

# EPA Tools and Resources Webinar: Treating Contaminants of Emerging Concern

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EPA Office of Research and Development

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## 1) Nitrate/Perchlorate

- 1) Anion exchange
- 2) Point of Use (POU) membranes
- 3) Biological treatment (anaerobic)

## 2) Microcystins

- 1) Cell removal
- 2) Powdered activated carbon
- 3) Disinfection/Oxidation

## 3) PFAS

- 1) Activated carbon
- 2) Anion exchange
- 3) Reverse osmosis



# Research: Treatment

## Publically Available Drinking Water Treatability Database

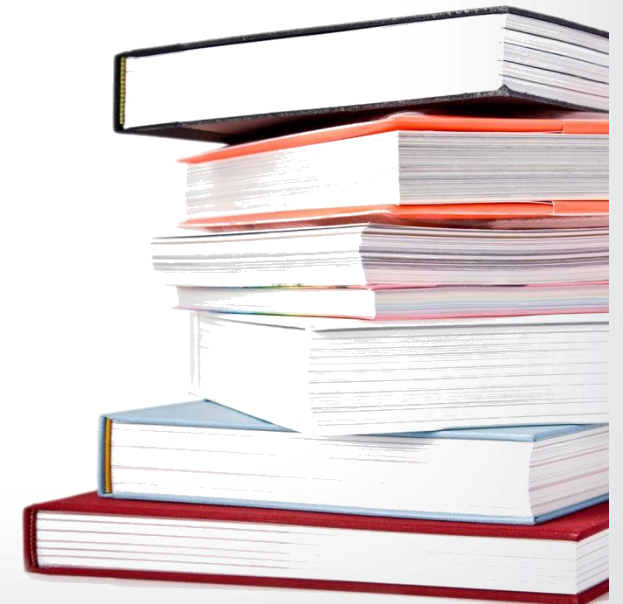
Interactive literature review database that contains over 65 regulated and unregulated contaminants and covers 34 treatment processes commonly employed or known to be effective (thousands of sources assembled on one site)

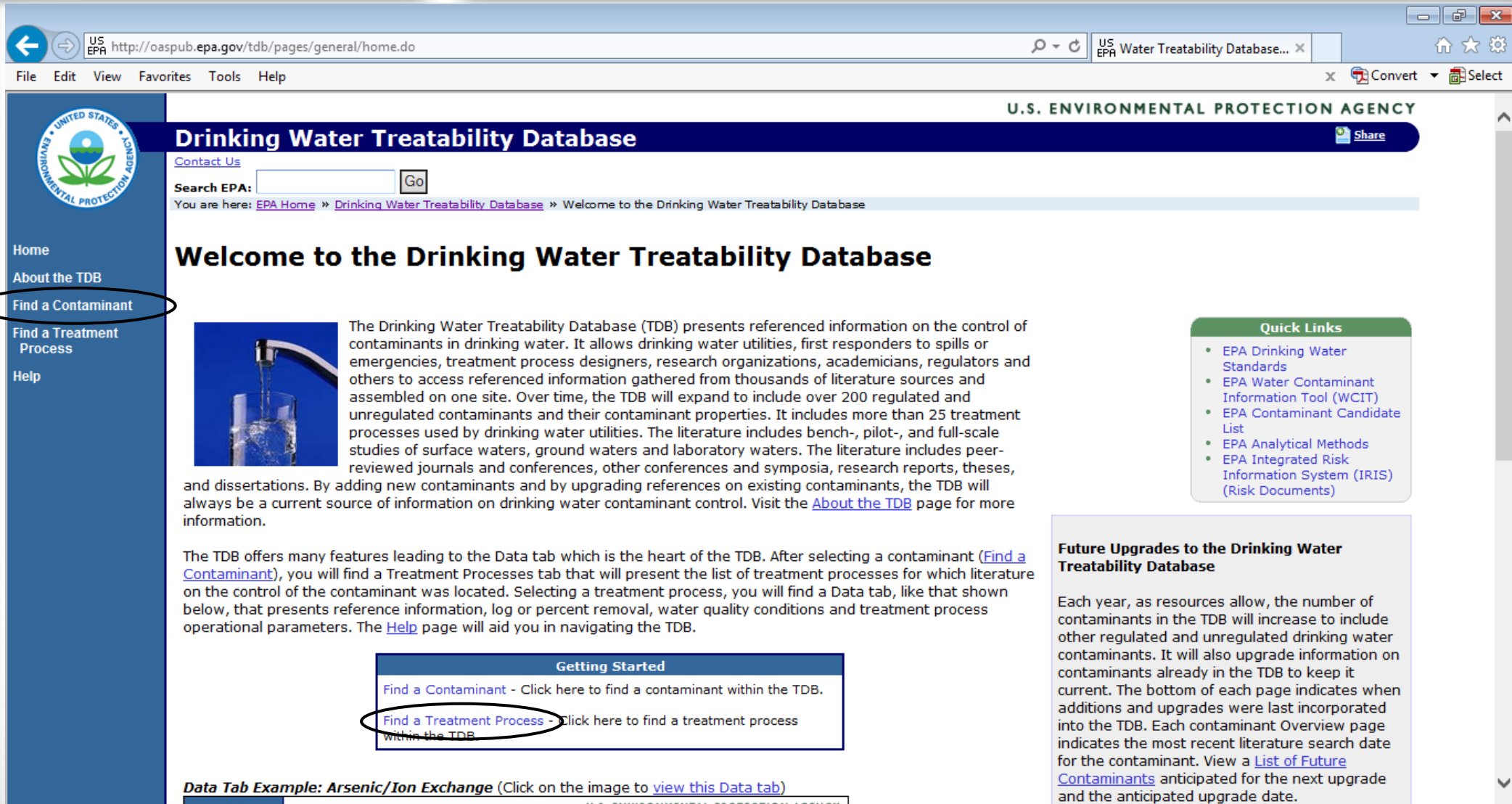
### Currently available:

- Nitrate
- Perchlorate
- Microcystins
- PFOA, PFOS, PFNA, PFHxA, PFHxS, PFBS, Gen-X

<http://iaspub.epa.gov/tdb/pages/general/home.do>

Search: EPA TDB





The screenshot shows the EPA Drinking Water Treatability Database website. The browser address bar displays <http://oaspub.epa.gov/tdb/pages/general/home.do>. The page header includes the EPA logo and the text "U.S. ENVIRONMENTAL PROTECTION AGENCY". The main heading is "Drinking Water Treatability Database". A search bar is present with the text "Search EPA:" and a "Go" button. Below the search bar, a breadcrumb trail reads: "You are here: [EPA Home](#) » [Drinking Water Treatability Database](#) » Welcome to the Drinking Water Treatability Database".

The main content area features a large heading: "Welcome to the Drinking Water Treatability Database". To the left of this heading is an image of water being poured from a tap into a glass. The text below the heading describes the database's purpose: "The Drinking Water Treatability Database (TDB) presents referenced information on the control of contaminants in drinking water. It allows drinking water utilities, first responders to spills or emergencies, treatment process designers, research organizations, academicians, regulators and others to access referenced information gathered from thousands of literature sources and assembled on one site. Over time, the TDB will expand to include over 200 regulated and unregulated contaminants and their contaminant properties. It includes more than 25 treatment processes used by drinking water utilities. The literature includes bench-, pilot-, and full-scale studies of surface waters, ground waters and laboratory waters. The literature includes peer-reviewed journals and conferences, other conferences and symposia, research reports, theses, and dissertations. By adding new contaminants and by upgrading references on existing contaminants, the TDB will always be a current source of information on drinking water contaminant control. Visit the [About the TDB](#) page for more information."

Below this text, a section titled "Getting Started" contains two links: "Find a Contaminant - Click here to find a contaminant within the TDB." and "Find a Treatment Process - Click here to find a treatment process within the TDB." The "Find a Treatment Process" link is circled in red.

On the right side of the page, there is a "Quick Links" section with a list of links: "EPA Drinking Water Standards", "EPA Water Contaminant Information Tool (WCIT)", "EPA Contaminant Candidate List", "EPA Analytical Methods", and "EPA Integrated Risk Information System (IRIS) (Risk Documents)".

At the bottom right, a section titled "Future Upgrades to the Drinking Water Treatability Database" states: "Each year, as resources allow, the number of contaminants in the TDB will increase to include other regulated and unregulated drinking water contaminants. It will also upgrade information on contaminants already in the TDB to keep it current. The bottom of each page indicates when additions and upgrades were last incorporated into the TDB. Each contaminant Overview page indicates the most recent literature search date for the contaminant. View a [List of Future Contaminants](#) anticipated for the next upgrade and the anticipated upgrade date."

The left sidebar contains navigation links: "Home", "About the TDB", "Find a Contaminant" (circled in red), "Find a Treatment Process", and "Help".



# Treatability Database

The screenshot shows a web browser window displaying the EPA's Drinking Water Treatability Database. The browser's address bar shows the URL: <https://oaspub.epa.gov/tdb/pages/contaminant/treatmentSummary.do>. The page header includes the EPA logo and the text "U.S. ENVIRONMENTAL PROTECTION AGENCY". The main heading is "Drinking Water Treatability Database" with a "Share" button. Below this is a search bar with the text "Search EPA:" and a "Go" button. The breadcrumb trail reads: "You are here: EPA Home » Drinking Water Treatability Database » Perfluorooctanoic Acid". The main content area is titled "Perfluorooctanoic Acid" and has several tabs: "Overview", "Treatment Processes", "Properties", "Fate and Transport", and "References". The "Treatment Processes" tab is selected. The text under this tab describes the effectiveness of various treatment processes for removing perfluorooctanoic acid. It lists effective processes: GAC (up to 99 percent removal), membrane separation (up to > 98 percent), powdered activated carbon (88 percent), and ion exchange (73 to 95 percent). It also notes that UV irradiation at wavelengths in the 185-220 nm range and/or at long irradiation times (up to 72 hours) could potentially be effective (62 to 90 percent), and membrane filtration varied in effectiveness (22 to 56 percent). It then lists processes not considered effective: conventional treatment (no removal) and UV at wavelengths outside of the 185-220 nm range (4 percent to 10 percent removal). UV/hydrogen peroxide treatment (35 percent removal) was less effective in comparison to UV alone (45 percent) after 24 hours of irradiation. Finally, it states that studies were identified evaluating the following treatment technologies for the removal of perfluorooctanoic acid:   

- [Conventional Treatment](#) - Multiple full-scale studies reported insignificant removal of PFOA by conventional treatment. PFOA levels after conventional drinking water treatment were found to correlate to the PFOA levels detected in their surface waters source...
- [GAC Isotherm](#) - Adsorption was observed for PFOA detected in a contaminated groundwater. It was found to be nonlinear.
- [Granular Activated Carbon](#) - Removal of PFOA by GAC can be effective. Bench scale tests, including rapid small scale column tests, showed removals from less than zero to 95 percent, depending on carbon type and background TOC concentrations [1700, 2423, 2441]. At one full sca...
- [Ion Exchange](#) - Removal of PFOA using anion exchange resins was found to be effective (73 to 95 percent removal) in a bench study [2427], and in a full scale application [2424; 2441] that used a resin designed for arsenic removal. A full scale application using...



