct the BCT cost test.

8.2.4 New Source Performance Standards (NSPS)

TABLE 8-12

NSPS EFFLUENT LIMITATIONS ELECTRONIC CRYSTALS

Pollutant	LTA (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit (mg/l)
pH in range 6-9					
Total Toxic Organics					1.37
Arsenic*	0.51	1.62	0.83	4.09	2.09
Fluoride	14.5	1.2	17.4	2.2	32
Total Suspended Solids	18.2	1.26	23	3.35	61.0

* Arsenic limitations are applicable only to discharges from gallium arsenide and indium arsenide crystal manufacturing operations.

NSPS limitations are based on solvent management, neutralization, and end-of-pipe precipitation/clarification. These technologies are equivalent to BAT for toxic pollutants and fluoride, and are equivalent to BPT/BCT for conventional pollutants. Other options were not selected for reasons presented under BAT.

NSPS effluent limitations for electronic crystals producers are the same as BPT/BAT for toxic pollutants and fluoride and BPT/BCT for pH and suspended solids. The bases for those limitations are presented in Sections 8.2.1 and 8.2.3.

8.2.5 Pretreatment Standards for New and Existing Sources
(PSNS and PSES)

TABLE 8-13

PSES AND PSNS EFFLUENT LIMITATIONS
 ELECTRONIC CRYSTALS

Pollutant	LTA (mg/l)	30-day Average		Daily Maximum	
		VF	Limit (mg/l)	VF	Limit (mg/l)
Total Toxic Organics					1.37
Arsenic*	0.51	1.62	0.83	4.09	2.09

* Arsenic limitations are applicable only to discharges from gallium arsenide and indium arsenide crystal manufacturing operations.

Both TTO and arsenic will be removed to a greater extent by BAT than by biological treatment at well operated POTWs achieving secondary treatment. Effective solvent management can reduce TTO by over 99 percent while a POTW will remove 13 to 97 percent of these same pollutants. Similarly, precipitation/clarification of arsenic will remove over 92 percent of this pollutant while a POTW will only remove 35 percent. Therefore, PSES and PSNS are required to prevent pass through. For PSES and PSNS, EPA is promulgating limitations based on solvent management, neutralization, and end-of-pipe precipitation/clarification (Option 2) for the facilities which manufacture gallium or indium arsenide crystals. For facilities which only manufacture other types of crystals, PSES and PSNS are based on solvent management (Option 1). Option 2 will assure control of arsenic in addition to controlling toxic organics.

The Agency is not promulgating pretreatment standards for fluoride. Fluoride is not a toxic pollutant under the Act and EPA has more discretion concerning the establishment of pretreatment standards for such pollutants. In this particular instance fluoride is not a pollutant of concern for indirect dischargers. The flow for the electronic crystals subcategory is 29,000 gallons per day and the concentration of fluoride in the wastewater entering the POTW is 129 mg/l. EPA's environmental assessment, based on a substantial body of scientific literature, shows that there is little likelihood of health or environmental effects from the introduction of fluoride at these flows and concentrations into a POTW. For these reasons, EPA believes it is not appropriate to establish nationally applicable categorical pretreatment standards.

PSES and PSNS limitations for electronic crystals producers are the same as those for BPT except that pH and TSS are not regulated for pretreatment. The bases for limitations were presented in Section 8.2.1.

8.3 STATISTICAL ANALYSIS

Statistical analysis of discharge monitoring data allows a quantitative assessment of the variability of effluent concentrations following wastewater treatment. Long term data, collected on a daily basis, reflect the fact that even properly operating treatment systems experience fluctuations in pollutant concentrations discharged. These fluctuations result from variations in process flow, raw waste loading of pollutants, treatment chemical feed, mixing effectiveness during treatment, and combinations of these or other factors.

It is found that the day-to-day variability in effluent concentrations includes occasional large changes while averages for each month's data experience smaller fluctuations. The variability in the monthly average is usually found to be well described by the normal distribution, with values evenly distributed around the mean. However, daily fluctuations are most often described by a lognormal or asymmetric distribution. This reflects the fact that an effluent value may rise considerably from the mean level, but may fall only to the value of zero.

In the development of effluent limitations and standards, allowance for the variation in the effluent concentration of a pollutant is accounted for by the establishment of a variability factor which is always greater than 1.0. This factor, calculated based on the type of distribution of daily or monthly average concentrations, is then multiplied by the mean pollutant concentration to yield a performance standard or effluent limitation that is reasonable for a particular treatment technology and a particular type of waste.

The following paragraphs describe the statistical methodology used to calculate the variability factors and to establish limitations for pollutant concentrations.

8.3.1 Calculation of Variability Factors

Variability factors are used to account for effluent concentration fluctuations in the establishment of reasonable effluent limitations. Calculation of these factors is discussed here, while their application is discussed under the next heading.

Daily Pollutant Level Measurements -- These calculations were based on the following three assumptions: (1) the daily

pollutant concentration data are lognormally distributed; (2) monitoring was conducted in a responsible fashion, such that the resulting measurements can be considered statistically independent and amenable to standard statistical procedures; (3) treatment facilities and monitoring techniques were substantially constant throughout the monitoring period. The lognormality assumption is well established for daily sampling and has been demonstrated in the analysis of effluent samples from many industries. The other two assumptions, which concern self-consistency of the data, were supported by direct examination of the data and by consideration of supplemental information accompanying the data.

The variability factor is especially useful with lognormally distributed pollutant levels because its value is independent of the long-term average, and depends only upon the day-to-day variability of the treatment process and the expected number of unusually high discharge periods. For a lognormal population the variability factor (P/A), the performance standard P, and the long-term average A, are related by

$$\ln (P/A) = S'(Z - S'/2)$$

where \ln represents the natural logarithm, S' is the estimated standard deviation of the natural logarithms of pollutant concentrations, and Z is a factor derived from the standard normal distribution.

The value of Z selected for the calculation of daily performance standards is 2.326, which corresponds to the 99th percentile of the lognormal distribution. Thus only one percent pollutant concentrations is expected greater than the performance standard P . This assumes the continued proper operation of the wastewater treatment procedures, and is equivalent to allowing a plant in normal operation 3 or 4 exceedances per year.

To estimate the variability factor for a particular set of monitoring data, where the method of moments is used, S' is calculated as the square root of $\ln (1.0 + (CV^2))$. Here CV is the sample coefficient of variation, and is the ratio of sample standard deviation to sample mean.

30-Day Averages of Pollutant Levels -- While individual pollutant concentrations are assumed to be lognormally distributed, 30-day averages are not assumed to fit this model. Instead, the statistical "Central Limit Theorem" provides justification for using the normal distribution as the appropriate model. Thus the 30-day average values are expected to behave approximately as random data from a normal distribution, with mean A and standard deviation S'' .

For any probability (k percent) that a particular monthly average will not exceed the performance standard P, there corresponds a value Z such that

$$P = A + Z (S'')$$

The variability factor is

$$P/A = 1.0 + Z(S''/A)$$

and is estimated by

$$P/A = 1.0 + Z(CV)$$

In this equation, Z is frequently given the value of 1.64, to correspond with a probability, k, of 95 percent that a monthly average is within guidelines. CV is the estimated coefficient of variation of the 30-day averages. It may be computed by Sx/A , where S is the standard deviation of sample measurements and x is the mean of sample measurements.

Hence one obtains the performance standard P by multiplying the mean of the 30-day averages by the variability factor. An interpretation is that for the selected value of $Z = 1.64$ corresponding to the 95th percentile of a normal distribution, 19 of every 20 30-day averages will not exceed P.

8.3.2 Calculation of Effluent Limitations

The effluent limitations are based on the premise that a plant's treatment system can be operated to maintain average (mean) effluent concentrations equal to those determined from the sampled data from visited plants. As explained in the introduction, the day-to-day concentrations will fluctuate below and above these average concentrations. Thus the effluent daily limitations must be set far enough above the average daily concentrations that plants with properly operated treatment systems will not exceed them (99 percent of the time), and the 30-day average limitations must be set sufficiently above the mean of 30-day averages so that no more than 5 percent of 30-day averages will exceed the limitations, again assuming a properly operated treatment system. The effluent limitations were obtained for each parameter by multiplying the average concentration (based on visit data) by the appropriate daily and 30-day variability factors (based on historical data) to obtain the effluent limitations. Expressed as equations,

$$\begin{aligned} \text{Daily maximum limitation} &= VF_D \times A \\ \text{30-day average limitation} &= VF_{30} \times A \end{aligned}$$

In these equations, VF_D is the daily maximum variability factor, VF_{30} is the 30-day average variability factor, and A is the average concentration based on plant visit data.

SECTION 9

COST OF WASTEWATER TREATMENT AND CONTROL

This section presents estimates of the costs of implementation of wastewater treatment and control systems for the Semiconductor and Electronic Crystals subcategories of the Electrical and Electronic Components category. The systems for which cost estimates are presented are those options selected by the Agency as the technical bases for discharge regulations as presented in Section 8. The cost estimates then provide the basis for probable economic impact of regulation on the industry.

The general approach or methodology for cost estimating is presented below followed by the treatment and control option costs. Finally, this section addresses non-water quality aspects of wastewater treatment and control including air pollution, noise pollution, solid wastes and energy considerations.

9.1 COST ESTIMATING METHODOLOGY

Costs involved in setting up and operating a wastewater treatment unit are comprised of investment costs for construction, equipment, engineering design, and land, and operating costs for energy, labor, and chemicals. There are also costs for disposing of sludge and for routine analysis of the treated effluent.

The costs presented in this section are based on model plants which closely resemble the types and capacities of waste treatment facilities needed for each product subcategory. Model plants are not set up as exemplary plants, but as typical of sufficient design to represent the range of plants and treatment facilities present in the industry. Data are based on plant visits and contacts with industries to verify treatment practices and to obtain data on size, wastewater flow, and solid waste disposal systems. The differences in treatment capacities are reflected in the choice of model plants which are presented for different flow rates covering the existing range of flows at average concentrations of pollutants.

