

**BOILER NUMBER ONE  
1st QUARTER 2017  
HYDROGEN CHLORIDE  
EMISSIONS TEST REPORT**

**22 MARCH 2017**



**L'ANSE WARDEN ELECTRIC COMPANY, LLC.**  
157 South Main Street  
L'Anse, Michigan 49946

May 2017

W.O. No. 14464.007.006

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## **RENEWABLE OPERATING PERMIT REPORT CERTIFICATION**

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MICHIGAN DEPARTMENT OF ENVIRONMENTAL QUALITY  
AIR QUALITY DIVISION

**RENEWABLE OPERATING PERMIT  
REPORT CERTIFICATION**

Authorized by 1994 P.A. 451, as amended. Failure to provide this information may result in civil and/or criminal penalties.

Reports submitted pursuant to R 336.1213 (Rule 213), subrules (3)(c) and/or (4)(c), of Michigan's Renewable Operating (RO) Permit program must be certified by a responsible official. Additional information regarding the reports and documentation listed below must be kept on file for at least 5 years, as described in General Condition No. 22 in the RO Permit and be made available to the Department of Environmental Quality, Air Quality Division upon request.

Source Name L'Anse Warden Electric Company LLC County Baraga

Source Address 157 S. Main Street City L'Anse

AQD Source ID (SRN) B4260 RO Permit No. MI-ROP-B4260-2011 RO Permit Section No. \_\_\_\_\_

Please check the appropriate box(es):

**Annual Compliance Certification (General Condition No. 28 and No. 29 of the RO Permit)**

Reporting period (provide inclusive dates): From \_\_\_\_\_ To \_\_\_\_\_

1. During the entire reporting period, this source was in compliance with **ALL** terms and conditions contained in the RO Permit, each term and condition of which is identified and included by this reference. The method(s) used to determine compliance is/are the method(s) specified in the RO Permit.

2. During the entire reporting period this source was in compliance with all terms and conditions contained in the RO Permit, each term and condition of which is identified and included by this reference, **EXCEPT** for the deviations identified on the enclosed deviation report(s). The method used to determine compliance for each term and condition is the method specified in the RO Permit, unless otherwise indicated and described on the enclosed deviation report(s).

**Semi-Annual (or More Frequent) Report Certification (General Condition No. 23 of the RO Permit)**

Reporting period (provide inclusive dates): From \_\_\_\_\_ To \_\_\_\_\_

1. During the entire reporting period, **ALL** monitoring and associated recordkeeping requirements in the RO Permit were met and no deviations from these requirements or any other terms or conditions occurred.

2. During the entire reporting period, all monitoring and associated recordkeeping requirements in the RO Permit were met and no deviations from these requirements or any other terms or conditions occurred, **EXCEPT** for the deviations identified on the enclosed deviation report(s).

**Other Report Certification**

Reporting period (provide inclusive dates): From \_\_\_\_\_ To \_\_\_\_\_

Additional monitoring reports or other applicable documents required by the RO Permit are attached as described:  
Emissions Test Report

I certify that, based on information and belief formed after reasonable inquiry, the statements and information in this report and the supporting enclosures are true, accurate and complete.

<u>James R. Richardson</u>	<u>Technical Manager</u>	<u>907-885-7187</u>
Name of Responsible Official (print or type)	Title	Phone Number
<u>James R. Richardson</u>	<u>Consultant to LWEC</u>	<u>9-May-2017</u>
Signature of Responsible Official		Date

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# **1. INTRODUCTION**

Weston Solutions, Inc. (WESTON) has been retained by L'Anse Warden Electric Company, LLC (LWEC) to perform an emissions testing program on the Boiler No. 1 exhaust duct at the LWEC facility located in L'Anse, Baraga County, Michigan. Boiler No. 1 was previously a coal, oil, and gas-fired steam generating station and has been converted to burn biomass. The objective of this test program is to satisfy the requirements set forth by the Michigan Department of Environmental Quality (MDEQ) Air Quality Division (AQD) Consent Order (AQD No. 35-2016). The Consent Order contains provisions requiring four successive quarters of emission stack testing for hydrogen chloride (HCl) on EUBOILER No. 1, followed by two semi-annual HCl emission stack tests for HCl, followed by one further HCl emission stack test within the succeeding three years thereafter. Boiler No. 1 is identified as EUBOILER No. 1, and the facility currently operates under the State of Michigan Renewable Operating Permit (ROP) No. MI-ROP-B4260-2011 and Permit to Install (PTI) 67-16.

WESTON's Integrated Air Services (IAS) group completed the first quarter 2017 required testing on 22 March 2017. Mr. Tom Gasloli of the MDEQ was present throughout the testing.

## **1.1 PLANT INFORMATION**

L'Anse Warden Electric Company, LLC  
157 South Main Street  
L'Anse, Michigan 49946  
Mr. JR Richardson  
Phone: 906-885-7187

## **1.2 TESTING FIRM INFORMATION**

Weston Solutions, Inc.  
1400 Weston Way  
West Chester, PA 19380  
Mr. Ken Hill  
Phone: 610-701-3043

### 1.3 SUMMARY OF TEST PARAMETERS

Table 1-1 provides the test parameters, associated test methods, and reporting units for this test program.

**Table 1-1  
Summary of Test Parameters**

<b>Test Parameter<sup>(1)</sup></b>	<b>Test Method<sup>(2)</sup></b>	<b>Reporting Units</b>
Volumetric Flow Rate (VFR)	EPA M1-4	dscfm
Hydrogen Chloride (HCl)	EPA M26A (modified)	ppmvd, lb/hr

1. VFR measurements were performed in conjunction with each HCl test run.
2. EPA Method 26A modified by collecting a non-isokinetic sample from a single traverse point similar to EPA Method 26.

Following this introduction, Section 2 provides a summary of the test results. Section 3 provides a description of the process and sampling locations. Section 4 provides a description of the sampling and analytical procedures. Section 5 outlines the fuel processing, fuel sampling and analytical procedures used during the test program. Section 6 provides quality assurance and quality control procedures (QA/QC). Detailed test results, raw test data, boiler operating data, laboratory reports, fuel sample results, quality control records, example calculations, and a list of project participants are provided in Appendices A through H, respectively.



## **2. SUMMARY OF TEST RESULTS**

### **2.1 TEST RESULTS DISCUSSION**

Table 2-1 provides a summary of the hydrogen chloride (HCl) test results. Any differences in the test results summary tables and detailed test results shown in the appendices are due to rounding the results for presentation purposes.

As discussed in the Test Protocol (Revision 1, November 2016), WESTON conducted testing to measure the stack gas moisture, temperature, and velocity measurements concurrent with the HCl sampling at the ESP outlet duct. The measurements were used to calculate stack gas volumetric flow rates and hydrogen chloride mass rates. Detailed results tables are presented in Appendix A.

As an additional quality assurance measure, LWEC conducted fuel sampling and analysis during the test program. The chlorine results for each fuel sample collected can be found in Appendix E.

There were no sampling or operational issues that impacted the field testing and the results presented are believed to be representative of the emissions encountered during the test periods.

**Table 2-1**

**Summary of 1<sup>st</sup> Quarter 2017 HCl Test Results**

<b>Parameter</b>	<b>Date</b>	<b>Time</b>	<b>Unit of Measure</b>	<b>Result</b>	<b>PTI 168-07D Emissions Limit</b>
HCl (EPA 26A)	03/22/17	0900-1000	lb/hr	1.56	2.17
	03/22/17	1015-1115	lb/hr	1.63	2.17
	03/22/17	1130-1230	lb/hr	1.81	2.17
<b>Average</b>				<b>1.67</b>	<b>2.17</b>

### **3. DESCRIPTION OF PROCESS AND SAMPLING LOCATIONS**

#### **3.1 PROCESS OVERVIEW**

LWEC is a cogeneration facility, consisting of a single boiler generating process steam and electric power to the grid firing primarily biomass materials. The boiler typically produces steam at 180,000 lbs/hr and gross power generation from 14 to 17.7 MW/hr.

##### **3.1.1 Basic Operating Parameters**

The fuel feed to the boiler is regulated to meet process steam and electrical generation requirements. The fuel blend and excess air were modified to improve combustion characteristics. Adjustments to air, fuel blend or load were made as necessary to conform to emissions monitoring limits.

##### **3.1.2 Boiler Operations**

The hourly boiler operating limit is 324 million British thermal units (MMBtu). The maximum annual heat input is 2,656,800 MMBtu, based on 8,200 hours of operation per year.

The boiler load was determined by the demand for process steam and electricity. The boiler load was maintained at 90% of capacity during the test program.

##### **3.1.3 Test Program Fuel Mix and Firing Rates**

The fuel mix during testing consisted of wood, creosote treated railroad ties, and TDF. The firing rates for each of the fuels was within the range consistent for safe normal operations.

#### **3.2 AIR POLLUTION CONTROL EQUIPMENT**

Particulate emissions are controlled with a multi-cyclone followed by a single chamber, three-field electrostatic precipitator (ESP).

### **3.2.1 ESP Operating Parameters**

The precipitator electrical controls and rapping sequence, intensity and frequency were set for optimum performance and were not modified after optimization unless emissions issues are observed.

### **3.3 REFERENCE METHOD TEST LOCATION**

The HCl sampling and pitot traverse were conducted in the first set of sample ports (secondary sample ports) located on a section of rectangular ductwork that runs horizontally from the exit of the ESP prior to the exhaust stack. The pitot traverse was conducted at twelve traverse points in the secondary ports (three traverse points in four ports). The HCl sample train was located at a single point in the port that was located second from the top. The rectangular ductwork is six feet by six feet six inches (6' x 6½') and has a straight run of fifty-seven feet (57'). All dimensions and port locations were verified prior to testing.

Figure 3-1 presents a diagram of the sample port and traverse point location.

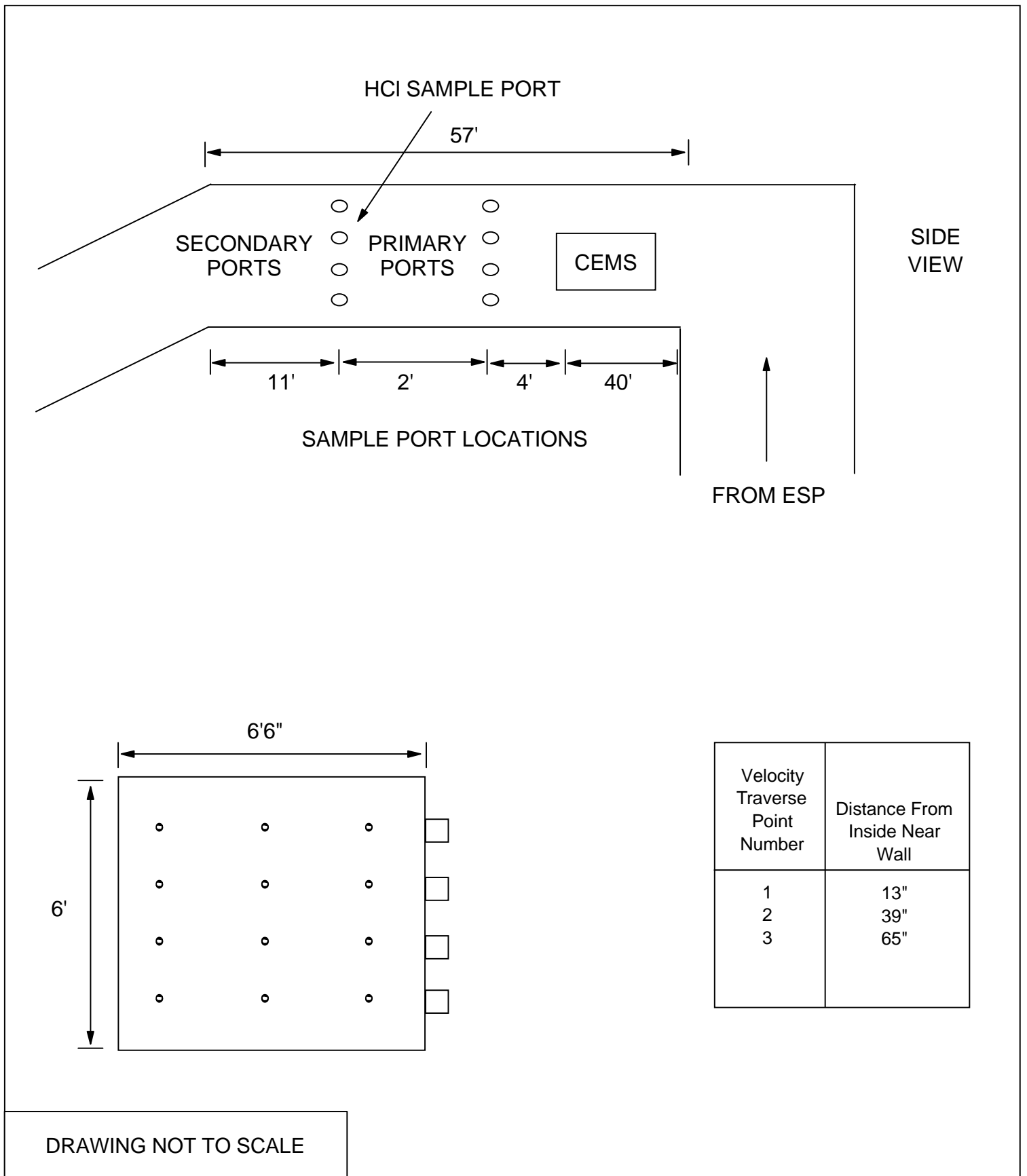
#### **3.3.1 Flue Gas Parameters**

The measured flue gas parameters at this location are as follows:

Temperature: approximately 420 - 450 °F, load dependent

Moisture: approximately 15% v/v, fuel moisture dependent

Volumetric Flow Rate: Up to about 150,000 ACFM, load dependent



**FIGURE 3-1**  
**SAMPLE PORT AND TRAVERSE POINT LOCATION**

## **4. SAMPLING AND ANALYTICAL PROCEDURES**

This section details the stack sampling and analytical procedures that were utilized during the test program. Table 4-1 summarizes the sampling and analytical methods.

### **4.1 PRE-TEST DETERMINATIONS**

Preliminary test data was obtained at the sampling location. Geometry measurements were measured and recorded, and traverse point distances verified. A preliminary velocity traverse was performed utilizing a calibrated "S" type pitot tube and a Dwyer inclined manometer to determine velocity profiles. Flue gas temperatures were observed with a calibrated direct readout pyrometer equipped with a chromel-alumel thermocouple. Water vapor content was measured by performing an EPA Method 4 moisture test, or was based on previous test data (preliminary only).

A check for the presence or absence of cyclonic flow was conducted at the test location. The average cyclonic flow check angle was  $< 20^\circ$ , thus verifying the suitability of the test site for obtaining representative samples.

Pre-test calibration of probe nozzles, pitot tubes, metering systems, and temperature measurement devices were performed as specified in Section 5 of EPA Method 5 test procedures.