# What is a Work Breakdown?

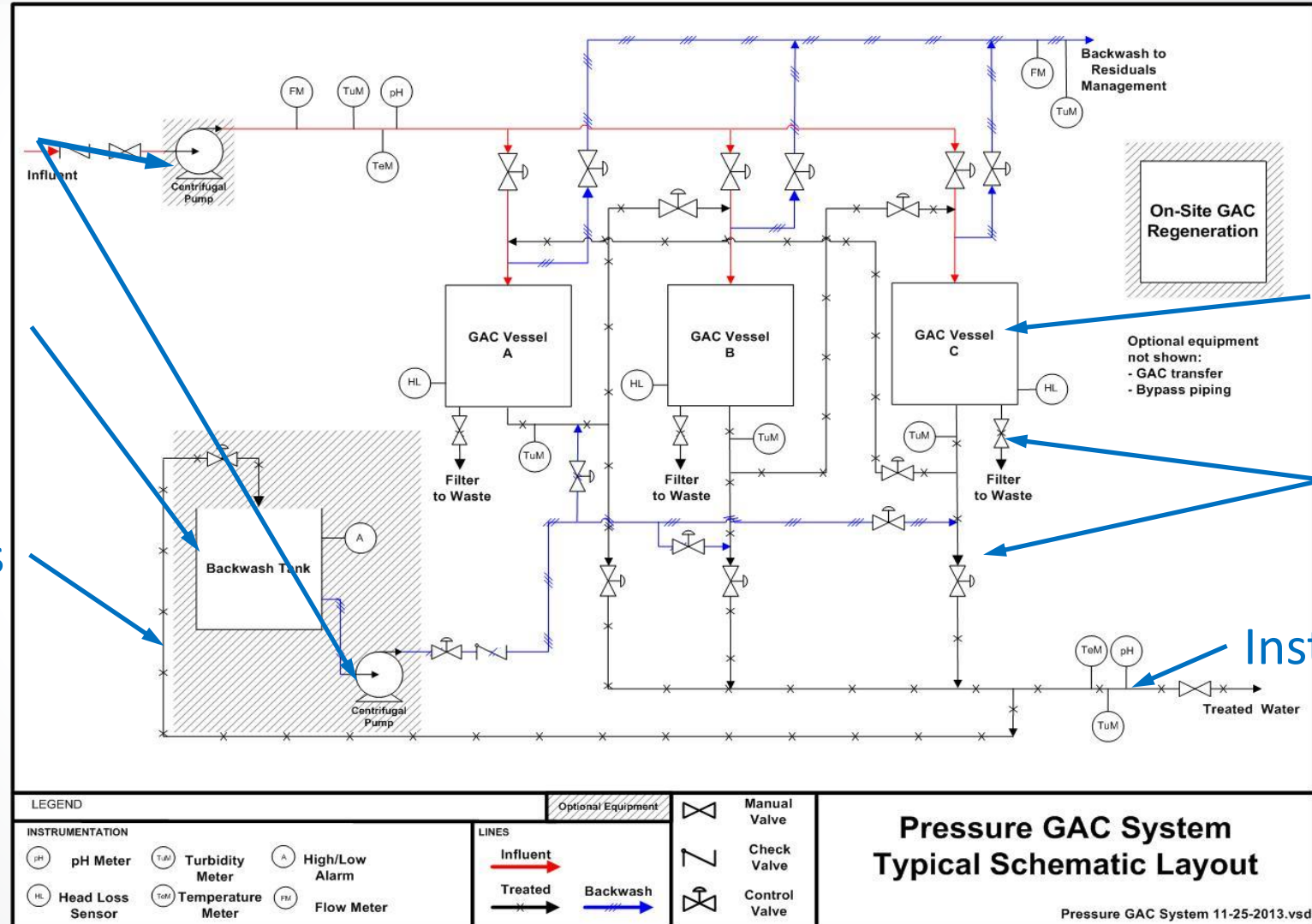
## Work Breakdown Structure (WBS) Approach?

A treatment technology is broken down into discrete components that can be measured to estimate costs. Components include specific equipment (e.g., tanks, vessels, pipes, instruments) and identifiable cost elements such as annual labor expenses, chemicals, and energy.

Pumps

Tanks

Pipes



Pressure Vessels

Valves

Instruments



# What Costs Do the WBS Models Estimate?

## Capital Costs

- Equipment costs
  - Pumps
  - Tanks/vessels
  - Pipes
  - Instruments
- Buildings
- Add-on costs
  - Pilot study
  - Permits
  - Land
- Indirect costs
  - Engineering
  - Construction management
  - Sitework/electrical

## Annual Operating Costs

- Labor
  - Technical
  - Managerial
  - Administrative
- Materials and supplies
  - Chemicals
  - Equipment maintenance
- Residuals management
  - Publicly owned treatment work (POTW)
  - Granular Activated Carbon (GAC) regeneration
  - RCRA Subtitle D or C landfill
- Energy
  - Operating (e.g., pumps, blowers)
  - Heating, ventilation, and air conditioning (HVAC)

- Adsorptive media
- **Anion exchange\***
- **Biological treatment\***
- Cation exchange
- **GAC\***
- Greensand filtration
- Microfiltration/ultrafiltration
- Multi-stage bubble aeration\*



- Non-treatment
- Packed tower aeration
- **POU/POE (Point of Entry) #**
- Reverse Osmosis/Nanofiltration
- UV disinfection
- UV Advanced Oxidation

\* **Search: EPA WBS** <http://www2.epa.gov/dwregdev/drinking-water-treatment-technology-unit-cost-models-and-overview-technologies>

# **For POU/POE search: EPA small system compliance help**  
<http://water.epa.gov/type/drink/pws/smallsystems/compliancehelp.cfm>



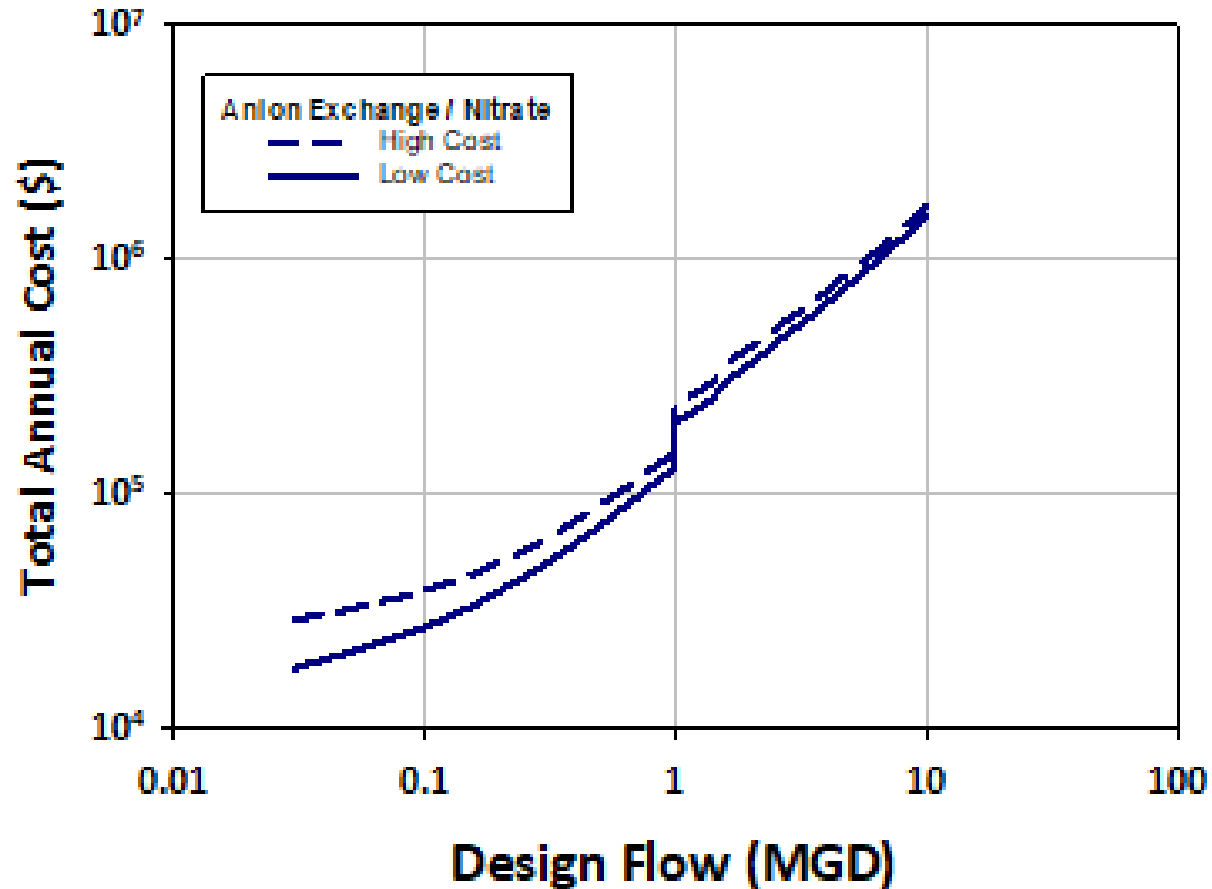
## Why Nitrate and Perchlorate?

- Nitrate: A number of utilities exceed the nitrate Maximum Contaminant Level (MCL), particularly small systems
- Perchlorate: New state regulations and federal regulation consideration
- Both are fully oxidized – oxidation processes including aerobic biotreatment will not work
- The treatment processes that will work are pretty much the same
  - Anion exchange resin
  - High pressure membranes: reverse osmosis or nanofiltration
  - Anaerobic biological treatment (novel technology)



# Cost: Nitrate/Anion Exchange

Typical cost curve with high and low cost



## Primary Assumptions

- 20.3 mg N/L Influent
- Nitrate selective resin
- 420 Bed volumes before regeneration
- 2 minute Empty Bed Contact Time (EBCT)
- Parallel contactors
- Brine discharge to POTW

## Specific Design Modifications for Smaller Systems within the Cost Model



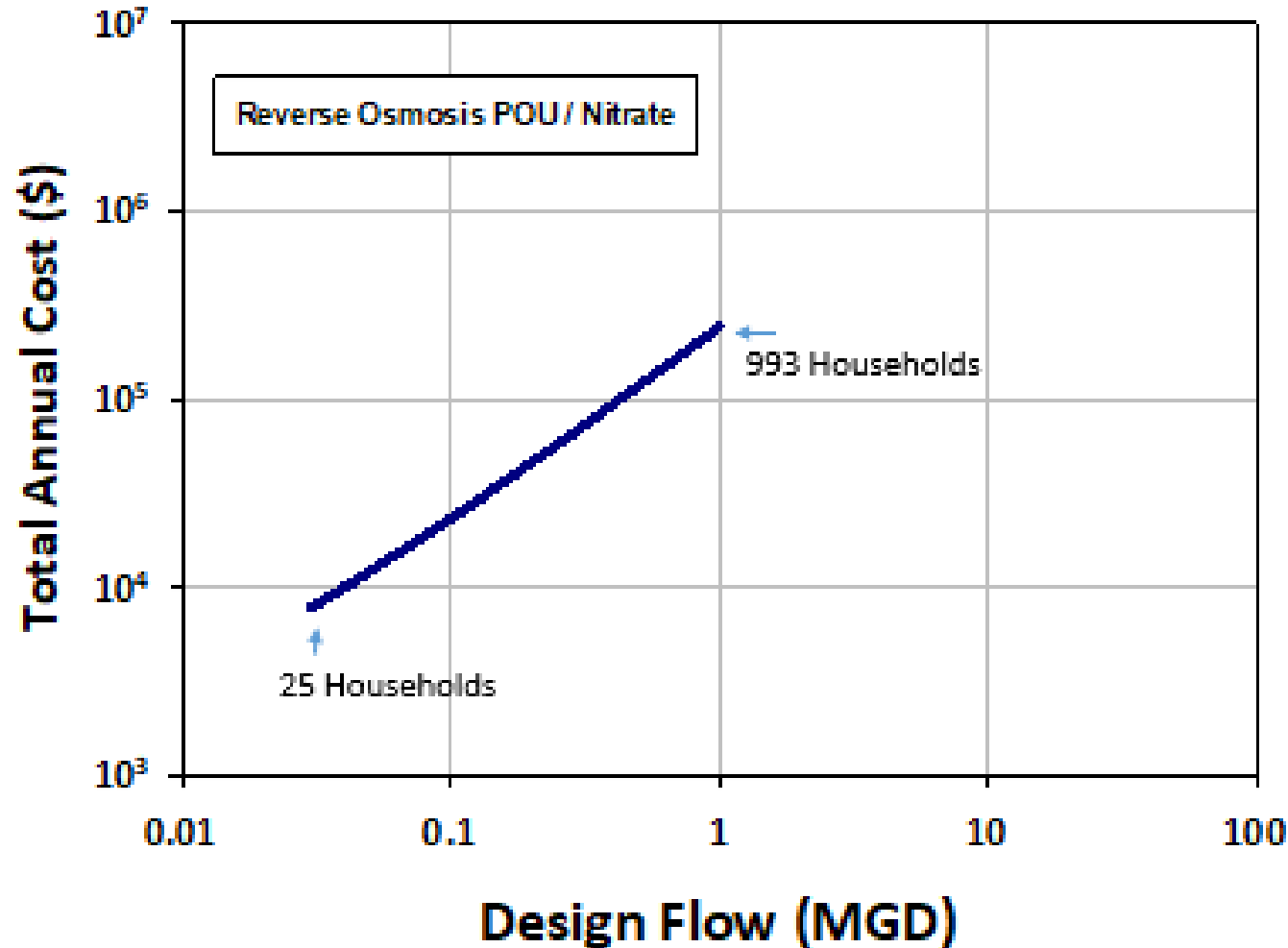
*(Considers flows under 1 MGD)*

- Construction issues (building)
- Residual handling flexibility
- Reduced spacing between vessels
- Smaller and no redundant vessels
- Reduced instrumentation
- No booster pumps
- No backwash pumps
- Reduced concrete pad thickness
- Reduced indirect costs



