

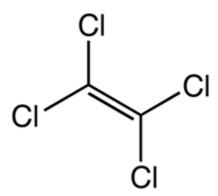
Office of Chemical Safety and Pollution Prevention

# Final Risk Evaluation for Perchloroethylene

## Systematic Review Supplemental File:

### Data Extraction Tables of Environmental Fate and Transport Studies

## CASRN: 127-18-4



December 2020

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Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Water								
Aerobic serum bottle test	Aerobic bottle test initial mass = 37 µg	Digested sludge: Preconditioned with methane or phenol bacteria, municipal digester sludge from the batch reactor cultures	aerobic	20 days	Biodegradation parameter: percent removal (based on active bottles and expected mass): Methane culture: 0%/20d (reported as -5%); phenol culture: -15%/20d	The reviewer agreed with this study's overall quality level.	Long et al. (1993)	High
Pure culture aerobic biodegradation	5 mg/L	other: Pure culture: T. versicolor (ATCC#45230); maintained by subculturing on 2% malt extract agar slants (pH 4.5) at room temperature. Subcultures were routinely made every 30 days.	aerobic	9 days	Biodegradation parameter: rate constant: 0.20 to 0.28 nmol h <sup>-1</sup> mg <sup>-1</sup> dry weight of biomass/3day	The reviewer agreed with this study's overall quality level.	<u>Marco-</u> <u>Urrea et al.</u> (2006)	High

#### Table 1. Biodegradation Study Summary for Perchloroethylene

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Reductive dechlorination in semi-continuous digester using a mixed culture inoculum	50 µmol/L	anaerobic microorganisms: Mixed culture isolated from sludge of urban wastewater treatment plant (Bourg-en-Bresse France); methanogenic, sulfate-reducing, acetogenic bacteria	anaerobic	37 days	Biodegradation parameter: percent dechlorination: 100%/37 days Biodegradation parameter: dechlorination rate: 5.2 nmol/mg protein/day Biodegradation parameter: removal rate: 3.3 μM day <sup>-1</sup> , calc. during first 9 days; conc. of trichloroethylene increased at the same rate during this time	The reviewer agreed with this study's overall quality level.	<u>Cabirol et</u> al. (1996)	High
Non-guideline; closed system, incubated in the dark at 35°C, inverted to minimize gas leakage	≤200 μg/L	activated sludge, adapted; To the medium was added 10ml/L of a methanogenic mixed culture, grown in a laboratory-scale digester fed waste-activated sludge, xylan and cellulose.	anaerobic	57 days	<u>Biodegradation</u> <u>parameter:</u> <u>concentration</u> <u>(initial</u> <u>concentration: 200</u> <u>ug/L):</u> 160 μg/L after 19d; <lod 57="" days<br="">(&lt;0.1μg/L);</lod>	The reviewer agreed with this study's overall quality level.	Bouwer and Mccarty (1983)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					Biodegradation parameter: percent removal: 100%/8w			
Anaerobic serum bottle test	80 µg/L	digested sludge; Municipal anaerobic digester sludge from batch reactor cultures	anaerobic	60 days	Biodegradation parameter: percent removal: 100%/60d	The reviewer agreed with this study's overall quality level.	<u>Long et al.</u> (1993)	High
Removal of chlorinated organic compounds by fixed-film bacteria in a continuous-flow system	10 µg/L	other: Biofilm resulting from primary sewage seed	aerobic	2 years	Biodegradation parameter: test system influent/effluent <u>comparison:</u> -1% Mean concentration; influent = 9.8±3.7 ug/L; effluent = 9.9±3.1 ug/L	The reviewer agreed with this study's overall quality level.	Bouwer and Mccarty (1982)	High
Reductive dechlorination in a semi-continuous reactor with an anaerobic enrichment culture	≥1.63 to ≤10.7 µmol/L	other: First- generation and sixth-generation enrichment culture obtained from semi- continuous reactor, operated at 35°C with digested sludge obtained from anaerobic digester seeded from Ithaca wastewater treatment plant	anaerobic	Sixth generation: Methanol 42 to 110 days; Glucose 25 to 114 days; semi- continuous operation 16-90 days followed by incubation period 7-30 days	Biodegradation         parameter: percent         removal via         dechlorination:         >99% (0.15 to         0.81%         remaining);         Biodegradation         parameter:         dechlorination         parameter:         dechlorination         parameter:         dechlorination         products:         Ultimate product:         ethene;         intermediates         products:         vinyl	The reviewer agreed with this study's overall quality level.	Freedman and Gossett (1989)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					chloride(major), TCE, 1,2-DCEs; <u>Biodegradation</u> <u>parameter:</u> <u>dechlorination</u> <u>rate:</u> after 309 days 36.4 µmol PCE consumed, 4.90 µmol VC and 12.0 µmol ethene remained. Reductive dechlorination was not suitable unless an electron donor was present			
Non-guideline incubation	10 mg/L	other: aquifer material - Traverse City Microcosms	anaerobic	250 days	$\frac{\text{Biodegradation}}{\text{parameter:}}$ $\frac{\text{transformation}}{\text{rate:}-4.43+/-0.95}$ $\mu \text{M/day. Similar}$ results obtained in the presence of toluene, ethylbenzene or benzoate.	The reviewer agreed with this study's overall quality level.	Edwards et al. (1992)	High
Non-guideline incubation	<10 mg/L	other; aquifer material - Pensacola Microcosms	anaerobic	250 days	Biodegradation parameter: transformation rate: -1.45+/-0.34 μM/day (day 1- 40) and -2.16+/- 0.34 μM/day (day 116-131). Similar	The reviewer agreed with this study's overall quality level.	Edwards et al. (1992)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					results obtained in the presence of toluene under the same conditions but not when no other organic compounds were present with a stable toluene degrading suspended consortia derived from the Pensacola microcosms.			
Non-guideline incubation	<10 mg/L	other; aquifer material - Tyndall Airforce Base Microcosms	anaerobic	85 days	Biodegradation           parameter:           transformation           rate in toluene           amended sample:           0.013+/-0.066           μM/day. PCE not           transformed in the           presence of           benzoate or           phenol.	The reviewer agreed with this study's overall quality level.	<u>Edwards et</u> <u>al. (1992)</u>	High
Non-guideline; screening test	6.5 µm	other; Hanford soil microcosms	aerobic	30 hours	Biodegradation parameter: test substance transformation rate: 0.0 μMol/mg total suspended solids/h	The reviewer agreed with this study's overall quality level.	<u>Kim et al.</u> (2000)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Non-guideline anaerobic biodegradation experiment	17-176 μg/mL	other; methanogenic mixed culture grown in a laboratory-scale digester fed waste-activated sludge	anaerobic	16 weeks	Biodegradation parameter: concentration: 56 μg/L after 16w for 176 μg/L initial concentration; 20 μg/L after 16w for 36 μg/L initial concentration; 7 μg/L after 16w for 17 μg/L initial concentration; limited degradation	The reviewer agreed with this study's overall quality level.	<u>Bouwer et</u> <u>al. (1981)</u>	High
Static-culture flask- screening test	5 and 10 mg/L	sewage, domestic, non-adapted; Settled domestic wastewater	aerobic	28 days (includes 7- day static incubation and 3 weekly subcultures)	Biodegradation parameter: percent removal 45%/7d and 87%/28d (5 mg/L initial concentration); 30%/7d and 84%/28d (10 mg/L initial concentration) Significant degradation with rapid adaptation	The reviewer agreed with this study's overall quality level.	<u>Tabak et</u> <u>al. (1981)</u>	High
Granular sludge from USAB reactor treating sugar beet refinery wastewater	1000-1500 nmol/batch	activated sludge, industrial, non- adapted; Granular sludge from USAB reactor treating sugar beet	anaerobic	65 days	Biodegradation parameter: rate constant (first order transformation kinetics): 0.048	The reviewer agreed with this study's overall quality level.	<u>van Eekert</u> <u>et al.</u> (2001)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
		refinery wastewater			day <sup>-1</sup> g <sup>-1</sup> volatile suspended solids; <u>Biodegradation</u> <u>parameter:</u> <u>product:</u> trichloroethene; Some disappearance of test substance was seen in presence of autoclaved sludge.			
Other; Static microcosms	4.2 mg/L	natural water/sediment: freshwater; Sediment included muck from vegetation decay and marl (carbonaceous precipitate of algal growth). 4g muck or 6g marl used.	aerobic	16 weeks	Biodegradation parameter: <u>Concentration</u> (ug/L) of degradation products (95% <u>confidence</u> interval in parentheses): Trichloroethene: Week 0 - ND; Week 2 - 510(4,175); Week 16 - Trace. Cis- DCE: Wk 0 - ND, wk 2 - 300(1,360), wk 16 ND. Trans-DCE: Wk 0 - ND, weeks 2 and 4 - trace, wk 16 - ND.	The reviewer agreed with this study's overall quality level.	<u>Parsons et</u> al. (1985)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					Transformation was not observed in nonviable microcosms			
Pure culture anaerobic dehalogenation; isolation of strain and optimization of growth	50-200 mmol/L	anaerobic microorganisms; Pure cultures (strain TT4B) obtained from anaerobic sediments from a stream in Massachusetts contaminated with TCE and toluene	anaerobic	10 days	Biodegradation parameter: percent removal: 100%/10d; Biodegradation parameter: degradation products: trichloroethylene and cis- dichloroethylene. No abiotic dehalogenation was observed.	The reviewer agreed with this study's overall quality level.	<u>Krumholz</u> <u>et al.</u> (1996)	High
Non-guideline	3.33 µg/mL (approx.)	other; muck from the Everglades	anaerobic	30 days	Biodegradation parameter: Half- life: 34 days		<u>Wood et al.</u> (1981)	High
Reductive dechlorination in a semi-continuous reactor with an anaerobic enrichment culture	$\geq 3.5 \leq 55$ $\mu mol/L$	other; Methanol- PCE Enrichment culture using a methanogenic bacterium and a method reference (HERO2802294)	anaerobic	195 days	Biodegradation parameter: percent removal via dechlorination: 100%/2-4d; Biodegradation parameter: degradation products: ethene with very little or no vinyl chloride and negligible	The reviewer agreed with this study's overall quality level.	DiStefano et al. (1991)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					quantities of trichloroethene and dichloroethene isomers.			
Non-guideline; other		other: Anaerobic mixed culture known to be capable of dechlorinating PCE to ethene seeded with aquifer material from a PCE- contaminated site in Victoria, TX	anaerobic	1 hour	<u>Biodegradation</u> <u>parameter:</u> <u>Dechlorination</u> <u>rate</u> : 77 μM/day	The reviewer agreed with this study's overall quality level.	Haston and Mccarty (1999)	High
Non-guideline; other	6.98 mg/L	activated sludge (adaptation not specified); enrichment cultures from contaminated groundwaters from site former solvent-recycling factor in Germany	aerobic/ anaerobic	14 days	Biodegradation parameter: percent removal: anaerobic: 0%/14d Biodegradation parameter: removal: aerobic changed to anaerobic conditions: some transformation/ 14d; Biodegradation parameter: degradation products: 100% transformation of	The reviewer agreed with this study's overall quality level.	<u>Kästner</u> (1991)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					PCE to cis-1,2- dichloroethylene (cDCE) in aerobically prepared medium with nitrogen atm			
Influents and effluents of 27 Korean WWTPs screened for 22 chemicals		activated sludge, industrial (adaptation not specified); Various treatment processes used: 15 used activated sludge process, remaining used sequencing batch reactors, biological nutrient removal process, chemical coagulation and sedimentation, or contact oxidation.	not specified	3 months	<u>Biodegradation</u> <u>parameter: percent</u> <u>removal (average,</u> <u>estimated from</u> <u>graph):</u> 72% Removal Percentage % (degradation, volatilization, sorption to solids, all included in "removal")	The reviewer agreed with this study's overall quality level.	<u>Lee et al.</u> (2015)	High
Continuous flow column study	110 mg/L	anaerobic microorganisms	anaerobic	2- or 4-days detention time	Biodegradation parameter: percent removal: 99.98% Biodegradation parameter: Degradation products: trichloroethylene, dichloroethylene, vinyl chloride, and carbon	The reviewer downgraded this study's overall quality rating. They noted: Control groups were not reported, limiting study evaluation.	<u>Vogel and</u> <u>Mccarty</u> (1985)	Low