### **4.2 FORMAL TESTING**

#### **4.2.1 Gas Volumetric Flow Rate**

A series of three test runs were performed. The gas velocity was measured using EPA Methods 1 and 2. Velocity measurements were performed using an "S-type" pitot tube. The stack gas pressure differential was measured with inclined manometers. Flue gas temperatures were measured with calibrated digital temperature readouts equipped with chromel-alumel (type-K) thermocouples. Velocity measurements and stack gas temperatures were performed in conjunction with the HCl sampling train and traversed across the duct diameter. The velocity and volumetric flow rate were used for determining the HCl mass rate calculations. Likewise

**Table 4-1  
Summary of Sampling and Analytical Methods**

<b>Sample</b>	<b>No. of Test Runs</b>	<b>Sampling Duration</b>	<b>Sampling Method</b>	<b>Sample Size</b>	<b>Analytical Parameters</b>	<b>Analytical Method</b>
Stack Gas	3	1-hr composite sample per run	Modified M26A	~ 40 ft <sup>3</sup>	HCl	Ion Chromatography (SW846-9057)
		Concurrent	M1-2	NA	Temperature	Temperature
					Velocity	Pitot Tube
		Integrated with M26A	M4	~ 40 ft <sup>3</sup>	Moisture	Volumetric
Concurrent	M3/3A	30 Liter Bag	O <sub>2</sub> /CO <sub>2</sub>	Continuous Emission Monitor		

moisture content was determined concurrently with each test. The moisture content of the gas stream was determined by the volume increase of the impinger water and weight increase of the silica gel in comparison to the volume of gas sampled.

The gas stream composition [oxygen (O<sub>2</sub>) and carbon dioxide content (CO<sub>2</sub>)] of the flue gas was measured according to EPA Method 3/3A procedures using a Reference Method Continuous Emission Monitoring (CEM) system. EPA protocol gas standards were used to calibrate the O<sub>2</sub>/CO<sub>2</sub> analyzer.

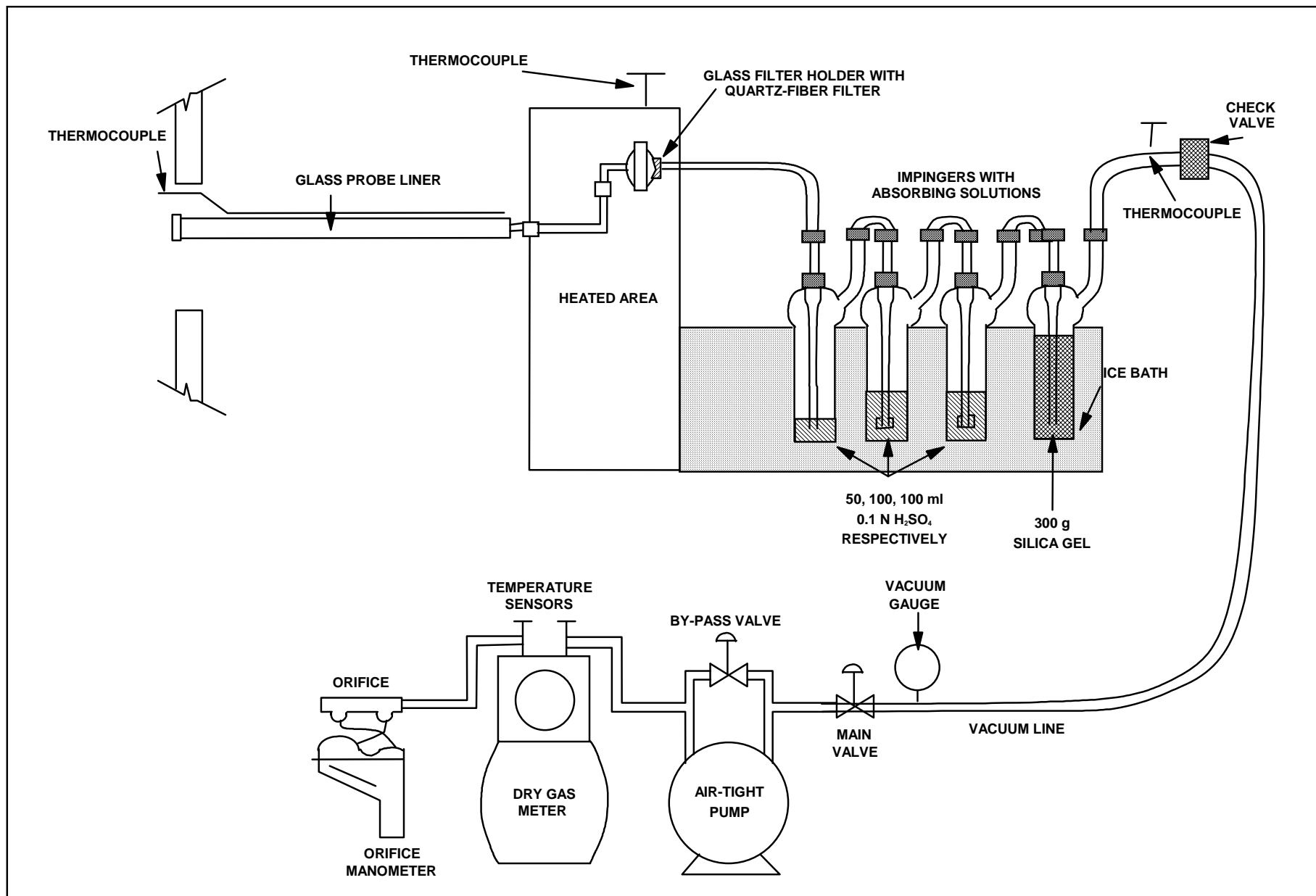
A Tedlar bag sample technique was used to determine the gas stream composition. The Tedlar bag samples of O<sub>2</sub> and CO<sub>2</sub> were collected from the exhaust of the control console calibrated orifice at a constant rate of ~0.5 liters per minute. This provides an integrated, conditioned (dry) sample. The gas passing through the control console orifice is conditioned by the impinger train. The sample is also integrated with respect to time and location in the stack.

Analysis of the Tedlar bag samples were performed using EPA Reference Method 3A analytical procedures. The conditioned Tedlar bag samples was analyzed directly by calibrated analyzers such as a paramagnetic O<sub>2</sub> analyzer and a non-dispersive Infrared (NDIR) CO<sub>2</sub> analyzer. The O<sub>2</sub> and CO<sub>2</sub> analyzers were configured and calibrated in accordance with the gas analyzer requirements outlined in EPA Reference Method 3A. The dry molecular weight of the gas stream was calculated using the measured oxygen and carbon dioxide concentrations. The balance of the gas stream was assumed to be nitrogen.

#### **4.3 EPA METHOD 26A – HYDROGEN CHLORIDE SAMPLING TRAIN**

The sampling train utilized to perform the hydrogen chloride sampling was configured as an EPA Reference Method 26A full-size sampling train with the exception of no borosilicate nozzle attached to the sample probe (see Figure 4-1). This modification was implemented to allow non-isokinetic sampling from a single traverse point similar to EPA Method 26. A heated ( $\geq 248^{\circ}\text{F}$ ) borosilicate probe was attached to a heated ( $\geq 248^{\circ}\text{F}$ ) borosilicate filter holder containing a 9-cm quartz filter. The filter folder was connected to the first of four impingers by means of rigid glass connectors. The first moisture knockout impinger contained 50 ml of 0.1 normal sulfuric acid.





**FIGURE 4-1**  
**EPA METHOD 26A (MODIFIED)**  
**HYDROGEN CHLORIDE SAMPLING TRAIN**

The second and third impingers each contained 100 ml of 0.1 N sulfuric acid. The fourth impinger contained 300 grams of dry silica gel. The second and third impingers were a standard Greenburg-Smith type and all other impingers were of a modified design. All impingers were maintained in an ice bath. A control console with a leakless vacuum pump, a calibrated dry gas meter, a calibrated orifice, and inclined manometers were connected to the final impinger via an umbilical cord to complete the train. Probe, filter box, and impinger exit gas temperatures were monitored with a calibrated direct read-out pyrometer equipped with a chromel-alumel thermocouples.

The HCl sample was collected in conjunction with independent stack gas velocities and stack gas composition ( $O_2/CO_2$  content) in order to calculate the stack gas volumetric flow rate and HCl mass rates.

#### **4.3.1 Hydrogen Chloride Sample Recovery**

At the conclusion of each test, the sampling train was dismantled, the openings sealed, and the components transported to the field laboratory.

A consistent procedure was employed for sample recovery as follows:

1. The quartz fiber filter or thimble was removed from its filter holder with tweezers and discarded.
2. The total liquid content of impingers one, two and three (0.1 N  $H_2SO_4$ ) were measured and the sample placed in a polyethylene container fitted with a Teflon-lined closure (Sample type 1). Also included in this sample was a distilled water rinse of the impingers and connectors. The sample was labeled for chloride analysis.
3. The silica gel impinger was immediately weighed to the nearest 0.5 g.
4. Samples of sulfuric acid and distilled water used for this program were retained for blank analysis.

Each sample bottle was labeled to clearly identify its contents. The height of the fluid level was marked on each bottle. The samples were then transported to the subcontract laboratory. Sample integrity was assured by maintaining chain-of-custody records.

### **4.3.2 Hydrogen Chloride Analysis**

The samples from the H<sub>2</sub>SO<sub>4</sub> impingers were analyzed for chloride (Cl<sup>-</sup>) by the procedures outlined in EPA SW-846 Method 9057 (ion chromatography) and reported as HCl. Maxxam Analytics of Mississauga, Ontario, Canada conducted the analysis. A blind audit sample developed by Environmental Resource Associates as per EPA's Stationary Source Audit Sample Program was submitted and analyzed with the stack samples.

## **5. FUEL SAMPLING AND ANALYSIS**

LWEC fuel is supplied by M.A. Energy Resources LLC (MAER). MAER operates a fuel aggregation facility where raw materials are processed then conveyed to the facility.

Fuel samples were collected during the test program during each test run in accordance with 40 CFR 63 Subpart 7521(c and d) by LWEC designated personnel from a point where each fuel drops onto the conveyor belt feeding the boiler. Three samples of each fuel type were submitted for analysis as listed in Table 5-1.

**Table 5-1  
Fuel Sample Analytical Methods**

<b>Fuel Type</b>	<b>Required Analysis</b>	<b>Analytical Methods</b>	<b>Minimum Detection Level</b>
TDF	Moisture Content	<u>ASTM D3173</u> , “ <i>Standard Test Method for Moisture in the Analysis Sample of Coal and Coke</i> ”	Not Applicable
	Chlorine Concentration	<u>EPA 5050/9056</u> , “ <i>Determination of Inorganic Anions by Ion Chromatography</i> ”	~50 ppm
Wood	Moisture Content	<u>ASTM D3173</u> , “ <i>Standard Test Method for Moisture in the Analysis Sample of Coal and Coke</i> ”	Not Applicable
	Chlorine Concentration	<u>EPA 5050/9056</u> , “ <i>Determination of Inorganic Anions by Ion Chromatography</i> ”	~50 ppm
Creosote Ties	Moisture Content	<u>ASTM D3173</u> , “ <i>Standard Test Method for Moisture in the Analysis Sample of Coal and Coke</i> ”	Not Applicable
	Chlorine Concentration	<u>EPA 5050/9056</u> , “ <i>Determination of Inorganic Anions by Ion Chromatography</i> ”	~50 ppm

## 6. QUALITY ASSURANCE/QUALITY CONTROL

### 6.1 QUALITY CONTROL PROCEDURES

As part of the HCl quarterly program, WESTON implemented a QA/QC program. QA and QC are defined as follows:

- Quality Control: The overall system of activities whose purpose is to provide a quality product or service: for example, the routine application of procedures for obtaining prescribed standards of performance in the monitoring and measurement process.
- Quality Assurance: A system of activities whose purpose is to provide assurance that the overall quality control is being done effectively. Further,

The field team manager for stack sampling was responsible for implementation of field QA/QC procedures. Individual laboratory managers were responsible for implementation of analytical QA/QC procedures. The overall project manager oversaw all QA/QC procedures to ensure that sampling and analyses met the QA/QC requirements and that accurate data resulted from the test program.

### 6.2 GAS STREAM SAMPLING QA PROCEDURES

General QA checks were conducted during testing and apply to all methods including the following:

- Performance of leak checks.
- Use of standardized forms, labels and checklists.
- Maintenance of sample traceability.
- Collection of appropriate blanks.
- Use of calibrated instrumentation.
- Review of data sheets in the field to verify completeness.
- Use of validated spreadsheets for calculation of results.

The following section details specific QA procedures applied to the test methods.

### **6.2.1 Stack Gas Velocity/Volumetric Flow Rate QA Procedures**

The QA procedures followed for velocity/volumetric flow rate determinations followed guidelines set forth by EPA Method 2. Incorporated into this method, were sample point determinations by EPA Method 1, and gas moisture content determination by EPA Method 4. QA procedures for Methods 1 and 2 are discussed below.

Volumetric flow rates were determined during the isokinetic flue gas tests. The following QC steps were followed during these tests:

- The S-type pitot tube was visually inspected before sampling.
- Both legs of the pitot tube were leak checked before sampling.
- Proper orientation of the S-type tube was maintained while making measurements. The yaw and pitch axes of the S-type pitot tube were maintained at 90° to the flow.
- The manometer oil was leveled and zeroed before each run.
- Pitot tube coefficients were determined based on physical measurement techniques as delineated in Method 2.

### **6.2.2 Moisture and Sample Gas Volume QA Procedures**

Gas stream moisture was determined as part of the HCl test trains. The following QA procedures were followed in determining the volume of moisture collected:

- Preliminary impinger train tare weights were weighed or measured volumetrically to the nearest 0.1 g or 1.0 ml.
- The balance was leveled and placed in a clean, motionless, environment for weighing.
- The indicating silica gel was fresh for each run and periodically inspected and replaced during runs if needed.
- The silica gel impinger gas temperature was maintained below 68°F.

The QA procedures that were followed in regards to accurate sample gas volume determination were:

- The dry gas meter was fully calibrated annually using an EPA approved intermediate standard device.

- Pre-test, port-change, and post-test leak-checks were completed (must be less than 0.02 cfm or 4 percent of the average sample rate).
- The gas meter was read to the thousandth of a cubic foot for all initial and final readings.
- Readings of the dry gas meter, meter orifice pressure (Delta H) and meter temperatures were taken at every sampling point.
- Accurate barometric pressures were recorded at least once per day.
- Pre- and Post-test dry gas meter checks were completed to verify the accuracy of the meter calibration constant (Y).

### **6.2.3 HCl Sampling Train QA Procedures**

The Quality Assurance procedures outlined in this section were designed to ensure collection of representative, high quality test parameter (HCl) concentrations and mass emissions data. The sampling QA procedures followed to ensure representative measurements were:

- All glassware was prepared per reference method procedures.
- Recovery procedures were completed in a clean environment.
- Sample containers for liquids and filters were constructed of borosilicate or polyethylene with Teflon®-lined lids.
- At least one reagent blank of each type of solution or filter was retained and analyzed.
- All test train components from the probe tip through the last impinger were constructed of glass (with the exception of the filter support pad which is Teflon®).
- All recovery equipment (i.e., brushes, graduated cylinders, etc.) were non-metallic.

### **6.2.4 Sample Identification and Custody**

Sample custody procedures for this program were based on EPA recommended procedures. Since samples were analyzed at remote laboratories, the custody procedures emphasized careful documentation of sample collection and field analytical data and the use of chain-of-custody records for samples being transferred. These procedures are discussed below.