Unit process equipment costs were assembled from vendors and other commercial sources. Information on the costs of equipment, the present costs of chemicals and average costs for hauling sludge was developed with data from industry, engineering firms, and equipment suppliers. Appropriate factors were applied to determine total investment costs and annual costs.

The costs which will actually be incurred by an individual plant may be more or less than presented in the cost estimate. The major variations in treatment costs between plants result from differences in pollutant concentrations and site dependent conditions, as reflected in piping lengths, climate, land availability, water and power supply, and the location of the point of final discharge. In addition, solids disposal costs and material costs will vary depending on geographical locations.

The following assumptions were employed in the cost development:

1. All non-contact cooling water was excluded from treatment and treatment costs. This source of wastewater is not covered by these regulations.
2. Source water treatment, cooling tower and boiler blowdown discharges were not considered process wastewater.
3. Sanitary sewage flow is excluded.
4. The treatment facilities were assumed to operate 8 hrs/per day, 260 days per year for small plants (below 60,000 GPD); 24 hrs/day, 260 days per year for medium-sized plants (60,000 GPD to 200,000 GPD); and 24 hrs/day 350 days per year for large plants (greater than 200,000 GPD). Treatment facilities operations are based on industry provided data.
5. Excluded from the estimates were any costs associated with permits, reports, or hearings required by regulatory agencies.

Investment costs are expressed in end of year 1979 dollars to construct facilities at various wastewater flow rates. Operation, maintenance, and amortization of the investment are expressed as base level annual costs.

9.1.1 Direct Investment Costs for Land and Facilities

Types of direct investment costs for waste treatment facilities and criteria for estimating major components of the model plants are presented below.

Construction Costs -- Construction costs include site preparation, grading, enclosures, buildings, foundations, earthworks, roads, paving, and concrete. Since few if any buildings will be utilized, construction costs have been

calculated using a factor of 1.15 applied to the installed equipment cost or 2.0 applied to the equipment cost.

Equipment Cost -- Equipment for wastewater treatment consists of a combination of items such as pumps, chemical feed systems, agitators, flocculant feed systems, tanks, clarifiers and thickeners. Cost tables for these items were developed from vendor's quotations for a range of sizes, capacities, and motor horsepowers. Except for large size tanks and chemical storage bins, the cost represents packaged, factory-assembled units.

Critical equipment is assumed to be installed in a weatherproof structure. Chemical storage feeders and feedback controls include such items as probes, transmitters, valves, dust filters, and accessories. Critical pumps are furnished in duplicate as a duty and a spare each capable of handling the entire flow.

Installation Costs -- Installation is defined to include all services, activities, and miscellaneous material necessary to implement the described wastewater treatment and control system, including piping, fittings, and electrical work. Many factors can impact the cost of installing equipment modules. These include wage rates, manpower availability, who does the job (outside contractor or regular employees), new construction versus modification of existing systems, and site-dependent conditions (e.g., the availability of sufficient electrical service). In these estimates, installation costs were chosen for each model based upon average site conditions taking into consideration the complexity of the system being installed. An appropriate cost is allowed for interconnecting piping, power circuits and controls.

Monitoring Equipment -- Monitoring equipment will be installed at the treated effluent discharge point. It will consist of an indicating, integrating, and recording type flow meter, pH meter, sensor, recorder, alarms, controls, and an automatic sampler. This equipment will be used for the purpose of operating the treatment system as well as complying with discharge requirements.

Land -- Land availability and cost of land can vary significantly, depending upon geographical location, degree of urbanization and the nature of adjacent development. Land for waste treatment is assumed to be contiguous with the production plant site. For the purpose of the report land is valued at \$12,000 per acre.

Investment Costs for Supporting Services -- Engineering design and inspection are typical services necessary to advance a project from a concept to an operating system. Such services broadly include laboratory and pilot plant work to establish

design parameters, site surveys to fix elevation and plant layout, foundation and groundwater investigation, and operating instructions, in addition to design plans, specifications and inspection during construction. These costs, which vary with job conditions, are often estimated as percentages of construction costs, with typical ranges as follows:

Preliminary survey and construction surveying	1 to 2%
Soils and groundwater investigation	1 to 2%
Laboratory and pilot process work	2 to 4%
Engineering design and specifications	7 to 12%
Inspection during construction	2 to 3%
Operation and maintenance manual	1 to 3%
TOTAL	14 to 26%

From these totals of 14 to 26 percent, a mid-value of 20 percent of in-place construction (installed equipment and construction) cost has been used in this study to represent the engineering and design cost applied to model plant cost estimates.

The contractor's fee and contingency, usually expressed as a percentage of in-place construction cost, includes such general items as temporary utilities, small tools, field office overhead and administrative expense. The contractor is entitled to a reasonable profit on his activities and to the cost of interest on capital tied up during construction. Although not all of the above cost will be incurred on every job, an additional 50 percent of the in-place construction cost has been used to cover related cost broadly described as contractor's fees, incidentals, overhead, and contingencies.

9.1.2 Annual Costs

Operation and Maintenance Costs -- Annual operation and maintenance costs are described and calculated as follows:

Labor and Supervision Costs:

Personnel costs are based on an hourly rate of \$20.00. This includes fringe benefits and an allocated portion of costs for management, administration, and supervision. Personnel are assigned for specific activities as required by the complexity of the system, ranging from 1-8 hours per day.

Energy Costs:

Energy costs are based on the cost of \$306.00 per horsepower operating 24 hours per day and 350 days per year. For batch processes appropriate adjustments were made to suit the

production schedule. The cost per horsepower year is computed as follows:

$$Cy = 1.1 (0.745 \text{ HP} \times \text{Hr.} \times \text{Ckw}) / (E \times P)$$

where Cy = Cost per year
HP = Total Horsepower Rating of Motor (1 HP = 0.7457 kw)
E = Efficiency Factor (0.9)
P = Power Factor (1.00)
Hr. = Annual Operating Hours (350 x 24 = 8,400)
Ckw = Cost per Kilowatt-Hour of Electricity (\$0.040)

Note: The 1.1 factor in the equation represents allowance for incidental energy used such as lighting, etc. It is assumed that no other forms of energy are used in the waste treatment system.

Chemicals:

Prices for the chemicals were obtained from vendors and the Chemical Marketing Reporter. Unit costs of common chemicals delivered to the plant site are based on commercial grade of the strength or active ingredient percentage with prices as follows:

Hydrated Lime (Calcium Hydroxide) Bulk	\$80/ton
Flocculant	\$ 2/lb

Maintenance:

The annual cost of maintenance is estimated as ten percent (10%) of the investment cost, excluding land.

Taxes and Insurance:

An annual provision of three percent of the total investment cost has been included for taxes and insurance.

Residual Waste Disposal:

Sludge disposal costs can vary widely. Chief cost determinants include the amount and type of waste. Off-site hauling and disposal costs are taken as \$20/YD³ for bulk hauling, with appropriate increases for small quantities in steel containers. Information available to the Agency indicates that the selected treatment technologies for controlling pollutants in this industry will not result in hazardous wastes as defined by RCRA. (Solvents collected by solvent management will be subject to RCRA; see Section 9.2.1 and Table 9-3 for EPA's discussion of these costs.)

Monitoring, Analysis, and Reporting

The manpower requirements covered by the annual labor and supervision costs include those activities associated with the operation and maintenance of monitoring instruments, recorder, and automatic samplers as well as the taking of periodic grab samples. Additional costs for analytical laboratory services have been estimated for each subcategory assuming that sampling takes place three times a week at the point of discharge. A cost of \$7,500/year has been used for monitoring analyses and reporting for Option 2 and 3.

Amortization -- Amortization of capital costs (investment costs) are computed as follows:

$$CA = B (r(1+r)^n) / ((1+r)^n - 1)$$

where CA = Annual Cost
 B = Initial amount invested excluding cost of land
 r = Annual interest rate (assumed 13 percent)
 n = Useful life in years

The multiplier for B in equation (1) is often referred to as the capital recovery factor and is 0.2843 for the assumed overall useful life of 5 years. No residual or sludge value is assumed.

9.1.3 Items Not Included In Cost Estimate

Although specific plants may encounter extremes of climate, flood hazards and lack of water, the cost of model plants have been estimated for average conditions of temperature, drainage, and natural resources. It is assumed that any necessary site drainage, roads, water development, security, environmental studies and permit costs are already included in production facilities costs. Therefore, the model costs are only for facilities, suppliers, and services directly related to the treatment and disposal of waterborne wastes, including land needed for treatment and on-site sludge disposal. Air pollution control equipment is not included, except for dust collectors associated with treatment, chemical transfer and feeding. Raw wastes from various sources are assumed to be delivered to the treatment facility at sufficient head to fill the influent equalization basin, and final effluent is discharged by gravity. Cost of pumps, pipes, lines, etc., necessary to deliver raw wastewater to the treatment plant or to deliver the treated effluent to the point of discharge are not included in the cost estimates.

9.2 COST ESTIMATES FOR TREATMENT AND CONTROL OPTIONS

Table 9-1 summarizes the treatment and control options selected as the bases for effluent limitations and standards for the semiconductor and electronic crystals subcategories.

TABLE 9-1. TREATMENT AND CONTROL OPTIONS
SELECTED AS BASES FOR
EFFLUENT LIMITATIONS

<u>Subcategory</u>	<u>BPT</u>	<u>BAT</u>	<u>BCT/NSPS</u>	<u>Pretreatment</u>
Semiconductors	1	3	1 3	1
Electronic Crystals	2	2	2 2	1+2

9.2.1 Option 1

This treatment option consists of neutralization of the plant discharge and solvent management to control toxic organics. Since all direct dischargers in both the semiconductor and electronic crystals subcategories currently neutralize their discharges, no costs of neutralization will be incurred by the industry. Costs associated with Option 1 result from the collection of additional solvents and monitoring. These costs are explained below.

