

SUMMARY

A residue method for quantitation of Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone in soil test system was validated.

The test substance containing a mixture of Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone (1:1:1) was analyzed using external standardization by Liquid Chromatography with Tandem Mass Spectrometry Detection (LC-MS/MS). Soil samples were fortified with 1 ppb (LOQ) and 10 ppb (10X LOQ) of each compound. The limit of detection was defined as approximately 20% LOQ using the current methodology.

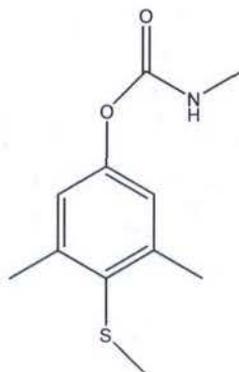
The experiment for soil matrix was conducted with two reagent blank, two untreated controls and five control samples spiked for each fortification level: one at LOQ level and another at 10X LOQ level. Target compounds were extracted from soil samples, followed by liquid partition and SPE (solid phase extraction) clean-up. The final concentrated extract was reconstituted with methanol and analyzed by LC-MS/MS.

Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone contents were quantitated against separate 1/x weighted linear curves of the reference substances Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone whose concentrations ranged from 2 ng/mL to 100 ng/mL for each compound. The calibration for each compound yielded acceptable linearity (correlation coefficient $r > 0.995$) over the range examined. The quantitation of each compound was based on the peak area response and concentration of the calibration standards. The amount of Methiocarb was determined with the quantitation MS/MS ion transition from m/z 226 to m/z 169 and the confirmation MS/MS ion transition from m/z 226 to m/z 121. The amount of Methiocarb sulfoxide was determined with the quantitation MS/MS ion transition from m/z 242.5 to m/z 185 and the confirmation MS/MS ion transition from m/z 242.5 to m/z 170. The amount of Methiocarb sulfone was determined with the quantitation MS/MS ion transition from m/z 258 to m/z 107 and the confirmation MS/MS ion transition from m/z 258 to m/z 202. Recoveries from fortified samples were determined by calculating the found concentration of individual compound and dividing the concentration by the relevant fortification level.

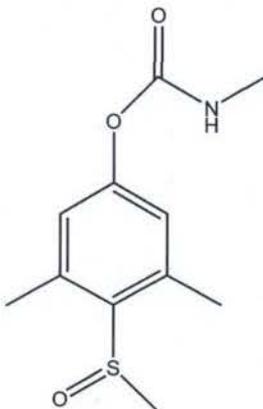
The LOD in soil matrix is estimated to be 0.2 ppb of each compound using either MS/MS transition.

MATERIAL AND METHODS**Test and Reference Substances**

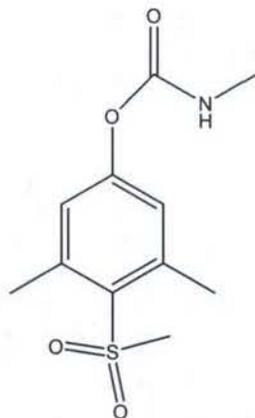
Name: Methiocarb
Supplier: Ricerca Biosciences, LLC
Lot No.: 436200075
IUPAC name: 3,5-Dimethyl-4-(methylthio)phenyl methylcarbamate
CAS No.: 2032-65-7
Molecular formula: $C_{11}H_{15}NO_2S$
Molecular weight: 225.3 grams/mole
Purity: 98.97%
Expiration Date: November 2014
Structure:



Name: Methiocarb sulfoxide
Supplier: Sigma-Aldrich
Lot No.: SZBC248XV
IUPAC name: 3,5-Dimethyl-4-(methylsulfinyl)phenyl methylcarbamate
CAS No.: 2635-10-1
Molecular formula: $C_{11}H_{15}NO_3S$
Molecular weight: 241.3 grams/mole
Purity: 99.5%
Expiration Date: September 4, 2017
Structure:



Name:	Methiocarb sulfone
Supplier:	ChemService
Lot No.:	2352100
IUPAC name:	3,5-Dimethyl-4-(methylsulfonyl)phenyl methylcarbamate
CAS No.:	2179-25-1
Molecular formula:	C ₁₁ H ₁₅ NO ₄ S
Molecular weight:	257.3 grams/mole
Purity:	98.7%
Expiration Date:	June 30, 2015
Structure:	



Certificates of Analysis for the reference and test substances are provided in Appendix B.