The Field Team Manager was responsible for ensuring that all stack samples taken were accounted for and that all proper custody and documentation procedures were followed for the



field sampling and field analytical efforts. The Field Team Manager was assisted in this effort by key sampling personnel involved in sample recovery.

Following sample collection, all stack samples were given a unique sample identification code. Stack sample labels were completed and affixed to the sample container. The sample volumes were determined and recorded and the liquid levels on each bottle were marked. Sample bottle lids were sealed on the outside with Teflon® tape to prevent leakage. Additionally, the samples were stored in a secure area until they are shipped.

As the samples were packed for travel, chain-of-custody forms were completed for each shipment. The chain-of-custody forms specifying the treatment of each sample were also enclosed in the sample shipment container.

### **6.2.5 Data Reduction and Validation QC Checks**

All data and/or calculations for flow rates, moisture contents, and isokinetic rates, were made using a computer software program validated by an independent check. In addition, all calculations were spot checked for accuracy and completeness by the Field Team Leader.

In general, all measurement data was validated based on the following criteria:

- Process conditions during sampling or testing.
- Acceptable sample collection procedures.
- Consistency with expected or other results.
- Adherence to prescribed QC procedures.

Any suspect data was flagged and identified with respect to the nature of the problem and potential effect on the data quality.

A pre and post-test calibration were performed on the O<sub>2</sub>/CO<sub>2</sub> analyzer using a zero gas, mid gas and high gas as required by the reference methods.

The O<sub>2</sub>/CO<sub>2</sub> tedlar bag was collected at a constant rate during the HCl testing period.

All calibration gases used met EPA Protocol standards.

### **6.3 LABORATORY AUDIT SAMPLES**

A laboratory audit sample for HCl was obtained from Environmental Resource Associates (ERA) which is an accredited Stationary Source Audit Sample (SSAS) provider. The audit sample was analyzed in conjunction with the stack samples. The audit result is reported in the Maxxam analytical report along with the source emission results. Additionally, the reported audit result is compared to the assigned value in the ERA submittal included in Appendix D. The ERA laboratory report indicates passing results for the audit sample.

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**APPENDIX A**  
**DETAILED TEST RESULTS**

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**L'Anse Warden Electric Company**  
**L'Anse, MI**  
**Boiler No. 1**  
**Summary of Hydrogen Chloride Test Data and Test Results**

**TEST DATA**

	1	2	3
Test run number	1	2	3
Location		Boiler No. 1	
Test date	03/22/17	03/22/17	03/22/2017
Test time period	0900-1000	1015-1115	1130-1230

**SAMPLING DATA**

Sampling duration, min.	60	60	60
Barometric pressure, in. Hg	30.00	30.00	30.00
Avg. orifice press. diff., in H2O	1.80	1.80	1.80
Avg. dry gas meter temp., deg F	33.8	51.8	67.4
Avg. abs. dry gas meter temp., deg. R	494	512	527
Total liquid collected by train, ml	170.0	149.2	146.1
Std. vol. of H2O vapor coll., cu.ft.	8.003	7.024	6.878
Dry gas meter calibration factor	0.9999	0.9999	0.9999
Sample vol. at meter cond., dcf	40.029	41.396	41.170
Sample vol. at std. cond., dscf <sup>(1)</sup>	43.088	42.992	41.487

**GAS STREAM COMPOSITION DATA**

CO2, % by volume, dry basis	12.3	12.1	12.0
O2, % by volume, dry basis	7.8	8.0	7.9
N2, % by volume, dry basis	79.9	79.9	80.1
Molecular wt. of dry gas, lb/lb mole	30.28	30.26	30.24
H2O vapor in gas stream, prop. by vol.	0.157	0.140	0.142
Mole fraction of dry gas	0.843	0.860	0.858
Molecular wt. of wet gas, lb/lb mole	28.36	28.53	28.50

**GAS STREAM VELOCITY AND VOLUMETRIC FLOW DATA**

Static pressure, in. H2O	-12.00	-12.50	-12.00
Static pressure, in. Hg	-0.882	-0.919	-0.882
Absolute pressure, in. Hg	29.12	29.08	29.12
Avg. temperature, deg. F	444	445	448
Avg. absolute temperature, deg.R	904	905	908
Pitot tube coefficient	0.84	0.84	0.84
Duct Avg. gas stream velocity, ft./sec.	59.7	58.9	59.4
Duct cross sectional area, sq.ft.	39.00	39.00	39.00
Avg. gas stream volumetric flow, wacf/min.	139678	137764	138908
Avg. gas stream volumetric flow, dscf/min. <sup>(2)</sup>	66896	67194	67457

**HCl LABORATORY REPORT DATA**

Total HCl, mg	7.60	7.90	8.40
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**HCl EMISSIONS**

				Average	Limit
Concentration, lb/dscf	3.89E-07	4.05E-07	4.46E-07	4.13E-07	---
Concentration, ppm/v	4.11	4.28	4.72	4.37	---
Mass rate, lb/hr	1.56	1.63	1.81	1.67	2.17

(1) Standard conditions = 68 deg. F. (20 deg. C.) and 29.92 in Hg (760 mm Hg)

(2) Volumetric flow rate from EPA Method 2 velocity measurements.

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**APPENDIX B  
RAW TEST DATA**

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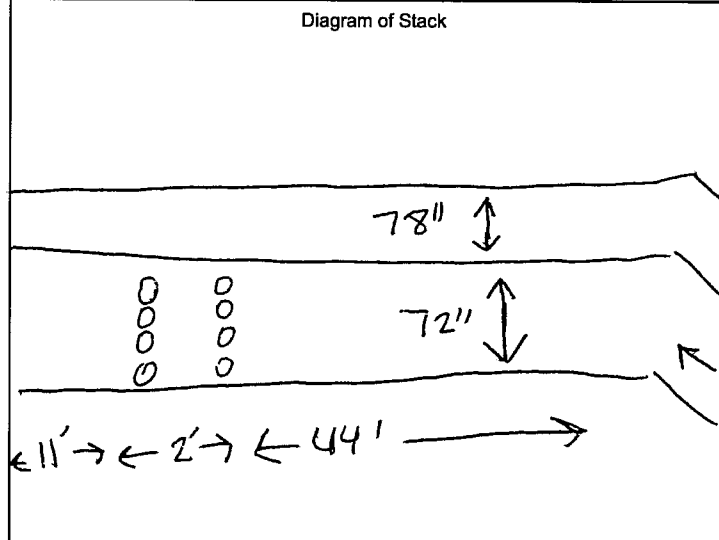
Client LWEC  
 Location/Plant L'Anse, MI  
 Source ESP Outlet

Operator Schwartz  
 Date 3/21/17  
 W.O. Number 14464.007.006.0001

**Duct Type**  Circular  Rectangular Duct Indicate appropriate type  
**Traverse Type**  Particulate Traverse  Velocity Traverse  CEM Traverse

Distance from far wall to outside of port (in.) = C	90
Port Depth (in.) = D	12
Depth of Duct, diameter (in.) = C-D	78
Area of Duct (ft <sup>2</sup> )	39
Total Traverse Points	12
Total Traverse Points per Port	3
Port Diameter (in.) —(Flange-Threaded-Hole)	
Monorail Length	
<b>Rectangular Ducts Only</b>	
Width of Duct, rectangular duct only (in.)	72
Total Ports (rectangular duct only)	4
Equivalent Diameter = (2*L*W)/(L+W)	74.89

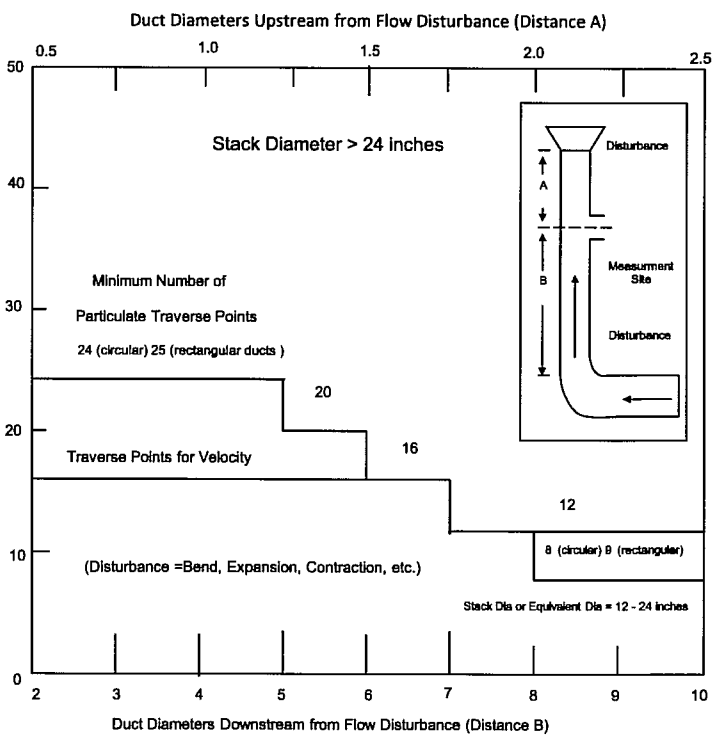
Flow Disturbances	
Upstream - A (ft)	11.0
Downstream - B (ft)	44.0
Upstream - A (duct diameters)	1.8
Downstream - B (duct diameters)	7.1



Traverse Point Locations			
Traverse Point	% of Duct	Distance from Inside Duct Wall (in)	Distance from Outside of Port (in)
1	0.167	13.026	25.026
2	0.500	39	51
3	0.833	64.974	76.974
4			
5			
6			
7			
8			
9			
10			
11			
12			

CEM 3 Point(Long Measurement Line) Stratification Point Locations		
	% of Duct	
1	0.167	
2	0.50	
3	0.833	



Note: If stack dia < 12 inch use EPA Method 1A (Sample port upstream of pitot port)  
 Note: If stack dia > 24" then adjust traverse point to 1 inch from wall  
 If stack dia < 24" then adjust traverse point to 0.5 inch from wall

Traverse Point Location Percent of Stack -Circular													
		Number of Traverse Points											
		1	2	3	4	5	6	7	8	9	10	11	12
T r a v e r s e P o i n t	1		14.6		6.7		4.4		3.2		2.6		2.1
	2		85.4		25		14.6		10.5		8.2		6.7
	3			75		29.6		19.4		14.6		11.8	
	4				93.3		70.4		32.3		22.6		17.7
	5					85.4		67.7		34.2		25	
	6						95.6		80.6		65.8		35.6
	7							89.5		77.4		64.4	
	8								96.8		85.4		75
	9									91.8		82.3	
	10										97.4		88.2
	11											93.3	
	12												97.9

Traverse Point Location Percent of Stack -Rectangular													
		Number of Traverse Points											
		1	2	3	4	5	6	7	8	9	10	11	12
T r a v e r s e P o i n t	1		25.0	16.7	12.5	10.0	8.3	7.1	6.3	5.6	5.0	4.5	4.2
	2		75.0	50.0	37.5	30.0	25.0	21.4	18.8	16.7	15.0	13.6	12.5
	3			83.3	62.5	50.0	41.7	35.7	31.3	27.8	25.0	22.7	20.8
	4				87.5	70.0	58.3	50.0	43.8	38.9	35.0	31.8	29.2
	5					90.0	75.0	64.3	56.3	50.0	45.0	40.9	37.5
	6						91.7	78.6	68.8	61.1	55.0	50.0	45.8
	7							92.9	81.3	72.2	65.0	59.1	54.2
	8								93.8	83.3	75.0	68.2	62.5
	9									94.4	85.0	77.3	70.8
	10										95.0	86.4	79.2
	11											95.5	87.5
	12												95.8

# Determination of Stack Gas Velocity - Method 2

Client LWEC  
 Location/Plant L'Anse, MI  
 Source Boiler 2

Operator KS/TB  
 Date 3/22/17  
 W.O. Number 14464.007

Pitot Coeff (Cp) .84  
 Stack Area, ft<sup>2</sup> (As) 39  
 Pitot Tube/Thermo ID P-676

Run Number	1	2	3
Time	019-928	1017-1026	1136-1144
Barometric Press, in Hg (Pb)	30.00	30.00	30.00
Static Press, in H <sub>2</sub> O (Pstatic)	-12-0	-12-5	-12-0
Source Moisture, % (BWS)	15.7	14.0	14.2
O <sub>2</sub> , %	7.8	8	7.9
CO <sub>2</sub> , %	12.3	12.1	12.0

Cyclonic Flow Determination	Angle yielding zero Delta P	Leak Check good ?		Leak Check good ?		Leak Check good ?			
		Port	Point	Delta P	Source Temp, F° (Ts)	Delta P	Source Temp, F° (Ts)	Delta P	Source Temp, F° (Ts)
A		A	1	0.84	442	0.83	445	0.83	448
			2	0.85	440	0.83	440	0.85	446
			3	0.82	436	0.80	436	0.82	443
		B	1	0.72	446	0.72	442	0.74	447
			2	0.69	448	0.71	446	0.71	452
			3	0.53	437	0.62	441	0.55	446
		C	1	0.59	449	0.55	447	0.59	444
			2	0.61	452	0.57	450	0.59	449
			3	0.60	450	0.54	450	0.58	451
D	1	0.47	440	0.45	444	0.45	442		
	2	0.48	443	0.44	447	0.43	447		
	3	0.46	446	0.44	446	0.46	452		
Avg Angle		Avg Delta P & Temp		0.63833	444.08	0.62500	444.5	0.63583	447.5
		avg √Delta P		0.79430		0.78533		0.79047	
		Average gas stream velocity, ft/sec.		59.69147		58.87340		59.36878	
		Vol. flow rate @ actual conditions, wacf/min		139678		137764		138909	
		Vol. flow rate at standard conditions, dscf/min		66897		67194		67457	

$$MWd = (0.32 \cdot O_2) + (0.44 \cdot CO_2) + (0.28 \cdot (100 - (CO_2 + O_2)))$$

$$MWs = (MWd \cdot (1 - (BWS/100))) + (18 \cdot (BWS/100))$$

$$Tsa = Ts + 460$$

$$Ps = Pb + (Pstatic/13.6)$$

$$Vs = 85.49 \cdot Cp \cdot \text{avg} \sqrt{\Delta P} \cdot \sqrt{Tsa / (Ps \cdot MWs)}$$

$$Qs(\text{act}) = 60 \cdot Vs \cdot As$$

$$Qs(\text{std}) = 17.64 \cdot (1 - (BWS/100)) \cdot (Ps/Tsa) \cdot Qs(\text{act})$$

MWd = Dry molecular weight source gas, lb/lb-mole.

MWs = Wet molecular weight source gas, lb/lb-mole.

Tsa = Source Temperature, absolute (°R)

Ps = Absolute stack static pressure, inches Hg.

Vs = Average gas stream velocity, ft/sec.

Qs(act) = Volumetric flow rate of wet stack gas at actual, wacf/min

Qs(std) = Volumetric flow rate of dry stack gas at standard conditions, dscf/min

Note: Micromanometer is required if:

(A) The average Delta P readings are less than 0.05 inches of water.

(B) For traverses of 12 or more points, more than 10% of the Delta P readings are below 0.05 inches of water.

(C) For traverses of less than 12 points, more than one Delta P readings is below 0.05 inches of water.