Many plants already meet the TTO limit. Those plants that are not already in compliance will have to improve the effectiveness of their solvent management program. EPA has assumed the real costs of compliance for such plants are minimal. Primarily, this is because the costs are small increments above existing costs. That is, a discharger who is currently handling and disposing solvents contained in drums or tanks may have some additional amounts of solvents to deal with. He already would have incurred the basic costs of setting up such systems. However, to the extent that there may be incremental costs they would be offset by the resale value of the additional solvents. Data in the record show that resale of spent solvents is commonly practiced.

For monitoring, EPA has estimated the annual costs for 47 percent of the plants to conduct quarterly monitoring. It is difficult to predict precisely how many plants will take advantage of the certification alternative to monitoring, although we expect most plants will want to do so. For purposes of costing, based upon our estimate that 53 percent of existing plants already meet the toxic organic limit, we are assuming the same percentage, at a minimum, will also choose to certify. While it is difficult to estimate monitoring frequency for total toxic organics in the

absence of significant historical experience, based on a survey of state and regional permitting authorities, we estimate that, on average, monitoring for TTO will be required once per quarter. The costs for quarterly monitoring are presented in Table 9-2.

In some cases plants may be required to monitor as frequently as once a month. Thus, EPA has done an economic sensitivity analysis to assess the impact of monthly monitoring costs as part of its economic impact analysis. The capital and annual costs of both quarterly and monthly monitoring for TTO, in 1983 dollars, are presented in Table 9-2.

EPA has also performed an economic sensitivity analysis for RCRA costs. As stated above, EPA believes that minimal costs are associated with TTO compliance. Nevertheless, EPA has costed out and assessed the economic impact if plants presently not in compliance sent the additional solvents to hazardous waste disposal facilities covered by the Resource Conservation and Recovery Act. These costs represent the worst case compliance costs associated with solvent management. They are presented in 1983 dollars in Table 9-3.

9.2.2 Option 2

The capital and annual costs of adding this end-of-pipe precipitation/clarification system to Option 1 treatment are presented in Table 9-4. The range of model plant wastewater flows reflect the range of flows that currently exist for the subcategory. Figure 9-1 graphically presents the annual costs versus plant wastewater flow for this option. The costs are incremental and therefore only reflect the additional costs of adding end-of-pipe precipitation/clarification technology to Option 1 technology.

9.2.3 Option 3

The capital and annual costs of adding this in-plant precipitation/clarification treatment system for fluoride acid wastes to Option 1 treatment are presented in Table 9-5. The range of model plant waste flows reflect the range of flows for this stream as they currently exist in both subcategories. Figure 9-2 graphically presents the annual costs versus waste stream flow for this option. The costs are incremental and therefore only reflect the additional costs of adding in-plant precipitation/clarification technology to Option 1 technology.

9.2.4 Option 5

The capital and annual costs of adding filtration to end-of-pipe precipitation/clarification (Option 2) are presented in Table

9-6. These costs are incremental and therefore only reflect the additional costs of adding filtration technology to Option 2 technology.

9.3 ENERGY AND NON-WATER QUALITY ASPECTS

Compliance with the proposed regulations will have no effect on air, noise, or radiation pollution and will only result in minimal energy usage. The amount of solid waste generated will be 7,700 metric tons per year. Available information indicates that the solid waste generated will not be hazardous as defined in the Resource Conservation and Recovery Act (RCRA). Energy requirements associated with these regulations will be 100,000 kilowatt-hours per year or only 7.5 kilowatt-hours per day per facility.

Based on the absence of any significant non-water quality impacts from these regulations, EPA has concluded that the regulations best serve overall national environmental goals.

TABLE 9-2
 PLANT MONITORING COSTS
 FOR ORGANICS(1)

INVESTMENT COSTS			
Automatic Sampler - Complete		\$ <u>2,500</u>	\$ <u>2,500</u>
TOTAL INVESTMENT COST		\$ 2,500	\$ 2,500
ANNUAL COSTS		<u>Quarterly</u>	<u>Monthly</u>
Analysis Cost	\$ 860	\$ 3,440	\$ 10,320
Sample kit	\$ 50	200	600
Sampling personnel @ \$22/hr x 8 hrs/episode	\$ 176	<u>704</u>	\$ 2,112
TOTAL OPERATION AND MAINTENANCE COST		\$ 4,344	\$ 13,032
AMORTIZATION OF INVESTMENT COST		<u>711</u>	<u>711</u>
TOTAL ANNUAL COST		\$ 5,055	\$ 13,743

(1) 1983 Dollars

TABLE 9-3

INCREMENTAL COST OF SOLVENT DISPOSAL IN ACCORDANCE WITH RCRA

OPTION 1

<u>Plant Number</u>	<u>Gal/90 Days</u> ⁽¹⁾	<u>Number of 55 gal drums</u>	<u>Disposal Cost (\$)</u>	<u>Transportation Cost (\$)</u>	<u>Total Qrtly Cost (\$)</u>	<u>Total Year Cost (\$)</u> ⁽²⁾
02040	193.4	4	400	200	600	2400
02347	1921	35	3500	200	3700	14800
04294	798	15	1500	200	1700	6800
04296	12.4	1	100	200	300	1200
06143	85.7 ⁽³⁾	2	200	200	400	1600
35035	15.7 ⁽³⁾	1	100	200	300	1200
41061	210.8	4	400	200	600	2400
304	190.4	4	400	200	600	2400
404	72.8	2	200	200	400	1600
308	69.7	2	200	200	400	1600

(1) Based on a 24 hour/production day, 250 days production per year; however, some plants operate less hours per day and days per year.

(2) Disposal cost is based on a 1983 estimate from an EPA-approved disposal facility.

(3) 3-day sampling average.

TABLE 9-4

MODEL PLANT TREATMENT COSTS
OPTION 2

	Flow, gpd (1/day)				
	2,000 (7,570)	10,000 (37,850)	60,000 (227,000)	150,000 (568,000)	200,000 (757,000)
A. INVESTMENT COSTS					
Construction	\$ 2,500	\$ 7,000	\$ 12,000	\$ 17,000	\$ 20,200
Equipment in place including piping, fittings, electrical work and controls	28,000	83,000	142,000	202,500	244,600
Monitoring equipment in place	6,000	6,000	6,000	6,000	6,000
Engineering design and inspection	6,500	18,000	31,000	44,000	53,000
Incidentals, overhead, fees, contingencies	15,500	45,000	77,000	110,000	132,500
Land	3,000	3,000	6,000	6,000	6,000
TOTAL INVESTMENT COST	<u>\$61,500</u>	<u>\$162,000</u>	<u>\$274,000</u>	<u>\$385,500</u>	<u>\$462,300</u>
B. OPERATION AND MAINTENANCE COST					
Labor and supervision	11,000	11,000	11,000	11,000	11,000
Energy	600	1,000	5,000	6,000	7,000
Chemicals	200	1,100	4,000	9,500	12,500
Maintenance	6,000	16,000	27,500	38,000	46,000
Taxes and insurance	2,000	5,000	8,500	12,000	13,800
Residual waste disposal	1,500	8,500	52,000	108,000	128,500
Monitoring, analysis and reporting	7,500	7,500	7,500	7,500	7,500
TOTAL OPERATION AND MAINTENANCE COST	<u>\$28,800</u>	<u>\$50,100</u>	<u>\$115,500</u>	<u>\$192,000</u>	<u>\$226,300</u>
C. AMORTIZATION OF INVESTMENT COST					
	16,632	45,206	76,196	107,897	129,733
TOTAL ANNUAL COST	<u>\$45,432</u>	<u>\$95,306</u>	<u>\$191,696</u>	<u>\$399,897</u>	<u>\$356,033</u>