# Cost: Nitrate / Point of Use

*Only for 1 MGD design flow and below*

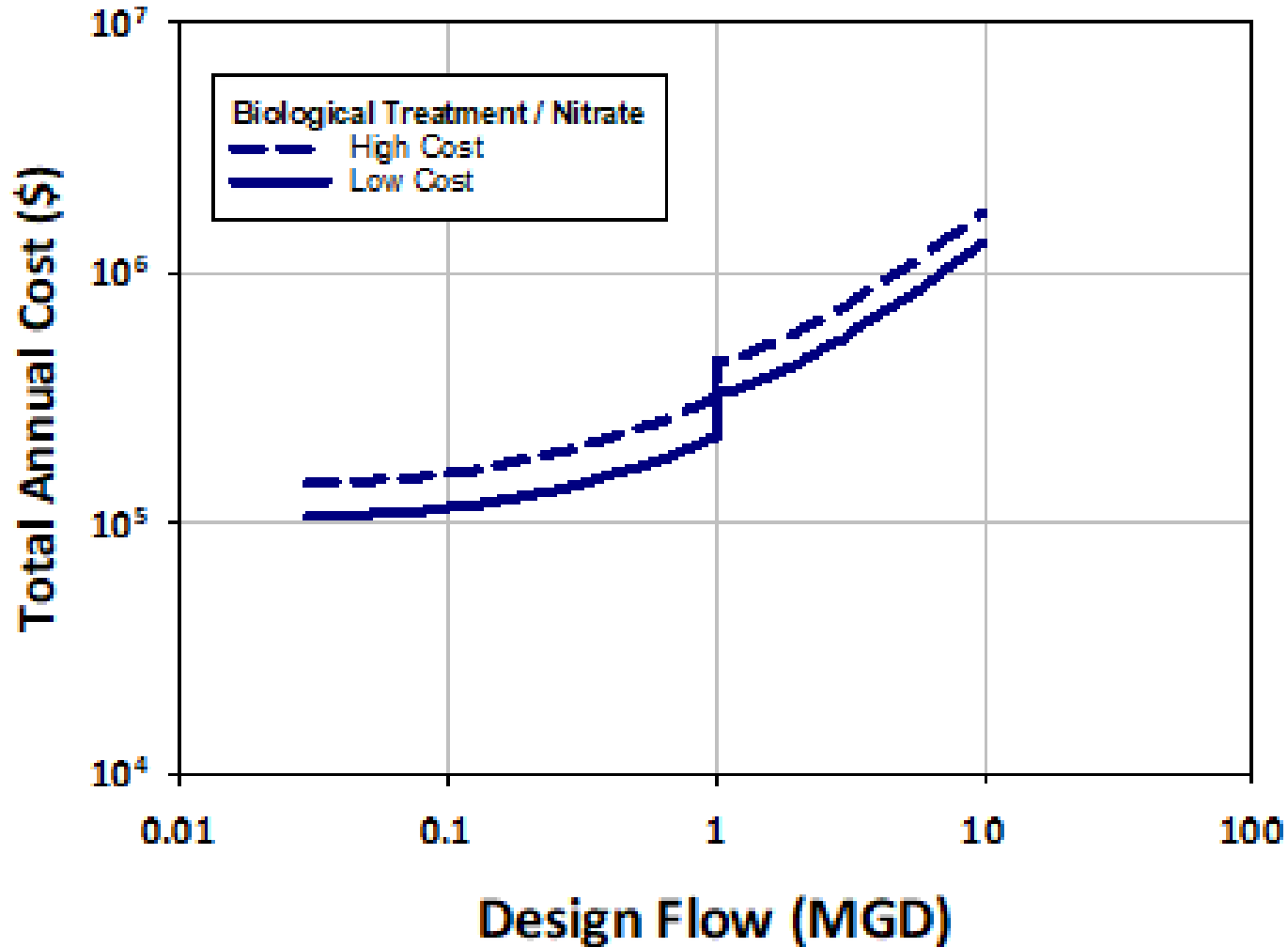


## Primary Assumptions

- 20.3 mg N/L Influent
- Reverse osmosis (RO) treatment
- Replacement frequency:
  - RO membrane: 3 years
  - Pre filters: 9 months
  - Post filter: 12 months
- Groundwater
- No post UV disinfection



# Cost: Nitrate / Anaerobic Biological Treatment

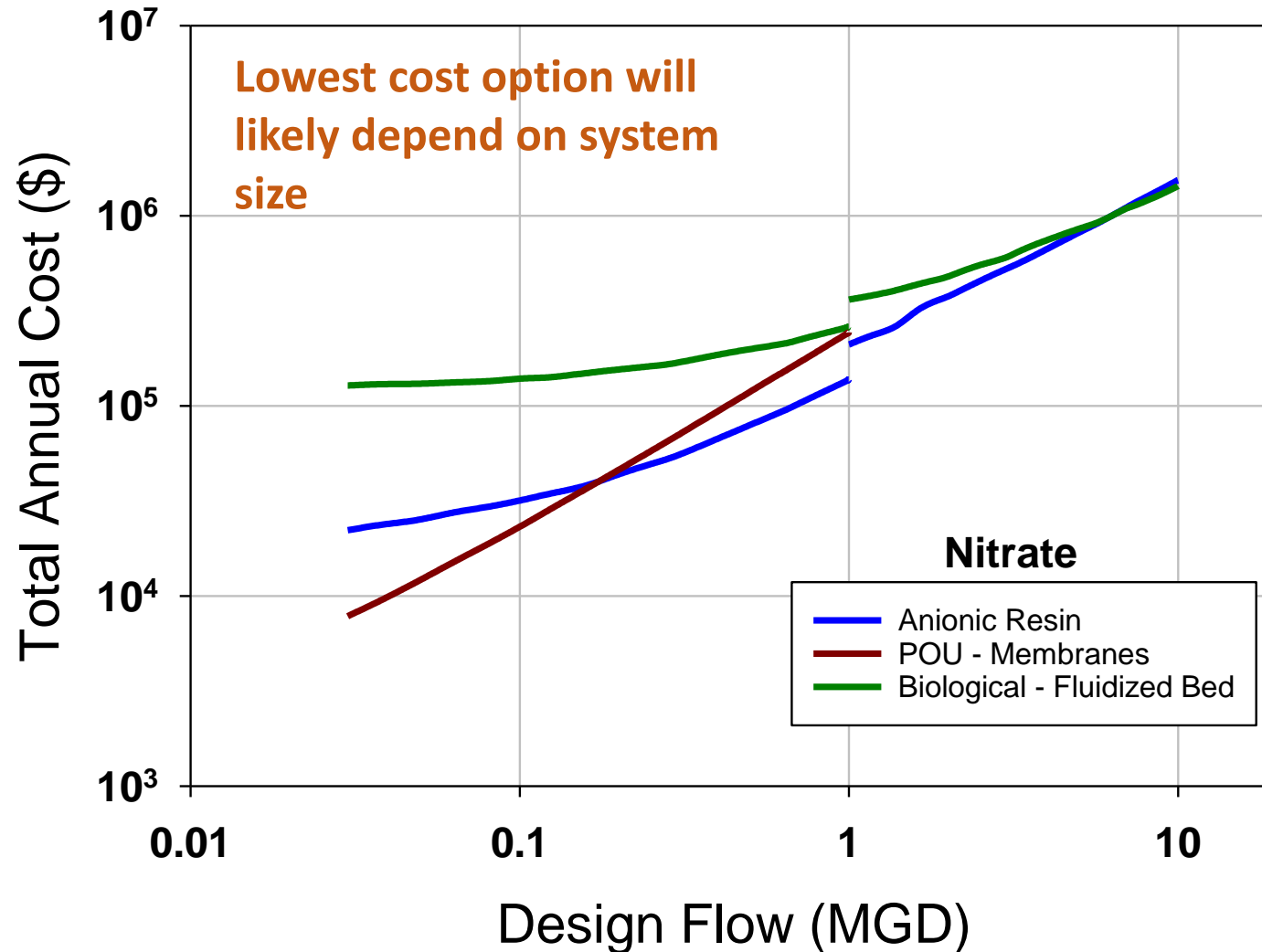


## Primary Assumptions

- 20.3 mg N/L
- Fluidized bed reactor
- 28.5 mg/L acetic acid
- 2 mg P/L phosphoric acid
- 10 minute EBCT
- Post treatment aeration
- Post treatment filtration
- Recycle of spent backwash



# Cost: Nitrate (combined)



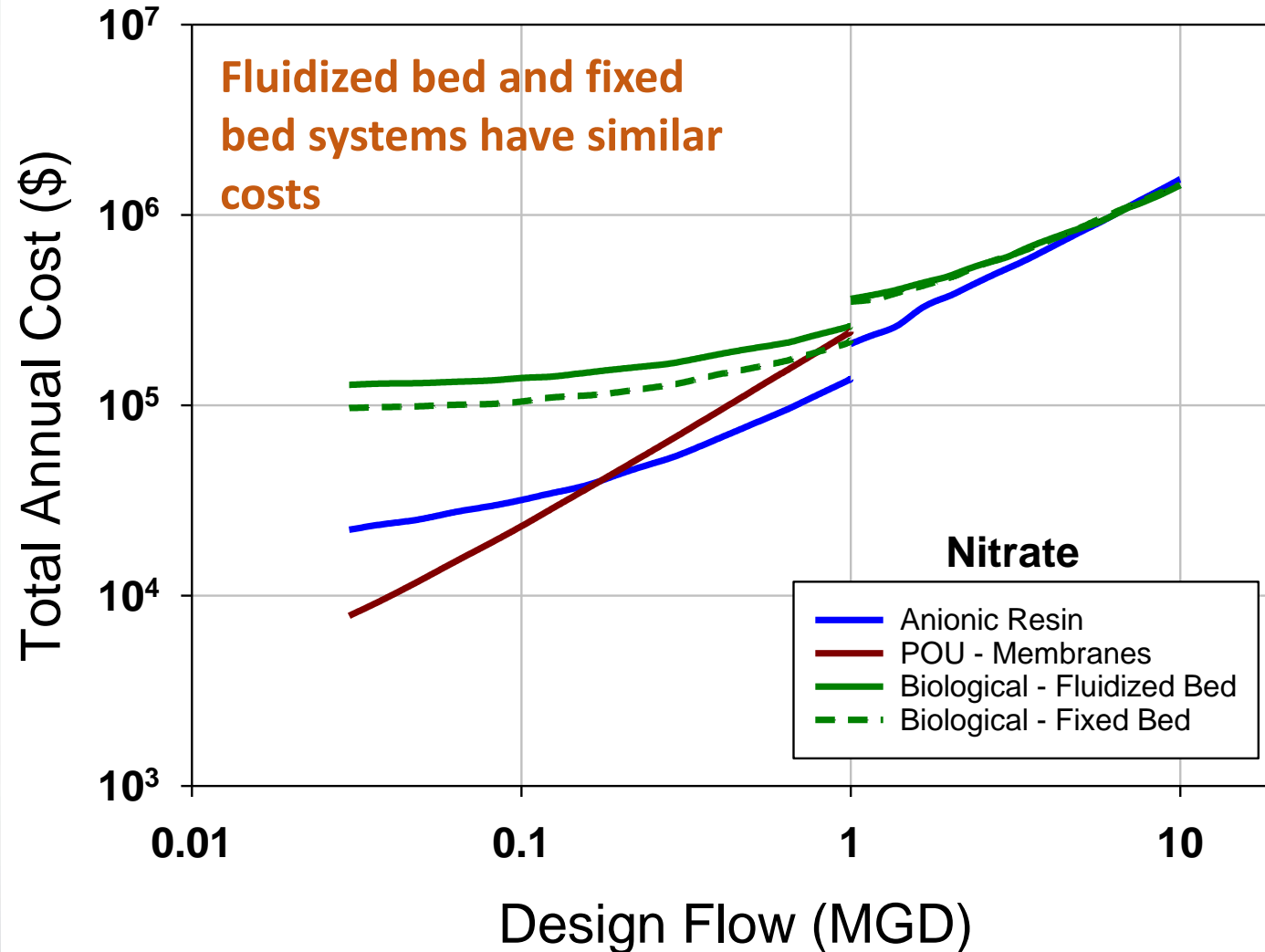
## Conditions Same as Previous Slides:

- Medium cost option
- Influent 20.3 mg N/L
- Groundwater
- Ion Exchange (IEX): Nitrate selective
- Biological: Fluidized bed
- POU: Reverse Osmosis



# Cost: Nitrate (combined)

*Includes both fluidized bed and fixed bed for anaerobic biological treatment*



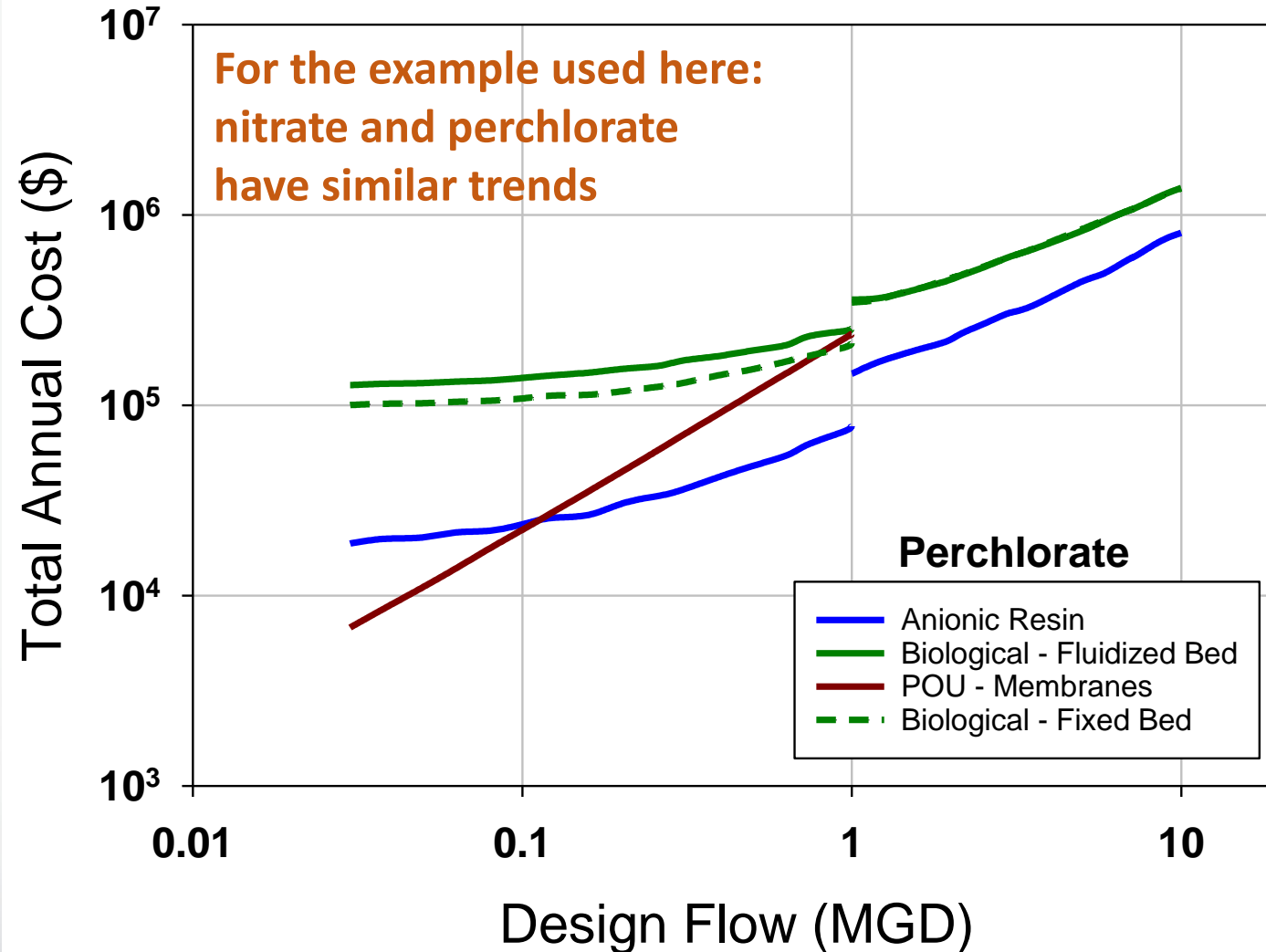
### Conditions Same as Previous Slides:

- Medium cost option
- Influent 20.3 mg N/L
- Groundwater
- IEX: Nitrate selective
- Biological: Fluidized bed
- POU: Reverse Osmosis



# Cost: Perchlorate (combined)

*Includes both fluidized bed and fixed bed for anaerobic biological treatment*



## Conditions Same as Previous Slides:

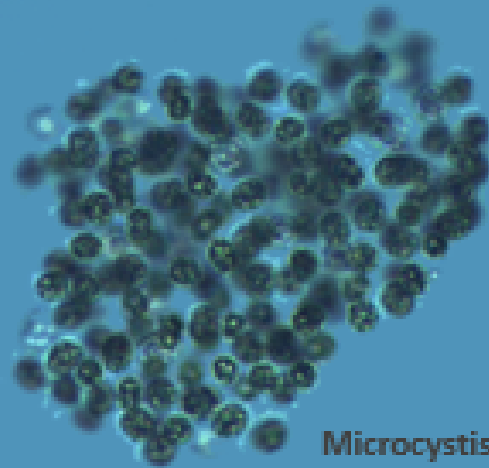
- Medium cost option
- Influent 24 ug/L
- Groundwater
- IEX: Perchlorate selective
- Biological: Fluidized & fixed bed
- POU: Reverse Osmosis



## Toxin within the cell and those that are dissolved require different treatment processes

### Particulates (toxin in cell)

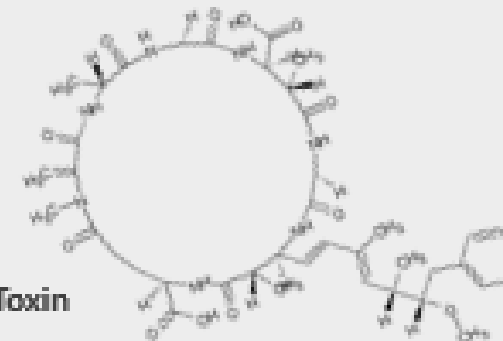
- Solids removal processes effective
- Do not want to lyse cell or toxin will be released



Microcystis (cells)

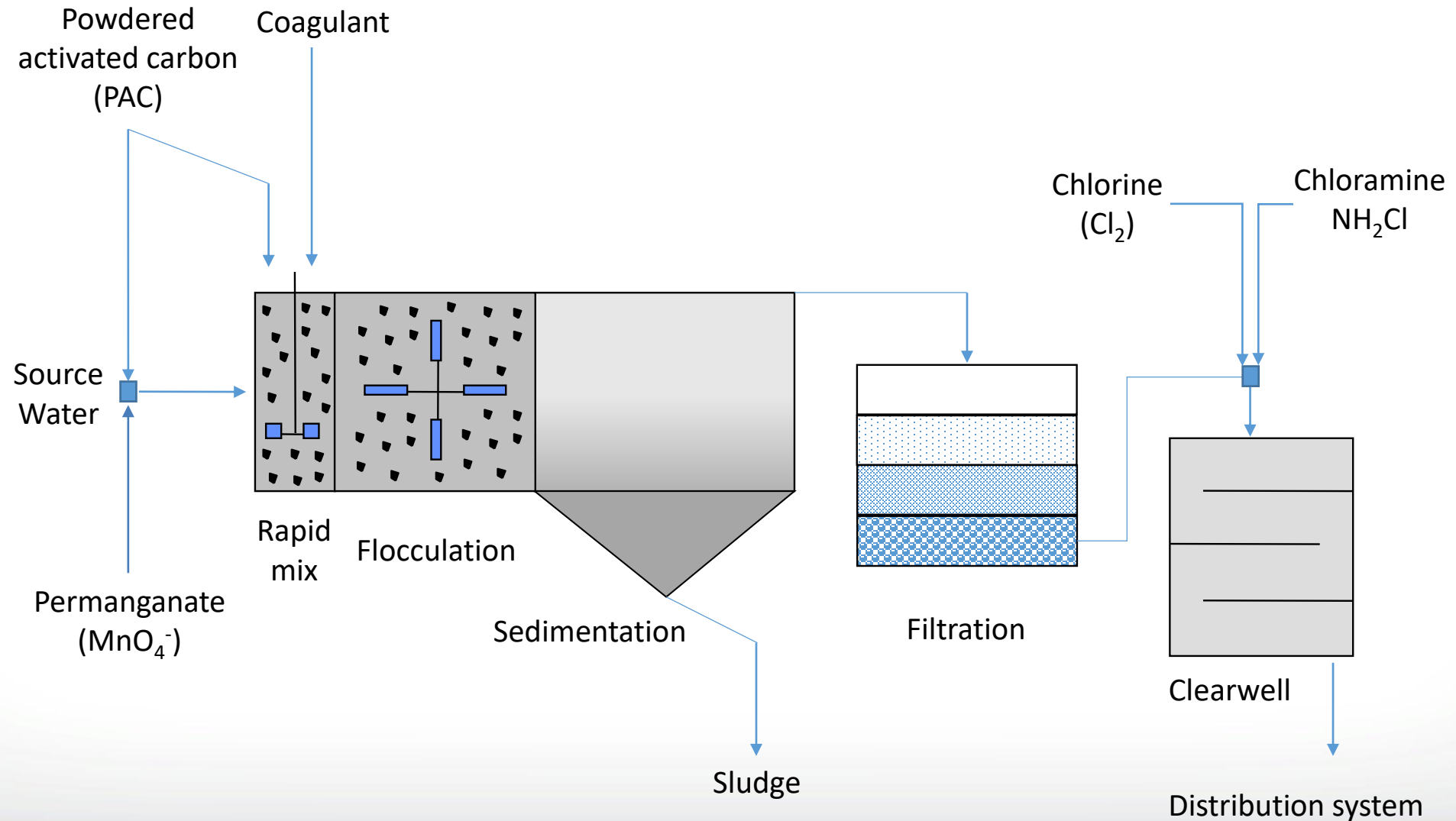
### Dissolved (toxin released from cell)

- Solids removal processes ineffective
- Typical disinfectants may not be effective enough (e.g., permanganate, chlorine)
- More effective treatments are expensive and plants typically do not have them in place (e.g., GAC, Ozone)