**L'Anse Warden Electric Company  
Inputs for Hydrogen Chloride Calculations**

**Test Data**

	1	2	3
Run number			
Location		Boiler No. 1	
Date	03/22/17	03/22/17	03/22/2017
Time period	0900-1000	1015-1115	1130-1230
Operator	KS	KS	KS

**Inputs For Calcs.**

Delta H	1.8000	1.8000	1.8000
Stack temp. (deg.F)	444.1	444.5	447.5
Meter temp. (deg.F)	33.8	51.8	67.4
Sample volume (act.)	40.029	41.396	41.170
Barometric press. (in.Hg)	30.00	30.00	30.00
Volume H2O imp. (ml)	160.0	136.0	136.0
Weight change sil. gel (g)	10.0	13.2	10.1
% CO2	12.3	12.1	12.0
% O2	7.8	8.0	7.9
% N	79.9	79.9	80.1
Area of stack (sq.ft.)	39.000	39.000	39.000
Sample time (min.)	60	60	60
Static pressure (in.H2O)	-12.00	-12.50	-12.00
Meter box cal.	0.9999	0.9999	0.9999
Cp of pitot tube	0.84	0.84	0.84
Traverse Points	12	12	12

**HCl Laboratory Report Data**

HCl, mg	7.60	7.90	8.40
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# ISOKINETIC FIELD DATA SHEET

# PADEP EPA Method 26A - HCl

Client: L'Anse Warden  
 W.O.#: 14464.007.005.0001  
 Project ID: LWEC  
 Mode/Source ID: Boiler  
 Samp. Loc. ID: ESP-OUT  
 Run No. ID: 1  
 Test Method ID: M26A  
 Date ID: 22MAR2017  
 Source/Location: Boiler Stack  
 Sample Date: 3/22/17  
 Baro. Press (in Hg): 30.00  
 Operator: KS

**Stack Conditions**

Assumed	Actual
<u>14</u>	
	<u>160</u>
	<u>10</u>
<u>11</u>	<u>12.3</u>
<u>7</u>	<u>7.6</u>
<u>36</u>	
<u>-12</u>	<u>-12</u>
	<u>15°F</u>

Meter Box ID: 13  
 Meter Box Y: 0.9999  
 Meter Box Del H: 6.8715  
 Probe ID / Length: 6'  
 Probe Material: Boro  
 Pitot / Thermocouple ID: 0.84  
 Pitot Coefficient: —  
 Nozzle ID: —  
 Avg Nozzle Dia (in): —  
 Area of Stack (ft²): 39  
 Sample Time: 60min  
 Total Traverse Pts: 1

**Leak Checks**

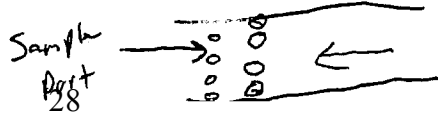
Sample Train (ft³): 0.006  
 Leak Check @ (in Hg): 15  
 Pitot good: N/A  
 Orsat good: —  
**Temp Check**  
 Meter Box Temp: 13  
 Reference Temp: 13.7  
 Pass/Fail (+/- 2°): Pass / Fail  
 Temp Change Response: yes / no

K Factor	<u>N/A</u>		
	Initial	Mid-Point	Final
	<u>0.006</u>		<u>0.004</u>
	<u>15</u>		<u>10</u>
	yes / no	yes / no	yes / no
	<u>yes</u> / no	yes / no	<u>yes</u> / no
	Pre-Test Set		Post-Test Set
	<u>13</u>		<u>16</u>
	<u>13.7</u>		<u>16.2</u>
	Pass / Fail		Pass / Fail
	<u>yes</u> / no		<u>yes</u> / no

TRAVERSE POINT NO.	SAMPLE TIME (min)	CLOCK TIME (plant time)	VELOCITY PRESSURE Delta P (in H2O)	ORIFICE PRESSURE Delta H (in H2O)	DRY GAS METER READING (ft³)	STACK TEMP (°F)	DGM INLET TEMP (°F)	DGM OUTLET TEMP (°F)	PROBE TEMP (°F)	FILTER BOX TEMP (F)	IMPINGER EXIT TEMP (°F)	SAMPLE TRAIN VAC (in Hg)	COMMENTS
	<u>0</u>	<u>0900</u>			<u>688.260</u>								
<u>131</u>	<u>5</u>			<u>1.8</u>	<u>692.0</u>		<u>N/A</u>	<u>30</u>	<u>261</u>	<u>263</u>	<u>34</u>	<u>3.0</u>	
	<u>10</u>			<u>1.8</u>	<u>695.1</u>			<u>30</u>	<u>263</u>	<u>263</u>	<u>34</u>	<u>3.0</u>	
	<u>15</u>			<u>1.8</u>	<u>698.5</u>			<u>31</u>	<u>262</u>	<u>266</u>	<u>35</u>	<u>3.0</u>	
	<u>20</u>			<u>1.8</u>	<u>701.6</u>			<u>31</u>	<u>266</u>	<u>268</u>	<u>34</u>	<u>3.0</u>	
	<u>25</u>			<u>1.8</u>	<u>704.8</u>			<u>32</u>	<u>265</u>	<u>267</u>	<u>34</u>	<u>3.0</u>	
	<u>30</u>			<u>1.8</u>	<u>708.3</u>			<u>33</u>	<u>265</u>	<u>266</u>	<u>34</u>	<u>3.0</u>	
	<u>35</u>			<u>1.8</u>	<u>711.6</u>			<u>34</u>	<u>264</u>	<u>266</u>	<u>35</u>	<u>3.0</u>	
	<u>40</u>			<u>1.8</u>	<u>715.0</u>			<u>35</u>	<u>261</u>	<u>262</u>	<u>34</u>	<u>3.0</u>	
	<u>45</u>			<u>1.8</u>	<u>718.3</u>			<u>36</u>	<u>264</u>	<u>267</u>	<u>34</u>	<u>3.0</u>	
	<u>50</u>			<u>1.8</u>	<u>721.6</u>			<u>37</u>	<u>264</u>	<u>265</u>	<u>35</u>	<u>3.0</u>	
	<u>55</u>			<u>1.8</u>	<u>725.0</u>			<u>37</u>	<u>261</u>	<u>263</u>	<u>34</u>	<u>3.0</u>	
	<u>60</u>	<u>1000</u>		<u>1.8</u>	<u>728.289</u>			<u>39</u>	<u>260</u>	<u>264</u>	<u>35</u>	<u>3.0</u>	

Avg Sqrt Delta P: 1.8    Avg Delta H: 1.8    Total Volume: 40.029    Avg Ts: N/A    Avg Tm: 33.75    Min/Max: 260/268    Min/Max: 262/268    Max Temp: 35    Max Vac: 3.0    Max Temp: 35

Avg Sqrt Del H: 1.3464    Comments: Sample Port 28



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# ISOKINETIC FIELD DATA SHEET

# PADEP EPA Method 26A - HCl

Client	L'Anse Warden	Stack Conditions	Meter Box ID	13	K Factor	N/A
W.O.#	14464.007.005.0001	Assumed	Meter Box Y	0.9994	Initial	0.008
Project ID	LWEC	Actual	Meter Box Del H	1.8915	Mid-Point	
Mode/Source ID	Boiler	% Moisture	Probe ID / Length		Final	0.008
Samp. Loc. ID	ESP-OUT	Impinger Vol (ml)	Probe Material	Boro	Leak Checks	
Run No.ID	2	Silica gel (g)	Pitot / Thermocouple ID		Sample Train (ft³)	15
Test Method ID	M26A	CO2, % by Vol	Pitot Coefficient	0.84	Leak Check @ (in Hg)	10
Date ID	22MAR2017	O2, % by Vol	Nozzle ID		Pitot good	yes / no
Source/Location	Boiler Stack	Temperature (°F)	Avg Nozzle Dia (in)		Orsat good	yes / no
Sample Date	3/22/17	Meter Temp (°F)	Area of Stack (ft²)	39	Temp Check	Pre-Test Set
Baro. Press (in Hg)	30.00	Static Press (in H2O)	Sample Time	60	Meter Box Temp	22
Operator	KS	Ambient Temp (°F)	Total Traverse Pts	1	Reference Temp	22.8
					Pass/Fail (+/- 2°)	Pass / Fail
					Temp Change Response	yes / no

TRAVERSE POINT NO.	SAMPLE TIME (min)	CLOCK TIME (plant time)	VELOCITY PRESSURE Delta P (in H2O)	ORIFICE PRESSURE Delta H (in H2O)	DRY GAS METER READING (ft³)	STACK TEMP (°F)	DGM INLET TEMP (°F)	DGM OUTLET TEMP (°F)	PROBE TEMP (°F)	FILTER BOX TEMP (F)	IMPING EXIT TEMP (°F)	SAMPLE TRAIN VAC (in Hg)	COMMENTS
	0	1015			729.049								
B1	5			1.8	732.4		N/A	44	261	263	39	3.0	
	10			1.8	735.2			45	260	265	38	3.0	
	15			1.8	739.2			46	262	266	38	3.0	
	20			1.8	742.5			48	261	264	40	3.0	
	25			1.8	745.2			49	263	264	41	3.0	
	30			1.8	749.2			52	266	268	43	3.0	
	35			1.8	752.6			53	267	269	43	3.0	
	40			1.8	756.1			55	265	266	44	3.0	
	45			1.8	759.6			56	266	268	46	3.0	
	50			1.8	763.1			57	264	268	47	3.0	
	55			1.8	766.6			58	266	269	46	3.0	
	60	1115		1.8	770.445			58	266	268	46	3.0	
			Avg Sqrt Delta P	Avg Delta H	Total Volume	Avg Ts		Avg Tm	Min/Max	Min/Max	Max Temp	Max Vac	Max Temp
				1.8	41.396			51.18 TB	260/267	263/269	47	3.0	
			Avg Sqrt Del H	Comments:				51.8 ✓	EPA 26A from 40CFR Part 60 App A				
			1.34164										



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# ISOKINETIC FIELD DATA SHEET

# PADEP EPA Method 26A - HCl

Client	L'Anse Warden	<b>Stack Conditions</b>	
W.O.#	14464.007.005.0001	Assumed	Actual
Project ID	LWEC	% Moisture	14
Mode/Source ID	Boiler	Impinger Vol (ml)	136
Samp. Loc. ID	ESP-OUT	Silica gel (g)	10.1
Run No.ID	3	CO2, % by Vol	11
Test Method ID	M26A	O2, % by Vol	7
Date ID	22MAR2017	Temperature (°F)	65
Source/Location	Boiler Stack	Meter Temp (°F)	-12
Sample Date	3/22/17	Static Press (in H <sub>2</sub> O)	-12
Baro. Press (in Hg)	30.00	Ambient Temp (°F)	25.9
Operator	KS		

Meter Box ID	13
Meter Box Y	0.9999
Meter Box Del H	1.8915
Probe ID / Length	Boro
Probe Material	
Pitot / Thermocouple ID	
Pitot Coefficient	0.84
Nozzle ID	-
Avg Nozzle Dia (in)	-
Area of Stack (ft <sup>2</sup> )	39
Sample Time	6:00
Total Traverse Pts	1

<b>Leak Checks</b>
Sample Train (ft <sup>3</sup> )
Leak Check @ (in Hg)
Pitot good
Orsat good
<b>Temp Check</b>
Meter Box Temp
Reference Temp
Pass/Fail (+/- 2°)
Temp Change Response

K Factor	N/A		
	Initial	Mid-Point	Final
	0.008		0.004
	15		10
	yes / no	yes / no	yes / no
	yes / no	yes / no	yes / no
	Pre-Test Set		Post-Test Set
	25		27
	26.5		27.4
	(Pass / Fail)		(Pass / Fail)
	yes / no		yes / no

TRAVERSE POINT NO.	SAMPLE TIME (min)	CLOCK TIME (plant time)	VELOCITY PRESSURE Delta P (in H <sub>2</sub> O)	ORIFICE PRESSURE Delta H (in H <sub>2</sub> O)	DRY GAS METER READING (ft <sup>3</sup> )	STACK TEMP (°F)	DGM INLET TEMP (°F)	DGM OUTLET TEMP (°F)	PROBE TEMP (°F)	FILTER BOX TEMP (F)	IMPINGER EXIT TEMP (°F)	SAMPLE TRAIN VAC (in Hg)	COMMENTS
	0	11:30			770.835								
B1	5			1.8	774.4		N/A	64	265	265	53	3.0	
	10			1.8	778.0			65	266	264	50	3.0	
	15			1.8	781.2			66	263	267	48	3.0	
	20			1.8	784.7			66	264	267	47	3.0	
	25			1.8	788.1			66	267	264	47	3.0	
	30			1.8	791.5			67	269	268	46	3.0	
	35			1.8	795.0			68	267	264	47	3.0	
	40			1.8	798.3			69	266	265	47	3.0	
	45			1.8	801.8			69	264	267	48	3.0	
	50			1.8	805.2			69	263	269	48	3.0	
	55			1.8	808.6			70	265	269	48	3.0	
	60	12:30		1.8	812.005		N/A	70	265	268	49	3.0	

Avg Sqrt Delta P	Avg Delta H	Total Volume	Avg Ts	Avg Tm	Min/Max	Min/Max	Max Temp	Max Vac	Max Temp
	1.8	41.170		67.42	263/269	264/269	53	3.0	
	Avg Sqrt Del H	Comments:							
	1.34164								

EPA 26A from 40CFR Part 60 App A



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# SAMPLE RECOVERY FIELD DATA

PADEP EPA Method 26A - HCl

Client L'Anse Warden W.O. # 14464.007.005.0001  
 Location/Plant L'Anse MI Source & Location Boiler Stack

Run No. 1 Sample Date 3/22/17 Recovery Date 3/22/17  
 Sample I.D. LWEC - Boiler - ESP-OUT - 1 - M26A - Analyst TB Filter Number N/A

	Impinger							Imp.Total	8	Total
	1	2	3	4	5	6	7			
Contents	H2SO4	H2SO4	H2SO4	Empty					Silica Gel	
Final	158	170	82					410	310.0	
Initial	50	100	100					250	300	
Gain	108	70	-18					160	10.0	

Impinger Color Clear Labeled?  540 Tot Vol  
 Silica Gel Condition 3/4 Blue Sealed?  N/A

Run No. 2 Sample Date 3/22/17 Recovery Date 3/22/17  
 Sample I.D. LWEC - Boiler - ESP-OUT - 2 - M26A - Analyst TB Filter Number N/A

	Impinger							Imp.Total	8	Total
	1	2	3	4	5	6	7			
Contents	H2SO4	H2SO4	H2SO4	Empty					Silica Gel	
Final	148	146	92					386	313.2	
Initial	50	100	100					250	300	
Gain	98	46	-8					136	13.2	

Impinger Color Clear Labeled?  526 Tot Vol  
 Silica Gel Condition mostly blue Sealed?  N/A

Run No. 3 Sample Date 3/22/17 Recovery Date 3/22/17  
 Sample I.D. LWEC - Boiler - ESP-OUT - 3 - M26A - Analyst TB Filter Number N/A

	Impinger							Imp.Total	8	Total
	1	2	3	4	5	6	7			
Contents	H2SO4	H2SO4	H2SO4	Empty					Silica Gel	
Final	160	122	104					386	310.1	
Initial	50	100	100					250	300	
Gain	110	22	4					136	10.1	

