### 1. Introduction

### Background and Objective:

The objective of this study was to independently validate an analytical method for the determination of d-phenothrin in surface water. The method was originally validated by Smithers Viscient using liquid/liquid clean-up with subsequent LC-MS/MS analysis to achieve a limit of quantitation (LOQ) of  $0.005~\mu g/L$ .

#### 2. EXPERIMENTAL

# 2.1 Test System

Surface water (River Brenz) was obtained locally. The surface water was characterized for its inorganic load (e.g. pH, conductivity, hardness) and its organic load (e.g. DOC/TOC), whereby these characterizations were performed (non-GLP) by Institut Alpha, Ulm, Germany, following appropriate DIN/EN procedures (see APPENDIX 3 for detailed characteristics).

# 2.2 Analytical Test and Reference Item(s)

One analytical standard of d-phenothrin obtained by Sumitomo Chemical Co., Ltd. (see Appendix 1 for Certificate of Analysis) was used as test / reference item:

#### d-phenothrin

Chemical Formula: C<sub>23</sub>H<sub>26</sub>O<sub>3</sub> Molecular Mass: 350.458 g/moL

CAS no: 26046-85-5 Expiry date: 27-Feb-2019

Batch/Lot no: C170227 Purity: 99.8%

The analytical standard was stored in a refrigerator when not in use.

# 2.3 Analytical Method

## 2.3.1 Apparatus

#### 2.3.1.1 Laboratory Equipment

Mettler-Toledo XS205DU analytical balance for analytical standard.

Rotary evaporators Büchi Rotavapor R 200 V800 and Büchi Rotavapor R 210 V850.

Ultrasonic bath Sonorex RK100, BANDELIN electronic.

Nitrogen evaporator, Thermo Scientific.

Vortex mixer REAX top, Heidolph.

Typical glassware and laboratory equipment.

All the glassware was cleaned in a laboratory dishwasher and air-dried before use.

### 2.3.1.2 LC-MS/MS System

Agilent 1290 infinity series LC system (vacuum solvent degasser, binary LC pump, column oven) and CTC Analytics HTC-Pal Autosampler.

Columns:

Agilent Poroshell C<sub>8</sub> column: Length: 50 mm, i.d.: 3.0 mm, particle size: 2.7 μm.

Pre-column: Phenomenex C<sub>18</sub>, 4 mm length, 3 mm i.d.

Applied Biosystems API5500 Q-trap LC-MS/MS system with TurboIonspray (ESI) source.

Analyst 1.6.3 Instrument control and data acquisition software.

### 2.3.2 Solvents, Chemicals and Miscellaneous

Acetone, for Pesticide Residue Analysis, Promochem.

Acetonitrile, HPLC grade, Promochem.

n-Hexane, for Pesticide Residue Analysis, Promochem.

Water, LC-MS grade, Merck.

Methanol, LC-MS grade, Merck.

Millipore water, supply at EAG Laboratories GmbH.

Ammonium acetate, ≥ 98%, Sigma-Aldrich.

Nitric acid, 65%, Fluka.

### 2.3.3 Preparation of Standard Solutions

Two stock solutions of the analyte were prepared in acetone as exemplified:

Substance name	Weight* [mg]	Dissolve in [mL]	Obtain*[m g/mL]
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	5.05	5.04	1.0
d-phenothrin (purity 99.8%)	50.38	5.03	10

<sup>(\*):</sup> Purity was taken into account.

Fortification solutions with concentrations of 10 and  $0.10 \,\mu\text{g/mL}$  were prepared in acetone diluting the 10 mg/mL d-phenothrin stock solution in acetone and a further subsequent dilution.

Calibration solutions were prepared from intermediate solutions in acetonitrile with concentrations of 10000, 100 and 10 ng/mL by volumetric dilution to obtain concentrations in a range from 0.005 ng/mL to 0.10 ng/mL in acetonitrile/water (8/2; v/v). For preparation of matrix-matched standards final extracts of residue-free control specimens (processed together with the validation specimens) were used. Aliquots of the final extracts were fortified with the analyte using an intermediate solution in acetonitrile with a concentration of 1.0 ng/mL. Matrix-matched standards with a matrix content of at least 90% were prepared in concentrations of 0.005, 0.010, 0.025, 0.050, 0.080 and 0.10 ng/mL.

All solutions containing the analyte were stored refrigerated, when not in use.

#### 2.3.4 Residue Analysis

#### Preparation of non-disposable glassware:

- Rinse all non-disposable glassware either three times with equivalent amounts of 20% nitric acid or by soaking for 15 minutes.
- 2. Rinse the glassware three times with millipore water until no nitric acid is left.
- 3. Rinse the glassware two additional times with millipore water, followed by three equivalent volumes of acetone.
- 4. Finally rinse the glassware with the final dilution solvent.