Other Chemicals

HPLC grade water, deionized water, acetone, methanol, hexanes, dichloromethane and ethyl acetate were obtained from Burdick & Jackson; formic acid, obtained from Fisher Scientific.

Equipment List

Laboratory Balances

Whatman No. 4 filter

SPE Cartridge, (20mL/5g FL, Agilent Mega Bond Elut)

HDPE Wide-Mouth Bottles

Thermometers

Pasteur pipettes

Beakers

Suction flask, (125mL capacity)

Buchner Glass funnels (6 cm diameter)

Wrist-action Shaker

Graduated glass cylinders

Hamilton glass precision syringes

Volumetric flasks

Pipetmen with plastic disposable tips

Separatory funnels (250 mL capacity)

Concentration Flasks

Amber bottles and vials with Teflon® lined caps

Vortex mixer

Büchi rotavapor with water bath

Turbovap® LV nitrogen evaporator

AB Sciex API 4000 Series Triple Quad Mass Spectrometer with Thermo Scientific
Dionex Ultimate 3000 Liquid Chromatograph (LC-MS/MS)

Test System

Source of Test System

Soil test system is a sandy loam collected at Vacaville, CA. The soil sample (Inventory No. 2439W-046) was stored refrigerated (typically < 4°C) in the dark prior to the beginning of the study and kept at room temperature during the short period (about one week) when the validation was conducted.

Characterization of the Test System

The soil (Inventory No. 2439W-046) used in the study was characterized by Agvise Laboratories, Inc. (604 Highway 15 West, Northwood, North Dakota). The characterization reports are presented in Appendix C.

Test Method

The analytical method for the analysis of three compounds validated at PTRL West by Liquid Chromatography with Tandem Mass Spectrometry Detection (LC-MS/MS) was based on the analytical method developed in this study.

The soil samples were spiked with known concentrations of three compounds (Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone (1:1:1)). Compounds were extracted from samples, followed by liquid partition and SPE (solid phase extraction) clean-up. The final concentrated extract was reconstituted with methanol and analyzed by LC-MS/MS. The percent recovery was determined using external standardization where linear curves for calibration standards were analyzed along with the samples.

Preparation of Methiocarb Stock Solution

A stock solution of the Methiocarb reference substance was prepared by weighing an aliquot (26.14 mg) of the reference substance (Inventory No. 2465W-001) onto glass boat and then into a 50 mL volumetric flask. The stock solution was dissolved and diluted with 51.7 mL acetonitrile to yield a nominal concentration of 0.500 mg/mL. The concentration of stock solution was corrected for the purity of the reference substance (98.97%). The stock solution was transferred into an amber bottle and stored in the freezer (typically < -10°C) when not in use.

Preparation of Methiocarb sulfoxide Stock Solution

A stock solution of the Methiocarb sulfoxide reference substance was prepared by weighing an aliquot (25.44 mg) of the reference substance (Inventory No. 2465W-002A) onto glass boat and then into a 50 mL volumetric flask. The stock solution was dissolved and diluted with 50.6 mL acetonitrile to yield a nominal concentration of 0.500 mg/mL. The concentration of stock solution was corrected for the purity of the reference substance (99.5%). The stock solution was transferred into an amber bottle and stored in the freezer (typically $< -10^{\circ}\text{C}$) when not in use.

Preparation of Methiocarb sulfone Stock Solution

A stock solution of the Methiocarb sulfone reference substance was prepared by weighing an aliquot (21.85 mg) of the reference substance (Inventory No. 2465W-003A) onto glass boat and then into a 50 mL volumetric flask. The stock solution was dissolved and diluted with 50.0 mL acetonitrile to yield a nominal concentration of 0.431 mg/mL. The concentration of stock solution was corrected for the purity of the reference substance (98.7%). The stock solution was transferred into an amber bottle and stored in the freezer (typically $< -10^{\circ}\text{C}$) when not in use.

Preparation of Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone Mixed Stock and Fortification Solutions

The 100 $\mu\text{g}/\text{mL}$ mixed stock solution was prepared by measuring 5 mL of each stock Methiocarb and Methiocarb sulfoxide and 5.8 mL of Methiocarb sulfone solution into 25 mL volumetric flask. Final solution was diluted to the mark with methanol. This mixed stock solution contains 98 $\mu\text{g}/\text{mL}$ of each compound.