Impinger Color Clear Labeled?  532 Tot Vol  
 Silica Gel Condition 3/4 Blue Sealed?  N/A

Check COC for Sample IDs of Media Blanks



*Handwritten signature*

# Source Gas Analysis Data Sheet - Modified Method 3/3A

Client LWEC Analyst TB  
 Location/Plant L'ANSE, MI Date 3/22/17  
 Source Boiler 2 Analyzer Make & Model Servomex 1400  
 W.O. Number 14464.007.006.0001

Calibration 1

Analysis Number	Span	Calibration Gas Value O <sub>2</sub> (%)	Calibration Gas Value CO <sub>2</sub> (%)	Analyzer Response O <sub>2</sub> (%)	Analyzer Response CO <sub>2</sub> (%)
1	Zero	0	0	0	0
2	Mid	11.98	8.908	12.1	9.0
3	High	21.65	16.63	21.7	16.6
Average					

Run Number	Analysis Time	Analyzer Response O <sub>2</sub> (%)	Analyzer Response CO <sub>2</sub> (%)
1	1030	7.8	12.3
2	1145	8.0	12.1
3	1240	7.9	12.0
Average			

Run Number	Analysis Time	Analyzer Response O <sub>2</sub> (%)	Analyzer Response CO <sub>2</sub> (%)
Average			

Span	Cylinder ID
Mid	CC 61928
High	CC 452229



\*\*Report all values to the nearest 0.1 percent

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**APPENDIX C**  
**BOILER OPERATING DATA**

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**Fuel Feed Rates**  
**3-22-17 Compliance Testing**  
**L'Anse Warden Electric Company, LLC**

	<b>Bin #1</b> RR Ties (% Rake Speed)	<b>Bin #2</b> Woodchips (% Rake Speed)	<b>Bin #3</b> RR Ties (% Rake Speed)	<b>TDF Bin</b> Totalizer (Tons)	<b>Main Fuel Belt</b> Totalizer (Tons)	<b>Time</b>
	16	19	16	11.07	189.43	8:15
	16	19	16	11.45	194.72	8:30
	17	20	17	11.83	200.25	8:45
	16	19	16	12.19	207.51	9:00
	14	17	14	12.61	214.91	9:15
	14	17	14	12.98	221.83	9:30
	14	17	14	13.34	227.05	9:45
	15	18	15	13.72	232.36	10:00
	14	17	14	14.11	239.43	10:15
	14	17	14	14.49	246.49	10:30
	14	17	14	14.89	253.52	10:45
	14	17	14	15.26	260.33	11:00
	13	16	13	15.66	265.99	11:15
	12	15	12	16.04	272.73	11:30
	12	15	12	16.43	279.05	11:45
	13	16	13	16.79	284.3	12:00
	13	16	13	17.17	290.28	12:15
	13	16	13	17.57	295.88	12:30
	13	16	13	17.95	301.95	12:45
Average Rake Speed (%) =	14.1	17.1	14.1			
Fraction of Total =	0.31	0.38	0.31			
Total TDF (wet tons) =				6.88		
Total Fuel (wet tons) =					112.52	
Elapsed Time (hours) =						4.5
Total Fuel - Total TDF (wet tons) =	105.6					
TDF Feed Rate (tons/hour) =	1.53					
Approximate RR Ties to Woodchips Ratio =	1.65					
Approximate RR Ties Tons/Hr =	14.6					
Approximate Woodchips Tons/Hr =	8.9					
Average Laboratory RR Tie Moisture =	25.53%					
Average Laboratory Woodchips Moisture =	48.28%					

Note: Assumes rake speeds directly correlate with weight being dropped on the belt.



PROCESS OPERATING DATA LOG SHEET FOR EMISSIONS TESTING

POWER BOILERS--RECOMENDED EVERY 15 MINUTES; START 15 MIN B4 TEST, AND CONTINUE 15 MIN AFTER TEST

Oper. Initials	Start Time	Stop Time	FUEL STORAGE BUILDING						TDF BIN		Main Belt	Oxygen Levels		Production Rates		Fan	Air		
			Bin #1 Fuel In place	Rake Speed	TPH	Bin #2 Fuel In place	Rake Speed	TPH	Bin #3 Fuel In place	Rake Speed		TPH	Screw Speed	Totalizer	TPH Rate (If TDF)			Main Fuel belt	BOILER O2
Plant	DATE	TIME (use military time)	R/R Ties Totalizer	SPECIFY UNITS %	Wood Chips Totalizer	SPECIFY UNITS %	R/R Ties Totalizer	SPECIFY UNITS %	SPECIFY UNITS %			Totalizer	% (specify wet or dry)	% (specify wet or dry)	KPPH	CrSS	%	%	
LWEC	3/22/2017	800																	
LWEC	3/22/2017	815	1	16		19	16	17	11.07		189.43	12.5		161	16.58	85.5	100		
LWEC	3/22/2017	830		16		19	16	17	11.45		194.72	12.3	4.3	179	16.60	84.9	97.4		
LWEC	3/22/2017	845		16		19	16	17	11.83		200.25	12.8	8.5	181	16.54	96	97.9		
LWEC	3/22/2017	900		16		19	16	17	12.19		207.51	12.3	7.7	181	16.62	87.8	100		
LWEC	3/22/2017	915		14		17	14	17	12.61		214.91	12.1	6.5	179	16.56	86.1	93.1		
LWEC	3/22/2017	930		14		17	14	17	12.98		221.83	12.4	6.6	179	16.58	83.6	96.6		
LWEC	3/22/2017	945		14		17	14	17	13.34		227.05	12.4	7.4	180	16.54	85.8	94.0		
LWEC	3/22/2017	1000		15		18	15	17	13.72		232.38	12.4	6.6	178	16.54	83.6	94.0		
LWEC	3/22/2017	1015		14		17	14	17	14.11		239.43	12.4	6.6	181	16.61	83.2	95.3		
LWEC	3/22/2017	1030		14		17	14	17	14.49		246.49	12.3	5.5	181	16.63	85.6	94.1		
LWEC	3/22/2017	1045		14		17	14	17	14.87		253.52	12.6	5.1	175	16.59	86.7	85.2	84.7	
LWEC	3/22/2017	1100		13		16	13	17	15.26		260.33	12.9	7.7	178	16.68	85.8	95.7	84.7	
LWEC	3/22/2017	1115		13		16	13	17	15.64		267.99	12.5	7.3	177	16.58	85.1	94.9		
LWEC	3/22/2017	1130		13		16	13	17	16.02		272.73	12.4	8.3	178	16.56	83.2	95.3		
LWEC	3/22/2017	1145		13		16	13	17	16.40		279.05	12.3	8.0	178	16.64	83.4	97.2		
LWEC	3/22/2017	1200		13		16	13	17	16.79		284.30	12.2	6.6	178	16.62	82.9	100		
LWEC	3/22/2017	1215		13		16	13	17	17.17		290.24	12.6	6.6	178	16.60	84.6	94		
LWEC	3/22/2017	1230		13		16	13	17	17.57		295.84	12.7	6.1	179	16.55	84.0	94.1		
LWEC	3/22/2017	1245		13		16	13	17	17.95		301.95	12.6	6.1	176	16.62	84.0	97.6		
LWEC	3/22/2017	1300																	
LWEC	3/22/2017	1315																	
LWEC	3/22/2017	1330																	
LWEC	3/22/2017	1345																	
LWEC	3/22/2017	1400																	
LWEC	3/22/2017	1415																	
LWEC	3/22/2017	1430																	
LWEC	3/22/2017	1445																	
LWEC	3/22/2017	1500																	
LWEC	3/22/2017	1515																	
LWEC	3/22/2017	1530																	
LWEC	3/22/2017	1545																	
LWEC	3/22/2017	1600																	
LWEC	3/22/2017	1615																	
LWEC	3/22/2017	1630																	
LWEC	3/22/2017	1645																	
LWEC	3/22/2017	1700																	
LWEC	3/22/2017	1715																	
LWEC	3/22/2017	1730																	
LWEC	3/22/2017	1745																	
LWEC	3/22/2017	1800																	
LWEC	3/22/2017	1815																	
LWEC	3/22/2017	1830																	
LWEC	3/22/2017	1845																	
LWEC	3/22/2017	1900																	
LWEC	3/22/2017	1915																	
LWEC	3/22/2017	1930																	
LWEC	3/22/2017	1945																	
LWEC	3/22/2017	2000																	

PROCESS OPERATING DATA LOG SHEET

POWER BOILERS-RECOMMENDED EMISSIONS

*Handwritten notes:*  
 HAD 2/29  
 Compliance Test  
 Major Permitting  
 EP License

Oper. Initials	Start Time	Stop Time	Stack Monitors			ESP POWER DATA				ESP POWER DATA				ESP POWER DATA				TIME
			OPACITY	CO	Flue Gas Temp	Primary	ESP Field 1 (Inlet)	ESP Field 1 (Inlet)	Spark Rate	Primary	ESP Field 2 (Center)	ESP Field 2 (Center)	Spark Rate	Primary	ESP Field 3 (Outlet)	ESP Field 3 (Outlet)	Spark Rate	
Plant	DATE	TIME (use military time)	%	lb/mmbtu	(Precip Gas Out)	Voltage	Secondary kV	Secondary mA	SPM	Voltage	Secondary kV	Secondary mA	SPM	Voltage	Secondary kV	Secondary mA	SPM	Use military time
LWEC	3/22/2017	800																800
LWEC	3/22/2017	815	2.6	112	432	252	34	644	1	220	29	644	1	208	27	540	2	815
LWEC	3/22/2017	830	2.6	94	432	254	34	648	1	226	26	644	1	188	27	496	2	830
LWEC	3/22/2017	845	2.6	94	431	252	34	628	1	221	29	650	1	184	27	396	13	845
LWEC	3/22/2017	900	2.6	76	431	254	32	648	1	226	30	644	1	210	27	536	1	900
LWEC	3/22/2017	915	2.6	181	431	254	31	630	1	224	30	648	1	210	27	526	1	915
LWEC	3/22/2017	930	2.6	148	432	262	36	650	1	224	30	622	1	202	28	520	5	930
LWEC	3/22/2017	945	2.6	134	432	238	31	650	13	224	30	648	1	208	28	526	4	945
LWEC	3/22/2017	1000	2.6	14	434	246	33	650	1	226	29	650	1	208	28	520	9	1000
LWEC	3/22/2017	1015	2.6	83	432	252	34	648	1	220	29	650	1	188	28	530	10	1015
LWEC	3/22/2017	1030	2.7	127	430	262	35	626	1	226	30	644	1	216	29	510	1	1030
LWEC	3/22/2017	1045	2.7	84	432	248	33	650	1	226	31	640	1	212	29	510	0	1045
LWEC	3/22/2017	1100	2.7	133	434	230	32	550	1	226	30	650	1	196	27	508	8	1100
LWEC	3/22/2017	1115	2.7	103	432	260	38	560	1	234	31	650	1	210	29	546	3	1115
LWEC	3/22/2017	1130	2.7	104	435	244	33	514	12	234	30	640	1	188	27	540	3	1130
LWEC	3/22/2017	1145	2.7	117	434	262	35	644	1	234	30	648	1	204	28	484	12	1145
LWEC	3/22/2017	1200	2.7	151	436	260	35	648	1	226	30	648	1	190	26	386	13	1200
LWEC	3/22/2017	1215	2.7	93	437	260	35	650	1	230	30	650	1	190	28	480	14	1215
LWEC	3/22/2017	1230	2.7	119	435	236	32	650	1	230	30	644	1	204	28	340	4	1230
LWEC	3/22/2017	1245	2.7	107	438	260	35	554	5	234	31	634	1	212	29	532	4	1245
LWEC	3/22/2017	1300																1300
LWEC	3/22/2017	1315																1315
LWEC	3/22/2017	1330																1330
LWEC	3/22/2017	1345																1345
LWEC	3/22/2017	1400																1400
LWEC	3/22/2017	1415																1415
LWEC	3/22/2017	1430																1430
LWEC	3/22/2017	1445																1445
LWEC	3/22/2017	1500																1500
LWEC	3/22/2017	1515																1515
LWEC	3/22/2017	1530																1530
LWEC	3/22/2017	1545																1545
LWEC	3/22/2017	1600																1600
LWEC	3/22/2017	1615																1615
LWEC	3/22/2017	1630																1630
LWEC	3/22/2017	1645																1645
LWEC	3/22/2017	1700																1700
LWEC	3/22/2017	1715																1715
LWEC	3/22/2017	1730																1730
LWEC	3/22/2017	1745																1745
LWEC	3/22/2017	1800																1800
LWEC	3/22/2017	1815																1815
LWEC	3/22/2017	1830																1830
LWEC	3/22/2017	1845																1845
LWEC	3/22/2017	1900																1900
LWEC	3/22/2017	1915																1915
LWEC	3/22/2017	1930																1930
LWEC	3/22/2017	1945																1945
LWEC	3/22/2017	2000																2000

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**APPENDIX D  
LABORATORY REPORTS**

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## HCI

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Your Project #: 14464.007.005.001  
Site#: L'ANSE, MI  
Site Location: L'ANSE WARDEN

**Attention: Ken Hill**  
Weston Solutions Inc  
1400 Weston Way  
West Chester, PA  
USA 19380

**Report Date: 2017/03/27**  
Report #: R4404678  
Version: 1 - Final

**CERTIFICATE OF ANALYSIS**

**MAXXAM JOB #: B758294**  
Received: 2017/03/23, 15:12

Sample Matrix: Stack Sampling Train  
# Samples Received: 6

Analyses	Quantity	Date Extracted	Date Analyzed	Laboratory Method	Reference
Hydrogen Halides in H2SO4 Imp.	6	2017/03/24	2017/03/24	BRL SOP-00108	EPA 26A m
Volume of Sulfuric Acid Impinger	5	N/A	2017/03/24		

**Remarks:**

Maxxam Analytics' laboratories are accredited to ISO/IEC 17025:2005 for specific parameters on scopes of accreditation. Unless otherwise noted, procedures used by Maxxam are based upon recognized Provincial, Federal or US method compendia such as CCME, MDDELCC, EPA, APHA.

All work recorded herein has been done in accordance with procedures and practices ordinarily exercised by professionals in Maxxam's profession using accepted testing methodologies, quality assurance and quality control procedures (except where otherwise agreed by the client and Maxxam in writing). All data is in statistical control and has met quality control and method performance criteria unless otherwise noted. All method blanks are reported; unless indicated otherwise, associated sample data are not blank corrected.

Maxxam Analytics' liability is limited to the actual cost of the requested analyses, unless otherwise agreed in writing. There is no other warranty expressed or implied. Maxxam has been retained to provide analysis of samples provided by the Client using the testing methodology referenced in this report. Interpretation and use of test results are the sole responsibility of the Client and are not within the scope of services provided by Maxxam, unless otherwise agreed in writing.

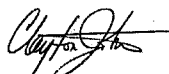
Solid sample results, except biota, are based on dry weight unless otherwise indicated. Organic analyses are not recovery corrected except for isotope dilution methods.

Results relate to samples tested.

This Certificate shall not be reproduced except in full, without the written approval of the laboratory.

Reference Method suffix "m" indicates test methods incorporate validated modifications from specific reference methods to improve performance.