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					dioxide; a stepwise degradation scheme was proposed.			
Continuous-flow column study	0.76 mg/L	anaerobic microorganisms; Methanogenic bacteria	anaerobic	10 and 22 days	Biodegradation parameter: concentration: 300 μg/L to 5 μg/L. Biodegradation parameter: Degradation products: trichloroethylene, dichloroethylene, vinyl chloride, and carbon dioxide; a stepwise degradation scheme was proposed.	The reviewer agreed with this study's overall quality level.	<u>Vogel and</u> <u>Mccarty</u> (1985)	High
		other; Pure cultures: Pseudomonas stutzeri OX1, Luria-Bertani growth medium at 30°C	aerobic	24 hours	Biodegradation parameter: Degradation products: 65% chloride ion generated Initial degradation rates determined by sampling every 7 min for up to 2 hours	The reviewer agreed with this study's overall quality level.	<u>Ryoo et al.</u> (2000)	High

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Non-guideline	91 mg/L	PCE-MeOH enrichment culture, yeast extract	anaerobic	60 days	Biodegradation parameter: reaction conditions: Hydrogen was able to serve as the electron donor in reductive dechlorination of perchloroethylene to vinyl chloride and ethylene over periods of 14-40 days.	The reviewer agreed with this study's overall quality level.	DiStefano et al. (1992)	High
Simulated seasonal field study		natural water: marine; NR	aerobic		<u>Biodegradation</u> <u>parameter: half-</u> <u>lives:</u> 25 d (spring), 14 d (summer), 12 d (winter) Volatilization dominated the loss of test material	The reviewer agreed with this study's overall quality level.	<u>Wakeham</u> <u>et al.</u> (1983)	Medium
Simulated seasonal field study	0.5 to 2 μmol/L	natural water: marine; NR	aerobic	24 hours	Biodegradation parameter: half- lives poisoned with HgCl <sub>2</sub> and not poisoned, respectively: 12.1 d and 12.0 d; Based on the half- lives with and without HgCl2	The reviewer agreed with this study's overall quality level.	<u>Wakeham</u> <u>et al.</u> (1983)	Medium

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					poisoning, the loss of test material was determined to be primarily by volatilization.			
Degradation in muck-water	200 µg	natural water; Everglades muck and water collected on Feb 12, 1979	not specified	70 days	Biodegradation parameter: percent removal: 100% / 70 days Biodegradation parameter: half- life: 33.8 days (primary degradation)	The reviewer downgraded this study's overall quality rating. They noted: The starting material had reported impurities.	Dow Chemical (1980)	Medium
The influent and effluent concentrations of several VOCs were measured at two WWTPs and compared to a model developed by the authors that estimated VOC removal by volatilization, adsorption, and biodegradation based on the WWTP operational conditions.		activated sludge, industrial, adapted	aerobic		Biodegradation parameter: predicted DCM biodegradation removal: WWTP #1: 84% (mainly by volatilization); WWTP #2: 83.1% (mainly by volatilization)	The reviewer agreed with this study's overall quality level.	<u>Namkung</u> and <u>Rittmann</u> (1987)	Medium

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Degradation/dechlor ination using an enrichment culture	300 and 600 μmol/L	other; Enrichment culture of over 98% Dehalococcoides, obtained in 3 phases; microcosm set up with groundwater from chlorinated ethene- contaminated anaerobic aquifer in Bitterfeld Germany; preparation methods described briefly and referenced	anaerobic	120 days	Biodegradation parameter: percent removal: 100%/110 d; Biodegradation parameter: degradation products: vinyl chloride, ethene	The reviewer agreed with this study's overall quality level.	<u>Cichocka</u> et al. (2010)	Medium
Calculation		not specified	anaerobic		Biodegradation parameter: reduction potential for transformation half-reactions: PCE to TCE ~0.7 volts at 0°C and 100°C.	The reviewer downgraded this study's overall quality rating. They noted: Study reports calculated estimates with limited details for endpoints related to fate (thermodynamic property).	<u>Haas and</u> <u>Shock</u> (1999)	Low

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Anaerobic biodegradation	1.06 mg/L	anaerobic sludge; anaerobic microorganisms from a municipal waste treatment plant	anaerobic	100 days	Biodegradation parameter: percent removal: 100%/40d	The reviewer downgraded this study's overall quality rating. They noted: Due to limited information, evaluation of the reasonableness of the study results was not possible.	<u>Gossett</u> (1985)	Low
Non-guideline aerobic biodegradation experiment	8.8+/-15% and 74+/-15% μg/mL	other; Primary sewage effluent Palo Alto, CA, Water Pollution Control Facility	aerobic	25 weeks	Biodegradation parameter: percent remaining: 120-125% relative to control samples; limited degradation	The reviewer downgraded this study's overall quality rating. They noted: Greater than 100% of test substance was remaining relative to the controls after 25 weeks.	<u>Bouwer et</u> <u>al. (1981)</u>	Low
Pure culture study using anaerobic methanogenic bacteria	$\geq 1$ to $\leq 3$ mg/L	Methanosarcina sp. strain	anaerobic	13 days	Biodegradationparameter: rateconstant:5.7-6.0 nmol/day;Biodegradationparameter: CH4production rateconstant:180 nmol/day	Species specific biodegradation study excluded.	Fathepure and Boyd (1988)	Unacceptable

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					during the active phase of methanogenesis			
Batch sewage sludge microcosm experiment		digested sludge; digester sludge from several industrial and municipal sewage treatment plants canton de Vaud (Switzerland)	anaerobic		<u>Biodegradation</u> <u>parameter: percent</u> <u>removal in test</u> <u>system:</u> No degradation or loss occurs	Biodegradation results were not reported for Perchloroethylene	<u>Balsiger et</u> <u>al. (2005)</u>	Unacceptable
Anaerobic biodegradation with municipal solid waste (MSW), biowaste, and/or compost	4 to 5 mg/kg	other; Digester filled with MSW, biowaste, or compost; MSW from unsorted waste from households, trade and communities, organic waste from private households of Hamburg, Germany	anaerobic	Not specified; likely >130 days	Biodegradation parameter: degradation rate in test system: PCE was degraded at a low rate under acidic conditions and at a higher rate in the methane phase (however, only very low concentrations of perc initially added were found in the gas phase, attributed to adsorption and rapid decomposition	The study did not include or report control groups to validate the system used.	Deipser and Stegmann (1997)	Unacceptable

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Degradation in open and closed systems	≥ 0.1 to ≤ 1 ppm	natural water: marine; Seawater collected from outside Lysekil on the Swedish Coast	not specified	14 days	Biodegradation parameter: percent removal in open- light, closed-light, and closed-dark systems, respectively: 50%, 40%, 54%	Serious uncertainties or limitations were identified in sampling methods of the outcome of interest. In addition, loss from leaks in valves and open test systems were likely to have a substantial impact on the results, making the study unusable.	Jensen and Rosenberg (1975)	Unacceptable
Non-guideline; Modified shake flask closed bottle biodegradation test		not specified	aerobic		Biodegradation parameter: percent removal: 0%/21d Possibly multiple studies (1) Acclimation period with adaptive transfers after 48 or 72h with and without lactose; (2) No biodegradation after 21d acclimation period in a river die- away study without co- substrate	Testing methods and conditions were not reported, and data provided were insufficient to interpret results in this secondary source; citing HERO ID 18157, Mudder, T. I. and J. L. Musterman (1982).	<u>ECHA</u> (2017b)	Unacceptable