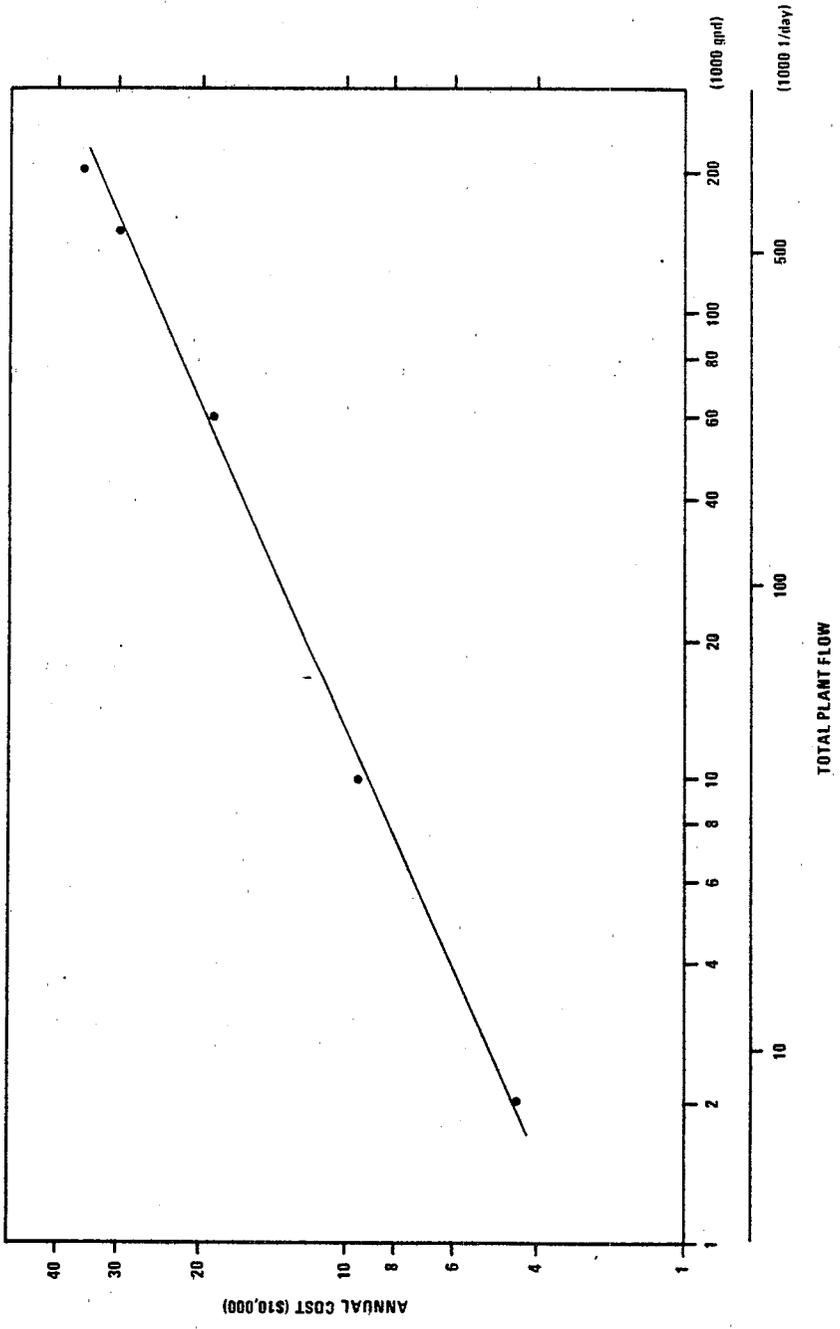


FIGURE 9-1 ANNUAL COST VS. FLOW FOR OPTION 2 TECHNOLOGY

TABLE 9-5

MODEL PLANT TREATMENT COSTS
OPTION 3

	Fluoride Stream Flow, gpd (1/day)			
	100 (378)	500 (1,890)	2,500 (9,460)	6,000 (22,700)
A. INVESTMENT COSTS				
Construction	\$ 3,300	\$ 3,300	\$ 5,500	\$ 10,100
Equipment in place including piping, fittings, electrical work and controls	40,600	40,600	67,200	121,900
Monitoring equipment in place	0	0	0	0
Engineering design and inspection	8,800	8,800	14,500	19,800
Incidentals, overhead, fees, contingencies	8,800	8,800	14,500	26,400
Land	0	0	0	0
TOTAL INVESTMENT COST	<u>\$61,500</u>	<u>\$ 61,500</u>	<u>\$101,700</u>	<u>\$178,200</u>
B. OPERATION AND MAINTENANCE COST				
Labor and supervision	5,000	20,000	20,000	20,000
Energy	50	200	350	700
Chemicals	200	1,000	5,000	12,000
Maintenance	3,100	3,100	5,100	8,900
Taxes and insurance	1,900	1,900	3,050	5,300
Residual waste disposal	700	3,500	17,500	42,000
Monitoring, analysis and reporting	1,200	1,200	1,200	1,200
TOTAL OPERATION AND MAINTENANCE COST	<u>\$12,150</u>	<u>\$30,900</u>	<u>\$52,200</u>	<u>\$90,100</u>
C. AMORTIZATION OF INVESTMENT COST				
	17,500	17,500	28,900	50,700
TOTAL ANNUAL COST	<u>\$29,650</u>	<u>\$48,400</u>	<u>\$81,100</u>	<u>\$140,800</u>

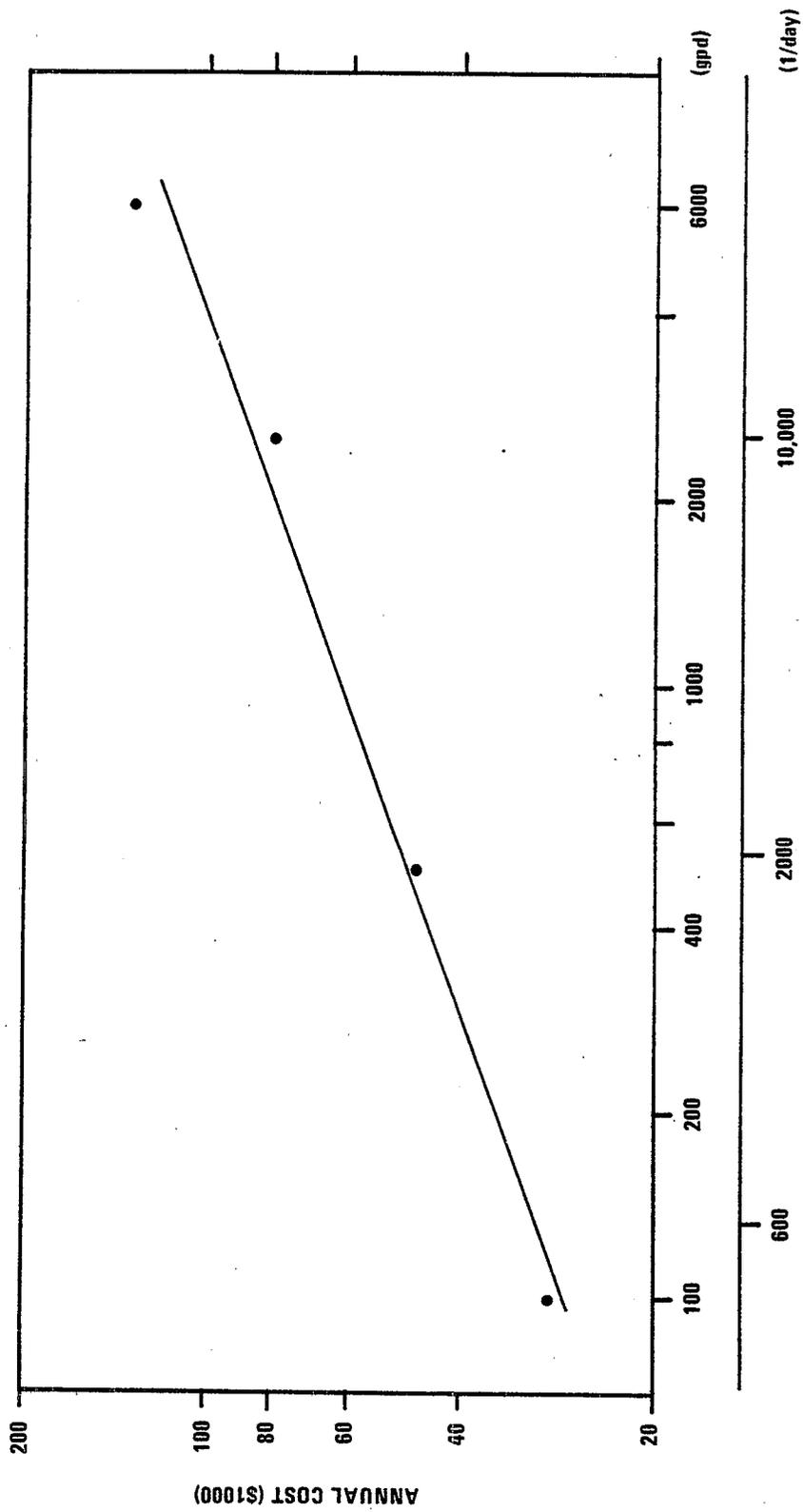


FIGURE 9-2 ANNUAL COST VS. FLOW FOR OPTION 3 TECHNOLOGY

TABLE 9-6

MODEL PLANT TREATMENT COSTS
OPTION 5, INCREMENTAL COSTS

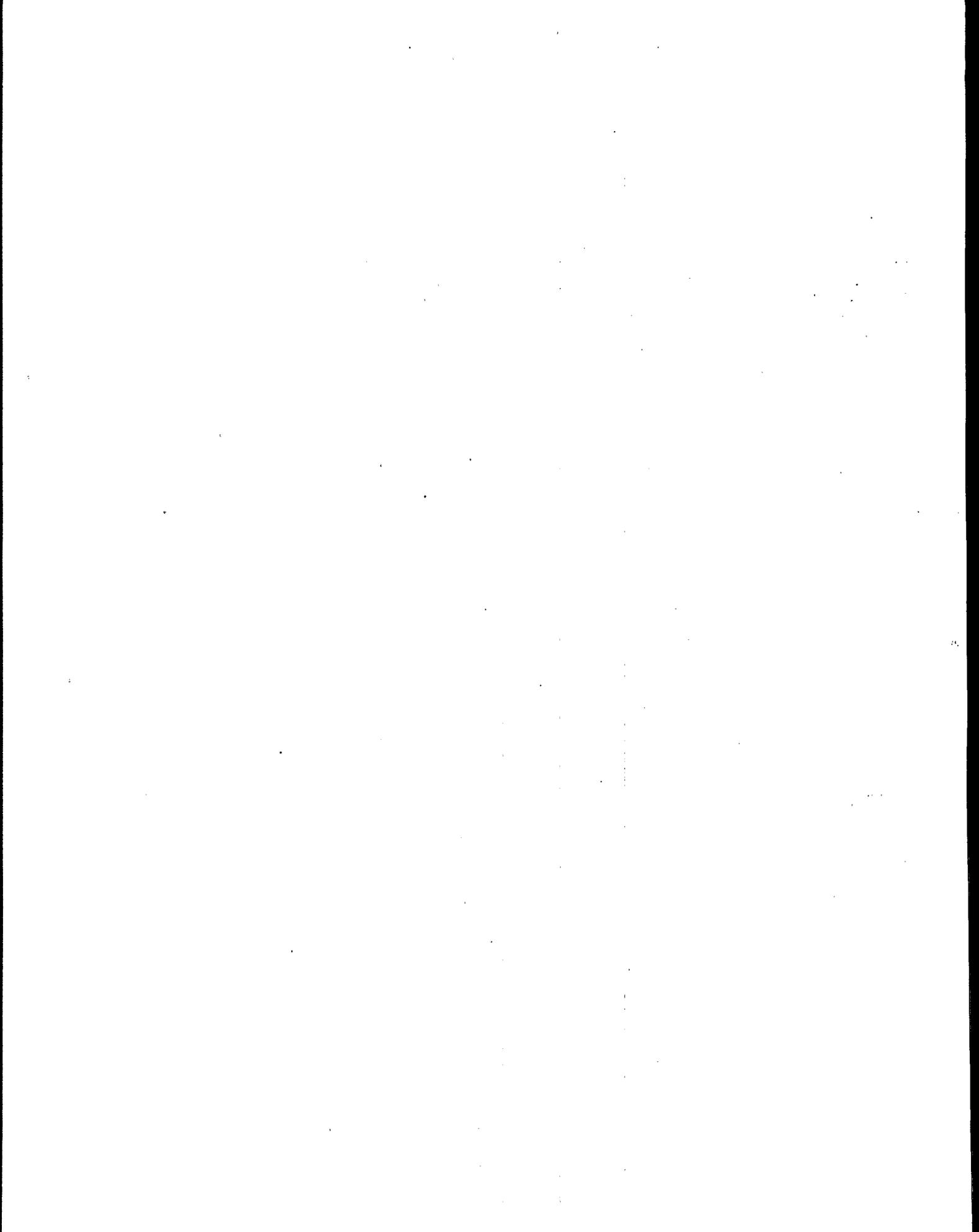
	Flow, gpd (1/day)				
	2,000 (7,570)	10,000 (37,850)	60,000 (227,000)	150,000 (568,000)	200,000 (757,000)
A. INVESTMENT COSTS					
Construction	\$ 700	\$ 800	\$ 1,600	\$ 3,300	\$ 3,800
Equipment in place including piping, fittings, electrical work and controls	6,700	7,900	16,000	33,000	38,000
Monitoring equipment in place	---	---	---	---	---
Engineering design and inspection	1,500	1,700	3,500	7,200	8,400
Incidentals, overhead, fees, contingencies	3,700	4,400	8,800	18,200	20,900
Land	---	---	---	---	---
TOTAL INVESTMENT COST	<u>\$12,600</u>	<u>\$14,800</u>	<u>\$29,900</u>	<u>\$61,700</u>	<u>\$71,100</u>
B. OPERATION AND MAINTENANCE COST					
Labor and supervision	2,000	2,000	3,000	4,000	4,000
Energy	300	500	2,500	3,000	3,500
Chemicals	---	---	---	---	---
Maintenance	1,260	1,480	3,000	6,200	7,100
Taxes and insurance	380	440	900	1,850	2,130
Residual waste disposal	---	---	---	---	---
Monitoring, analysis and reporting	---	---	---	---	---
TOTAL OPERATION AND MAINTENANCE COST	<u>\$3,940</u>	<u>\$4,420</u>	<u>\$9,400</u>	<u>\$15,050</u>	<u>\$16,730</u>
C. AMORTIZATION OF INVESTMENT COST					
	3,580	4,210	8,500	17,540	20,210
TOTAL ANNUAL COST	<u>\$7,520</u>	<u>\$8,630</u>	<u>\$17,900</u>	<u>\$32,590</u>	<u>\$36,940</u>