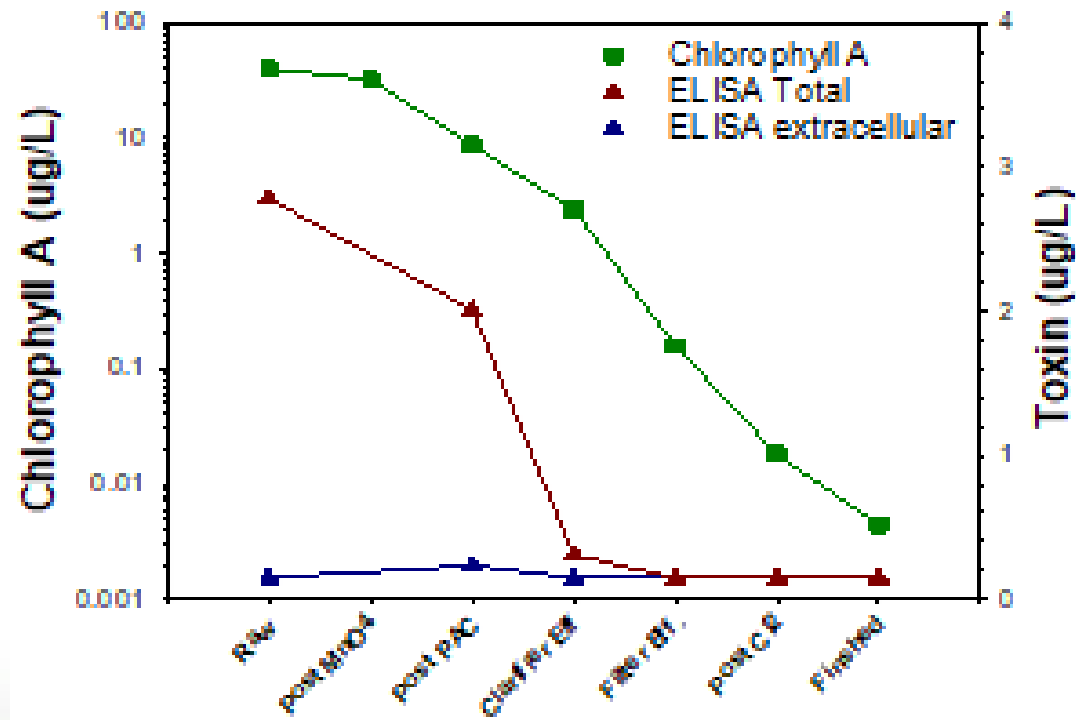


Microcystin Toxin

# Typical Treatment Train



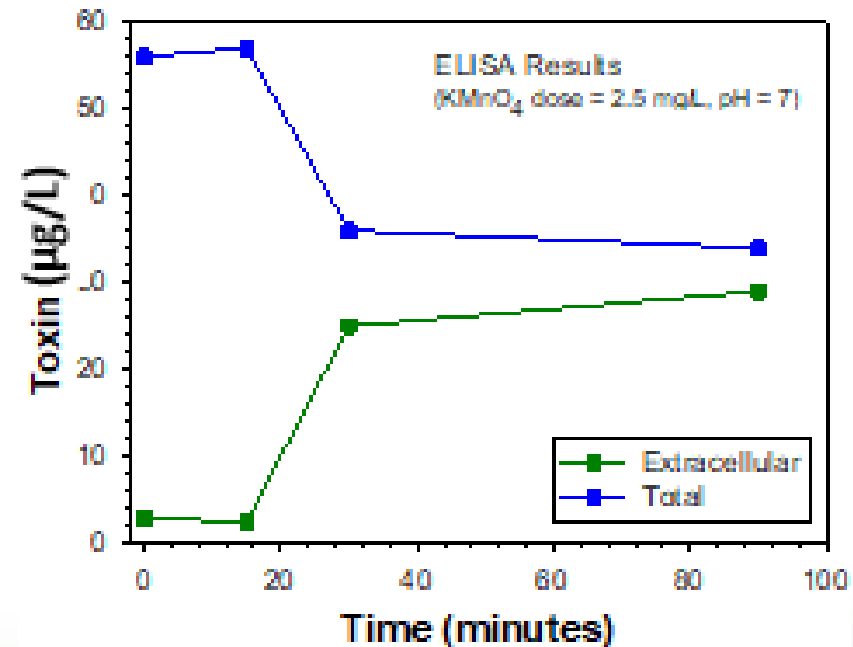
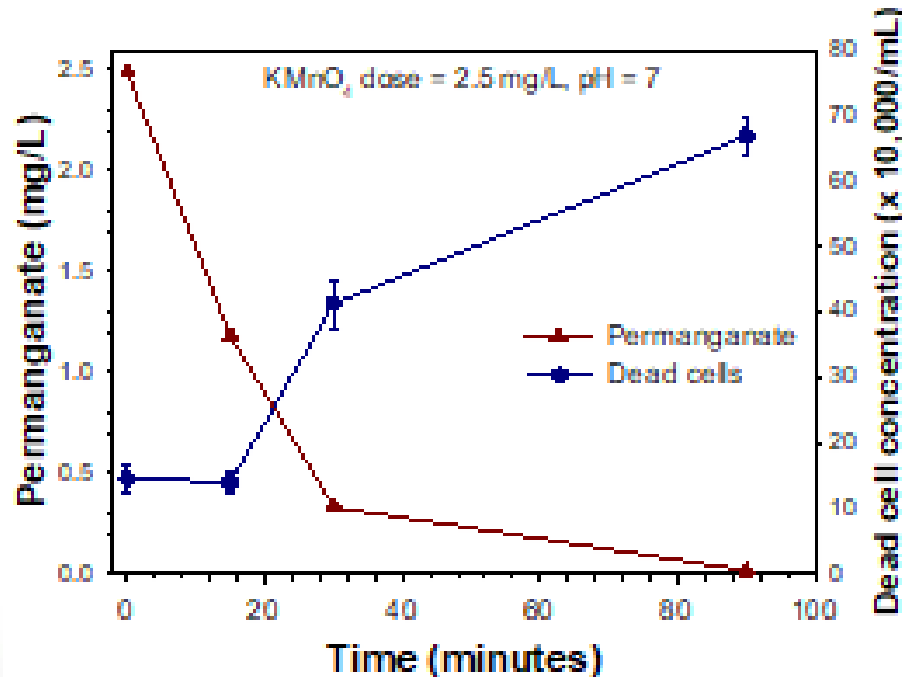
If toxin remains in the cell, most of it is removed before the filter



# Effect of Permanganate

Inactivates cells

Releases toxins into solution while at the same time destroying them



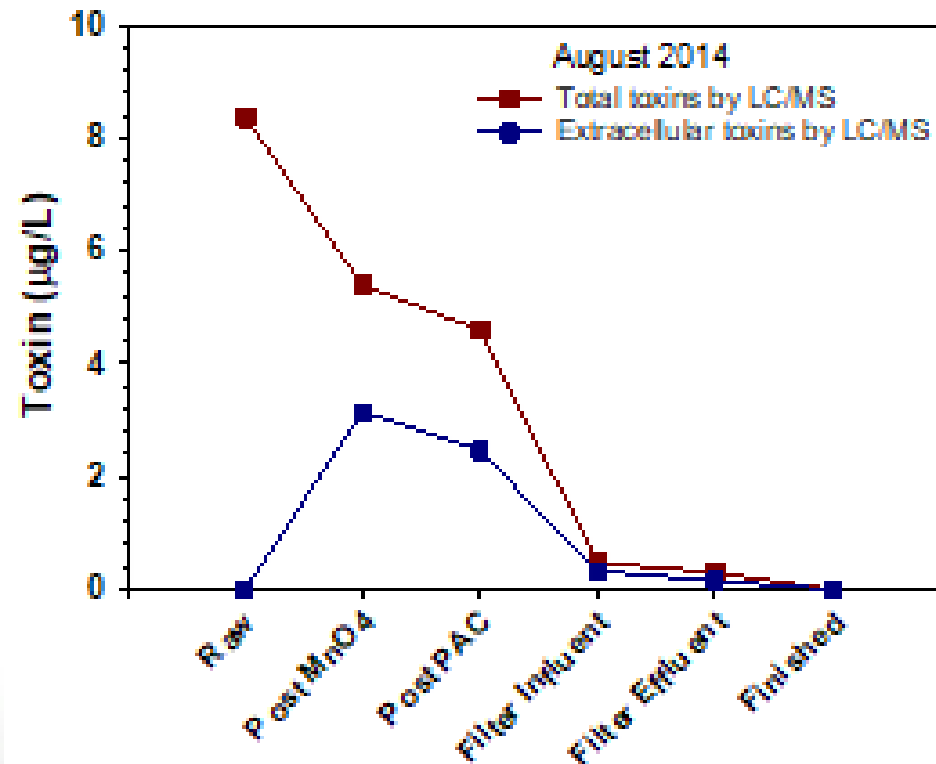


# Through Treatment (Microcystin Toxin)

**Permanganate reducing total and increasing extracellular toxin**

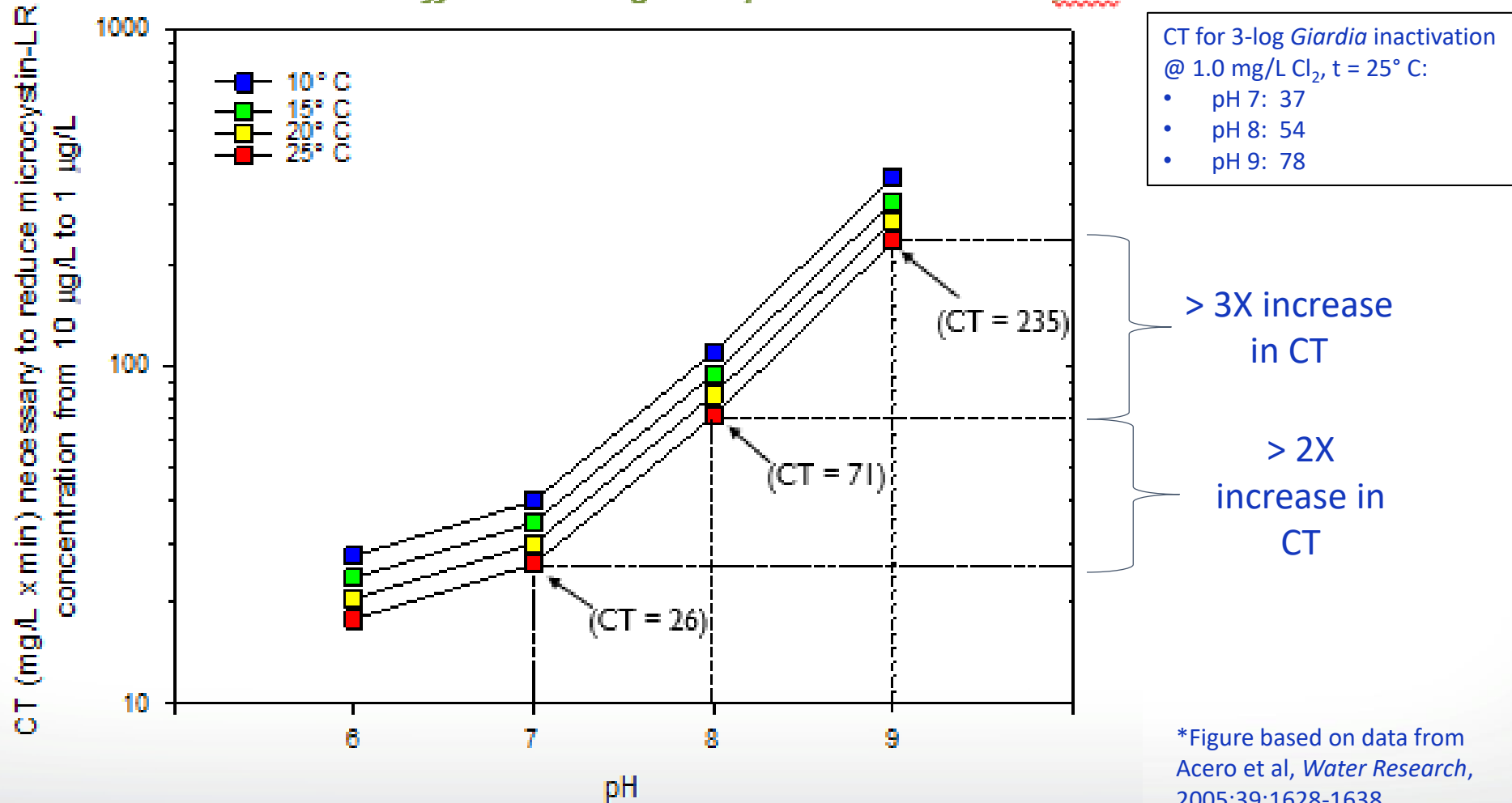
**Powdered activated carbon reducing the extracellular toxin**