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Inhibition of gas production to anaerobic sludge from an operating municipal sludge digester	1000 mg/L	sewage, domestic (adaptation not specified); Mt Pleasant MI WWTP	anaerobic	48 hours	Parameter: inhibition of gas production: 7.3% inhibition at 10 mg/L; 20% inhibition at 100 mg/L; 36% at 300 mg/L; 50% at 550 mg/L after 48h	Study describes inhibition of gas production not biodegradation rates or transformation pathways.	<u>Dow Chem</u> <u>Co (1977)</u>	Low
Sediment								
Other; Non- guideline	500 μg/L	natural water/sediment: freshwater	aerobic/ anaerobic	436d	Biodegradation parameter: percent removal: 90% on days 338- 436; 16% during day 140-337	The reviewer agreed with this study's overall quality level.	<u>Cheng et</u> <u>al. (2010)</u>	High
Static microcosm with muck and surface water in sealed septum bottles	100	sewage, domestic, non-adapted	anaerobic	21 days	Biodegradation parameter: percent removal: 72.2%/21d	The reviewer agreed with this study's overall quality level.	<u>Parsons et</u> al. (1984)	Medium

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Other; Reductive dechlorination in anaerobic sediment; column and batch studies	≥0.6 to ≤9 µmol/L	natural water/ sediment	anaerobic	Continuous- flow column >300 days; batch >44 days	Biodegradation parameter: degradation product: 95-98% PCE was recovered as ethane under steady-state column conditions in the presence of lactate	The reviewer downgraded this study's overall quality rating. They noted: No control groups or validation were reported.	<u>de Bruin et</u> <u>al. (1992)</u>	Low
Anaerobic biodegradation in Ni contaminated sediment	600 μmol/L	Anaerobic, nutrient poor sediment, contaminated with chloroethene and nickel, collected at a depth of 15m near a former metalworking industrial plant in the Netherlands; microbial pop range 10 <sup>2</sup> -10 <sup>3</sup> cells/g of sediment under sulfate reducing conditions	anaerobic	1 year (52 wks)	Biodegradation parameter: degradation products: Ethene/ethane production was observed after 31 weeks; the addition of sulfate and lactate were necessary for microbial activity; approximately 1 mM chloride was detected in the effluent (background from control was 0.5mM)	The reviewer downgraded this study's overall quality rating. They noted: Specific results stating degradation rates and/or half-lives were not reported.	<u>Drzyzga et</u> <u>al. (2002)</u>	Low

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Microbial reductive dechlorination of PCE in sediment microcosms	55 mmol/L	sewage, domestic, non-adapted	anaerobic		Biodegradation parameter: dechlorination products: Reductive dechlorination of PCE in sediments from 3 locations generated more trans-DCE than cis-DCE; sediment from one location generated a lower trans-/cis-DCE ratio	The reviewer downgraded this study's overall quality rating. They noted: Due to limited information, evaluation of the reasonableness of the study results was not possible.	<u>Cheng et</u> <u>al. (2010)</u>	Low
Soil								
Other; Biodegradation in a continuous flow system. Long-term operation of a sand column fed methanol and concentrations of PCE under anaerobic continuous flow conditions.	≥12 to ≤720 µmol/L	Biomass from an anaerobic digester at the Metropolitan Toronto Main Treatment Plant	anaerobic	2.5 years	$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	The reviewer downgraded this study's overall quality rating. They noted: Continuous flow reactor with a sand column that's fed PCE and methanol; experiment a treatment system, the study may not be relevant to fate and environmental degradation and	<u>Isalou et al.</u> (1998)	Low

Study Type (year)	Initial Concentration	Inoculum Source	(An)aerobic Status	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
						therefore not applicable to fate assessment.		

## Table 2. Bioconcentration Study Summary for Perchloroethylene

Study Type (year)	Initial Concentration	Species	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Bioconcentration in Bluegill sunfish: Aquarium with well-water and modified continuous-flow proportional dilution apparatus for chemical introduction	3.43±1.53 μg/L	Bluegill sunfish (Lepomis macrochirus) 100 per aquarium	Tetrachloroethylene 21 days; Test: 28 days or until equilibrium; water and fish samples collected periodically until apparent equilibrium was reached or the max exposure of 28 days was reached	Bioconcentration parameter: BCF: 49 (bluegill); Bioconcentration parameter: half-life: < 1 day	The reviewer agreed with this study's overall quality level. This study is related to another study, HERO ID 3970785, Echa. Bioaccumulation: aquatic/sediment: Tetrachloroethylene. 2017.	<u>Barrows et</u> <u>al. (1980)</u>	High
Kinetic uptake measured via a <sup>13</sup> C method and ECD-GC method	0.22 or 0.50 mg/L	Skeletonema costatum	72 hours	<u>Bioconcentration</u> <u>parameter: BCF D(U):</u> 118 (algae, uptake); <u>Bioconcentration</u> <u>parameter:</u> BCF D(T): 113 (algae); derived from exp data	The reviewer agreed with this study's overall quality level.	<u>Wang et</u> al. (1996)	High

Study Type (year)	Initial Concentration	Species	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
				for % inhibition and exposure concentration with time			
Kinetic uptake				Bioconcentration parameter: BCF D(U): 312 (algae, uptake); Bioconcentration	The reviewer agreed		
measured via a <sup>13</sup> C method and ECD-GC method	0.22 or 0.50 mg/L	Heterosigma akashiwo	72 hours	<u>parameter:</u> BCF D(T): 280 (algae); derived from exp data for % inhibition and exposure concentration with time	with this study's overall quality level.	<u>Wang et</u> <u>al. (1996)</u>	High
"Bioaccumulation test of a chemical substance in fish or shellfish" provided in "the Notice on the Test Method Concerning New Chemical Substances"	0.1 and 0.01 ppm	Cyprinus carpio	8 weeks	Bioconcentration parameter: BCF: 25.8-77.1 (high); 28.4-75.7 (low) (carp)	The reviewer agreed with this study's overall quality level. The BCF study is also available from the NITE website.	<u>Kawasaki</u> <u>(1980)</u>	High
Uptake-clearance of a Perc commercial product	0.06121 and 0.650 ppm	Rainbow trout (Salmo gairdneri)	48-hour exposure followed by 144- hour clearance	Bioconcentration parameter: BCF: 39.7 (rainbow trout)	The reviewer agreed with this study's overall quality level.	Dow Chemical (1973)	High
Method described by Branson et al 1974 for determination of BCF in Rainbow trout	Two concentrations an order of magnitude apart	Salmo gairdneri		Bioconcentration parameter: BCF: 39.6 ± 5.5 (rainbow trout)	The reviewer agreed with this study's overall quality level.	<u>Neely et</u> <u>al. (1974)</u>	Medium

Study Type (year)	Initial Concentration	Species	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Not specified		Golden ide (Leuciscus idus)		Bioconcentration parameter: unspecified: 90 (Golden ide)	Limited study information provided (i.e. study controls not reported).	<u>Freitag et</u> <u>al. (1985)</u>	Unacceptable
Monitoring of halogenated hydrocarbons in mollusks and fish and comparison to concentrations in seawater		Bassinum undatum; Modiolus, Pecten maximus; Conger, Gadus morhua, Pollachius birens, Scylliorhinus canicula, Trisopterus luscus		Bioconcentration parameter: monitoring: Relative to their seawater levels these compounds were only slightly enriched; usually by a factor of 2 - 25x (snails, scallops, eels, marine fish) Bioconcentration parameter: concentration in organs: brain>gill>liver>muscle; exceptions were noted	The test substance concentration in seawater was not reported. Results provided are a range of BCF (2-25x) that are not test compound or organism specific.	<u>Dickson</u> and Riley (1976)	Unacceptable
Long term accumulation experiment in apparatus similar to that used for acute tox determination	0.03 and 0.3 mg/L	Dab	10 days and up to 3 months (3-35 days)	Bioconcentration parameter: accumulation <u>factor in organs:</u> 5-9 times for flesh and 200-400 times for liver (dab)	The study did not report crucial details on method, sampling and organisms.	Pearson and McConnell (1975)	Unacceptable
Bioaccumulation; purge-trap method using a Dean-stark apparatus	5 and 100 µg/L	Mytilus edulis	8 and 21 days	Bioconcentration parameter: BAF: 25.7 (mussel)	Foreign language paper with abstract and data tables in English. Full text article review needed when available in English.	<u>Saisho et</u> <u>al. (1994)</u>	Unacceptable

Study Type (year)	Wavelength Range	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Air						
Other; calculation Atmospheric lifetime in three regimes of the troposphere based on OH Reaction rate constant			Photodegradation parameter: Tropospheric lifetime at 288 K, 263 K, and 260 K, respectively: 80, 119, and 251 days	The reviewer agreed with this study's overall quality level.	<u>U.S. EPA (1987)</u>	High
Other; Test material injected into quartz flask containing air to give concentrations in the range of $10^{-7}$ to $10^{-4}$ by mass.	Outdoor light		Photodegradation parameter: direct photolysis half-life: 50%/12 weeks	Testing conditions were not reported, and data provided were very general; concentration of test material not specified. Ambient air used for experiment was not subject to any pretreatment or analysis; climate and conditions were not controlled.	Pearson and McConnell (1975)	Unacceptable
Water						
Direct photolysis in quartz tubes	366 nm	5 days	Photodegradation parameter: direct photolysis: Negligible	The reviewer agreed with this study's overall quality level.	<u>Chodola et al.</u> (1989)	High

