



NO_x emissions from urban centers mix with emissions from vegetation, facilitating PM_{2.5} formation in many parts of the world, including downwind of Atlanta, GA (pictured here).

EPA Study Shows Combustion Emissions Influence Rate and Amount of PM from Vegetation

by Carrie Holz and Emily Smith

Results from a study that examined how atmospheric oxidation from vegetation influences the formation of fine particulate matter.

Recent research led by the U.S. Environmental Protection Agency (EPA) examined how the rate of atmospheric oxidation of volatile organic compounds (VOCs) influences the formation of fine particulate matter (PM_{2.5}). The team's research included laboratory experiments, atmospheric observations, and model calculations, and found that when nitrogen oxides (NO_x)—which is dominantly man-made and comes from combustion activities—is reduced, there is a corresponding reduction in how quickly VOCs from vegetation form PM_{2.5}. This, in turn, reduces the amount of PM_{2.5} in that region.

The research estimates that a quarter of the decreases in organic PM_{2.5} seen over the past 20 years in the southeastern United States are likely due to reductions in NO_x emissions. This means that air quality mitigation strategies targeting reductions in ozone and inorganic PM_{2.5} formation have also led to decreases in organic PM_{2.5} formation.

The study, published in the *Proceedings of the National Academy of Sciences (PNAS)*,¹ focuses on a class of compounds from vegetation called monoterpenes, which are chemicals emitted from certain types of plants. Monoterpenes all have the same chemical formula, but can have different structures. This study focused specifically on the “alpha-pinene” structure, which is one of the major species emitted from pine trees that gives rise to their pine scent. Prior research showed that monoterpene chemistry is responsible for most of the organic component of PM_{2.5} in the southeastern United States.²

“Since monoterpenes are emitted in the gas phase, chemistry is required to process them into lower-volatility forms that can enter PM_{2.5},” said Havalala Pye, EPA scientist and lead author of the study. “This study was unique, in that it focused on the mechanism by which NO_x affect PM_{2.5} and was able to verify the mechanism with experimental data.”

The research team used datasets developed from laboratory environmental chamber experiments to understand the chemical interactions taking place. Pye noted that the study was motivated by concern about a “NO_x penalty” that could potentially increase PM_{2.5} as NO_x emissions are reduced.

“Monoterpenes are known to undergo a unique type of reaction called autoxidation, which leads to efficient PM_{2.5} formation,” she explained. “Autoxidation exhibits a NO_x penalty in terms of PM_{2.5} yield when NO_x emissions decline, since NO_x generally acts to suppress autoxidation reactions.”

To determine the atmospheric impact of NO_x, researchers first showed that autoxidation reactions were the main driver of PM_{2.5} for monoterpenes by building a chemical mechanism that would describe in detail what was taking place at each stage of the chemical reaction. The highly detailed mechanism, consisting of hundreds of reactions and products, was able to reproduce PM_{2.5} concentrations observed in previous lab experiments.

Researchers then looked at the mechanism in the context of changing NO_x emissions. “While the NO_x yield penalty was indeed present, the mechanism indicated that oxidant abundance was actually more important for dictating the PM_{2.5} from monoterpenes,” Pye said. “In fact, oxidant reductions due to decreasing NO_x emissions can overcome the yield penalty from autoxidation.”

Models, including EPA’s Community Multiscale Air Quality Modeling (CMAQ; <https://www.epa.gov/cmaq/learn-about-cmaq>) System, and data from a 2013 Southeast Nexus (SENEX; <https://www.esrl.noaa.gov/csd/projects/senex/>) field study were also used to confirm the research team’s understanding of how PM_{2.5} changes in the ambient environment.

During the SENEX study, an aircraft was flown in and out of the urban plume downwind of Atlanta, measuring the effects of NO_x. Researchers noticed that inside the plume, NO_x

concentrations, ozone, organic PM, and monoterpene autoxidation products were enhanced. Outside the plume, in the regional background, autoxidation products were lower in concentration, which confirmed that emissions from human activities were enhancing PM_{2.5} from monoterpene autoxidation.

According to Pye, research dating back to the 1990s has proposed that monoterpenes were important for PM_{2.5} formation, but that science is only now beginning to reveal the nuances of this relationship. She says there’s still much work to be done before researchers will fully understand monoterpene chemistry.

“The abundance of monoterpene secondary organic aerosol and interaction with NO_x is only now being elucidated and quantified,” Pye said. “Mechanisms offer a path forward to determine robust associations of biogenic VOCs with anthropogenic emissions and the resulting implications for air quality today, and in the future.”

About the Collaborators

EPA scientists collaborated on this study with scientists from the University of Washington; Pacific Northwest National Laboratory; National Oceanic and Atmospheric Administration; and University of Colorado, Boulder. The research was made possible by a Presidential Early Career Award for Scientists and Engineers (PECASE; <https://obamawhitehouse.archives.gov/the-press-office/2017/01/09/president-obama-honors-federally-funded-early-career-scientists>) award to Havala Pye. The PECASE award is the highest honor bestowed by the U.S. government on science and engineering professionals in the early stages of their independent research careers. The award comes with a financial stipend that allows investigators to pursue questions of high societal importance to protecting public health and the environment. Advances in this work provide the scientific underpinnings of future National Ambient Air Quality Standards and air quality management strategies to achieve them. **em**

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More Information

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