**Particulate removal removes the intracellular toxin**



# Impact of Chlorination

Chlorination most effective at high temperatures and low pHs



## Powdered Activated Carbon (PAC)

Removes some harmful algal bloom (HAB) toxins better than others  
Carbon choice  
Choosing the correct dose quickly  
Reduced filter times and sludge disposal

## Granular Activated Carbon (GAC)

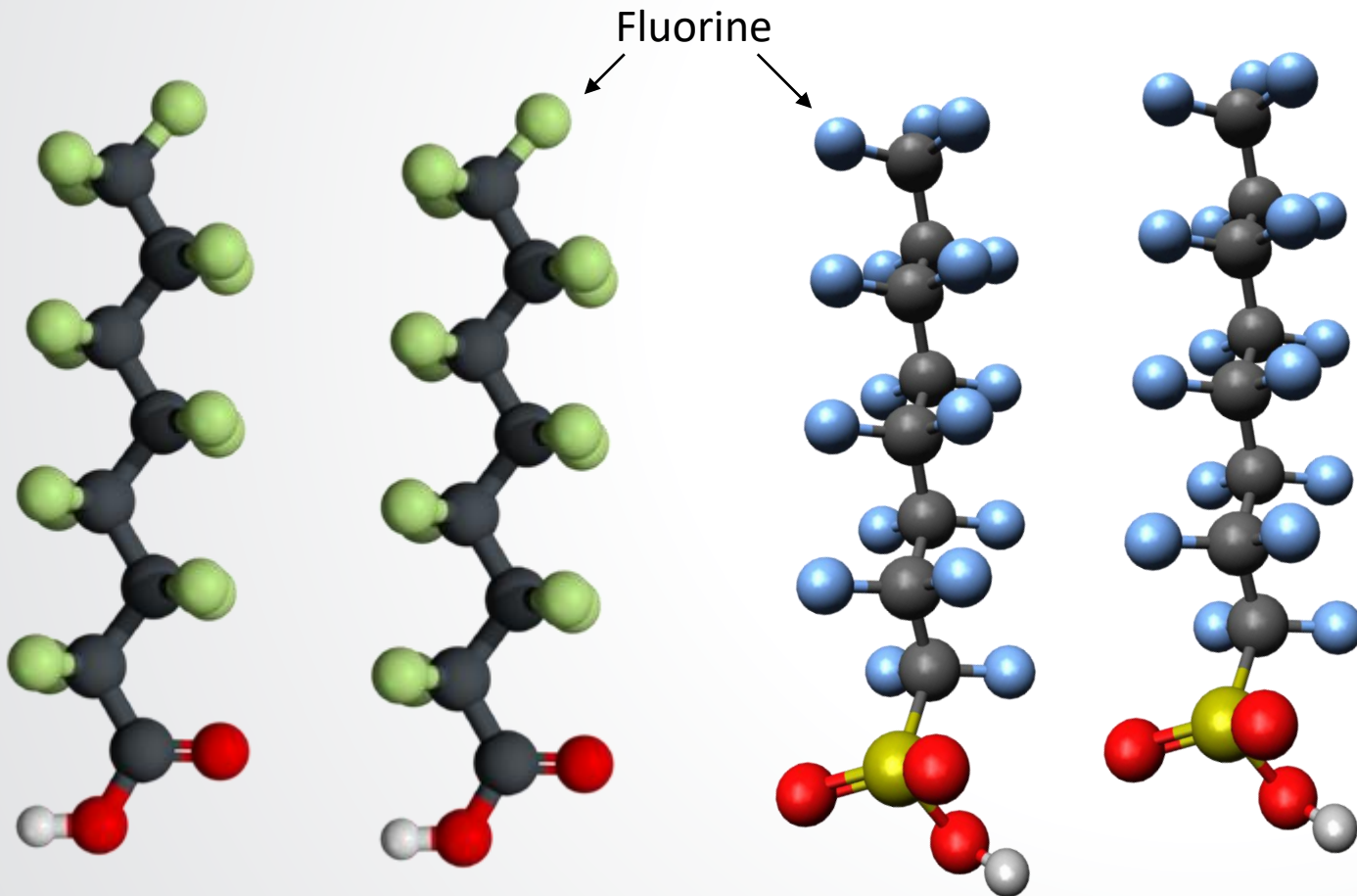
Removes some HAB toxins better than others  
Removal depends on amount of preloading  
High capital cost  
Reactivation/removal frequency – cost and operation

## UV (After treatment)

Needed UV doses are much higher than that required for 2-log disinfection of *Cryptosporidium* = 5.8 mJ/cm<sup>2</sup>, *Giardia* = 5.2 mJ/cm<sup>2</sup>, viruses = 100 mJ/cm<sup>2</sup>.

- Permanganate** Applied early in the treatment process where concentrations of cyanobacterial cells in are still high – potential to stimulate toxin release
- Chlorine** Degradation rate increases significantly with lower pH – need to balance corrosion compliance
- Ozone** High capital cost  
If applied fairly early in treatment - potential for toxin release
- Chlorine Dioxide** Not considered effective against microcystins





*Perfluorooctanoic acid (PFOA)*

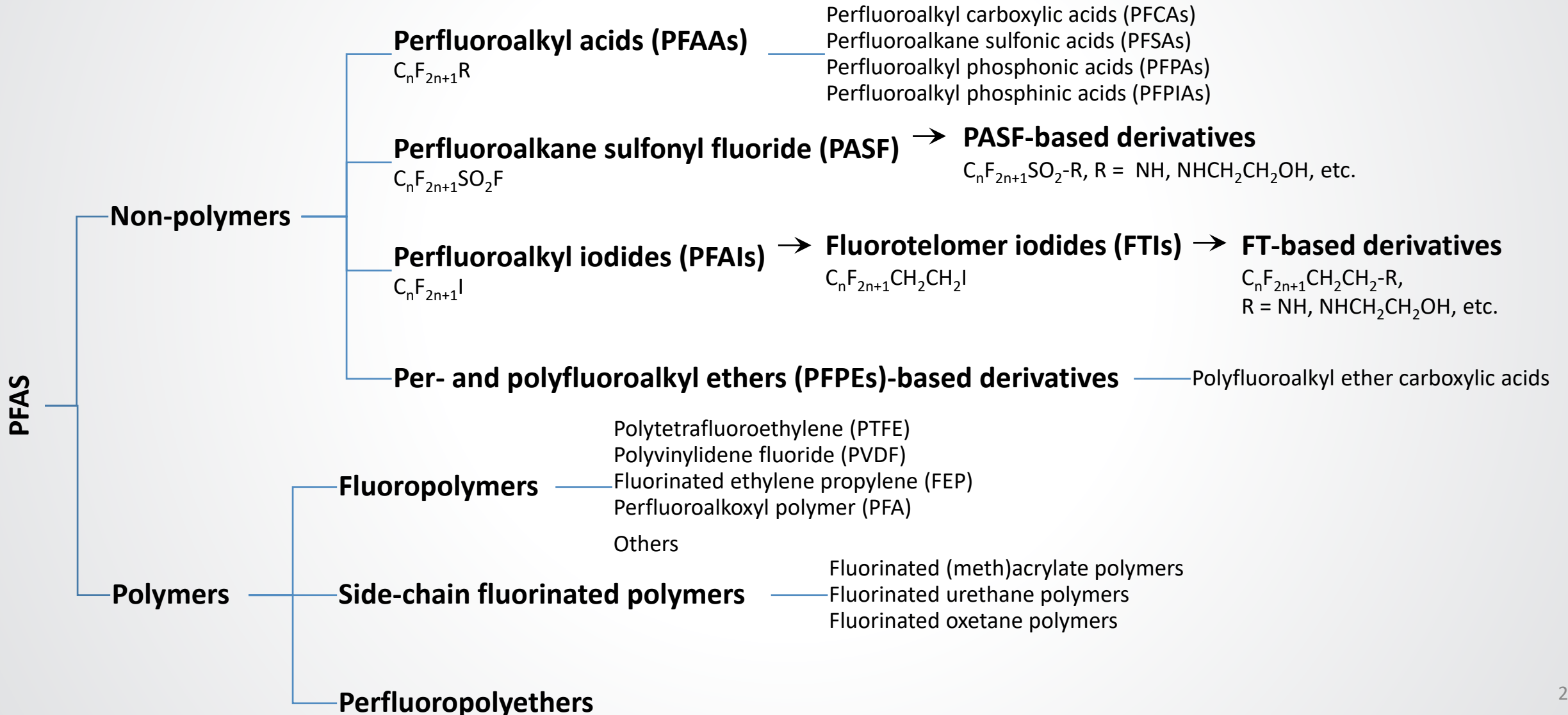
*Perfluorooctanesulfonic acid (PFOS)*

## ➤ A class of chemicals

- Chains of carbon (C) atoms surrounded by fluorine (F) atoms
  - Water-repellent (hydrophobic body)
  - Stable C-F bond
- Some PFAS include oxygen, hydrogen, sulfur and/or nitrogen atoms, creating a polar end



# Thousands of Chemicals: More Than Just PFOA and PFOS



- **Problem:** Utilities lack treatment technology cost data for PFAS removal
- **Action:**
  - Gather performance and cost data from available sources (DOD, utilities, industry, etc.)
  - Conduct EPA research on performance of treatment technologies including home treatment systems
  - Update EPA's Treatability Database and Unit Cost Models
  - Connect EPA's Treatability Database to EPA's Unit Cost Models for ease of operation
  - Model performance and cost, and then extrapolate to other scenarios
    - Variable source waters
    - Variable PFAS concentrations in source water
    - Different reactivation/disposal options
    - Document secondary benefits
    - Address treatment impact on corrosion
  - Evaluate reactivation of granular activated carbon
- **Impact:** Enable utilities to make informed decisions about cost-effective treatment strategies for removing PFAS from drinking water





# Drinking Water Treatment for PFOS

## Ineffective Treatments

- Conventional Treatment
- Low Pressure Membranes
- Biological Treatment (including slow sand filtration)
- Disinfection
- Oxidation
- Advanced oxidation