SECTION 10

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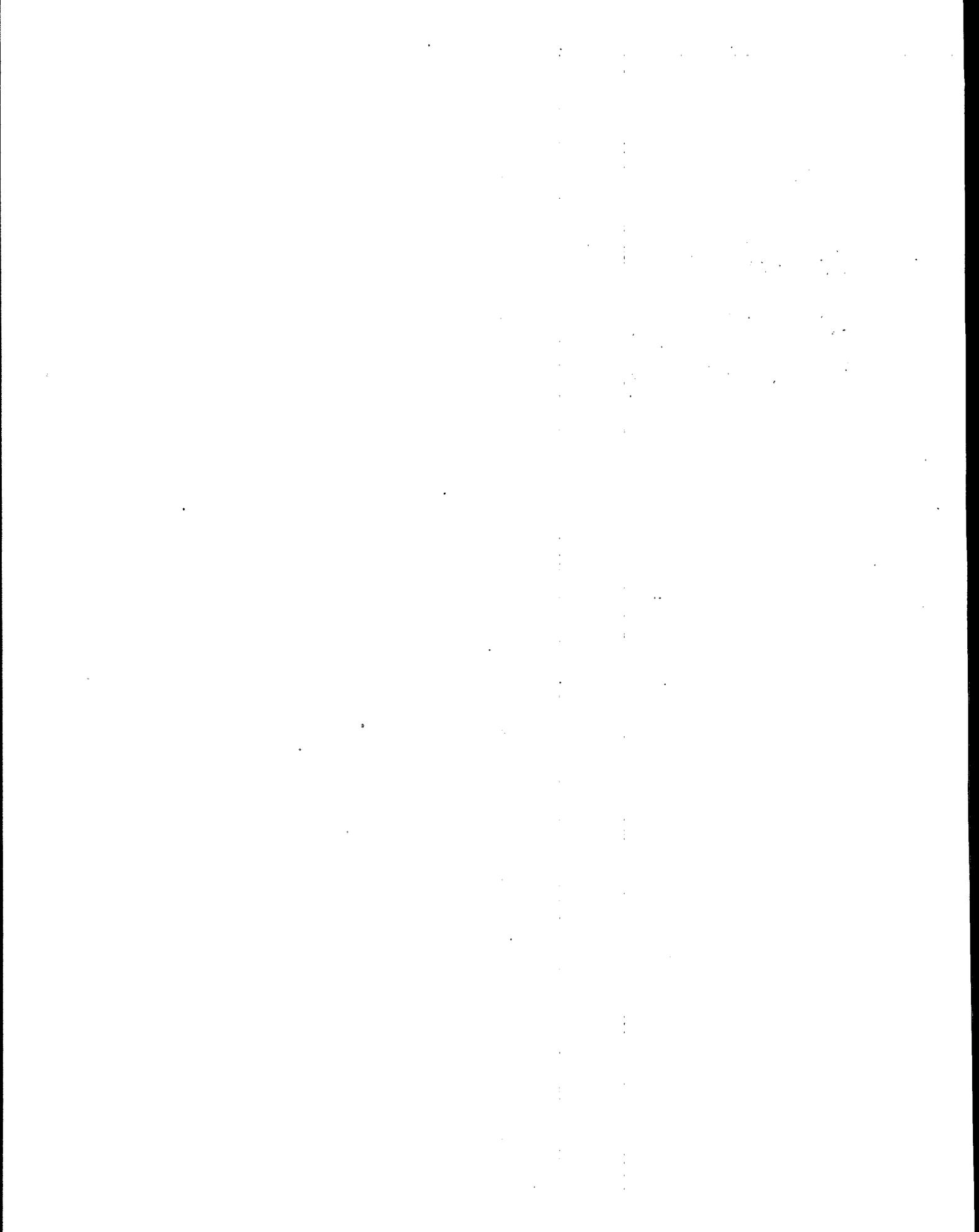
SECTION 11

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SECTION 12

GLOSSARY

Absorb - To take up matter or radiation.

Act - Federal Water Pollution Control Act.

Activate - To treat the cathode or target of an electron tube in order to create or increase the emission of electrons.

Adjustable Capacitor - A device capable of holding an electrical charge at any one of several discrete values.

Adsorption - The adhesion of an extremely thin layer of molecules (of gas, liquid) to the surface of solids (granular activated carbon for instance) or liquids with which they are in contact.

Aging - Storage of a permanent magnet, capacitor, meter or other device (sometimes with a voltage applied) until the characteristics of the device become essentially constant.

Algicide - Chemicals used to retard the growth of phytoplankton (algae) in bodies of water.

Aluminum Foil - Aluminum in the form of a sheet of thickness not exceeding 0.005 inch.

Anneal - To treat a metal, alloy, or glass by a process of heating and slow cooling in order to remove internal stresses and to make the material less brittle.

Anode - The collector of electrons in an electron tube. Also known as plate; positive electrode.

Anodizing - An electrochemical process of controlled aluminum oxidation producing a hard, transparent oxide up to several mils in thickness.

Assembly or Mechanical Attachment - The fitting together of previously manufactured parts or components into a complete machine, unit of a machine, or structure.

Autotransformer - A power transformer having one continuous winding that is tapped; part of the winding serves as the primary coil and all of it serves as the secondary coil, or vice versa.

Ballast - A circuit element that serves to limit an electric current or to provide a starting voltage, as in certain types of lamps, such as in fluorescent ceiling fixtures.

Binder - A material used to promote cohesion between particles of carbon or graphite to produce solid carbon and graphite rods or pieces.

Biochemical Oxygen Demand (BOD) - (1) The quantity of oxygen used in the biochemical oxidation of organic matter in a specified time, at a specified temperature, and under specified conditions. (2) Standard test used in assessing wastewater quality.

Biodegradable - The part of organic matter which can be oxidized by bioprocesses, e.g., biodegradable detergents, food wastes, animal manure, etc.

Biological Wastewater Treatment - Forms of wastewater treatment in which bacteria or biochemical action is intensified to stabilize, oxidize, and nitrify the unstable organic matter present. Intermittent sand filters, contact beds, trickling filters, and activated sludge processes are examples.

Breakdown Voltage - Voltage at which a discharge occurs between two electrodes.

Bulb - The glass envelope which incloses an incandescent lamp or an electronic tube.

Busbar - A heavy rigid, metallic conductor, usually uninsulated, used to carry a large current or to make a common connection between several circuits.

Bushing - An insulating structure including a central conductor, or providing a central passage for a conductor, with provision for mounting on a barrier (conducting or otherwise), for the purpose of insulating the conductor from the barrier and conducting current from one side of the barrier to the other.

Calcining - To heat to a high temperature without melting or fusing, as to heat unformed ceramic materials in a kiln, or to heat ores, precipitates, concentrates or residues so that hydrates, carbonates or other compounds are decomposed and volatile material is expelled, e.g., to heat limestone to make lime.

Calibration - The determination, checking, or correction of the graduation of any instrument giving quantitative measurements.

Capacitance - The ratio of the charge on one of the plates of a capacitor to the potential difference between the plates.

Capacitor - An electrical circuit element used to store charge temporarily, consisting in general of two conducting materials separated by a dielectric material.

Carbon - A nonmetallic, chiefly tetravalent element found native or as a constituent of coal, petroleum, asphalt, limestone, etc.

Cathode - The primary source of electrons in an electron tube; in directly heated tubes the filament is the cathode, and in indirectly heated tubes a coated metal cathode surrounds a heater.

Cathode Ray Tube - An electron-beam tube in which the beam can be focused to a small cross section on a luminescent screen and varied in position and intensity to produce a visible pattern.