The 10 $\mu\text{g}/\text{mL}$ mixed fortification solution was prepared by measuring 1.02 mL of 98 $\mu\text{g}/\text{mL}$ mixed stock solution into 10 mL volumetric flask. Final solution was diluted to the mark with methanol. This mixed fortification solution contains 10 $\mu\text{g}/\text{mL}$ of each compound.

The 1 $\mu\text{g}/\text{mL}$ mixed fortification solution was prepared by measuring 1 mL of 10 $\mu\text{g}/\text{mL}$ mixed stock solution into 10 mL volumetric flask. Final solution was diluted to the mark with methanol. This mixed fortification solution contains 1 $\mu\text{g}/\text{mL}$ of each compound.

The mixed stock and fortification solutions were vortexed to mix, transferred into amber bottle and stored in the freezer (typically $< -10^{\circ}\text{C}$) when not in use.

Preparation of Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone Mixed Standard Solutions

Due to the instability of Methiocarb sulfone, it is recommended that the calibration solutions be freshly prepared for the analysis. Seven calibration standard solutions were prepared by mixing an appropriate volume of 1 $\mu\text{g}/\text{mL}$ mixed fortification solution via pipetman with an appropriate volume of methanol. Final calibrants were transferred into autosampler vials and vortexed to mix, for LC-MS/MS analysis. The standard solutions ranged from 2 ng/mL to 100 ng/mL can be prepared as shown below:

Theoretical Conc. (ng/mL) Each compound	Volume of 1 $\mu\text{g}/\text{mL}$ Solution (μL)	Volume of Methanol (μL)	Final Volume (mL)
100	100	900	1.0
75	75	925	1.0
50	50	950	1.0
25	25	975	1.0
10	10	990	1.0
5	5	995	1.0
2	2	998	1.0

Fortification Procedure

Fortification of untreated soil samples was conducted at two fortification levels as shown below:

Fortification Level (ppb or $\mu\text{g}/\text{L}$)	Each compound Solution
1.0	25 μL of 1 $\mu\text{g}/\text{mL}$ in 25 grams of soil
10	25 μL of 10 $\mu\text{g}/\text{mL}$ in 25 grams of soil

Fortification was conducted to determine the percent recovery within the method validation. This procedure was performed in quintuplicate during method validation at each fortification level for each compound.

Extraction Method for Methiocarb, Methiocarb Sulfoxide and Methiocarb Sulfone in Soil

1. Weigh 25g of soil into a 125mL new plastic disposable centrifuge bottle.
2. Fortify the sample as necessary.
3. Add 40 mL of acetone and 10mL deionized water. Place on a wrist-action shaker for 30 minutes.
4. Filter the mixture through a Whatman No. 4 filter into a 125mL suction flask with the aid of a vacuum (water aspirator). The filter is supported in a Buchner funnel connected to the suction flask.
5. Rinse the 125mL plastic bottle and filter cake with 10mL acetone.
6. Transfer the filtrate and rinse to a 250mL separatory funnel.
7. Rinse the 125mL suction flask with 50mL of 1:1 Dichlormethane:Hexane, (v:v), sonicate and transfer to the 250mL separatory funnel containing the filtrate.
8. Shake the funnel vigorously by hand for 2 minutes.
9. Drain the lower aqueous layer back into the suction flask, collect the upper organic layer into a clean 250 mL concentration flask. .
10. Return the aqueous layer to the separatory flask and repeat previous three steps.
11. Roto-evaporate the combined extract at 30 °C to dryness.
12. Reconstitute the sample with 3mL 1:4 Ethyl acetate/Hexane, (v:v), sonicate well.
13. Condition SPE cartridge (20mL/5g FL, Agilent Mega Bond Elut) with 10mL of 1:4 Ethyl acetate/Hexane (v:v).
14. Load the concentrated sample onto the conditioned cartridge, collecting the eluate in a clean 50 mL concentration flask.
15. Rinse the 250 mL flask with 30mL of 1:4 Ethyl acetate/Hexane, (v:v), and add to the cartridge collecting into the same 50 mL flask.
16. Rinse the 250 mL flask with 10mL of acetone and add to the cartridge collecting and combining in the same 50mL flask.
17. Roto-evaporate the combined extract at 30 °C to dryness.
18. Reconstitute the sample in 2.5 mL methanol and sonicate well.
19. Aliquot in GC-vials for analysis by LC-MS/MS.

A schematic diagram of the soil extraction method is presented in Figure 1.