Encryption Key



Clayton Johnson  
Project Manager - Air Toxics, Source Evaluation  
27 Mar 2017 16:46:59

Please direct all questions regarding this Certificate of Analysis to your Project Manager.  
Clayton Johnson, Project Manager - Air Toxics, Source Evaluation  
Email: CJohnson@maxxam.ca  
Phone# (905)817-5769

Maxxam has procedures in place to guard against improper use of the electronic signature and have the required "signatories", as per section 5.10.2 of ISO/IEC 17025:2005(E), signing the reports. For Service Group specific validation please refer to the Validation Signature Page.

**EPA M26A HYDROGEN HALIDES AND HALOGENS (STACK SAMPLING TRAIN)**

Maxxam ID		ECH997	ECH998	ECH999			
Sampling Date		2017/03/22	2017/03/22	2017/03/22			
	UNITS	M26A- ESP OUT- SB H2SO4	M26A- ESP OUT- SB DI	M26A- ESP OUT- R1	RDL	MDL	QC Batch
Sulfuric Acid Volume	ml	112	133	540	1	1	4912915
Hydrochloric Acid	ug	<250	<250	7600	250	75	4912919
RDL = Reportable Detection Limit QC Batch = Quality Control Batch							

Maxxam ID		ECH999	ECI000	ECI001			
Sampling Date		2017/03/22	2017/03/22	2017/03/22			
	UNITS	M26A- ESP OUT- R1 Lab-Dup	M26A- ESP OUT- R2	M26A- ESP OUT- R3	RDL	MDL	QC Batch
Sulfuric Acid Volume	ml	N/A	526	532	1	1	4912915
Hydrochloric Acid	ug	7500	7900	8400	250	75	4912919
RDL = Reportable Detection Limit QC Batch = Quality Control Batch Lab-Dup = Laboratory Initiated Duplicate N/A = Not Applicable							

Maxxam ID		ECI002			
Sampling Date					
	UNITS	AUDIT- 021517T- 1440	RDL	MDL	QC Batch
Hydrochloric Acid	ug	12	0.10	0.030	4912919
RDL = Reportable Detection Limit QC Batch = Quality Control Batch					

**TEST SUMMARY**

**Maxxam ID:** ECH997  
**Sample ID:** M26A- ESP OUT- SB H2SO4  
**Matrix:** Stack Sampling Train

**Collected:** 2017/03/22  
**Shipped:**  
**Received:** 2017/03/23

Test Description	Instrumentation	Batch	Extracted	Date Analyzed	Analyst
Hydrogen Halides in H2SO4 Imp.	IC/SPEC	4912919	2017/03/24	2017/03/24	Ann-Marie Stern
Volume of Sulfuric Acid Impinger		4912915	N/A	2017/03/24	Frank Mo

**Maxxam ID:** ECH998  
**Sample ID:** M26A- ESP OUT- SB DI  
**Matrix:** Stack Sampling Train

**Collected:** 2017/03/22  
**Shipped:**  
**Received:** 2017/03/23

Test Description	Instrumentation	Batch	Extracted	Date Analyzed	Analyst
Hydrogen Halides in H2SO4 Imp.	IC/SPEC	4912919	2017/03/24	2017/03/24	Ann-Marie Stern
Volume of Sulfuric Acid Impinger		4912915	N/A	2017/03/24	Frank Mo

**Maxxam ID:** ECH999  
**Sample ID:** M26A- ESP OUT- R1  
**Matrix:** Stack Sampling Train

**Collected:** 2017/03/22  
**Shipped:**  
**Received:** 2017/03/23

Test Description	Instrumentation	Batch	Extracted	Date Analyzed	Analyst
Hydrogen Halides in H2SO4 Imp.	IC/SPEC	4912919	2017/03/24	2017/03/24	Ann-Marie Stern
Volume of Sulfuric Acid Impinger		4912915	N/A	2017/03/24	Frank Mo

**Maxxam ID:** ECH999 Dup  
**Sample ID:** M26A- ESP OUT- R1  
**Matrix:** Stack Sampling Train

**Collected:** 2017/03/22  
**Shipped:**  
**Received:** 2017/03/23

Test Description	Instrumentation	Batch	Extracted	Date Analyzed	Analyst
Hydrogen Halides in H2SO4 Imp.	IC/SPEC	4912919	2017/03/24	2017/03/24	Ann-Marie Stern

**Maxxam ID:** ECI000  
**Sample ID:** M26A- ESP OUT- R2  
**Matrix:** Stack Sampling Train

**Collected:** 2017/03/22  
**Shipped:**  
**Received:** 2017/03/23

Test Description	Instrumentation	Batch	Extracted	Date Analyzed	Analyst
Hydrogen Halides in H2SO4 Imp.	IC/SPEC	4912919	2017/03/24	2017/03/24	Ann-Marie Stern
Volume of Sulfuric Acid Impinger		4912915	N/A	2017/03/24	Frank Mo

**Maxxam ID:** ECI001  
**Sample ID:** M26A- ESP OUT- R3  
**Matrix:** Stack Sampling Train

**Collected:** 2017/03/22  
**Shipped:**  
**Received:** 2017/03/23

Test Description	Instrumentation	Batch	Extracted	Date Analyzed	Analyst
Hydrogen Halides in H2SO4 Imp.	IC/SPEC	4912919	2017/03/24	2017/03/24	Ann-Marie Stern
Volume of Sulfuric Acid Impinger		4912915	N/A	2017/03/24	Frank Mo



Maxxam Job #: B758294  
Report Date: 2017/03/27

Weston Solutions Inc  
Client Project #: 14464.007.005.001  
Site Location: L'ANSE WARDEN

### TEST SUMMARY

**Maxxam ID:** ECI002  
**Sample ID:** AUDIT- 021517T- 1440  
**Matrix:** Stack Sampling Train

**Collected:**  
**Shipped:**  
**Received:** 2017/03/23

Test Description	Instrumentation	Batch	Extracted	Date Analyzed	Analyst
Hydrogen Halides in H2SO4 Imp.	IC/SPEC	4912919	2017/03/24	2017/03/24	Ann-Marie Stern



**GENERAL COMMENTS**

Sample ECI002 [AUDIT- 021517T- 1440] : Audit reported in mg/l

**Results relate only to the items tested.**

**QUALITY ASSURANCE REPORT**

QA/QC				Date		%		
Batch	Init	QC Type	Parameter	Analyzed	Value	Recovery	UNITS	QC Limits
4912919	A_S	Matrix Spike(ECH999)	Hydrochloric Acid	2017/03/24		94	%	80 - 120
4912919	A_S	Spiked Blank	Hydrochloric Acid	2017/03/24		99	%	90 - 110
4912919	A_S	Method Blank	Hydrochloric Acid	2017/03/24	<250		ug	
4912919	A_S	RPD - Sample/Sample Dup	Hydrochloric Acid	2017/03/24	0.53		%	20

Matrix Spike: A sample to which a known amount of the analyte of interest has been added. Used to evaluate sample matrix interference.

Spiked Blank: A blank matrix sample to which a known amount of the analyte, usually from a second source, has been added. Used to evaluate method accuracy.

Method Blank: A blank matrix containing all reagents used in the analytical procedure. Used to identify laboratory contamination.

**VALIDATION SIGNATURE PAGE**

The analytical data and all QC contained in this report were reviewed and validated by the following individual(s).



---

Frank Mo, B.Sc., Inorganic Lab. Manager

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Maxxam has procedures in place to guard against improper use of the electronic signature and have the required "signatories", as per section 5.10.2 of ISO/IEC 17025:2005(E), signing the reports. For Service Group specific validation please refer to the Validation Signature Page.

Lab Tracking Number



### Chain-of-Custody Record/Lab Work Request

<b>Client</b>	L'Anse Warden, L'Anse MI		
<b>Work Order Number</b>	14464.007.005.0001	<b>Phone Number</b>	610-701-3043
<b>Contact Person</b>	Ken Hill	<b>Turn Around Time</b>	3 Day <del>Standard</del>

Lab ID	Field Sample ID	Sample Collection Date	Analysis	Analyses Requested/Other Info		Sample Check-off
				Total Pptg/L Vol		
	LWEC - Boiler - ESP-OUT - 1 - M26A - H2SO4	3/22/17	M26A	540		
	LWEC - Boiler - ESP-OUT - 2 - M26A - H2SO4	3/22/17	M26A	526		
	LWEC - Boiler - ESP-OUT - 3 - M26A - H2SO4	3/22/17	M26A	532		
	LWEC Boiler - ESP-OUT - SB - M26A - H2SO4	3/22/17	M26A			
	LWEC Boiler - ESP-OUT - SB - M26A - DIH2O	3/22/17	M26A			
	LWEC-HCL-Q1-2017-Audit		M26A			

**Notes:** M26A - Chlorides by IC per Method 9057. Analyze samples in duplicate as per Method. Starting Impinger volumes; H2SO4 250 ml. Approximate Final pH \_\_\_\_\_

Relinquished By	Received By	Date	Time	Lab Use Only	
<i>[Signature]</i>	Fedex	3/22/17	1600	Shipper	Air Bill #
				Opened By	Date/Time
				Temp °C	Condition
				Custody Seals: Yes No None N/A	

**Laboratory Comments:**

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## SSAS

---



March 27, 2017

Ken Hill  
Weston Solutions  
1400 Weston Way  
West Chester, PA 19341

Enclosed is your final report for ERA's Stationary Source Audit Sample (SSAS) Program. Your final report includes an evaluation of all results submitted by your laboratory to ERA.

Data Evaluation Protocols: All analytes in ERA's SSAS Program have been evaluated comparing the reported result to the acceptance limits generated using the criteria contained in the TNI SSAS Table.

For any "Not Acceptable" results, please contact your state regulator for any corrective action requirements.

Thank you for your participation in ERA's SSAS Program. If you have any questions, please contact our Proficiency Testing Department at 1-800-372-0122.

Sincerely,

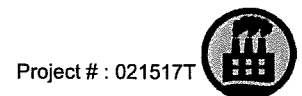
Patrick Larson  
Quality Officer

cc: Project File Number 021517T



A Waters Company

Recipient Type	Report Recipient	Contact	Project ID
Agency	MI-DEQ-Air Quality Division (SSAS) 525 West Allegan St 3rd Floor Constitution Hall PO Box 30437 Lansing, MI 48933 USA	David Patterson pattersond2@michigan.gov Phone: 517-241-7469	
Facility	L'Anse Warden Electric Company 157 South Main St L'Anse, MI 49946 USA	JR Richardson jr.richardson@pmpowergroup.com Phone: 907-885-7187	
Lab	Maxxam Analytics Inc 6740 Campobello Rd Mississauga, ON L5N 2L8 Canada	Clayton Johnson Sr. Project Manager cjohnson@maxxam.ca Phone: (905) 817-5769	
Tester	Weston Solutions 1400 Weston Way West Chester, PA 19341 USA	Ken Hill k.hill@westonsolutions.com Phone: 610-721-6521	LWEC HCI- Q1 2017





A Waters Company

# 021517T Laboratory Exception Report

Clayton Johnson  
Sr. Project Manager  
Maxxam Analytics Inc  
6740 Campobello Rd  
Mississauga, ON L5N 2L8  
(905) 817-5769

EPA ID:  
ERA Customer Number:

Not Reported  
M748564

## Evaluation Checks

There are no values reported with < where the assigned value was greater than 0.

## Not Acceptable Evaluations

There were no Not Acceptable evaluations for this study.







A Waters Company

# Final Report Results For Laboratory Maxxam Analytics Inc



16341 Table Mountain Pkwy • Golden, CO 80403 • 800.372.0122 • 303.431.8454 • fax 303.421.0159 • www.eraqc.com

Project # 0015073





## **SSAP Evaluation Report**

Project Number: **021517T**

ERA Customer Number: **M748564**

Laboratory Name: **Maxxam Analytics Inc**

### **Inorganic Results**





A Waters Company

# 021517T Evaluation Final Complete Report

Clayton Johnson  
Sr. Project Manager  
Maxxam Analytics Inc  
6740 Campobello Rd  
Mississauga, ON L5N 2L8  
(905) 817-5769

EPA ID:  
ERA Customer Number:

Not Reported  
M748564

TNI Analyte Code	Analyte	Units	Reported Value	Assigned Value	Acceptance Limits	Performance Evaluation	Method Description	Analysis Date	Analyst Name
------------------	---------	-------	----------------	----------------	-------------------	------------------------	--------------------	---------------	--------------

*SSAP Hydrogen Halides in Impinger Solution (cat# 1440, lot# 021517T) Study Dates: 02/15/17 - 03/27/17*

1770	Hydrogen Chloride	mg/L	12.0	11.8	10.6 - 13.0	Acceptable	EPA 26A 2000	3/24/2017	
1775	Hydrogen Fluoride	mg/L		14.5	13.0 - 16.0	Not Reported			



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**APPENDIX E**  
**FUEL SAMPLE RESULTS**

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April 20, 2017

Service Request No:T1700520

Mr. J.R. Richardson  
L'Anse Warden Electric Co., LLC  
157 S. Main St.  
L'Anse, MI 49946

**Laboratory Results for: Nov. & Dec. 2016 Composite**

Dear Mr.Richardson,

Enclosed are the results of the sample(s) submitted to our laboratory March 24, 2017  
For your reference, these analyses have been assigned our service request number **T1700520**.

All analyses were performed according to our laboratory's quality assurance program. All results are intended to be considered in their entirety, and ALS Environmental is not responsible for use of less than the complete report. Results apply only to the items submitted to the laboratory for analysis and individual items (samples) analyzed, as listed in the report.

Please contact me if you have any questions. My extension is 7102. You may also contact me via email at [Wendy.Hyatt@alsglobal.com](mailto:Wendy.Hyatt@alsglobal.com).