Central Treatment Facility - Treatment plant which co-treats process wastewaters from more than one manufacturing operation or co-treats process wastewaters with noncontact cooling water or with non-process wastewaters (e.g., utility blow-down, miscellaneous runoff, etc.).

Centrifuge - The removal of water in a sludge and water slurry by introducing the water and sludge slurry into a centrifuge. The sludge is driven outward with the water remaining near the center. The dewatered sludge is usually landfilled.

Ceramic - A product made by the baking or firing of a nonmetallic mineral such as tile, cement, plaster, refractories, and brick.

Chemical Coagulation - The destabilization and initial aggregation of colloidal and finely divided suspended matter by the addition of a floc-forming chemical.

Chemical Oxidation - The addition of chemical agents to wastewater for the purpose of oxidizing pollutant material, e.g., removal of cyanide.

Chemical Oxygen Demand (COD) - (1) A test based on the fact that all organic compounds, with few exceptions, can be oxidized to carbon dioxide and water by the action of strong oxidizing agents under acid conditions. Organic matter is converted to carbon dioxide and water regardless of the biological

assimilability of the substances. One of the chief limitations is its inability to differentiate between biologically oxidizable and biologically inert organic matter. The major advantage of this test is the short time required for evaluation (2 hours). (2) The amount of oxygen required for the chemical oxidation of organics in a liquid.

Chemical Precipitation - (1) Formation of insoluble materials generated by addition of chemicals to a solution. (2) The process of softening water by the addition of lime and soda ash as the precipitants.

Chlorination - The application of chlorine to water or wastewater generally for the purpose of disinfection, but frequently for accomplishing other biological or chemical results.

Circuit Breaker - Device capable of making, carrying, and breaking currents under normal or abnormal circuit conditions.

Cleaning - The removal of soil and dirt (including grit and grease) from a workpiece using water with or without a detergent or other dispersing agent.

Coil - A number of turns of wire used to introduce inductance into an electric circuit, to produce magnetic flux, or to react mechanically to a changing magnetic flux.

Coil-Core Assembly - A unit made up of the coil windings of a transformer placed over the magnetic core.

Coking - (1) Destructive distillation of coal to make coke. (2) A process for thermally converting the heavy residual bottoms of crude oil entirely to lower-boiling petroleum products and by-product petroleum coke.

Colloids - A finely divided dispersion of one material called the "dispersed phase" (solid) in another material called the "dispersion medium" (liquid). Normally negatively charged.

Composite Wastewater Sample - A combination of individual samples of water or wastewater taken at selected intervals and mixed in proportion to flow or time to minimize the effect of the variability of an individual sample.

Concentric Windings - Transformer windings in which the low-voltage winding is in the form of a cylinder next to the core, and the high-voltage winding, also cylindrical, surrounds the low-voltage winding.

Conductor - A wire, cable, or other body or medium suitable for carrying electric current.

Conduit - Tubing of flexible metal or other material through which insulated electric wires are run.

Contamination - A general term signifying the introduction into water of microorganisms, chemicals, wastes or sewage which renders the water unfit for its intended use.

Contractor Removal - The disposal of oils, spent solutions, or sludge by means of a scavenger service.

Conversion Coating - As metal-surface coating consisting of compound of the base metal.

Cooling Tower - A device used to cool manufacturing process water before returning the water for reuse.

Copper - A common, reddish, chiefly univalent and bivalent metallic element that is ductile and malleable and one of the best conductors of heat and electricity.

Core (Magnetic Core) - A quantity of ferrous material placed in a coil or transformer to provide a better path than air for magnetic flux, thereby increasing the inductance of the coil or increasing the coupling between the windings of a transformer.

Corona Discharge - A discharge of electricity appearing as a bluish-purple glow on the surface of and adjacent to a conductor when the voltage gradient exceeds a certain critical value; caused by ionization of the surrounding air by the high voltage.

Curing - A heating/drying process carried out in an elevated-temperature enclosure.

Current Carrying Capacity - The maximum current that can be continuously carried without causing permanent deterioration of electrical or mechanical properties of a device or conductor.

Dag (Aquadag) - A conductive graphite coating on the inner and outer side walls of some cathode-ray tubes.

Degreasing - The process of removing grease and oil from the surface of the basis material.

Dewatering - A process in which water is removed from sludge.

Dicing - Sawing or otherwise machining a semiconductor wafer into small squares or dice from which transistors and diodes can be fabricated.

Die - A tool or mold used to cut shapes to or form impressions on materials such as metals and ceramics.

Die Cutting (Also Blanking) - Cutting of plastic or metal sheets into shapes by striking with a punch.

Dielectric - A material that is highly resistant to the conductance of electricity; an insulator.

Di-n-octyl-phthalate - A liquid dielectric that is presently being substituted for a PCB dielectric fluid.

Diode (Semiconductor), (Also Crystal Diode, Crystal Rectifier) - A two-electrode semiconductor device that utilizes the rectifying properties of a p-n junction or point contact.

Discrete Device - Individually manufactured transistor, diode, etc.

Dissolved Solids - Theoretically the anhydrous residues of the dissolved constituents in water. Actually the term is defined by the method used in determination. In water and wastewater treatment, the Standard Methods tests are used.

Distribution Transformer - An element of an electric distribution system located near consumers which changes primary distribution voltage to a lower consumer voltage.

Dopant - An impurity element added to semiconductor materials used in crystal diodes and transistors.

Dragout - The solution that adheres to the part or workpiece and is carried past the edge of the tank.

Dry Electrolytic Capacitor - An electrolytic capacitor with a paste rather than liquid electrolyte.

Drying Beds - Areas for dewatering of sludge by evaporation and seepage.

Dry Slug - Usually refers to a plastic-encased sintered tantalum slug type capacitor.

Dry Transformer - Having the core and coils neither impregnated with an insulating fluid nor immersed in an insulating oil.

Effluent - The quantities, rates, and chemical, physical, biological and other constituents of waters which are discharged from point sources.

Electrochemical Machining - Shaping of an anode by the following process: The anode and cathode are placed close together and electrolyte is pumped into the space between them. An electrical potential is applied to the electrodes causing anode metal to be dissolved selectively, producing a shaped anode that complements the shape of the cathode.

Electrolyte - A nonmetallic electrical conductor in which current is carried by the movement of ions.

Electron Beam Lithography - Similar to photolithography - A fine beam of electrons is used to scan a pattern and expose an electron-sensitive resist in the unmasked areas of the object surface.

Electron Discharge Lamp - An electron lamp in which light is produced by passage of an electric current through a metallic vapor or gas.

Electron Gun - An electrode structure that produces and may control, focus, deflect and converge one or more electron beams in an electron tube.

Electron Tube - An electron device in which conduction of electricity is accomplished by electrons moving through a vacuum or gaseous medium within a gas-tight envelope.

Electroplating - The production of a thin coating of one metal on another by electrode position.

Emissive Coating - An oxide coating applied to an electrode to enhance the emission of electrons.

Emulsion Breaking - Decreasing the stability of dispersion of one liquid in another.

End-of-Pipe Treatment - The reduction and/or removal of pollutants by chemical treatment just prior to actual discharge.

Epitaxial Layer - A (thin) semiconductor layer having the same crystalline orientation as the substrate on which it is grown.

Epitaxial Transistor - Transistor with one or more epitaxial layers.

Equalization - The process whereby waste streams from different sources varying in pH, chemical constituents, and flow rates are collected in a common container. The effluent stream from this equalization tank will have a fairly constant flow and pH level, and will contain a homogeneous chemical mixture. This tank will help to prevent unnecessary shock to the waste treatment system.

Etch - To corrode the surface of a metal in order to reveal its composition and structure.

Extrusion - Forcing the carbon-binder-mixture through a die under extreme pressure to produce desirable shapes and characteristics of the piece.

Field-effect Transistors - Transistors made by the metal-oxide-semiconductor (MOS) technique, differing from bipolar ones in that only one kind of charge carrier is active in a single device. Those that employ electrons are called n-MOS transistors; those that employ holes are p-MOS transistors.

Filament - (1) Metallic wire which is heated in an incandescent lamp to produce light by passing an electron current through it. (2) A cathode in a fluorescent lamp that emits electrons when electric current is passed through it.

Filtering Capacitor - A capacitor used in a power-supply filter system to provide a low-reactance path for alternating currents and thereby suppress ripple currents, without affecting direct currents.

Fixed Capacitor - A capacitor having a definite capacitance value that cannot be adjusted.

Float Gauge - A device for measuring the elevation of the surface of a liquid, the actuating element of which is a buoyant float that rests on the surface of the liquid and rises or falls with it. The elevation of the surface is measured by a chain or tape attached to the float.

Floc - A very fine, fluffy mass formed by the aggregation of fine suspended particles.

Flocculation - In water and wastewater treatment, the agglomeration of colloidal and finely divided suspended matter after coagulation by gentle stirring by either mechanical or hydraulic means. In biological wastewater treatment where coagulation is not used, agglomeration may be accomplished biologically.

Flocculator - An apparatus designed for the formation of floc in water or sewage.

Flow-proportioned Sample - A sampled stream whose pollutants are apportioned to contributing streams in proportion to the flow rates of the contributing streams.

Fluorescent Lamp - An electric discharge lamp in which phosphor materials transform ultraviolet radiation from mercury vapor ionization to visible light.

Forming - Application of voltage to an electrolytic capacitor, electrolytic rectifier or semiconductor device to produce a desired permanent change in electrical characteristics as part of the manufacturing process.

Frit Seal - A seal made by fusing together metallic powders with a glass binder for such applications as hermetically sealing ceramic packages for integrated circuits.

Funnel - The rear, funnel-shaped portion of the glass enclosure of a cathode ray tube.

Fuse - Overcurrent protective device with a circuit-opening fusible part that would be heated and severed by overcurrent passage.

Gate - One of the electrodes in a field effect transistor.

Getter - A metal coating inside a lamp which is activated by an electric current to absorb residual water vapor and oxygen.

Glass - A hard, amorphous, inorganic, usually transparent, brittle substance made by fusing silicates, and sometimes borates and phosphates, with certain basic oxides and then rapidly cooling to prevent crystallization.