### PAC Dose to Achieve

50% Removal	16 mg/l
90% Removal	>50 mg/L

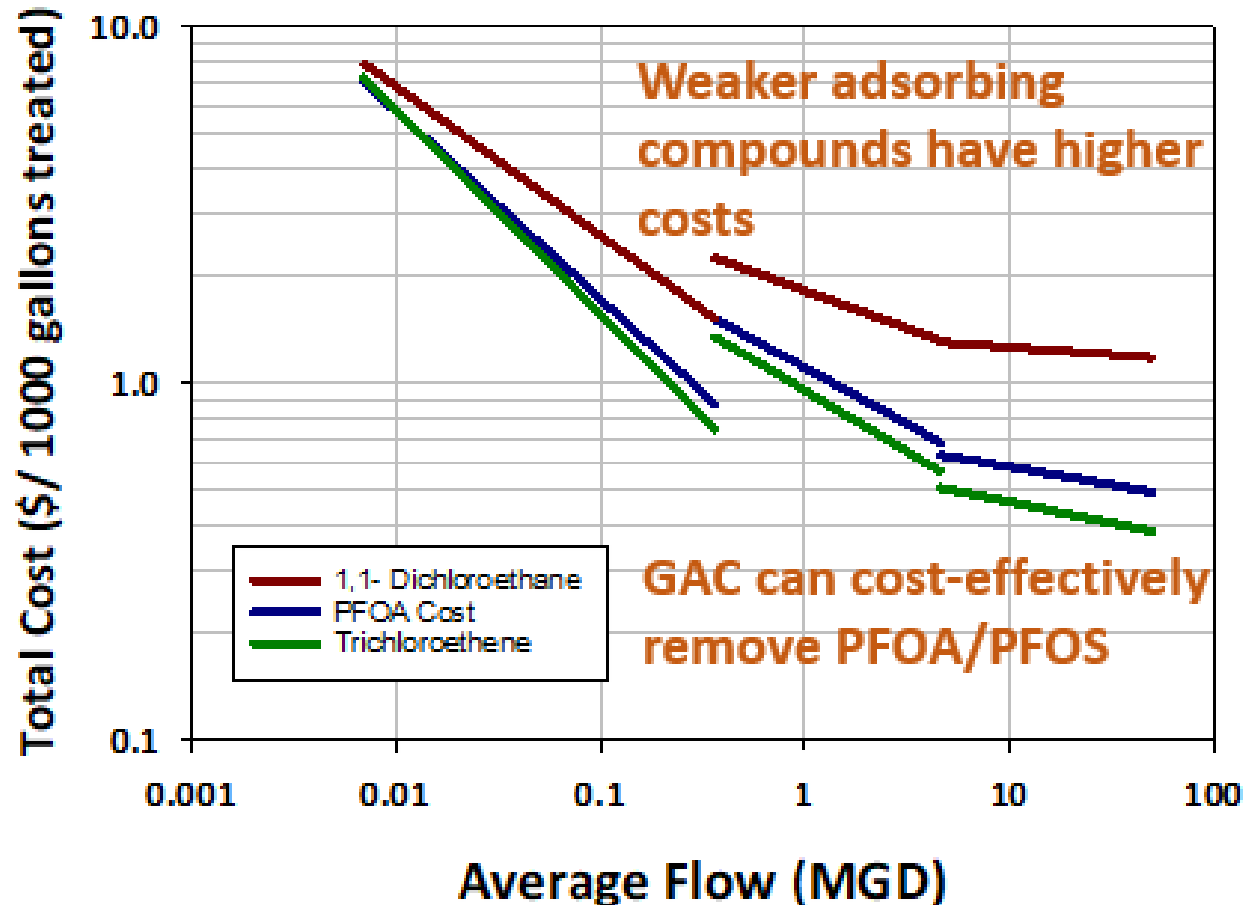
*Dudley et al., 2015*

## Effective Treatments

	Percent Removal	
Anion Exchange Resin (IEX)	90 to 99	- <b>Effective</b>
High Pressure Membranes	93 to 99	- <b>Effective</b>
Powdered Activated Carbon (PAC)	10 to 97	- <b>Effective for only select applications</b>
Granular Activated Carbon (GAC)		
Extended Run Time	0 to 26	- <b>Ineffective</b>
Designed for PFAS Removal	> 89 to > 98	- <b>Effective</b>



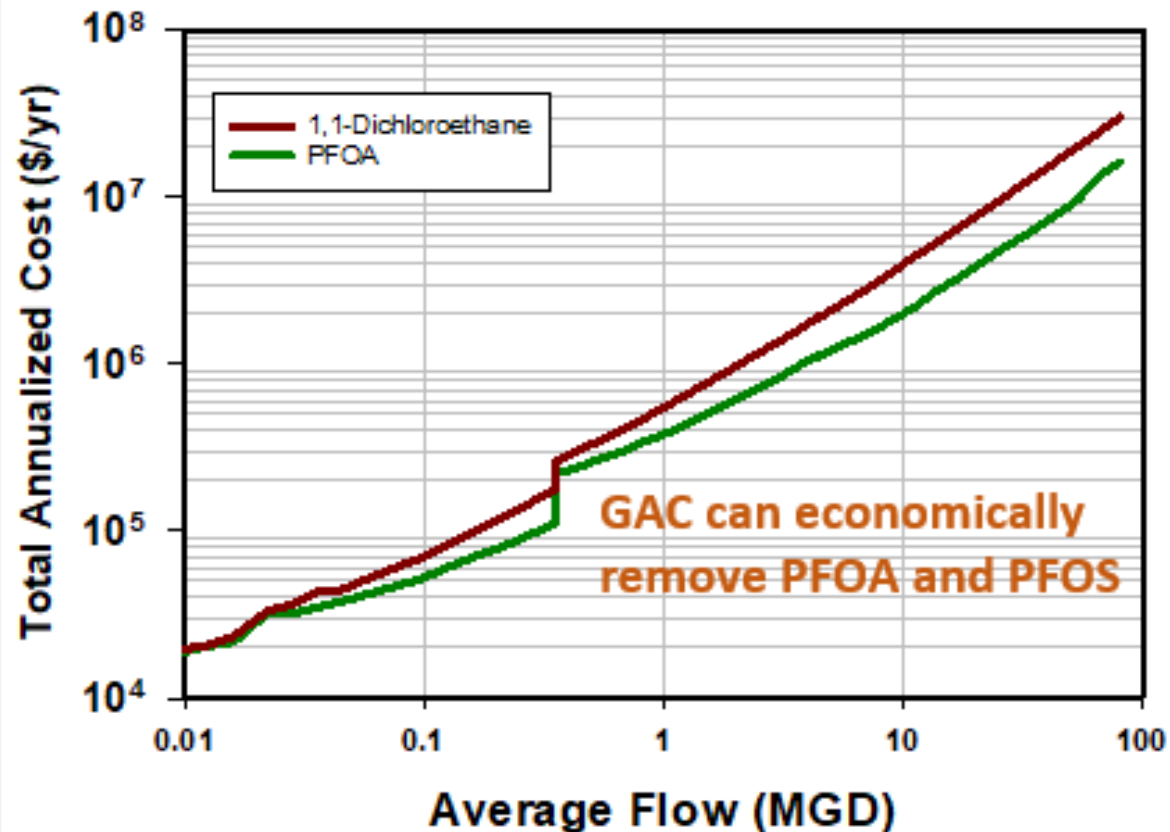
# GAC Treatment Cost: PFOA, TCE, 11 DCA



**EPA will be evaluating additional water qualities and designs**

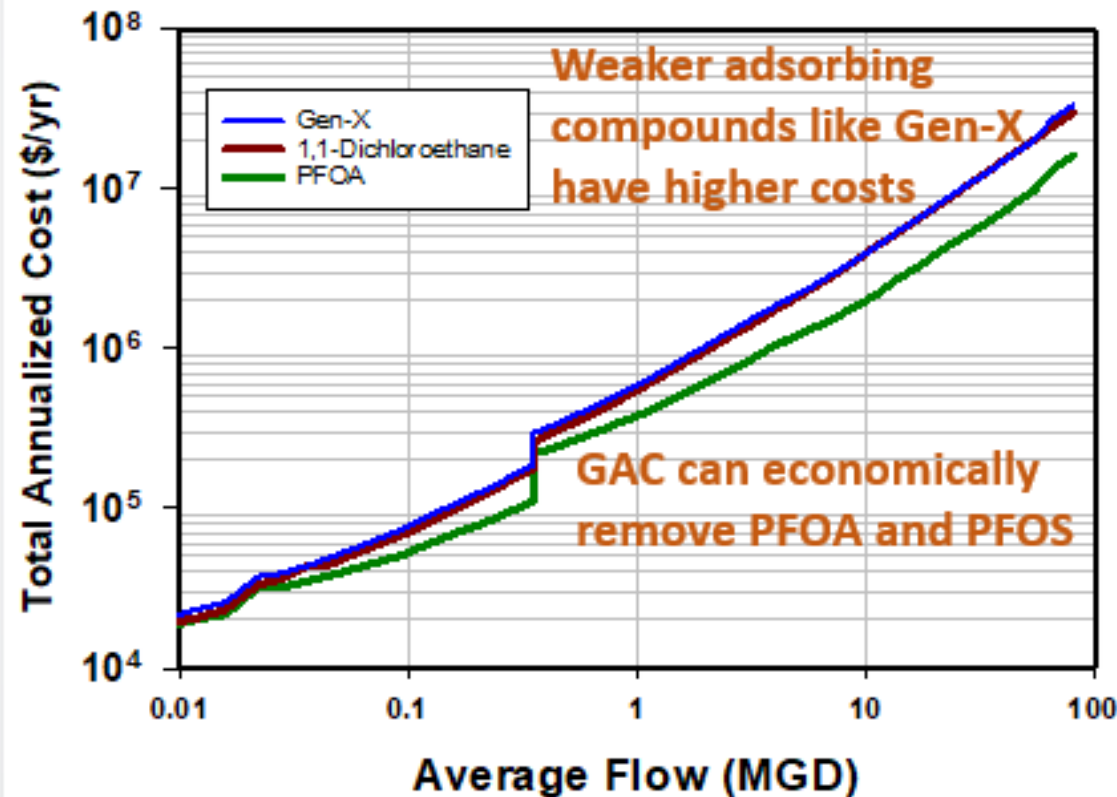
- Full Scale
- 26 min EBCT
- Lead-Lag configuration
- F600 Calgon carbon
- 1.5 m<sup>3</sup>/min flow
- Full automation
- POTW residual discharge
- Off site regeneration
- 135,000, 70,000, and 11,000 bed volumes to breakthrough for trichloroethylene (TCE), PFOA, and 11DCA, respectively.

# Costs for Additional PFAS



- Pilot Scale Performance Data
- 20 min EBCT
- F400 Calgon carbon
- Full automation
- POTW residual discharge
- Off site regeneration
- 31,000, 7,100, and 5,560 bed volumes to breakthrough for PFOA, Gen-X, and 11-DCA, respectively.

# Cost for Additional PFAS

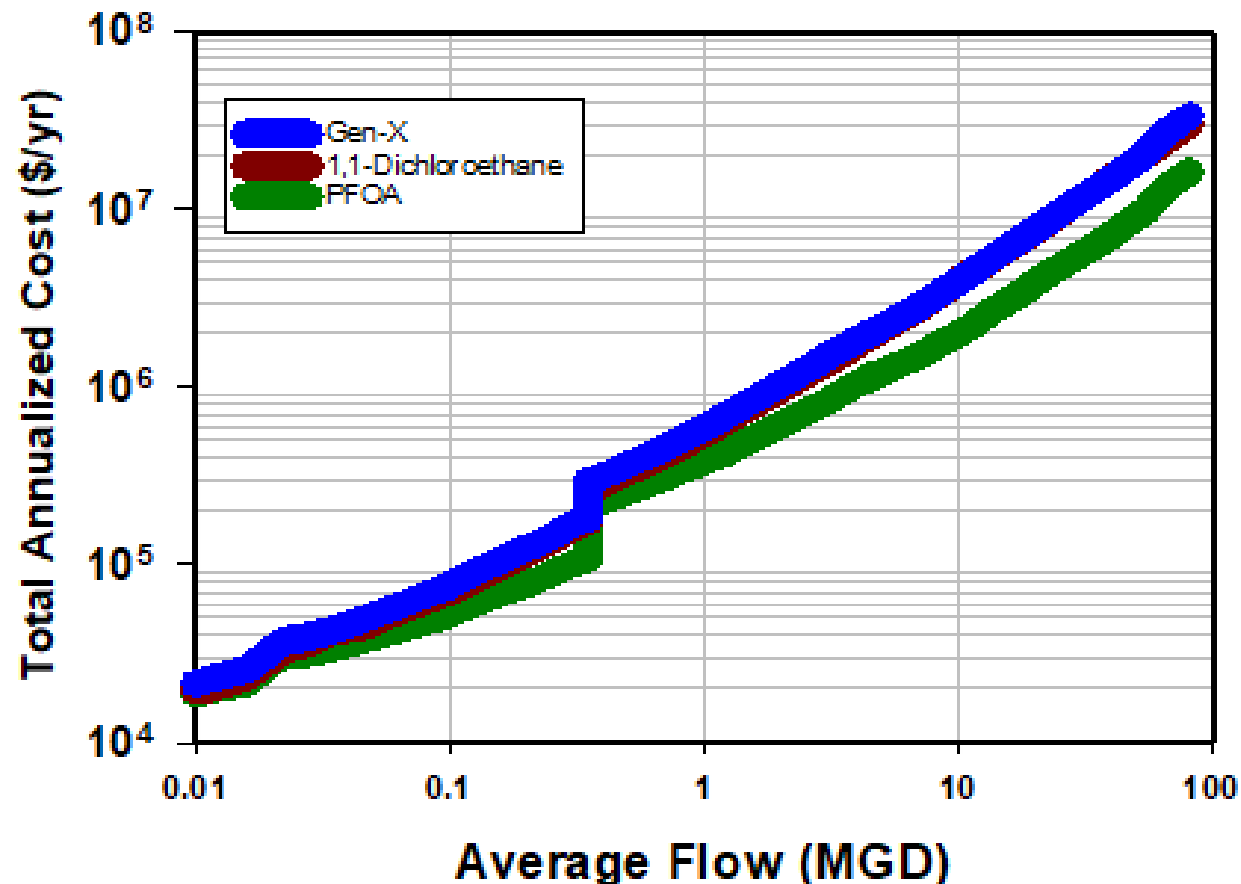


- Pilot Scale Performance Data
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- Full automation
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# Cost for Additional PFAS