#### d-phenothrin

- 1. 500 mL surface water were transferred into a 1 L separation funnel.
- 2. Specimens were fortified, if necessary.
- 3. 200 mL of n-hexane were added and shaken for about 1 minute.
- 4. The n-hexane phase was transferred into a 500 mL round-bottom flask.
- 5. Step 3 and 4 were repeated.
- 6. The water was discarded after the second extraction.
- 7. The separation funnel was rinsed with 50 mL n-hexane which were added to the appropriate round-bottom flask.
- 8. The volume was concentrated to about 2 mL using a rotary evaporator at < 35°C.
- 9. 100 mL of acetone were added to the round-bottom flask.
- 10. The volume was concentrated to about 5 mL using a rotary evaporator at < 35°C.
- 11. The remaining solution was transferred into a 15 mL glass centrifuge tube.

- 12. The round-bottom flask was rinsed with 5 mL of n-hexane followed by 5 mL of acetone.
- 13. The sample was concentrated to incipient dryness using a gentle stream of nitrogen at room temperature.
- 14. 1.6 mL of acetonitrile were added, vortexed for 30 seconds and sonicated for 5 minutes.
- 15. 0.4 mL of millipore water were added, vortexed for 30 seconds and sonicated for 5 minutes.
- 16. 250  $\mu$ L of the extract (or 500  $\mu$ L for blanks) were transferred into a vial containing 4.75 mL (or 9.5 mL for blanks) acetonitrile/water (8/2; v/v). High concentration level samples were additionally diluted into the standard curve range with acetonitrile/water (8/2; v/v).

# 2.4 LC-MS/MS Analysis

Specimen extracts and calibration solutions were analyzed by liquid chromatography with tandem mass spectrometric detection (LC-MS/MS) with the following methods:

Method for d-phenothrin:

LC System	Agilent 1290 infinity series LC system (vacuum solvent degasser, binary LC pump, column oven) and CTC Analytics HTC-Pal Autosampler.				lumn
LC Column	Agilent Poroshell C <sub>8</sub> column: Length: 50 mm, i.d.: 3.0 mm, particle size: 2.7 μm.				
	Pre-column: Phenomenex C <sub>18</sub> , 4 mm length, 3 mm i.d.				
LC Injection Volume	100 μL				Salt.
Temperature	40 °C				Par
LC Method	Solvent A: Water containing 10mM ammonium acetate Solvent B: Methanol Mobile Phase Composition:				
	Time (min)	Flow rate (mL/min)	% A	% B	
	0.00	0.400	70	30	
	5.00	0.400	10	90	
	7.00	0.400	0	100	
	9.00	0.400	0	100	
	9.10	0.400	70	30	
DE BEN THE	10.00	0.400	70	30	S.E.
Retention time	≈ 6.3 min for d-phenothrin				
MS/MS System	Applied Biosystems API5500 Q-trap LC-MS/MS system with TurboIonspray (ESI) source.				urce.
Ion Source	Source temper	rature:	550°C		
Conditions	GS 1:		50 (arbitrary units)		
ESI Positive	GS 2:				
Polarity	Curtain gas: 20 (arbitrary units)				
			Medium		
			10 V		
	IonSpray voltage:		5500 V		
	Resolution:		Q1: Unit, Q3: Low		

MS/MS Conditions	MS/MS transition for quantification: Collision energy (CE):	351 m/z → 183 m/z 27 V
for d-phenothrin	Cell exit potential (CXP):  Dwell time:	12 V 300 ms
	Declustering potential (DP):	126 V
	MS/MS transition for confirmation: Collision energy (CE): Cell exit potential (CXP): Dwell time:	$351 \text{ m/z} \rightarrow 249 \text{ m/z}$ 21  V 20  V 300  ms
	Declustering potential (DP):	126 V
	MS/MS transition for confirmation:	$351 \text{ m/z} \rightarrow 305 \text{ m/z}$
	Collision energy (CE): Cell exit potential (CXP): Dwell time:	15 V 24 V 300 ms
	Declustering potential (DP):	126 V

The quantitative determination was carried out by external standardization using matrix-matched standards. Calibration functions ( $\geq$  5 levels) for d-phenothrin with three ion mass transitions ranging from 0.005 ng/mL to 0.10 ng/mL were used to evaluate the extracts (Figure 2) except for the transitions 351 m/z -> 249 m/z and 351 m/z -> 305 m/z where the lowest calibration of 0.005 ng/mL gave no detectable response (Figure 3). The lowest calibration standard used for these MRMs was 0.010 ng/mL.

Representative LC-MS/MS ion chromatograms of matrix-matched standard solutions, reagent blanks and for extracts of fortified and control specimens are presented in Figure 4 to Figure 8. The product ion spectrum of d-phenothrin is presented in Figure 1.

#### 2.5 Calculations

Recovery results for d-phenothrin with full validation for surface water derived from LC-MS/MS and calculations are shown in detail in Table 1.

The following equation was used to calculate the individual residues R in µg/L:

 $R = C_{End} \times DF \times (V_1 \times V_{End}) / (V_{Sample} \times V_{Aliquot})$ 

=  $C_{End} \times DF \times Multiplier M$ 

R: Analyte residue in μg/L.