Glow Lamp - An electronic device, containing at least two electrodes and an inert gas, in which light is produced by a cloud of electrons close to the negative electrode when a voltage is applied between the electrodes.

Grab Sample - A single sample of wastewater taken at an "instant" in time.

Graphite - A soft black lustrous carbon that conducts electricity and is a constituent of coal, petroleum, asphalt, limestone, etc.

Grease - In wastewater, a group of substances including fats, waxes, free fatty acids, calcium and magnesium soaps, mineral oil and certain other nonfatty materials. The type of solvent and method used for extraction should be stated for quantification.

Grease Skimmer - A device for removing grease or scum from the surface of wastewater in a tank.

Green Body - An unbaked carbon rod or piece that is usually soft and quite easily broken.

Grid - An electrode located between the cathode and anode of an electron tube, which has one or more openings through which electrons or ions can pass, and which controls the flow of electrons from cathode to anode.

Grinding - The process of removing stock from a workpiece by the use of abrasive grains held by a rigid or semi-rigid binder.

Hardness - A characteristic of water, imparted by calcium, magnesium, and ion salts such as bicarbonates, carbonates, sulfates, chlorides, and nitrates. These cause curdling of soap, deposition of scale in boilers, damage in some industrial processes and sometimes objectionable taste. Hardness may be determined by a standard laboratory procedure or computed from the amounts of calcium and magnesium as well as iron, aluminum, manganese, barium, strontium, and zinc, and is expressed as equivalent calcium carbonate.

Heavy Metals - A general name given to the ions of metallic elements such as copper, zinc, chromium, and nickel. They are normally removed from wastewater by an insoluble precipitate (usually a metallic hydroxide).

Holding Tank - A reservoir to contain preparation materials so as to be ready for immediate service.

Hybrid Integrated Circuits - A circuit that is part integrated and part discrete.

Impact Extrusion - A cold extrusion process for producing tubular components by striking a slug of the metal, which has been placed in the cavity of the die, with a punch moving at high velocity.

Impregnate - To force a liquid substance into the spaces of a porous solid in order to change its properties.

Incandescent Lamp - An electric lamp producing light in which a metallic filament is heated white-hot in a vacuum by passage of an electric current through it.

Industrial Wastes - The liquid wastes from industrial processes as distinct from domestic or sanitary wastes.

Influent - Water or other liquid, either raw or partly treated, flowing into a reservoir basin or treatment plant.

In-Process Control Technology - The regulation and conservation of chemicals and rinse water at their point of use as opposed to end-of-pipe treatment.

Insulating Paper - A standard material for insulating electrical equipment, usually consisting of bond or kraft paper coated with black or yellow insulating varnish on both sides.

Insulation (Electrical Insulation) - A material having high electrical resistivity and therefore suitable for separating adjacent conductors in an electric circuit or preventing possible future contact between conductors.

Insulator - A nonconducting support for an electric conductor.

Integrated Circuit - Assembly of electronic devices interconnected into circuits.

Interleaved Winding - An arrangement of winding coils around a transformer core in which the coils are wound in the form of a disk, with a group of disks for the low-voltage windings stacked alternately with a group of disks for the high-voltage windings.

Intermittent Filter - A natural or artificial bed of sand or other fine-grained material onto which sewage is intermittently flooded and through which it passes, with time allowed for filtration and the maintenance of aerobic conditions.

Ion Exchange - A reversible chemical reaction between a solid (ion exchanger) and a fluid (usually a water solution) by means of which ions may be interchanged from one substance to another. The superficial physical structure of the solid is not affected.

Ion Exchange Resins - Synthetic resins containing active groups (usually sulfonic, carboxylic, phenol, or substituted amino groups) that give the resin the ability to combine with or exchange ions with a solution.

- Ion Implantation - A process of introducing impurities into the near surface regions of solids by directing a beam of ions at the solid.
- Junction - A region of transition between two different semiconducting regions in a semiconductor device such as a p-n junction, or between a metal and a semiconductor.
- Junction Box - A protective enclosure into which wires or cables are led and connected to form joints.
- Knife Switch - Form of switch where moving blade enters stationary contact clips.
- Klystron - An evacuated electron-beam tube in which an initial velocity modulation imparted to electrons in the beam results subsequently in density modulation of the beam; used as an amplifier in the microwave region or as an oscillator.
- Lagoon - A man-made pond or lake for holding wastewater for the removal of suspended solids. Lagoons are also used as retention ponds after chemical clarification to polish the effluent and to safeguard against upsets in the clarifier; for stabilization of organic matter by biological oxidation; for storage of sludge; and for cooling of water.
- Landfill - The disposal of inert, insoluble waste solids by dumping at an approved site and covering with earth.
- Lapping - The mechanical abrasion or surface planing of the semiconductor wafer to produce desired surface and wafer thickness.
- Lime - Any of a family of chemicals consisting essentially of calcium hydroxide made from limestone (calcite) which is composed almost wholly of calcium carbonates or a mixture of calcium and magnesium carbonates.
- Limiting Orifice - A device that limits flow by constriction to a relatively small area. A constant flow can be obtained over a wide range of upstream pressures.
- Machining - The process of removing stock from a workpiece by forcing a cutting tool through the workpiece and removing a chip of basis material. Machining operations such as turning, milling, drilling, boring, tapping, planing, broaching, sawing and cutoff, shaving, threading, reaming, shaping, slotting, hobbing, filling, and chambering are included in this definition.

Magnaflux Inspection - Trade name for magnetic particle test.

Make-up Water - Total amount of water used by any process/process step.

Mandrel - A metal support serving as a core around which the metals are wound and annealed to form a central hole.

Mask (Shadow Mask) - Thin sheet steel screen with thousands of apertures through which electron beams pass to a color picture tube screen. The color of an image depends on the balance from each of three different electron beams passing through the mask.

Metal Oxide Semiconductor Device - A metal insulator semiconductor structure in which the insulating layer is an oxide of the substrate material; for a silicon substrate, the insulating layer is silicon dioxide (SiO₂).

Mica - A group of aluminum silicate minerals that are characterized by their ability to split into thin, flexible flakes because of their basal cleavage.

Miligrams Per Liter (mg/l) - This is a weight per volume designation used in water and wastewater analysis.

Mixed Media Filtration - A filter which uses two or more filter materials of differing specific gravities selected so as to produce a filter uniformly graded from coarse to fine.

MOS - (See Metal Oxide Semiconductor).

Mount Assembly - Funnel neck ending of picture tube holding electron gun(s).

National Pollutant Discharge Elimination System (NPDES) - The federal mechanism for regulating point source discharge by means of permits.

Neutralization - Chemical addition of either acid or base to a solution such that the pH is adjusted to approximately 7.

Noncontact Cooling Water - Water used for cooling which does not come into direct contact with any raw material, intermediate product, waste product or finished product.

Oil-Filled Capacitor - A capacitor whose conductor and insulating elements are immersed in an insulating fluid that is usually, but not necessarily, oil.

- Outfall - The point or location where sewage or drainage discharges from a sewer, drain, or conduit.
- Oxide Mask - Oxidized layer of silicon wafer through which "windows" are formed which will allow for dopants to be introduced into the silicon.
- Panel - The front, screen portion of the glass enclosure of a cathode ray tube.
- PCB (Polychlorinated Biphenyl) - A colorless liquid, used as an insulating fluid in electrical equipment. (The future use of PCB for new transformers was banned by the Toxic Substances Control Act of October 1976).
- pH - The negative of the logarithm of the hydrogen ion concentration. Neutral water has a pH value of 7. At pH lower than 7, a solution is acidic. At pH higher than 7, a solution is alkaline.
- pH Adjustment - A means of maintaining the optimum pH through the use of chemical additives. Can be manual, automatic, or automatic with flow corrections.
- Phase - One of the separate circuits or windings of a polyphase system, machine or other apparatus.
- Phase Assembly - The coil-core assembly of a single phase of a transformer.
- Phosphate Coating - A conversion coating on metal, usually steel, produced by dipping it into a hot aqueous solution of iron, zinc, or manganese phosphate.
- Phosphor - Crystalline inorganic compounds that produce light when excited by ultraviolet radiation.
- Photolithography - The process by which a microscopic pattern is transferred from a photomask to a material layer (e.g., SiO₂) in an actual circuit.
- Photomask - A film or glass negative that has many high-resolution images, used in the production of semiconductor devices and integrated circuits.
- Photon - A quantum of electromagnetic energy.
- Photoresist - A light-sensitive coating that is applied to a substrate or board, exposed, and developed prior to chemical etching; the exposed areas serve as a mask for selective etching.

Picture Tube - A cathode ray tube used in television receivers to produce an image by varying the electron beam intensity as the beam scans a fluorescent screen.

Plate - (1) Preferably called the anode. The principal electrode to which the electron stream is attracted in an electron tube. (2) One of the conductive electrodes in a capacitor.

Polar Capacitor - An electrolytic capacitor having an oxide film on only one foil or electrode which forms the anode or positive terminal.

Pole Type Transformer - A transformer suitable for mounting on a pole or similar structure.

Poling - A step in the production of ceramic piezoelectric bodies which orients the axes of the crystallites in the preferred direction.

Polishing - The process of removing stock from a workpiece by the action of loose or loosely held abrasive grains carried to the workpiece by a flexible support. Usually, the amount of stock removed in a polishing operation is only incidental to achieving a desired surface finish or appearance.

Pollutant - The term "pollutant" means dredged spoil,, solid wastes, incinerator residue, sewage, garbage, sewage sludge, munitions, chemical wastes, biological materials, radioactive materials, heat, wrecked or discarded equipment, rock, sand, cellar dirt and industrial, municipal and agricultural waste discharged into water.

Pollutant Parameters - Those constituents of wastewater determined to be detrimental and, therefore, requiring control.

Pollution Load - A measure of the unit mass of a wastewater in terms of its solids or oxygen-demanding characteristics, or in terms of harm to receiving waters.

Polyelectrolytes - Synthetic or natural polymers containing ionic constituents, used as a coagulant or a coagulant aid in water and wastewater treatment.