#### Table 3. Photolysis Study Summary for Perchloroethylene

Study Type (year)	Wavelength Range	Duration	Result	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Photodegradation in water (indirect photolysis)	185-254 nm	60 minutes	Photodegradation parameter: indirect photolysis rate constants: 0.221 kdeg min <sup>-1</sup> (oxygenated) and 0.287 kdeg min <sup>-1</sup> (oxygen free) Photodegradation parameter: indirect photolysis half- lives: 3.75 min (oxygenated) and 3.39 min (oxygen free)	The reviewer agreed with this study's overall quality level.	<u>Shirayama et al.</u> (2001)	High
Nonguideline lab study - direct photolysis	sunlight	1 year	Photodegradation parameter: direct photolysis half-life: 50%/8.8 months	The reviewer agreed with this study's overall quality level. Related HERO ID, Echa. Phototransformation in water: Tetrachloroethylene. 2017.	<u>Dilling et al. (1975)</u>	High
Photoinduced reductive dechlorination in water containing ferrous and/or sulfide ions	Intensity 530±20 lux	33 days	Photodegradation parameter: indirect: reductive dechlorination: No appreciable degradation was observed with any medium with or without photolysis	The reviewer agreed with this study's overall quality level.	<u>Doong and Wu</u> (1992)	High

Study Type (year)	рН	Temperature	Duration	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Nonguideline; hydrolysis study	4 and 9.2	4 and 50°C (at both pH values)	7 days	Hydrolysis parameter: conditions for reaction:Under acidic and basic conditions, the percent difference of areas at 4 and 50°C ranged between - 14% to +1% indicating that the chance for transformation of tetrachloroethylene by hydrolysis under environmentally relevant conditions of temperature and pH is minimal.	The reviewer agreed with this study's overall quality level.	<u>Chodola et</u> <u>al. (1989)</u>	High
Nonguideline lab study in Pyrex tubes with light- proof container, shaken every 2-weeks, water purged with air for 15 min prior to addition of chlorinated compounds		approx. 25°C	1 year	<u>Hydrolysis</u> parameter: half-life: 50% / 8.8 months	The reviewer agreed with this study's overall quality level.	<u>Dilling et al.</u> (1975)	High

## Table 4. Hydrolysis Study Summary for Perchloroethylene

Study Type (year)	рН	Temperature	Duration	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Alkaline homogenous hydrolysis experiments; a range of pH and temperature evaluated. Arrhenius temperature dependence assumed.	2-14	130-170°C	30 min to several days (for all test materials; specific duration for tetrachloroethylene not specified)	<u>Hydrolysis</u> parameter: half-life (pH 7 at 25°C): 50% / 9.9×10 <sup>8</sup> years	The reviewer agreed with this study's overall quality level.	<u>Jeffers et al.</u> (1989)	Medium

## Table 5. Sorption Study Summary for Perchloroethylene

Study Type (year)	Sorbent Source	Sorbent Qualities (clay/silt/sand, OC, pH)	Duration	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Other; Non-guideline study using bottle-point method for isotherm experiments	Filtrasorb 400 activated carbon at 500 and 1500 µg/L; Background organic matter was treated old municipal landfill leachate in Dover, Delaware	DOC 300 mg/L	2 weeks	$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	The reviewer agreed with this study's overall quality level.	<u>Sorial et al.</u> (1994)	High

Study Type (year)	Sorbent Source	Sorbent Qualities (clay/silt/sand, OC, pH)	Duration	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
OECD Guideline 106 (Adsorption - Desorption Using a Batch Equilibrium Method)	7 samples of Danish clayey till from three sites at depths of 2.4 to 9.5 m below the surface (4 contaminated; 3 uncontaminated)	foc% 0.02-0.08; % Clay content 23.0- 27.0; 4 samples reduced clayey till; 3 samples oxidized clayey till		$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	The reviewer agreed with this study's overall quality level.	<u>Lu et al.</u> (2011)	High

Study Type (year)	Sorbent Source	Sorbent Qualities (clay/silt/sand, OC, pH)	Duration	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
				Sorption parameter: log Kd (calculated from log Kd $\equiv 0.590 \log$ Kow-1.561) (oxidized clay): 3.29, 3.54, and 3.62 Sorption parameter: log Kd (calculated from log Kd $\equiv 0.590 \log$ Kow-1.561) (reduced clay): 3.78, 3.86, 3.69, and 3.41			
Other; Non-guideline study	pulverized sediment (5-50 g) with a known volume of synthetic groundwater (5- 17 mL) and mass of PCE, from methanol stock solution, was either flame sealed into a	Three samples consisted of sand and gravel and three samples consisted of medium to fine sand.		Sorption parameter: Kd: sand and gravel: 0.82, 1.16, and 1.92 Sorption parameter: Kd: medium to fine sand:	The reviewer agreed with this study's overall quality level.	<u>Wang et al.</u> (2013)	High

Study Type (year)	Sorbent Source	Sorbent Qualities (clay/silt/sand, OC, pH)	Duration	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
	glass ampoule or sealed with a PTFE-lined butyl rubber septum into serum vial			0.8, 1.16, and 0.82			
Other; Sorption on wastewater solids (isotherm test)	Wastewater solids collected from three different municipal WWTP near Cincinnati, OH	Mixed-liquor solids, primary sludge, anaerobically digested sludge		Sorption parameter: adsorption coefficient K: primary sludge, mixed-liquor solids and digested, sludge, respectively: 0.60, 0.90, 0.70 Sorption parameter: log Kp: primary sludge, mixed-liquor solids and digested, sludge, mixed-liquor solids and digested, sludge, respectively: 2.90, 3.01, 3.09	The reviewer agreed with this study's overall quality level.	<u>Dobbs et al.</u> (1989)	High

Study Type (year)	Sorbent Source	Sorbent Qualities (clay/silt/sand, OC, pH)	Duration	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Other; Vapor partitioning/adsorption in aquifer solids	Aquifer solids from contaminated aquifers in Livermore Valleys CA with low natural organic matter sediment with sand-sized aggregates in clay and silt	Sand clay and silt: Surface area: 13-29 m <sup>2</sup> /g; Organic content 0.0064- 0.11%; particle diameter: <75-150 µm; Sand portion: internal porosity 0.54 mL/g; pore diameter: 69 Angstrom		$\frac{\text{Sorption}}{\text{parameter:}}$ $\frac{\text{Kf:}}{\text{Kf:}} 0.0048$ (nonlinear isotherm 1/n = 0.62)	The reviewer agreed with this study's overall quality level.	Farrell and <u>Reinhard</u> (1994)	High
Other; Vapor partitioning/adsorption in aquifer solids	Aquifer solids from contaminated aquifers in Santa Clara CA consisting of fragments of sedimentary rocks (54%), single grains (30%), igneous and metamorphic rock fragments (16%)	Column solids were in equilibrium with 100% relative humidity; desorption isotherms measured with stepwise batch techniques; solids allowed to adsorb PCE vapors at ca. 79% saturation and equilibrated for 1 week, purged and equilibrate for another 1-3 week		Sorption parameter: <u>Kf:</u> 0.0054 (nonlinear isotherm $1/n$ = 0.43)	The reviewer agreed with this study's overall quality level.	Farrell and Reinhard (1994)	High
Secondary source, based on several experimental studies				$\frac{\text{Sorption}}{\text{parameter:}}$ $\frac{\text{Koc:}}{141}$ $(\log \text{Koc} = 2.15)$	The reviewer downgraded this study's overall quality rating. They noted: Limited information reported in this secondary source and unable to	<u>ECHA</u> (2017a)	Unacceptable

Study Type (year)	Sorbent Source	Sorbent Qualities (clay/silt/sand, OC, pH)	Duration	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
					confirm study results with cited reference HEROID 3839195, ECB (2005). European Union risk assessment report: Tetrachloroethylene. Part 1 - Environment. United Kingdom, European Commission – Joint Research Centre Institute for Health and Consumer Protection European Chemicals Bureau. 57.		

## Table 6. Other Fate Endpoints Study Summary for Perchloroethylene

System	Study Type (year)	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Measured concentrations of tetrachloroethylene in sediments of Scheldt Estuary and water samples from Southern North Sea	Partitioning based on measurements in sediments of Scheldt Estuary and water Southern North Sea	Parameter: log Koc (sw, eq): 1.58 (Log Koc, sw: 0.94, 1.83, 1.99, 2.34; Log Koc 0.92, 1.82, 1.96, 2.26)	The reviewer agreed with this study's overall quality level.	<u>Roose et al.</u> (2001)	High
Experiment was conducted at the H1 wetland at the Tres Rios Demonstration	Volatilization rates and half- lives for VOCs in	Parameter: mass flux to the <u>atmosphere for</u>	The reviewer agreed with this study's overall quality level.	<u>Keefe et al.</u> (2004)	High