C<sub>End</sub>: Final concentration of analyte in specimen extract, in ng/mL.

DF: Dilution factor.

V<sub>1</sub>: Solvent volume for reconstitution after evaporation to dryness: 2.0 mL.

V<sub>End:</sub> End volume: 5.0 mL (10 mL for blanks).

V<sub>Sample</sub>: Sample volume: 500 mL.

V<sub>Aliquot:</sub> Aliquot volume of V<sub>1</sub>: 0.25 mL (0.50 mL for blanks).

Recoveries (Rec.) were calculated for the fortified specimens as follows:

Rec. = 
$$(R/R_{fort}) \times 100\%$$

Example for Calculation:

The calculation is exemplified with the surface water specimen P4686-59 fortified at  $0.0050 \,\mu g/L$  (LOQ). The undiluted final extract was examined by LC-MS/MS in run file P4686API5#061 (Figure 7) to give a peak area of 1.74e+005 counts for d-phenothrin at the transition 351 m/z to 183 m/z. Using the respective calibration curve a final concentration of  $0.051 \, ng/mL$  is calculated (see Table 1).

Thus:

$$R = C_{End} \times DF \times (V_1 \times V_{End}) / (V_{Sample} \times V_{Aliquot})$$

- =  $C_{End} \times DF \times Multiplier M$
- = 0.051 ng/mL x 1 x (2.0 mL x 5.0 ml) / (500 mL x 0.25 mL)
- $= 0.0041 \, \mu g/L$

Rec. = 
$$(R/R_{fort}) \times 100\%$$

=  $(0.0041 \,\mu\text{g/L} / 0.0050 \,\mu\text{g/L}) \times 100\% = 82\%$ .

Calculations were performed with full precision. Thus minor insignificant discrepancies may arise when recalculated.

### 3.6 Deviation to Method

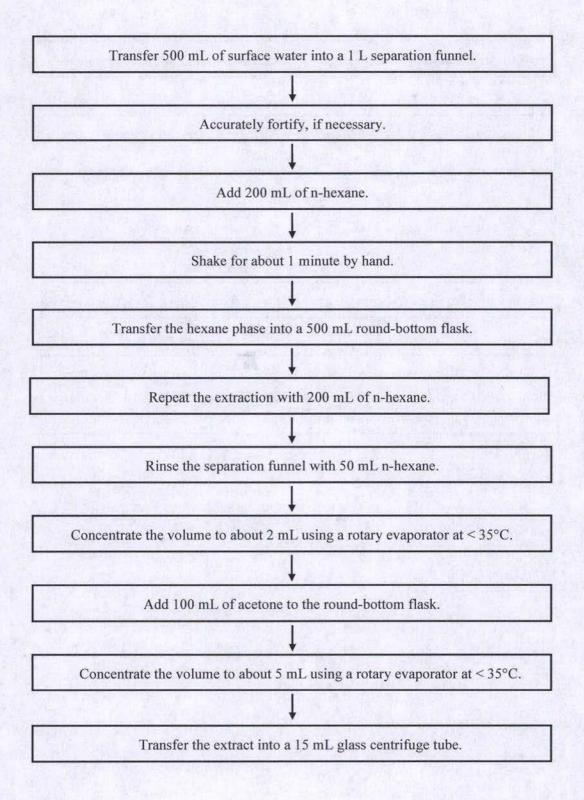
Minor deviation from the original method:

Due to different technical equipment samples were kept at room temperature and not cooled to 5°C while standing in the autosampler. This minor deviation has no impact on the outcome of the study.

# 3.7 Deviation to Study Plan

Matrix-matched standards were used for the evaluation of the results as a strong suppression on instrument response was observed for surface water. For the same reason the lowest calibration of 0.005  $\mu$ g/L in surface water gave no detectable response for both confirmatory transitions. Furthermore, this also implicated that calibration functions obtained from injections of matrix-matched standards consisted of  $\geq 5$  different concentrations and not  $\geq 6$  concentration levels as required by the Study Plan. Nevertheless, the existing calibrations range cover the range from no more than 30% of the LOQ and at least +20% of the highest analyte concentration level detected in a fortified sample as described in the protocol. There is no impact on the outcome of the study.

### Appendix 4 Method Flow Chart



# Appendix 4 Method Flow Chart (continued)

Rinse the round-bottom flask with 5 mL of hexane followed by 5 mL of acetone.

Concentrate the combined solutions to dryness using a stream of nitrogen at room temperature.

Add 1.6 mL of acetonitrile, vortex for 30 seconds and sonicate for 5 min.

Add 0.4 mL of purified reagent water, vortex for 30 seconds and sonicate for 5 min.

Transfer 250  $\mu$ L of the extract (or 500  $\mu$ L for blanks) into a vial containing 4.75 mL acetonitrile / water (8/2, v/v) (or 9.5 mL for blanks). Dilute high concentration level samples into the standard curve range with acetonitrile/water (8/2; v/v).