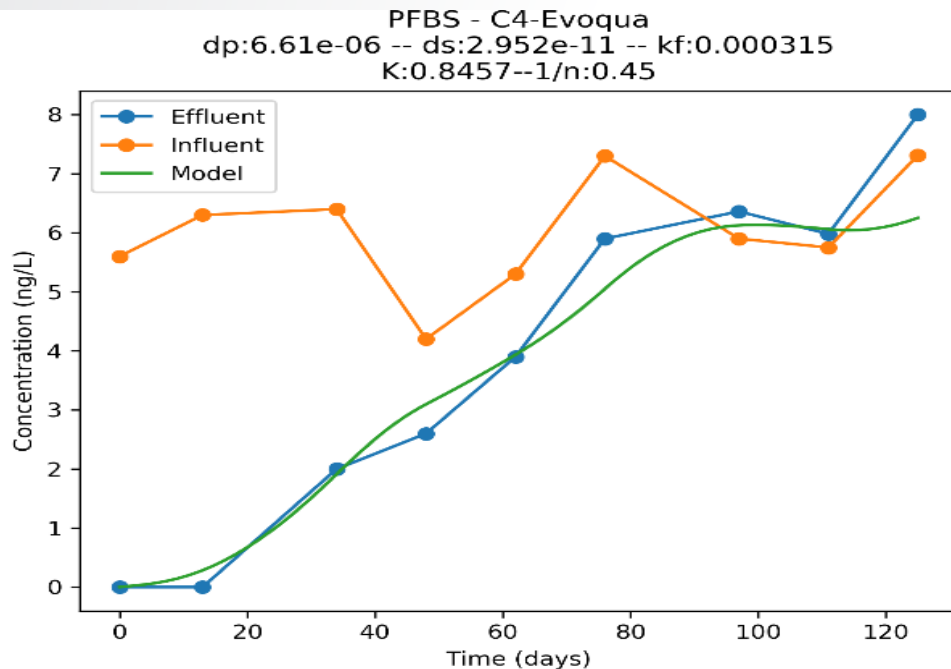
Compounds will have a range of costs depending on water quality and other factors that impact design and operation



- Pilot Scale Performance Data
- 20 min EBCT
- F400 Calgon carbon
- Full automation
- POTW residual discharge
- Off site regeneration
- 31,000, 7,100, and 5,560 bed volumes to breakthrough for PFOA, Gen-X, and 11-DCA, respectively.



- Fitting pilot- or full-scale data



- Predicting Results for Consistent Design
  - Allows for comparison across technologies by cost
- Allows for Predicting other Scenarios
  - Other designs: Number of contactors, contactor EBCTs, different treatment goals, etc.
  - Other influent conditions: Changing concentrations of PFAS or background constituents, changing demand, etc.



# Advantages of Select Treatments

## **Granular Activated Carbon (GAC)**

**Most studied technology**

**Will remove 100% of the contaminants, for a time**

**Good capacity for some PFAS**

Will remove a significant number of disinfection byproduct precursors

Will help with maintaining disinfectant residuals

Will remove many co-contaminants

Likely positive impact on corrosion (lead, copper, iron)

## **Anion Exchange Resin (PFAS selective)**

**Will remove 100% of the contaminants, for a time**

**High capacity for some PFAS**

**Smaller beds compared to GAC**

Can remove select co-contaminants

## **High Pressure Membranes (Reverse Osmosis or Nanofiltration)**

**High PFAS rejection**

Will remove many co-contaminants

Will remove a significant number of disinfection byproduct precursors

Will help with maintaining disinfectant residuals



# Issues to Consider

**EPA is evaluating these issues to document where and when they will be an issue**

**Granular Activated Carbon**  
(GAC)

**GAC run time for short-chained PFAS (shorter run times)**  
**Potential overshoot of poor adsorbing PFAS, if not designed correctly**  
**Reactivation/removal frequency**  
Disposal or reactivation of spent carbon

**Anion Exchange Resin**  
(PFAS selective)

**Run time for select PFAS (shorter run times)**  
**Overshoot of poor adsorbing PFAS, if not designed correctly**  
Unclear secondary benefits  
Disposal of resin

**High Pressure Membranes**  
(Reverse osmosis or  
Nanofiltration)

**Capital and operations costs**  
Membrane fouling  
Corrosion control  
Lack of options for concentrate stream treatment or disposal



# EPA PFAS Data and Tools

- Links to data and tools that include information related to PFAS and are available on EPA's website:

<https://www.epa.gov/pfas>

<https://www.epa.gov/pfas/epa-pfas-data-and-tools>

EPA PFAS Data and Tools

Environmental Topics Laws & Regulations About EPA Search EPA.gov

PFOA, PFOS and Other PFASs CONTACT US SHARE

PFAS Home  
Basic Information on PFAS  
EPA Actions  
PFAS Infographic  
**Data and Tools**  
State Information

## EPA PFAS Data and Tools

Below are links to data and tools that include information on PFAS and are currently available on the agency's website.

**Chemistry**

- [Chemistry Dashboard](#)
- [ChemView](#)

**Drinking Water**

- [Drinking Water Treatability Database](#)
  - [PFOA](#)
  - [PFOS](#)
- [Drinking Water Laboratory Methods](#)
- [Data from EPA's Third Unregulated Contaminant Monitoring Rule \(UCMR\)](#)

**Toxicity**

- [GenX Chemicals Studies](#)
- [Health & Environmental Research Online \(HERO\)](#)
- [Toxics Release Inventory](#)

**Waste**

- [Sampling and Laboratory Methods \(SW-486 Compendium\)](#)

[Contact Us](#) to ask a question, provide feedback, or report a problem.



# Drinking Water Goals

**For utilities that have a CEC in their source water at concentrations of health concern**

- 1) Eliminate source of the CECs to the source water
- 2) Either choose a new source of water or choose a **technology, design, and operational scheme** that will reduce the CECs to safe levels at the lowest possible cost in a **robust, reliable, and sustainable manner** that avoids unintended consequences

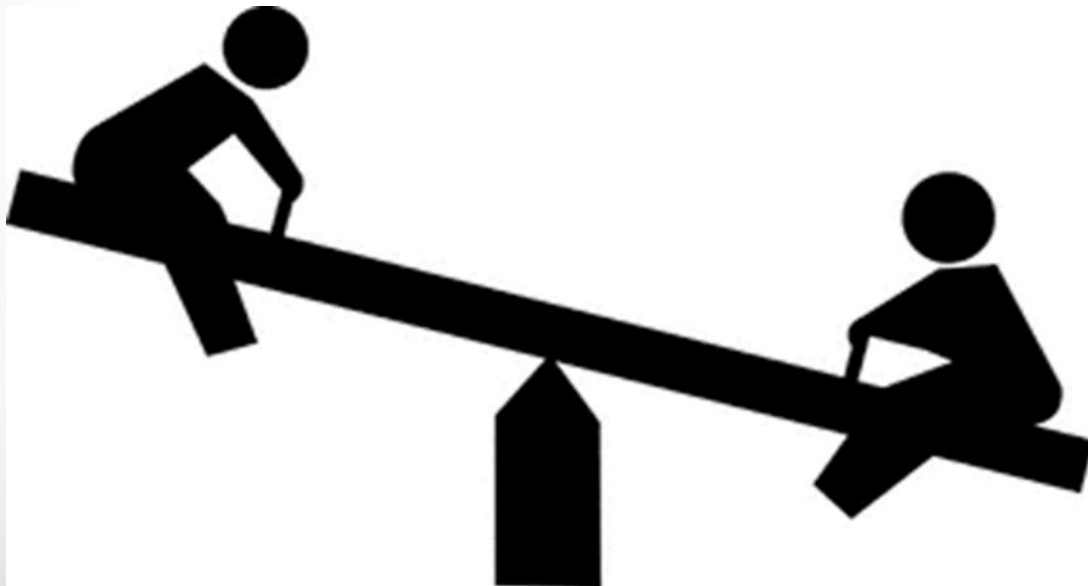


## **Issues to address (not inclusive)**

- 1) Capital and operating costs are affordable
- 2) Staff can handle operational scheme over the long term
- 3) Technology can operate long term under a reasonable maintenance program
- 4) Technology and treatment train can handle source water quality changes
- 5) Any waste stream generated can be treated or disposed in a sustainable and cost-effective manner over the long term

## Choice of technology, design, and operations can lead to...

- 1) Negative impacts on the performance of the rest of the **treatment system** for other parameters (e.g., decreased control of particulates/pathogens, taste & odor compounds, other source water contaminants)
- 2) Negative impacts on the **distribution system** (e.g., increased lead, copper, or iron corrosion; disinfection residual maintenance difficulties)



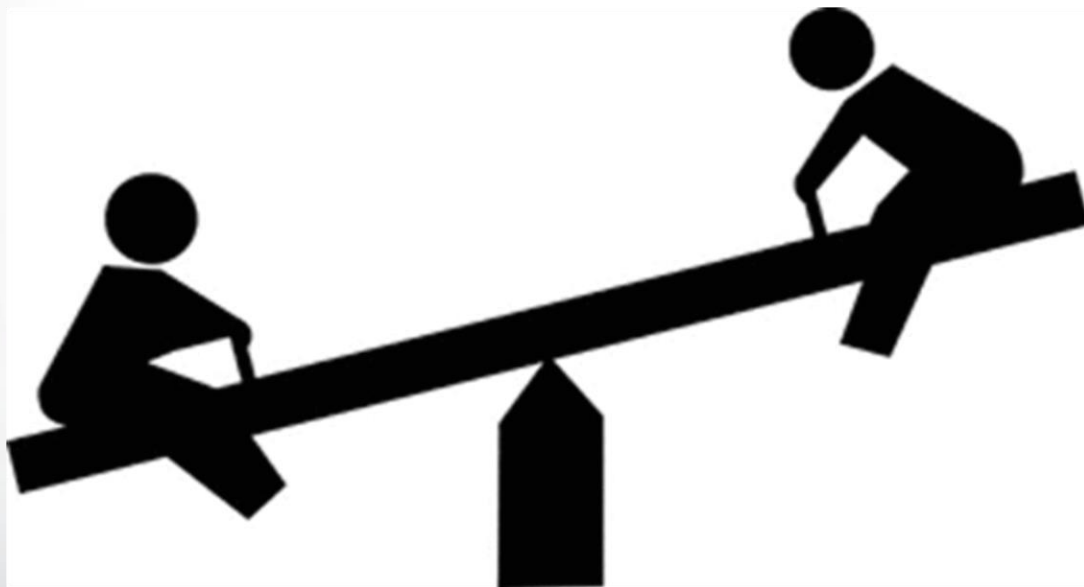
**EPA is conducting  
research on optimizing  
CEC treatment**



# To Achieve other Positive Benefits

Choice of technology, design, and operation can have...

- 1) **Positive impacts** on the performance of the rest of the **treatment system** for other parameters (e.g., improved control of particulates/pathogens, taste & odor compounds, industrial contaminants, pesticides, pharmaceuticals, personal care products, endocrine disruptors)
- 2) **Positive impacts** on the **distribution system** (e.g., decreased lead, copper, or iron corrosion; better disinfection residual maintenance; fewer disinfection byproducts)



**Improved Treatment**  
**Improved Disinfection**  
**Decreased Corrosion**



**EPA is a resource for  
communities, states and regions**



# Contact

## **Tom Speth, Ph.D., P.E.**

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US EPA Office of Research and Development  
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## **Acknowledgements**

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**ABT Associates:** Pat Ransom