System	Study Type (year)	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Wetlands, located near the Salt River in Maricopa County, southwest of Phoenix, AZ	constructed wastewater treatment wetlands.	<u>tetrachloroethene:</u> 0.45g/d/ha			
Anaerobic batch fed reactor	digested sludge; Municipal anaerobic digester sludge	<u>Biodegradation parameter:</u> <u>test reactor</u> <u>influent/effluent</u> <u>comparison:</u> _Average reactor influent of Perc = 120 ug/L, average reactor effluent = 3 ug/L	The reviewer agreed with this study's overall quality level.	<u>Long et al.</u> (1993)	High
Aerobic batch fed reactor	digested sludge; Municipal anaerobic digester sludge	<u>Biodegradation parameter:</u> <u>test reactor</u> <u>influent/effluent</u> <u>comparison:</u> Average reactor influent = 160 ug/L; average reactor effluent = 42 ug/L	The reviewer agreed with this study's overall quality level.	<u>Long et al.</u> (1993)	High
Analysis of NYC municipal wastewaters; Influent and effluent samples were collected from 14 Water Pollution Control Plants in New York City	Analysis of NYC municipal wastewaters from 1989-1993	Parameter: WWTPinfluent/effluentcomparison:Tetrachloroethene wasdetected in 94% of influentsamples and 74% ofeffluent samples; theconcentration rangedetected in influent was 5-78 µg/L and effluent was1-69 µg/L.Tetrachloroethene was oneof five commonlyobserved volatilesdetected; present in at leasta third or more of all thesamples analyzed.	The reviewer agreed with this study's overall quality level.	<u>Stubin et al.</u> <u>(1996)</u>	High

System	Study Type (year)	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
VOCs injected into water line of shower and glass syringes were used to collect air and water samples.		Parameter: percent volatilization at 25°C, <u>33°C, and 42°C,</u> respectively: 56% +/-7%, 66% +/-7% and 68% +/-7% Parameter: percent volatilization at 42°C by <u>flow rates</u> : 69% +/-7% at flow rate 9.7 L/min and 68% +/-7% at flow rate 13.5 L/min	The reviewer downgraded this study's overall quality rating. They noted: Study investigated volatilization from shower water. Study results may not be relevant to a specific/designated Fate endpoint.	Tancrede et al. (1992)	Low
Desorption profiles obtained from PCE alone and PCE mixed with TCE in sediment	Diffusion/desorption on natural solids	Parameter: Diffusion rate <u>constant of PCE alone and</u> <u>PCE in mix with TCE,</u> <u>respectively:</u> 5.00E-4/min (95%CI: 0.02E-4, initial sorbed mass: 8.1 μg/g) and 7.33E-4/min (95%CI: 0.05E-4, initial sorbed mass: 9.2 μg/g)	The reviewer agreed with this study's overall quality level.	Li and Werth (2004)	High
Modified EPA method 624	Stripping of volatile organics from wastewater	Parameter: WWTPinfluent/effluentcomparison:88 and 139 μg/m³ max offgas samples; avg influentand effluent: 0.2μg/L inwater and 40 and 43μg/m3 in off gas atskyway. influent andeffluent: 0.3 and 0.2 μg/Lin water and 82 and 115μg/m³ in off gas athighland creek	The reviewer agreed with this study's overall quality level.	<u>Bell et al.</u> (1993)	High

System	Study Type (year)	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
VOC's were measured in water samples between November 2006-June 2008; by comparing STE and post-RO samples matched to plant and date the overall treatment efficiency was calculated as a proportion of the removal	Monitoring of water samples and correlation to treatment efficiency	Parameter: WWTP removal efficiency: The median removal efficiency was 91.2% for tetrachloroethene; STE samples (n=29): 86.2% detections; post-MF samples (n=9): 88.9% detections; post-RO samples (n=27): 14.8% detections	The reviewer agreed with this study's overall quality level.	<u>Rodriguez et</u> <u>al. (2012)</u>	High
Gravimetric measurements by a Mettler H54 balance	evaporation rates of solutes from water	Parameter: volatilization rates at 23.2°C and 3.5°C, respectively: 1.36E5 g/cm <sup>2</sup> -s and 4.77E4 g/cm <sup>2</sup> -s	The reviewer agreed with this study's overall quality level.	<u>Chiou et al.</u> (1980)	High
Refers to Hill et al.	Volatilization rate study for high-volatility compounds	Parameter: volatilization rate constant ratios <u>kvC/kvo:</u> 0.52 +/- 0.09	The reviewer agreed with this study's overall quality level.	<u>Smith et al.</u> (1980)	High
Concentrations in air, water and sludge phases analyzed under four different operational circumstances evaluating single and combined effects of aeration and sludge addition on phase distributions; sludge added prior to experiments; aeration 3rd-10th hour	Fugacity model approach to VOC fate in WWTP	Parameter: partitioning: The concentrations of the VOCsIn the air, water, and sludge phases of the bioreactor were analyzed regularly. Mass distributions indicated that Perc was mainly present in the four treatment stages; less than 1% of the total mass was subject to biological sorption and/or degradation by the sludge;	The reviewer agreed with this study's overall quality level.	<u>Chen et al.</u> (2014)	High

System	Study Type (year)	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
		water aeration resulted in increased partitioning to the air phase with a negative impact on biological removal; Perc mass distribution throughout the 4 stages: ~99% water, ~0.4% air, less than 0.1% sludge			
Reductive dehalogenation measured in static microcosms with 8 mL leachate and 10 mM PCE under 80%-N2: 10%-CO2: 10%-H2 atmosphere with and without organic carbon amendments	Dehalogenation of PCE in landfill leachates with and without organic carbon amendments under anaerobic conditions	Extent of dehalogenation was correlated with the availability of organic carbon; 19% degradation after 14 days in low carbon Cell 8 leachate and almost all PCE and some TCE degradation after 30 days; greater than 75% degradation after 14 days in high car	The reviewer agreed with this study's overall quality level.	Leahy and Shreve (2000)	High
	Concentration in seawater and air	$\frac{\text{Parameter: seawater to air}}{\text{flux:}}$ 0.02-160.4 (mean 34.3) nmol m <sup>-2</sup> d <sup>-1</sup>	The reviewer agreed with this study's overall quality level.	<u>He et al.</u> (2013)	High
200 rpm stirring of the solution with a shallow pitch propeller stirrer in still air		Parameter: volatilization half-life: 27.1 min	The reviewer agreed with this study's overall quality level.	<u>Dilling (1977)</u>	High
Transformation study under anaerobic conditions	non-guideline	Parameter: abiotic dechlorination in the presence of iron half-lives <u>at 50°C and 23°C</u> respectively:	The reviewer agreed with this study's overall quality level.	Schreier and Reinhard (1994)	Medium

System	Study Type (year)	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
		20d+/-4 and 38d+/-25; dechlorination did not occur in the presence of manganese			
wastewater flow: 41.5, 21, 852, 2390, 499, 110 and 30.5 l/min. Volatile organic loading rate: 14.6, 4.6, 292, 286, 19, 5.29, 0.395 kg/L. Feed ratio: 9.6, 10.5, 28.8, NA, 14.7, 7.1, 1.4 kg/kg for plants A-G respectively	7 steam stripper operations are reported	Parameter: percent removal from steam stripper operations: >99.2 and >99.89 removal from plant C and D, respectively. no removal plants A, B, E-G.	The reviewer agreed with this study's overall quality level.	<u>Blaney (1989)</u>	Medium
Mill Creek, Little Miami River, and Muddy Creek treatment plants		Parameter: 8h TWA in air: 150, 280, 3800, 380 ppb; Parameter: air concentration: 0-20 ppm v/v; Parameter: WW concentration: 0.19-47 μg/L	The reviewer downgraded this study's overall quality rating. They noted: The volatility is reported for 3 sites in open systems.	Dunovant et al. (1986)	Low
Continuous release of chemicals and steady hydrological parameters assumed to develop a steady-state model for estimating concentration in river	Hydrological data and monitoring samples are used to calculate volatilization	Parameter half-life in river: 4-6 d; Perc release from the river is variable with an average value of 0.6%. Mainly removed by volatilization.	The analytical method used for detection of the test substance was not reported.	Brüggemann and Trapp (1988)	Unacceptable
Full scale Wet Air Oxidation (WAO) of solvent still bottoms and general organic waste details are reported by	Wet air oxidation performance data	Parameter: percent removal from test system: 95.35% Effluent concentration solvent still bottoms of	Due to limited information, evaluation of the reasonableness of the study results was not possible.	<u>Matienzo</u> (1989)	Unacceptable

System	Study Type (year)	Results	Comments	Affiliated Reference	Data Quality Evaluation Results of Full Study Report
Radimsky and Shah 1985; parameters used to measure performance were COD, BOD, and DOC; off-gas emissions measured by THC as methane;		tetrachloroethylene = 40 mg/L			
Performance data was collected on full scale batch fractional distillation systems as referenced in the source document	Distillation performance data	Parameter: performance of <u>test system:</u> Mean solvent concentration of distillation residues: tetrachloroethylene 12 reported values, mean concentration = 4; feasible treatment level = 3.0 mg/kg	Due to limited information, evaluation of the reasonableness of the study results was not possible.	<u>Matienzo</u> (1989)	Unacceptable
Samples taken from WWTP to evaluate air stripping and removal of Perc/PCE	Partitioning in activated sludge plant	Parameter: percent removal from WWTP: 78% removal of Perc/ PCE by full scale aeration basin <u>Parameter: gas-liquid</u> phase partition coefficient: avg. 600 (+/-86) and 611 (+/-369) (ng/L)/(μg/L)	Study evaluates removal based on air stripping. The extent of air stripping is a function of the compound p-chem properties and a function of WWTP design and operation.	<u>Parker et al.</u> (1993)	Unacceptable

## EPI Suite<sup>TM</sup> Model Outputs

(<u>U.S. EPA, 2012</u>)

SMILES : C(=C(Cl)Cl)(Cl)Cl CHEM : TETRACHLOROETHENE MOL FOR: C2 CL4 MOL WT : 165.83 ------ EPI SUMMARY (v4.11) ------

Physical Property Inputs: Log Kow (octanol-water): 3.40 Boiling Point (deg C) : 121.30 Melting Point (deg C) : -22.30 Vapor Pressure (mm Hg) : 18.5 Water Solubility (mg/L): 206 Henry LC (atm-m3/mole) : 0.0177