Respectfully submitted,

**ALS Group USA, Corp. dba ALS Environmental**

Wendy Hyatt  
Client Services  
Manager

ADDRESS 3860 S. Palo Verde Road, Suite 302, Tucson, AZ 85714  
PHONE +1 520 573 1061 | FAX +1 520 573 1063  
ALS Group USA, Corp.  
dba ALS Environmental

**Client:** L'Anse Warden Electric Co., LLC  
**Project:** Nov. & Dec. 2016 Composite

**Service Request:**T1700520

**SAMPLE CROSS-REFERENCE**

<u>SAMPLE #</u>	<u>CLIENT SAMPLE ID</u>	<u>DATE</u>	<u>TIME</u>
T1700520-001	1 Comp. B. of R/R Tie		0000
T1700520-002	1 Comp. B. Wood Chips		0000
T1700520-003	1 Comp. B. of TDF		0000



**ALS Environmental - Tucson**  
 ADDRESS 3860 S. Palo Verde Road, Suite 302, Tucson, AZ 85714  
 PHONE +1 520 573 1061 FAX +1 520 573 1063  
 ALS Group

## Chain of Custody

**T1700520** 5  
 L'Anse Warden Electric Co., LLC  
 Nov. & Dec. 2016 Composite

Work Order No.:



Project Manager: JR Richardson				Bill to: Midge Axley																																																																																																																												
Client Name: L'Anse Warden Electric Co., LLC				Company: L'Anse Warden Electric Co., LLC																																																																																																																												
Address: 157 S. Main St.				Address: PO Box 695																																																																																																																												
City, State ZIP: L'Anse, MI 49946				City, State ZIP: White Pine, MI 49971																																																																																																																												
Email: jr.richardson@pmpowergroup.com Phone: 906.885.7187				Email: midge.axley@pmpowergroup.co Phone: 906.885.7402																																																																																																																												
Project Name: Nov. & Dec. 2016 Composite				<b>REQUESTED ANALYSIS</b>																																																																																																																												
Project Number:				<table border="1" style="width: 100%; height: 100%; text-align: center;"> <tr><td>No. of Containers</td><td>Prep Grind</td><td>Moisture, Total - ASTM E871</td><td>Moisture, Total - ASTM D3173</td><td>Sulfur, Total - ASTM D6700-01</td><td>Sulfur, Total - ASTM D4239</td><td>Chlorine, Total - EPA 5050/9056</td><td>Heat Content - ASTM D5865</td><td>Wire Content - ASTM D6700-01</td><td>Carbon Percent ASTM D6316</td><td>Total Lead - EPA 6020</td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> <tr><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr> </table>				No. of Containers	Prep Grind	Moisture, Total - ASTM E871	Moisture, Total - ASTM D3173	Sulfur, Total - ASTM D6700-01	Sulfur, Total - ASTM D4239	Chlorine, Total - EPA 5050/9056	Heat Content - ASTM D5865	Wire Content - ASTM D6700-01	Carbon Percent ASTM D6316	Total Lead - EPA 6020																																																																																																														
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Print Name	Signature	Date/Time	Print Name	Signature	Date/Time																																																																																																																											
John Polkky	<i>[Signature]</i>	03-20-17/1600	U.P.S	<i>[Signature]</i>	3/20/2017																																																																																																																											
			Hilary Kaminski <i>[Signature]</i> 3/24/17 1105																																																																																																																													



3860 S. Palo Verde Road, Suite 302  
 Tucson, AZ 85714  
 T: +1 520 573 1061  
 F: +1 520 573 1063  
 www.alsglobal.com

Sample Receipt Form

T1700520 5

L'Anse Warden Electric Co., LLC  
 Nov. & Dec. 2016 Composite

Client/Project: **L'Anse Warden Electric Co** Work Order Number 

Received by: **Hilary Kaminski** Date & Time: **3/24/17 1105** Matrix: **Solid**

Samples were received via: **UPS** Samples were received in: **Box**

Were custody seals on containers?  Yes  No  NA If yes, how many and where?

If present were custody seals intact?  Yes  No If present, were they signed and dated?  Yes  No

Arrival Temp C	Temp Blank C	Tracking Number
ambient	na	1Z526F7E0394223037

Packing material used? **Paper**

Did all the bottles arrive in good condition (unbroken)?  Yes  No  NA If No, record comments below

Did all sample labels and tags agree with COC?  Yes  No  NA If No, record discrepancies below

Were all the appropriate containers and volumes received for the tests indicated?  Yes  No  NA

Are samples received deemed acceptable?  Yes  No

Comments:  
 3-gallon ziploc bags

Notes, discrepancies, & resolutions:

As a part of ISO 17025 protocols, ALS must notify clients that the quoted analytical methods performed by ALS may have minor modifications from the methods as published. These modifications are written into our Standard Operating Procedures and do not impact the quality of the data. Receipt of this document will be considered an acceptance of the procedures used by the laboratory for analysis unless notified by the client.

Modifications may include, but are not limited to:

- The analysis of a sample matrix that differs from that stated in the published method (example - ASTM D5865 Standard Test Method for Gross Calorific Value of Coal and Coke is used for other matrices such as biomass, Tire Derived Fuel, etc.).
- Analyzing a sample mass that differs from those in the published method (example - to accommodate samples with high concentrations of analyte, samples of limited volume, or to comply with the instrument manufacturer's operating guidelines).
- Instruments used for the analysis may differ from those listed in the published method (example - using ICP-OES when the method references flame Atomic Absorption Spectroscopy)

RIGHT SOLUTIONS | RIGHT PARTNER





Client: L'Anse Warden Electric Co., LLC  
 157 S. Main St.  
 L'Anse, MI 49946

Attn: J.R. Richardson

Project: Nov. & Dec. 2016 Composite

Date Received: 3/24/17

### Certificate of Analysis

Sample ID:	Sample Date and Time:	Lab #:	Moisture, Total	Chlorine	Sulfur	Lead, Total	Heating Value (Gross)	
			E871 wt%	5050/9056 Moist. Free mg/kg	D4239 Moist. Free wt%	3050/6010 Moist. Free mg/kg	E711 As Received BTU/lb	Moist Free BTU/lb
1 Comp. B. of R/R Tie	n/a	T1700520-001	31.93	54	0.074	4	6,204	8,965
1 Comp. B. Wood Chips	n/a	T1700520-002	38.57	42	0.023	<1	5,076	8,530

ADDRESS 3860 S. Palo Verde Road, Suite 302, Tucson, AZ 85714  
 PHONE +1 520 573 1061

Rpt-T1700520 L'Anse Warden Electric Co., LLC Richardson 001 002,  
 4/20/2017



Client: L'Anse Warden Electric Co., LLC  
 157 S. Main St.  
 L'Anse, MI 49946

Attn: J.R. Richardson

Project: Nov. & Dec. 2016 Composite

Date Received: 3/24/17

### Certificate of Analysis

Sample ID:	Sample Date:	Lab #:	Moisture, Total D3173 wt%	Heating Value		Heating Value		Wire Content D6700 Air Dried wt%
				Wire Free D5865 As Received BTU/lb	Moist. Free BTU/lb	With Wire calculated As Received BTU/lb	Moist. Free BTU/lb	
1 Comp. B. of TDF	n/a	T1700520-003	2.36	16,016	16,404	15,312	15,682	4.4

Sample ID:	Sample Date:	239	Sulfur, Total	Carbon, Total	Chlorine	Lead, Total			
			Wire Free D4239 Moist. Free wt%	Wire Free D5373 Moist. Free wt%	Wire Free 5050/9056 As Received mg/kg	Wire Free 3050/6010 Moist. Free mg/kg			
1 Comp. B. of TDF	n/a	T1700520-003	1.82	81.49	498	8			

**Notes:**

Solid samples were air dried at 40°C for several days, measured for moisture loss, coarse ground to < 6mm, and split into sub-samples, one for storage and one for further grinding to < 1mm. TDF sample required freezing with liquid nitrogen prior to the coarse and fine grinding steps. The wire was removed from the coarse ground TDF sample using magnetic separation. Analyses of TDF sample performed on a wire free sample.

ADDRESS 3860 S. Palo Verde Road, Suite 302, Tucson, AZ 85714  
 PHONE +1 520 573 1061

Rpt-T1700520 L'Anse Warden Electric Co., LLC Richardson 001 002,  
 4/20/2017

---

**APPENDIX F**  
**QUALITY CONTROL RECORDS**

---

## Long Cal and Temperature Cal Datasheet for Standard Dry Gas Meter Console

Calibrator PM

Meter Box Number 13

Ambient Temp 72

Date 3-Dec-16

Wet Test Meter Number P-2952

Temp Reference Source Thermocouple Simulator (Accuracy +/- 1°F)

Dry Gas Meter Number 6837013

Baro Press, in Hg (Pb)	29.75
------------------------	-------

Setting	Gas Volume		Temperatures				Time, min (O)	Calibration Results	
	Wet Test Meter	Dry gas Meter	Wet Test Meter	Dry Gas Meter				Y	ΔH
in H <sub>2</sub> O (ΔH)	ft <sup>3</sup> (Vw)	ft <sup>3</sup> (Vd)	°F (Tw)	Outlet, °F (Td <sub>o</sub> )	Inlet, °F (Td <sub>i</sub> )	Average, °F (Td)			
0.5	6.0	562.225	72.0	75.00	75.00	75.5	15.2	0.9910	1.8071
		568.312		76.00	76.00				
		6.087		75.50	75.50				
1.0	5.0	569.330	72.0	76.00	76.00	77.5	9.1	1.0008	1.8585
		574.365		79.00	79.00				
		5.035		77.50	77.50				
1.5	10.0	574.365	72.0	79.00	79.00	80.5	15.4	1.0037	1.9849
		584.450		82.00	82.00				
		10.085		80.50	80.50				
2.0	10.0	584.450	72.0	82.00	82.00	83.0	13.1	1.0044	1.9062
		594.562		84.00	84.00				
		10.112		83.00	83.00				
3.0	10.0	594.562	72.0	84.00	84.00	85.0	10.7	0.9996	1.9006
		604.735		86.00	86.00				
		10.173		85.00	85.00				
<b>Average</b>								<b>0.9999</b>	<b>1.8915</b>

Vw - Gas Volume passing through the wet test meter  
 Vd - Gas Volume passing through the dry gas meter  
 Tw - Temp of gas in the wet test meter  
 Tdi - Temp of the inlet gas of the dry gas meter  
 Tdo - Temp of the outlet gas of the dry gas meter  
 Td - Average temp of the gas in the dry gas meter

O - Time of calibration run  
 Pb - Barometric Pressure  
 ΔH - Pressure differential across orifice  
 Y - Ratio of accuracy of wet test meter to dry gas meter

$$Y = \frac{Vw * Pb * (td + 460)}{Vd * \left[ Pb + \frac{(\Delta H)}{13.6} \right] * (tw + 460)}$$

$$\Delta H = \left[ \frac{0.0317 * \Delta H}{Pb * (td + 460)} \right] * \left[ \frac{(tw + 460) * O}{Vw} \right]^2$$

Reference Temperature	Temperature Reading from Individual Thermocouple Input <sup>1</sup>						Average Temperature Reading	Temp Difference <sup>2</sup> (%)
	Select Temperature	Channel Number						
<input type="radio"/> °C <input checked="" type="radio"/> °F	1	2	3	4	5	6		
32	32	32	32	32	32	32	32.0	0.0%
212	212	212	212	212	212	212	212.0	0.0%
932	932	932	932	932	932	932	932.0	0.0%
1832	1833	1833	1833	1833	1833	1833	1833.0	0.0%

<sup>1</sup> - Channel Temps must agree with +/- 5°F or 3°C

<sup>2</sup> - Acceptable Temperature Difference less than 1.5 %

$$\text{Temp Diff} = \left[ \frac{(\text{Reference Temp}^{\circ\text{F}} + 460) - (\text{Test Temp}^{\circ\text{F}} + 460)}{\text{Reference Temp}^{\circ\text{F}} + 460} \right]$$

## Y Factor Calibration Check Calculation

METHOD 26A (HCl) TEST TRAIN

METER BOX NO. 13

RUN NO. 3 3/22/17

MWd = Dry molecular weight source gas, lb/lb-mole.	
0.32 = Molecular weight of oxygen, divided by 100.	
0.44 = Molecular weight of carbon dioxide, divided by 100.	
0.28 = Molecular weight of nitrogen or carbon monoxide, divided by 100.	
% CO <sub>2</sub> = Percent carbon dioxide by volume, dry basis.	12.0
% O <sub>2</sub> = Percent oxygen by volume, dry basis.	7.9

$$MWd = (0.32 * O_2) + (0.44 * CO_2) + (0.28 * (100 - (CO_2 + O_2)))$$

$$MWd = (0.32 * 7.9) + (0.44 * 12) + (0.28 * (100 - (12 + 7.9)))$$

$$MWd = (2.53) + (5.28) + (22.43)$$

$$MWd = 30.24$$

Tma = Source Temperature, absolute(°R)	
Tm = Average dry gas meter temperature, deg F.	67.4

$$Tma = Ts + 460$$

$$Tma = 67.42 + 460$$

$$Tma = 527.42$$

Ps = Absolute meter pressure, inches Hg.	
13.60 = Specific gravity of mercury.	
delta H = Avg pressure drop across the orifice meter during sampling, in H <sub>2</sub> O	1.800
Pb = Barometric Pressure, in Hg.	30.00

$$Pm = Pb + (\text{delta H} / 13.6)$$

$$Pm = 30 + (1.8 / 13.6)$$

$$Pm = 30.13$$

Yqa = dry gas meter calibration check value, dimensionless.	
0.03 = (29.92/528)(0.75) <sup>2</sup> (in. Hg/°R) cfm <sup>2</sup> .	
29.00 = dry molecular weight of air, lb/lb-mole.	
Vm = Volume of gas sample measured by the dry gas meter at meter conditions, dcf.	41.170
Y = Dry gas meter calibration factor (based on full calibration)	0.9999
Delta H@ = Dry Gas meter orifice calibration coefficient, in. H <sub>2</sub> O.	1.8915
Q SQRT Delta H = Avg SQRT press. drop across the orifice meter during sampling, in. H <sub>2</sub> O	1.3416
O = Total sampling time, minutes.	60

$$Yqa = (O / Vm) * SQRT (0.0319 * Tma * 29) / (\text{Delta H}@ * Pm * MWd) * \text{avg SQRT Delta H}$$

$$Yqa = (60.00 / 41.17) * SQRT (0.0319 * 527.42 * 29) / (1.89 * 30.13 * 30.24) * 1.34$$

$$Yqa = 1.457 * SQRT 487.913 / 1,723.177 * 1.34$$

$$Yqa = 1.040$$

Diff = Absolute difference between Yqa and Y	
--	--

$$\text{Diff} = ((Y - Yqa) / Y) * 100$$

$$\text{Diff} = ((0.9999 - 1.040) / 0.9999) * 100$$

$$\text{Diff} = 4.01$$

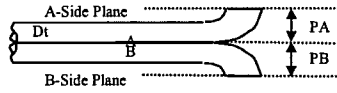
# Type S Pitot Tube Inspection Data Form

Pitot Tube Identification Number: P-676

If all Criteria PASS  
Cp is equal to 0.84

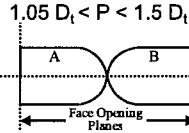
Inspection Date 11/16/16 Individual Conducting Inspection KS

**PASS/FAIL**



Distance to A Plane (PA) - inches 0.483  
 Distance to B Plane (PB) - inches 0.483  
 Pitot OD (D<sub>t</sub>) - inches 0.375

PASS  
PASS

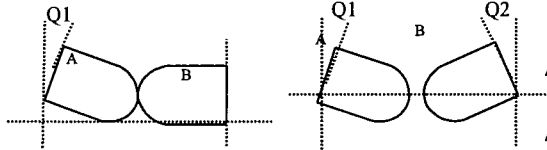


PA must Equal PB

Are Open Faces Aligned  
Perpendicular to the Tube Axis

YES  NO

PASS



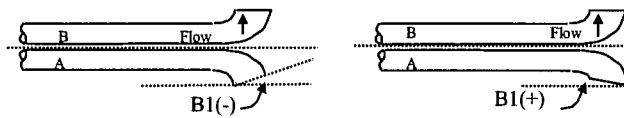
Angle of Q1 from vertical A Tube-  
degrees (absolute) 0