Liquid Chromatography with Tandem Mass Spectrometry Analytical Method (LC-MS/MS)

LC conditions

Column: Agilent Zorbax® SB-CN, 3.5µm, 4.6mm X 75mm (S/N USMC004586)

Injection volume: 10 µL

Autosampler tray temperature: 8 °C

Flow rate: 0.5 mL/min

Run time: 12.5 minutes

Mobile Phase:

- A: 0.05% Formic acid in HPLC grade water
- B: 0.05% Formic acid in HPLC grade Methanol

Gradient Program:

Time (minutes)	%A	%B	Flow rate (mL/min)
0.0	80	20	0.5
0.5	80	20	0.5
4.0	10	90	0.5
6.5	10	90	0.5
7.0	80	20	0.5
12.5	80	20	0.5

MS conditions

APCI Positive mode

APCI Parameters

Collision Gas (CAD)	9
Curtain Gas (CUR)	40
Gas 1 (GS1)	90
Gas 2 (GS2)	40
Temperature (TEM)	300
Declustering Potention (DP)	50
Exit Potential (EP)	10
Q1 resolution	unit
Q3 resolution	low

MRM Parameters

Compound name	Precursor ion	Product ion	Collision Energy (CE)	Cell Exit Potential (CXP)	Dwell (msec)
Methiocarb	226	169	14	14	100
Methiocarb	226	121	27	10	100
Methiocarb sulfoxide	242.5	185	21	9.8	100
Methiocarb sulfoxide	242.5	170	32	15	100
Methiocarb sulfone	258	107	54	9	100
Methiocarb sulfone	258	202	12.5	11	100

LC-MS/MS Analysis

Samples were analyzed interspersed between the calibrants so as to assess the response of the calibrants if they had been affected by matrix samples (signal suppression or enhancement). Since separate linear curves were prepared for each compound, samples were interspersed between each calibration standards. Calibrants and samples were analyzed in single injection. Due to the instability of Methiocarb Sulfone, the analysis should be done as soon as the samples are prepared.

Methods of Calculation***Quantitation***

Separation of Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone was achieved by LC-MS/MS. The compounds were identified by the coincidence of their retention times with their respective reference standards and MS characteristics. The quantitation of Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone was conducted by peak area of each compound relative to the theoretical concentration of the calibrants. The content of Methiocarb, Methiocarb sulfoxide and Methiocarb sulfone in samples was quantitated against separate 1/x weighted linear curves ($y = mx + b$) of *Methiocarb*, *Methiocarb sulfoxide* and *Methiocarb sulfone* calibrants where:

y = peak area

x = ng/mL compound injected

m = slope

b = intercept

Weighting of the calibration curve of each compound was applied so as to provide better curve fit at the lower concentration levels of each compound. The calculation of weighted curve equations (linear regression) and concentration (ng/mL) present in samples and calibrants was conducted using Analyst® software.

Recoveries from fortified samples were determined by averaging the found concentration of both compounds and dividing by the relevant fortification level.

Transcriptions (spreadsheets) of the raw data to support calculations for this study are presented in Appendix D.

Calibration Range

The calibration curve was generated by Analyst® software range from 2 ng/mL to 100 ng/mL of each compound for soil validation.

Limit of Quantitation

The limit of quantitation (LOQ) was set at 1 ppb for each in soil which represented 10 ng/mL of each compound in calibration standard solution as validated in this study.

Limit of Detection

The limit of detection (LOD) was defined as approximately 20% of LOQ which represented 2 ng/mL of each compound in calibration standard solution. So the LOD for soil is estimated to be 0.2 ppb of each compound.

Time Required for Completion of a Sample Set

A sample set can be divided into two subsample sets for efficient handling. A subsample set consisted of a reagent blank, one controls (untreated soil samples) and five fortified soil samples (at one level i.e. LOQ). Time required for one subsample set from initiation of extraction until the completion of instrumental analysis and data evaluation is as follows:

- Sample preparation takes approximately 6 hours

- LC-MS/MS analysis and data processing (one or two MS/MS transitions for each compound) take approximately 6 hours

TOTAL = approximately 12 hours for one analyst to complete a subsample set (approximately one a half calendar days) or 24 hours (3 calendar days) to complete two subsample sets to satisfy the validation requirements.

Statistical Methods

Means, standard deviation, relative standard deviation, and 1/x linear regression were the only statistical methods employed in this study.

Figure 1. Schematic Diagram of the Analytical Method-Soil.