Log Octanol-Water Partition Coef (SRC): Log Kow (KOWWIN v1.69 estimate) = 2.97 Log Kow (Exper. database match) = 3.40 Exper. Ref: HANSCH,C ET AL. (1995)

Boiling Pt, Melting Pt, Vapor Pressure Estimations (MPBPVP v1.43):
Boiling Pt (deg C): 114.28 (Adapted Stein & Brown method)
Melting Pt (deg C): -60.56 (Mean or Weighted MP)
VP(mm Hg, 25 deg C): 17.8 (Mean VP of Antoine & Grain methods)
VP (Pa, 25 deg C): 2.37E+003 (Mean VP of Antoine & Grain methods)
MP (exp database): -22.3 deg C
BP (exp database): 121.3 deg C
VP (exp database): 1.85E+01 mm Hg (2.47E+003 Pa) at 25 deg C

Water Solubility Estimate from Log Kow (WSKOW v1.42): Water Solubility at 25 deg C (mg/L): 134.3 log Kow used: 3.40 (user entered) melt pt used: -22.30 deg C Water Sol (Exper. database match) = 206 mg/L (25 deg C) Exper. Ref: HORVATH,AL ET AL. (1999)

Water Sol Estimate from Fragments: Wat Sol (v1.01 est) = 247.12 mg/L

ECOSAR Class Program (ECOSAR v1.11): Class(es) found: Vinyl/Allyl Halides

Henrys Law Constant (25 deg C) [HENRYWIN v3.20]: Bond Method : 1.65E-002 atm-m3/mole (1.67E+003 Pa-m3/mole) Group Method: 1.77E-002 atm-m3/mole (1.80E+003 Pa-m3/mole) Exper Database: 1.77E-02 atm-m3/mole (1.79E+003 Pa-m3/mole) For Henry LC Comparison Purposes: User-Entered Henry LC: 1.770E-002 atm-m3/mole (1.793E+003 Pa-m3/mole) Henrys LC [via VP/WSol estimate using User-Entered or Estimated values]: HLC: 1.960E-002 atm-m3/mole (1.986E+003 Pa-m3/mole) VP: 18.5 mm Hg (source: User-Entered) WS: 206 mg/L (source: User-Entered)

Log Octanol-Air Partition Coefficient (25 deg C) [KOAWIN v1.10]: Log Kow used: 3.40 (user entered) Log Kaw used: -0.140 (user entered) Log Koa (KOAWIN v1.10 estimate): 3.540 Log Koa (experimental database): 3.480

Probability of Rapid Biodegradation (BIOWIN v4.10): Biowin1 (Linear Model) : 0.2230 Biowin2 (Non-Linear Model) : 0.0012 Expert Survey Biodegradation Results: Biowin3 (Ultimate Survey Model): 2.1400 (months ) Biowin4 (Primary Survey Model) : 3.2060 (weeks ) MITI Biodegradation Probability: Biowin5 (MITI Linear Model) : 0.3626 Biowin6 (MITI Non-Linear Model) : 0.0227 Anaerobic Biodegradation Probability: Biowin7 (Anaerobic Linear Model): 0.7775 Ready Biodegradability Prediction: NO

Hydrocarbon Biodegradation (BioHCwin v1.01): Structure incompatible with current estimation method!

Sorption to aerosols (25 Deg C)[AEROWIN v1.00]: Vapor pressure (liquid/subcooled): 2.47E+003 Pa (18.5 mm Hg) Log Koa (Exp database): 3.480 Kp (particle/gas partition coef. (m3/ug)): Mackay model : 1.22E-009 Octanol/air (Koa) model: 7.41E-010 Fraction sorbed to airborne particulates (phi): Junge-Pankow model : 4.39E-008 Mackay model : 9.73E-008 Octanol/air (Koa) model: 5.93E-008

Atmospheric Oxidation (25 deg C) [AopWin v1.92]: Hydroxyl Radicals Reaction: OVERALL OH Rate Constant = 0.2139 E-12 cm3/molecule-sec Half-Life = 49.998 Days (12-hr day; 1.5E6 OH/cm3) Ozone Reaction: OVERALL Ozone Rate Constant = 0.000073 E-17 cm3/molecule-sec Half-Life = 15660.362 Days (at 7E11 mol/cm3) Fraction sorbed to airborne particulates (phi): 7.06E-008 (Junge-Pankow, Mackay avg) 5.93E-008 (Koa method) Note: the sorbed fraction may be resistant to atmospheric oxidation Soil Adsorption Coefficient (KOCWIN v2.00): Koc : 94.94 L/kg (MCI method) Log Koc: 1.977 (MCI method) Koc : 892.2 L/kg (Kow method) Log Koc: 2.950 (Kow method) Experimental Log Koc: 2.4 (database)

Aqueous Base/Acid-Catalyzed Hydrolysis (25 deg C) [HYDROWIN v2.00]: Rate constants can NOT be estimated for this structure!

Bioaccumulation Estimates (BCFBAF v3.01): Log BCF from regression-based method = 1.910 (BCF = 81.34 L/kg wet-wt) Log Biotransformation Half-life (HL) = 0.5257 days (HL = 3.355 days) Log BCF Arnot-Gobas method (upper trophic) = 1.663 (BCF = 46.04) Log BAF Arnot-Gobas method (upper trophic) = 1.663 (BAF = 46.04) log Kow used: 3.40 (user entered)

Volatilization from Water: Henry LC: 0.0177 atm-m3/mole (entered by user) Half-Life from Model River: 1.357 hours Half-Life from Model Lake : 122.8 hours (5.116 days)

Removal in Wastewater Treatment: Total removal: 87.91 percent Total biodegradation: 0.05 percent Total sludge adsorption: 6.32 percent Total to Air: 81.53 percent (using 10000 hr Bio P,A,S)

Removal in Wastewater Treatment: Total removal: 88.20 percent Total biodegradation: 2.12 percent Total sludge adsorption: 6.29 percent Total to Air: 79.79 percent (using Biowin/EPA draft method)

Level III Fugacity Model: (MCI Method) Mass Amount Half-Life Emissions (percent) (hr) (kg/hr) Air 46 1.6e+003 1000 Water 46.8 1.44e+003 1000 Soil 6.87 2.88e+003 1000 Sediment 0.314 1.3e+004 0 Persistence Time: 181 hr

Level III Fugacity Model: (MCI Method with Water percents) Mass Amount Half-Life Emissions (percent) (hr) (kg/hr) Air 46 1.6e+003 1000 Water 46.8 1.44e+003 1000 water (46.8) biota (0.00587) suspended sediment (0.00666) Soil 6.87 2.88e+003 1000 Sediment 0.314 1.3e+004 0 Persistence Time: 181 hr

Level III Fugacity Model: (EQC Default) Mass Amount Half-Life Emissions (percent) (hr) (kg/hr) Air 29.1 1.6e+003 1000 Water 30.4 1.44e+003 1000 water (30.3) biota (0.00381) suspended sediment (0.0469) Soil 38.4 2.88e+003 1000 Sediment 2.03 1.3e+004 0 Persistence Time: 279 hr