PASS

Angle of Q2 from vertical B Tube-  
degrees (absolute) 0

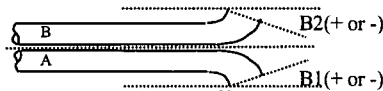
PASS

Q1 and Q2 must be  $\leq 10^\circ$



Angle of B1 from  
vertical A Tube-  
degrees (absolute) 1

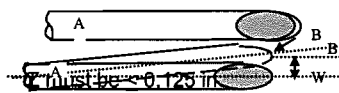
PASS



Angle of B1 from  
vertical B Tube-  
degrees (absolute) 0

PASS

B1 or B2 must be  $\leq 5^\circ$



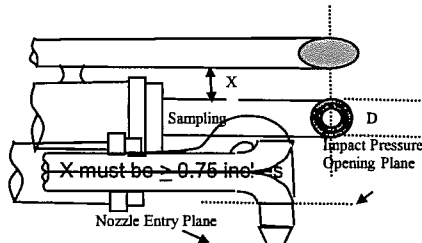
Horizontal offset between A and  
B Tubes (Z) - inches 0.009

PASS

Vertical offset between A and B  
Tubes (W) - inches 0.005

PASS

W must be  $\leq 0.03125$  inches

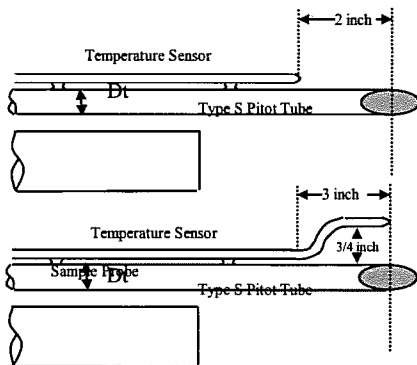


Distance between Sample  
Nozzle and Pitot (X) - inches N/A

PASS

Impact Pressure  
Opening Plane is  
above the Nozzle  
Entry Plane

YES  NO  
 NA



Thermocouple meets  
the Distance Criteria  
in the adjacent figure

YES  NO  
 NA

Thermocouple meets  
the Distance Criteria  
in the adjacent figure

YES  NO  
 NA

Sample Probe

# CERTIFICATE OF ANALYSIS

## Grade of Product: EPA Protocol

Part Number:	E03NI79E15A00E4	Reference Number:	82-124547137-1
Cylinder Number:	CC61928	Cylinder Volume:	150.5 CF
Laboratory:	124 - Riverton - NJ	Cylinder Pressure:	2015 PSIG
PGVP Number:	B52016	Valve Outlet:	590
Gas Code:	CO2,O2,BALN	Certification Date:	Mar 29, 2016

**Expiration Date: Mar 29, 2024**

Certification performed in accordance with "EPA Traceability Protocol for Assay and Certification of Gaseous Calibration Standards (May 2012)" document EPA 600/R-12/531, using the assay procedures listed. Analytical Methodology does not require correction for analytical interference. This cylinder has a total analytical uncertainty as stated below with a confidence level of 95%. There are no significant impurities which affect the use of this calibration mixture. All concentrations are on a volume/volume basis unless otherwise noted.

Do Not Use This Cylinder below 100 psig, i.e. 0.7 megapascals.

ANALYTICAL RESULTS					
Component	Requested Concentration	Actual Concentration	Protocol Method	Total Relative Uncertainty	Assay Dates
CARBON DIOXIDE	9.000 %	8.908 %	G1	+/- 0.7% NIST Traceable	03/29/2016
OXYGEN	12.00 %	11.98 %	G1	+/- 0.4% NIST Traceable	03/29/2016
NITROGEN	Balance				

CALIBRATION STANDARDS					
Type	Lot ID	Cylinder No	Concentration	Uncertainty	Expiration Date
NTRM	13060819	CC417106	24.04 % CARBON DIOXIDE/NITROGEN	+/- 0.6%	May 16, 2019
NTRMplus	09060208	CC262337	9.961 % OXYGEN/NITROGEN	+/- 0.3%	Nov 08, 2018

ANALYTICAL EQUIPMENT		
Instrument/Make/Model	Analytical Principle	Last Multipoint Calibration
Horiba VIA 510-CO2-LDH9LRNS	NDIR	Mar 03, 2016
Horiba MPA 510-O2-7TWMJ041	Paramagnetic	Mar 03, 2016

Triad Data Available Upon Request



\_\_\_\_\_  
**Signature on file**  
**Approved for Release**

# CERTIFICATE OF ANALYSIS

## Grade of Product: EPA Protocol

Part Number: E03NI62E15A0224	Reference Number: 82-124489131-1
Cylinder Number: CC452229	Cylinder Volume: 157.2 CF
Laboratory: ASG - Riverton - NJ	Cylinder Pressure: 2015 PSIG
PGVP Number: B52015	Valve Outlet: 590
Gas Code: CO2,O2,BALN	Certification Date: Apr 24, 2015

**Expiration Date: Apr 24, 2023**

Certification performed in accordance with "EPA Traceability Protocol for Assay and Certification of Gaseous Calibration Standards (May 2012)" document EPA 600/R-12/531, using the assay procedures listed. Analytical Methodology does not require correction for analytical interference. This cylinder has a total analytical uncertainty as stated below with a confidence level of 95%. There are no significant impurities which affect the use of this calibration mixture. All concentrations are on a volume/volume basis unless otherwise noted.

Do Not Use This Cylinder below 100 psig, i.e. 0.7 megapascals.

ANALYTICAL RESULTS					
Component	Requested Concentration	Actual Concentration	Protocol Method	Total Relative Uncertainty	Assay Dates
CARBON DIOXIDE	17.00 %	16.63 %	G1	+/- 0.7% NIST Traceable	04/24/2015
OXYGEN	21.00 %	21.65 %	G1	+/- 0.6% NIST Traceable	04/24/2015
NITROGEN	Balance				

CALIBRATION STANDARDS					
Type	Lot ID	Cylinder No	Concentration	Uncertainty	Expiration Date
NTRM	13060739	CC414621	16.939 % CARBON DIOXIDE/NITROGEN	+/- 0.6%	May 08, 2019
NTRM	09061414	CC273509	22.53 % OXYGEN/NITROGEN	+/- 0.4%	Mar 08, 2019

ANALYTICAL EQUIPMENT		
Instrument/Make/Model	Analytical Principle	Last Multipoint Calibration
Horiba VIA 510-CO2-LDH9LRNS	NDIR	Apr 17, 2015
Siemens Oxymat 6E-O2-N1-M1-0603	Paramagnetic	Apr 10, 2015

Triad Data Available Upon Request



Signature on file  
 Approved for Release



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## APPENDIX G

### EXAMPLE CALCULATIONS

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**EXAMPLE CALCULATIONS FOR  
SAMPLE VOLUME, MOISTURE AND VOLUMETRIC FLOWRATE**

Client: L'Anse, Warden Electric Company

Facility: L'Anse, MI

Test Number: Run 1

Test Date: 03/22/17

Test Location: Boiler No. 1

Test Period: 0900-1000

**1. Volume of dry gas sampled at standard conditions (68 deg F, 29.92 in. Hg), dscf.**

$$Vm(std) = \frac{17.64 \times Y \times Vm \times \left( Pb + \frac{\text{delta H}}{13.6} \right)}{(Tm + 460)}$$

$$Vm(std) = \frac{17.64 \times 0.9999 \times 40.029 \times \left( 30.00 + \frac{1.800}{13.6} \right)}{33.75 + 460} = 43.088$$

Where:

- Vm(std) = Volume of gas sample measured by the dry gas meter, corrected to standard conditions, dscf.
- Vm = Volume of gas sample measured by the dry gas meter at meter conditions, dcf.
- Pb = Barometric Pressure, in Hg.
- delt H = Average pressure drop across the orifice meter, in H<sub>2</sub>O
- Tm = Average dry gas meter temperature , deg F.
- Y = Dry gas meter calibration factor.
- 17.64 = Factor that includes ratio of standard temperature (528 deg R) to standard pressure (29.92 in. Hg), deg R/in. Hg.
- 13.6 = Specific gravity of mercury.

## 2. Volume of water vapor in the gas sample corrected to standard conditions, scf.

$$Vw(std) = (0.04707 \times Vwc) + (0.04715 \times Wwsg)$$

$$Vw(std) = (0.04707 \times 160.0) + (0.04715 \times 10.0) = 8.003$$

Where:

$Vw(std)$  = Volume of water vapor in the gas sample corrected to standard conditions, scf.

$Vwc$  = Volume of liquid condensed in impingers, ml.

$Wwsg$  = Weight of water vapor collected in silica gel, g.

0.04707 = Factor which includes the density of water (0.002201 lb/ml), the molecular weight of water (18.0 lb/lb-mole), the ideal gas constant 21.85 (in. Hg) (ft<sup>3</sup>/lb-mole)(deg R); absolute temperature at standard conditions (528 deg R), absolute pressure at standard conditions (29.92 in. Hg), ft<sup>3</sup>/ml.

0.04715 = Factor which includes the molecular weight of water (18.0 lb/lb-mole), the ideal gas constant 21.85 (in. Hg) (ft<sup>3</sup>/lb-mole)(deg R); absolute temperature at standard conditions (528 deg R), absolute pressure at standard conditions (29.92 in. Hg), and 453.6 g/lb, ft<sup>3</sup>/g.

## 3. Moisture content

$$bws = \frac{Vw(std)}{Vw(std) + Vm(std)}$$

$$bws = \frac{8.003}{8.003 + 43.088} = 0.157$$

Where:

$bws$  = Proportion of water vapor, by volume, in the gas stream, dimensionless.

#### 4. Mole fraction of dry gas.

$$M_d = 1 - b_{ws}$$

$$M_d = 1 - 0.157 = 0.843$$

Where:

$$M_d = \text{Mole fraction of dry gas, dimensionless.}$$

#### 5. Dry molecular weight of gas stream, lb/lb-mole.

$$MW_d = (0.440 \times \% \text{CO}_2) + (0.320 \times \% \text{O}_2) + (0.280 \times (\% \text{N}_2 + \% \text{CO}))$$

$$MW_d = (0.440 \times 12.3) + (0.320 \times 7.8) + (0.280 \times (79.9 + 0.00))$$

$$= 30.28$$

Where:

$$MW_d = \text{Dry molecular weight, lb/lb-mole.}$$

$$\% \text{CO}_2 = \text{Percent carbon dioxide by volume, dry basis.}$$

$$\% \text{O}_2 = \text{Percent oxygen by volume, dry basis.}$$

$$\% \text{N}_2 = \text{Percent nitrogen by volume, dry basis.}$$

$$\% \text{CO} = \text{Percent carbon monoxide by volume, dry basis.}$$

$$0.440 = \text{Molecular weight of carbon dioxide, divided by 100.}$$

$$0.320 = \text{Molecular weight of oxygen, divided by 100.}$$

$$0.280 = \text{Molecular weight of nitrogen or carbon monoxide, divided by 100.}$$

#### 6. Actual molecular weight of gas stream (wet basis), lb/lb-mole.

$$MW_s = (MW_d \times M_d) + (18 \times (1 - M_d))$$

$$MW_s = (30.28 \times 0.843) + (18 \times (1 - 0.843)) = 28.36$$

Where:

$$MW_s = \text{Molecular weight of wet gas, lb/lb-mole.}$$

$$18 = \text{Molecular weight of water, lb/lb-mole.}$$

**7. Average velocity of gas stream at actual conditions, ft/sec.**

$$V_s = 85.49 \times C_p \times ((\Delta p)^{1/2})_{\text{avg}} \times \left( \frac{T_s (\text{avg})}{P_s \times MW_s} \right)^{1/2}$$

$$V_s = 85.49 \times 0.84 \times 0.794300 \times \left( \frac{904}{29.12 \times 28.36} \right)^{1/2} = 59.7$$

Where:

- $V_s$  = Average gas stream velocity, ft/sec.  
 $85.49$  = Pitot tube constant, ft/sec  $\times \frac{(\text{lb/lb-mole})(\text{in. Hg})^{1/2}}{(\text{deg R})(\text{in H}_2\text{O})}$   
 $C_p$  = Pitot tube coefficient, dimensionless.  
 $T_s$  = Absolute gas stream temperature, deg R =  $T_s, \text{ deg F} + 460$ .  
 $P_s$  = Absolute gas stack pressure, in. Hg. =  $P_b + \frac{P(\text{static})}{13.6}$   
 $\Delta p$  = Velocity head of stack, in. H<sub>2</sub>O

**8. Average gas stream volumetric flowrate at actual conditions, wacf/min.**

$$Q_s(\text{act}) = 60 \times V_s \times A_s$$

$$Q_s(\text{act}) = 60 \times 59.69 \times 39.00 = 139678$$

Where:

- $Q_s(\text{act})$  = Volumetric flowrate of wet stack gas at actual conditions, wacf/min.  
 $A_s$  = Cross-sectional area of stack, ft<sup>2</sup>.  
 $60$  = Conversion factor from seconds to minutes.

**9. Average gas stream dry volumetric flowrate at standard conditions, dscf/min.**

$$Qs(std) = 17.64 \times Md \times \frac{Ps}{Ts} \times Qs(act)$$
$$Qs(std) = 17.64 \times 0.843 \times \frac{29.12}{904} \times 139678$$
$$= 66896$$

Where:

$Qs(std)$  = Volumetric flowrate of dry stack gas at standard conditions, dscf/min.

**EXAMPLE CALCULATIONS FOR  
CONCENTRATIONS AND EMISSION RATES OF HCl**

Client: L'Anse, Warden Electric Company  
Test Number: Run 1  
Test Location: Boiler No. 1

Facility: L'Anse, MI  
Test Date: 03/22/17  
Test Period: 0900-1000

**1. Hydrogen chloride concentration, lb/dscf.**

$$C1(HCl) = \frac{W(HCl) \times 2.2046 \times 10^{-6}}{V_{dm}(std)}$$

$$C1(HCl) = \frac{7.6000 \times 2.2046 \times 10^{-6}}{43.088}$$

$$C1(HCl) = 3.89E-07$$

Where:

W(HCl) = Weight of hydrogen chloride collected in sample, mg.  
C1(HCl) = Hydrogen chloride concentration, lbs/dscf.  
 $2.2046 \times 10^{-6}$  = Conversion factor from mg to lbs.

**2. Hydrogen chloride concentration, ppmv.**

$$C2(HCl) = \frac{385.35 \times 10^6}{MW} \times C1(HCl)$$

$$C2(HCl) = \frac{385.35 \times 10^6}{36.45} \times 0.0000003888$$

$$= 4.11$$

where:

C2(HCl) = Concentration of HCl in stack gas, parts per million by volume (dry basis).  
 $385.35 \times 10^6$  = Conversion factor from lbs/ppm.

**2. Hydrogen chloride mass emission rate, lb/hr.**

$$PMR1(HCl) = C1(HCl) \times Qs(std) \times 60$$

$$PMR1(HCl) = 0.0000003888 \times 66896 \times 60$$

$$= 1.56$$

where:

PMR1(HCl) = Hydrogen chloride mass emission rate, lb/hr.

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## APPENDIX H PROJECT PARTICIPANTS

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<b>Team Member</b>	<b>Title</b>	<b>Company</b>
JR Richardson	Technical Manager	LWEC
John Polky	Plant Fuels Supervisor	
Chris Anderson	Operations/Maintenance Manager	
Al Clishe	Senior Consultant	
Steve Kohl	Legal Counsel	Warner Norcross & Judd LLP
Jed Chrestensen	Project Engineer	Mannik Smith Group
Ken Hill	Senior Project Manager	Weston Solutions, Inc.
Brian Allan	Report Coordinator	
Tyson Belknap	Project Scientist	
Kyle Schweitzer	Technician III	