Power Regulators - Transformers used to maintain constant output current for changes in temperature output load, line current, and time.

Power Transformer - Transformer used at a generating station to step up the initial voltage to high levels for transmission.

Prechlorination - (1) Chlorination of water prior to filtration.
(2) Chlorination of sewage prior to treatment.

Precipitate - The discrete particles of material settled from a liquid solution.

Pressure Filtration - The process of solid/liquid phase separation effected by passing the more permeable liquid phase through a mesh which is impenetrable to the solid phase.

Pretreatment - Any wastewater treatment process used to reduce pollution load partially before the wastewater is introduced into a main sewer system or delivered to a treatment plant for substantial reduction of the pollution load.

Primary Feeder Circuit (Substation) Transformers - These transformers (at substations) are used to reduce the voltage from the subtransmission level to the primary feeder level.

Primary Treatment - A process to remove substantially all floating and settleable solids in wastewater and partially to reduce the concentration of suspended solids.

Primary Winding - Winding on the supply (i.e. input) side of a transformer.

Priority Pollutant - The 129 specific pollutants established by the EPA from the 65 pollutants and classes of pollutants as outlined in the consent decree of June 8, 1976.

Process Wastewater - Any water which, during manufacturing or processing, comes into direct contact with or results from the production or use of any raw materials, intermediate product, finished product, by-product, or waste product.

Process Water - Water prior to its direct contact use in a process or operation. (This water may be any combination of a raw water, service water, or either process wastewater or treatment facility effluent to be recycled or reused.)

Pyrolysis - The breaking apart of complex molecules into simpler units by the use of heat, as in the pyrolysis of heavy oil to make gasoline.

Quenching - Shock cooling by immersion of liquid or molten material in a cooling medium (liquid or gas). Used in metallurgy, plastics forming, and petroleum refining.

Raceway - A channel used to hold and protect wires, cables or busbars.

Rapid Sandfilter - A filter for the purification of water where water which has been previously treated, usually by coagulation and sedimentation, is passed through a filtering medium consisting of a layer of sand or prepared anthracite coal or other suitable material, usually from 24 to 30 inches thick and resting on a supporting bed of gravel or a porous medium such as carborundum. The filtrate is removed by a drain system. The filter is cleaned periodically by reversing the flow of the water through the filtering medium. Sometimes supplemented by mechanical or air agitation during backwashing to remove mud and other impurities.

Raw Wastewater - Plant water prior to any treatment or use.

Rectifier - (1) A device for converting alternating current into direct current. (2) A nonlinear circuit component that, ideally, allows current to flow in one direction unimpeded but allows no current to flow in the other direction.

Recycled Water - Process wastewater or treatment facility effluent which is recirculated to the same process.

Resistor - A device designed to provide a definite amount of resistance, used in circuits to limit current flow or to provide a voltage drop.

Retention Time - The time allowed for solids to collect in a settling tank. Theoretically retention time is equal to the volume of the tank divided by the flow rate. The actual retention time is determined by the purpose of the tank. Also, the design residence time in a tank or reaction vessel which allows a chemical reaction to go to completion, such as the reduction of hexavalent chromium or the destruction of cyanide.

Reused Water - Process wastewater or treatment facility effluent which is further used in a different manufacturing process.

Rinse - Water for removal of dragout by dipping, spraying, fogging etc.

Sanitary Sewer - A sewer that carries liquid and water wastes from residences, commercial buildings, industrial plants, and institutions together with ground, storm, and surface waters that are not admitted intentionally.

Sanitary Water - The supply of water used for sewage transport and the continuation of such effluents to disposal.

Secondary Settling Tank - A tank through which effluent from some prior treatment process flows for the purpose of removing settleable solids.

Secondary Wastewater Treatment - The treatment of wastewater by biological methods after primary treatment by sedimentation.

Secondary Winding - Winding on the load (i.e. output) side of a transformer.

Sedimentation - Settling of matter suspended in water by gravity. It is usually accomplished by reducing the velocity of the liquid below the point at which it can transport the suspended material.

Semiconductor - (1) A solid crystalline material whose electrical conductivity is intermediate between that of a metal and an insulator. (2) A solid state electrical device that performs functions such as information processing and display, power handling, and interconversion between light energy and electrical energy.

Settleable Solids - (1) That matter in wastewater which will not stay in suspension during a preselected settling period, such as one hour, but either settles to the bottom or floats to the top. (2) In the Imhoff cone test, the volume of matter that settles to the bottom of the cone in one hour.

Sewer - A pipe or conduit, generally closed, but normally not flowing full, for carrying sewage and other waste liquids.

Silvering - The deposition of thin films of silver on glass, etc. carried by one of several possible processes.

Skimming Tank - A tank so designed that floating matter will rise and remain on the surface of the wastewater until removed, while the liquid discharges continuously under walls or scum boards.

Sludge - The solids (and accompanying water and organic matter) which are separated from sewage or industrial wastewater.

Sludge Cake - The material resulting from air drying or dewatering sludge (usually forkable or spadable).

Sludge Disposal - The final disposal of solid wastes.

Sludge Thickening - The increase in solids concentration of sludge in a sedimentation or digestion tank.

Snubber - Shock absorber.

Soldering - The process of joining metals by flowing a thin (capillary thickness) layer of nonferrous filler metal into the space between them. Bonding results from the intimate contact produced by the dissolution of a small amount of base metal in the molten filler metal, without fusion of the base metal.

Solvent - A liquid capable of dissolving or dispersing one or more other substances.

Solvent Degreasing - The removal of oils and grease from a workpiece using organic solvents or solvent vapors.

Sputtering - A process to deposit a thin layer of metal on a solid surface in a vacuum. Ions bombard a cathode which emits the metal atoms.

Stacked Capacitor - Device containing multiple layers of dielectric and conducting materials and designed to store electrical charge.

Stamping - Almost any press operations including blanking, shearing, hot or cold forming, drawing, blending, or coining.

Steel - An iron-based alloy, malleable under proper conditions, containing up to about 2% carbon.

Step-Down Transformers - (Substation) - A transformer in which the AC voltages of the secondary windings are lower than those applied to the primary windings.

Step-Up Transformer - Transformer in which the energy transfer is from a low-voltage primary (input) winding to a high-voltage secondary (output) winding or windings.

Studs - Metal pins in glass of picture tube onto which shadow mask is hung.

Substation - Complete assemblage of plant, equipment, and the necessary buildings at a place where electrical energy is received (from one or more power-stations) for conversion (e.g. from AC to DC by means of rectifiers, rotary converters), for stepping-up or down by means of transformers, or for control (e.g. by means of switch-gear, etc.).

Subtransmission (Substation) Transformers - At the end of a transmission line, the voltage is reduced to the subtransmission level (at substations) by subtransmission transformers.

Suspended Solids - (1) Solids that are either floating or in suspension in water, wastewater, or other liquids, and which are largely removable by laboratory filtering. (2) The quantity of material removed from wastewater in a laboratory test, as prescribed in "Standard Methods for the Examination of Water and Wastewater" and referred to as non-filterable residue.

Tantalum - A lustrous, platinum-gray ductile metal used in making dental and surgical tools, penpoints, and electronic equipment.

Tantalum Foil - A thin sheet of tantalum, usually less than 0.006 inch thick.

Terminal - A screw, soldering lug, or other point to which electric connections can be made.

Testing - A procedure in which the performance of a product is measured under various conditions.

Thermoplastic Resin - A plastic that solidifies when first heated under pressure, and which cannot be remelted or remolded without destroying its original characteristics; examples are epoxides, melamines, phenolics and ureas.

Transformer - A device used to transfer electric energy, usually that of an alternating current, from one circuit to another; especially, a pair of multiply-wound, inductively coupled wire coils that effect such a transfer with a change in voltage, current, phases, or other electric characteristics.

Transistor - An active component of an electronic circuit consisting of a small block of semiconducting material to which at least three electrical contacts are made; used as an amplifier, detector, or switch.

Trickling Filter - A filter consisting of an artificial bed of coarse material, such as broken stone, clinkers, slats, or brush over which sewage is distributed and applied in drops, films, or spray, from troughs, drippers, moving distributors or fixed nozzles and through which it trickles to the underdrain giving opportunity for the formation of zoogical slimes which clarify the oxidized sewage.

Trimmer Capacitors - These are relatively small variable capacitors used in parallel with larger variable or fixed capacitors to permit exact adjustment of the capacitance of the parallel combination.

Vacuum Filter - A filter consisting of a cylindrical drum mounted on horizontal axis, covered with a filter cloth revolving with a partial submergence in liquid. A vacuum is maintained under the cloth for the larger part of a revolution to extract moisture and the cake is scraped off continuously.

Vacuum Metalizing - The process of coating a workpiece with metal by flash heating metal vapor in a high-vacuum chamber containing the workpiece. The vapor condenses on all exposed surfaces.

Vacuum Tube - An electron tube vacuated to such a degree that its electrical characteristics are essentially unaffected by the presence of residual gas or vapor.

Variable Capacitor - A device whose capacitance can be varied continuously by moving one set of metal plates with respect to another.

Voltage Breakdown - The voltage necessary to cause insulation failure..

Voltage Regulator - Like a transformer, it corrects changes in current to provide continuous, constant current flow.

Welding - The process of joining two or more pieces of material by applying heat, pressure or both, with or without filler material, to produce a localized union through fusion or recrystallization across the interface.

Wet Air Scrubber - Air pollution control device which uses a liquid or vapor to absorb contaminants and which produces a wastewater stream.

Wet Capacitor - (See oil-filled capacitor).

Wet Slug Capacitor - Refers to a sintered tantalum capacitor where the anode is placed in a metal can, filled with an electrolyte and then sealed.

Wet Tantalum Capacitor - A polar capacitor the cathode of which is a liquid electrolyte (a highly ionized acid or salt solution).

Wet Transformer - Having the core and coils immersed in an insulating oil.

Yoke A set of coils placed over the neck of a magnetically deflected cathode-ray tube to deflect the electron beam horizontally and vertically when suitable currents are passed through the coils.