## References

- Balsiger, C; Holliger, C; Höhener, P. (2005). Reductive dechlorination of chlorofluorocarbons and hydrochlorofluorocarbons in sewage sludge and aquifer sediment microcosms. Chemosphere 61: 361-373. http://dx.doi.org/10.1016/j.chemosphere.2005.02.087
- Barrows, ME; Petrocelli, SR; Macek, KJ; Carroll, JJ. (1980). Bioconcentration and elimination of selected water pollutants by bluegill sunfish (Lepomis macrochirus). In R Haque (Ed.), Dynamics, Exposure and Hazard Assessment of Toxic Chemicals (pp. 379-392). Ann Arbor, MI: Ann Arbor Science.
- Bell, J; Melcer, H; Monteith, H; Osinga, I; Steel, P. (1993). Stripping of volatile organic compounds at full-scale municipal wastewater treatment plants. Water Environ Res 65: 708-716. http://dx.doi.org/10.2175/WER.65.6.2
- Blaney, BL. (1989). Applicability of steam stripping to organics removal from wastewater streams. In Third International Conference on New Frontiers for Hazardous Waste Management: Proceedings (pp. 415-424). (EPA/600/9-89/072). Cincinnati, OH: U.S. Environmental Protection Agency. https://nepis.epa.gov/Exe/ZyPURL.cgi?Dockey=30005CR3.txt
- Bouwer, EJ; Mccarty, PL. (1982). Removal of trace chlorinated organic compounds by activated carbon and fixedfilm bacteria. Environ Sci Technol 16: 836–843. <u>http://dx.doi.org/10.1021/es00106a003</u>
- Bouwer, EJ; Mccarty, PL. (1983). Transformations of 1- and 2-carbon halogenated aliphatic organic compounds under methanogenic conditions. Appl Environ Microbiol 45: 1286-1294.
- Bouwer, EJ; Rittmann, BE; McCarty, PL. (1981). Anaerobic degradation of halogenated 1- and 2-carbon organic compounds. Environ Sci Technol 15: 596-599. <u>http://dx.doi.org/10.1021/es00087a012</u>
- Brüggemann, R; Trapp, S. (1988). Release and fate modelling of highly volatile solvents in the river Main. Chemosphere 17: 2029-2041. <u>http://dx.doi.org/10.1016/0045-6535(88)90013-6</u>
- Cabirol, N; Perrier, J; Jacob, F; Fouillet, B; Chambon, P. (1996). Role of methanogenic and sulfate-reducing bacteria in the reductive dechlorination of tetrachloroethylene in mixed culture. Bull Environ Contam Toxicol 56: 817-824. http://dx.doi.org/10.1007/s001289900119
- <u>Chen, WH; Yang, WB; Yuan, CS; Yang, JC; Zhao, QL.</u> (2014). Fates of chlorinated volatile organic compounds in aerobic biological treatment processes: the effects of aeration and sludge addition. Chemosphere 103: 92-98. <u>http://dx.doi.org/10.1016/j.chemosphere.2013.11.039</u>
- Cheng, D; Chow, WL; He, J. (2010). A Dehalococcoides-containing co-culture that dechlorinates tetrachloroethene to trans-1,2-dichloroethene. ISME J 4: 88-97. <u>http://dx.doi.org/10.1038/ismej.2009.90</u>
- Chiou, CT; Freed, VH; Peters, LJ; Kohnert, RL. (1980). Evaporation of solutes from water. Environ Int 3: 231-236. http://dx.doi.org/10.1016/0160-4120(80)90123-3
- <u>Chodola, GR; Biswas, N; Bewtra, JK; St. Pierre, CC; Zytner, RG.</u> (1989). Fate of selected volatile organic substances in aqueous environment. Water Pollut Res J Can 24: 119-142.
- <u>Cichocka, D; Nikolausz, M; Haest, PJ; Nijenhuis, I.</u> (2010). Tetrachloroethene conversion to ethene by a Dehalococcoides-containing enrichment culture from Bitterfeld. FEMS Microbiol Ecol 72: 297-310. <u>http://dx.doi.org/10.1111/j.1574-6941.2010.00845.x</u>
- de Bruin, WP; Kotterman, MJ; Posthumus, MA; Schraa, G; Zehnder, AJ. (1992). Complete biological reductive transformation of tetrachloroethene to ethane. Appl Environ Microbiol 58: 1996-2000.
- Deipser, A: Stegmann, R. (1997). Biological degradation of VCCs and CFCs under simulated anaerobic landfill conditions in laboratory test digesters. Environ Sci Pollut Res Int 4: 209-216. http://dx.doi.org/10.1007/BF02986348
- Dickson, AG; Riley, JP. (1976). The distribution of short-chain halogenated aliphatic hydrocarbons in some marine organisms. Mar Pollut Bull 7: 167-169. <u>http://dx.doi.org/10.1016/0025-326X(76)90212-5</u>
- Dilling, WL. (1977). Interphase transfer processes. II. Evaporation rates of chloro methanes, ethanes, ethylenes, propanes, and propylenes from dilute aqueous solutions. Comparisons with theoretical predictions. Environ Sci Technol 11: 405-409. <u>http://dx.doi.org/10.1021/es60127a009</u>
- Dilling, WL; Tefertiller, NB; Kallos, GJ. (1975). Evaporation rates and reactivities of methylene chloride, chloroform, 1,1,1-trichloroethane, trichloroethylene, tetrachloroethylene, and other chlorinated compounds in dilute aqueous solutions. Environ Sci Technol 9: 833-838. <u>http://dx.doi.org/10.1021/es60107a008</u>
- DiStefano, TD; Gossett, JM; Zinder, SH. (1991). Reductive dechlorination of high concentrations of tetrachloroethene to ethene by an anaerobic enrichment culture in the absence of methanogenesis. Appl Environ Microbiol 57: 2287-2292.

- DiStefano, TD; Gossett, JM; Zinder, SH. (1992). Hydrogen as an electron donor for dechlorination of tetrachloroethene by an anaerobic mixed culture. Appl Environ Microbiol 58: 3622-3629.
- Dobbs, RA; Wang, L; Govind, R. (1989). Sorption of toxic organic compounds on wastewater solids: Correlation with fundamental properties. Environ Sci Technol 23: 1092-1097. <u>http://dx.doi.org/10.1021/es00067a004</u>
- Doong, RA; Wu, SC. (1992). Reductive dechlorination of chlorinated hydrocarbons in aqueous solutions containing ferrous and sulfide ions. Chemosphere 24: 1063-1075. <u>http://dx.doi.org/10.1016/0045-6535(92)90197-Y</u>
- Dow Chem Co. (1977). THE INHIBITION OF ANAEROBIC SLUDGE GAS PRODUCTION BY 1,1,1-TRICHLOROETHANE, METHYLENE CHLORIDE, TRICHLOROETHYLENE AND PERCHLOROETHYLENE, Part 2. (OTS: OTS0517178; 8EHQ Num: NA; DCN: 86-870002089; TSCATS RefID: 309930; CIS: NA).
- Dow Chemical (Dow Chemical Company). (1973). Uptake, clearance and bioconcentration of dow-per (perchloroethylene) in rainbow trout, Salmo gairdneri richardson. (8EHQ Num: NA; DCN: 86-870002077; TSCATS RefID: 309906; CIS: NA).
- Dow Chemical (Dow Chemical Company). (1980). Introductory study of the biodegradation of the chlorinated methane, ethane and ethene compounds: Progress report CR806890-01 coop agreement [TSCA Submission]. (Experimental Design No. 12. OTS0509177. 40-8024098. 47004 F1-2A. TSCATS/200511). https://ntrl.ntis.gov/NTRL/dashboard/searchResults/titleDetail/OTS0509177.xhtml
- Drzyzga, O; El Mamouni, R; Agathos, SN; Gottschal, JC. (2002). Dehalogenation of chlorinated ethenes and immobilization of nickel in anaerobic sediment columns under sulfidogenic conditions. Environ Sci Technol 36: 2630-2635. <u>http://dx.doi.org/10.1021/es010184x</u>
- Dunovant, VS; Clark, CS; Que Hee, SS; Hertzberg, VS; Trapp, JH. (1986). Volatile organics in the wastewater and airspaces of three wastewater treatment plants. J Water Pollut Control Fed 58: 886-895.
- ECHA (European Chemicals Agency). (2017a). Adsorption/desorption: Tetrachloroethylene. Helsinki, Finland. Retrieved from <u>https://echa.europa.eu/registration-dossier/-/registered-dossier/14303/5/5/2#</u>
- ECHA (European Chemicals Agency). (2017b). Biodegradation in water: screening tests: Tetrachloroethylene. Helsinki, Finland. Retrieved from <u>https://echa.europa.eu/registration-dossier/-/registered-dossier/14303/5/3/2</u>
- Edwards, EA; Liang, LN; Dunia, GG. (1992). Anaerobic microbial transformation of aromatic hydrocarbons and mixtures of aromatic hydrocarbons and halogenated solvents. (CE319). Arlington, VA: Air Force Office of Scientific Research. https://ntrl.ntis.gov/NTRL/dashboard/searchResults.xhtml?searchQuery=ADA260498
- Farrell, J; Reinhard, M. (1994). Desorption of halogenated organics from model solids, sediments, and soil under unsaturated conditions. 1. Isotherms. Environ Sci Technol 28: 53-62. <u>http://dx.doi.org/10.1021/es00050a009</u>
- Fathepure, BZ; Boyd, SA. (1988). Dependence of tetrachloroethylene dechlorination on methanogenic substrate consumption by Methanosarcina sp. strain DCM. Appl Environ Microbiol 54: 2976-2980.
- Freedman, DL; Gossett, JM. (1989). Biological reductive dechlorination of tetrachloroethylene and trichloroethylene to ethylene under methanogenic conditions. Appl Environ Microbiol 55: 2144-2151.
- Freitag, D; Ballhorn, L; Geyer, H; Korte, F. (1985). Environmental hazard profile of organic chemicals: an experimental method for the assessment of the behaviour of organic chemicals in the ecoshpere by means of simple laboratory tests with 14C labelled chemicals. Chemosphere 14: 1589-1616.
- <u>Gossett, JM.</u> (1985). Anaerobic degradation of C1 and C2 chlorinated hydrocarbons. (ESL-TR-85-38). Tyndal AFB, FL: Air Force Engineering & Services Center. <u>https://apps.dtic.mil/docs/citations/ADA165005</u>
- Haas, JR; Shock, EL. (1999). Halocarbons in the environment: Estimates of thermodynamic properties for aqueous chloroethylene species and their stabilities in natural settings. Geochim Cosmo Act 63: 3429-3441. http://dx.doi.org/10.1016/S0016-7037(99)00276-8
- Haston, ZC; Mccarty, PL. (1999). Chlorinated ethene half-velocity coefficients (KS) for reductive dehalogenation. Environ Sci Technol 33: 223-226. <u>http://dx.doi.org/10.1021/es9805876</u>
- He, Z; Yang, G; Lu, X; Zhang, H. (2013). Distributions and sea-to-air fluxes of chloroform, trichloroethylene, tetrachloroethylene, chlorodibromomethane and bromoform in the Yellow Sea and the East China Sea during spring. Environ Pollut 177: 28-37. <u>http://dx.doi.org/10.1016/j.envpol.2013.02.008</u>
- Isalou, M; Sleep, BE; Liss, SN. (1998). Biodegradation of high concentrations of tetrachloroethene in a continuous flow column system. Environ Sci Technol 32: 3579-3585.
- Jeffers, PM; Ward, LM; Woytowitch, LM; Wolfe, NL. (1989). Homogeneous hydrolysis rate constants for selected chlorinated methanes, ethanes, ethanes, and propanes. Environ Sci Technol 23: 965-969. http://dx.doi.org/10.1021/es00066a006

Jensen, S; Rosenberg, R. (1975). Degradability of some chlorinated aliphatic hydrocarbons in sea water and sterilized water. Water Res 9: 659-661.

- <u>Kästner, M.</u> (1991). Reductive dechlorination of tri- and tetrachloroethylenes depends on transition from aerobic to anaerobic conditions. Appl Environ Microbiol 57: 2039-2046.
- Kawasaki, M. (1980). Experiences with the test scheme under the chemical control law of Japan: An approach to structure-activity correlations. Ecotoxicol Environ Saf 4: 444-454. <u>http://dx.doi.org/10.1016/0147-6513(80)90046-9</u>
- Keefe, SH; Barber, LB; Runkel, RL; Ryan, JN. (2004). Fate of volatile organic compounds in constructed wastewater treatment wetlands. Environ Sci Technol 38: 2209-2216. http://dx.doi.org/10.1021/es034661i
- Kim, Y; Arp, DJ; Semprini, L. (2000). Chlorinated solvent cometabolism by butane-grown mixed culture. J Environ Eng 126: 934-942. <u>http://dx.doi.org/10.1061/(ASCE)0733-9372(2000)126:10(934</u>)
- Krumholz, LR; Sharp, R; Fishbain, SS. (1996). A freshwater anaerobe coupling acetate oxidation to tetrachloroethylene dehalogenation. Appl Environ Microbiol 62: 4108-4113.
- Leahy, JG; Shreve, GS. (2000). The effect of organic carbon on the sequential reductive dehalogenation of tetrachloroethylene in landfill leachates. Water Res 34: 2390-2396. <u>http://dx.doi.org/10.1016/S0043-1354(99)00389-9</u>
- Lee, W; Park, SH; Kim, J; Jung, JY. (2015). Occurrence and removal of hazardous chemicals and toxic metals in 27 industrial wastewater treatment plants in Korea. Desalination Water Treat 54: 1141-1149. http://dx.doi.org/10.1080/19443994.2014.935810
- Li, J; Werth, CJ. (2004). Slow desorption mechanisms of volatile organic chemical mixtures in soil and sediment micropores. Environ Sci Technol 38: 440-448. <u>http://dx.doi.org/10.1021/es034830z</u>
- Long, JL; Stensel, HD; Ferguson, JF; Strand, SE; Ongerth, JE. (1993). Anaerobic and aerobic treatment of chlorinated aliphatic compounds. J Environ Eng 119: 300-320. <u>http://dx.doi.org/10.1061/(ASCE)0733-9372(1993)119:2(300)</u>
- Lu, C; Bjerg, PL; Zhang, F; Broholm, MM. (2011). Sorption of chlorinated solvents and degradation products on natural clayey tills. Chemosphere 83: 1467-1474. <u>http://dx.doi.org/10.1016/j.chemosphere.2011.03.007</u>
- Marco-Urrea, E; Gabarrell, X; Sarra, M; Caminal, G; Vicent, T; Reddy, CA. (2006). Novel aerobic perchloroethylene degradation by the white-rot fungus Trametes versicolor. Environ Sci Technol 40: 7796-7802. http://dx.doi.org/10.1021/es0622958
- Matienzo, LV. (1989). Staff report on development of treatment standards for non-RCRA solvent waste. Sacramento, CA: Toxic Substances Control Program. <u>http://infohouse.p2ric.org/ref/17/16884.pdf</u>
- Namkung, E; Rittmann, BE. (1987). Estimating volatile organic compound emissions from publicly owned treatment works (pp. 670-678). (NIOSH/00172323). Namkung, E; Rittmann, BE.
- <u>Neely, WB; Branson, DR; Blau, GE.</u> (1974). Partition coefficient to measure bioconcentration potential of organic chemicals in fish. Environ Sci Technol 8: 1113-1115. <u>http://dx.doi.org/10.1021/es60098a008</u>
- Parker, WJ; Thompson, DJ; Bell, JP; Melcer, H. (1993). Fate of volatile organic compounds in municipal activated sludge plants. Water Environ Res 65: 58-65.
- Parsons, F; Lage, GB; Rice, R. (1985). Biotransformation of chlorinated organic solvents in static microcosms. Environ Toxicol Chem 4: 739-742. <u>http://dx.doi.org/10.1002/etc.5620040604</u>
- Parsons, F; Wood, PR; Demarco, J. (1984). Transformations of tetrachloroethene and trichloroethene in microcosms and groundwater. J Am Water Works Assoc 762: 56-59. <u>http://dx.doi.org/10.1002/j.1551-</u> 8833.1984.tb05282.x
- Pearson, CR; McConnell, G. (1975). Chlorinated C1 and C2 hydrocarbons in the marine environment. Proc Biol Sci 189: 305-332. <u>http://dx.doi.org/10.1098/rspb.1975.0059</u>
- Rodriguez, C; Linge, K; Blair, P; Busetti, F; Devine, B; Van Buynder, P; Weinstein, P; Cook, A. (2012). Recycled water: potential health risks from volatile organic compounds and use of 1,4-dichlorobenzene as treatment performance indicator. Water Res 46: 93-106. <u>http://dx.doi.org/10.1016/j.watres.2011.10.032</u>
- Roose, P; Dewulf, J; Brinkman, UAT; Van Langenhove, H. (2001). Measurement of volatile organic compounds in sediments of the Scheldt Estuary and the Southern North Sea. Water Res 35: 1478-1488. http://dx.doi.org/10.1016/S0043-1354(00)00410-3
- Ryoo, D; Shim, H; Canada, K; Barbieri, P; Wood, TK. (2000). Aerobic degradation of tetrachloroethylene by toluene-o-xylene monooxygenase of Pseudomonas stutzeri OX1. Nat Biotechnol 18: 775–778.
- Saisho, K; Hasegawa, Y; Saeki, M; Toyoda, M; Saito, Y. (1994). [Bioaccumulation of volatile chlorinated hydrocarbons in blue mussel, Mytilus edulis and killifish, Oryzias latipes]. Jpn J Toxicol Environ Health 40: 274-278. <u>http://dx.doi.org/10.1248/jhs1956.40.274</u>

- <u>Schreier, CG; Reinhard, M.</u> (1994). Transformation of chlorinated organic compounds by iron and manganese powders in buffered water and in landfill leachate. Chemosphere 29: 1743-1753. <u>http://dx.doi.org/10.1016/0045-6535(94)90320-4</u>
- Shirayama, H; Tohezo, Y; Taguchi, S. (2001). Photodegradation of chlorinated hydrocarbons in the presence and absence of dissolved oxygen in water. Water Res 35: 1941-1950. <u>http://dx.doi.org/10.1016/S0043-1354(00)00480-2</u>
- Smith, JH; Bomberger, DC, Jr; Haynes, DL. (1980). Prediction of the volatilization rates of high-volatility chemicals from natural water bodies. Environ Sci Technol 14: 1332-1337. http://dx.doi.org/10.1021/es60171a004
- Sorial, GA; Papadimas, SP; Suidan, MT; Speth, TF. (1994). Competitive adsorption of VOCs and BOM: Oxic and anoxic environments. Water Res 28: 1907-1919. <u>http://dx.doi.org/10.1016/0043-1354(94)90166-X</u>
- Stubin, AI; Brosnan, TM; Porter, KD; Jimenez, L; Lochan, H. (1996). Organic priority pollutants in New York City municipal wastewaters: 1989-1993. Water Environ Res 68: 1037-1044. http://dx.doi.org/10.2175/106143096X128108
- Tabak, HH; Quave, SA; Mashni, CI; Barth, EF. (1981). Biodegradability studies with organic priority pollutant compounds. J Water Pollut Control Fed 53: 1503-1518.
- Tancrede, M; Yanagisawa, Y; Wilson, R. (1992). Volatilization of volatile organic compounds from showers: I. Analytical method and quantitative assessment. Atmos Environ A 26: 1103-1111. http://dx.doi.org/10.1016/0960-1686(92)90042-J
- U.S. EPA (U.S. Environmental Protection Agency). (1987). Atmospheric persistence of eight air toxics [EPA Report]. (EPA-600/3-87/004). Research Triangle Park, NC.
- U.S. EPA (U.S. Environmental Protection Agency). (2012). Estimation Programs Interface Suite<sup>™</sup> for Microsoft® Windows, v 4.11. Washington, DC. <u>https://www.epa.gov/tsca-screening-tools/epi-suitetm-estimation-program-interface</u>
- van Eekert, MHA; Schröder, TJ; van Rhee, A; Stams, AJM; Schraa, G; Field, JA. (2001). Constitutive dechlorination of chlorinated ethenes by a methanol degrading methanogenic consortium. Bioresour Technol 77: 163-170. http://dx.doi.org/10.1016/S0960-8524(00)00149-8
- Vogel, TM; Mccarty, PL. (1985). Biotransformation of tetrachloroethylene to trichloroethylene, dichloroethylene, vinyl chloride, and carbon dioxide under methanogenic conditions. Appl Environ Microbiol 49: 1080-1083.
- Wakeham, SG; Davis, AC; Karas, JA. (1983). Mesocosm experiments to determine the fate and persistence of volatile organic compounds in coastal seawater. Environ Sci Technol 17: 611-617. http://dx.doi.org/10.1021/es00116a009
- Wang, G; Allen-King, RM; Choung, S; Feenstra, S; Watson, R; Kominek, M. (2013). A practical measurement strategy to estimate nonlinear chlorinated solvent sorption in low foc sediments. Ground Water Monit Remediat 33: 87-96. <u>http://dx.doi.org/10.1111/j.1745-6592.2012.01413.x</u>
- Wang, X; Harada, S; Watanabe, M; Koshikawa, H; Sato, K; Kimura, T. (1996). Determination of bioconcentration potential of tetrachloroethylene in marine algae by 13C. Chemosphere 33: 865-877. http://dx.doi.org/10.1016/0045-6535(96)00230-5
- Wood, PR; Parsons, FZ; DeMarco, J; Harween, HJ; Lang, RF; Payan, IL; Ruiz, MC. (1981). Introductory study of the biodegradation of the chlorinated methane, ethane and ethene compounds. Paper presented at American Water Works Association Annual Conference and Exposition, June 7-11, 1981, St. Louis, MO.